

**Stereoselective Synthesis of 2-Deoxy-2-Iodo-*O*-Glycoside
in Biphasic Medium and BDMS Mediated Facile Access
to 2,3-Unsaturated-*O*-Glycoside, Imidazopyridine and
 α -Aminoamidine**

*A Dissertation Submitted to the
Indian Institute of Technology Guwahati
As Partial Fulfillment for the Degree of*

DOCTOR OF PHILOSOPHY



by

R. Sidick Basha
Roll No. 09612214

**Department of Chemistry
Indian Institute of Technology Guwahati
Guwahati-781 039
July 2014**



*The Unique Challenge which Chemical Synthesis provides for the Creative Imagination
and the skilled hands ensure that it will endure as long as men Write books, Paint pictures,
and Fashion things which are beautiful, or practical, or both*

R. B. Woodward



INDIAN INSTITUTE OF TECHNOLOGY, GUWAHATI

Department of Chemistry

STATEMENT

I do hereby declare that the matter embodied in this thesis entitled “*Stereoselective Synthesis of 2-Deoxy-2-Iodo-O-Glycoside in Biphasic Medium and BDMS Mediated Facile Access to 2,3-Unsaturated-O-Glycoside, Imidazopyridine and α -Aminoamidine*” is the result of investigations carried out by me under the supervision of Prof. Abu T. Khan in the Department of Chemistry, Indian Institute of Technology Guwahati, India.

In keeping with the general practice of reporting scientific observations, due acknowledgements have been made wherever the work described is based on the findings of other investigators.

IIT Guwahati
August 08, 2014

R. Sidick Basha



Indian Institute of Technology Guwahati

Guwahati – 781039, India

Tel. No.: +91-361-2582305

Fax No.: +91-361-2582349

e-mail: atk@iitg.ernet.in

Dr. Abu T. Khan

Professor of Chemistry

CERTIFICATE

This is to certify that Mr. R. Sidick Basha has been working in my research group since January, 2011 as a regular registered Ph. D. student. I am forwarding his thesis entitled “*Stereoselective Synthesis of 2-Deoxy-2-Iodo-O-Glycoside in Biphasic Medium and BDMS Mediated Facile Access to 2,3-Unsaturated-O-Glycoside, Imidazopyridine and α -Amino amidine*” being submitted for the Ph. D. (Science) Degree of this Institute. I certify that he has fulfilled all the requirements according to the rules of this Institute regarding the investigations embodied in his thesis and this work has not been submitted elsewhere for a degree.

IIT Guwahati
August 08, 2014

Prof. A. T. Khan
(Thesis Supervisor)



Indian Institute of Technology Guwahati

Guwahati – 781039, India
Tel. No.: +91-361-2582307
Fax No.: +91-361-2582349
e-mail: patel@iitg.ernet.in

Dr. Bhisma K. Patel
Professor and Head
Department of Chemistry

CERTIFICATE

This is to certify that Mr. R. Sidick Basha has completed his Ph. D. Thesis work from 01 January, 2011 as a regular registered Ph. D. student under my colleague Prof. Abu T. Khan. I have been appointed as a Co-Supervisor when Prof. Khan joined as Vice-Chancellor of Aliah University in West Bengal on deputation from IIT Guwahati. I am forwarding his thesis as a Co-supervisor entitled “*Stereoselective Synthesis of 2-Deoxy-2-Iodo-O-Glycoside in Biphasic Medium and BDMS Mediated Facile Access to 2,3-Unsaturated-O-Glycoside, Imidazopyridine and α -Amino amidine*” being submitted for the Ph. D. (Science) Degree of this Institute. I also certify that he has fulfilled all the requirements according to the rules of this Institute regarding the investigations embodied in his thesis and this work has not been submitted elsewhere for a degree.

IIT Guwahati
August 08, 2014

Prof. B. K. Patel
(Thesis Co-Supervisor)

Acknowledgement

I would like to express my deepest gratitude to my research supervisor Professor Abu T. Khan for his valuable suggestion, perceptive guidance and decisive insights during the entire course of Ph.D. work. His Moral scientific spirit and ever-friendly nature has uplift me a lot during my research work which enrich my ideas. I am extremely thankful to him for giving me a liberty to pursue my own ideas and conceptual quest of knowledge toward research. I find myself privileged to have worked under his kind guidance. My endless gratitude goes towards him. I will always be grateful to my Co-supervisor Prof. B. K. Patel for providing me all sort of necessary vision and eternal support during absence of my Thesis Supervisor. I wish to convey my sincere gratitude to Prof. T. Punniyamurthy Chairman of Doctoral Committee, and other members Dr. B. Mandal, and Prof. V. V. Dasu for their intellectual support, encouragement and helpful suggestions. My genuine regards to all the faculty of the Department of Chemistry for their encouragement and inspiration.

I am grateful to Council of Scientific and Industrial Research (CSIR), India for financial support and IIT Guwahati for all the facilities that were made available to me. I extend my gratitude to Central Instrument Facility, IIT Guwahati for Instrument facility and DST for X-ray facility.

I wish to acknowledge Dr. Md. Musawwer Khan, Dr. Mohan Lal, Dr. Deb K. Das, Prasanta, Ajaz, Arindam, Satavisha, Kobirul, Suchandra, Belal, Karuna, Gattu and my other labmates for their direct and indirect help. I express my profound gratefulness to all friends from IITG for their help and encouragement. Special thanks to Dr. Babulal Das regard XRD data collection.

It is a great pleasure to express my gratitude to our staff of CIF Mr. Chandan Borgohain, Mr. Kula Kamal Senapati and Mr. Kh. Kesho Singh for their assistance and co-operation. I express my heartiest thanks to all the research scholars, Department of Chemistry, IIT Guwahati for their help and endless support.

I am grateful to Dr. Aiyagari Ramesh, Department of Biotechnology, IIT Guwahati and Suman Khan, Department of Chemical Science, Gauhati University for their necessary dissertation corrections.

I owe my sincere thanks to our technical staff Abhilasha, Lipika, Parikshit, Aniruddha, Diganta, Shyamal and office staff Nilyotpal, Imdadul, Santanu for their diplomatic help and professional support.

I am extremely grateful for the everlasting support, love and blessings from my Parents, who have always motivated me in all stages of my life.

R. Sidick Basha

Contents of the Thesis

Chapter I	Chapter II	Chapter III	Chapter IV	Chapter V	Chapter VI
● Chapter I	General Introduction to <i>O</i>-glycosylation of Sugars				1-16
●	1.1 Introduction				1-2
	1.2 Synthetic Strategies and Development of <i>O</i> -Glycosylation on Glycal Assembly				2-5
	1.3 Synthetic Strategies and Development of Iodoglycosylation				5-11
	1.4 Ferrier rearrangement Importance				11
	1.4a Synthetic Strategies and Development of Ferrier rearrangement on glucal				11-16
● Chapter II	Aqueous Biphasic Mediated Stereoselective Synthesis of 2-deoxy-2-iodo-<i>O</i>-glycoside through <i>in situ</i> generated Iodonium ion Using Oxobis(acetylacetonato)vanadium				17-56
	Results and discussion				17-30
	Experimental				31-56
● Chapter III	Bromodimethylsulfonium bromide (BDMS) Catalyzed Synthesis of 2,3-unsaturated-<i>O</i>-glycosides via Ferrier rearrangement				57-75
	Results and discussion				57-63
	Experimental				64-75
	References (Chapter I-III)				76-81
● Chapter IV	Brief Literature Survey on Isocyanide-based Multicomponent Reactions				82-96
●	1.1 Introduction				82-83
	1.2 Recent Development of MCRs				83-84
	1.3 Isocyanide Based Multicomponent reactions				84
	1.3a Synthetic Utility of Three-Component IMCRs				85-86
	1.3b Synthetic Utility of Four-Component IMCRs				87-88
	1.4 Imidazo[1,2- <i>a</i>]pyridine Importance				88

	1.4a Synthetic utility of Imidazo[1,2- <i>a</i>]pyridines	89-93
	1.5 Amidine & Aminoamidine Importance	93-94
	1.5a Amino Amidines Synthetic Methods	94-96
● Chapter V	Synthesis of Imidazo[1,2-<i>a</i>]pyridines in One-pot three component reaction using Bromodimethylsulfonium bromide (BDMS) as a Catalyst	97-114
	Results and discussion	97-103
	Experimental	104-114
● Chapter VI	AB²C type UGI reaction for the Synthesis of α-amino Amidines in presence of Bromodimethylsulfonium bromide (BDMS)	115-135
	Results and discussion	115-123
	Experimental	124-135
	References (Chapter 4-6)	136-141
□ Appendix	Conclusion and Future perspectives	142-143
	List of author's publications	144-146

Abbreviations

Ac	acetyl
AIBN	azobisisobutyronitrile
BDMS	bromodimethylsulfonium bromide
Bn	benzyl
Boc	<i>tert</i> -butoxycarbonyl
Bu	butyl
<i>t</i> -Bu	<i>tert</i> -Butyl
Bz	benzoyl
CAN	cerium(IV)ammonium nitrate
Cbz	benzyloxycarbonyl
CCDC	Cambridge Crystallographic Data Centre
COSY	correlation spectroscopy
DCE	1,1-dichloroethane
DCM	dichloromethane
DMAP	<i>N,N</i> -4-dimethylaminopyridine
DME	dimethyl ether
DMF	<i>N,N</i> -dimethylformamide
DMSO	dimethylsulfoxide
EDCI	1-ethyl-3-(3-dimethylaminopropyl)carbodiimide
Et	ethyl
HFIP	hexafluoroisopropanol
IR	infrared
MCR	multicomponent reaction
Me	methyl
Mp	melting point
MS	molecular sieves
MW	microwave
NMR	nuclear magnetic resonance
NOESY	Nuclear Overhauser Enhancement Spectroscopy
<i>i</i> -Pr	isopropyl
Ph	phenyl

ppm	parts per million
Pr	propyl
<i>p</i> -TSA	<i>p</i> -toluenesulfonic acid
Py	pyridine
rt	room temperature
TBS	<i>tert</i> -butyldimethylsilyl
Tf	trifluoromethanesulfonyl
TFA	trifluoroacetic acid
THF	tetrahydrofuran
TLC	thin layer chromatography
TMS	trimethylsilyl
TMSOTf	trimethylsilyl trifluoromethanesulfonate



*General
Introduction to
O-glycosylation of
Sugars*

1.1 Introduction

Carbohydrates play an essential role in living organism¹ and support a diverse array of biological functions. They are involved in storage of energy from natural resource of glycogen and starch, which lead to reliable materials as cellulose and proteoglycans.²⁻⁴ Moreover, they reside in every cell surface in form of glycolipids and glycoproteins. They also acts as glycode, depends on α or β glycosyl unit assembly. Carbohydrates possess a significant role in cell-cell interactions and cell communication for various biomedical investigations.⁵⁻⁷ They also act as ligands for numerous regulatory processes.^{8,9} A classical polysaccharide unit heparin¹⁰ is found to be a potent anticoagulant. Carbohydrate-based drugs are found to be the most important pharmacophores due to their lack of toxicity and minimal immunogenicity. Carbohydrate-protein interactions in neoglycoconjugates help to analyze the functions of the glycan for targeted diagnostic medicines. Numerous methods have been developed for targeted glycovaccines and glycoconjugates. Carbohydrates such as glycans are also involved in many biological processes¹¹ such as cell and tumor growth, fertilization and tumor metastasis.

Carbohydrates interact with proteins through *O*-glycoside linkage and alter the biological process of conjugate in terms of stability and activity toward enzymes. It also governs the protein three-dimension orientation.¹² Oligosaccharides associate in many biological functions such as blood group determination, binding ability of lectins and hormone regulation.¹³⁻¹⁵ Indeed, synthetic oligosaccharides exist as drugs in pharmaceuticals, and are usually achieved through glycosylation methods. The challenging task for synthetic chemist is to attain a lucid synthetic pathway for complex carbohydrate molecules. The significant problem in glycosylation is stereochemistry control at the anomeric centre (α or β configuration) followed by regioselective glycosylation using orthogonal protecting group. There are eight different classical methods used for glycosylation which is shown in Figure 1. Among them glycosylation on glucal assembly is highly desirable due to their diverse synthetic utility for the construction of natural products,¹⁶ oligosaccharide¹⁷ and *C*-glycosides synthesis.¹⁸ In 1873, Michael¹⁹ first synthesized a phenyl glucoside using acetobromoglucose and sodium phenoxide. Later on, Koenigs and Knorr²⁰ first demonstrated a silver based promoter²¹ for controlled glycosylation from glycosyl

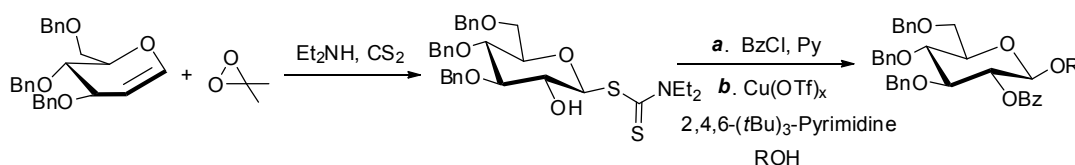
halides. In the last decade, rapid growth on glycosylation synthesis has been achieved. However, till date there is no general method available for *O*-glycosylation.



Figure 1. Methods used for glycosylation

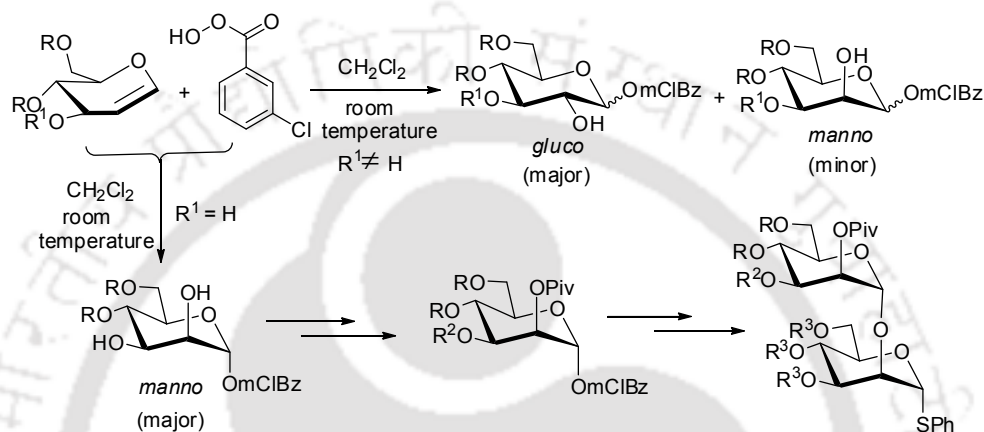
➤ 1.2 Synthetic Strategies and Development of *O*-Glycosylation on Glycal Assembly

Wei *et al.*²² developed the stereoselective synthesis of β -*O*-glycoside from 3,4,6-tri-*O*-benzyl-*D*-glucal through *O*-glycosylation. In this methodology epoxyglycal reacts with diethylamine and carbon disulfide (CS_2) leading to *in situ* generation of glycosyl dithiocarbamates, which subsequently triggered by copper triflate along with acid scavenger 2,4,6-*(t*Bu)₃-pyrimidine forms β -*O*-glycoside as shown in Scheme 1.



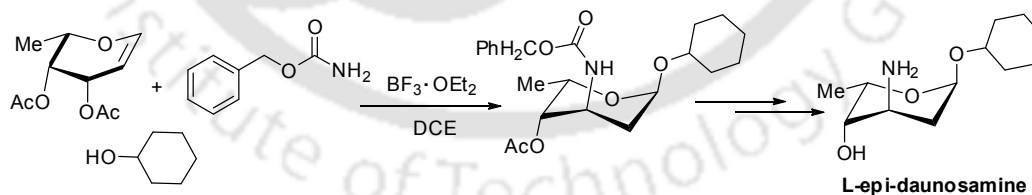
Scheme 1

Castillon *et al.*²³ reported that fully protected glucal react with MCPBA in dichloromethane at room temperature which attain a stereoselective glucopyranoside derivative. However, in case of 4,6-Di-O-protected glucal it led to mannopyranoside derivative. The corresponding mannopyranoside were further extended for the synthesis of complex disaccharide unit which is represented in Scheme 2.



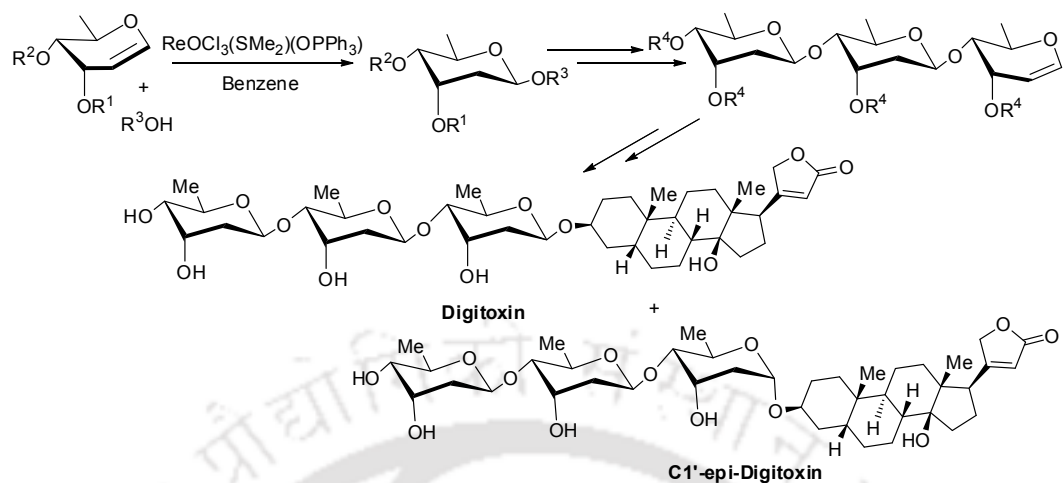
□ Scheme 2

Liu *et al.*²⁴ demonstrated that 3,4-di-O-acetyl-6-deoxy-L-galactal react with benzyl carbamate, cyclohexanol through O-glycosylation in presence of boron trifluoride as catalyst in DCE furnish to 3-amino-2,3,6-trideoxyhexoses, which subsequently underwent two more steps to achieve L-epi-daunosamine as shown in Scheme 3.



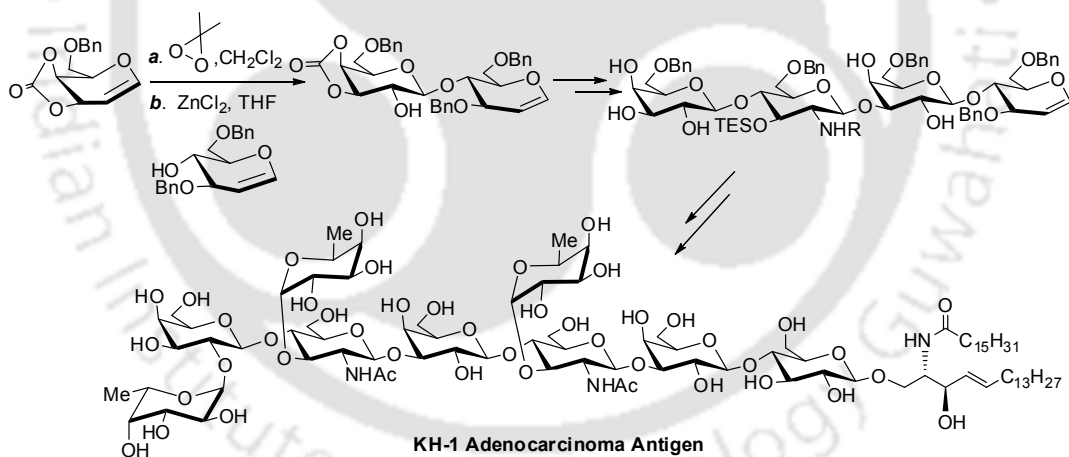
□ Scheme 3

Zhu *et al.*²⁵ developed a methodology wherein 6-deoxy-D-allal react with glycosyl donor in presence of rhenium(V) as a catalyst to afford stereoselective β -D-Digitoxosides. In addition to that, the method was further extended for the synthesis of Digitoxin and C1'-epi-Digitoxin which is shown in Scheme 4.



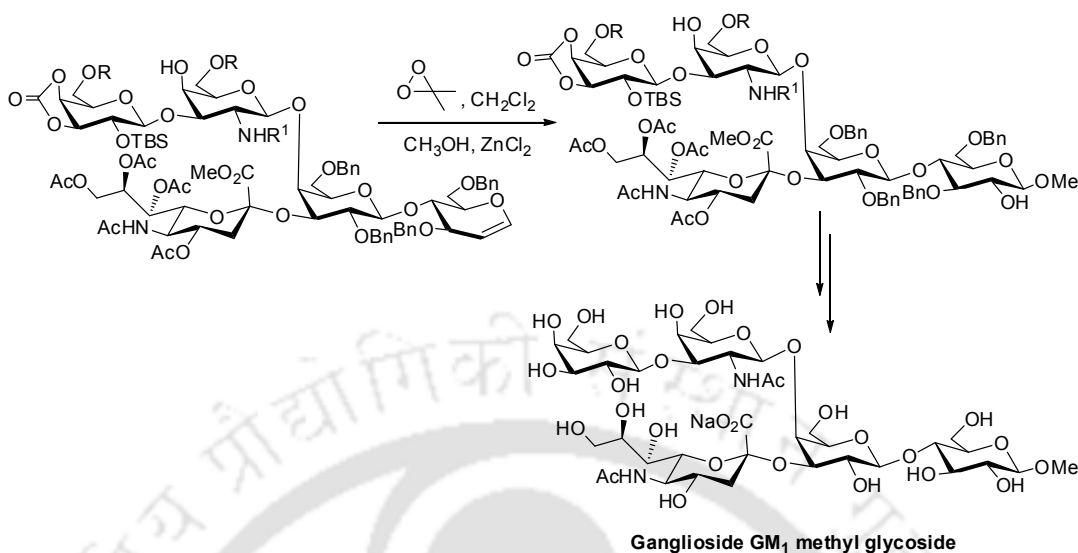
□ **Scheme 4**

Danishefsky *et al.*²⁶ demonstrated the total synthesis of KH-1 adenocarcinoma antigen, a potential anticancer vaccine by utilizing *O*-glycosylation on glucal assembly as a key synthetic intermediate step which is shown in Scheme 5.



□ **Scheme 5**

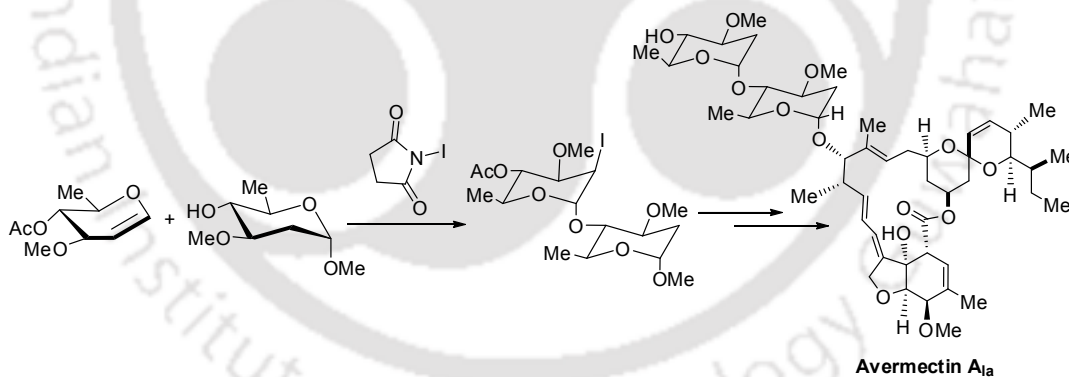
Pentasaccharide glucal moiety reacts with dimethyldioxirane/acetone in dichloromethane followed by zinc chloride in methanol leading to *O*-glycosylation of pentasaccharide, which subsequently undergoes several steps to attain the target molecule of Ganglioside GM₁ methyl glycoside²⁷ which is represented in Scheme 6.



□ Scheme 6

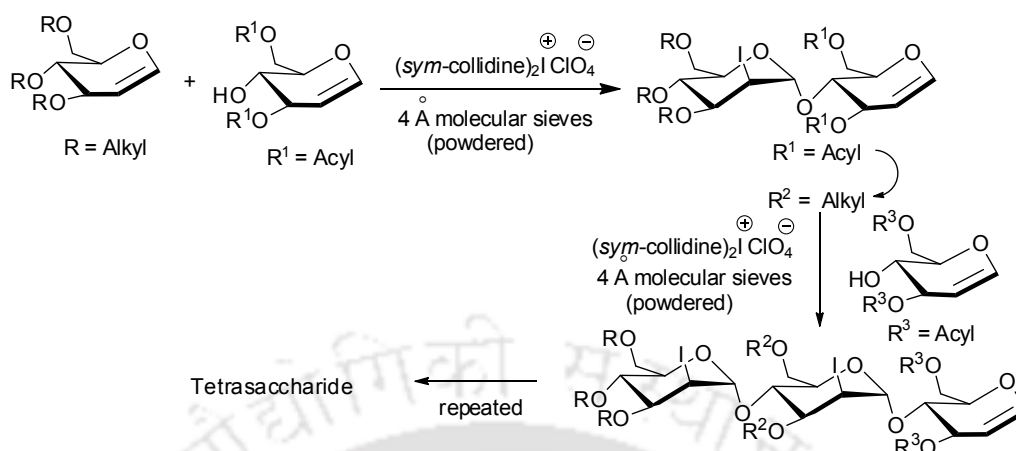
1.3 Synthetic Strategies and Development of Iodoglycosylation

Danishefsky *et al.*²⁸ demonstrated the total synthesis of Avermectin A_{1a} by utilizing Iodoglycosylation as a key step which is shown in Scheme 7.



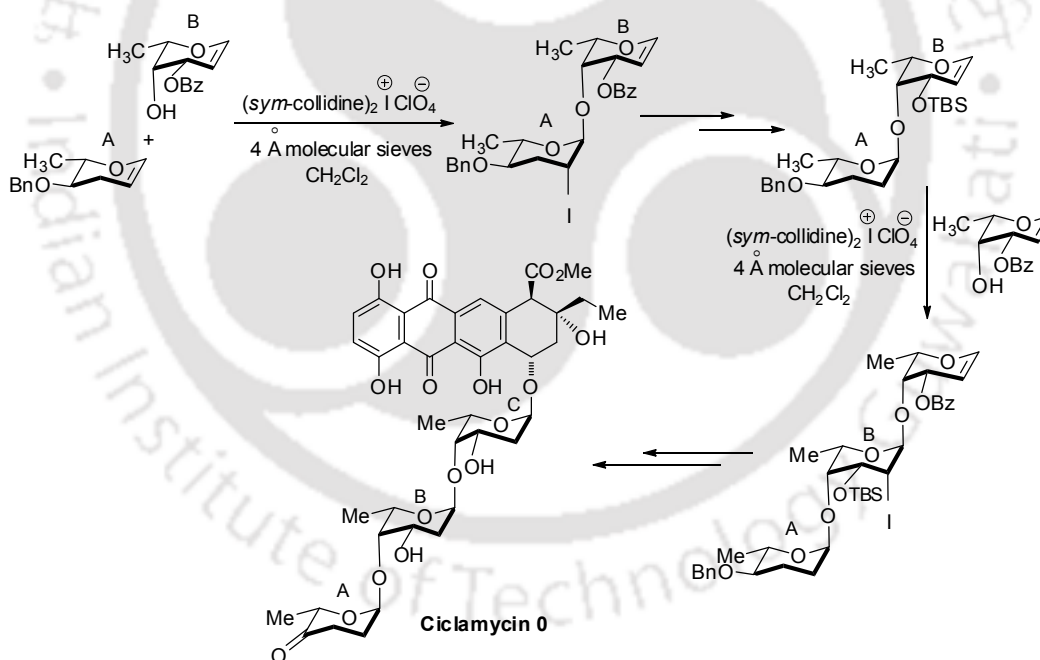
□ Scheme 7

Tetrasaccharide were synthesized from two different glucal moiety by using (*sym*-collidine)₂I⁺ClO₄⁻ as an iodinating agent.²⁹ The key fact of this methodology is that one of the glucal moiety which deactivate the ring in presence of an acetyl substituent led to controlled stereoselective iodoglycosylation which is shown in Scheme 8.



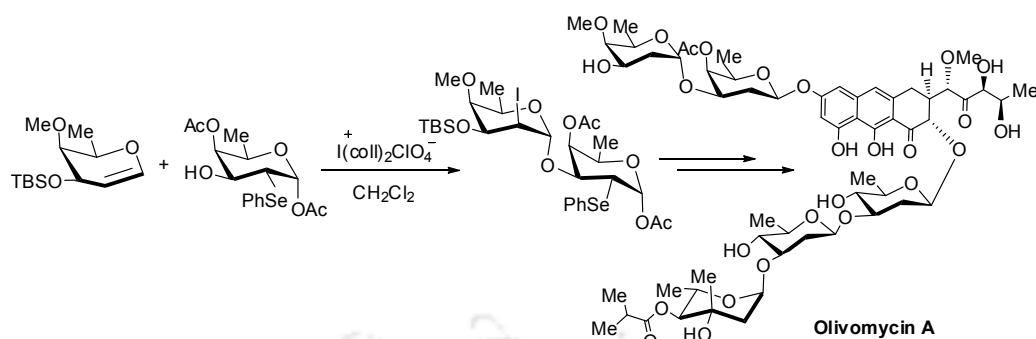
□ Scheme 8

Danishefsky *et al.*³⁰ achieved the total synthesis of ciclamycin 0 by using two stereoselective iodo-glycosylation as the key step *via* $(\text{sym-collidine})_2\text{I}^+\text{ClO}_4^-$ as an iodinating agent as depicted in Scheme 9.



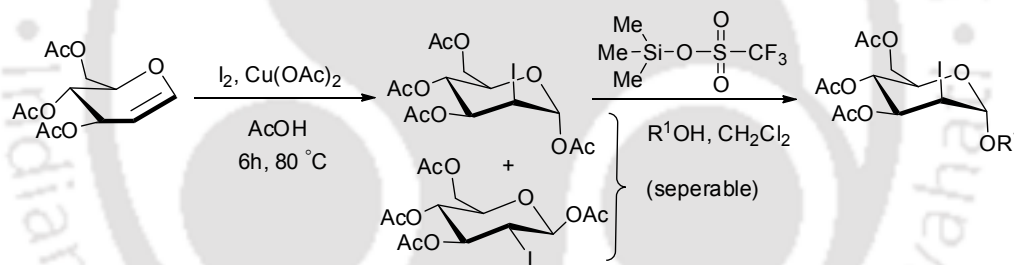
□ Scheme 9

Roush *et al.*³¹ developed that mono silylated-D-fucal reacted with 3-hydroxy-galacto-2-phenylseleno acetate in presence of $\text{I}^+(\text{coll})_2\text{ClO}_4^-$ in dichloromethane at room temperature leading to a stereoselective α -iodoglycosylation of β,α -disaccharide, which underwent several steps to attain the target molecule of olivomycin A which is represented in Scheme 10.



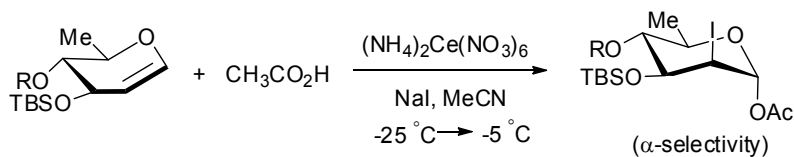
□ Scheme 10

Lafont *et al.*³² demonstrated that 3,4,6-tri-*O*-acetyl-*D*-glucal react with acetic acid in presence of iodine/copper acetate which afford a separable 2-Iodo- β -glucopyranoside and 2-Iodo- α -mannopyranoside. Typically, the obtained 2-Iodo- α -mannopyranoside further undergoes reaction with alcohol, TMSOTf in dichloromethane to furnish alkyl-3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo- α -*D*-mannopyranoside which is shown in Scheme 11.



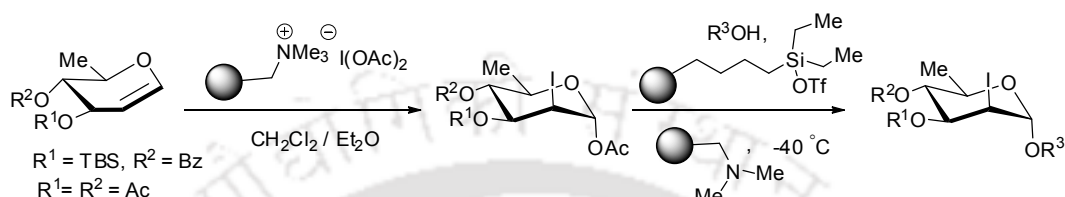
□ Scheme 11

Roush *et al.*³³ mentioned a stereoselective synthesis of 2-deoxy-2-iodo-mannopyranosyl acetates from glycal using ceric(IV) ammonium nitrate, sodium iodide and acetic acid in acetonitrile at -25 °C to -5 °C as shown in Scheme 12. The Major outputs of the results showed a superior selectivity than NIS in acetic acid.



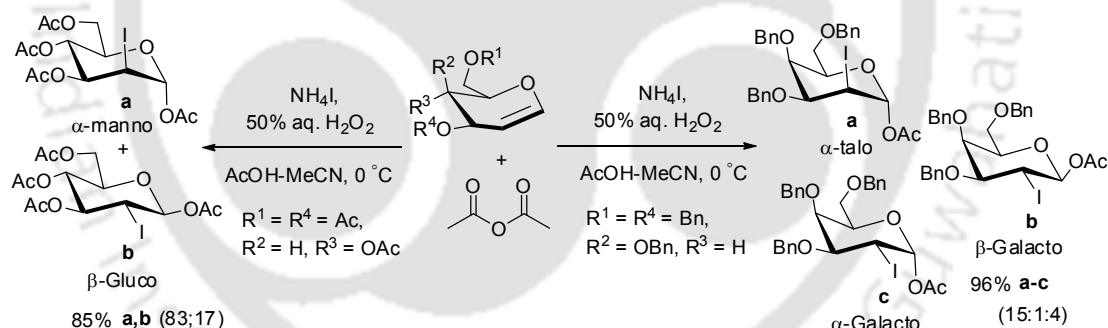
□ Scheme 12

Kirschning *et al.*³⁴ demonstrated that polymer-bound bis(acetoxy)iodate(I) reacted with glycal in dichloromethane/diethyl ether leading to 2-iodo- α -glycosyl acetates, which is activated by polymer-bound silyl triflate with different glycosyl acceptors at $-40\text{ }^{\circ}\text{C}$ to afford 2-deoxy-2-iodo- α -glycosides with a good yield which is shown in Scheme 13.



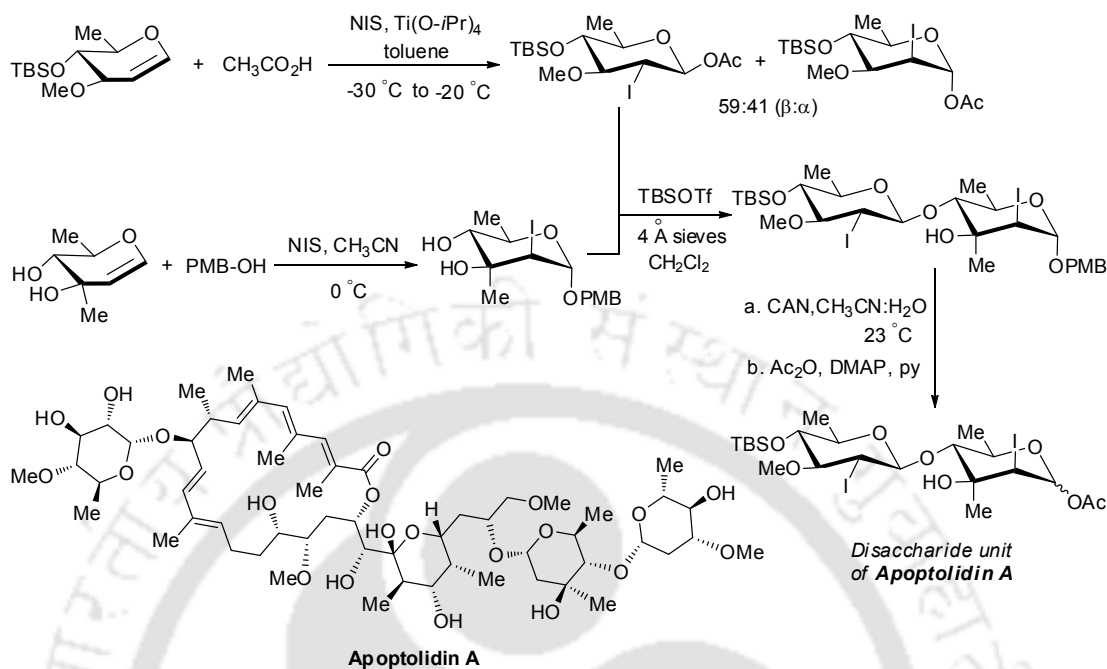
□ Scheme 13

Gammon *et al.*³⁵ developed a stereoselective synthesis of iodoacetoxylation from glugal using ammonium iodide, 50% aqueous H_2O_2 and $\text{Ac}_2\text{O}/\text{AcOH}$ in acetonitrile at $0\text{ }^{\circ}\text{C}$. Moreover, in case of 3,4,6-tri-*O*-benzyl-*D*-galactal it afforded a mixture of *talo* and *galacto* pyranosides as depicted in Scheme 14.



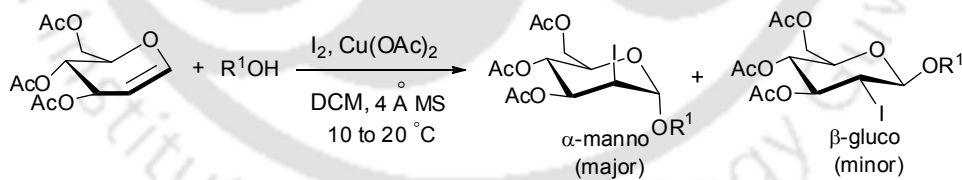
□ Scheme 14

Roush *et al.*³⁶ developed a disaccharide unit of Apoptolidin A via stereoselective synthesis of 2-iodo- β -glucopyranoside from glycal using NIS, $\text{Ti}(\text{O}i\text{Pr})_4$ at $-30\text{ }^{\circ}\text{C}$ to $-20\text{ }^{\circ}\text{C}$. Moreover, 3-*C*-methyl-*L*-rhamnal reacts with *p*-methoxybenzyl alcohol in NIS at $0\text{ }^{\circ}\text{C}$ leading to 4-hydroxy-3-*C*-methyl-2-iodo- α -mannopyranoside. The resulted 2-iodo- β -glucopyranoside react with acceptor 4-hydroxy-3-*C*-methyl-2-iodo- α -mannopyranoside in TBSOTf to furnish diiododisaccharide assembly, which further undergoes deprotection of *p*-methoxybenzyl group using CAN followed by protection of hydroxyl group by Ac_2O , pyridine to deliver the target molecule which is shown in Scheme 15.



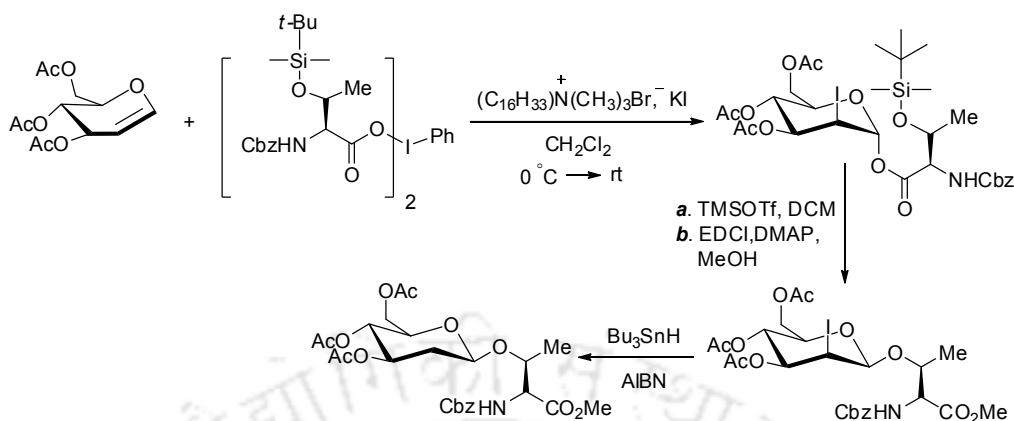
□ Scheme 15

Saeng *et al.*³⁷ illustrated that 3,4,6-tri-*O*-acetyl-D-glucal reacted with variety of alcohol using I_2 , $Cu(OAc)_2$ in dichloromethane at 10 to 20 °C direct to stereoselective synthesis of 2-deoxy-2-iodo- α -mannopyranoside as major one which is shown in Scheme 16.



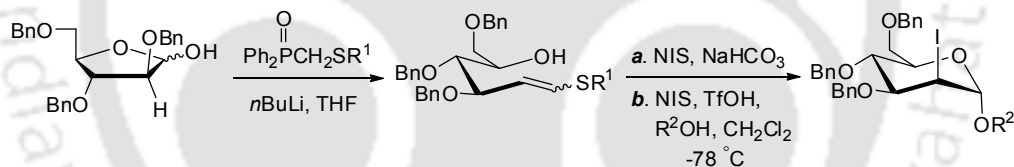
□ Scheme 16

Hotha *et al.*³⁸ demonstrated that 3,4,6-tri-*O*-acetyl-D-glucal react with $PhI(OCOR)_2$ in presence of CTAB, KI in dichloromethane at 0 °C to room temperature to afford 2-deoxy-2-iodo- α -mannopyranoside, which consequently involved in neighboring group participation in presence of Lewis acid leading to stereoselective β -iodo-mannopyranoside. However, the method was further extended for the synthesis of 2-deoxy-*O*-glycoside as shown in Scheme 17.



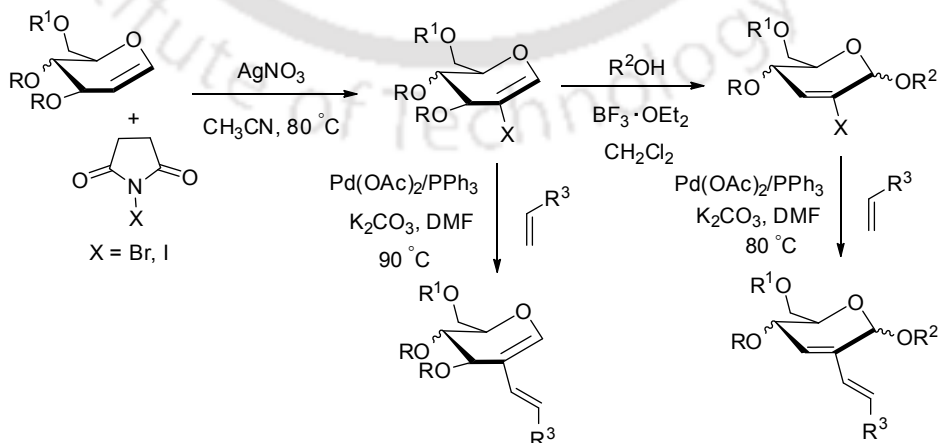
Scheme 17

Díaz and Boutureira *et al.*³⁹ represented the olefination reaction on 2,3,5-tri-O-benzyl-D-arabinofuranose using $Ph_2P(O)CH_2SR^1$ and $nBuLi$. The resulting open chain alkenyl sulfide react with NIS, $NaHCO_3$ promote I^+ endocyclization leading to 2-deoxy-2-iodo-1-thiomannopyranoside, which subsequently react with alcohol in NIS, TfOH at $-78^\circ C$ to offer the stereoselective 2-deoxy-2-iodo- α -mannopyranosides which is shown in Scheme 18.



Scheme 18

Vankar *et al.*⁴⁰ demonstrated that glucal react with NIS/NBS, $AgNO_3$ in MeCN at $80^\circ C$ furnish to 2-iodoglycals/2-bromoglycals. The synthesized 2-haloglycals were further



Scheme 19

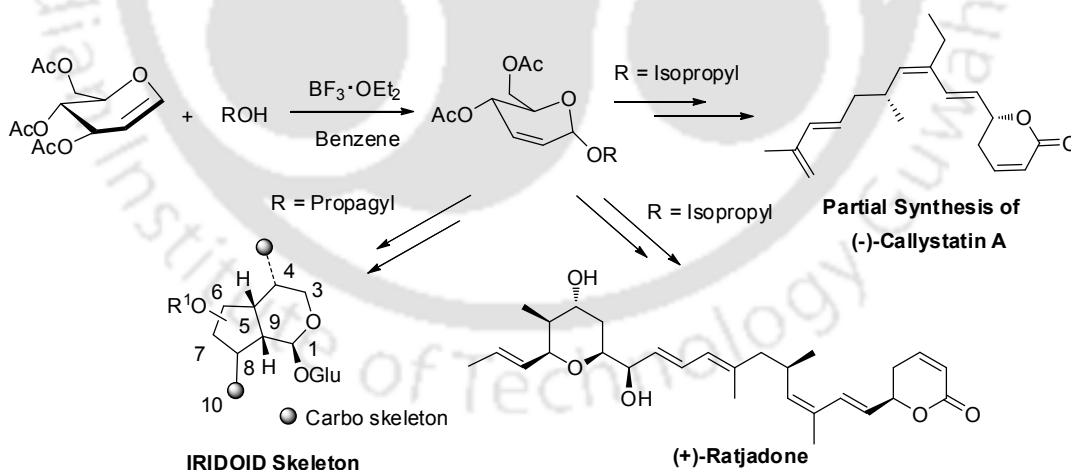
utilized for Heck coupling reaction which offered 2C branched glycols. In contrast to that, Ferrier reaction along with Heck coupling reaction resulted to 2C-branched-*O*-glycosides as represented in Scheme 19.

1.4 Ferrier rearrangement Importance

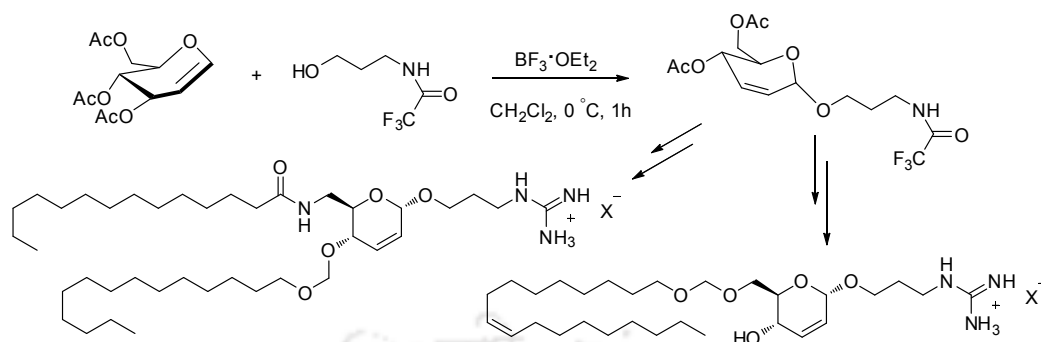
Ferrier rearrangement⁴¹ was found to be an elegant approach for the synthesis of 2,3-unsaturated-*O*-glycosides, which reside as constructive subunit for several antibiotics.⁴² Moreover, it is further utilized as glycosyl donors for numerous oligosaccharide synthesis.⁴³ The stereochemistry point of view in Ferrier rearrangement reveals that the formation of carbenium ion plays a crucial role on regio- and stereoselectivity. Indeed, the regioisomeric product depends upon the reaction condition and catalyst used in the medium. However, in case of acidic condition it offers stereoselective synthesis of 2,3-unsaturated-*O*-glycosides.

1.4a Synthetic Strategies and Development of Ferrier rearrangement on glucal

Ferrier reaction are extensively utilized for the construction of IRIDOID skeleton,⁴⁴ (+) Ratjadone⁴⁵ and partial synthesis of (-) Callystatin A⁴⁶ as depicted in Scheme 20.

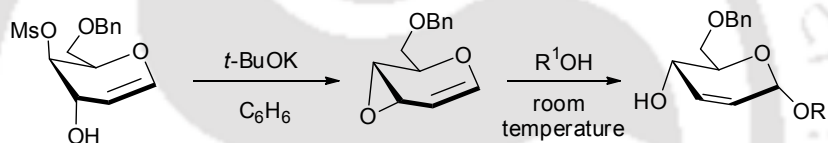


Herscovici *et al.*⁴⁷ have described the synthesis of novel cationic lipids using 2,3-unsaturated-*O*-glycoside as key step which is shown in Scheme 21. The resulted amide/acetal units which reside on 2,3-unsaturated-*O*-glycoside at C6/C4 position direct to several mono- and bicatenar derivatives.



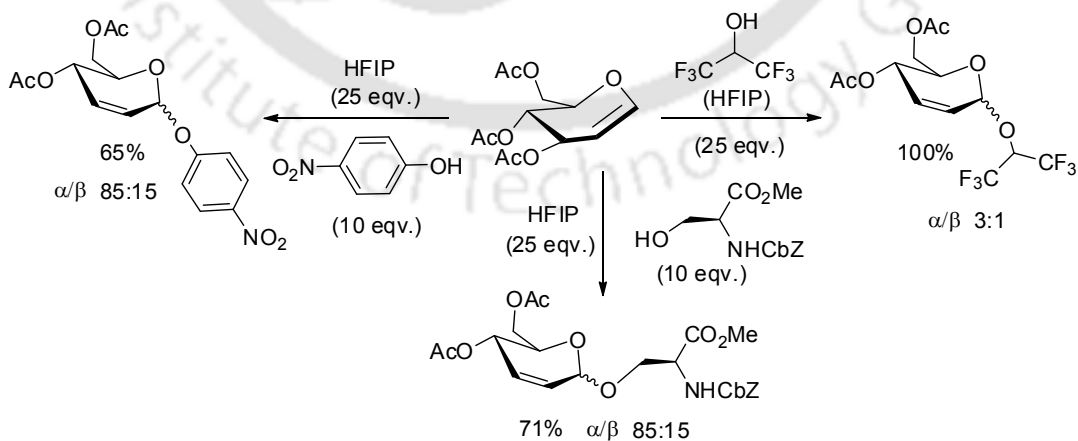
□ Scheme 21

Crotti *et al.*⁴⁸ demonstrated that hydroxy-mesylated glucal on reaction with potassium *tert*-butoxide led to unstable *in situ* generated vinyl oxirane, which further react with various alcohol at room temperature to furnish 4-hydroxy-2,3-unsaturated- α -O-glycoside which is depicted in Scheme 22.



□ Scheme 22

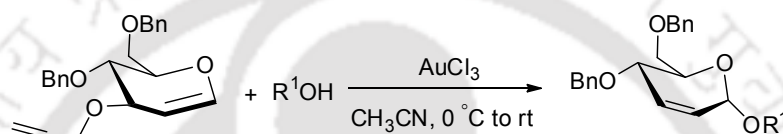
Bonnet-Delpon *et al.*⁴⁹ described that glucal on reaction with 25 equivalent of hexafluoro isopropanol (HFIP) at 80°C afford hexafluoroisopropyl-2,3-unsaturated- α -O-glycoside as major one. However in presence of 10 equivalent of other alcohol such as



□ Scheme 23

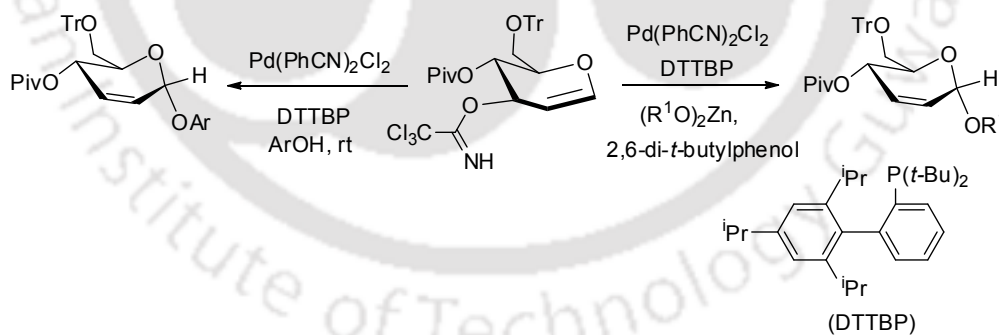
4-nitrophenol and *N*-protected serine alcohol which deliver 4-nitrophenyl-2,3-unsaturated- α -*O*-glycoside and *N*-protected serine-2,3-unsaturated- α -*O*-glycoside with 65% and 71% yield. It is important to note that in this case, hexafluoro isopropanol act as solvent-cum-abetting agent in the reaction medium as shown in Scheme 23.

Hotha *et al.*⁵⁰ developed a method wherein 4,6-di-*O*-benzyl-3-*O*-propargyl- β -glucal react with nucleophilic alcohol in presence of AuCl₃ as catalyst at 0 °C to room temperature to afford stereoselective 2,3-unsaturated- α -*O*-glycoside which is shown in Scheme 24.



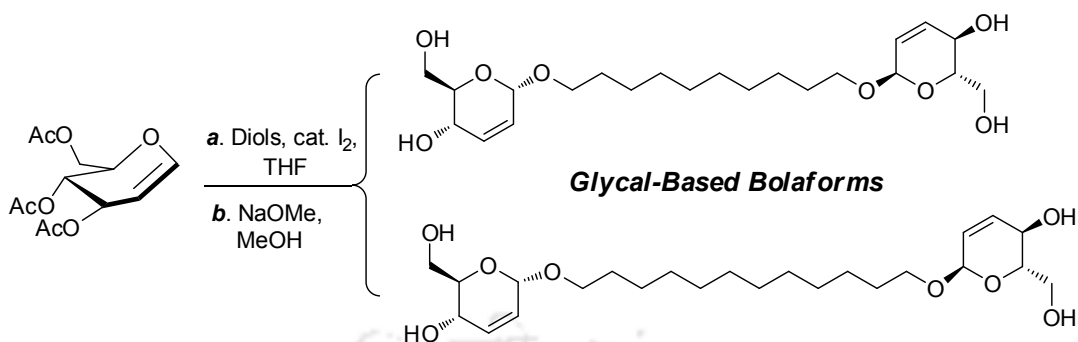
□ Scheme 24

Nguyen *et al.*⁵¹ reported that glucal containing trichloroacetimidate group at the C-3 position react with zinc(II) alkoxide/ArOH using palladium(II)-biaryl phosphine as a catalyst at room temperature which offer 2,3-unsaturated- α -*O*-glycoside as shown in Scheme 25.



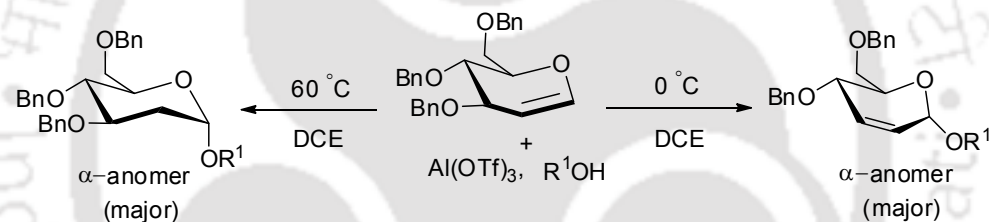
□ Scheme 25

Bozell *et al.*⁵² demonstrated iodine catalyzed synthesis of *bis*-2,3-unsaturated-*O*-glycoside from 3,4,6-tri-*O*-acetyl- β -glucal and diols in THF at room temperature. The corresponding 2,3-unsaturated-*O*-glycoside were further deprotected to endure as glycal based bolaforms which is shown in Scheme 26.



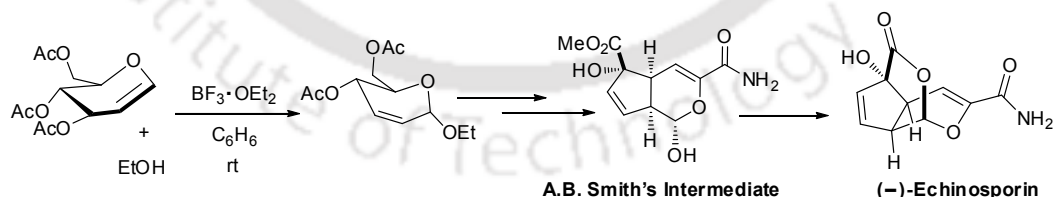
□ Scheme 26

Kinfe *et al.*⁵³ described the temperature mediated *O*-glycosides which revealed that 3,4,6-tri-*O*-benzyl-D-glucal react with alcohol in presence of $\text{Al}(\text{OTf})_3$ at 0 °C in DCE leading to stereoselective 2,3-unsaturated- α -*O*-glycoside. However, at 60 °C it afforded 2-deoxy-*O*-glycoside of α -anomer which is depicted in Scheme 27.



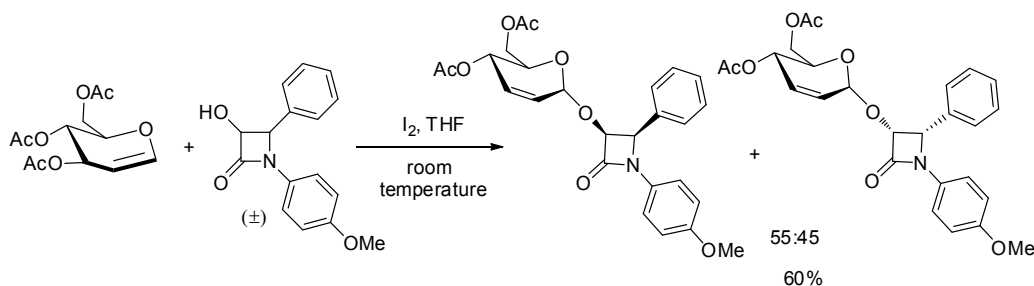
□ Scheme 27

Hale *et al.*⁵⁴ achieved the total synthesis of (-)-Echinospirin by utilizing 2,3-unsaturated-*O*-glycoside as a key step which is shown in Scheme 28.



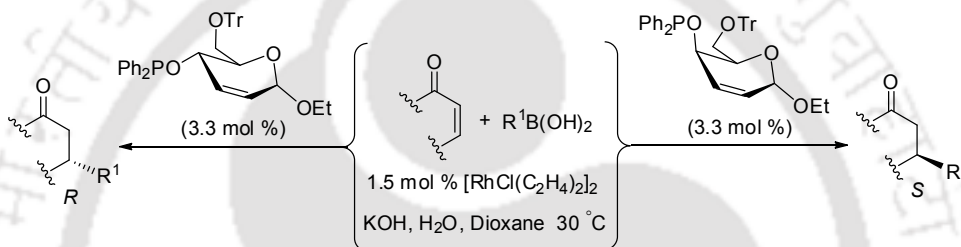
□ Scheme 28

Banik *et al.*⁵⁵ demonstrated the diastereomeric synthesis of 2,3-unsaturated-*O*-glycoside from 3,4,6-tri-*O*-acetyl-D-glucal and racemic *cis*-1-(*p*-anisyl)-3-hydroxy-4-phenyl-2-azetidinone in presence of iodine as catalyst in THF at room temperature which is depicted in Scheme 29.



□ Scheme 29

Boysen *et al.*⁵⁶ utilized the phosphinite hybrid of 2,3-unsaturated-*O*-glycoside as a novel diastereomeric ligand for asymmetric 1,4 addition as shown in Scheme 30.



□ Scheme 30

Other methods have also been reported for the synthesis of 2,3-unsaturated-*O*-glycosides⁴¹ⁱ via Ferrier rearrangement using a variety of catalysts such as protic acids⁵⁷ or Lewis acids^{41a-h} or metal triflates,⁵⁸ metal salts⁵⁹ and solid supported heterogeneous catalyst.⁶⁰ However, the drawbacks of these reported methods are: tedious work-up procedure, requirement of expensive catalysts, involvement of expensive microwave equipment, longer reaction time and harsh reaction conditions. Though, these methods are quite useful but still chemists are keen to look for milder reaction conditions with high stereoselectivity.

Overall literature survey reveals the importance of iodoglycosylation and Ferrier rearrangement on glucal, which constitute a major component in natural products and offer a diverse range of biomedical application. From this viewpoint, the challenging task to the synthetic chemists is to develop a new synthetic methodology for iodoglycosylation and Ferrier rearrangement, which are highly desirable. In the present thesis work a new methodology for the synthesis of 2-deoxy-2-iodo-*O*-glycoside and 2,3-unsaturated-*O*-glycosides has been developed, which are discussed in Chapter 2 and Chapter 3.

The methodology described in chapter 2 stems from the shadow of peroxovanadium chemistry. Vanadium haloperoxidase (isolated enzyme)⁶¹ which catalyze the oxidation of halide⁶² for the production of bromoorganics.⁶³ The reaction mechanism of this enzyme opens a new concept in synthetic platform.⁶⁴ Typically, vanadium peroxo complex formed in aqueous solution proceeds through interaction of vanadium metal with oxidant hydrogen peroxide.⁶⁵ Few years ago, Butler and her co-workers demonstrated the oxidation of bromide ion to bromonium ion using peroxovanadium complex and they extensively utilized the method for the bromination of organic substrates.⁶⁶ Taking cues from these reported method, the endeavor in the thesis work was to carry forward the novel concepts in carbohydrate chemistry.



Chapter II

Aqueous Biphasic Mediated Stereoselective Synthesis of 2-deoxy-2-iodo-O-glycoside through in situ generated Iodonium ion Using Oxobis(acetylacetonato)vanadium

Results and Discussion

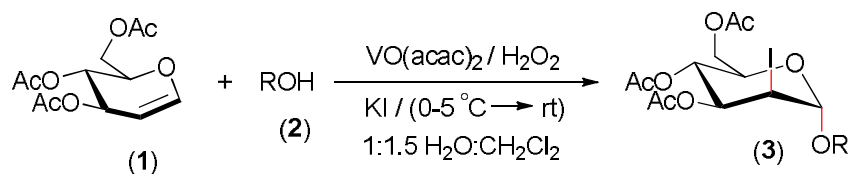
product **3a** was characterized by ^1H NMR which shows the peak at 5.10 as broad singlet for H1, doublet of doublet for H2 at 4.46 ($J = 4.4, 1.2$ Hz) and the C1-alkoxy, C2-iodo in the ^{13}C NMR which appears at 101.4, 29.9 ppm respectively.

Table 1. Optimization of the reaction conditions ^a

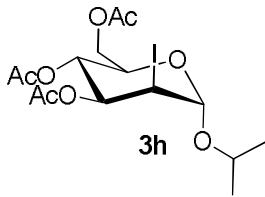
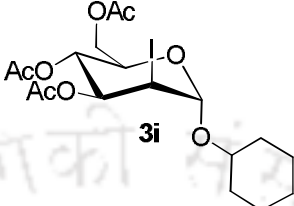
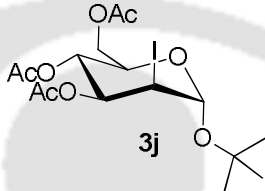
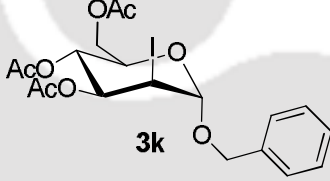
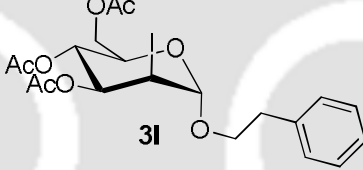
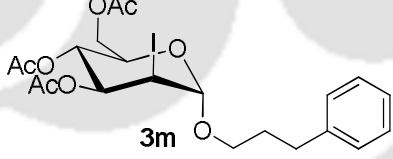
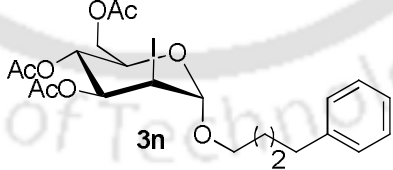
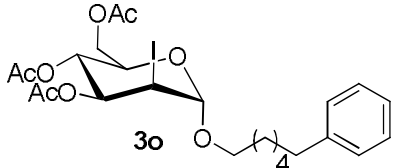
Entry	VO(acac) ₂ mol % / 30 % H ₂ O ₂ mL	Solvent	MX (mmol)	Time (h)	Yield (%) ^b	$\alpha : \beta$ ^c
1	-/0.12	H ₂ O/CH ₂ Cl ₂	NaI (0.5)	24	NR	-
2	10/0.06	H ₂ O/ CH ₂ Cl ₂	NaI (0.75)	14	<10	2:1
3	15/0.06	H ₂ O/CH ₂ Cl ₂	NaI (0.75)	12	<20	2:1
4	-/0.12	H ₂ O/CH ₂ Cl ₂	KI (0.75)	34	NR	-
5	10/0.06	H ₂ O/CH ₂ Cl ₂	KI (0.5)	12	42	3:1
6	15/0.06	H ₂ O/CH ₂ Cl ₂	KI (0.5)	10	45	4:1
7	20/0.12	H₂O/CH₂Cl₂	KI (0.75)	3	75	12:1
8	25/0.12	H ₂ O/CH ₂ Cl ₂	KI (0.75)	4	70	9:1
9	20/0.12	H ₂ O/CH ₃ CN	KI (0.75)	3.5	65	7:1
10	20/0.12	H ₂ O/EtOAc	KI (0.75)	5	68	5:1
11	20/0.12	H ₂ O/Acetone	KI (0.75)	4	62	6:1
12	20/0.12	H ₂ O/THF	KI (0.75)	4.5	58	4:1

^aThe reactions were performed using 0.5 mmol scale of 3,4,6-tri-*O*-acetyl-*D*-glucal (**1**) and *n*-butanol (**2a**). ^bIsolated yield. ^cThe ratio of $\alpha:\beta$ was determined by anomeric proton integration from ^1H NMR spectra.

We have performed the reactions with a variety of alcohols (**2b-f**) and 3,4,6-tri-*O*-acetyl-*D*-glucal (**1**) under similar reaction conditions which readily afforded the product (**3b-f**) with moderate to good yields (Table 2, entries 2-6). In the case of ethylene glycol (**2g**), the yield was only 52% (Table 2, entry 7). Encouraged by these results, the reaction was carried out using secondary and tertiary alcohols such as isopropanol (**2h**), cyclohexanol (**2i**) and *tert*-butanol (**2j**) with 3,4,6-tri-*O*-acetyl-*D*-glucal (**1**) which gave the desired product (**3h-j**) of 2-deoxy-2-iodo-*O*-glycoside in 62-88% yield (Table 2, entries 8-10). To examine the compatibility of the present protocol, we have conducted the reaction with phenyl alkyl alcohols such as benzyl alcohol (**2k**), 2-phenylethanol (**2l**), 3-phenylpropol (**2m**), 4-phenylbutanol (**2n**) and 6-phenylhexanol (**2o**) with 3,4,6-tri-*O*-acetyl-*D*-glucal (**1**) under identical reaction condition which gave the corresponding product (**3k-o**) in 40-72% yield (Table 2, entries 11-15).

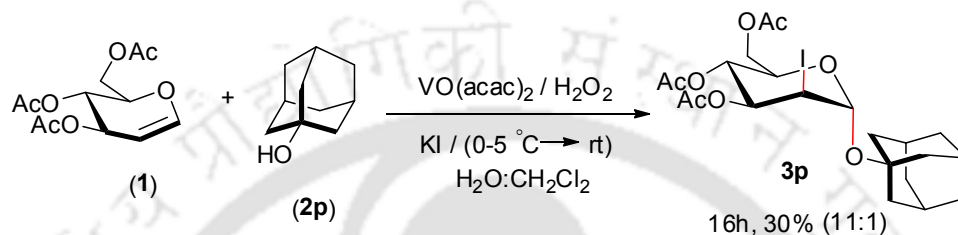
Table 2. Iodoglycosidation of 3,4,6-tri-*O*-acetyl-D-glucal^a

S.No	Alcohol (2)	Product (3)	Time (h)	Yield (%) ^b	$\alpha:\beta^c$
01	<i>n</i> -butanol 2a		2.5	87	11:1
02	Amyl alcohol 2b		3.5	82	12:1
03	Isoamylalcohol 2c		10	63	α
04	Heptan-1-ol 2d		4	72	14:1
05	Ethanol 2e		3.5	88	12:1
06	<i>n</i> -propanol 2f		3	85	10:1
07	Ethylene glycol 2g		12	52	8:1

08	Isopropanol 2h		5	82	11:1
09	Cyclohexanol 2i		3.5	88	^a
10	<i>tert</i> -butanol 2j		4.5	62	12:1
11	Benzyl alcohol 2k		8	72	12:1
12	2-Phenyl ethanol 2l		8.5	55	10:1
13	3-Phenyl prop-1-ol 2m		6.5	42	5:1
14	4-Phenyl butan-1-ol 2n		7	62	7:1
15	6-Phenyl hexan-1-ol 2o		8	40	4:1

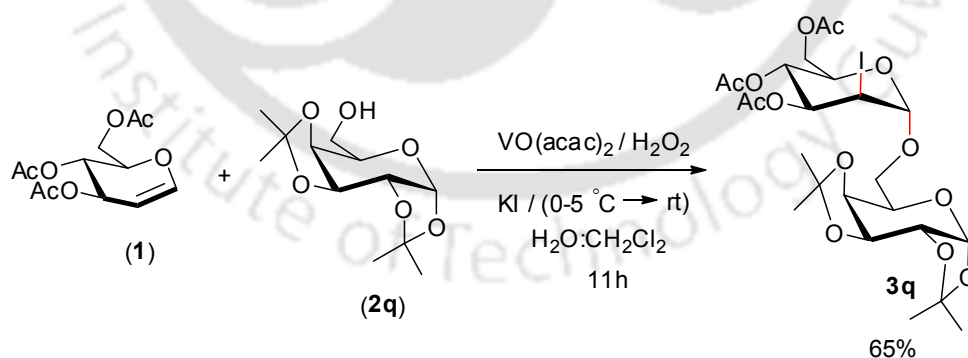
^aAll the reactions were performed using VO(acac)₂, H₂O₂, KI in H₂O:CH₂Cl₂ with 3,4,6-tri-*O*-acetyl-D-glucal (0.5 mmol) and alcohol (0.5 mmol). ^bIsolated yield. ^c The ratio $\alpha:\beta$ was determined by anomeric proton integration from ¹H NMR.

The successfully achieved results of 2-deoxy-2-iodo-*O*-glycoside derivative **3** α : β ratio, reaction time and yield are summarized in Table 2. All the isolated products were fully characterized by IR, NMR and elemental analysis. Moreover, the present protocol was further verified on reaction with sterically hindered tertiary alcohol such as adamantol (**2p**) under identical reaction condition and the isolated product **3p** was obtained with 30% yield which is shown in Scheme 32.



□ **Scheme 32.** Synthesis of adamantyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*O*-glycoside

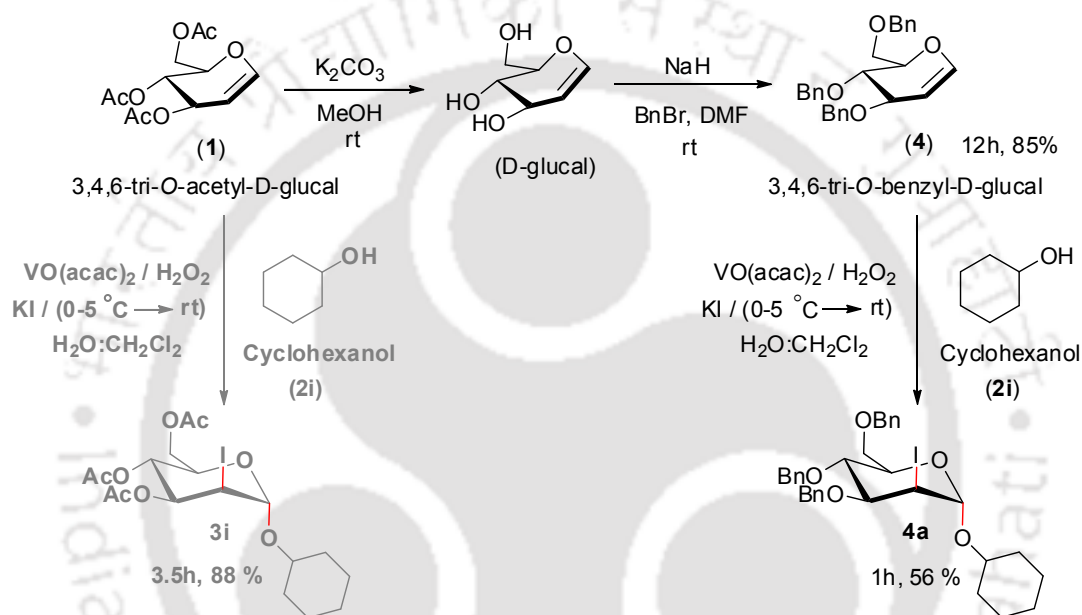
Furthermore on reaction with glucal (**1**), glycosyl acceptor 1,2,3,4-di-*O*-isopropylidene-*D*-galactopyranoside (**2q**) under similar reaction conditions and the time taken for the completion of the reaction were 11h and the corresponding product **3q** of α -anomer were obtained with 65% yield as shown in Scheme 33. It is thus clear that present protocol works well with acid sensitive glycosyl acceptor. In addition to that, Lafont *et al.* demonstrated the synthesis of Iodo-disaccharide **3q** with two consequent steps under drastic experimental conditions³² which attain the overall yield with only 58.3% respectively.



□ **Scheme 33.** Stereoselective synthesis of 2-deoxy-2-iodo-disaccharide

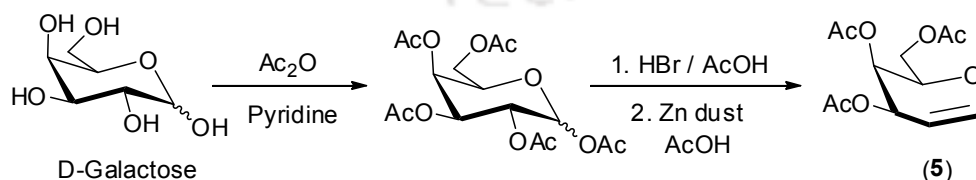
Next, we wanted to study whether different protecting group on the glucal moiety play an important role in reactivity and selectivity.^{29,35} For this study, we have synthesized 3,4,6-tri-*O*-benzyl-*D*-glucal (**4**) from 3,4,6-tri-*O*-acetyl-*D*-glucal (**1**) via two steps, *i.e.* deprotection with potassium carbonate in methanol followed by benzylation using

sodium hydride.⁶⁸ The reaction was conducted with 3,4,6-tri-*O*-benzyl-*D*-glucal (**4**) with cyclohexanol (**2i**) under similar experimental conditions which gave the required product **4a** in 56% yield as shown in Scheme 34. When we compared the output of the result with **3i**, the reaction time was shorter and the yield was lesser. From this, we may conclude that in our present protocol the reactivity of benzylated-*D*-glucal (**4**) is more as compare to acetylated-*D*-glucal (**1**). However, in case of stereoselectivity both **3i** & **4a** prefer to form only α anomer.



Scheme 34. Synthesis of cyclohexyl 3,4,6-tri-*O*-benzyl-2-deoxy-2-iodo-*O*-glycoside

We have prepared 3,4,6-tri-*O*-acetyl-*D*-galactal (**5**) from *D*-galactose via three steps⁶⁹ which is shown in Scheme 35. The obtained galactal (**5**) on reaction with isopropyl alcohol (**2h**) under analogous reaction conditions gave the desired product **6a** in 52% yield.



Scheme 35. Synthesis of 3,4,6-tri-*O*-acetyl-*D*-galactal

Likewise, a variety of alcohols such as cyclohexanol (**2i**), *n*-propanol (**2f**), amyl alcohol (**2b**) and isoamyl alcohol (**2c**) underwent reactions with 3,4,6-tri-*O*-acetyl-*D*-galactal (**5**)

under similar reaction condition which gave the corresponding products (**6b-e**) in 55-82% yield which is depicted in Table 3. Furthermore, the structure of the compound **6a** was confirmed through single X-ray diffraction data as shown in Figure 2a.

Table 3. Iodo-galactosidation of 3,4,6-tri-*O*-acetyl-D-galactal^a

S.No	ROH (2)	Product (6)	Time (h)	Yield (%) ^b	$\alpha:\beta$ ^c
01	Isopropanol 2h		12	52	α
02	Cyclohexanol 2i		2	77	α
03	<i>n</i> -propanol 2f		16	73	α
04	Amyl alcohol 2b		15	55	12:1
05	Isoamyl alcohol 2c		14	82	14:1

^aAll the reactions were performed using 3,4,6-tri-*O*-acetyl-D-galactal (0.5 mmol) and alcohol (0.5 mmol) in H₂O:CH₂Cl₂. ^bIsolated yield. ^cThe ratio of $\alpha:\beta$ was determined from ¹H NMR.

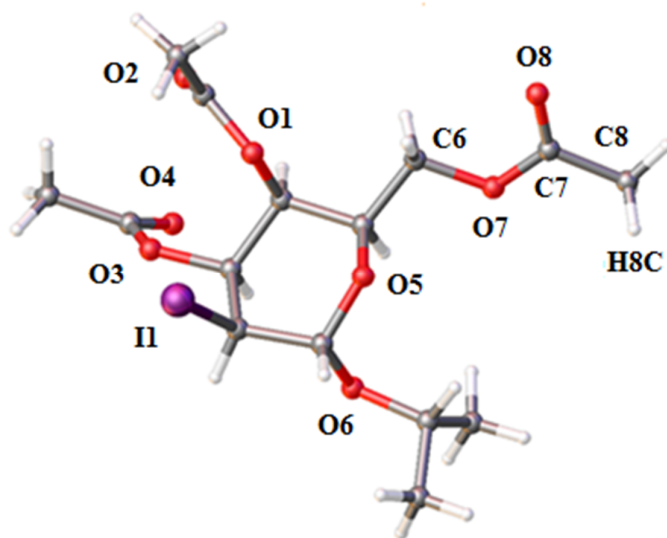


Figure 2a. X-ray crystal structure of **6a**

Crystal engineering perspectives in organic molecules have received particular attention in recent years.⁷⁰ However, the engineered structure of the crystal network mainly depend upon the weak interactions among the molecule. In addition, the molecule **6a** crystallized from chloroform and appeared as orthorhombic crystal system with $P2_12_12_1$ space group. Each unit cell contain four molecules of **6a** and the crystal packing revealed

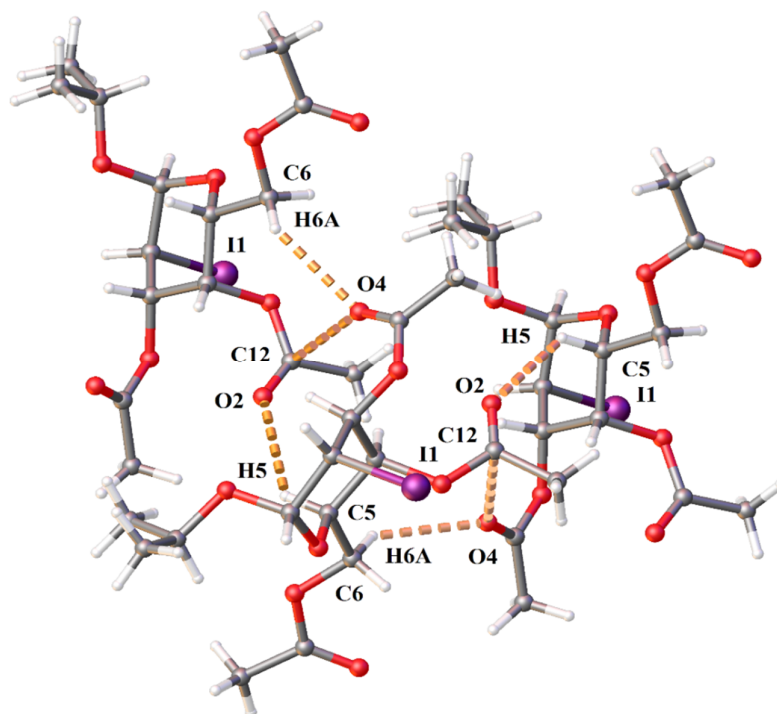


Figure 2b. Interactions of O4...H6A, O4...C12 and C5H5...O2 in **6a**

that an intermolecular interaction with O4...H6A, O4...C12 and H5...O2 exhibit an interesting fused bicyclo seven and eight membered supramolecular assembly which is shown in Figure 2b. On the other hand O5...H8C interactions in **6a** forms a fascinating zigzag structure as depicted in Figure 2c.

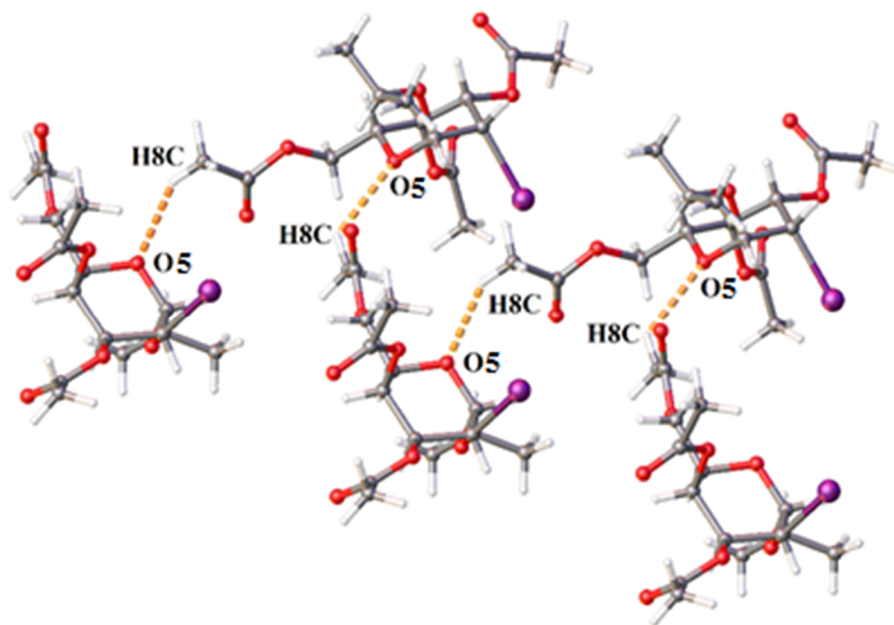
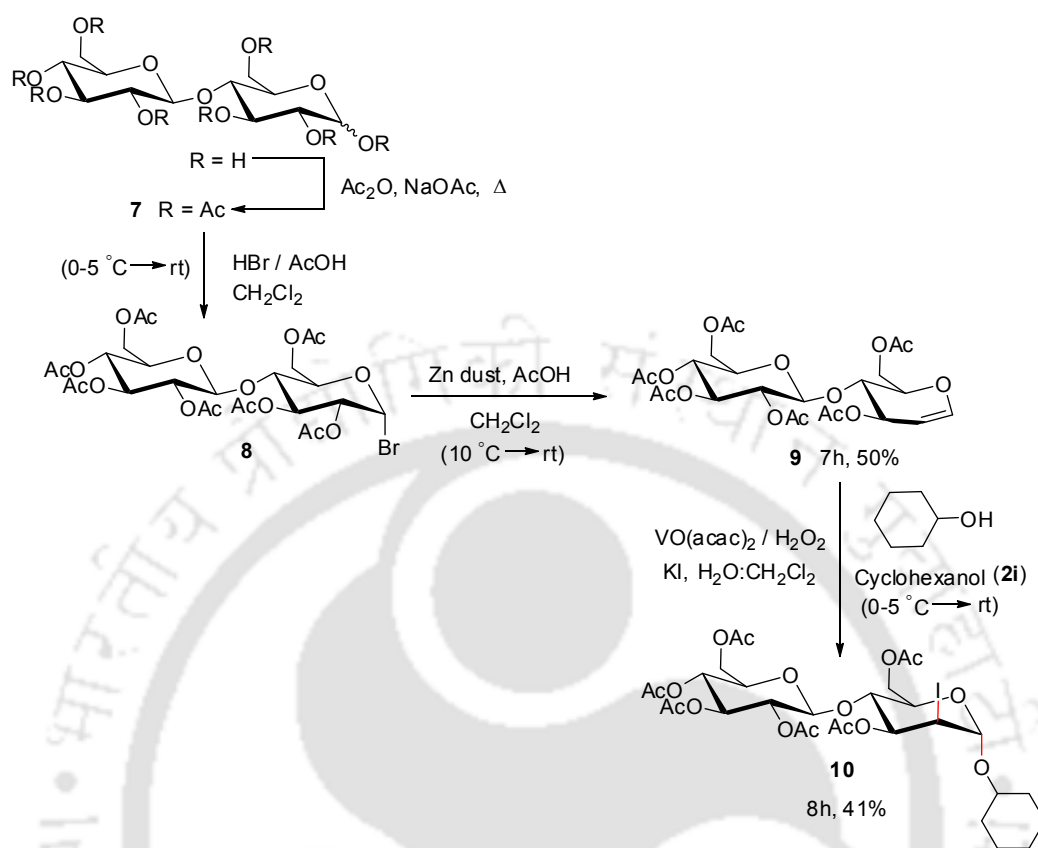


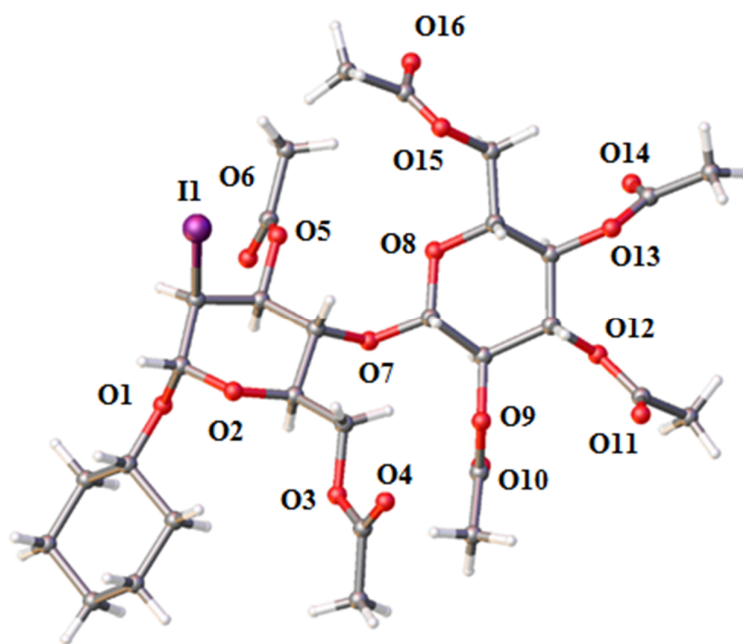
Figure 2c. Interactions of O5...H8C in **6a**

The framework of iodo-disaccharide unit might be useful for biomedical application in near future. We prefer to investigate our synthetic protocol on disaccharide unit assembly. At first hexa-*O*-acetyl-cellobial **9** was synthesized via three steps i.e., acetylation,^{71a} anomeric α -bromination followed by reduction with zinc dust in acetic acid.⁶⁹ Subsequently, the obtained disaccharide glucal **9** on reaction with cyclohexanol (**2i**) under identical reaction condition formed only α anomer of 2-deoxy-2-iododisaccharide **10** in 41% yield as shown in Scheme 36. In addition, the structure of compound **10** was confirmed through single x-ray crystal data which is depicted in Figure 3a. The molecular packing of **10** exists as layered structure which is represented in Figure 3b. Moreover, O6...I1 interaction in **10** revealed significant supramolecular architecture as shown in Figure 3c.

Our subsequent endeavour was to explore the present protocol on lactose unit. Initial reaction was performed with lactose **11**, acetic anhydride and sodium acetate under reflux condition to afford lactose octaacetate **12**.^{71a}



□ Scheme 36. Ido-glycosidation of Hexa-*O*-acetyl-cellobial



● Figure 3a. X-ray crystal structure of **10**

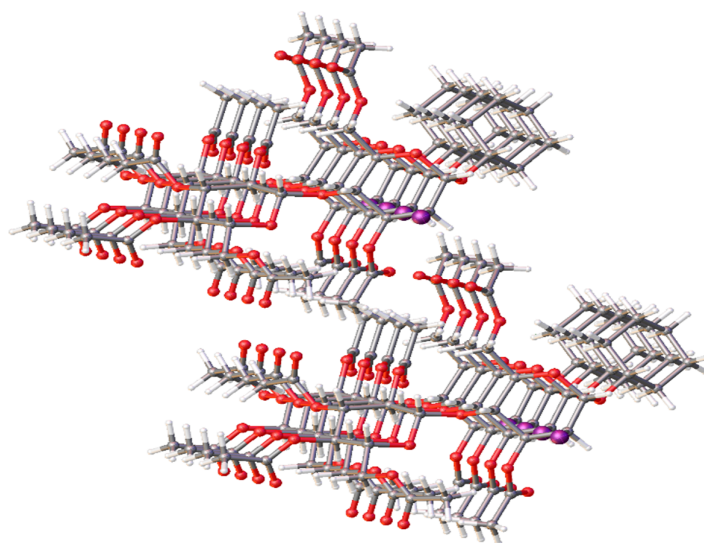


Figure 3b. Layered structure of 10

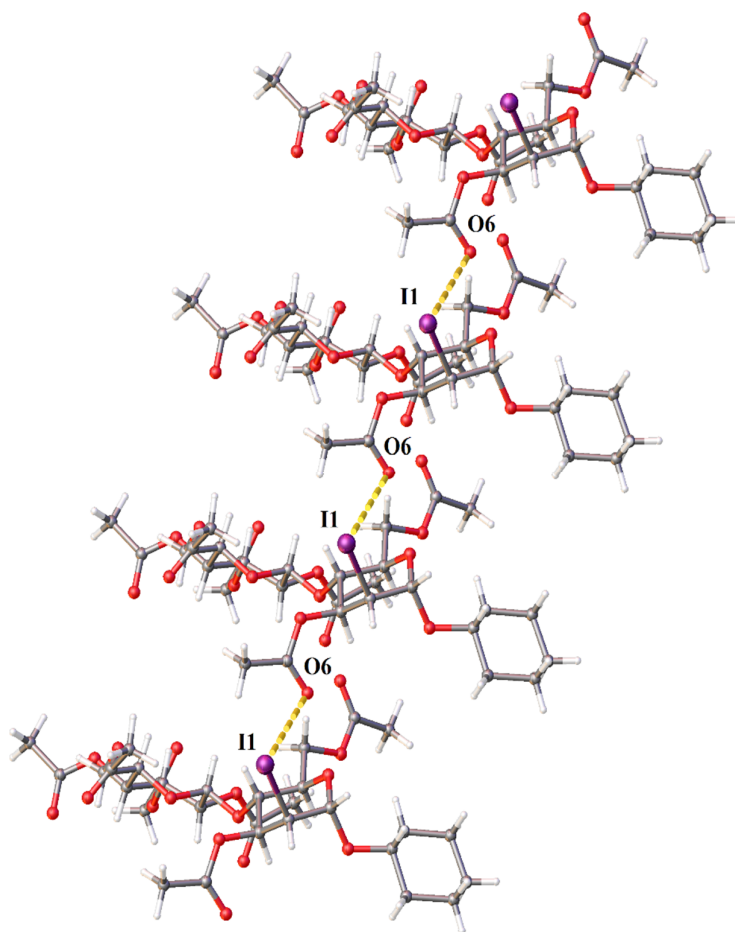
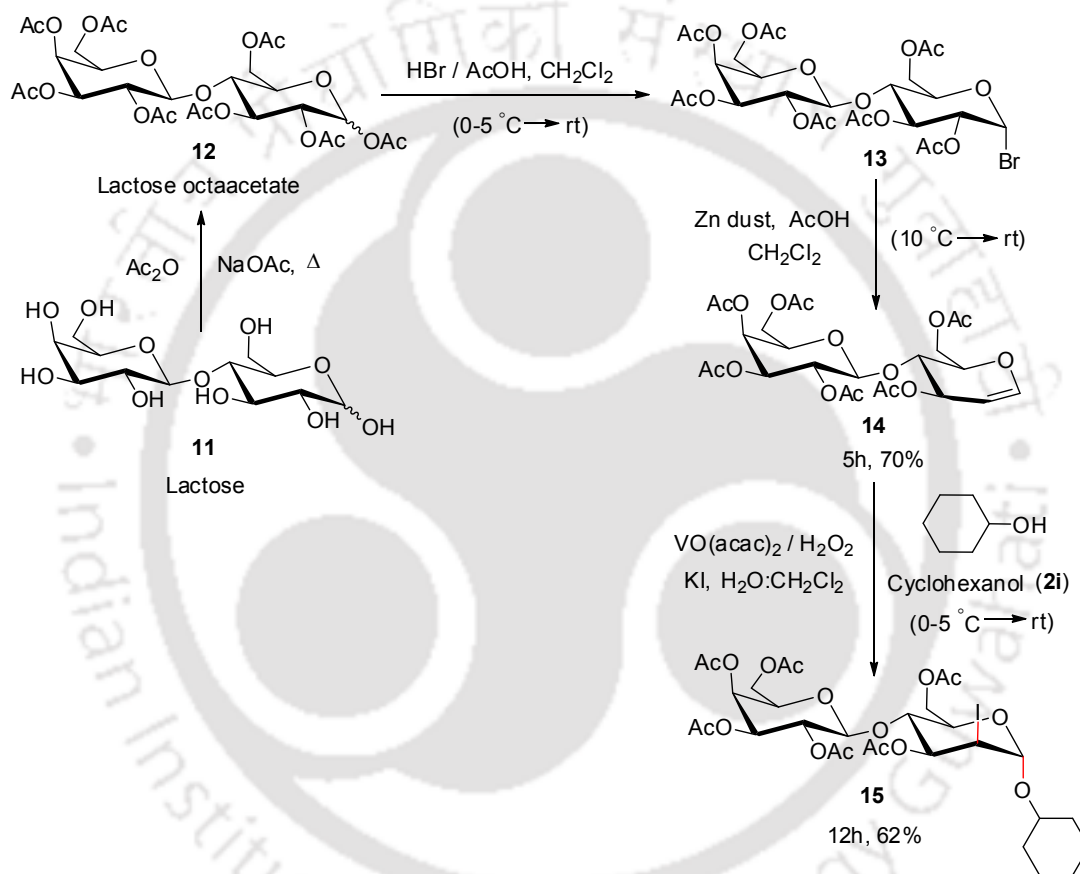


Figure 3c. O6...I1 interaction of 10

Later on, the compound **12** reacted with HBr in acetic acid and gave the desired product of acetobromo- α -D-lactose **13**. Subsequently, on reduction with Zn dust/acetic acid, hexa-*O*-acetyl-lactobial **14**^{71b} formed with 70% yield. The obtained lactobial **14** were treated with cyclohexanol (**2i**) in presence of VO(acac)₂/H₂O₂ and KI in H₂O:CH₂Cl₂ under similar reaction condition which provide the required product **15** in 62% yield with only α anomer which is shown in Scheme 37.

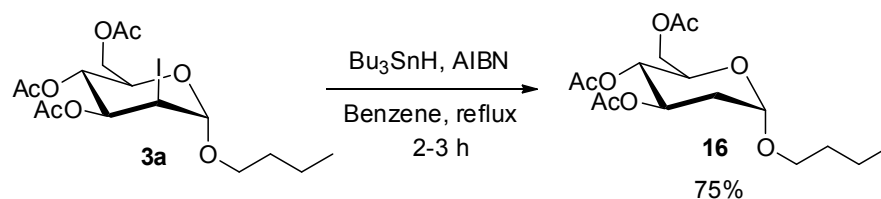


□ Scheme 37. Iodo-glycosidation of Hexa-*O*-acetyl-lactobial

Generally 2-deoxy-*O*-glycoside was synthesized from 2-deoxy-2-iodo-*O*-glycoside through reductive deiodination. From this outlook, we have conducted a reaction with butyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*O*-glycoside **3a**, tributyltin hydride in presence of radical initiator azobis isobutyronitrile in benzene under reflux condition which gave the corresponding product **16** in 75% yield as shown in Scheme 38.

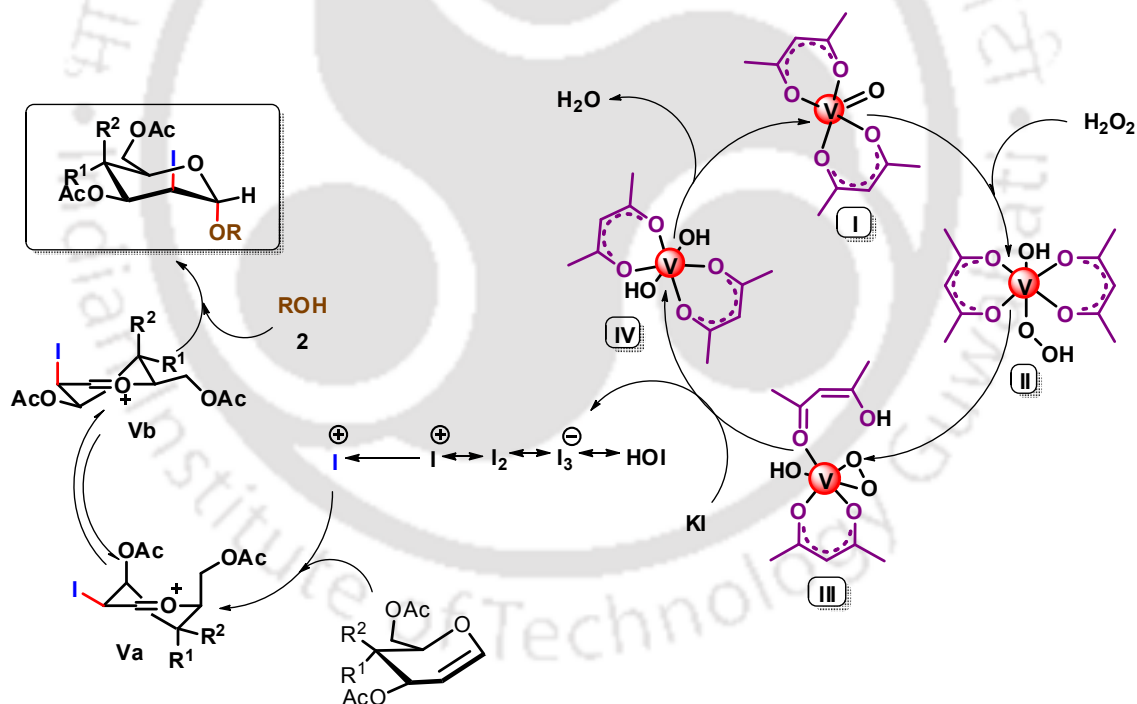
The formation of the product 2-deoxy-2-iodo-*O*-glycoside is described as follows, Initially VO(acac)₂ **I** reacts with H₂O₂ to form **II** which undergoes further rearrangement to form vanadium mono peroxocomplex **III**. The incoming halide ion I⁻ from potassium

iodide, which will oxidize from vanadium mono peroxocomplex **III** to form iodonium ion (which might exist as $I_3^- \leftrightarrow I_2 \leftrightarrow HOI$ in the solution).



□ **Scheme 38.** Synthesis of butyl 3,4,6-tri-*O*-acetyl-2-deoxy-*D*-glycopyranoside

Moreover, the liberated iodonium ion I^+ reacts with glucal to form the two possible conformers of **Va** and **Vb**. In terms of stability and its reactivity toward the incoming nucleophile it favours **Vb** for further attack with nucleophilic alcohol **2** to form highly stereoselective α anomer of 2-deoxy-2-iodo-*O*-glycoside as a predominant one as shown in Scheme 39.



□ **Scheme 39.** Plausible mechanism for the formation of 2-deoxy-2-iodo-*O*-glycoside

In summary, we have devised a greener protocol for the synthesis of 2-deoxy-2-iodo-*O*-glycoside using $\text{VO}(\text{acac})_2$, H_2O_2 and KI in aqueous based biphasic medium. The main

features of the present protocol are simple, eco-friendly, mild reaction condition, highly stereoselective and compatible with a wide range of substrate. Moreover, the present protocol is a straight forward methodology for the synthesis of 2-deoxy-2-iodo-*O*-glycosides and it was further explored for the synthesis of 2-deoxy-*O*-glycoside. However, the chemical synthesis of *O*-glycosylated product needs anhydrous reaction conditions. But the present protocol will proceed through aqueous biphasic medium. To the best of our knowledge, this is the first reported method of *O*-glycosylation in aqueous biphasic medium for the synthesis of 2-deoxy-2-iodo-*O*-glycosides. In addition to that, the molecule **6a** and **10** exhibit fascinating supramolecular scaffolds.



Experimental Section

Chapter II

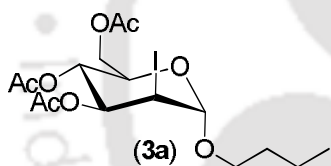
*Aqueous Biphasic Mediated
Stereoselective Synthesis of
2-deoxy-2-iodo-O-glycoside
through in situ generated
Iodonium ion Using Oxobis
(acetylacetonato)vanadium*

Experimental

General procedure for the synthesis of 2-deoxy-2-iodo-*O*-glycosides

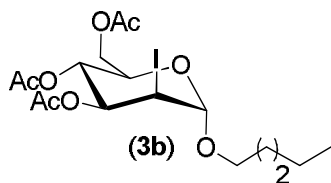
A mixture of VO(acac)₂ (0.1 mmol, 0.026 g) in 1 mL water followed by 30% hydrogen peroxide (1.07 mmol, 0.12 mL) was added at 0-5 °C and allow to stir till pale yellow color persisted. Then KI (0.75 mmol, 0.124 g)/1 mL H₂O was added to the reaction medium and a dark brown color appeared. Subsequently, glucal (0.5 mmol)/1.5 mL CH₂Cl₂ and alcohol (0.5 mmol)/1.5 mL CH₂Cl₂ was added and allowed to stir for appropriate time as shown in Table 2. Later on, the reaction mixture was extracted with DCM (2 x 25 mL) and it was washed with 10% sodium thiosulphate and dried over sodium sulfate and concentrate under reduced pressure to obtain the crude residue, which was further subjected to silica gel column chromatography to obtain the pure product of 2-deoxy-2-iodo-*O*-glycoside.

Butyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-mannopyranoside (3a)



$[\alpha]_D^{25} +12^\circ$ (*c* 0.106, CH₂Cl₂); **IR** (KBr) 2959, 2929, 2867, 1750, 1367, 1227, 1119, 1042 cm⁻¹; **¹H NMR** (400 MHz, CDCl₃): δ 5.30 (t, *J* = 9.6 Hz, 1H), 5.10 (bs, 1H), 4.57 (dd, *J* = 9.2 Hz, *J* = 4.0 Hz, 1H), 4.46 (dd, *J* = 4.4 Hz, *J* = 1.2 Hz, 1H), 4.16 (dd, *J* = 12.0 Hz, *J* = 4.8 Hz, 1H), 4.08 (dd, *J* = 12.0 Hz, *J* = 2.0 Hz, 1H), 3.95 (ddd, *J* = 10.0 Hz, *J* = 4.8 Hz, *J* = 2.4 Hz, 1H), 3.61 (dt, *J* = 10.0 Hz, *J* = 6.4 Hz, 1H), 3.40 (dt, *J* = 9.6 Hz, *J* = 6.4 Hz, 1H), 2.05 (s, 3H), 2.02 (s, 3H), 1.99 (s, 3H), 1.54-1.50 (m, 2H), 1.35-1.29 (m, 2H), 0.87 (t, *J* = 7.2 Hz, 3H); **¹³C NMR** (100 MHz, CDCl₃): δ 170.8, 170.0, 169.6, 101.4, 69.3, 69.2, 68.5, 67.7, 62.4, 31.5, 29.9, 21.0, 20.8, 20.7, 19.4, 13.9. **Anal.** Calcd for C₁₆H₂₅IO₈ (472.26): C, 40.69; H, 5.34. Found: C, 40.54; H, 5.22%.

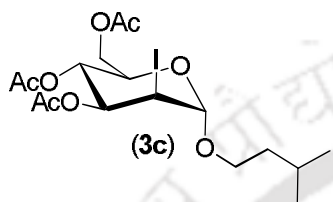
Pentyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-mannopyranoside (3b)



$[\alpha]_D^{25} +9^\circ$ (*c* 0.042, CH₂Cl₂); **IR** (KBr) 2921, 2850, 1736, 1654, 1438, 1365, 1221, 1116, 1042, 966 cm⁻¹; **¹H NMR** (400 MHz, CDCl₃): δ 5.36 (t, *J* = 9.6 Hz, 1H), 5.16 (bs, 1H), 4.64 (dd, *J* = 9.6 Hz, *J* = 4.4 Hz, 1H), 4.53 (dd, *J* = 4.4 Hz, *J* = 1.2 Hz, 1H), 4.22 (dd, *J* = 12.0 Hz, *J* = 4.8 Hz, 1H), 4.14 (dd, *J* = 12.0 Hz, *J* = 2.4 Hz, 1H), 4.01 (ddd, *J* = 10.0 Hz, *J* = 4.8 Hz, *J* = 2.4 Hz, 1H), 3.66 (dt, *J* = 9.6 Hz, *J*

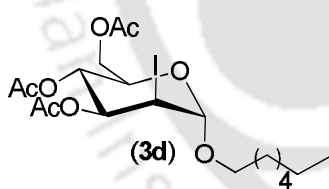
= 6.8 Hz, 1H), 3.45 (dt, $J = 9.6$ Hz, $J = 6.4$ Hz, 1H), 2.11 (s, 3H), 2.08 (s, 3H), 2.05 (s, 3H), 1.62-1.58 (m, 3H), 1.34-1.31 (m, 3H), 0.91 (t, $J = 6.8$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.9, 170.0, 169.7, 101.5, 69.3, 68.9, 67.8, 62.5, 30.0, 29.2, 28.3, 22.5, 21.1, 20.9, 20.8, 14.1. **Anal. Calcd** for $\text{C}_{17}\text{H}_{27}\text{IO}_8$ (486.29): C, 41.99; H, 5.60. Found: C, 41.87; H, 5.46%.

Isopentyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3c)



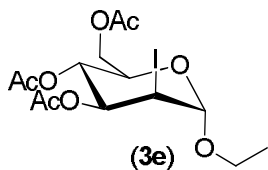
$[\alpha]_{\text{D}}^{25} +13^\circ$ (c 0.066, CH_2Cl_2); **IR** (KBr) 2957, 2924, 1745, 1466, 1367, 1225, 1118, 1050 cm^{-1} ; ^1H NMR (600 MHz, CDCl_3): δ 5.35 (t, $J = 9.6$ Hz, 1H), 5.15 (bs, 1H), 4.63 (dd, $J = 9.0$ Hz, $J = 4.2$ Hz, 1H), 4.51 (dd, $J = 4.2$ Hz, $J = 1.2$ Hz, 1H), 4.21 (dd, $J = 12.0$ Hz, $J = 4.8$ Hz, 1H), 4.14 (dd, $J = 12.0$ Hz, $J = 1.8$ Hz, 1H), 4.00 (ddd, $J = 9.6$ Hz, $J = 4.8$ Hz, $J = 1.8$ Hz, 1H), 3.70 (dt, $J = 10.2$ Hz, $J = 7.2$ Hz, 1H), 3.48 (dt, $J = 9.6$ Hz, $J = 7.2$ Hz, 1H), 2.10 (s, 3H), 2.07 (s, 3H), 2.05 (s, 3H), 1.70-1.67 (m, 1H), 1.51-1.46 (m, 2H), 0.91 (d, $J = 2.4$ Hz, 3H), 0.90 (d, $J = 2.4$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.8, 170.0, 169.7, 101.4, 69.2, 67.7, 67.2, 62.4, 38.2, 29.9, 25.0, 22.7, 22.5, 21.1, 20.9, 20.8. **Anal. Calcd** for $\text{C}_{17}\text{H}_{27}\text{IO}_8$ (486.29): C, 41.99; H, 5.60. Found: C, 41.89; H, 5.48%.

Heptyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3d)



$[\alpha]_{\text{D}}^{25} +11^\circ$ (c 0.082, CH_2Cl_2); **IR** (KBr) 2928, 2842, 1752, 1585, 1368, 1226, 1127, 1044 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 5.34 (t, $J = 9.6$ Hz, 1H), 5.14 (bs, 1H), 4.62 (dd, $J = 9.6$ Hz, $J = 4.4$ Hz, 1H), 4.50 (dd, $J = 4.4$ Hz, $J = 1.2$ Hz, 1H), 4.20 (dd, $J = 12.0$ Hz, $J = 4.8$ Hz, 1H), 4.12 (dd, $J = 12.0$ Hz, $J = 2.0$ Hz, 1H), 3.99 (ddd, $J = 10.0$ Hz, $J = 4.8$ Hz, $J = 2.4$ Hz, 1H), 3.63 (dt, $J = 9.6$ Hz, $J = 6.8$ Hz, 1H), 3.43 (dt, $J = 9.6$ Hz, $J = 6.4$ Hz, 1H), 2.09 (s, 3H), 2.06 (s, 3H), 2.03 (s, 3H), 1.60-1.56 (m, 2H), 1.29-1.23 (m, 8H), 0.87 (t, $J = 6.0$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.9, 170.0, 169.7, 101.4, 69.3, 69.2, 68.9, 67.8, 62.4, 31.9, 29.9, 29.5, 29.1, 26.2, 22.7, 21.1, 20.9, 20.8, 14.2. **Anal. Calcd** for $\text{C}_{19}\text{H}_{31}\text{IO}_8$ (514.34): C, 44.37; H, 6.07. Found: C, 44.23; H, 5.95%.

Ethyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3e)

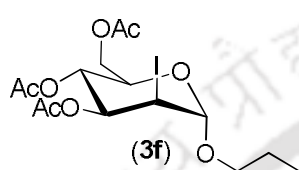


$[\alpha]_{\text{D}}^{25} +18^\circ$ (c 0.07, CH_2Cl_2); **IR** (KBr) 2975, 2924, 1743, 1367, 1225, 1123, 1048 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 5.27 (t,

$J = 9.6$ Hz, 1H), 5.09 (bs, 1H), 4.56 (dd, $J = 9.2$ Hz, $J = 4.4$ Hz, 1H), 4.43 (d, $J = 4.4$ Hz, 1H), 4.13 (dd, $J = 12.0$ Hz, $J = 4.4$ Hz, 1H), 4.04 (dd, $J = 12.0$ Hz, $J = 2.4$ Hz, 1H), 3.94 (ddd, $J = 10.0$ Hz, $J = 4.8$ Hz, $J = 2.4$ Hz, 1H), 3.66-3.62 (m, 1H), 3.48-3.43 (m, 1H), 2.02 (s, 3H), 1.99 (s, 3H), 1.96 (s, 3H), 1.15 (t, $J = 6.8$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.7, 169.8, 169.5, 101.1, 69.1, 67.7, 64.1, 62.3, 29.8, 20.1, 20.8, 20.7, 15.0.

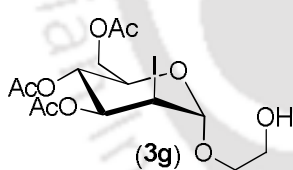
Anal. Calcd for $\text{C}_{14}\text{H}_{21}\text{IO}_8$ (444.21): C, 37.85; H, 4.76. Found: C, 37.72; H, 4.64%.

Propyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3f)



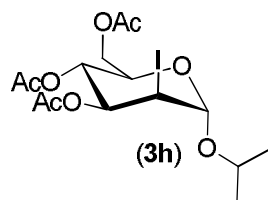
$[\alpha]_{\text{D}}^{25} +12^\circ$ (c 0.062, CH_2Cl_2); **IR** (KBr) 2964, 2935, 1747, 1437, 1367, 1228, 1121, 1040, 949 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 5.36 (t, $J = 10.0$ Hz, 1H), 5.16 (bs, 1H), 4.64 (dd, $J = 9.2$ Hz, $J = 4.0$ Hz, 1H), 4.52 (d, $J = 4.4$ Hz, 1H), 4.22 (dd, $J = 12.0$ Hz, $J = 4.8$ Hz, 1H), 4.14 (dd, $J = 12.4$ Hz, $J = 2.4$ Hz, 1H), 4.01 (ddd, $J = 10.0$ Hz, $J = 4.4$ Hz, $J = 2.0$ Hz, 1H), 3.61 (dt, $J = 9.6$ Hz, $J = 6.8$ Hz, 1H), 3.42 (dt, $J = 9.6$ Hz, $J = 6.4$ Hz, 1H), 2.11 (s, 3H), 2.08 (s, 3H), 2.05 (s, 3H), 1.64-1.59 (m, 2H), 0.93 (t, $J = 7.6$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.9, 170.0, 169.7, 101.5, 70.4, 69.3, 69.2, 67.8, 62.4, 29.9, 22.8, 21.1, 20.9, 20.8, 10.7. **Anal. Calcd** for $\text{C}_{15}\text{H}_{23}\text{IO}_8$ (458.24): C, 39.32; H, 5.06. Found: C, 39.20; H, 4.93%.

2-Hydroxy ethyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3g)



$[\alpha]_{\text{D}}^{25} +23^\circ$ (c 0.022, CH_2Cl_2); **IR** (KBr) 3414, 2959, 2923, 2845, 1736, 1653, 1438, 1367, 1226, 1122, 1045 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 5.36 (t, $J = 9.6$ Hz, 1H), 5.19 (bs, 1H), 4.60 (dd, $J = 9.2$ Hz, $J = 4.4$ Hz, 1H), 4.52 (dd, $J = 4.4$ Hz, $J = 1.2$ Hz, 1H), 4.24 (dd, $J = 12.4$ Hz, $J = 4.8$ Hz, 1H), 4.12 (dd, $J = 12.4$ Hz, $J = 2.0$ Hz, 1H), 4.09-4.03 (m, 1H), 3.88-3.79 (m, 2H), 3.62-3.61 (m, 2H), 2.09 (s, 3H), 2.04 (s, 3H), 2.00 (s, 3H), 1.69 (bs, -OH, 1H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.9, 169.8, 101.5, 69.5, 69.1, 67.6, 66.8, 62.3, 29.8, 21.1, 20.9. **Anal. Calcd** for $\text{C}_{14}\text{H}_{21}\text{IO}_9$ (460.21): C, 36.54; H, 4.60. Found: C, 36.42; H, 4.47%.

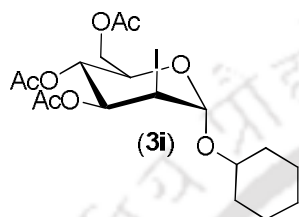
Isopropyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3h)



$[\alpha]_{\text{D}}^{25} +17^\circ$ (c 0.054, CH_2Cl_2); **IR** (KBr) 2973, 2928, 1746, 1437, 1368, 1226, 1117, 1036, 930, 838 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 5.29 (t, $J = 9.6$ Hz, 1H), 5.19 (bs, 1H), 4.59 (dd, $J = 9.6$ Hz, $J = 4.0$ Hz, 1H), 4.41 (d, $J = 4.4$ Hz, 1H), 4.15

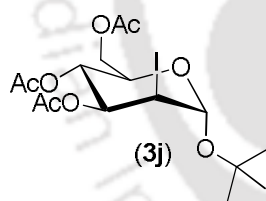
(dd, $J = 12.0$ Hz, $J = 4.4$ Hz, 1H), 4.06 (dd, $J = 12.4$ Hz, $J = 2.0$ Hz, 1H), 4.02 (ddd, $J = 10.0$ Hz, $J = 5.2$ Hz, $J = 2.4$ Hz, 1H), 3.85-3.80 (m, 1H), 2.04 (s, 3H), 2.02 (s, 3H), 1.99 (s, 3H), 1.16 (d, $J = 6.0$ Hz, 3H), 1.10 (d, $J = 6.4$ Hz, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.8, 170.0, 169.7, 99.7, 71.2, 69.3, 67.9, 62.5, 30.7, 23.2, 21.8, 21.1, 20.9, 20.8. **Anal. Calcd** for $\text{C}_{15}\text{H}_{23}\text{IO}_8$ (458.24): C, 39.32; H, 5.06. Found: C, 39.18; H, 4.95%.

Cyclohexyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3i)



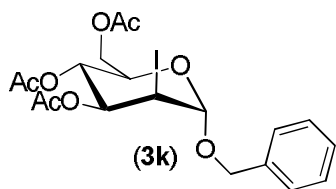
$[\alpha]_{\text{D}}^{25} +27^\circ$ (c 0.216, CH_2Cl_2); **IR** (KBr) 2933, 2855, 1748, 1449, 1367, 1228, 1116, 1039, 950, 601 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 5.34 (t, $J = 9.6$ Hz, 1H), 5.30 (bs, 1H), 4.66 (dd, $J = 9.2$ Hz, $J = 4.0$ Hz, 1H), 4.49 (dd, $J = 4.4$ Hz, $J = 1.2$ Hz, 1H), 4.22-4.16 (m, 2H), 4.11 (ddd, $J = 10.0$ Hz, $J = 4.8$ Hz, $J = 2.4$ Hz, 1H), 3.62-3.56 (m, 1H), 2.10 (s, 3H), 2.08 (s, 3H), 2.05 (s, 3H), 1.91-1.85 (m, 3H), 1.75-1.72 (m, 2H), 1.43-1.37 (m, 1H), 1.34-1.22 (m, 4H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.7, 169.9, 169.6, 99.6, 76.9, 69.2, 67.9, 62.4, 33.2, 31.6, 30.8, 25.5, 24.1, 23.9, 21.0, 20.7. **Anal. Calcd** for $\text{C}_{18}\text{H}_{27}\text{IO}_8$ (498.30): C, 43.39; H, 5.46. Found: C, 43.26; H, 5.34%.

t-Butyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3j)



$[\alpha]_{\text{D}}^{25} +10^\circ$ (c 0.016, CH_2Cl_2); **IR** (KBr) 2970, 2922, 1743, 1639, 1368, 1226, 1111, 1043, 1017 cm^{-1} ; ^1H NMR (600 MHz, CDCl_3): δ 5.37 (bs, 1H), 5.32 (t, $J = 9.0$ Hz, 1H), 4.65 (dd, $J = 9.0$ Hz, $J = 4.2$ Hz, 1H), 4.37 (dd, $J = 4.2$ Hz, $J = 1.2$ Hz, 1H), 4.21-4.16 (m, 2H), 4.05 (dd, $J = 11.4$ Hz, $J = 1.8$ Hz, 1H), 2.06 (s, 3H), 2.05 (s, 3H), 2.02 (s, 3H), 1.23 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3): 170.9, 170.1, 169.7, 96.3, 77.0, 69.3, 68.9, 68.1, 62.6, 32.2, 28.5, 21.1, 20.9. **Anal. Calcd** for $\text{C}_{16}\text{H}_{25}\text{IO}_8$ (472.26): C, 40.69; H, 5.34. Found: C, 40.55; H, 5.23%.

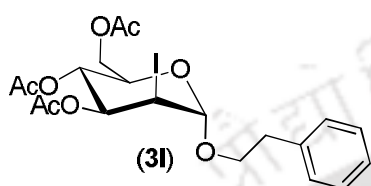
Benzyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3k)



$[\alpha]_{\text{D}}^{25} +28^\circ$ (c 0.18, CH_2Cl_2); **IR** (KBr) 2924, 2850, 1743, 1453, 1366, 1248, 1039, 749, 681 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.37-7.30 (m, 5H), 5.37 (t, $J = 9.6$ Hz, 1H), 5.23 (bs, 1H), 4.69 (d, $J = 10$ Hz, 1H), 4.65 (dd, $J = 10.0$ Hz, $J = 4.4$ Hz, 1H), 4.55 (dd, $J = 4.0$ Hz, $J = 1$ Hz, 1H), 4.53 (d, $J = 10$ Hz, 1H), 4.20 (dd, $J = 12.0$ Hz, $J = 4.4$ Hz, 1H), 4.06 (dd, $J = 12.4$ Hz, $J = 2.3$ Hz, 1H), 4.01 (ddd,

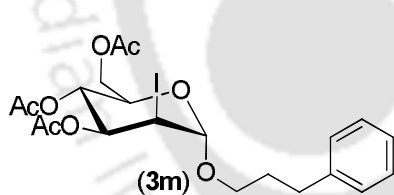
$J = 12.4$ Hz, $J = 4.4$ Hz, $J = 2.0$ Hz, 1H), 2.11 (s, 3H), 2.06 (s, 3H), 2.03 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.7, 169.9, 169.5, 136.4, 128.7, 128.3, 128.2, 100.5, 70.0, 69.3, 69.1, 67.5, 62.2, 29.6, 21.0, 20.8, 20.7. **Anal. Calcd** for $\text{C}_{19}\text{H}_{23}\text{IO}_8$ (506.28): C, 45.07; H, 4.58. Found: C, 44.92; H, 4.47%.

2-Phenylethyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-mannopyranoside (**3l**)



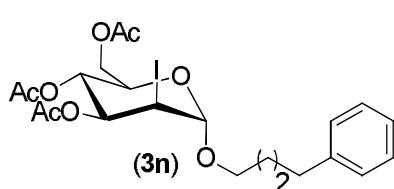
$[\alpha]_{\text{D}}^{25} +21^\circ$ (c 0.09, CH_2Cl_2); **IR** (KBr) 2920, 2845, 1744, 1367, 1224, 1120, 1046, 905, 701 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.31-7.19 (m, 5H), 5.28 (t, $J = 9.6$ Hz, 1H), 5.12 (bs, 1H), 4.56 (dd, $J = 9.2$ Hz, $J = 4.0$ Hz, 1H), 4.49 (dd, $J = 4.4$ Hz, $J = 1.2$ Hz, 1H), 4.04 (dd, $J = 12.0$ Hz, $J = 4.4$ Hz, 1H), 3.95 (dd, $J = 12.0$ Hz, $J = 2.0$ Hz, 1H), 3.82 (dt, $J = 9.6$ Hz, $J = 6.8$ Hz, 1H), 3.72 (dt, $J = 9.2$ Hz, $J = 6.4$ Hz, 1H), 3.46 (ddd, $J = 10.0$ Hz, $J = 4.4$ Hz, $J = 2.4$ Hz, 1H), 2.89 (t, $J = 6.8$ Hz, 2H), 2.07 (s, 6H), 2.02 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.7, 169.9, 169.5, 138.6, 129.0, 128.6, 126.5, 100.9, 69.1, 69.06, 69.03, 67.4, 62.0, 36.1, 29.6, 21.0, 20.8, 20.7. **Anal. Calcd** for $\text{C}_{20}\text{H}_{25}\text{IO}_8$ (520.31): C, 46.17; H, 4.84. Found: C, 46.05; H, 4.75%.

3-Phenylpropyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-mannopyranoside (**3m**)



$[\alpha]_{\text{D}}^{25} +5^\circ$ (c 0.25, CH_2Cl_2); **IR** (KBr) 3025, 2936, 2861, 1746, 1368, 1227, 1121, 1040, 910, cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.26-7.22 (m, 5H), 5.35 (t, $J = 9.6$ Hz, 1H), 5.10 (bs, 1H), 4.64 (dd, $J = 9.2$ Hz, $J = 4.4$ Hz, 1H), 4.50 (dd, $J = 4.4$ Hz, $J = 1.2$ Hz, 1H), 4.17 (dd, $J = 10.8$ Hz, $J = 4.8$ Hz, 1H), 4.08 (dd, $J = 12.0$ Hz, $J = 2.0$ Hz, 1H), 3.94 (ddd, $J = 10.0$ Hz, $J = 4.8$ Hz, $J = 2.4$ Hz, 1H), 3.59 (t, $J = 6.4$ Hz, 2H), 2.65 (t, $J = 7.6$ Hz, 2H), 2.05 (s, 3H), 2.03 (s, 3H), 2.02 (s, 3H), 1.93-1.83 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3): δ 171.1, 170.2, 169.9, 141.4, 128.7, 128.6, 126.2, 101.5, 69.4, 69.3, 68.0, 67.8, 62.5, 32.4, 30.9, 29.9, 21.1, 21.0, 20.9. **Anal. Calcd** for $\text{C}_{21}\text{H}_{27}\text{IO}_8$ (534.33): C, 47.20; H, 5.09. Found: C, 47.06; H, 4.96%.

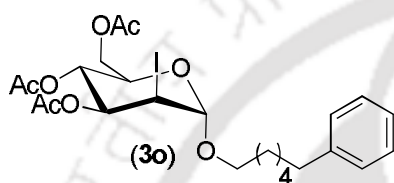
4-Phenylbutyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-mannopyranoside (**3n**)



$[\alpha]_{\text{D}}^{25} +10^\circ$ (c 0.19, CH_2Cl_2); **IR** (KBr) 3025, 2936, 2858, 1748, 1653, 1495, 1453, 1367, 1227, 1120, 1043, 748 cm^{-1} ; ^1H NMR (600 MHz, CDCl_3): δ 7.29-

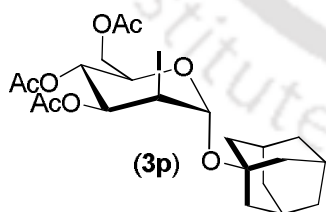
7.26 (m, 2H), 7.20-7.17 (m, 3H), 5.35 (t, $J = 9.6$ Hz, 1H), 5.14 (bs, 1H), 4.63 (dd, $J = 9.6$ Hz, $J = 4.8$ Hz, 1H), 4.51 (dd, $J = 4.8$ Hz, $J = 1.2$ Hz, 1H), 4.20 (dd, $J = 12.0$ Hz, $J = 4.8$ Hz, 1H), 4.12 (dd, $J = 12.6$ Hz, $J = 2.4$ Hz, 1H), 3.99 (m, 1H), 3.68-3.65 (m, 1H), 3.46-3.45 (m, 1H), 2.64 (t, $J = 7.8$ Hz, 2H), 2.09 (s, 3H), 2.07 (s, 3H), 2.04 (s, 3H), 1.71-1.62 (m, 4H); ^{13}C NMR (150 MHz, CDCl_3): δ 170.8, 170.0, 169.6, 142.1, 128.6, 128.5, 126.0, 101.4, 69.3, 69.2, 68.6, 67.7, 62.4, 35.7, 29.8, 29.0, 28.0, 21.1, 20.9, 20.8. **Anal.** Calcd for $\text{C}_{22}\text{H}_{29}\text{IO}_8$ (548.36): C, 48.19; H, 5.33. Found: C, 48.04; H, 5.25%.

6-Phenylhexyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-mannopyranoside (**3o**)



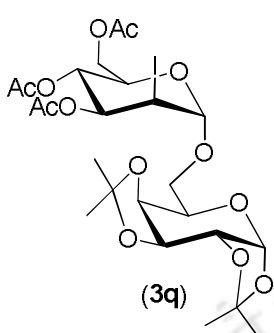
$[\alpha]_{\text{D}}^{25} +13^\circ$ (c 0.09, CH_2Cl_2); **IR** (KBr) 2927, 2848, 1751, 1654, 1366, 1226, 1121, 1044, 744 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 7.23-7.15 (m, 5H), 5.29 (t, $J = 9.6$ Hz, 1H), 5.07 (bs, 1H), 4.56 (dd, $J = 9.2$ Hz, $J = 4.4$ Hz, 1H), 4.44 (d, $J = 4.4$ Hz, 1H), 4.14 (dd, $J = 12.0$ Hz, $J = 4.0$ Hz, 1H), 4.05 (dd, $J = 12.4$ Hz, $J = 2.0$ Hz, 1H), 3.92 (ddd, $J = 10.0$ Hz, $J = 4.8$ Hz, $J = 2.4$ Hz, 1H), 3.59-3.51 (m, 1H), 3.38-3.34 (m, 1H), 2.55-2.50 (m, 2H), 2.02 (s, 3H), 2.00 (s, 3H), 1.97 (s, 3H), 1.56-1.46 (m, 4H), 1.29-1.28 (m, 4H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.9, 170.1, 169.7, 142.9, 128.6, 128.4, 125.8, 101.5, 69.3, 68.8, 67.8, 63.1, 62.5, 36.0, 31.5, 29.9, 29.5, 29.2, 25.8, 21.1, 20.9, 20.8. **Anal.** Calcd for $\text{C}_{24}\text{H}_{33}\text{IO}_8$ (576.41): C, 50.01; H, 5.77. Found: C, 49.87; H, 5.62%.

Adamantyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-mannopyranoside (**3p**)



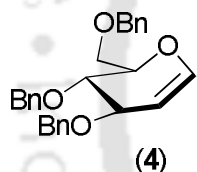
$[\alpha]_{\text{D}}^{25} +13^\circ$ (c 0.124, CH_2Cl_2); **IR** (KBr) 2909, 2851, 1748, 1451, 1367, 1228, 1034, 963 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 5.45 (bs, 1H), 5.27 (t, $J = 9.6$ Hz, 1H), 4.63 (dd, $J = 9.6$ Hz, $J = 4.4$ Hz, 1H), 4.33 (dd, $J = 4.0$ Hz, $J = 1.2$ Hz, 1H), 4.17-4.12 (m, 2H), 4.03 (dd, $J = 11.6$ Hz, $J = 2.0$ Hz, 1H), 2.08-2.06 (m, 4H), 2.02 (s, 3H), 2.00 (s, 3H), 1.98 (s, 3H), 1.73 (bs, 3H), 1.65-1.64 (m, 4H), 1.55-1.53 (m, 4H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.7, 169.9, 169.5, 94.7, 76.2, 69.2, 68.7, 68.0, 62.4, 42.3, 32.2, 30.6, 21.0, 20.7, 20.6. **Anal.** Calcd for $\text{C}_{22}\text{H}_{31}\text{IO}_8$ (550.38): C, 48.01; H, 5.68. Found: C, 47.92; H, 5.56%.

6-*O*-(3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo- α -*D*-mannopyranosyl)-1,2,3,4-di-*O*-isopropylidene- α -*D*-galactopyranoside (**3q**)



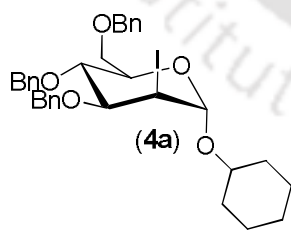
$[\alpha]_D^{25}$ -37° (c 0.042, CH_2Cl_2); **IR** (KBr) 2984, 2926, 2845, 1751, 1377, 1369, 1227, 1122, 1067, 1004, 891 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 5.50 (d, J = 5.2 Hz, 1H), 5.37 (t, J = 9.6 Hz, 1H), 5.24 (bs, 1H), 4.63-4.60 (m, 2H), 4.57 (dd, J = 4.8 Hz, J = 1.2 Hz, 1H), 4.32 (dd, J = 4.8 Hz, J = 2.4 Hz, 1H), 4.26-4.22 (m, 2H), 4.15-4.11 (m, 2H), 3.96 (ddd, J = 13.2 Hz, J = 6.4 Hz, J = 2.0 Hz, 1H), 3.79 (dd, J = 10.4 Hz, J = 6.4 Hz, 1H), 3.71 (dd, J = 10.0 Hz, J = 6.0 Hz, 1H), 2.12 (s, 3H), 2.08 (s, 3H), 2.05 (s, 3H), 1.55 (s, 3H), 1.43 (s, 3H), 1.34 (s, 6H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): δ 171.0, 170.0, 169.7, 109.6, 108.9, 101.5, 96.4, 71.0, 70.8, 70.7, 69.4, 69.3, 67.6, 67.1, 66.3, 62.3, 29.7, 26.3, 26.1, 25.1, 24.7, 21.1, 20.9, 20.8.

3,4,6-Tri-*O*-benzyl-*D*-glucal (**4**)



$^1\text{H NMR}$ (600 MHz, CDCl_3): δ 7.39-7.26 (m, 15H), 6.49 (d, J = 6.0 Hz, 1H), 4.93 (dd, J = 6.0 Hz, J = 2.4 Hz, 1H), 4.90 (d, J = 11.4 Hz, 1H), 4.72-4.68 (m, 2H), 4.66-4.60 (m, 3H), 4.28-4.27 (m, 1H), 4.14-4.11 (m, 1H), 3.93 (dd, J = 8.4 Hz, J = 6.0 Hz, 1H), 3.87 (dd, J = 10.2 Hz, J = 4.8 Hz, 1H), 3.82 (dd, J = 10.8 Hz, J = 2.4 Hz, 1H); **$^{13}\text{C NMR}$** (150 MHz, CDCl_3): δ 144.8, 138.5, 138.3, 138.1, 128.5, 128.0, 127.9, 127.8, 127.7, 100.1, 76.9, 75.8, 74.5, 73.8, 73.6, 70.5, 68.6.

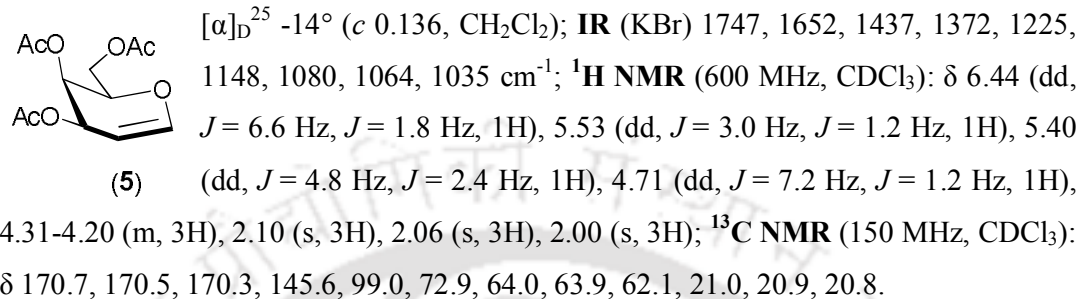
Cyclohexyl 3,4,6-tri-*O*-benzyl-2-deoxy-2-iodo-*D*-mannopyranoside (**4a**)



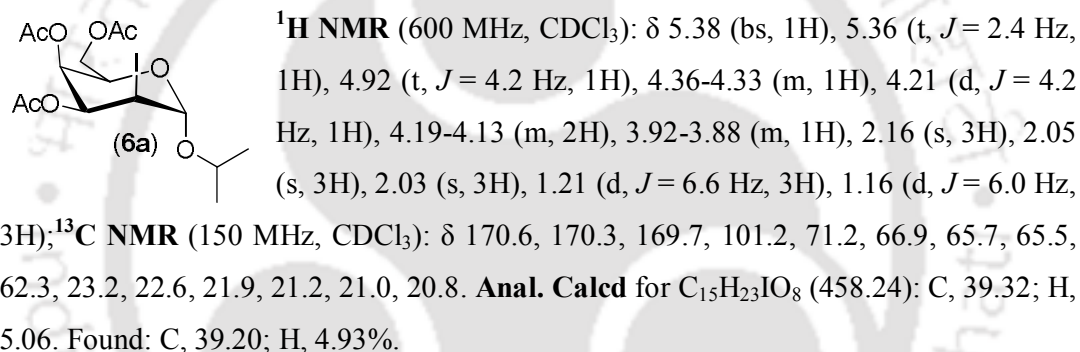
$[\alpha]_D^{25}$ $+8^\circ$ (c 0.050, CH_2Cl_2); **IR** (KBr) 3062, 3029, 2929, 2854, 1496, 1452, 1362, 1290, 1258, 1207, 1116, 1045, 1027, 999, 733, 696 cm^{-1} ; **$^1\text{H NMR}$** (600 MHz, CDCl_3): δ 7.34 (d, J = 7.2 Hz, 2H), 7.30-7.17 (m, 11H), 7.09 (dd, J = 7.8 Hz, J = 1.8 Hz, 2H), 5.27 (bs, 1H), 4.77 (d, J = 10.8 Hz, 1H), 4.63 (t, J = 12.0 Hz, 2H), 4.46-4.39 (m, 4H), 3.88-3.81 (m, 2H), 3.72 (dd, J = 10.8 Hz, J = 4.8 Hz, 1H), 3.62 (dd, J = 10.8 Hz, J = 1.2 Hz, 1H), 3.53-3.49 (m, 1H), 3.27 (dd, J = 8.4 Hz, J = 3.6 Hz, 1H), 1.75-1.74 (m, 2H), 1.61-1.59 (m, 2H), 1.43-1.41 (m, 1H), 1.18-1.14 (m, 5H); **$^{13}\text{C NMR}$** (150 MHz, CDCl_3): δ 138.6, 138.4, 138.0, 128.6, 128.5, 128.4, 128.3, 128.2, 128.0, 127.9, 127.8, 127.6, 99.6, 77.0, 76.3, 75.9, 75.5, 73.5, 72.3, 71.1, 69.2,

34.9, 33.4, 31.7, 25.7, 24.2, 24.0. **Anal. Calcd** for $C_{33}H_{39}IO_5$ (642.56): C, 61.68; H, 6.12. Found: C, 61.55; H, 6.02%.

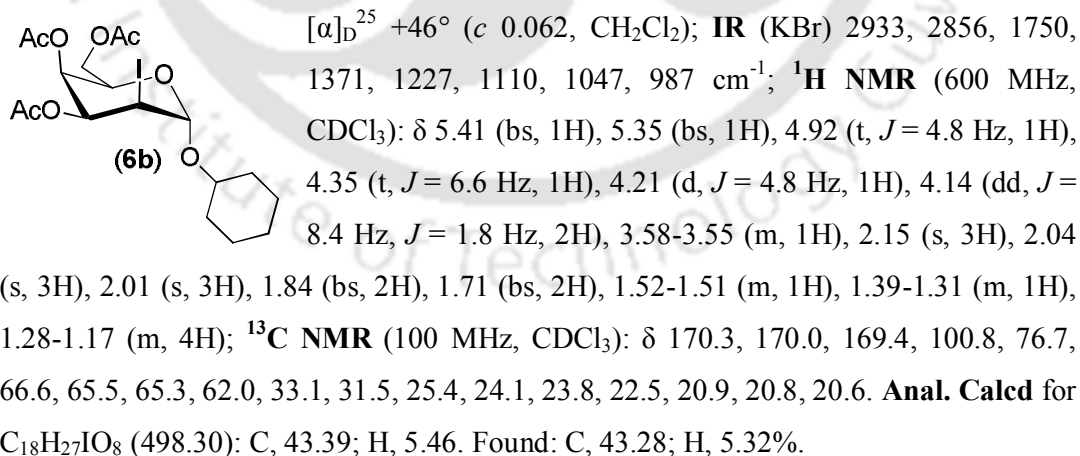
3,4,6-Tri-O-acetyl-D-galactal (5)

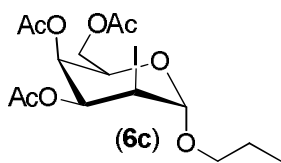


Isopropyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-talopyranoside (6a)

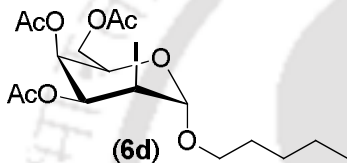


Cyclohexyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-talopyranoside (6b)

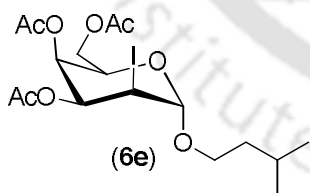


Propyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-talopyranoside (**6c**)

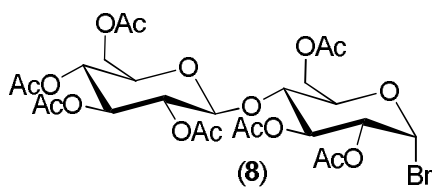
$[\alpha]_{\text{D}}^{25} +38^{\circ}$ (*c* 0.110, CH_2Cl_2); **IR** (KBr) 2964, 2926, 1750, 1372, 1228, 1170, 1116, 1049, 989, 727 cm^{-1} ; **^1H NMR** (600 MHz, CDCl_3): δ 5.35 (bs, 1H), 5.29 (bs, 1H), 4.90 (t, $J = 4.8$ Hz, 1H), 4.27-4.24 (m, 2H), 4.18-4.13 (m, 2H), 3.60 (dt, $J = 9.6$ Hz, $J = 6.6$ Hz, 1H), 3.40 (dt, $J = 9.6$ Hz, $J = 6.6$ Hz, 1H), 2.15 (s, 3H), 2.04 (s, 3H), 2.02 (s, 3H), 1.60-1.57 (m, 2H), 0.91 (t, $J = 7.8$ Hz, 3H); **^{13}C NMR** (150 MHz, CDCl_3): δ 170.6, 170.2, 169.7, 102.7, 70.3, 66.7, 65.6, 65.4, 62.2, 22.8, 21.7, 21.1, 21.0, 20.8, 10.7. **Anal. Calcd** for $\text{C}_{15}\text{H}_{23}\text{IO}_8$ (458.24): C, 39.32; H, 5.06. Found: C, 39.22; H, 4.94%.

Pentyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-talopyranoside (**6d**)

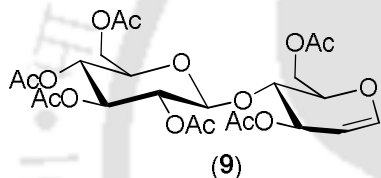
$[\alpha]_{\text{D}}^{25} +30^{\circ}$ (*c* 0.214, CH_2Cl_2); **IR** (KBr) 2957, 2927, 1750, 1370, 1225, 1111, 1051, 971 cm^{-1} ; **^1H NMR** (600 MHz, CDCl_3): δ 5.37 (bs, 1H), 5.29 (bs, 1H), 4.91 (t, $J = 4.2$ Hz, 1H), 4.28-4.25 (m, 2H), 4.19-4.13 (m, 2H), 3.65 (dt, $J = 9.6$ Hz, $J = 6.6$ Hz, 1H), 3.44 (dt, $J = 9.6$ Hz, $J = 6.6$ Hz, 1H), 2.17 (s, 3H), 2.06 (s, 3H), 2.05 (s, 3H), 1.60-01.55 (m, 2H), 1.31-1.30 (m, 4H), 0.89 (t, $J = 7.2$ Hz, 3H); **^{13}C NMR** (100 MHz, CDCl_3): δ 170.6, 170.3, 169.7, 102.7, 68.7, 66.7, 65.6, 65.4, 62.3, 29.2, 28.4, 22.5, 21.7, 21.1, 21.0, 20.8, 14.1. **Anal. Calcd** for $\text{C}_{17}\text{H}_{27}\text{IO}_8$ (486.29): C, 41.99; H, 5.60. Found: C, 41.89; H, 5.48%.

Isopentyl 3,4,6-tri-*O*-acetyl-2-deoxy-2-iodo-*D*-talopyranoside (**6e**)

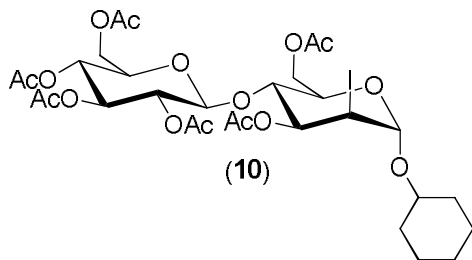
$[\alpha]_{\text{D}}^{25} +35^{\circ}$ (*c* 0.098, CH_2Cl_2); **IR** (KBr) 2957, 2928, 2872, 1750, 1467, 1371, 1229, 1113, 1054, 965, 728, 602 cm^{-1} ; **^1H NMR** (600 MHz, CDCl_3): δ 5.36 (bs, 1H), 5.28 (bs, 1H), 4.89 (t, $J = 4.2$ Hz, 1H), 4.27-4.23 (m, 2H), 4.19-4.12 (m, 2H), 3.69 (dt, $J = 9.6$ Hz, $J = 6.6$ Hz, 1H), 3.46 (dt, $J = 9.6$ Hz, $J = 6.6$ Hz, 1H), 2.16 (s, 3H), 2.04 (s, 3H), 2.02 (s, 3H), 1.69-1.65 (m, 1H), 1.48-1.44 (m, 2H), 0.89 (d, $J = 6.6$ Hz, 6H); **^{13}C NMR** (150 MHz, CDCl_3): δ 170.6, 170.3, 169.7, 102.8, 67.1, 66.9, 65.6, 65.5, 62.4, 38.3, 25.1, 22.8, 22.5, 21.8, 21.1, 21.0, 20.8. **Anal. Calcd** for $\text{C}_{17}\text{H}_{27}\text{IO}_8$ (486.29): C, 41.99; H, 5.60. Found: C, 41.87; H, 5.46%.

2,2',3,3',4',6,6'-Hepta-O-acetyl- α -bromo-D-cellobiose (**8**)

$[\alpha]_D^{25} +95^\circ$ (*c* 0.200, CH₂Cl₂); **IR** (KBr) 2956, 2923, 2852, 1752, 1736, 1366, 1214, 1033, 902, 597 cm⁻¹; **¹H NMR** (600 MHz, CDCl₃): δ 6.51 (d, *J* = 3.6 Hz, 1H), 5.52 (t, *J* = 9.6 Hz, 1H), 5.14 (t, *J* = 9.0 Hz, 1H), 5.06 (t, *J* = 10.2 Hz, 1H), 4.92 (t, *J* = 9.0 Hz, 1H), 4.75 (dd, *J* = 10.2 Hz, *J* = 4.2 Hz, 1H), 4.52 (dd, *J* = 10.8 Hz, *J* = 7.8 Hz, 2H), 4.35 (dd, *J* = 12.6 Hz, *J* = 4.8 Hz, 1H), 4.20-4.15 (m, 2H), 4.04 (dd, *J* = 12.6 Hz, *J* = 1.8 Hz, 1H), 3.83 (t, *J* = 9.6 Hz, 1H), 3.68-3.67 (m, 1H), 2.12 (s, 3H), 2.08 (s, 6H), 2.03 (s, 6H), 2.00 (s, 3H), 1.97 (s, 3H). **¹³C NMR** (150 MHz, CDCl₃): δ 170.6, 170.4, 170.3, 170.2, 169.5, 169.1, 100.7, 86.6, 75.4, 73.2, 73.1, 72.2, 71.8, 70.9, 69.6, 68.0, 61.8, 61.1, 21.0, 20.9, 20.8, 20.7.

3,6,2',3',4',6'-Hexa-O-acetyl-D-cellobial (**9**)

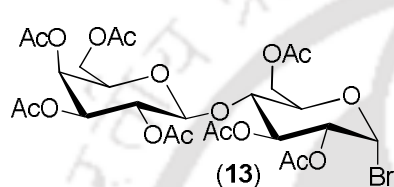
IR (KBr) 2959, 1743, 1651, 1436, 1369, 1226, 1166, 1062, 1037, 907 cm⁻¹; **¹H NMR** (400 MHz, CDCl₃): δ 6.36 (dd, *J* = 6.0 Hz, *J* = 1.2 Hz, 1H), 5.38-5.36 (m, 1H), 5.14 (t, *J* = 9.2 Hz, 1H), 5.03 (t, *J* = 9.6 Hz, 1H), 4.95-4.90 (m, 1H), 4.77 (dd, *J* = 6.0 Hz, *J* = 3.2 Hz, 1H), 4.65 (d, *J* = 8.0 Hz, 1H), 4.40 (dd, *J* = 11.2 Hz, *J* = 2.0 Hz, 1H), 4.26 (dd, *J* = 12.4 Hz, *J* = 4.4 Hz, 1H), 4.16-4.12 (m, 1H), 4.10-4.08 (m, 1H), 4.02 (dd, *J* = 12.4 Hz, *J* = 2.4 Hz, 1H), 3.94 (dd, *J* = 7.2 Hz, *J* = 5.2 Hz, 1H), 3.67-3.63 (m, 1H), 2.07 (s, 3H), 2.04 (s, 3H), 2.01 (s, 3H), 2.00 (s, 3H), 1.97 (s, 3H), 1.95 (s, 3H). **¹³C NMR** (100 MHz, CDCl₃): δ 170.3, 170.1, 169.9, 169.7, 169.1, 168.9, 145.2, 100.3, 98.9, 74.4, 74.2, 72.5, 71.7, 71.2, 68.5, 67.9, 61.6, 20.7, 20.5, 20.4, 20.3. **Anal.** Calcd for C₂₄H₃₂O₁₅ (560.50): C, 51.43; H, 5.75. Found: C, 51.31; H, 5.61%.

Cyclohexyl 3,6-Di-O-acetyl-4-O-(2,3,4,6,-tetra-O-acetyl- β -D-glucopyranosyl)-2-deoxy-2-iodo-D-mannopyranoside (**10**)

$[\alpha]_D^{25} +22.3^\circ$ (*c* 0.596, CH₂Cl₂); **IR** (KBr) 2934, 2857, 1750, 1450, 1368, 1230, 1167, 1117, 1039, 906 cm⁻¹; **¹H NMR** (600 MHz, CDCl₃): δ 5.22 (d, *J* = 1.2 Hz, 1H), 5.15 (t, *J* = 9.6 Hz, 1H), 5.04 (t, *J* = 9.6 Hz, 1H), 4.92 (dd, *J* = 9.0 Hz, *J* = 7.8 Hz, 1H), 4.69 (dd, *J* = 7.8 Hz, *J* = 3.6 Hz, 1H), 4.59 (d, *J* = 7.8 Hz, 1H), 4.44-4.41 (m, 2H), 4.29 (dd, *J* = 12.6 Hz, *J* = 5.4 Hz, 1H), 4.09 (dd,

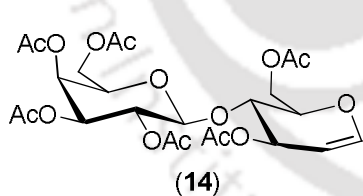
$J = 11.4$ Hz, $J = 4.8$ Hz, 1H), 4.05 (dd, $J = 12.6$ Hz, $J = 2.4$ Hz, 1H), 4.00-3.98 (m, 1H), 3.92 (t, $J = 8.4$ Hz, 1H), 3.70 (ddd, $J = 10.2$ Hz, $J = 5.4$ Hz, $J = 2.4$ Hz, 1H), 3.53-3.52 (m, 1H), 2.10 (s, 3H), 2.09 (s, 3H), 2.08 (s, 3H), 2.03 (s, 3H), 2.00 (s, 3H), 1.97 (s, 3H), 1.84-1.79 (m, 3H), 1.71-1.70 (m, 2H), 1.38-1.36 (m, 1H), 1.27-1.18 (m, 4H). ^{13}C NMR (100 MHz, CDCl_3): δ 170.3, 170.2, 169.9, 169.2, 169.1, 169.0, 100.8, 99.3, 76.7, 76.2, 72.8, 71.6, 71.5, 69.4, 69.2, 67.9, 62.1, 61.8, 33.0, 31.4, 29.4, 25.3, 23.9, 23.7, 20.8, 20.7, 20.6, 20.4. **Anal. Calcd** for $\text{C}_{30}\text{H}_{43}\text{IO}_{16}$ (786.55): C, 45.81; H, 5.51. Found: C, 45.67; H, 5.40%.

2,2',3,3',4',6,6'-Hepta-O-acetyl- α -bromo-D-lactose (13)



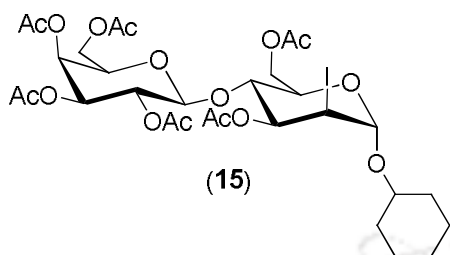
^1H NMR (400 MHz, CDCl_3): δ 6.49 (d, $J = 4.0$ Hz, 1H), 5.52 (t, $J = 9.6$ Hz, 1H), 5.32 (d, $J = 2.8$ Hz, 1H), 5.09 (dd, $J = 10.8$ Hz, $J = 8.0$ Hz, 1H), 4.92 (dd, $J = 10.0$ Hz, $J = 3.2$ Hz, 1H), 4.72 (dd, $J = 10.4$ Hz, $J = 4.0$ Hz, 1H), 4.48-4.45 (m, 2H), 4.17-4.09 (m, 3H), 4.04 (dd, $J = 11.2$ Hz, $J = 7.6$ Hz, 1H), 3.87-3.80 (m, 2H), 2.13 (s, 3H), 2.10 (s, 3H), 2.06 (s, 3H), 2.04 (s, 3H), 2.03 (s, 3H), 2.02 (s, 3H), 1.93 (s, 3H). ^{13}C NMR (150 MHz, CDCl_3): δ 170.6, 170.4, 170.3 (2C), 170.2, 169.5, 169.2, 100.9, 86.6, 75.1, 73.2, 71.2, 71.0 (2C), 69.8, 69.2, 66.8, 61.2, 61.1, 21.0, 20.8, 20.7.

3,6,2',3',4',6'-Hexa-O-acetyl-D-lactal (14)



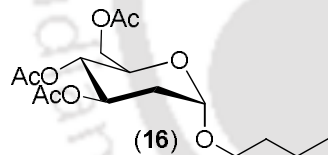
$[\alpha]_{\text{D}}^{25} +33^\circ$ (c 0.138, CH_2Cl_2); IR (KBr) 2926, 1746, 1432, 1370, 1234, 1133, 1051, 906 cm^{-1} ; ^1H NMR (600 MHz, CDCl_3): δ 6.37 (d, $J = 6.6$ Hz, 1H), 5.37 (t, $J = 3.6$ Hz, 1H), 5.32 (d, $J = 3.0$ Hz, 1H), 5.15 (dd, $J = 10.8$ Hz, $J = 7.8$ Hz, 1H), 4.97 (dd, $J = 10.2$ Hz, $J = 3.0$ Hz, 1H), 4.80 (dd, $J = 6.6$ Hz, $J = 3.6$ Hz, 1H), 4.63 (d, $J = 7.8$ Hz, 1H), 4.40 (dd, $J = 12.0$ Hz, $J = 2.4$ Hz, 1H), 4.16 (dd, $J = 11.4$ Hz, $J = 6.0$ Hz, 1H), 4.13-4.10 (m, 2H), 4.04 (dd, $J = 11.4$ Hz, $J = 7.2$ Hz, 1H), 3.96 (dd, $J = 7.2$ Hz, $J = 5.4$ Hz, 1H), 3.88 (t, $J = 6.6$ Hz, 1H), 2.12 (s, 3H), 2.08 (s, 3H), 2.05 (s, 3H), 2.03 (s, 3H), 2.01 (s, 3H), 1.94 (s, 3H). ^{13}C NMR (150 MHz, CDCl_3): δ 170.5, 170.3, 170.2, 170.0, 169.4, 145.6, 101.1, 99.1, 74.8, 74.3, 71.0, 70.8, 69.0, 68.9, 66.8, 61.9, 61.1, 21.2, 20.9, 20.8, 20.7, 20.6. **Anal. Calcd** for $\text{C}_{24}\text{H}_{32}\text{O}_{15}$ (560.50): C, 51.43; H, 5.75. Found: C, 51.33; H, 5.64%.

Cyclohexyl 3,6-Di-O-acetyl-4-O-(2,3,4,6,-tetra-O-acetyl- β -D-galactopyranosyl)-2-deoxy-2-iodo-D-mannopyranoside (**15**)



$[\alpha]_D^{25} +26^\circ$ (*c* 0.052, CH₂Cl₂); **IR** (KBr) 2928, 2854, 1750, 1369, 1222, 1118, 1080, 1046, 954, 602 cm⁻¹; **¹H NMR** (600 MHz, CDCl₃): δ 5.34 (d, *J* = 3.0 Hz, 1H), 5.22 (d, *J* = 1.8 Hz, 1H), 5.13 (dd, *J* = 10.2 Hz, *J* = 7.8 Hz, 1H), 4.97 (dd, *J* = 10.8 Hz, *J* = 3.6 Hz, 1H), 4.72 (dd, *J* = 7.8 Hz, *J* = 4.2 Hz, 1H), 4.59 (d, *J* = 8.4 Hz, 1H), 4.46-4.45 (m, 1H), 4.39 (dd, *J* = 12.0 Hz, *J* = 1.8 Hz, 1H), 4.14 (dd, *J* = 11.4 Hz, *J* = 6.6 Hz, 1H), 4.10-4.05 (m, 2H), 4.01-3.99 (m, 1H), 3.95-3.89 (m, 2H), 3.55-3.51 (m, 1H), 2.13 (s, 3H), 2.11 (s, 3H), 2.09 (s, 3H), 2.05 (s, 6H), 1.95 (s, 3H), 1.84-1.79 (m, 2H), 1.71-1.69 (m, 2H), 1.50-1.49 (m, 1H), 1.37-1.34 (m, 1H), 1.26-1.20 (m, 4H). **¹³C NMR** (150 MHz, CDCl₃): δ 170.7, 170.6, 170.3, 170.2, 169.6, 169.5, 101.6, 99.7, 77.0, 76.4, 71.2, 70.8, 70.2, 69.5, 69.4, 67.0, 62.5, 61.4, 33.3, 31.7, 29.8, 25.6, 24.2, 24.0, 21.3, 21.0, 20.9, 20.8, 20.7. **Anal. Calcd** for C₃₀H₄₃IO₁₆ (786.55): C, 45.81; H, 5.51. Found: C, 45.69; H, 5.38%.

Butyl 3,4,6-tri-O-acetyl-2-deoxy-D-glucopyranoside (**16**)



$[\alpha]_D^{25} +104^\circ$ (*c* 0.047, CH₂Cl₂); **IR** (KBr) 2959, 2925, 2867, 1745, 1438, 1367, 1229, 1129, 1047 cm⁻¹; **¹H NMR** (400 MHz, CDCl₃): δ 5.24 (ddd, *J* = 14.8 Hz, *J* = 9.6 Hz, *J* = 5.6 Hz, 1H), 4.91 (t, *J* = 9.6 Hz, 1H), 4.86 (d, *J* = 3.2 Hz, 1H), 4.22 (dd, *J* = 12.0 Hz, *J* = 4.0 Hz, 1H), 3.98 (dd, *J* = 12.0 Hz, *J* = 2.0 Hz, 1H), 3.89 (ddd, *J* = 10.4 Hz, *J* = 4.8 Hz, *J* = 2.4 Hz, 1H), 3.56 (dt, *J* = 9.6 Hz, *J* = 6.8 Hz, 1H), 3.32 (dt, *J* = 9.6 Hz, *J* = 6.4 Hz, 1H), 2.15 (ddd, *J* = 12.0 Hz, *J* = 5.6 Hz, *J* = 1.2 Hz, 1H), 2.01 (s, 3H), 1.97 (s, 3H), 1.94 (s, 3H), 1.74 (ddd, *J* = 15.2 Hz, *J* = 11.6 Hz, *J* = 3.6 Hz, 1H), 1.52-1.46 (m, 2H), 1.34-1.28 (m, 2H), 0.86 (t, *J* = 7.2 Hz, 3H); **¹³C NMR** (100 MHz, CDCl₃): δ 170.8, 170.3, 170.0, 97.0, 69.6, 69.3, 67.8, 67.6, 62.6, 35.2, 31.5, 21.0, 20.8, 19.4, 13.9. **Anal. Calcd** for C₁₆H₂₆O₈ (346.37): C, 55.48; H, 7.57. Found: C, 55.34; H, 7.45%.

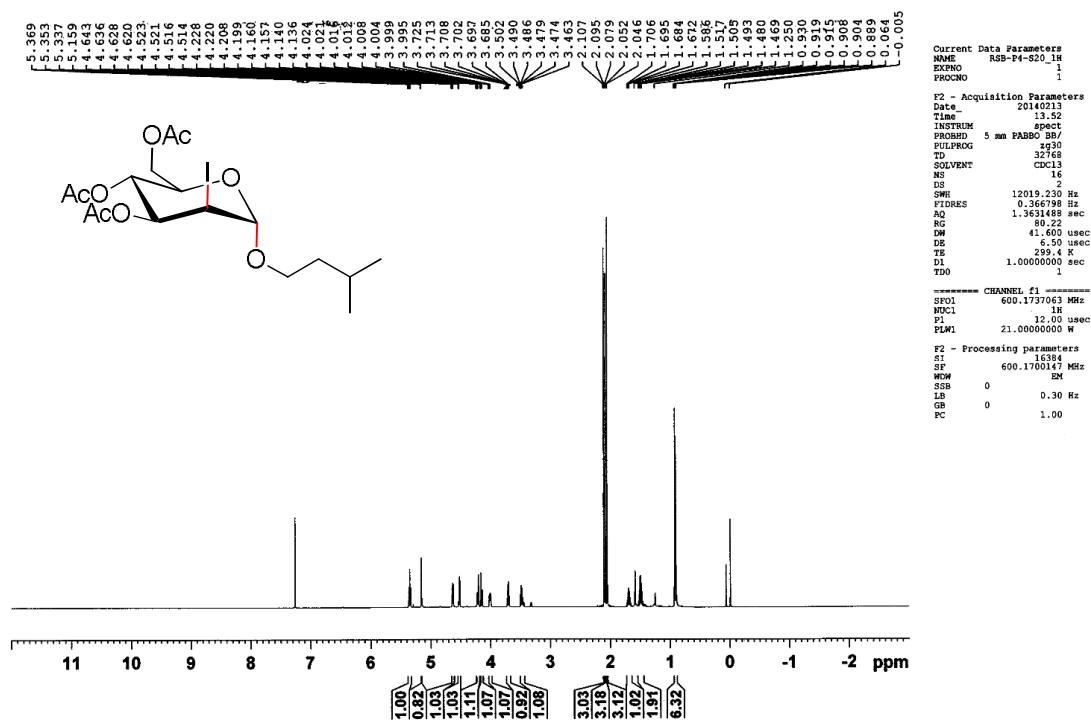
Complete crystallographic data of **6a** & **10** for the structural analyses have been deposited with the Cambridge Crystallographic Data Centre, CCDC No. 1006142 and 1006157. Copies of this information may be obtained free of charge from the Director,

Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK, (fax: +44-1223-336033, e-mail: deposit@ccdc.cam.ac.uk or via: www.ccdc.cam.ac.uk).

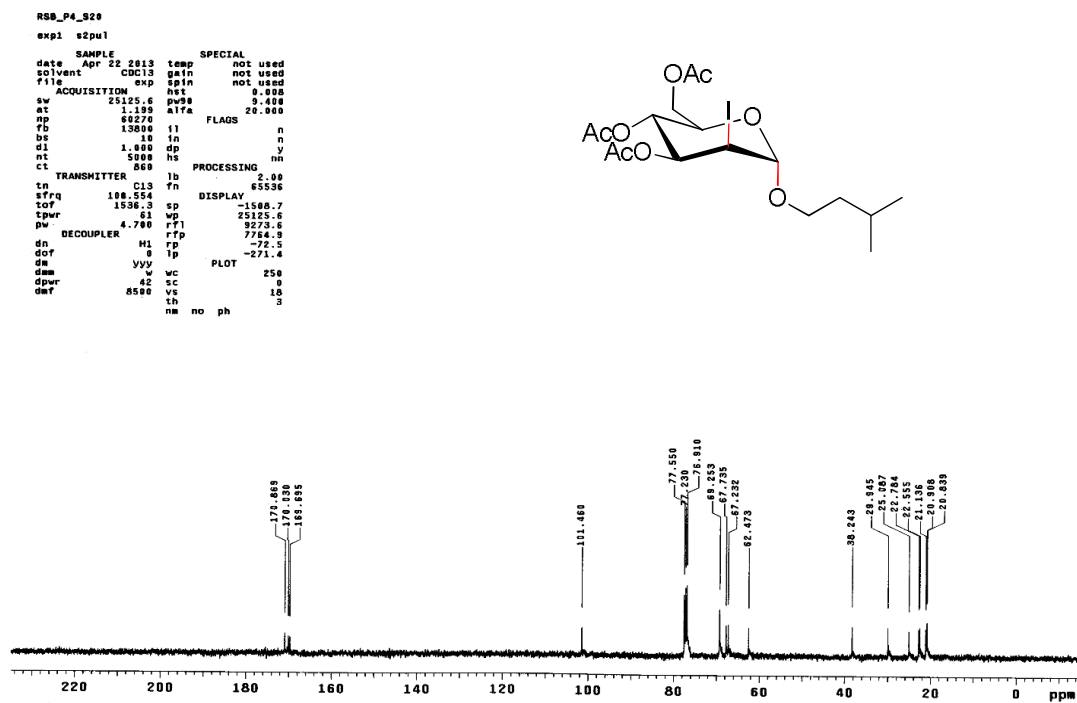
Table 4. Crystal data and structures refinement for the compound **6a** & **10**, atomic coordinates, equivalent isotropic displacement parameters and bond angles, please check the CIF.

Parameters	Compound 6a	Compound 10
Empirical Formula	'C ₁₅ H ₂₃ IO ₈ '	'C ₃₀ H ₄₃ IO ₁₆ '
Formula weight	458.23	786.54
Temperature	296 K	296 K
CCDC no	1006142	1006157
Wavelength (Å)	0.71073 Å	0.71073 Å
Crystal system	Orthorhombic	Monoclinic
Space group	P2(1)2(1)2(1)	'P 21'
Radiation type	Mo K α	Mo K α
Radiation source	'fine-focus sealed tube'	'fine-focus sealed tube'
<i>a</i> (Å)	8.4451 (2) Å	6.90610 (10) Å
<i>b</i> (Å)	9.2080 (2) Å	10.0119 (2) Å
<i>c</i> (Å)	25.7484 (6) Å	26.6266 (5) Å
α (°)	90.00°	90.00°
β (°)	90.00°	91.6930 (10)°
γ (°)	90.00°	90.00°
Cell Volume	2002.26 (8) Å ³	1840.24 (6) Å ³
<i>Z</i>	4	2
Density	1.527 g/cm ³	1.414 g/cm ³
<i>F</i> (0 0 0)	920	802
Theta range	1.58 to 28.59°	0.77 to 25.00°
Index ranges	-11 ≤ <i>h</i> ≤ 10, -12 ≤ <i>k</i> ≤ 11, -32 ≤ <i>l</i> ≤ 34	-7 ≤ <i>h</i> ≤ 8, -11 ≤ <i>k</i> ≤ 11, -31 ≤ <i>l</i> ≤ 31
Reflections collected	28963	21577
Independent reflections	5027 [R(int) = 0.0406]	6402 [R(int) = 0.0220]
Completeness to theta	28.59° 98.7 %	25.00° 99.7 %
Number of parameters	222	430
Number of restraints	0	1
Goodness-of-fit (GOF) on <i>F</i> ²	0.807	0.942
Refinement method	Full-matrix least-squares on <i>F</i> ²	Full-matrix least-squares on <i>F</i> ²

^1H NMR (600 MHz, CDCl_3): Isopentyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3c)

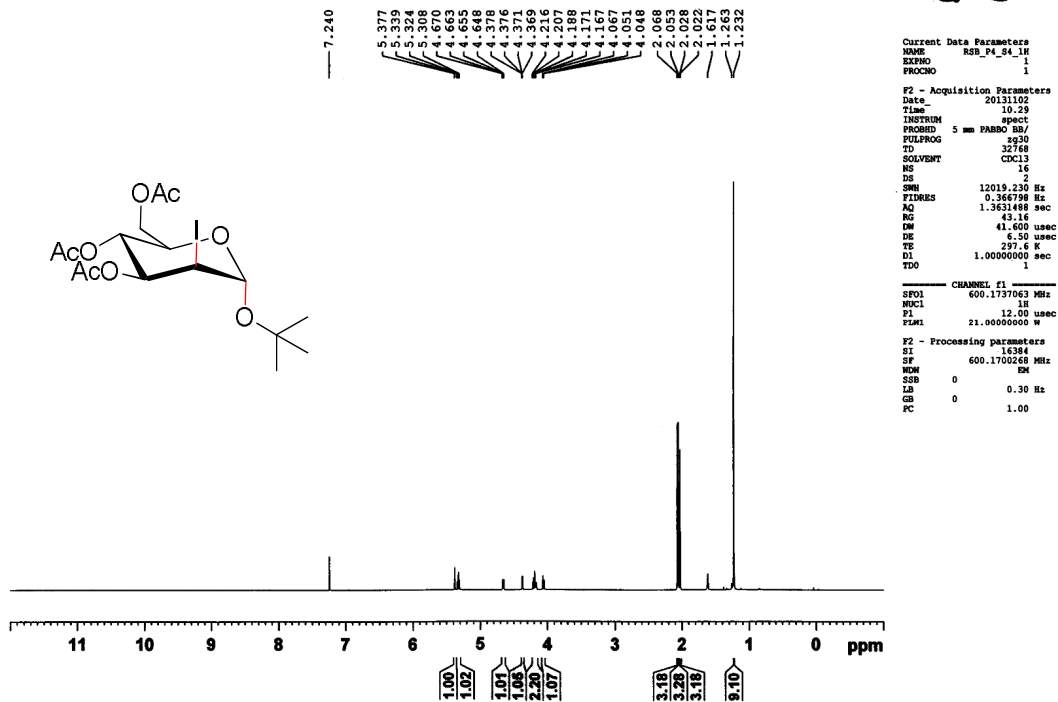


^{13}C NMR (100 MHz, CDCl_3): Isopentyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3c)



¹H NMR (600 MHz, CDCl₃): t-Butyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3j)

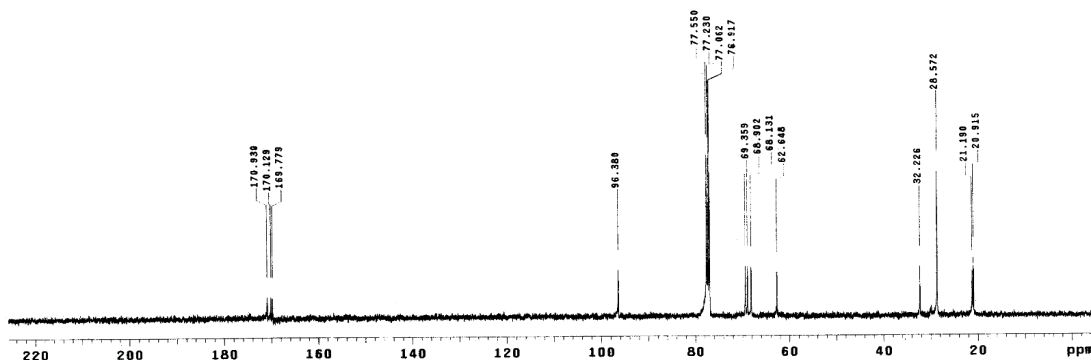
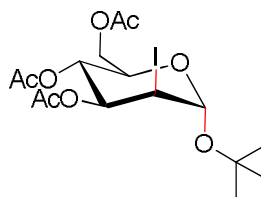
RSB_P4_S4_

**¹³C NMR (100 MHz, CDCl₃): t-Butyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3j)**

RSB_P4_TB0

exp1 s2pu1

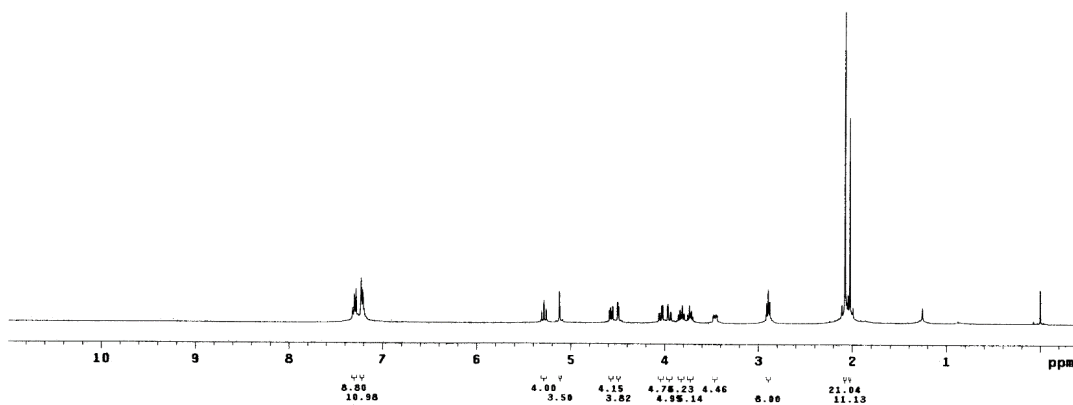
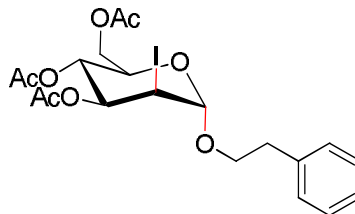
SAMPLE		SPECIAL	
date	Jul 11 2013	temp	not used
solvent	CDCl3	gain	not used
file	/export/home/~sp1n	spin	not used
mercury	RSB_P4_TB0	hst	0.000
ACQUISITION	pu09	h	5.400
sv	25125.6	alpha	20.000
at	1.199	FLDS	y
np	60270	fl	n
rb	13800	in	n
bs	10	dp	y
d1	1.000	hs	nn
nt	5000	lb	2.00
ct	5000	fn	65536
tn	TRANSMITTER	C13	DISPLAY
sfreq	100.554	sp	-549.5
tot	1936.3	wp	23245.1
tpwr	61	rfl	9271.3
pw	2.700	rfp	7764.3
DECOUPLER	H1	rp	-64.4
dn	0	lp	-271.4
dof	0	PLOT	3
da	yyy	wc	250
dsm	w	sc	0
dpr	42	vs	40
dwt	8500	th	3
	nm	no	ph



^1H NMR (400 MHz, CDCl_3): 2-Phenylethyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3I)

```

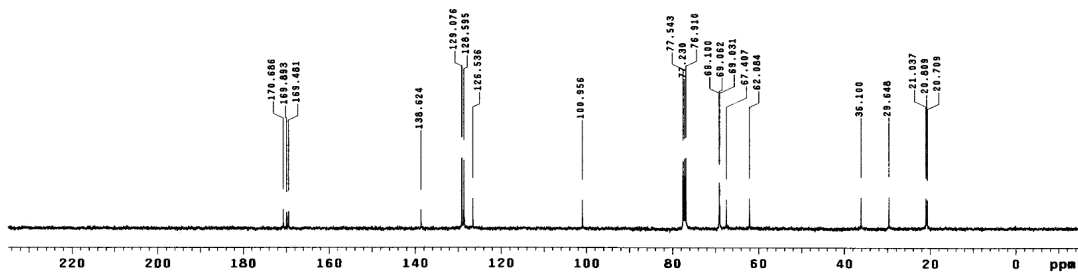
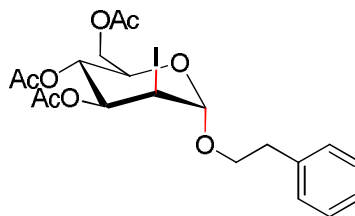
RSB_P4_S5
expl s2pu1
SAMPLE
date Dec 13 2011 temp not used
solvent CDCl3 gain not used
file not used
ACQUISITION exp spin not used
sw 6389.8 hst 0.000
at 1.998 a1fa 20.000
nd 25520
fb not used fl FLAGS n
ds 4 in n
d1 1.000 dp y
nt 32 hs PROCESSING nn
ct 32
TRANSMITTER lb fn 0.10
tn H1 f1 5536
sfreq 399.853 DISPLAY
tof 382.8 sp -166.0
tpr 57 wp 4559.3
pw 9.850 rF1 787.2
DECOUPLER C13 rf 0
dn C13 rp 110.2
dof 8 lp -95.9
dm nnn PLOT 250
dpr 50 sc 0
dwt 15900 vs 76
th 20
nm cdc ph
  
```



^{13}C NMR (100 MHz, CDCl_3): 2-Phenylethyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-mannopyranoside (3I)

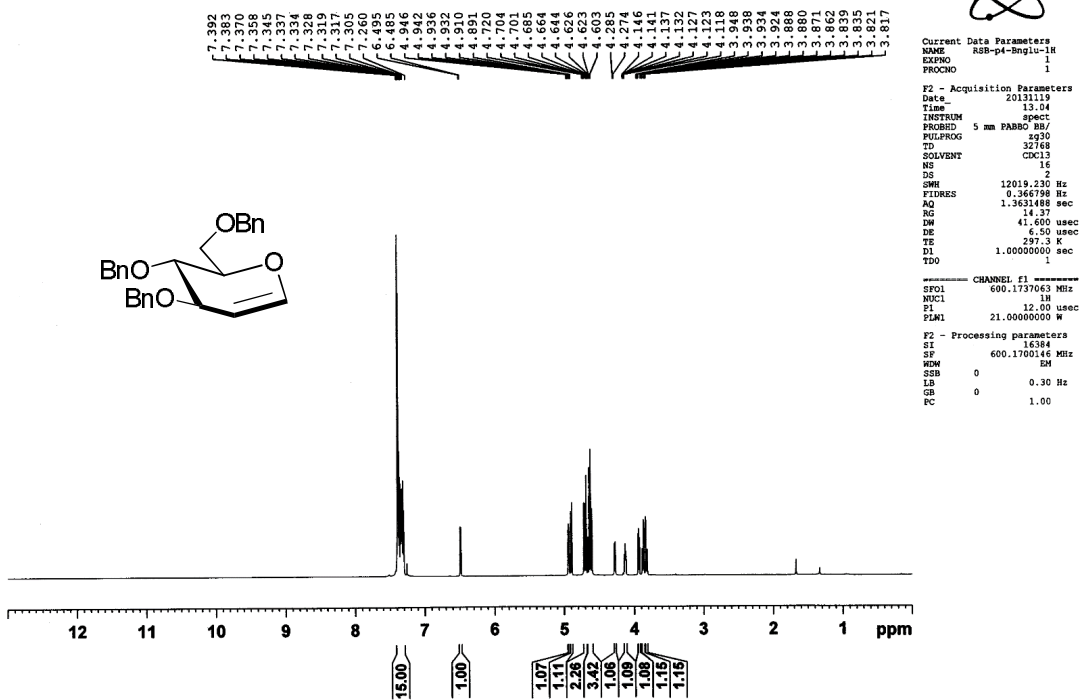
```

RSB_P4_S5
expl s2pu1
SAMPLE
date Dec 13 2011 temp not used
solvent CDCl3 gain not used
file not used
ACQUISITION exp spin not used
sw 25125.6 hst 0.000
at 1.199 a1fa 20.000
nd 80270
fb 13800 fl FLAGS n
bs 10 in n
d1 1.000 dp y
nt 5000 hs PROCESSING nn
ct 1420
TRANSMITTER lb fn 2.00
tn H1 f1 5536
sfreq 100.554 DISPLAY
tof 1536.3 sp -1318.4
tpr 61 wp 25125.6
pw 9.300 rF1 9281.3
DECOUPLER C13 rf 7764.3
dn H1 rp -58.6
dof 8 lp -339.8
dm yvy PLOT 250
dpr 42 sc 0
dwt 8900 vs 18
th nm no ph
  
```

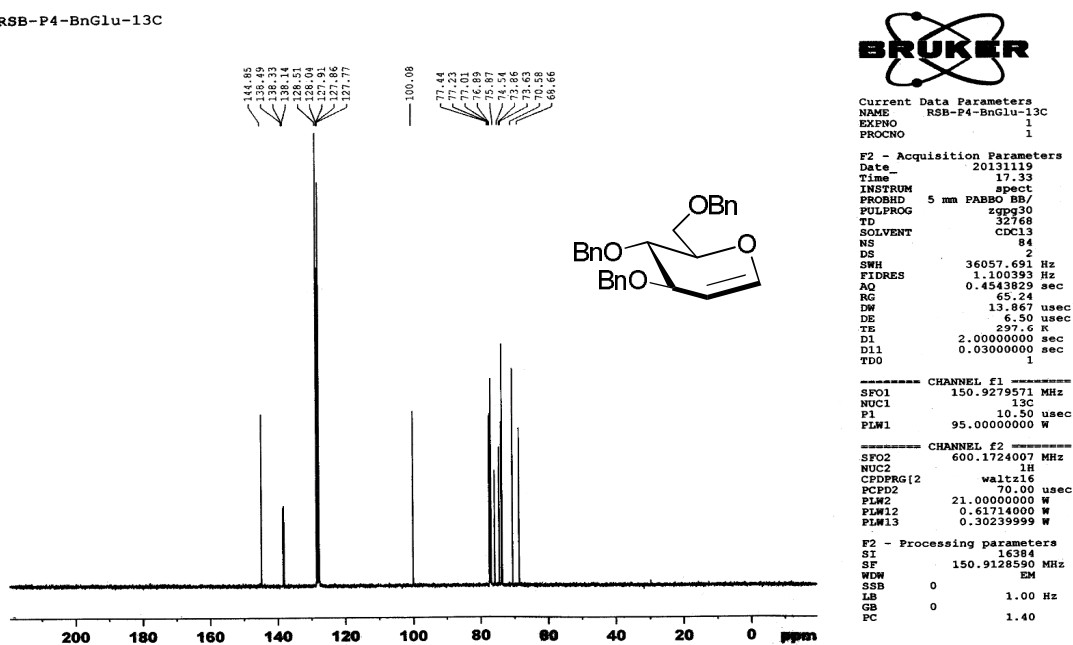


¹H NMR (600 MHz, CDCl₃): 3,4,6-Tri-O-benzyl-D-glucal (4)

RSB-p4-Bnglu-1H

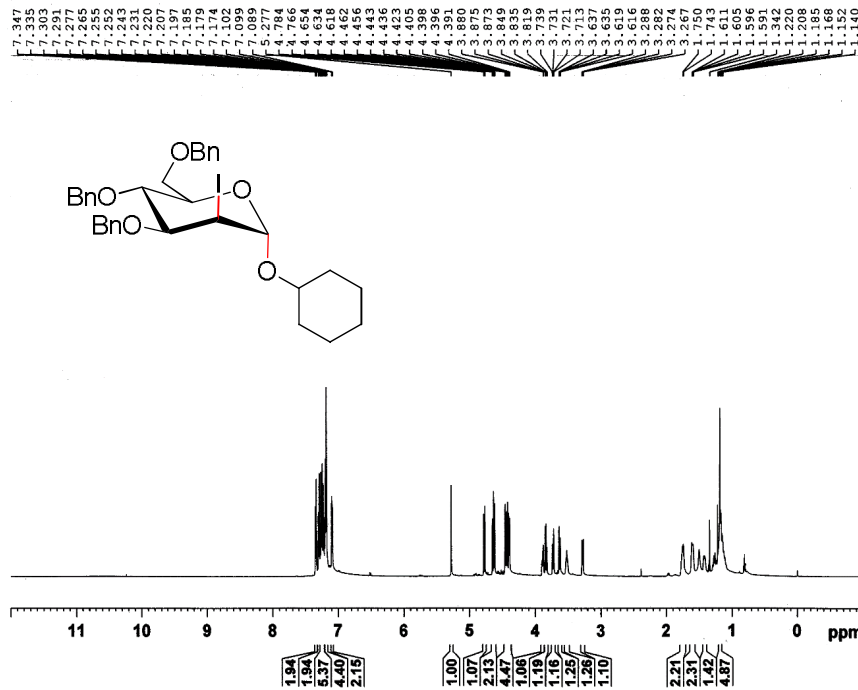
¹³C NMR (150 MHz, CDCl₃): 3,4,6-Tri-O-benzyl-D-glucal (4)

RSB-P4-BnGlu-13C



^1H NMR (600 MHz, CDCl_3): Cyclohexyl 3,4,6-tri-*O*-benzyl-2-deoxy-2-iodo-*D*-mannopyranoside (**4a**)

RSB-p4-s27-1H



Current Data Parameters
 NAME RSB-p4-s27-1H
 EXPNO 1
 PROCNO 1

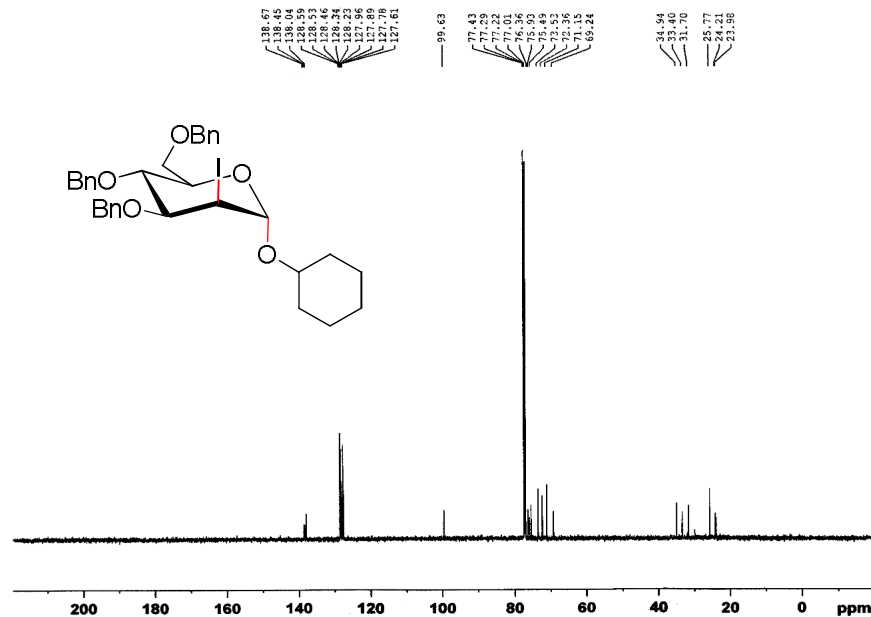
F2 - Acquisition Parameters
 Date_ 20131128
 Time 9.53
 INSTRUM spect
 PROBHD 5 mm PABBO BB/
 PULPROG zgpg30
 TD 32768
 SOLVENT CDCl_3
 NS 16
 DS 2
 SWH 12019.230 Hz
 FIDRES 0.366798 Hz
 AQ 1.3631488 sec
 RG 65.24
 DW 41.460 usec
 DE 6.50 usec
 TE 299.5 K
 D1 1.00000000 sec
 TDO 1

CHANNEL f1
 SFO1 600.1737063 MHz
 NUC1 ^1H
 P1 12.00 usec
 PLW1 21.00000000 W

F2 - Processing parameters
 SI 16384
 SF 600.1700635 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

^{13}C NMR (150 MHz, CDCl_3): Cyclohexyl 3,4,6-tri-*O*-benzyl-2-deoxy-2-iodo-*D*-mannopyranoside (**4a**)

RBS-P4-S27_13c



Current Data Parameters
 NAME RSB-p4-s27_13c
 EXPNO 1
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20131128
 Time 15.30
 INSTRUM spect
 PROBHD 5 mm PABBO BB/
 PULPROG zgpg30
 TD 32768
 SOLVENT CDCl_3
 NS 159
 DS 2
 SWH 36057.691 Hz
 FIDRES 1.100393 Hz
 AQ 0.4543829 sec
 RG 65.24
 DW 19.867 usec
 DE 6.50 usec
 TE 299.5 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TDO 1

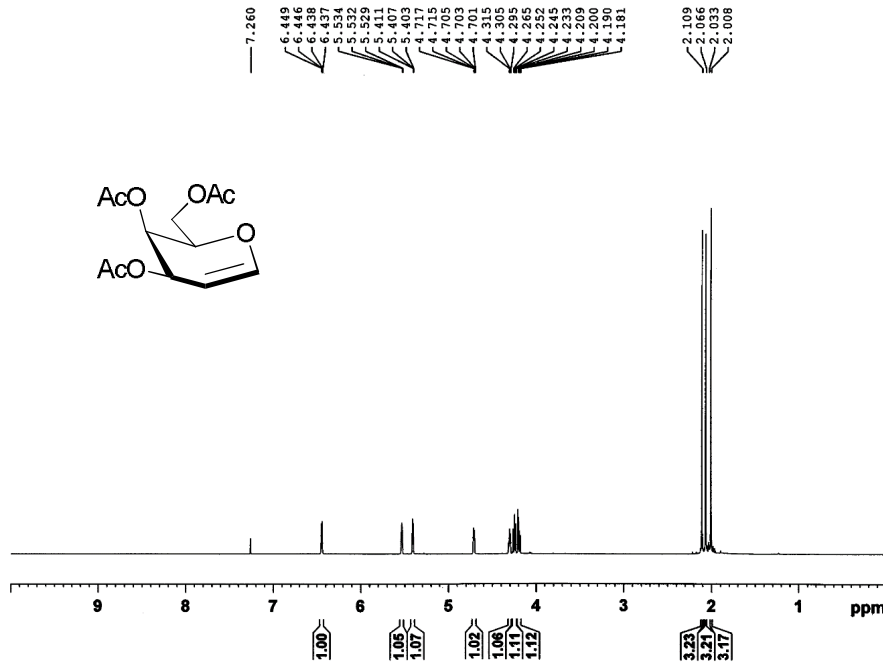
CHANNEL f1
 SFO1 150.9279571 MHz
 NUC1 ^{13}C
 P1 10.50 usec
 PLW1 95.00000000 W

CHANNEL f2
 SFO2 600.1724007 MHz
 NUC2 ^1H
 CPDPRG2 waltz16
 PCPD2 70.00 usec
 PLW2 21.00000000 W
 PLW12 0.61714000 W
 PLW13 0.30239999 W

F2 - Processing parameters
 SI 16384
 SF 150.9128392 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40

^1H NMR (600 MHz, CDCl_3): 3,4,6-Tri-O-acetyl-D-galactal (5)

RSB-D-Gal-1H



```

Current Data Parameters
NAME      RSB-D-Gal-1H
EXPNO    1
PROCNO   1

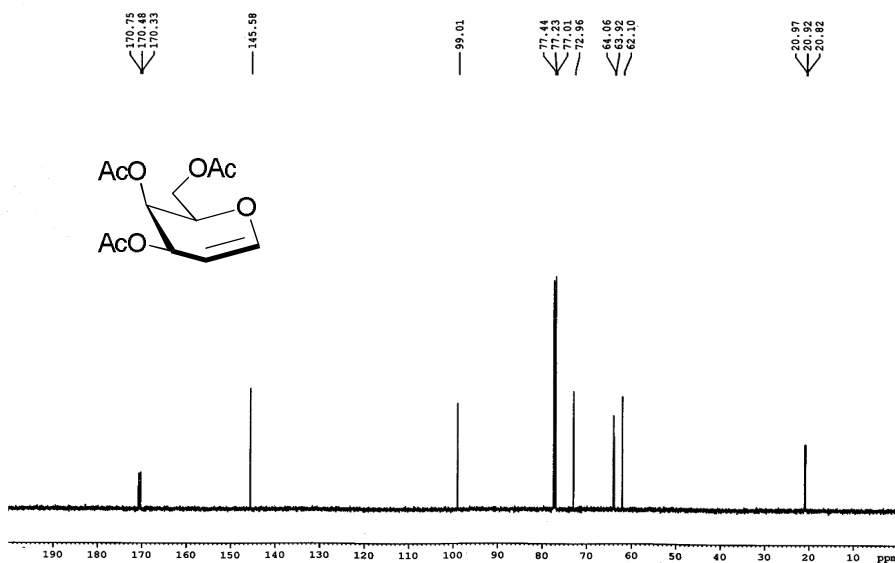
F2 - Acquisition Parameters
Date_    20131125
Time     9.58
INSTRUM  spect
PROBHD   5 mm PABBO BB/
PULPROG  zgpg30
TD       32768
SOLVENT  CDCl3
NS       16
DS       2
SWH      12019.230 Hz
FIDRES   0.366798 Hz
AQ       1.363168 sec
RG       43.16
DW       41.600 usec
DE       6.50 usec
TE       297.2 K
D1       1.00000000 sec
TDO      1

===== CHANNEL f1 =====
SFO1    600.137065 MHz
NUC1     1H
P1       12.00 usec
PL1      21.00000000 W

F2 - Processing parameters
SI       16384
SF       600.1700148 MHz
WDW      EM
SSB      0
LA       0
GB       0
PC       1.00
  
```

 ^{13}C NMR (150 MHz, CDCl_3): 3,4,6-Tri-O-acetyl-D-galactal (5)

RSB-DGOL-113C



```

Current Data Parameters
NAME      RSB-DGOL-113C
EXPNO    1
PROCNO   1

F2 - Acquisition Parameters
Date_    20131125
Time     15.49
INSTRUM  spect
PROBHD   5 mm PABBO BB/
PULPROG  zgpg30
TD       65536
SOLVENT  CDCl3
NS       32
DS       2
SWH      36057.692 Hz
FIDRES   1.100393 Hz
AQ       0.4543029 sec
RG       200.18
DW       13.967 usec
DE       6.50 usec
TE       297.2 K
D1       2.00000000 sec
D11      0.03000000 sec
TDO      3

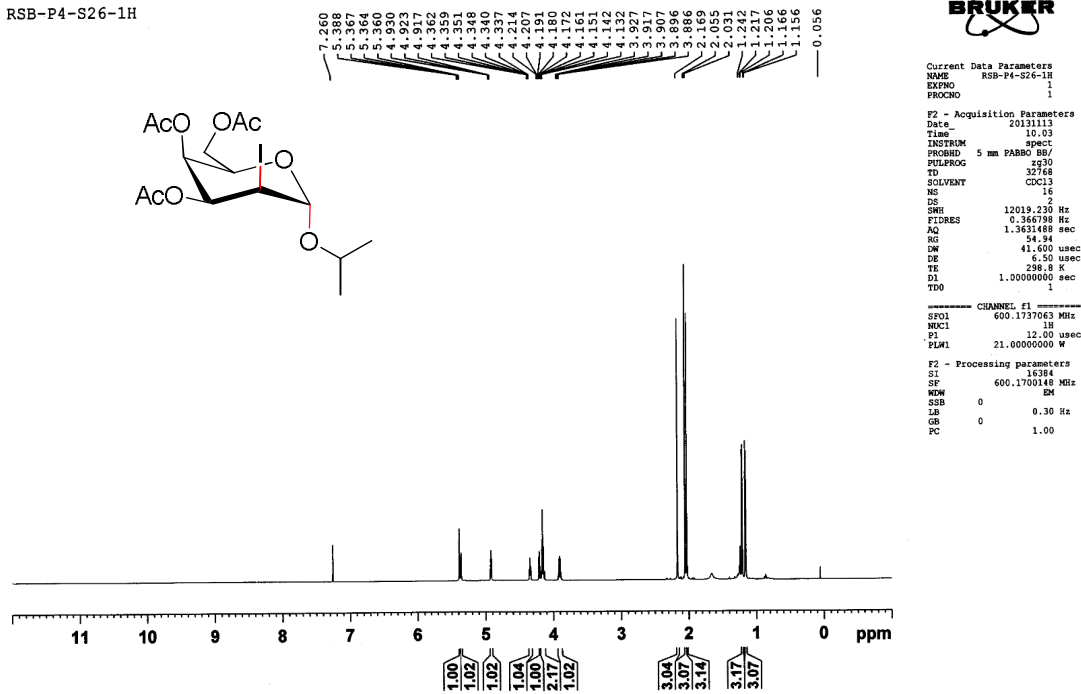
===== CHANNEL f1 =====
SFO1    150.9278571 MHz
NUC1     13C
P1       10.00 usec
PL1      95.00000000 W

===== CHANNEL f2 =====
SFO2    600.1724007 MHz
NUC2     1H
CPDPRG2  waltz16
PCPD2    70.00 usec
PLW2     21.00000000 W
PLW12    0.61714000 W
PLW13    0.30239999 W

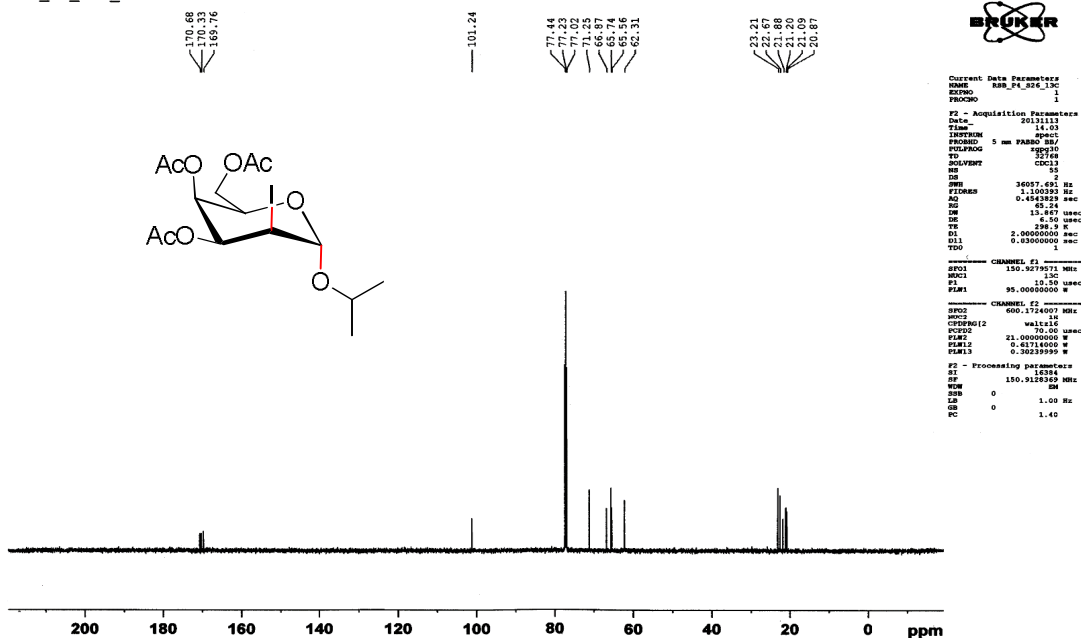
F2 - Processing parameters
SI       16384
SF       150.9124810 MHz
WDW      EM
SSB      0
LA       0
GB       0
PC       1.40
  
```

¹H NMR (600 MHz, CDCl₃): Isopropyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-talopyranoside (6a)

RSB-P4-S26-1H

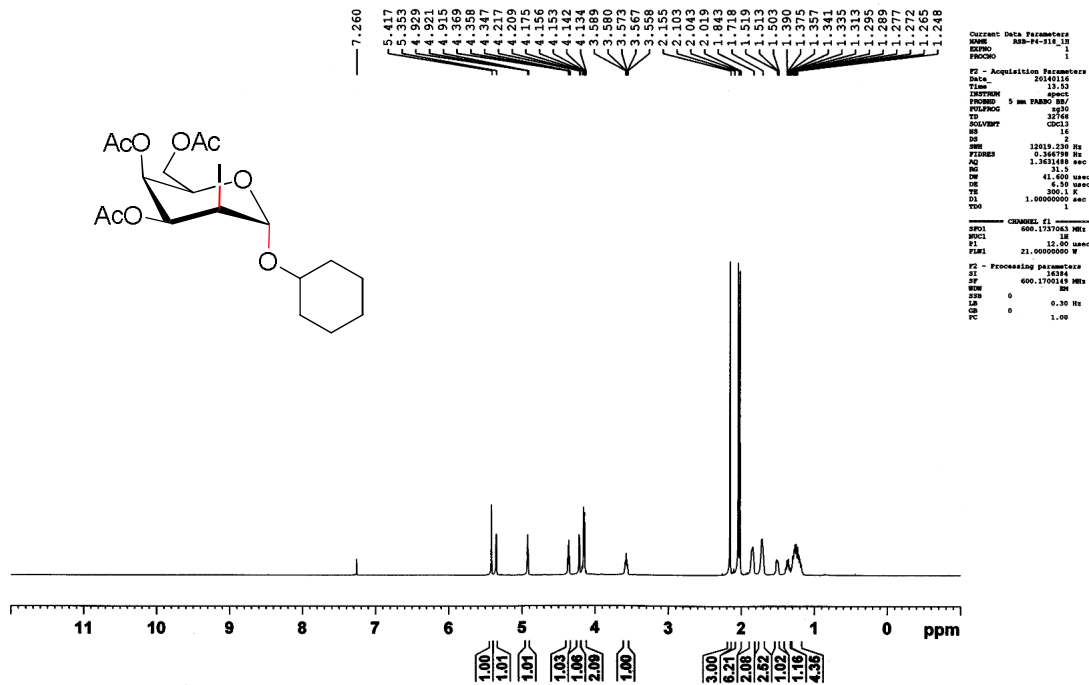
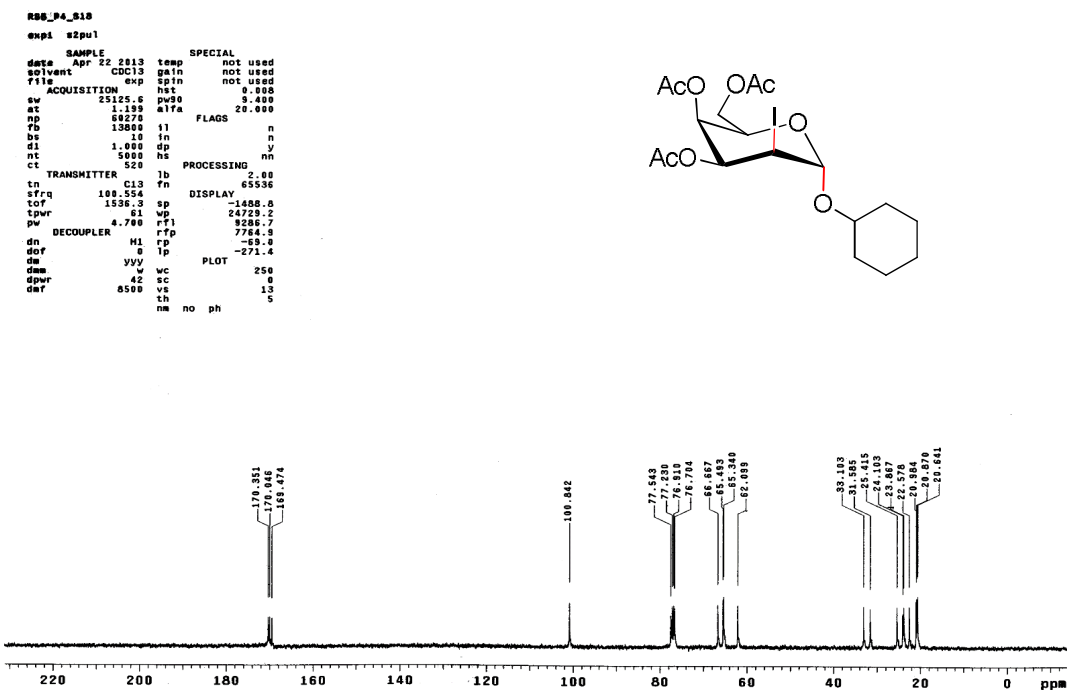
**¹³C NMR (150 MHz, CDCl₃): Isopropyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-talopyranoside (6a)**

RSB_P4_S26_13C



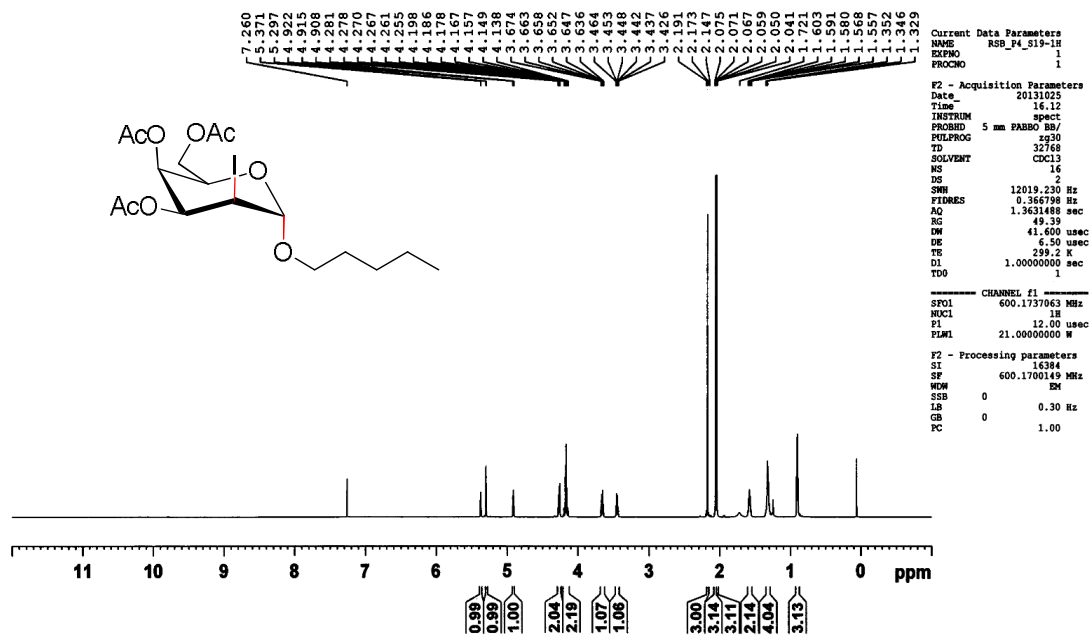
¹H NMR (600 MHz, CDCl₃): Cyclohexyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-talopyranoside (6b)

RSB-P4-S18_1H

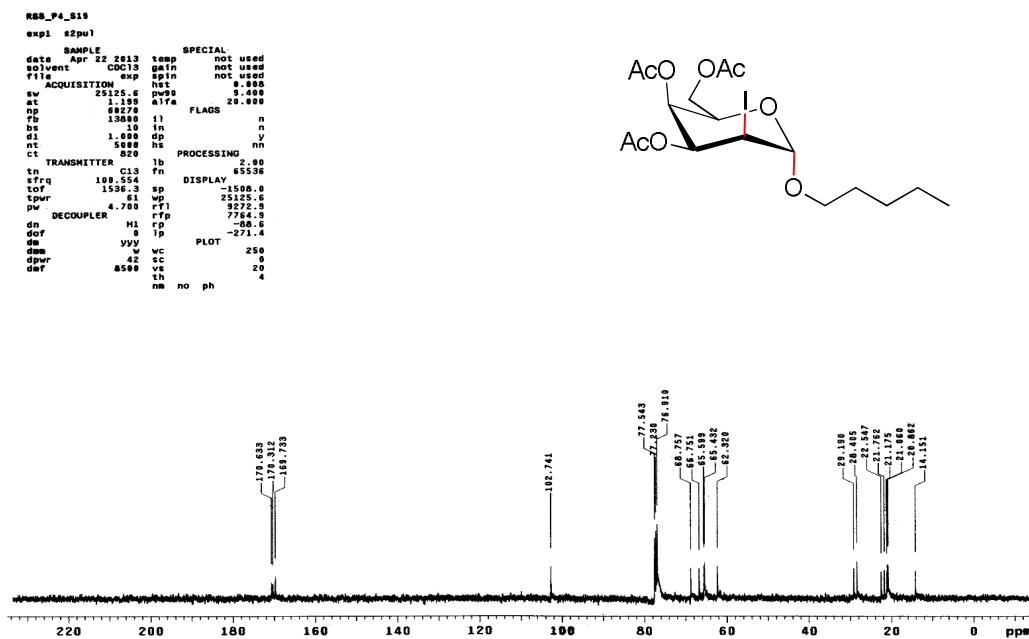
**¹³C NMR (100 MHz, CDCl₃): Cyclohexyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-talopyranoside (6b)**

¹H NMR (600 MHz, CDCl₃): Pentyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-talopyranoside (6d)

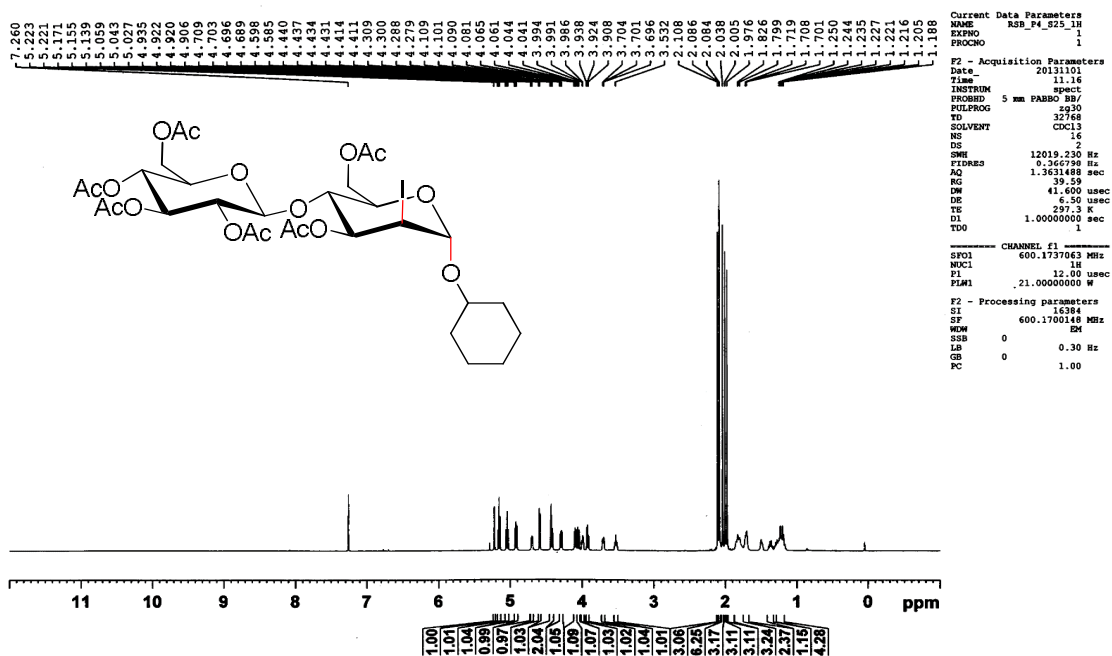
RSB_P4_S19



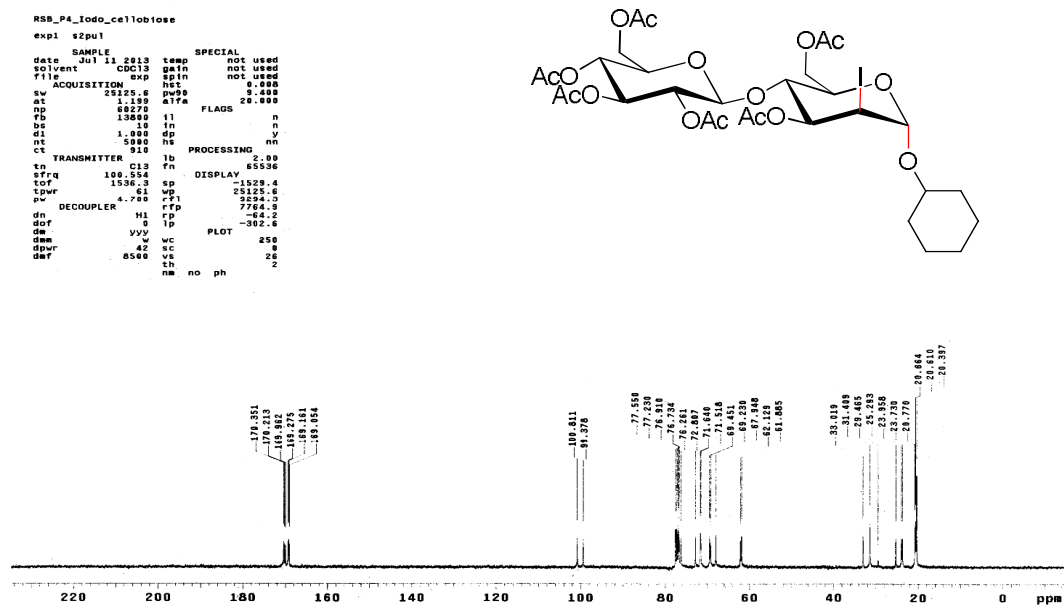
¹³C NMR (100 MHz, CDCl₃): Pentyl 3,4,6-tri-O-acetyl-2-deoxy-2-iodo-D-talopyranoside (6d)



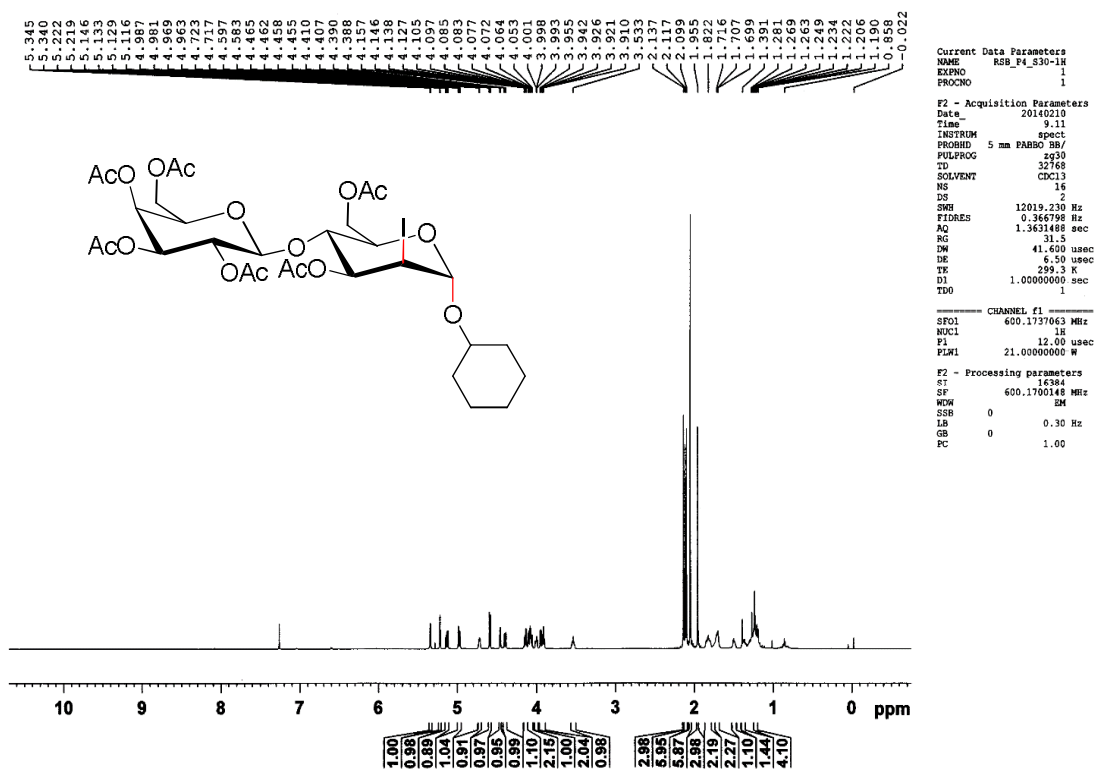
^1H NMR (600 MHz, CDCl_3): Cyclohexyl 3,6-Di-O-acetyl-4-O-(2,3,4,6,-tetra-O-acetyl- β -D-glucopyranosyl)-2-deoxy-2-iodo-D-mannopyranoside (**10**)



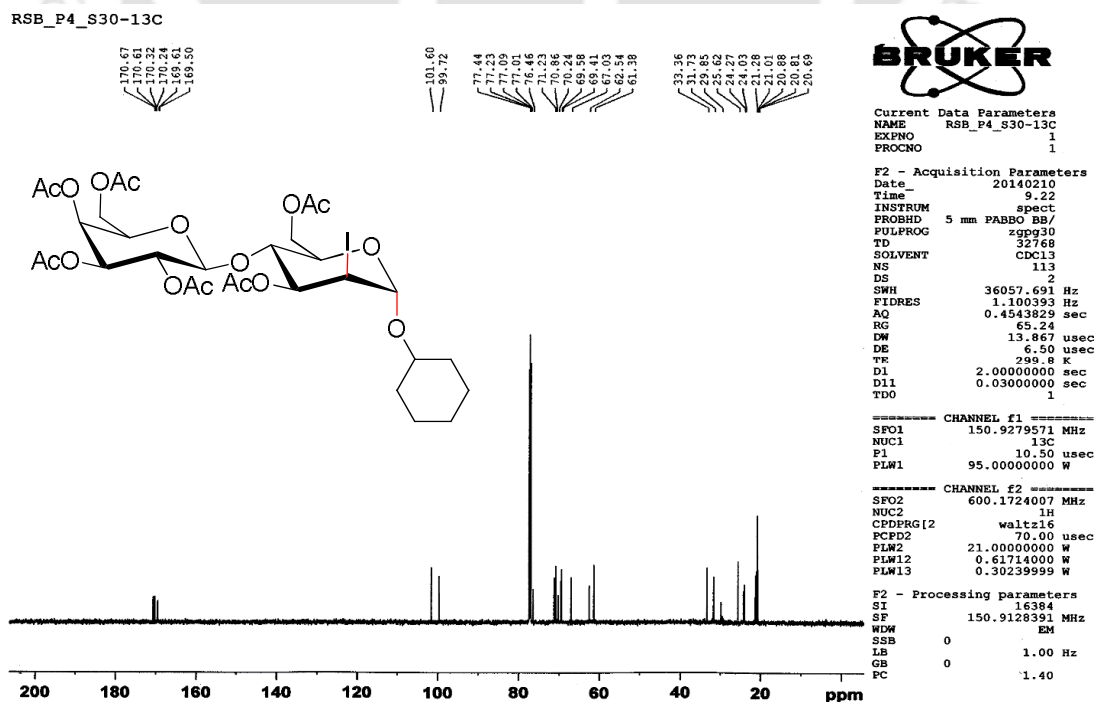
^{13}C NMR (100 MHz, CDCl_3): Cyclohexyl 3,6-Di-O-acetyl-4-O-(2,3,4,6,-tetra-O-acetyl- β -D-glucopyranosyl)-2-deoxy-2-iodo-D-mannopyranoside (**10**)



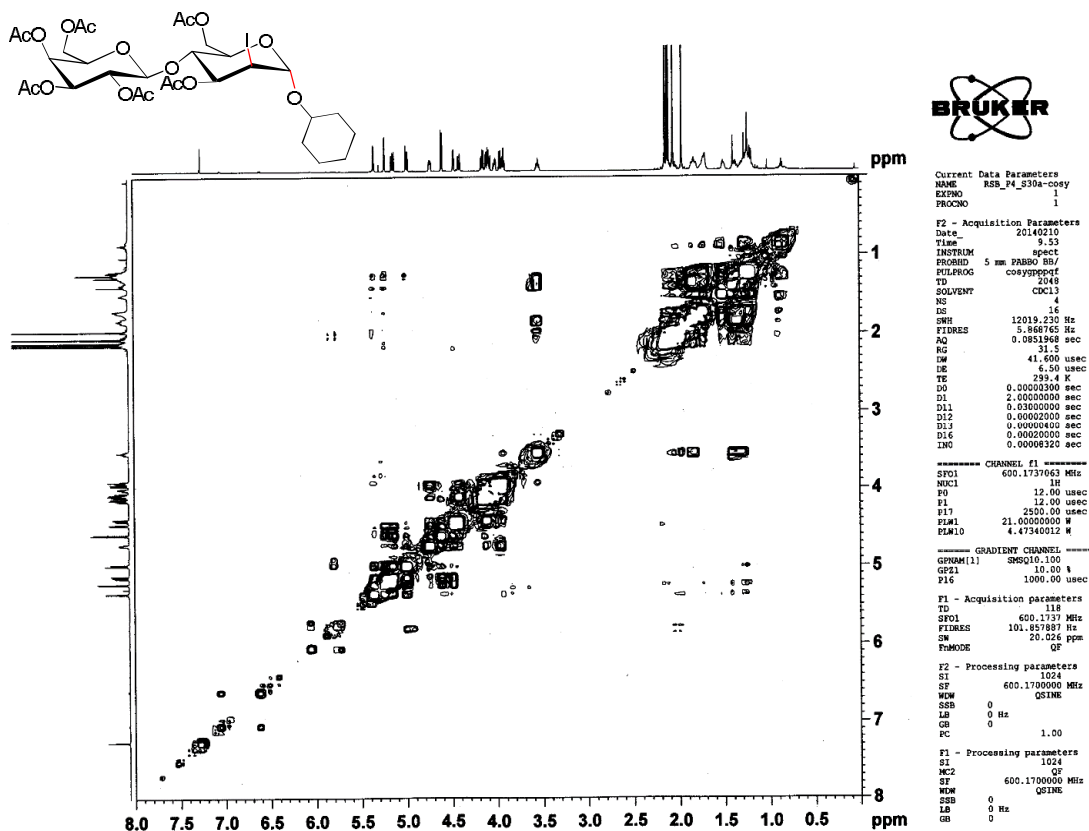
¹H NMR (600 MHz, CDCl₃): Cyclohexyl 3,6-Di-O-acetyl-4-O-(2,3,4,6,-tetra-O-acetyl-β-D-galactopyranosyl)-2-deoxy-2-iodo-D-mannopyranoside (15)



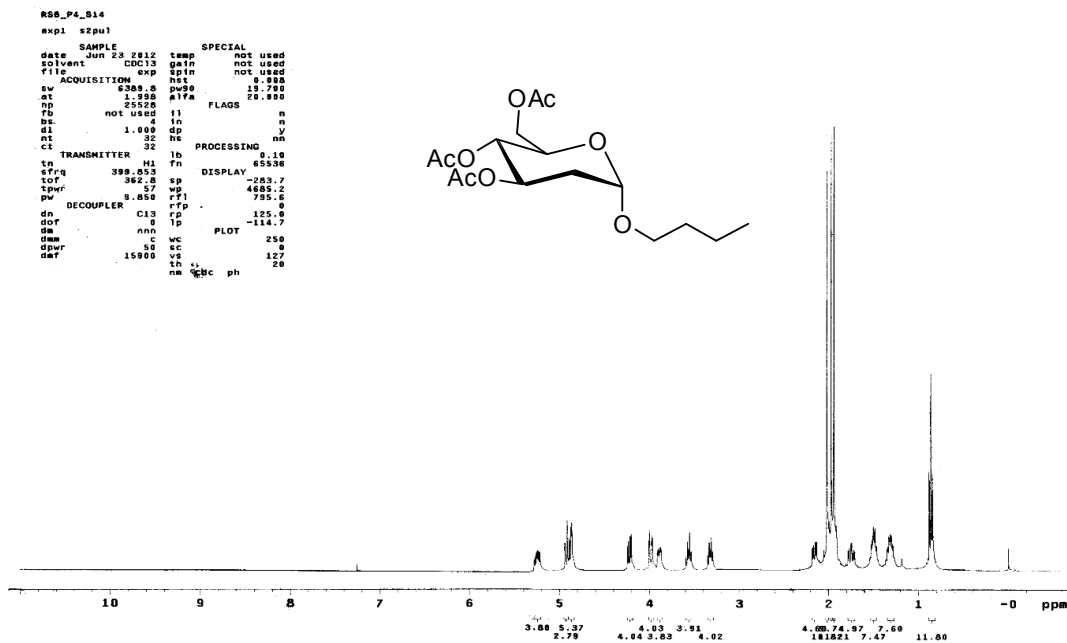
¹³C NMR (150 MHz, CDCl₃): Cyclohexyl 3,6-Di-O-acetyl-4-O-(2,3,4,6,-tetra-O-acetyl-β-D-galactopyranosyl)-2-deoxy-2-iodo-D-mannopyranoside (15)



^1H - ^1H COSY (600 MHz, CDCl_3): Cyclohexyl 3,6-Di-O-acetyl-4-O-(2,3,4,6-tetra-O-acetyl- β -D-galactopyranosyl)-2-deoxy-2-iodo-D-mannopyranoside (15)

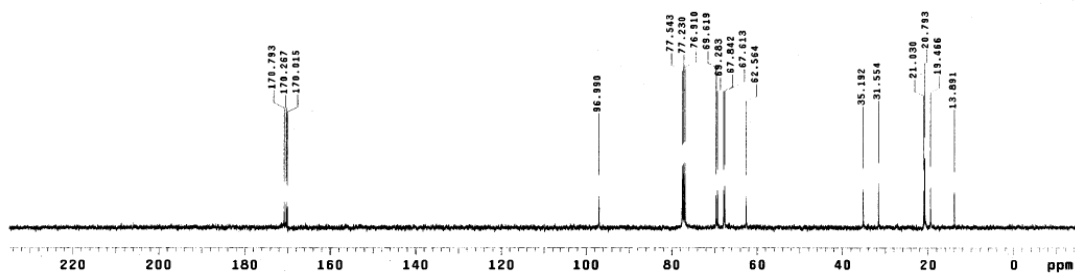
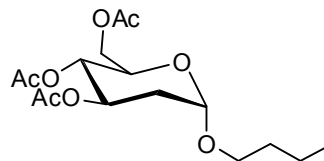


^1H NMR (400 MHz, CDCl_3): Butyl 3,4,6-tri-O-acetyl-2-deoxy-D-glucopyranoside (16)



^{13}C NMR (100 MHz, CDCl_3): Butyl 3,4,6-tri-*O*-acetyl-2-deoxy-*D*-glucopyranoside (16)

```
RSB_P4_S14
exp1 s2pu1
SAMPLE
date Jun 23 2012 temp SPECIAL
solvent CDC13 gain not used
file exp spin not used
ACQUISITION hst 0.000
sv 25125.6 pv90 18.000
at 1.130 aifa 20.000
np 08270 FLAGS
fb 13800 fl n
bs 10 in n
d1 1.000 dp y
nt 5000 hs nn
ct PROCESSING
tn C13 fb 2.00
sfrq 100.554 rn 65536
tof 1536.3 sp DISPLAY
tpr 61 wp 25125.6
pv 9.300 rF1 9275.2
dn DECOUPLER H1 rp 7784.9
dof 0 lp -84.3
dm yvj w PLOT -271.4
dam 42 sc 250
dpr 42 vs 0
dft 8900 th 17
nm no ph 4
```



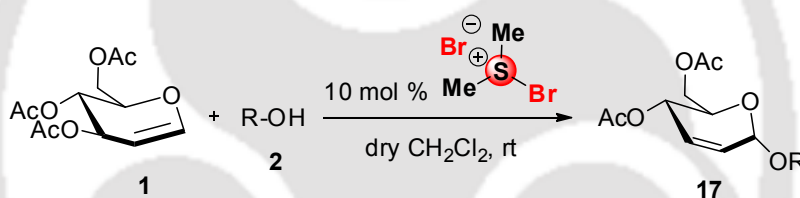


Results and Discussion

*Bromodimethyl
sulfonium bromide
(BDMS) Catalyzed
Synthesis of 2,3-
unsaturated-O-
glycosides via
Ferrier
rearrangement*

● Results and Discussion

The importance of Ferrier rearrangement on glucal and their emerging synthetic methods were discussed in Chapter I. Literature survey suggested that earlier reported methods suffered from the lack of anomeric selectivity. Thus there is a scope to develop an efficient and mild reaction conditions for the synthesis of 2,3-unsaturated-*O*-glycosides. The utilities of bromodimethylsulfonium bromide (BDMS) for various organic transformations have been demonstrated by our research group⁷² as well as by others.⁷³ BDMS performs a crucial role as promoters⁷⁴ for glycosylation reaction. Its application in numerous synthetic transformations is also reviewed by us.⁷⁵ In this Chapter, a simple and efficient synthetic protocol was accomplished for *O*-glycosidation of 3,4,6-tri-*O*-acetyl-*D*-glucal via Ferrier rearrangement using bromodimethylsulfonium bromide (BDMS) as a catalyst for the synthesis of 2,3-unsaturated-*O*-glycosides, which is shown in Scheme 40.



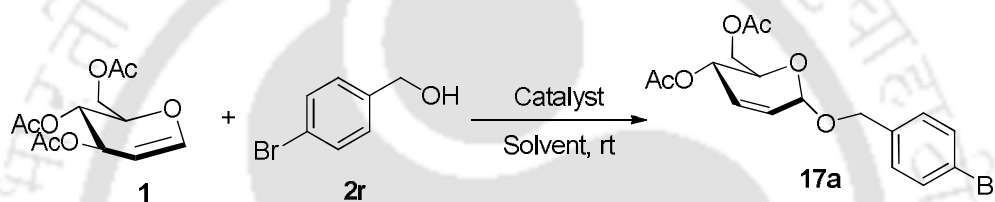
□ **Scheme 40.** Synthesis of 2,3-unsaturated-*O*-glycosides

Initial studies were performed on reaction with 3,4,6-tri-*O*-acetyl-*D*-glucal (**1**) and *p*-bromobenzylalcohol (**2r**) as model substrates to ascertain the optimal reaction conditions. However, the reactions were scrutinized with different mol % of BDMS in dry dichloromethane (Table 5, entries 1-3) and the isolated product **17a** was characterized by ¹H and ¹³C NMR spectroscopy. In addition, the reactions were performed with different solvents such as acetonitrile, dry ether and mixture of solvent system dichloromethane/acetonitrile (1:1) and their successive results are depicted in Table 5 (entries 4-6). The screening of the catalyst were also examined with metal triflates such as Co(OTf)₂ and Mn(OTf)₂ (Table 5, entries 7 and 8), which gave the required product **17a** with poor selectivity and low yield. Analyzing the obtained result revealed that 10 mol% BDMS in dichloromethane was found to be most effective for the synthesis of 2,3-unsaturated-*O*-glycosides. It is important to note that the reaction time decreases, on increasing the catalyst amount, but there is a significant lack of increase on yield. It is

also significant that the catalyst BDMS plays a vital role for high α -selectivity and good yield.

Furthermore, reaction were performed with 3,4,6-tri-*O*-acetyl-*D*-glucal (**1**) and benzyl alcohol (**2k**) in presence of 10 mol% BDMS under optimized reaction conditions and the expected product **17b** was obtained in 92% yield. Likewise, on reaction with a variety of alcohols such as 2-phenylethanol (**2l**), 3-phenylpropanol (**2m**), 4-phenylbutanol (**2n**) and 5-phenylpentanol (**2s**) in presence of 10 mol% catalyst under identical reaction conditions and the isolated products (**17c-f**) were obtained from 65-90% yield, which is shown in Table 6 (entries 3-6).

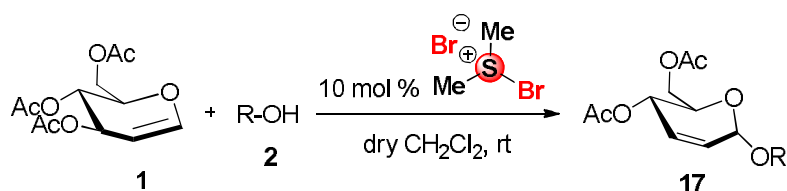
Table 5. Optimization of the reaction conditions^a



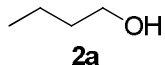
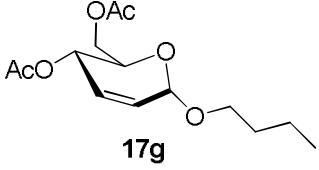
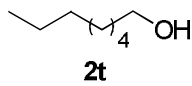
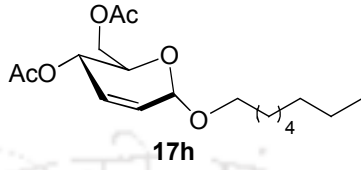
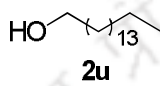
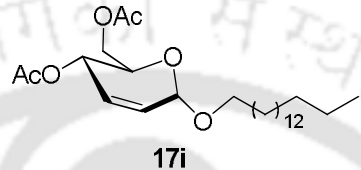
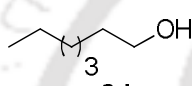
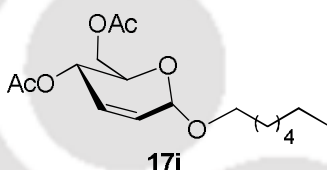

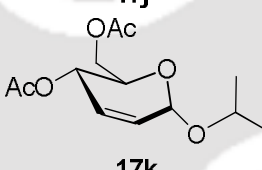
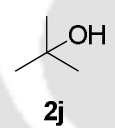
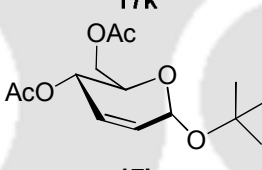
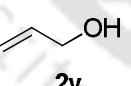
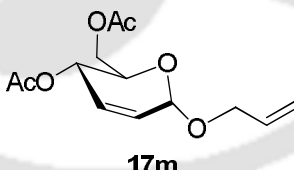
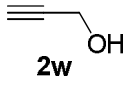
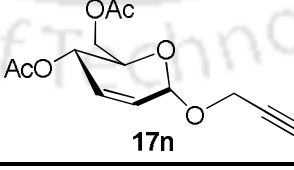
Entry	Catalyst	Solvent	Time (h)	% Yield ^b (α/β) ^c
1	5 mol% BDMS	DCM	3.5	78 (5:1)
2	10 mol% BDMS	DCM	2	85 (7:1)
3	15 mol% BDMS	DCM	1.5	83 (7:1)
4	5 mol% BDMS	CH ₃ CN	4	72 (5:1)
5	10 mol% BDMS	Et ₂ O	2.5	76 (6:1)
6	10 mol% BDMS	CH ₃ CN/DCM	2.5	80 (6:1)
7	5 mol% Co(OTf) ₂	CH ₃ CN	6	56 (1:1)
8	5 mol% Mn(OTf) ₂	CH ₃ CN	4	63 (2:1)

^aAll the reactions were performed using 1 mmol scale of 3,4,6-tri-*O*-acetyl-*D*-glucal and 1.1 *p*-bromobenzylalcohol at room temperature. ^bIsolated yield. ^cThe ratio $\alpha:\beta$ was determined from the ¹H-NMR.

Moreover, the present protocol was further investigated with alcohols such as primary, secondary, tertiary, allylic and propargyl alcohols with 3,4,6-tri-*O*-acetyl-*D*-glucal (**1**) in presence of catalytic amount of BDMS under similar reaction conditions and the obtained products (**17g-n**) were isolated in good yields which is shown in Table 6 (entries 7-14).

Table 6. BDMS catalyzed Ferrier rearrangement of 3,4,6-tri-*O*-acetyl-D-glucal (**1**) with alcohol (**2**)^a

S.No	Alcohol (2)	Product (17)	Time (h)	Yield (%) ^b	(α : β) ^c
1			2	85	7:1
2			0.5	92	8:1
3			0.5	90	9:1
4			2.5	75	12:1
5			2	72	4:1
6			3	65	3:1

7			0.5	86	6:1
8			1	78	8:1
9			2	60	α
10			1.5	62	4:1
11			1	88	10:1
12			1	85	α
13			0.5	87	18:1
14			0.5	90	α

^aAll the reactions were carried out with 1 mmol 3,4,6-tri-*O*-acetyl-D-glucal, 1.1 mmol alcohol using 10 mol% BDMS as catalyst. ^bIsolated yield. ^cThe ratio (α : β) was determined by anomeric proton integration from ¹H-NMR.

It is noteworthy that alcohols such as cetyl alcohol (**2u**), *tert*-butyl alcohol (**2j**), propargyl alcohol (**2w**) afford only α anomer of 2,3-unsaturated-*O*-glycosides. All the isolated products were characterized through ¹H NMR, ¹³C NMR, specific rotation, and

elemental analysis. Furthermore, the structure of product **17n** was determined by single-crystal X-ray data, which is shown in Figure 4a.

Crystal engineering attribute significant attention in recent years due to their immense applications in drug design and molecular biology. The impact of weak interactions between the molecules leads to different arrangement of the engineered crystal motif.⁷⁶ Interestingly, the intermolecular interaction between O6...H15 in **17n** which shows prominent cavity architecture and O4...C9 & O5...H14C interactions in **17n** reflect an extensive linear network assembly which is shown in Figure 4b & 4c.

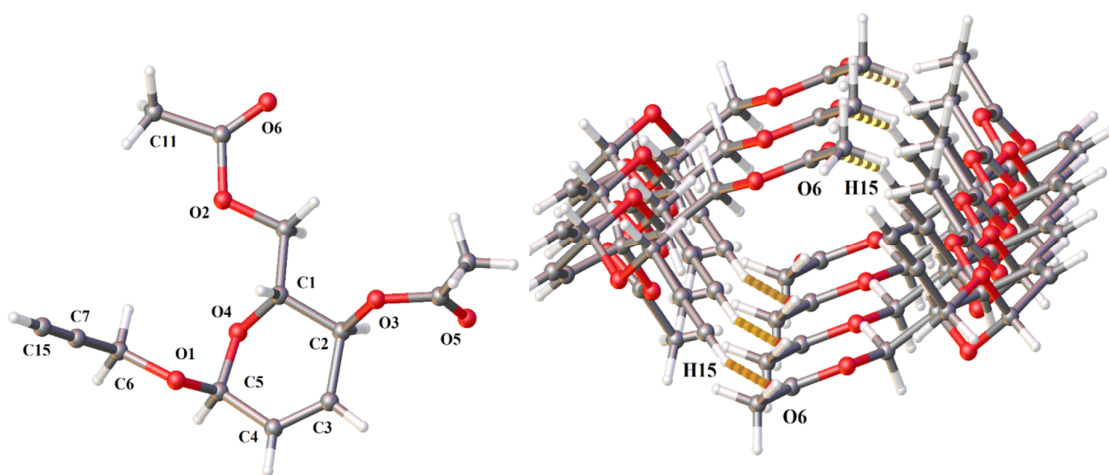


Fig 4a. X-Ray crystal structure of **17n**

Fig 4b. O6...H15 interaction in **17n**

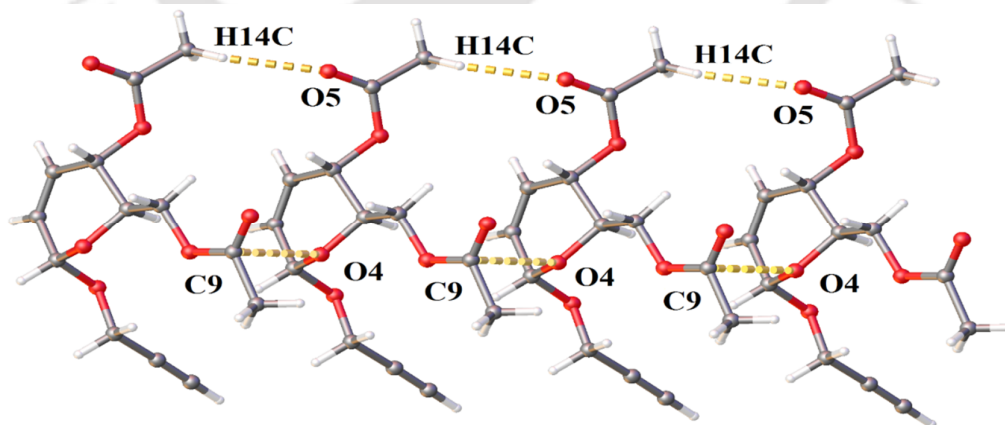
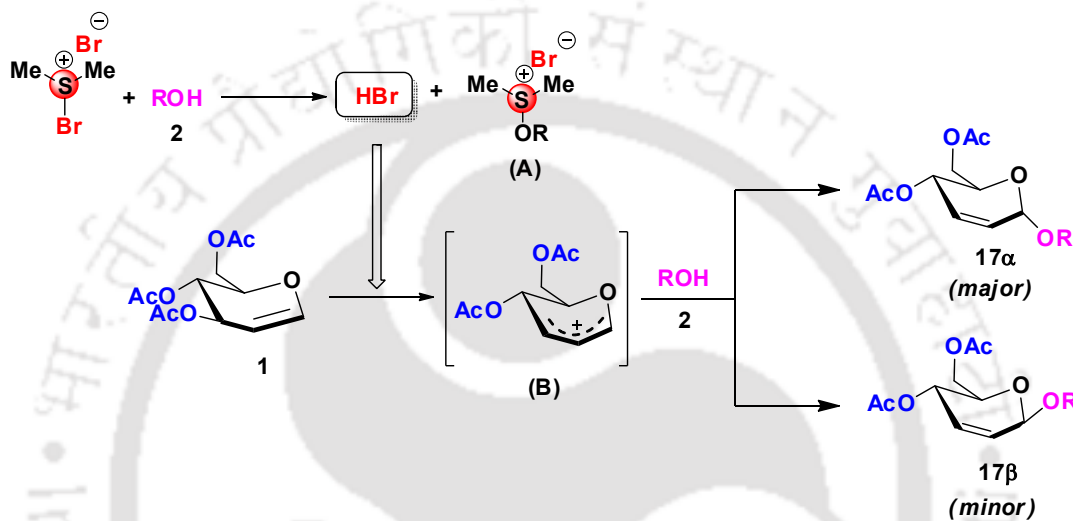


Fig 4c. O4...C9 & O5...H14C interactions in **17n**

A plausible mechanism for the formation of 2,3-unsaturated-*O*-glycoside can be described as follows: The alcohol **2** reacts with bromodimethylsulfonium bromide

(BDMS) to form the intermediate **A**, $[\text{Me}_2\text{S}^+\text{OR}] \text{Br}^-$ and dry HBr. Thus the *in situ* generated HBr which reacts with the acetyl group on C-3 position of 3,4,6-tri-*O*-acetyl-D-glucal **1** favours the formation of allylic oxocarbenium ion **B**, which was initially suggested by Ferrier. Subsequently, the alcohol **2** reacts with oxocarbenium ion **B** preferentially in a quasi-axial fashion which afford the major α -selectivity product of 2,3-unsaturated-*O*-glycosides **17 α** as shown in Scheme 41.

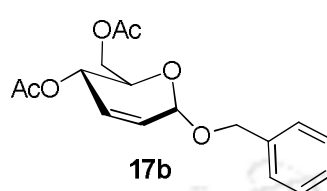
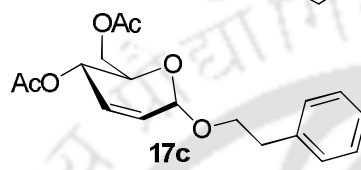
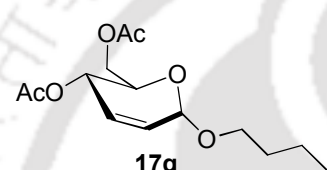
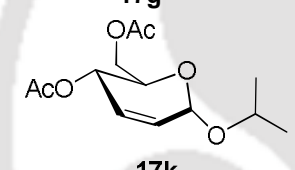
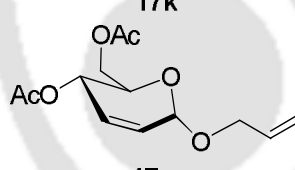
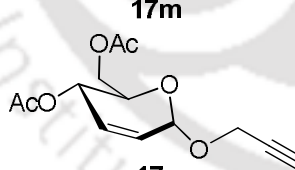


□ **Scheme 41.** Plausible mechanism for the formation of 2,3-unsaturated-*O*-glycosides

To compare the efficiency of the BDMS with aqueous HBr, a similar reaction was conducted with 3,4,6-tri-*O*-acetyl-D-glucal (**1**) and benzyl alcohol (**2k**) using 48% aqueous HBr under identical reaction condition. The output of the result shows a poor conversion of the product and the yield obtained is 30% of **17b** (based on starting material recovery).

The hydrated product is not observed while using 48% aqueous HBr. Another important aspect were a large scale reaction e.g. 10 mmol scale is carried out by using only 5 mol% catalyst. Besides that, the present protocol efficiency and generality is also compared with other recently reported method, which is shown in Table 7. The analyses of the comparative results conclude that the present protocol is equally efficient.

Table 7. Comparison of BDMS catalyzed Ferrier rearrangement with the existing reported method of 3,4,6-tri-*O*-acetyl-D-glucal with different alcohols.

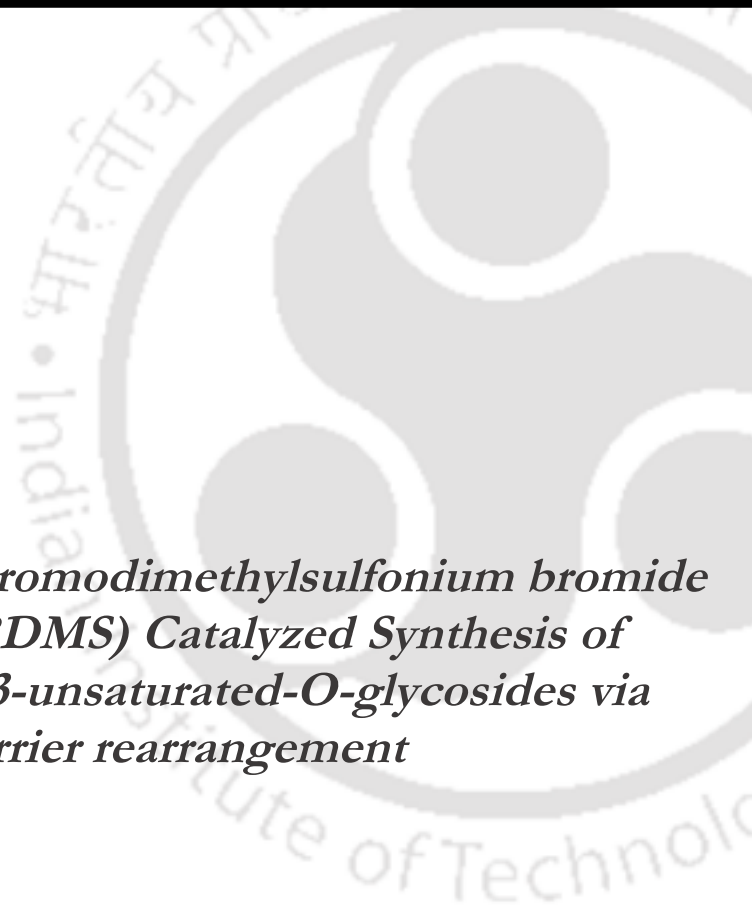
S.No	Product	Catalyst used	Time (h)	Yield (%) ^a	(α : β)
1		BDMS	0.5	92	8:1
		NaHSO ₄	5 min	91 ^{60h}	4:1
		Fe ₂ (SO ₄) ₃ . xH ₂ O	1.5	85 ^{59f}	-
2		BDMS	0.5	90	9:1
		H ₃ PO ₄	15 min	92 ^{57c}	8:1
3		BDMS	0.5	86	6:1
		NaHSO ₄	1	80 ^{60h}	7:1
		H ₃ PO ₄	10 min	93 ^{57c}	8:1
4		BDMS	1	88	10:1
		NbCl ₅	1.5 min	85 ^{59e}	4:1
5		BDMS	0.5	87	18:1
		NaHSO ₄	5 min	72 ^{60h}	9:1
		ZnCl ₂ /Al ₂ O ₃	20 min	83 ^{60f}	α
		NbCl ₅	2.0 min	95 ^{59e}	4:1
6		BDMS	0.5	90	α
		Bi(OTf) ₃	5 min	73 ^{58d}	α
		ZnCl ₂ /Al ₂ O ₃	10 min	88 ^{60f}	α
		NbCl ₅	2.5	92 ^{59e}	4:1

^aIsolated yield.

In summary, we have demonstrated an opportune synthetic protocol for *O*-glycosidation via Ferrier rearrangement using bromodimethylsulfonium bromide as a catalyst for the synthesis of 2,3-unsaturated-*O*-glycosides. The foremost advantages of the present methodology are: mild reaction condition, high α -selectivity, shorter reaction time, good yields and compatible with wide range of substrates. Moreover, the compound **17n** exhibit an interesting intermolecular interactions which shows a prominent cavity (O6...H15) and linear (O4...C9 & O5...H14C) supramolecular assembly.

Chapter III

Experimental Section



➤ *Bromodimethylsulfonium bromide (BDMS) Catalyzed Synthesis of 2,3-unsaturated-O-glycosides via Ferrier rearrangement*

Experimental

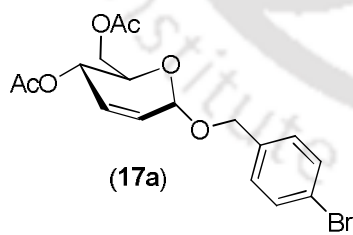
Preparation of bromodimethylsulfonium bromide (BDMS):

In a 100 mL round bottomed flask dimethyl sulfide (1.83 mL, 25 mmol) in 5 mL of dry dichloromethane was taken under ice cold condition (0-5 °C). To that bromine (25 mmol) in dry dichloromethane (5 mL) was added drop wise for a period of 10 minutes. During addition, bromodimethylsulfonium bromide began to form as light orange crystals and after complete addition of bromine, bromodimethylsulfonium bromide crystallized out, which was separated through Buchner funnel. The desired solid was washed twice with dry hexane and dried under reduced pressure and the obtained yield was 77% (4.3 g), m.p 80 °C.

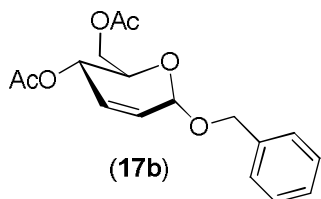
General procedure for the synthesis of 2,3-unsaturated-O-glycosides

A mixture of 3,4,6-tri-*O*-acetyl- α -D-glucal (1 equiv.) and the corresponding alcohol (1.1 equiv.) was taken in 2 mL of dry dichloromethane. Subsequently, catalytic amount of BDMS (0.023 g) was added and stirred at room temperature and the progress of the reactions was monitored by TLC. After completion of the reaction, the solvent was evaporated under reduced pressure and the residues were subjected to column chromatography (eluted with 1:9 ethyl acetate: hexane) to get the pure product of 2,3-unsaturated-*O*-glycosides.

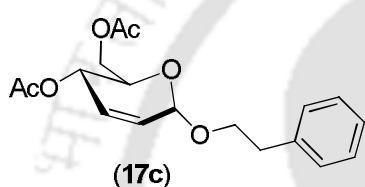
4-Bromobenzyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17a)



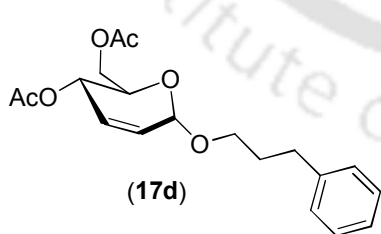
$[\alpha]_D^{25} +47^\circ$ (*c* 0.20, CH₂Cl₂); **IR** (KBr): 3456, 2924, 1742, 1488, 1369, 1231, 1069, 1038, 1010, 968, 803 cm⁻¹; **¹H NMR** (400 MHz, CDCl₃): δ 7.48 (d, *J* = 8.4 Hz, 2H), 7.23 (d, *J* = 8.4 Hz, 2H), 5.91 (d, *J* = 10.4 Hz, 1H), 5.86-5.82 (m, 1H), 5.33 (dd, *J* = 10 Hz, *J* = 1.2 Hz, 1H), 5.11 (s, 1H), 4.75 (d, *J* = 12.4 Hz, 1H), 4.56-4.53 (m, 1H), 4.24 (dd, *J* = 12 Hz, *J* = 5.6 Hz, 1H), 4.16-4.10 (m, 2H), 2.09 (s, 3H), 2.08 (s, 3H); **¹³C NMR** (100 MHz, CDCl₃): δ 170.9, 170.5, 136.8, 131.8, 129.8, 129.6, 127.7, 121.9, 93.8, 69.6, 67.3, 65.4, 63.0, 21.1, 20.9. **Anal. Calcd.** for C₁₇H₁₉BrO₆ (399.23): C, 51.14; H, 4.80. Found: C, 50.98; H, 4.70%.

Benzyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17b)

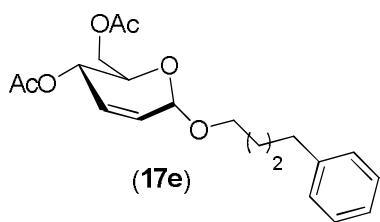
$[\alpha]_D^{25} +66^\circ$ (c 0.13, CH_2Cl_2); **IR** (KBr): 3031, 2903, 1746, 1497, 1454, 1405, 1371, 1236, 1186, 1102, 1041, 969, 909, 734, 699, 605 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 7.35-7.27 (m, 5H), 5.90-5.82 (m, 2H), 5.33 (dd, $J = 2.8$ Hz, $J = 1.6$ Hz, 1H), 5.13 (s, 1H), 4.80 (d, $J = 11.6$ Hz, 1H), 4.62-4.57 (m, 1H), 4.30-4.22 (m, 1H), 4.17-4.08 (m, 2H), 2.08 (s, 3H), 2.06 (s, 3H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): δ 170.7, 170.2, 137.6, 129.3, 128.4, 128.0, 127.8, 127.7, 93.6, 70.2, 67.0, 65.2, 62.9, 20.9, 20.7. **Anal. Calcd.** for $\text{C}_{17}\text{H}_{20}\text{O}_6$ (320.33): C, 63.74; H, 6.29. Found: C, 63.50; H, 6.21%.

2-Phenylethyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17c)

$[\alpha]_D^{25} +76^\circ$ (c 0.16, CH_2Cl_2); **IR** (KBr): 3028, 2928, 1743, 1496, 1370, 1228, 1105, 1039, 972, 909, 751, 700, 669 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 7.23-7.18 (m, 2H), 7.16-7.12 (m, 3H), 5.81 (dd, $J = 11.6$ Hz, $J = 11.2$ Hz, 1H), 5.75-5.71 (m, 1H), 5.21 (dd, $J = 9.6$ Hz, $J = 1.6$ Hz, 1H), 4.94 (s, 1H), 4.10 (dd, $J = 12$ Hz, $J = 5.2$ Hz, 1H), 3.96 (dd, $J = 12$ Hz, $J = 2.8$ Hz, 1H), 3.92-3.86 (m, 2H), 3.71-3.66 (m, 1H), 2.86 (t, 2H), 2.0 (s, 3H), 1.99 (s, 3H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): δ 170.9, 170.4, 138.9, 129.2, 129.0, 128.5, 127.8, 126.4, 94.5, 69.6, 67.0, 65.3, 63.0, 36.4, 21.1, 20.9. **Anal. Calcd.** for $\text{C}_{18}\text{H}_{22}\text{O}_6$ (334.36): C, 64.66; H, 6.63. Found: C, 64.50; H, 6.51%.

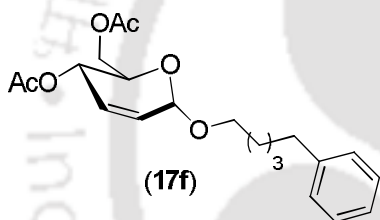
3-Phenylpropyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17d)

$[\alpha]_D^{25} +81^\circ$ (c 0.28, CH_2Cl_2); **IR** (KBr): 3060, 3026, 2942, 1744, 1603, 1496, 1453, 1370, 1235, 1106, 1038, 978, 909, 746, 700, 605 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 7.22-7.07 (m, 5H), 5.81 (d, $J = 10.4$ Hz, 1H), 5.78-5.74 (m, 1H), 5.23 (dd, $J = 9.2$ Hz, $J = 1.6$ Hz, 1H), 4.93 (s, 1H), 4.18-4.14 (m, 1H), 4.08-4.01 (m, 2H), 3.75-3.69 (m, 1H), 3.47-3.41 (m, 1H), 2.67-2.57 (m, 2H), 2.0 (s, 3H), 1.96 (s, 3H), 1.89-1.85 (m, 2H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): δ 170.8, 170.3, 141.7, 129.1, 128.5, 128.0, 125.9, 94.6, 68.2, 67.0, 65.3, 63.0, 32.5, 31.3, 21.0, 20.8. **Anal. Calcd.** for $\text{C}_{19}\text{H}_{24}\text{O}_6$ (348.39): C, 65.50; H, 6.94. Found: C, 65.40; H, 6.82%.

4-Phenyl butanyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (**17e**)

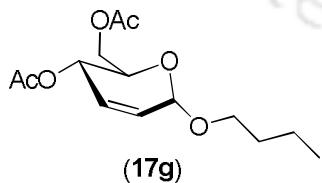
$[\alpha]_D^{25} +54^\circ$ (c 0.22, CH_2Cl_2); **IR** (KBr): 3026, 2936, 2861, 1744, 1496, 1453, 1370, 1233, 1041, 748, 699, 669 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 7.29-7.13 (m, 5H), 5.87 (d, $J = 11.6$ Hz, 1H), 5.83-5.81 (m, 1H), 5.30 (dd, $J = 9.6$ Hz, $J = 1.2$ Hz, 1H), 5.0 (s, 1H), 4.25-4.19

(m, 1H), 4.15 (dd, $J = 12$ Hz, $J = 2$ Hz, 1H), 4.10-4.06 (m, 1H), 3.82-3.76 (m, 1H), 3.54-3.47 (m, 1H), 2.67-2.59 (m, 2H), 2.07 (s, 3H), 2.06 (s, 3H), 1.72-1.62 (m, 4H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): δ 170.8, 170.3, 142.3, 129.0, 128.5, 128.4, 128.0, 125.8, 94.5, 68.8, 66.9, 65.3, 63.1, 35.7, 29.4, 28.2, 21.0, 20.8. **Anal. Calcd.** for $\text{C}_{20}\text{H}_{26}\text{O}_6$ (362.41): C, 66.28; H, 7.23. Found: C, 66.15; H, 7.14%.

5-Phenyl pentyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (**17f**)

$[\alpha]_D^{25} +12.8^\circ$ (c 0.40, CH_2Cl_2); **IR** (KBr): 3061, 3025, 2934, 2858, 1748, 1495, 1453, 1367, 1228, 1120, 1046, 907, 748, 700, 601 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 7.28-7.23 (m, 2H), 7.17-7.15 (m, 3H), 5.94-5.83 (m, 2H), 5.30 (d, $J = 9.2$ Hz, 1H), 5.0 (s, 1H), 4.22-4.17 (m,

1H), 4.03-3.99 (m, 1H), 3.87-3.73 (m, 1H), 3.52-3.44 (m, 2H), 2.63-2.59 (m, 2H), 2.07 (s, 3H), 2.06 (s, 3H), 1.66-1.62 (m, 4H), 1.44-1.39 (m, 2H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): 171.0, 170.5, 142.7, 129.2, 128.5, 128.1, 126.3, 125.9, 94.6, 69.0, 67.0, 65.5, 63.2, 36.0, 31.4, 29.8, 26.1, 21.2, 21.0. **Anal. Calcd.** for $\text{C}_{21}\text{H}_{28}\text{O}_6$ (376.44): C, 67.00; H, 7.50. Found: C, 66.85; H, 7.40%.

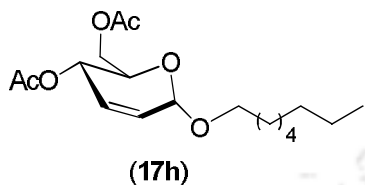
n-Butyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (**17g**)

$[\alpha]_D^{25} +106^\circ$ (c 0.23, CH_2Cl_2); **IR** (KBr): 2959, 2934, 2873, 1746, 1371, 1233, 1105, 1043, 907, 732, 605 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 5.82-5.75 (m, 1H), 5.72-5.67 (m, 1H), 5.16 (dd, $J = 9.6$ Hz, $J = 1.2$ Hz, 1H), 4.88 (s, 1H), 4.10 (dd,

$J = 12.4$ Hz, $J = 5.6$ Hz, 1H), 4.0 (dd, $J = 12$ Hz, $J = 2.4$ Hz, 1H), 3.98-3.94 (m, 1H), 3.68-3.61 (m, 1H), 3.40-3.33 (m, 1H), 1.95 (s, 3H), 1.94 (s, 3H), 1.48-1.41 (m, 2H), 1.29-1.22 (m, 2H), 0.79 (t, 3H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): δ 170.6, 170.2, 128.9,

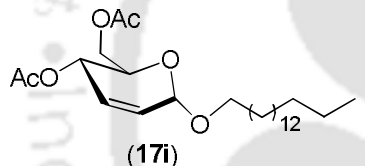
127.9, 94.3, 68.5, 66.8, 65.2, 63.0, 31.7, 20.8, 20.7, 19.3, 13.7. **Anal. Calcd.** for $C_{14}H_{22}O_6$ (286.32): C, 58.73; H, 7.74. Found: C, 58.50; H, 7.65%.

Octyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17h)



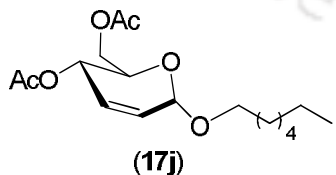
$[\alpha]_D^{25} +16.6^\circ$ (c 0.62, CH_2Cl_2); **IR** (KBr): 3060, 3025, 2933, 2858, 1748, 1602, 1494, 1453, 1370, 1230, 1120, 1047, 906, 746, 700 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 5.95-5.89 (m, 1H), 5.87-5.85 (m, 1H), 5.31 (d, $J = 9.6$ Hz, 1H), 5.03 (s, 1H), 4.28-4.22 (m, 1H), 4.19-4.16 (m, 1H), 4.13-4.09 (m, 1H), 3.80-3.74 (m, 1H), 3.53-3.46 (m, 1H), 2.10 (s, 3H), 2.08 (s, 3H), 1.63-1.57 (m, 5H), 1.32-1.29 (m, 7H), 0.87 (t, $J = 5.6$ Hz, 3H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 170.5, 170.4, 129.0, 128.1, 94.5, 69.1, 67.0, 65.4, 63.1, 31.9, 29.8, 29.5, 29.4, 26.3, 22.7, 21.0, 20.9, 14.2. **Anal. Calcd.** for $C_{18}H_{30}O_6$ (342.42): C, 63.14; H, 8.83. Found: C, 62.95; H, 8.70%.

Hexadecyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17i)

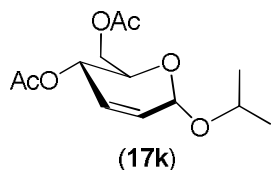


$[\alpha]_D^{25} +50^\circ$ (c 0.07, CH_2Cl_2); **IR** (KBr): 2924, 2853, 1748, 1466, 1369, 1233, 1105, 1040 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 5.88 (d, $J = 10.4$ Hz, 1H), 5.85-5.82 (m, 1H), 5.31 (dd, $J = 10$ Hz, $J = 1.2$ Hz, 1H), 5.02 (s, 1H), 4.25 (dd, $J = 12$ Hz, $J = 5.2$ Hz, 1H), 4.17 (dd, $J = 12$ Hz, $J = 2.4$ Hz, 1H), 4.12-4.09 (m, 1H), 3.78-3.73 (m, 1H), 3.53-3.47 (m, 1H), 2.10 (s, 3H), 2.08 (s, 3H), 1.63-1.59 (m, 4H), 1.25-1.20 (m, 24H), 0.88 (t, $J = 6.34$, 3H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 170.8, 170.3, 129.2, 128.1, 94.6, 69.2, 67.0, 65.5, 63.3, 32.1, 29.8 (9C), 29.6, 26.4, 22.9, 21.2, 21.0, 14.3. **Anal. Calcd.** for $C_{26}H_{46}O_6$ (454.64): C, 68.69; H, 10.20. Found: C, 68.55; H, 10.05%.

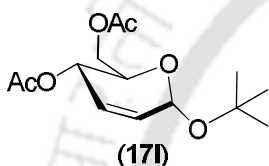
Heptyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17j)



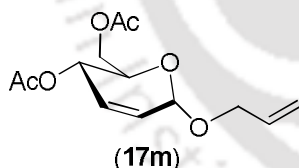
$[\alpha]_D^{25} +68^\circ$ (c 0.15, CH_2Cl_2); **IR** (KBr): 2928, 2857, 1745, 1369, 1234, 1105, 1039, 908, 802 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 5.98-5.83 (m, 2H), 5.33 (dd, $J = 11.2$ Hz, $J = 1.2$ Hz, 1H), 5.04 (s, 1H), 4.35-4.29 (m, 1H), 4.25 (dd, $J = 12$ Hz, $J = 2.4$ Hz, 1H), 3.87-3.81 (m, 1H), 3.61-3.48 (m, 2H), 2.18 (s, 3H), 2.16 (s, 3H), 1.70-1.63 (m, 3H), 1.38-1.36 (m, 7H), 0.95 (t, 3H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 170.9, 170.4, 129.1, 128.2, 94.5, 69.1, 67.0, 65.5, 63.2, 31.9, 29.8, 29.2, 26.3, 22.7, 21.1, 20.9, 14.2. **Anal. Calcd.** for $C_{17}H_{28}O_6$ (328.40): C, 62.17; H, 8.59. Found: C, 62.02; H, 8.44%.

Isopropyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17k)

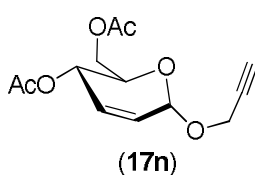
$[\alpha]_D^{25} +112^\circ$ (c 0.18, CH_2Cl_2); **IR** (KBr): 2972, 2931, 1745, 1453, 1371, 1236, 1184, 1125, 1100, 1036, 982, 909, 731, 606 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 5.82-5.73 (m, 2H), 5.23 (d, $J = 10$ Hz, 1H), 5.07 (s, 1H), 4.18 (dd, $J = 12$ Hz, $J = 6$ Hz, 2H), 4.14-4.07 (m, 1H), 3.96-3.90 (m, 1H), 2.03 (s, 3H), 2.02 (s, 3H), 1.20 (d, $J = 6$ Hz, 3H), 1.12 (d, $J = 6$ Hz, 3H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): δ 170.8, 170.3, 128.8, 128.5, 92.9, 70.8, 66.8, 65.5, 63.2, 23.6, 22.0, 21.0, 20.8. **Anal. Calcd.** for $\text{C}_{13}\text{H}_{20}\text{O}_6$ (272.29): C, 57.34; H, 7.40. Found: C, 57.20; H, 7.25%.

tert-Butyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17l)

$[\alpha]_D^{25} +102^\circ$ (c 0.14, CH_2Cl_2); **IR** (KBr): 2976, 2933, 1746, 1437, 1369, 1235, 1195, 1099, 1044, 982, 891, 774, 716, 605 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 5.84 (d, $J = 10$ Hz, 1H), 5.77-5.73 (m, 1H), 5.32 (s, 1H), 5.27 (d, $J = 8.8$ Hz, 1H), 4.26-4.21 (m, 1H), 4.19-4.13 (m, 2H), 2.08 (s, 3H), 2.07 (s, 3H), 1.29 (s, 9H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): δ 171.0, 170.6, 129.7, 128.3, 89.1, 75.5, 66.7, 65.5, 63.4, 28.9, 21.1, 21.0. **Anal. Calcd.** for $\text{C}_{14}\text{H}_{22}\text{O}_6$ (286.32): C, 58.73; H, 7.74. Found: C, 58.55; H, 7.57%.

Allyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17m)

$[\alpha]_D^{25} +102^\circ$ (c 0.12, CH_2Cl_2); **IR** (KBr): 3081, 2924, 1742, 1652, 1370, 1229, 1101, 1039, 812, 743, 603 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 5.94-5.81 (m, 3H), 5.30-5.24 (m, 2H), 5.18-5.15 (m, 1H), 5.04 (s, 1H), 4.24-4.19 (m, 2H), 4.14 (dd, $J = 12.4$ Hz, $J = 2.4$ Hz, 1H), 4.10-4.02 (m, 2H), 2.06 (s, 3H), 2.04 (s, 3H); **$^{13}\text{C NMR}$** (100 MHz, CDCl_3): δ 170.9, 170.4, 134.2, 129.3, 127.9, 117.6, 93.7, 69.4, 67.0, 65.4, 63.0, 21.0, 20.9. **Anal. Calcd.** for $\text{C}_{13}\text{H}_{18}\text{O}_6$ (270.28): C, 57.77; H, 6.71. Found: C, 57.59; H, 6.78%.

Propynyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17n)

$[\alpha]_D^{25} +161^\circ$ (c 0.18, CH_2Cl_2); **IR** (KBr): 3465, 3278, 2917, 1742, 1373, 1229, 1141, 1099, 1039, 965, 909, 731, 680, 613 cm^{-1} ; **$^1\text{H NMR}$** (400 MHz, CDCl_3): δ 5.92 (d, $J = 10.4$ Hz, 1H), 5.86-5.83 (m, 1H), 5.33 (dd, $J = 9.6$ Hz, $J = 1.2$ Hz, 1H), 5.24 (s, 1H), 4.31

(d, $J = 2.4$ Hz, 2H), 4.23 (d, $J = 4.8$ Hz, 1H), 4.18 (dd, $J = 12.4$ Hz, $J = 2.4$ Hz, 1H), 4.10-4.07 (m, 1H), 2.52 (t, 1H), 2.10 (s, 3H), 2.09 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 170.6, 170.1, 129.7, 127.2, 92.6, 79.0, 74.9, 67.1, 65.0, 62.7, 54.9, 20.8, 20.7. **Anal.** **Calcd.** for $\text{C}_{13}\text{H}_{16}\text{O}_6$ (268.26): C, 58.20; H, 6.01. Found: C, 58.01; H, 5.94%.

XRD for Compounds 17n

Complete crystallographic data of **17n** for the structural analyses have been deposited with the Cambridge Crystallographic Data Centre, CCDC No. 875412. Copies of this information may be obtained free of charge from the Director, Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK, (fax: +44-1223-336033, e-mail: deposit@ccdc.cam.ac.uk or via: www.ccdc.cam.ac.uk).

Table 8. Crystal data and structures refinement for the compound **17n**, atomic coordinates, equivalent isotropic displacement parameters and bond angles, please check the CIF.

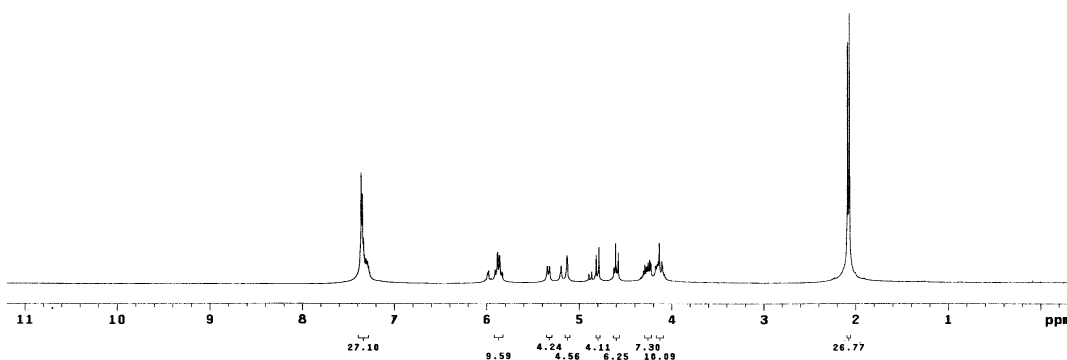
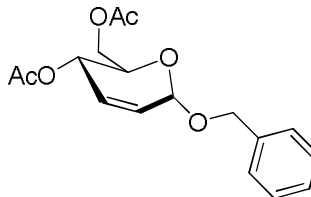
Parameters	Compound 17n
Empirical Formula	' $\text{C}_{13}\text{H}_{16}\text{O}_6$ '
Formula weight	268.26
Temperature	296 K
CCDC no	875412
Wavelength (\AA)	0.71073 \AA
Crystal system	Orthorhombic
Space group	P2(1)2(1)2(1)
Radiation type	Mo K α
Radiation source	'fine-focus sealed tube'
a (\AA)	5.2823 (6) \AA
b (\AA)	15.3302 (18) \AA
c (\AA)	17.207 (2) \AA
α ($^\circ$)	90.00 $^\circ$
β ($^\circ$)	90.00 $^\circ$
γ ($^\circ$)	90.00 $^\circ$
Absolute structure Flack	0.1 (1)
Cell Volume	1393.4 (3) \AA^3

Z	4
Density	1.279 g/cm ³
μ (mm ⁻¹)	0.102 mm ⁻¹
F(0 0 0)	568.0
Index ranges	-7<=h<=7, -17<=k<=20, -23<=l<=23
Number of unique reflections	3600
Number of parameters	174
Number of restraints	0
Cell Theta range	2.01-21.04°
Measured intensity theta range	1.78- 28.71°
Goodness-of-fit (GOF) ^a on F ²	1.060
Refinement method	Full-matrix least-squares on F ²

^1H NMR (400 MHz, CDCl_3): Benzyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-eno pyranoside (17b)

```

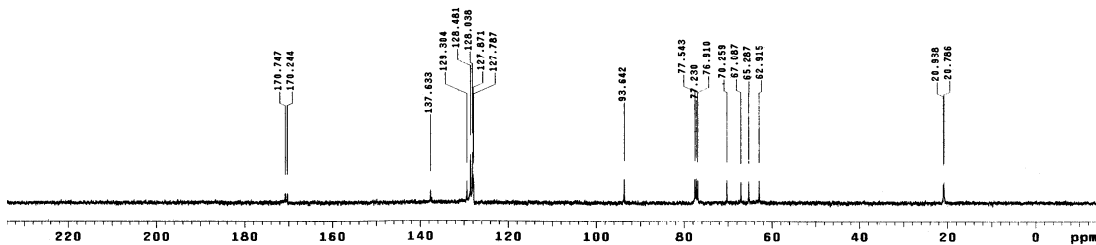
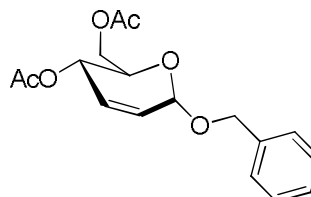
RSB_P5_S10
exp1 s2pu1
SAMPLE
date Jan 24 2012 temp not used
solvent CDCl3 ga1n not used
file exp1 sp1n not used
ACQUISITION
sw 6389.5 hst 8.988
at 1.398 a1fa 19.700
np 25520 a1fa 26.000
fb not used f1 n
bs 4 fn n
d1 1.000 dp y
nt 32 hs PROCESSING
ct 32 nn
TRANSMITTER
tn H1 fn 0.10
sfreq 399.853 sp DISPLAY 65536
tof 362.8 sp -152.5
tpwr 57 wp 4635.5
pw 9.850 rf1 788.0
DECOUPLER
dn C13 rp 113.9
dof 0 lp PLOT -97.5
dm nnn wc 250
dmm c sc 0
dpwr 50 vs 0
dnt 15900 vs 69
dmf 15900 vs 18
nm %h.c .ph 18
  
```



^{13}C NMR (100 MHz, CDCl_3): Benzyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-eno pyranoside (17b)

```

RSB_P5_S10
exp1 s2pu1
SAMPLE
date Jan 24 2012 temp not used
solvent CDCl3 ga1n not used
file exp1 sp1n not used
ACQUISITION
sw 25125.6 pw90 16.600
at 1.199 a1fa 20.000
np 68270 f1 n
bs 15000 fn n
d1 1.000 dp y
nt 5000 hs PROCESSING
ct 290 nn
TRANSMITTER
tn C13 fn 2.00
sfreq 100.534 sp DISPLAY 65536
tof 1536.3 sp -1525.6
tpwr 21 wp 25125.6
pw 9.300 rf1 8280.5
DECOUPLER
dn H1 rp 7764.9
dof 0 lp -98.6
dm yyy wc 250
dmm 42 sc 0
dpwr 8900 vs 13
dnt 8900 vs 13
dmf 8900 vs 2
nm no ph 2
  
```

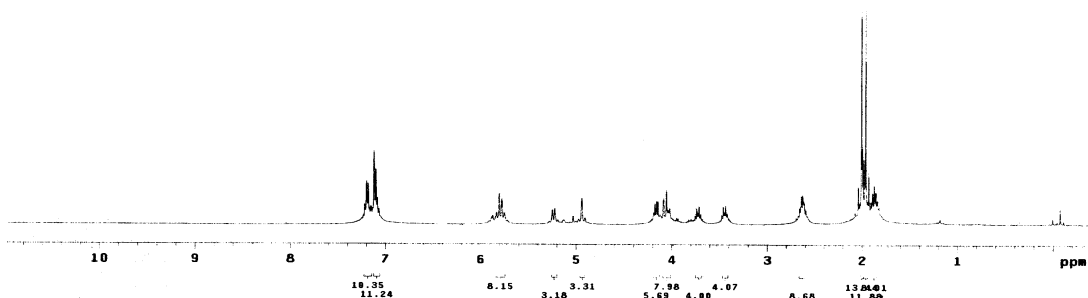
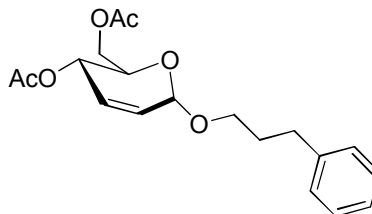


NMR (400 MHz, CDCl₃): 3-Phenylpropanyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-eno pyranoside (17d)

```

R8B_P5_S4
expt s2pu1
SAMPLE
date Dec 28 2011 temp not used
solvent CDCl3 gain not used
file /export/home/... spin not used
ciftemp/R8B_P5_S4_... hst 0.000
ACQUISITION
sw 6389.8 ppg9 15.700
at 1.199 alfa 20.000
np 25528 in n
fd not used dp y
bs 4 hs nn
dl 1.000 rp PROCESSING
nt 32 lb 6.10
ct 32 fn 65536
TRANSMITTER
tn H1 sp -161.1
sfrq 399.853 wp 4582.6
tof 362.8 rfl 824.3
tpwr 9.858 rfp 0
pw DECOUPLER rp 111.0
dn C13 lp PLOT -96.6
dof 0 wc 250
dm nm sc 8
dmm c vs 5B
dpwr 0 th 35
dof 15900 ne cdc ph

```

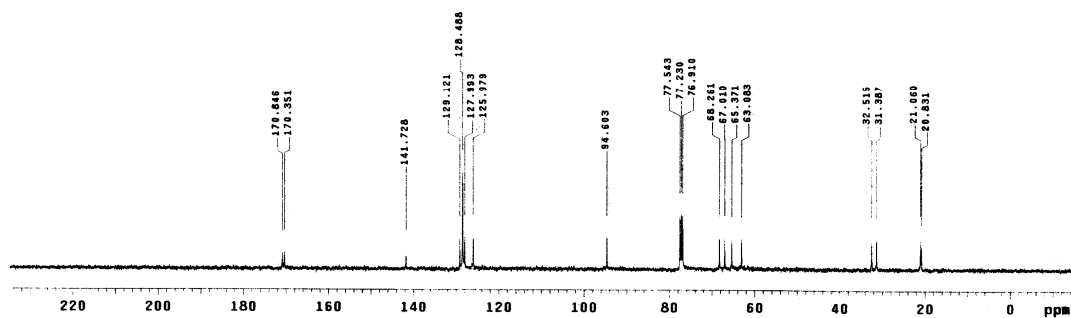
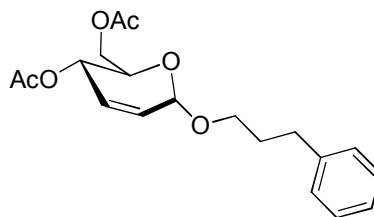


¹³C NMR (100 MHz, CDCl₃): 3-Phenylpropanyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17d)

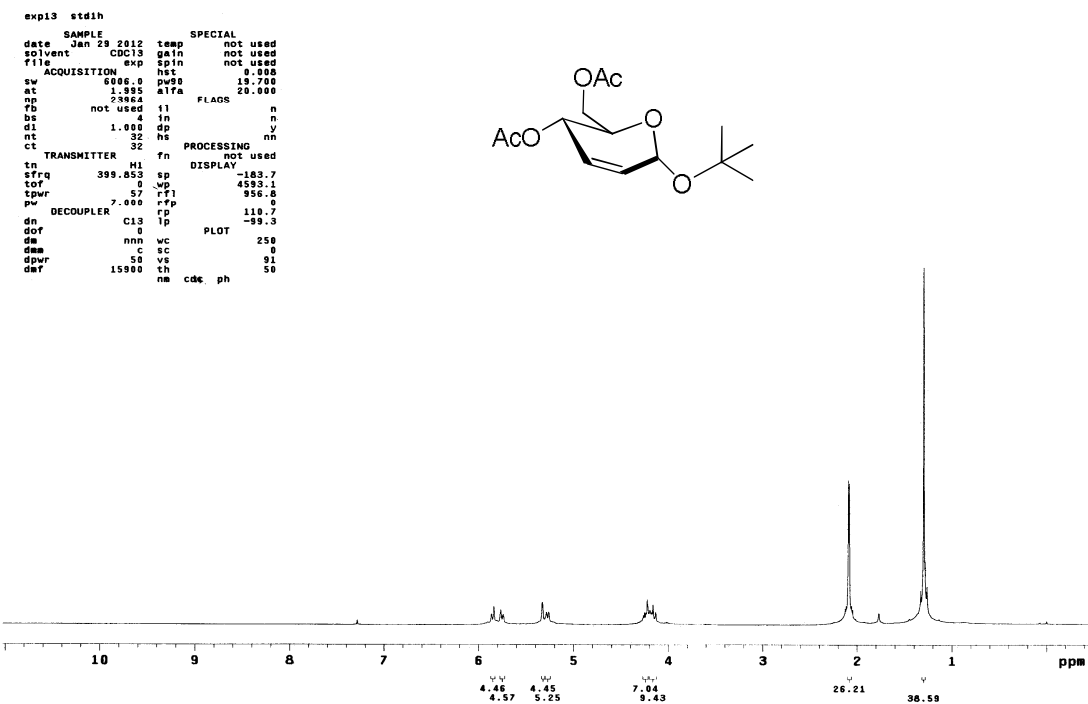
```

expt s2pu1
SAMPLE
date Dec 28 2011 temp not used
solvent CDCl3 gain not used
file exp spin not used
ACQUISITION
sw 25125.6 ppg9 16.000
at 1.199 alfa 20.000
np 58276 in n
fd 13600 ll n
bs 19 in n
dl 1.000 dp y
nt 5000 hs nn
ct 840 PROCESSING
TRANSMITTER
tn C13 fn 65536
sfrq 100.554 sp -1516.7
tof 1536.3 wp 25125.6
tpwr 61 rfp 9283.6
pw DECOUPLER rfp 7764.8
dn H1 rp -66.0
dof 0 lp PLOT -268.7
dm vvv wc 250
dmm w sc 8
dpwr 8900 vs 22
dof nm no ph 3

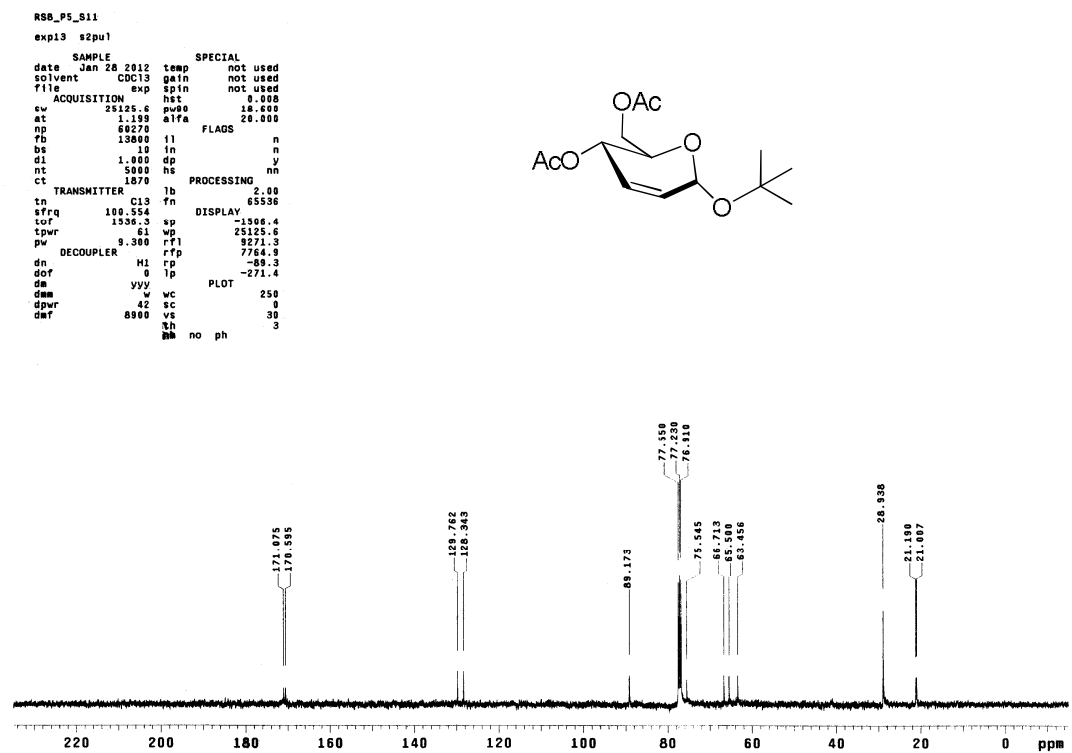
```



^1H NMR (400 MHz, CDCl_3): Tert-Butyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-eno pyranoside (17I)



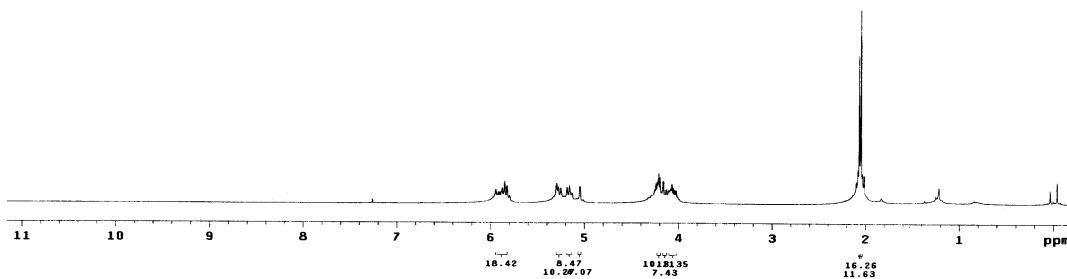
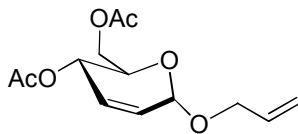
^{13}C NMR (100 MHz, CDCl_3): Tert-Butyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-eno pyranoside (17I)



^1H NMR (400 MHz, CDCl_3): Allyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17m)

```

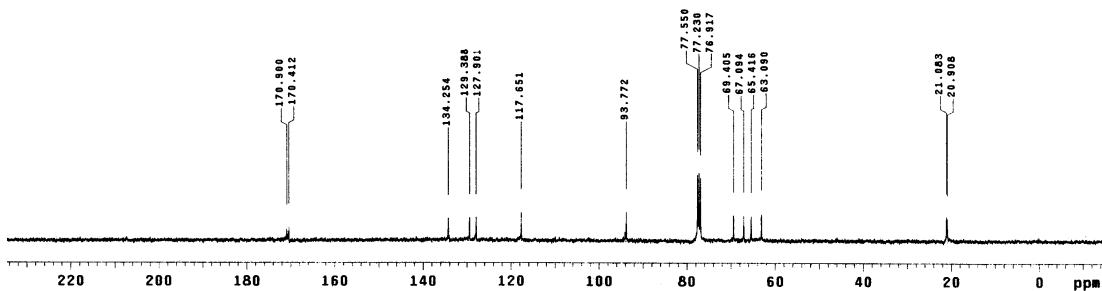
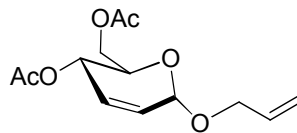
RSB_P5_S2
exp1 s2pu1
SAMPLE
date Dec 21 2011 temp not used
solvent CDCl3 gain not used
file exp spin not used
ACQUISITION hst 0.000
sv 25125.6 pu09 10.500
at 1.199 alpha 20.000
np 60270 FLAGS
fb not used f1 n
bs 4 in n
d1 1.000 dp y
nt 32 hs nn
ct 32 PROCESSING
tn TRANSMITTER H1 fb 0.10
sfrq 399.853 f0 65536
tof 362.8 sp -88.7
tpwr 37 wp 4588.4
pw 9.850 rf1 795.0
DECOUPLER rfp
dn C13 rp 110.0
dof 0 lp -84.0
dm c PLOT
dpm 50 wc 250
dpr 1500 sc 0
dft 1500 th 20
nm cdc ph
  
```



NMR (100 MHz, CDCl_3): Allyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-enopyranoside (17m)

```

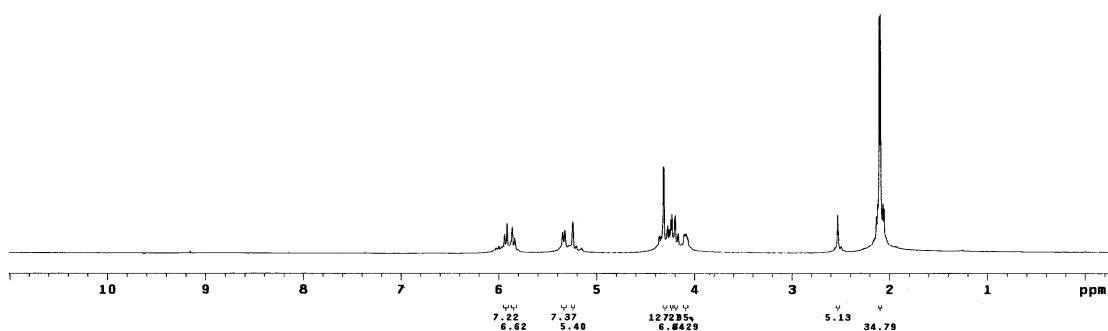
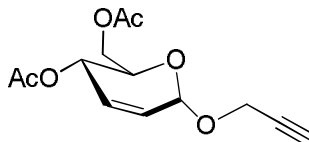
exp1 s2pu1
SAMPLE
date Dec 21 2011 temp not used
solvent CDCl3 gain not used
file exp spin not used
ACQUISITION hst 0.000
sv 25125.6 pu09 10.500
at 1.199 alpha 20.000
np 60270 FLAGS
fb 13680 f1 n
bs 10 in n
d1 1.000 dp y
nt 5080 hs nn
ct 1790 PROCESSING
tn TRANSMITTER C13 fb 2.00
sfrq 100.554 f0 65536
tof 1536.3 sp -1510.3
tpwr 61 wp 25125.6
pw 9.300 rf1 9275.2
DECOUPLER H1 rfp 7764.9
dn 0 lp -271.4
dm vvy PLOT
dpm 42 wc 250
dpr 8900 sc 0
dft 8900 vs 15
nm no ph th 2
  
```



^1H NMR (400 MHz, CDCl_3): Propynyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-eno pyranoside (17n)

```

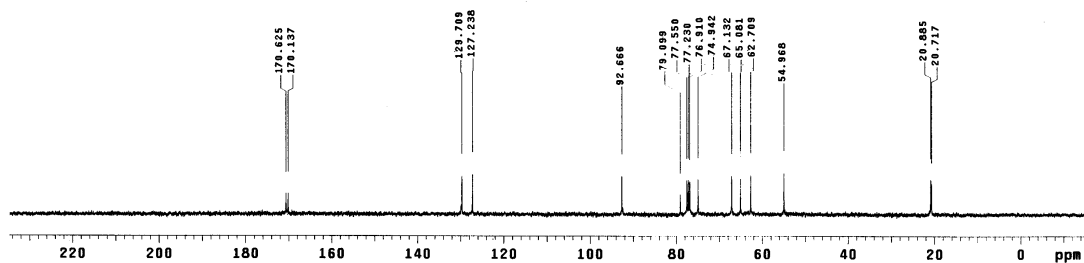
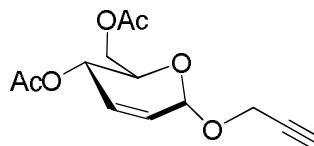
exp1 s2pu1
SAMPLE          SPECIAL
date Jan 31 2012 temp not used
solvent CDCl3 gain not used
file           sp1n not used
ACQUISITION    hst 0.008
sw 6388.8 pw90 19.700
at 1.398 a1fa 20.000
np 25528 FLAGS
fb not used f1 n
bs 4 f2 n
d1 1.000 dp y
nt 32 hs
ct 32 PROCESSING
tn TRANSMITTER lb 0.10
    HI fn 65536
sfrq 399.853 DISPLAY
tof 362.6 sp -186.3
tpwr 57 wp 4508.9
pw 9.850 rfp 752.7
DECOUPLER C13 rp 9
dn 0 rp 103.9
dof 0 lp -69.0
dm nnn PLOT
dmc c wc 250
dpcr 50 sc 0
daf 15900 vs 59
nm' sb 20
    cdc ph 20
  
```



^{13}C NMR (100 MHz, CDCl_3): Propynyl 4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-eno pyranoside (17n)

```

exp1 s2pu1
SAMPLE          SPECIAL
date Jan 31 2012 temp not used
solvent CDCl3 gain not used
file           sp1n not used
ACQUISITION    hst 0.008
sw 25125.6 pw90 18.600
at 1.199 a1fa 20.000
np 60270 FLAGS
fb 13800 f1 n
bs 10 f2 n
d1 1.000 dp y
nt 5000 hs
ct 420 PROCESSING
tn TRANSMITTER lb 2.00
    C13 fn 65536
sfrq 100.554 DISPLAY
tof 1536.3 sp -1521.8
tpwr 61 wp 25125.6
pw 9.300 rfp 3206.7
DECOUPLER C13 rp 7764.9
dn 0 rp -72.5
dof 0 lp -326.0
dm yyy PLOT
dmc w wc 250
dpcr 42 sc 0
daf 8900 vs 10
    th 3
    nm' bo ph
  
```





Chapter I-III
References



References

1. Stern, R.; Jedrzejewski, M. J. *Chem. Rev.* **2008**, *108*, 5061-5085.
2. Linhardt, R. J.; Toidam, T. *Carbohydrates in drug design*. Marcel Dekker: New York 1997.
3. (a) Goa, K. L.; Benfield, P. *Drugs* **1994**, *47*, 536-566. (b) Sashiwa, H.; Thompson, J. M.; Das, S. K.; Shigemasa, Y.; Tripathy, S.; Roy, R. *Biomacromolecules* **2000**, *1*, 303-305.
4. McAlindon, T. E.; LaValley, M. P.; Gulin, J. P.; Felson, D.T. *J. Am. Med. Assoc.* **2000**, *283*, 1469-1475.
5. Yarema, K. J.; Bertozzi, C. R. *Curr. Opin. Chem. Biol.* **1998**, *2*, 49-61.
6. Karlsson, K. A. *Trends Pharma. Sci.* **1991**, *12*, 265-272.
7. Wong, C-H. *Acc. Chem. Res.* **1999**, *32*, 376-385.
8. Ramström, O.; Bunyapaiboonsri, T.; Lohmann, S.; Lehn, J. M. *Biochimica et Biophysica Acta* **2002**, *1572*, 178-186.
9. Ramstrom, O.; Lehn, J. M. *Chembiochem* **2000**, *1*, 41-48.
10. Dwek, R. A. *Chem. Rev.* **1996**, *96*, 683-720.
11. Jelinek, R.; Kolusheva, S. *Chem. Rev.* **2004**, *104*, 5987-6015.
12. Varki, A. *Glycobiology* **1993**, *3*, 97-130.
13. Rini, J. M. *Annu. Rev. Biophys. Biomol. Struct.* **1995**, *24*, 551-577.
14. Weis, W. I.; Drickamer, K. *Annu. Rev. Biochem.* **1996**, *65*, 441-473.
15. Lis, H.; Sharon, N. *Chem. Rev.* **1998**, *98*, 637-674.
16. (a) Tanimoto, H.; Saito, R.; Chida, N. *Tetrahedron Lett.* **2008**, *49*, 358-362. (b) Sabitha, G.; Reddy, S. S. S.; Yadav, J. S. *Tetrahedron Lett.* **2010**, *51*, 6259-6261.
17. Danishefsky, S. J.; Bilodeau, M. T. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 1380-1419.
18. (a) Postema, M. H. D. *C-glycosides Synthesis*; CRC Press: London, 1995, pp 43-57. (b) Cusk, R.; Schaade, M.; Krieger, C. *Tetrahedron* **1996**, *52*, 6397-6408. (c) Takhi, M.; Rahman, A.; Schmidt, R. R. *Tetrahedron Lett.* **2001**, *42*, 4053-4056. (d) Tiwari, P.; Agnihotri, G.; Misra, A. K. *Carbohydr. Res.* **2005**, *340*, 749-752.
19. Michael, A. *Am. Chem. J.* **1879**, *1*, 305-312.
20. Koenigs, W.; Knorr, E. *Ber. Dtsch. Chem. Ges.* **1901**, *34*, 957-981.
21. Demchenko, A. V. *Synlett* **2003**, 1225-1240.

22. Padungros, P.; Alberch, L.; Wei, A. *Org. Lett.* **2012**, *14*, 3380-3383.
23. Marín, I.; Castilla, J.; Matheu, M. I.; Díaz, Y.; Castillon, S. *J. Org. Chem.* **2011**, *76*, 9622-9629.
24. Ding, F.; William, R.; Wang, F.; Ma, J.; Ji, L.; Liu, X-W. *Org. Lett.* **2011**, *13*, 652-655.
25. Baryal, K. N.; Adhikari, S.; Zhu, J. *J. Org. Chem.* **2013**, *78*, 12469-12476.
26. Deshpande, P. P.; Danishefsky, S. J. *Nature* **1997**, *387*, 164-166.
27. Bhattacharya, S. K.; Danishefsky, S. J. *J. Org. Chem.* **2000**, *65*, 144-151.
28. Danishefsky, S. J.; Selnick, H. G.; Armistead, D. M.; Wincott, F. E. *J. Am. Chem. Soc.* **1987**, *109*, 8119-8120.
29. Friesen, R. W.; Danishefsky, S. J. *J. Am. Chem. Soc.* **1989**, *111*, 6656-6660.
30. Suzuki, K.; Sulikowski, G. A.; Friesen, R. W.; Danishefsky, S. J. *J. Am. Chem. Soc.* **1990**, *112*, 8895-8902.
31. (a) Roush, W. R.; Lin, X-F. *J. Am. Chem. Soc.* **1995**, *117*, 2236-2250. (b) Roush, W. R.; Hartz, R. A.; Gustin, D. J. *J. Am. Chem. Soc.* **1999**, *121*, 1990-1991.
32. Lafont, D.; Boullanger, P.; Rosenzweig, M. *J. Carbohydr. Chem.* **1998**, *17*, 1377-1393.
33. Roush, W. R.; Narayan, S.; Bennett, C. E.; Briner, K. *Org. Lett.* **1999**, *1*, 895-897.
34. Kirschning, A.; Jesberger, M.; Schonberger, A. *Org. Lett.* **2001**, *3*, 3623-3626.
35. Gammon, D. W.; Kinf, H. H.; De Vos, D. E.; Jacobs, P. A.; Sels, B. F. *Tetrahedron Lett.* **2004**, *45*, 9533-9536.
36. Handa, M.; Smith III, W. J.; Roush, W. R. *J. Org. Chem.* **2008**, *73*, 1036-1039.
37. Sirion, U.; Purintawarrakun, S.; Sahakitpichan, P.; Saeeng, R. *Carbohydr. Res.* **2010**, *345*, 2401-2407.
38. Islam, M.; Tirukoti, N. D.; Nandi, S.; Hotha, S. *J. Org. Chem.* **2014**, *79*, 4470-4476.
39. Kövér, A.; Boutureira, O.; Matheu, M. I.; Díaz, Y.; Castillón, S. *J. Org. Chem.* **2014**, *79*, 3060-3068.
40. Dharuman, S.; Vankar, Y. D. *Org. Lett.* **2014**, *16*, 1172-1175.
41. (a) Ciment, D. M.; Ferrier, R. J. *J. Chem. Soc. C* **1966**, 441-445. (b) Ferrier, R. J.; Prasad, N. *J. Chem. Soc. C* **1969**, 570-575. (c) Ferrier, R. J.; Prasad, N. *J. Chem.*

- Soc. C* **1969**, 581-586. (d) Ferrier, R. J. *Adv. Carbohydr. Chem. Biochem.* **1969**, pp 199-266. (e) Descotes, G.; Martin, J. C. *Carbohydr. Res.* **1977**, *56*, 168-172. (f) Klaffke, W.; Pudlo, P.; Springer, D.; Thieme, J. *Liebigs Ann. Chem.* **1991**, 509-512. (g) Ferrier R. J. *Top. Curr. Chem.* **2001**, *215*, 153-175. (h) Ferrier R. J.; Zubkov, O. A. *Org. React.* **2003**, *62*, 569-736 and references therein. (i) Gómez, A. M.; Lobo, F.; Uriel, C.; López, J. C. *Eur. J. Org. Chem.* **2013**, 7221-7262 and references therein.
42. Williams, N. R.; Wander, J. D. *The Carbohydrates, Chemistry and Biochemistry*; Academic Press: New York, 1980; pp. 761-798.
43. López, J. C.; Gómez, A. M.; Valverde, S.; Fraser-Reid, B. *J. Org. Chem.* **1995**, *60*, 3851-3858.
44. Marco-Contelles, J.; Ruiz-Caro, J. *J. Org. Chem.* **1999**, *64*, 8302-8310.
45. Bhatt, U.; Christmann, M.; Quitschalle, M.; Claus, E.; Kalesse, M. *J. Org. Chem.* **2001**, *66*, 1885-1893.
46. Murakami, N.; Wang, W.; Aoki, M.; Tsutsui, Y.; Higuchi, K.; Aoki, S.; Kobayashi, M. *Tetrahedron Lett.* **1997**, *38*, 5533-5536.
47. Herscovici, J.; Egron, M. J.; Quenot, A.; Leclercq, F.; Leforestier, N.; Mignet, N.; Wetzler, B.; Scherman, D. *Org. Lett.* **2001**, *3*, 1893-1896.
48. Bussolo, V. D.; Caselli, M.; Romano, M. R.; Pineschi, M.; Crotti, P. *J. Org. Chem.* **2004**, *69*, 7383-7386.
49. Di Salvo, A.; David, M.; Crousse, B.; Bonnet-Delpon, D. *Adv. Synth. Catal.* **2006**, *348*, 118-124.
50. Kashyap, S.; Hotha, S. *Tetrahedron Lett.* **2006**, *47*, 2021-2023.
51. Schuff, B. P.; Mercer, G. J.; Nguyen, H. M. *Org. Lett.* **2007**, *9*, 3173-3176.
52. Bozell, J. J.; Tice, N. C.; Sanyal, N.; Thompson, D.; Kim, J-M.; Vidal, S. *J. Org. Chem.* **2008**, *73*, 8763-8771.
53. Williams, D. B. G.; Simelane, S. B.; Kinfu, H. H. *Org. Biomol. Chem.* **2012**, *10*, 5636-5642.
54. Flasz, J. T.; Hale, K. J. *Org. Lett.* **2012**, *14*, 3024-3027.
55. Banik, B. K.; Manhas, M. S. *Tetrahedron* **2012**, *68*, 10769-10779.
56. Grugel, H.; Albrecht, F.; Minuth, T.; Boysen, M. M. K. *Org. Lett.* **2012**, *14*, 3780-3783.

57. (a) Lin, H.-C.; Du, W.-P.; Chang, C.-C.; Lin, C.-H. *Tetrahedron Lett.* **2005**, *46*, 5071-5076. (b) Gorityala, B. K.; Cai, S.; Ma, J.; Liu, X.-W. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 3093-3095. (c) Gorityala, B. K.; Cai, S.; Lorpitthaya, R.; Ma, J.; Pasunooti, K. K.; Liu, X.-W. *Tetrahedron Lett.* **2009**, *50*, 676-679.
58. (a) Yadav, J. S.; Reddy, B. V.; Murthy, C. V.; Kumar, G. M. *Synlett* **2000**, 1450-1451. (b) Abdel-Rahman, A. A.-H.; Winterfeld, G. A.; Takhi, M.; Schmidt, R. R. *Eur. J. Org. Chem.* **2002**, 713-717. (c) Yadav, J. S.; Subba Reddy, B. V.; Sunder Reddy, J. S. *J. Chem. Soc., Perkin Trans. 1* **2002**, 2390-2394. (d) Babu, J. L.; Khare, A.; Vankar, Y. D. *Molecules* **2005**, *10*, 884-892. (e) Chen, P.; Wang, S. *Tetrahedron* **2012**, *68*, 5356-5362.
59. (a) Babu, B. S.; Balasubramanian, K. K. *Tetrahedron Lett.* **2000**, *41*, 1271-1274. (b) Masson, C.; Soto, J.; Bessodes, M. *Synlett* **2000**, 1281-1282. (c) Yadav, J. S.; Reddy, B. V. S.; Pandey, S. K. *New J. Chem.* **2001**, *25*, 538-540. (d) Yadav, J. S.; Reddy, V. S.; Reddy, K. B.; Satyanarayana, M. *Tetrahedron Lett.* **2002**, *43*, 7009-7012. (e) Hotha, S.; Tripathi, A. *Tetrahedron Lett.* **2005**, *46*, 4555-4558. (f) Zhang, G.; Liu, Q.; Shi, L.; Wang, J. *Tetrahedron* **2008**, *64*, 339-344. (g) Balamurugan, R.; Koppolu, S. R. *Tetrahedron* **2009**, *65*, 8139-8142. (h) Chen, P.; Lin, L. *Tetrahedron* **2013**, *69*, 4524-4531.
60. (a) Toshima, K.; Miyamoto, N.; Matsuo, G.; Nakata, M.; Matsumura, S. *Chem Commun.* **1996**, 1379-1380. (b) de Oliveira, R. N.; Filho, J. R. d. F.; Srivastava, R. M. *Tetrahedron Lett.* **2002**, *43*, 2141-2143. (c) Shanmugasundaram, B.; Bose, A. K.; Balasubramanian, K. K. *Tetrahedron Lett.* **2002**, *43*, 6795-6798. (d) Rafiee, E.; Tangestaninejad, S.; Habibi, M. H.; Mirkhani, V. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 3611-3614. (e) Agarwal, A.; Rani, S.; Vankar, Y. D. *J. Org. Chem.* **2004**, *69*, 6137-6140. (f) Gorityala, B. K.; Lorpitthaya, R.; Bai, Y.; Liu, X.-W. *Tetrahedron* **2009**, *65*, 5844-5848. (g) Levecque, P.; Gammon, D. W.; Jacobs, P.; De Vos, D.; Sels, B. *Green Chem.* **2010**, *12*, 828-835. (h) Kinfe, H. H.; Mebrahtu, F. M.; Sithole, K. *Carbohydr. Res.* **2011**, *346*, 2528-2532. (i) Chen, P.; Wang, S. *Tetrahedron* **2013**, *69*, 583-588.
61. Butler, A.; Walker, J. V. *Chem. Rev.* **1993**, *93*, 1937-1944.
62. Butler, A. *Coord. Chem. Rev.* **1999**, *187*, 17-35.

63. (a) Carter-Franklin, J. N.; Parrish, J. D.; Tschirret-Guth, R. A.; Little, R. D.; Butler, A. *J. Am. Chem. Soc.* **2003**, *125*, 3688-3689. (b) Carter-Franklin, J. N.; Butler, A. *J. Am. Chem. Soc.* **2004**, *126*, 15060-15066. (c) Sandy, M.; Carter-Franklin, J. N.; Martin, J. D.; Butler, A. *Chem. Commun.* **2011**, *47*, 12086-12088.
64. (a) Kimblin, C.; Bu, X.; Butler, A. *Inorg. Chem.* **2002**, *41*, 161-163. (b) Kadota, S.; Fantus, I. G.; Deragon, G.; Guyda, H. J.; Hersh, B.; Posner, B. I. *Biochem. Biophys. Res. Comm.* **1987**, *147*, 259-266. (c) Butler, A.; Carter-Franklin, J. N. *Nat. Prod. Rep.* **2004**, *21*, 180-188.
65. (a) Howarth, O. W.; Hunt, J. R. *J. Chem. Soc., Dalton Trans.* **1979**, 1388-1391. (b) Conte, V.; Di Furia, F.; Moro, S.; Rabbolini, S. *J. Mol. Cat A: Chem.* **1996**, *113*, 175-184.
66. Clague, M. J.; Butler, A. *J. Am. Chem. Soc.* **1995**, *117*, 3475-3484.
67. Khan, A. T.; Ali, S.; Basha, R. S.; Khan, M. M.; Lal, M. *Carbohydr. Res.* **2011**, *346*, 2629-2632.
68. (a) Parker, K. A.; Georges, A. T. *Org. Lett.* **2000**, *2*, 497-499. (b) Ghosh, R.; Chakraborty, A.; Maiti, D. K.; Puranik, V. G. *Org. Lett.* **2006**, *8*, 1061-1064.
69. Somsak, L. *Chem. Rev.* **2001**, *101*, 81-136 and references therein.
70. (a) Desiraju, G. R. *Crystal Engineering: The Design of Organic Solids*; Elsevier: Amsterdam, 1989. (b) Hunter, C. A. *Chem. Soc. Rev.* **1994**, *23*, 101-109.
71. (a) Bedini, E.; Forte, G.; De Castro, C.; Parrilli, M.; Dalla Cort, A. *J. Org. Chem.* **2013**, *78*, 7962-7969. (b) Spassova, M. K.; Bornmann, W. G.; Ragupathi, G.; Sukenick, G.; Livingston, P. O.; Danishefsky, S. J. *J. Org. Chem.* **2005**, *70*, 3383-3395.
72. (a) Khan, A. T.; Khan, M. M. *Carbohydr. Res.* **2010**, *345*, 154-159. (b) Khan, A. T.; Khan, M. M. *Carbohydr. Res.* **2010**, *345*, 2139-2145. (c) Sarkar, S.; Deka, J. K. R.; Hazra, J. P.; Khan, A. T. *Synlett* **2013**, *19*, 2601-2605. (d) Khan, A. T.; Choudhury, A.; Ali, S.; Khan, M. M. *Tetrahedron Lett.* **2012**, *53*, 4852-4857. (e) Khan, A. T.; Ali, S.; Dar, A. A.; Lal, M. *Tetrahedron Lett.* **2011**, *52*, 5157-5160. (f) Khan, A. T.; Parvin, T.; Choudhury, L. H. *J. Org. Chem.* **2008**, *73*, 8398-8402 and references cited therein.
73. (a) Gazi, S.; Ananthkrishnan, R. *RSC. Adv.* **2012**, *2*, 7781-7787. (b) Yadav, D. K.; Yadav, A. K.; Srivastava, V. P.; Watal, G.; Yadav, L. D. S. *Tetrahedron Lett.*

- 2012, 53, 2890-2893. (c) Yadav, D. K.; Patel, R.; Srivastava, V. P.; Watal, G.; Yadav, L. D. S. *Tetrahedron Lett.* **2010**, 51, 5701-5703. (d) Jiang, B.; Dou, Y.; Xu, X.; Xu, M. *Org. Lett.* **2008**, 10, 593-596. (e) Das, B.; Srinivas, Y.; Holla, H.; Laxminarayana, K.; Narender, R. *Tetrahedron Lett.* **2007**, 48, 6681-6683. (f) Bhalerao, D. S.; Akamanchi, K. G. *Synlett* **2007**, 19, 2952-2956.
74. (a) Zhang, Y.; Wang, P.; Song, N.; Li, M. *Carbohydr. Res.* **2013**, 381, 101-111. (b) Xiong, D. C.; Zhang, L. H.; Ye, X. S. *Adv. Synth. Catal.* **2008**, 350, 1696-1700.
75. Khan, A. T.; Parvin, T.; Choudhury, L. H. *Tetrahedron* **2009**, 65, 9513-9526.
76. (a) Desiraju, G. R.; Steiner, T. *The Weak Hydrogen Bond in Structural Chemistry and Biology*, Oxford University Press, 1999. (b) Steiner, T. *Angew. Chem. Int. Ed.* **2002**, 41, 48-76.



*Brief Literature
Survey on
Isocyanide-based
Multicomponent
Reactions*

Chapter IV

1.1 Introduction

Novel synthetic methodologies are highly desirable in recent times. Owing to their economic constraints and environmental implications, synthetic chemists are keen to look for an easier and simpler novel procedure.¹ Indeed, MCRs² are found to be well-designed synthetic strategies which can accomplish this challenging task in a single operation. By definition “*Multicomponent reactions (MCRs) are convergent reactions, in which three or more starting materials react to form a product, where basically all or most of the atoms contribute to the newly formed product.*”- Ivar Ugi.³ The schematic representation of the defined MCRs is shown in Figure 5.

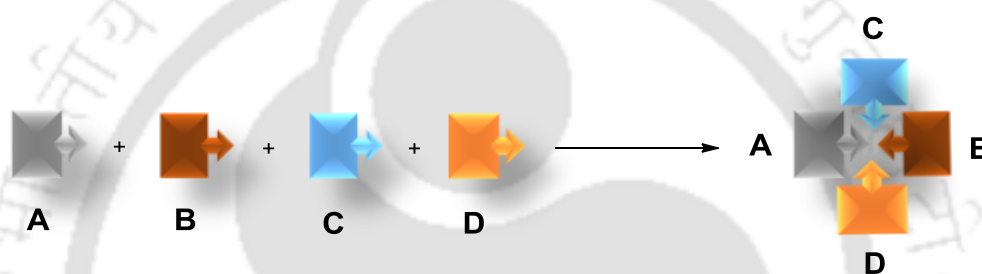


Figure 5. Multicomponent reactions

One-pot MCRs approaches are much easier to execute than a complicated multistep synthesis as depicted in Figure 6. The synthesis of the complex molecule with high stereo-, regio- and chemo selectivity is achieved through multicomponent reactions.⁴ MCRs render a facile access to high convergence and efficient synthesis of diverse molecules⁵ and this attribute of MCRs make them superior from other synthetic reactions.

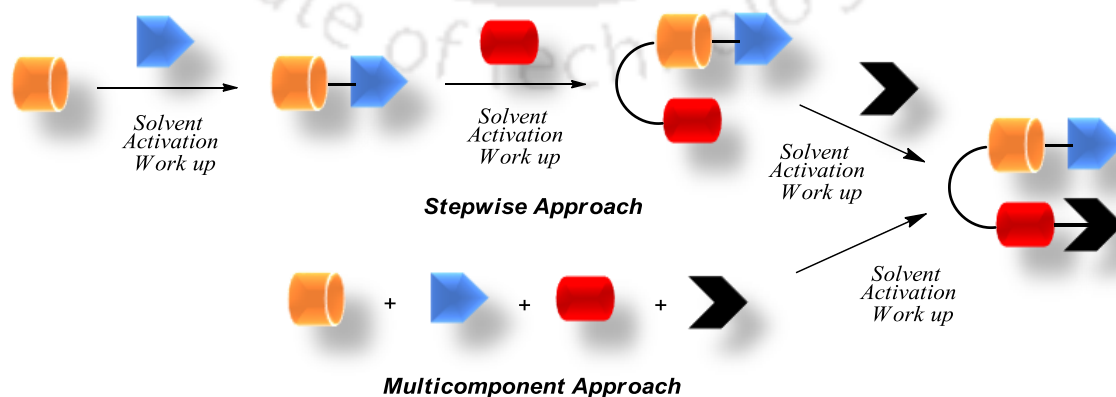


Figure 6. Stepwise vs Multicomponent Approach

Multicomponent reaction promote the synthesis of structurally complex architecture molecules in easier synthetic pathway.⁶ In addition to that, MCRs are particularly deployed in synthetic pharmaceuticals.⁷ Scaffolds assembled from MCRs are found to be potent drugs which are extensively utilized in biomedical applications.⁸ The historical overview of MCRs⁹⁻¹⁴ is shown in Figure 7.

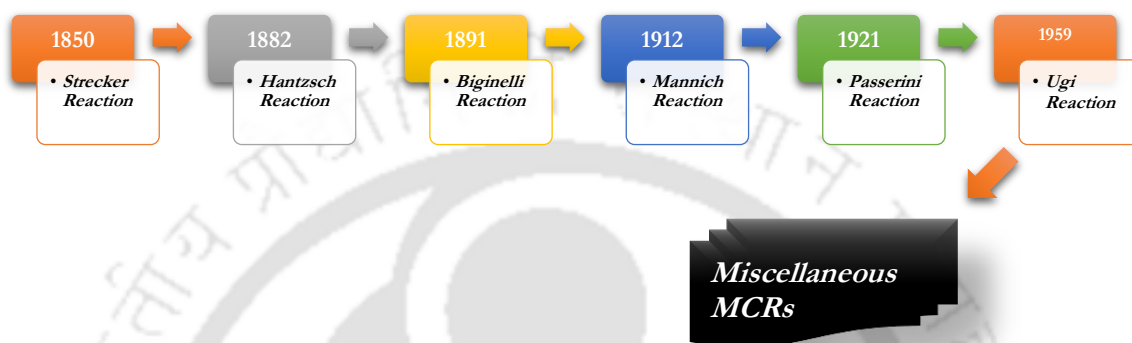
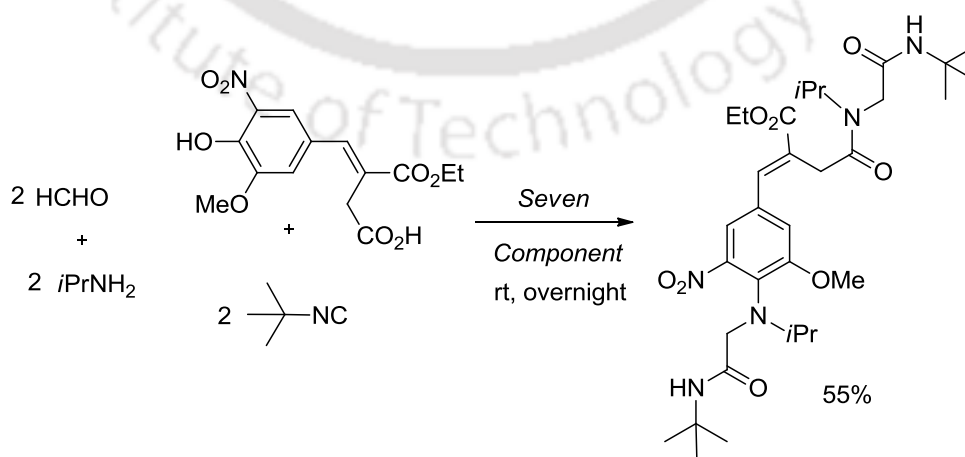


Figure 7. Historical overview of MCRs

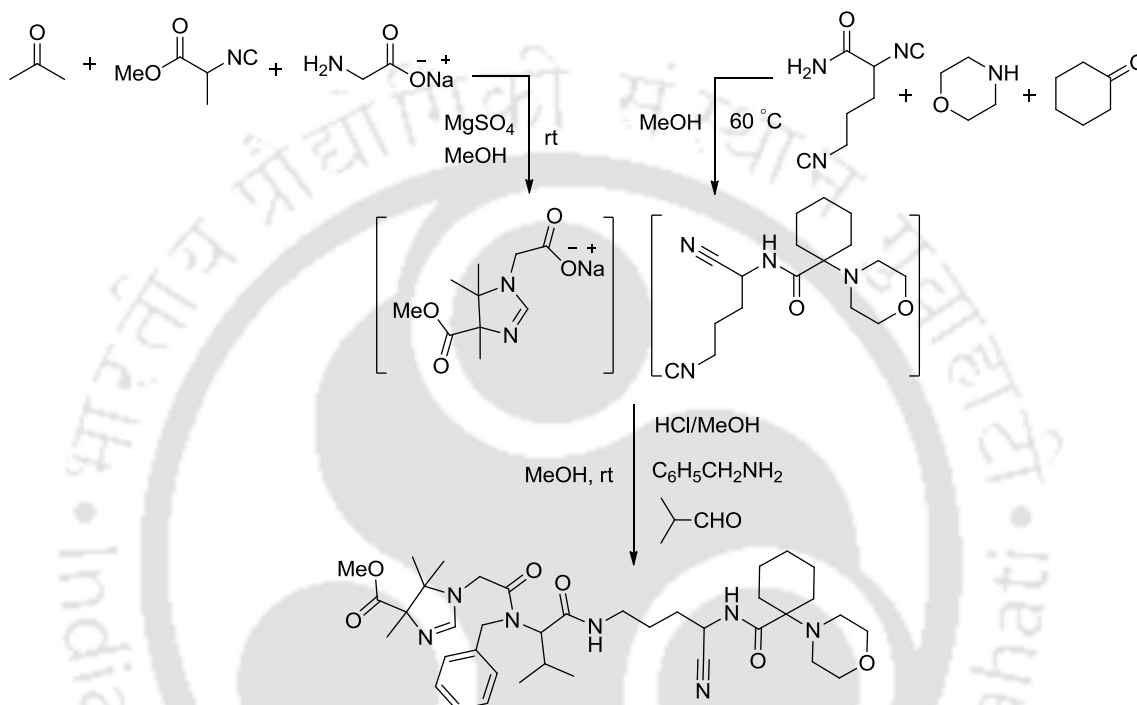
1.2 Recent Development of MCRs

Brauch *et al.*¹⁵ demonstrated that caffeic acid derivative reacts with formaldehyde, isopropyl amine and *tert*-butyl isonitrile in a one-pot sequential seven component reaction to afford Ugi-Mumm/Ugi-Smiles product in 55% yield. Overall eight new covalent bonds were formed in this seven component reaction which is shown in Scheme 42.



Scheme 42

Orru *et al.*¹⁶ developed a one-pot eight component reaction from two three-component followed by one four-component reactions which led to an overall eight component product. The resultant product endures with nine new covalent bonds formation. Eventually, this is the only methodology which reports highest number of component reactions i.e. eight till date in literature as depicted in Scheme 43.



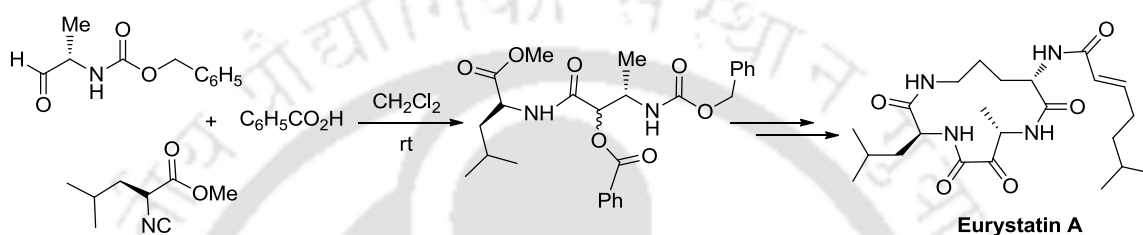
□ Scheme 43

1.3 Isocyanide Based Multicomponent reactions

Isocyanide are found to be most reactive, highly stereo and regioselective, and possess functional groups tolerance and higher bond forming efficiency. Due to their distinct feature, it is extensively utilized in multicomponent reactions. Moreover, it found to be the subclass of multicomponent reactions as Isocyanide based multicomponent reactions (IMCRs).¹⁷ In general, IMCRs are normally rapid and provide diversity for discovery of new heterocyclic entities. However, the pathways of the reacting isocyanides in MCRs might be as electrophile or nucleophile and in some cases as radical reactions. Nowadays, isocyanide based multicomponent reactions are examined as overall extension of Passerini and Ugi reactions.

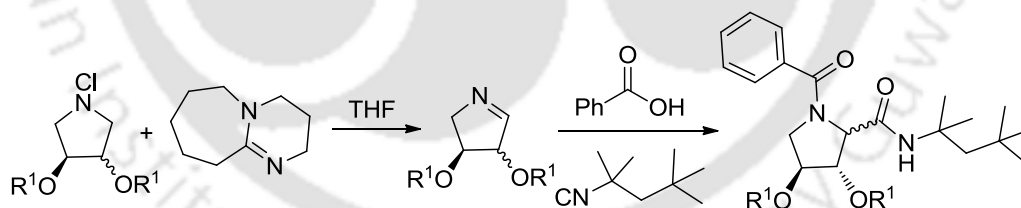
1.3a Synthetic Utility of Three-Component IMCRs

Schmidt and co-workers¹⁸ demonstrated that protected (*S*)-alaninal reacts with (*S*)-2-isocyano-4-methylpentanoate and benzoic acid in dichloromethane at room temperature subject to Passerini product of diastereomeric mixture, which eventually further undergoes three steps to attain a target molecule of eurystatin A as represented in Scheme 44.



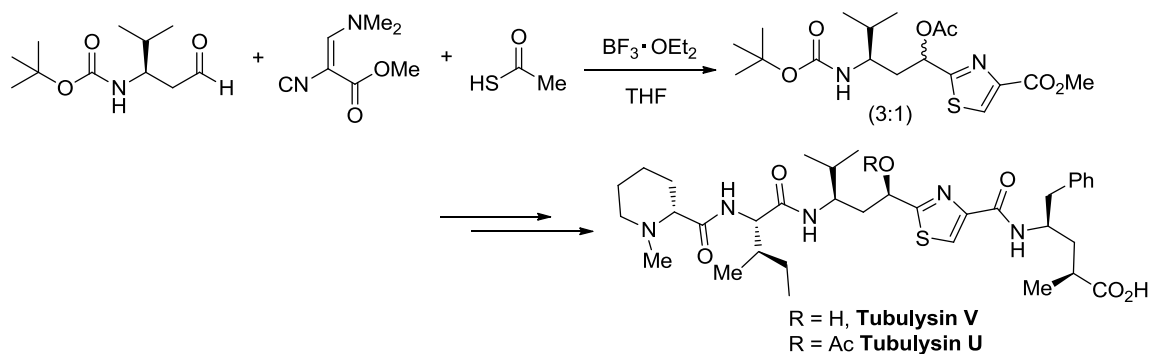
Scheme 44

Davis *et al.*¹⁹ used Joullie'-Ugi reaction to access functionalized azasugar moiety from five membered cyclic imine which is depicted in Scheme 45. Some of the synthesized compounds exhibit a selective antiviral activity.



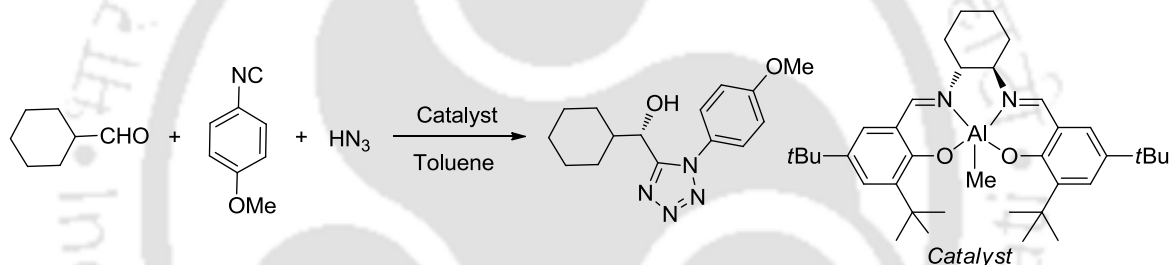
Scheme 45

Domling *et al.*²⁰ used Passerini-type three-component reaction as a key synthetic intermediate step for the synthesis of tubulysins V and U. Typically, *L*-*N*-Boc homovaline aldehyde react with thioacetic acid and Schollkopf's isonitrile in presence of boron trifluoride etherate at $-78\text{ }^{\circ}\text{C}$ to afford 3:1 diastereomeric mixture of Passerini adduct product which further undergoes several steps to attain target tubulysins V and U as shown in Scheme 46.



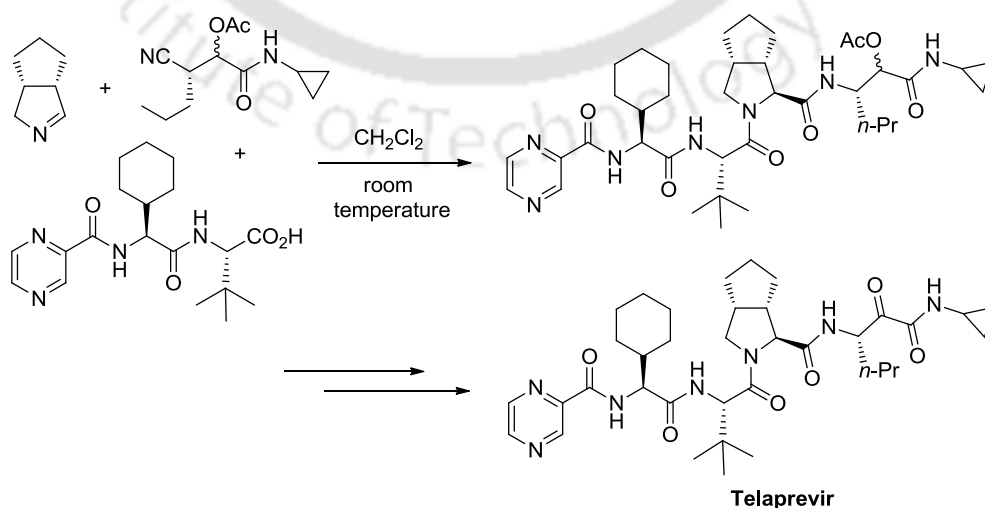
□ Scheme 46

Zhu *et al.*²¹ utilized a three-component Passerini reaction for the construction of enantioselective 5-(1-hydroxyalkyl)tetrazoles from aldehydes, isocyanides, and hydrazoic acid catalyzed by [(salen)Al(III)Me] in toluene with excellent yields as represented in Scheme 47.



□ Scheme 47

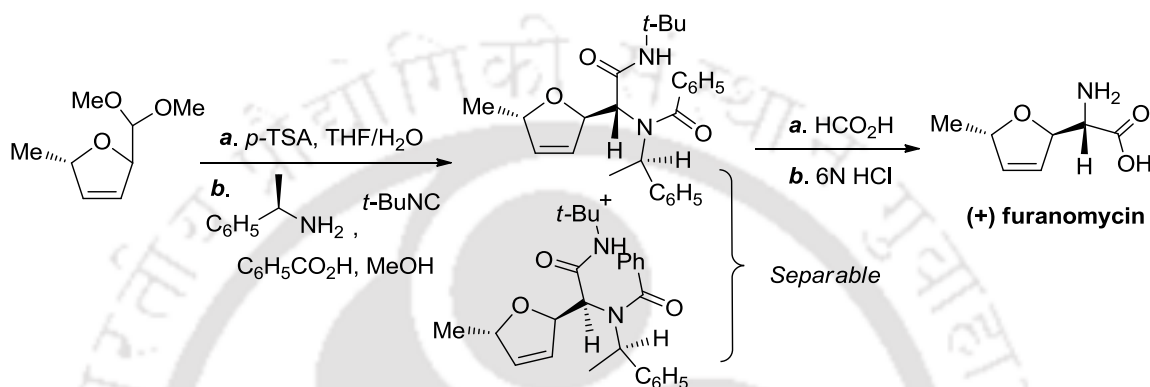
Ruijter *et al.*²² demonstrated a significant Ugi-type three-component approach used for the synthesis of biologically active synthetic pharmaceutical telaprevir which is shown in Scheme 48.



□ Scheme 48

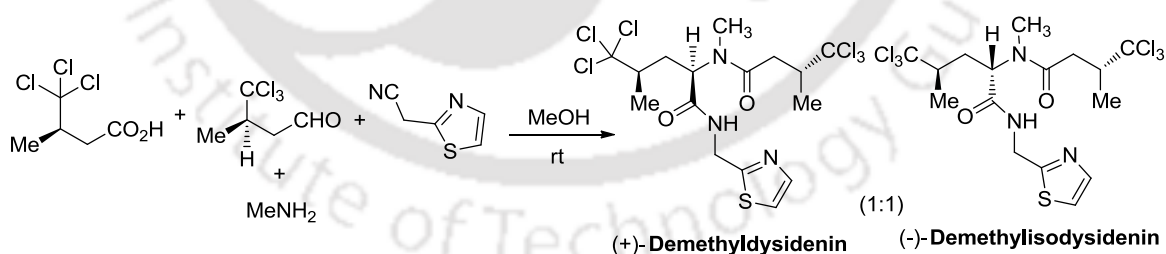
1.3b Synthetic Utility of Four-Component IMCRs

Joullié *et al.*²³ synthesized an antibiotic furanomycin using four component Ugi reaction as a key step. Deprotected furanoacetal react with *R*-methyl benzylamine, *tert*-butyl isocyanide and benzoic acid led to a mixture of diastereomeric products, which are separable and it further undergoes debenzoylation followed by acidic cleavage to furnish the target molecule as shown in Scheme 49.



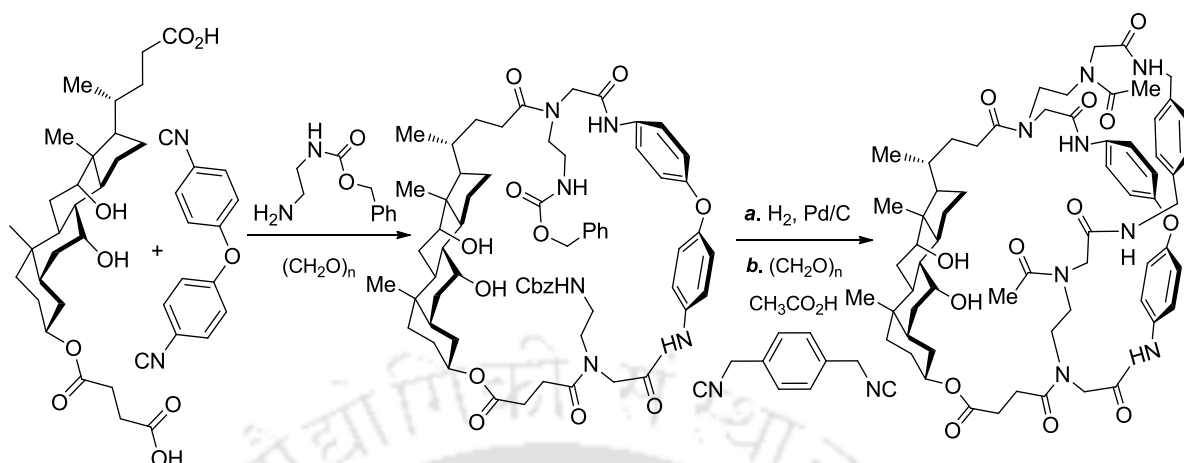
Scheme 49

Williard *et al.*²⁴ demonstrated the synthesis of (+)-demethyldysidenin and (-)-demethylisodysidenin from synthesized reactants of carboxylic acid and aldehyde with isocyanide and methylamine in methanol at room temperature to access a separable 1:1 diastereomeric products which is shown in Scheme 50.



Scheme 50

Rivera *et al.*²⁵ established that double Ugi four component reaction were explored for the synthesis of macrobicyclic from dicarboxylic acid, formaldehyde, diisocyanide and amine. The first obtained macrocycle containing protected amine were deprotected which led to reactive free amine and subsequently it reacted with different isocyanide for second double Ugi four component reaction which delivered a complex macrobicyclic molecule with 16 new covalent bonds as shown in Scheme 51.

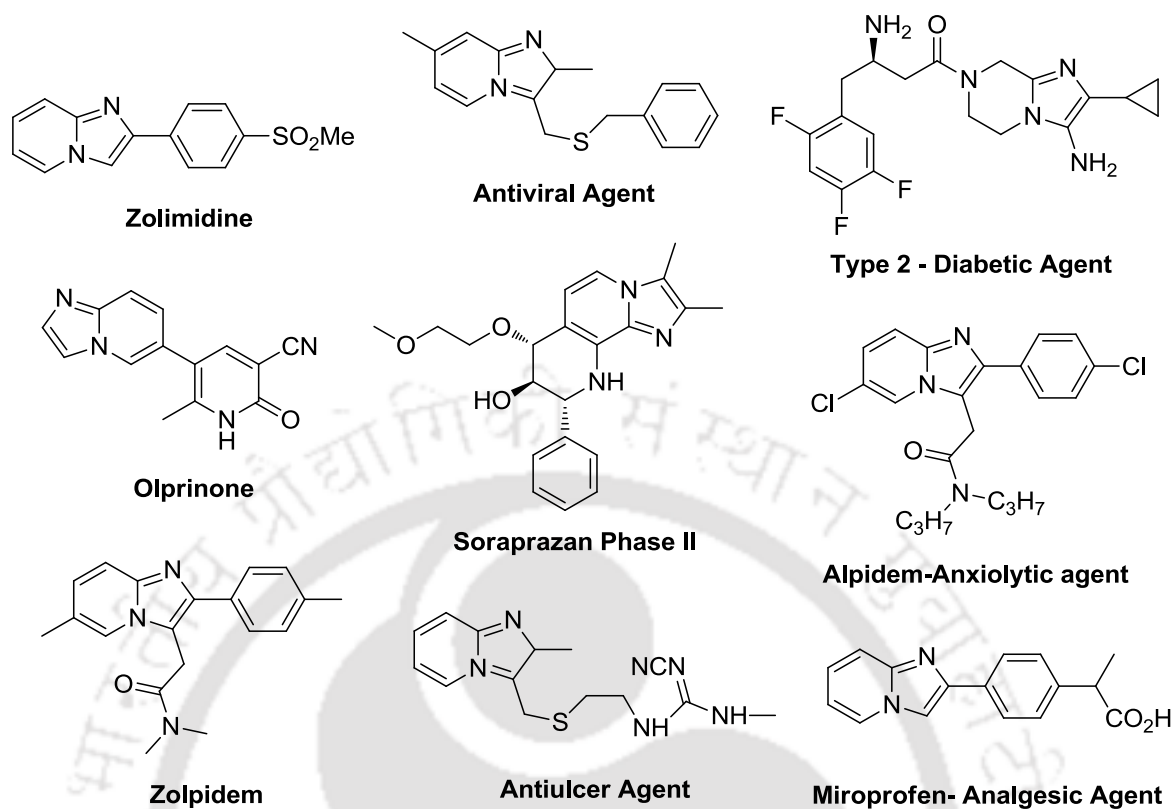


□ Scheme 51

In general, the above literature survey reveals the growth of the multicomponent reactions in challenging synthetic platform. The theme of the present research work is to synthesize imidazo[1,2-*a*]pyridines and aminoamidine derivatives. The following section discusses their importance along with the synthetic methods:

➤ 1.4 Imidazo[1,2-*a*]pyridine Importance

Imidazo[1,2-*a*]pyridine displays a significant role as a pharmacophore and it is extensively utilized in medicinal and synthetic pharmaceuticals.²⁶⁻²⁸ A diverse range of molecules possessing imidazo[1,2-*a*]pyridine scaffolds attribute a remarkable attention in biological activities such as antibacterial,²⁹ antiviral,³⁰ anti-inflammatory,³¹ antihepatitis C,³² anticonvulsant,³³ antipyretic,³⁴ antiulcer,³⁵ antianxiety,^{36a} anticancer^{36b} hypnotic and cardiotonic agent.³⁷ It also performs a unique role as calcium channel blockers,³⁸ cyclin-dependent kinase inhibitors,³⁹ bradykinin B2 receptor antagonists⁴⁰ and amyloid inhibitors.⁴¹ In addition to that, it also act as angiotensin II antagonists,⁴² dopamine D4 receptor agonists⁴³ and 5HT₃ antagonists.⁴⁴ The most common drugs containing imidazo[1,2-*a*]pyridine skeleton are Zolpidem-treatment of insomnia,⁴⁵ Zolimidine-antiulcer agent,³⁴ Alpidem-anxiolytic agent,⁴⁶ and Olprinone-treatment for heart failure⁴⁷ which is shown in Figure 8.

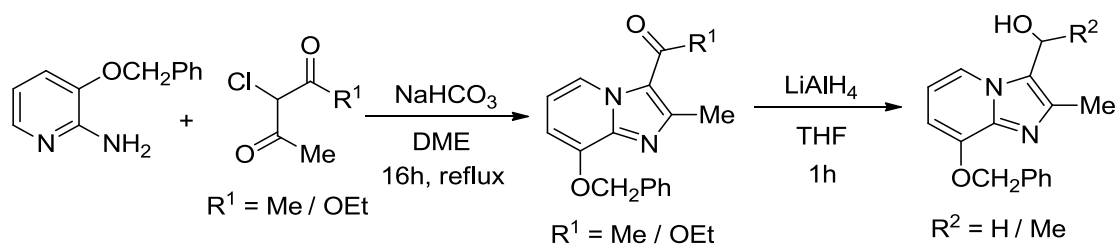


● **Figure 8.** Some of the biologically active compounds containing imidazo[1,2-*a*]pyridine framework

Despite these facts, imidazo[1,2-*a*]pyridine also exhibit an interesting application in material science.⁴⁸ Imidazo[1,2-*a*]pyridines are widely used as microglial cell visualization and as probe for benzodiazepine receptors.⁴⁹

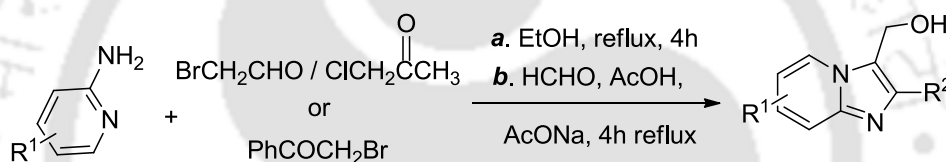
○ 1.4a Synthetic utility of Imidazo[1,2-*a*]pyridines

Starrett *et al.*⁵⁰ demonstrated that condensation of 2-aminopyridines with α -chloro-1,3-dicarbonyl compound in DME using sodium bicarbonate under reflux condition led to C-3 carbonyl imidazo[1,2-*a*]pyridine derivatives. The obtained product on reduction with lithium aluminium hydride in tetrahydrofuran which afford hydroxy alkyl imidazo[1,2-*a*]pyridine derivatives is shown in Scheme 52. Further it is elaborate as functionalized primary amine derivative at C-3 position to exhibit as effective antiulcer agents.



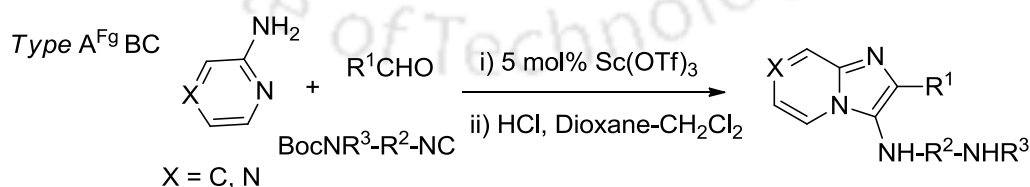
□ Scheme 52

Gueiffier *et al.*⁵¹ reported that direct hydroxymethylated imidazo[1,2-*a*]pyridines were synthesized using 2-aminopyridines, α -halo-(acetaldehyde/acetone)/phenacyl bromide in ethanol under reflux condition, followed by formaldehyde and sodium acetate in acetic acid which is depicted in Scheme 53. The resulted product tagged with thiol moiety is an important antiviral agent.



□ Scheme 53

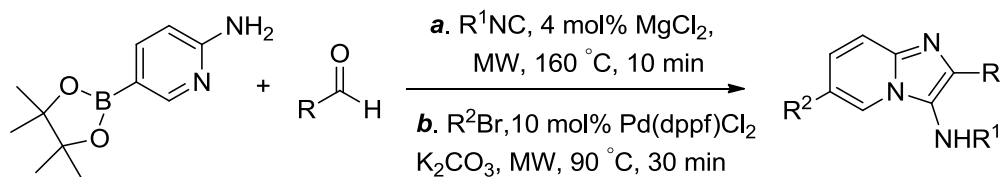
Bencsik *et al.*⁵² developed a diverse range of drug like library A^{FG}BC type imidazo[1,2-*a*]pyridines by employing Boc protected isocyanide, amidine and aldehyde in presence of catalytic amount of scandium triflate. In the subsequent step, cleavage of Boc in acidic medium afforded aminosubstituted imidazo[1,2-*a*]pyridines which is illustrated in Scheme 54.



□ Scheme 54

Dimauro *et al.*⁵³ explored a microwave-assisted one-pot synthesis of imidazo[1,2-*a*]pyridines via tandem cyclization followed by Suzuki coupling using bifunctionalized 2-aminopyridine-5-boronic ester as key starting material. The significant feature of this

reaction is that during cyclization process boronate ester remains unaffected and it is extensively utilized for Suzuki coupling as shown in Scheme 55.



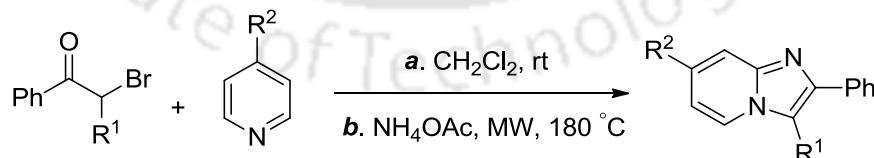
□ Scheme 55

Guchhait *et al.*^{54,36b} demonstrated a microwave assisted Ugi three component reaction for the synthesis of de-*tert*-butylated imidazo[1,2-*a*]pyridines in presence of catalytic amount of Zirconium(IV) chloride along with one equivalent of fluoroboric acid which is shown in Scheme 56.



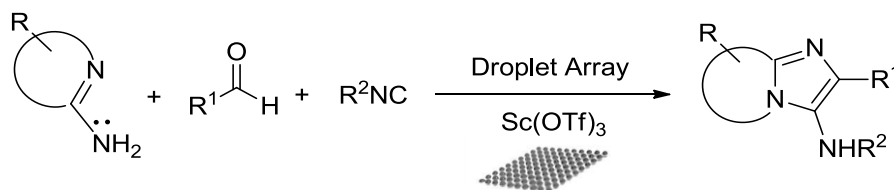
□ Scheme 56

Adib *et al.*⁵⁵ developed a microwave mediated synthesis of imidazo[1,2-*a*]pyridines through in situ generated *N*-phenacylpyridinium bromides, followed by nucleophilic ammonium acetate addition under solvent-free conditions with good yields which is depicted in Scheme 57.



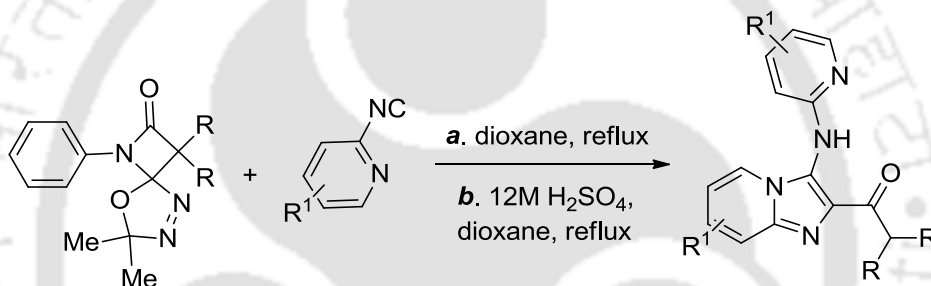
□ Scheme 57

Balakirev *et al.*⁵⁶ developed a drug like fluorophore molecule of imidazo[1,2-*a*]pyridines from Ugi three component reaction of aromatic aldehyde, amidine and isocyanide using droplet array method which is shown in Scheme 58.



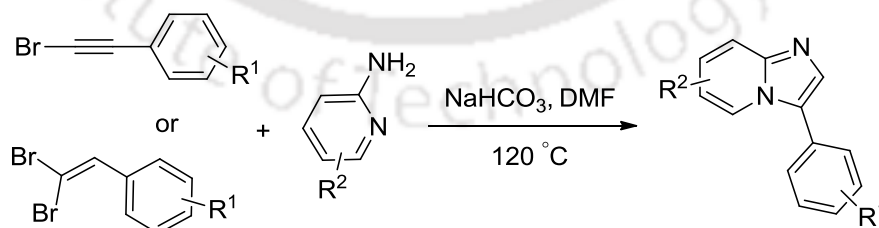
□ Scheme 58

Cheng *et al.*⁵⁷ reported that β -lactam carbenes generated from spiro[β -lactam-4,2'-oxadiazolines] react with 2-pyridyl isocyanides in dioxane under reflux condition, which subsequently undergoes an acid hydrolysis leading to pyridyl based imidazo[1,2-*a*]pyridines as shown in Scheme 59. The resulting product serves as a fluorescence probe for mercury ion.



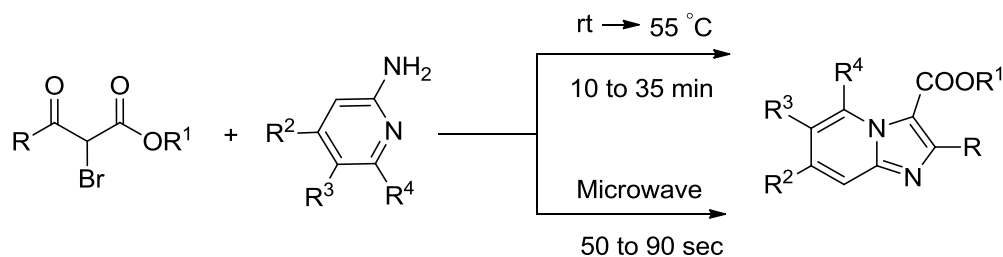
□ Scheme 59

Zhou *et al.*⁵⁸ demonstrated the synthesis of 3-arylimidazo[1,2-*a*]pyridines from 2-aminopyridine, alkynyl/alkenyl halide with sodium bicarbonate in DMF at 120 °C with good yields as depicted in Scheme 60.



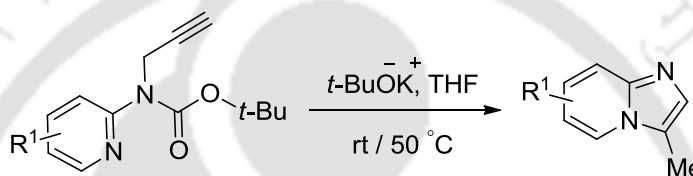
□ Scheme 60

Adimurthy *et al.*⁵⁹ accomplished a solvent-free synthesis of highly substituted imidazo[1,2-*a*]pyridines via thermal and microwave irradiated method using α -bromo- β -keto esters and aminopyridines in shorter reaction time as shown in Scheme 61.



□ Scheme 61

Savic *et al.*⁶⁰ developed an efficient route for the synthesis of imidazo[1,2-*a*]pyridines through base assisted intramolecular cyclization from protected N-propargylated-2-aminopyridine derivatives as represented in Scheme 62.



□ Scheme 62

Over the years, some other methods were developed for the synthesis of imidazo[1,2-*a*]pyridines using $\text{Sc}(\text{OTf})_3$,⁶¹ ZnCl_2 ,⁶² Montmorillonite clay K10,⁶³ ionic liquid,⁶⁴ glyoxalic acid⁶⁵ and *p*-toluenesulfonic acid.⁶⁶ Adib *et al.* developed the synthesis of imidazo[1,2-*a*]pyridine in water.⁶⁷ However, the above methods suffer from some of the drawbacks such as harsh reaction conditions, need of expensive and excess amount of catalyst, longer reaction time and awkward work-up procedure. Eventually, still there is a scope to upgrade a new methodology which might work better in terms of yield, reaction condition and substrate scope compatibility.

➤ 1.5 Amidine & Aminoamidine Importance

Amidines are extensively utilized for the constructions of nitrogen containing hetero cycles owing to well-designed structural units of bioactive scaffolds.⁶⁸ Naturally occurring amidines are Birnbaumins A and B, Efraeptins, Neofraeptins and Prostatins A and B which are isolated from microorganisms and Fungi.⁶⁹ Flustramine C and Perophoramidine are isolated from Marine Invertebrates.⁷⁰ Glomerulatin A-C isolated from higher plants⁷¹ are widely used natural products as well as pharmaceuticals. Hordatine B, Batzelladine B and Agmatine also possess attractive bioactive applications

which are all depicted in Figure 9. The important building blocks of the subclass of amidines are α -amino amidines, widely used for the synthesis of antibiotic, antitumor, iminopeptides,⁷² Bleomycin,⁷³ bioluminescence of *Cypridina hilgendorffii*⁷⁴ and novacaine derivatives.⁷⁵

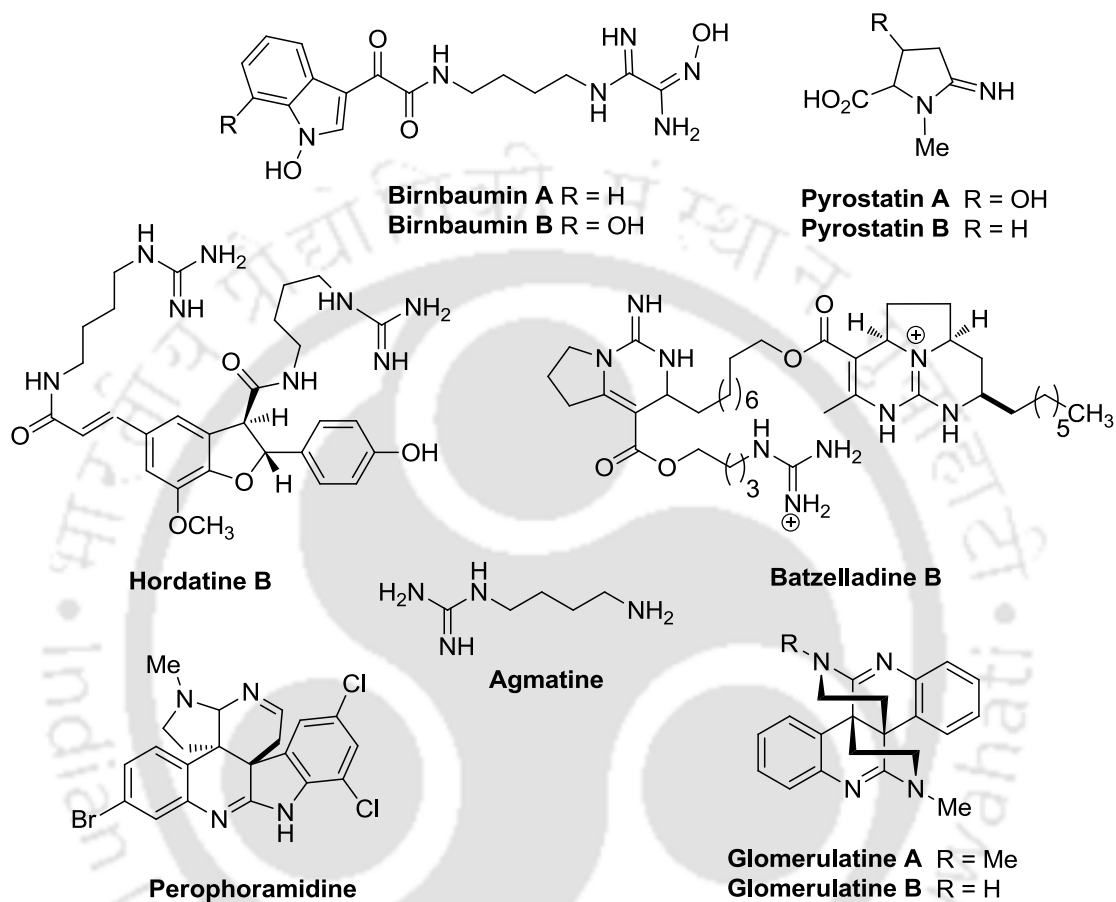
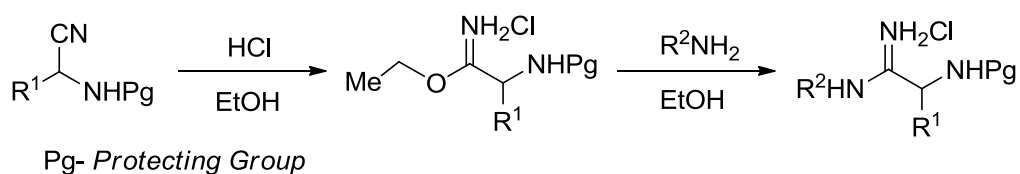


Figure 9. Biologically active compounds containing amidine skeleton

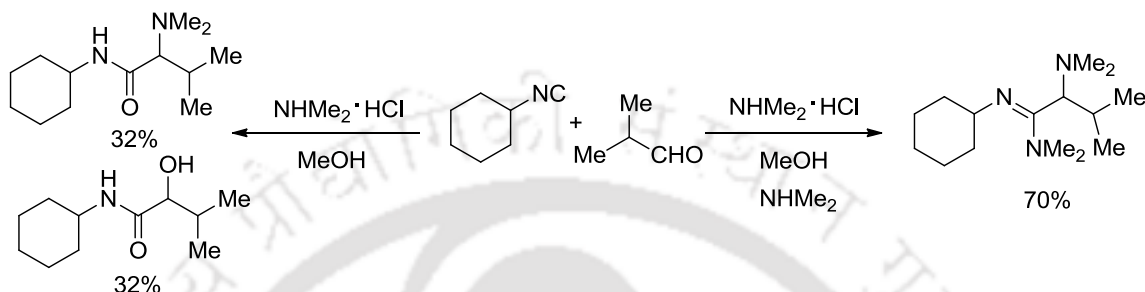
1.5a Amino Amidines Synthetic Methods

Mengelberg *et al.*⁷⁶ demonstrated the synthesis of α -amino amidine from N-protected α -amino iminoethers as intermediate followed by amination in ethanol as shown in Scheme 63.

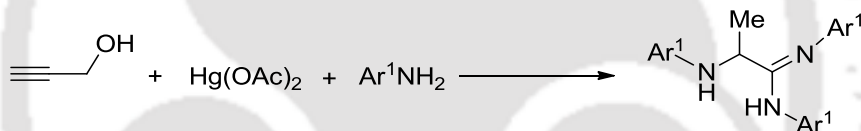


Scheme 63

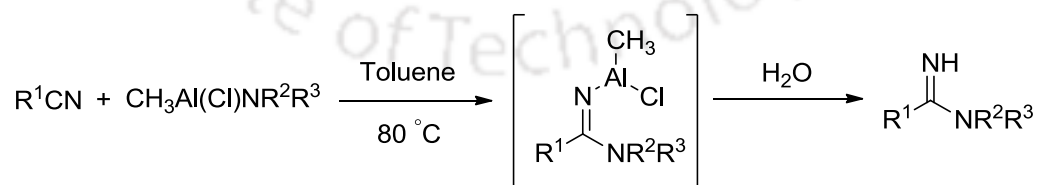
McFarland *et al.*⁷⁷ developed the synthesis of α -amino amidine using cyclohexyl isocyanide, isobutyraldehyde and dimethyl amine in presence of dimethylamine hydrochloride as a catalyst in methanol. However, in absence of dimethyl amine it afforded α -(hydroxyl/dimethylamino) amide compounds with 32% yield each as depicted in Scheme 64.



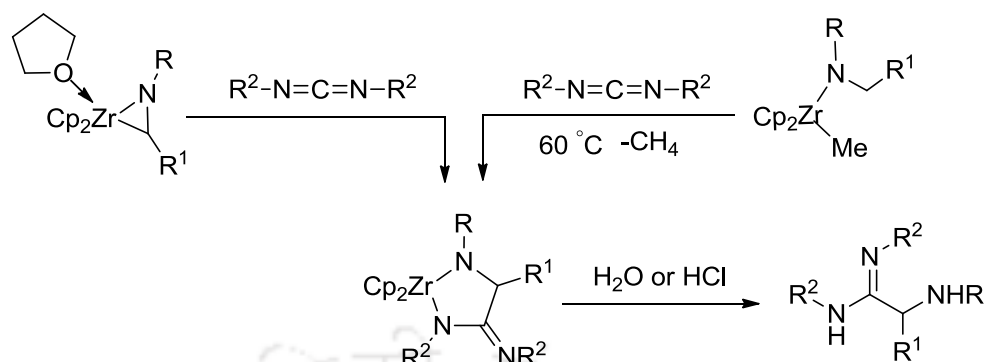
Barluenga *et al.*⁷⁸ accomplished 2-amino-*NN'*-triarylpropionamidines using aromatic amines, prop-2-ynol and mercury(II) acetate which is shown in Scheme 65.



Garigipati *et al.*⁷⁹ established an efficient method for the conversion of nitriles to amidines using alkyl chloroaluminium amide in toluene at 80 °C which is shown in Scheme 66. However, it is also extensively utilized for the synthesis of guanidines from N-substituted cyanamides

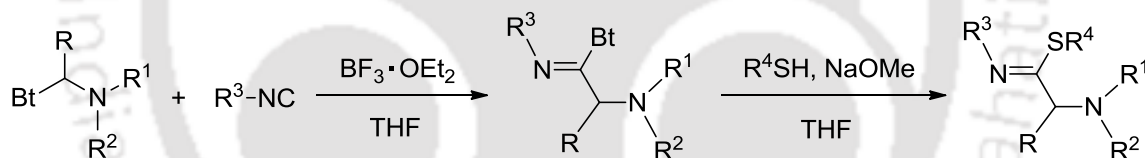


Norton *et al.*⁸⁰ reported that zirconaaziridines reacts with carbodiimides subject to zirconocene-imine complex which eventually undergoes protic cleavage leading to the formation of α -amino amidine. However, zirconaaziridines were also generated *in situ* on thermolysis of $\text{Cp}_2\text{Zr}[\text{N}(\text{R})\text{CH}_2(\text{R}^1)]\text{Me}$ as represented in Scheme 67.



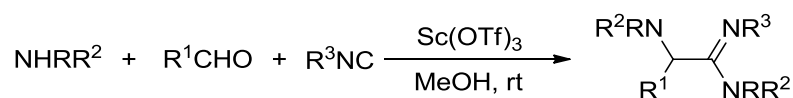
□ Scheme 67

Katritzky *et al.*⁸¹ described synthesis of highly substituted α -amino-imidoylbenzotriazole using N-(aminoalkyl)benzotriazoles and isocyanide in presence of boron trifluoride as catalyst in THF. The resulting α -amino-imidoylbenzotriazole further undergoes nucleophilic displacement with thiol to form a functionalized α -amino-thioacetimidates as depicted in Scheme 68.



□ Scheme 68

Recently Keung *et al.*⁸² demonstrated a one-pot three-component Ugi condensation reaction for the synthesis of α -amino amidines using secondary amine or primary amine, isobutyraldehyde or aromatic aldehyde and cyclohexyl isocyanide by employing scandium(III) triflate as a catalyst which is shown in Scheme 69.



□ Scheme 69

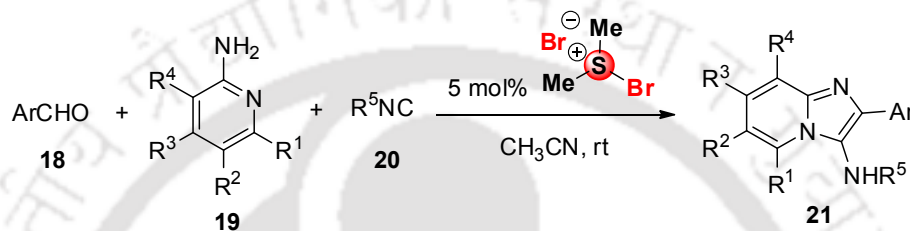
Results and Discussion

*Synthesis of Imidazo[1,2-a]pyridines in
One-pot three component reaction using
Bromodimethylsulfonium bromide (BDMS)
as a Catalyst*

Chapter V

● Results and Discussion

Imidazo[1,2-*a*]pyridine derivatives importance and their synthetic strategies have already been discussed in Chapter IV. In this Chapter an opportune method was devised for the synthesis of imidazo[1,2-*a*]pyridine by employing one-pot three-component reaction by using aromatic aldehyde, aromatic amidine and isocyanide in presence of 5 mol% of bromodimethylsulfonium bromide (BDMS) at room temperature, which is shown in Scheme 70.



Amidines

- 19a:** R¹ = R² = R³ = R⁴ = H
19b: R¹ = R³ = R⁴ = H, R² = Me
19c: R¹ = R² = R³ = H, R⁴ = Me
19d: R¹ = R³ = Me, R² = Br, R⁴ = H

Isocyanides

- 20a:** *tert*-Butyl
20b: Cyclohexyl
20c: 2-Morpholinoethyl

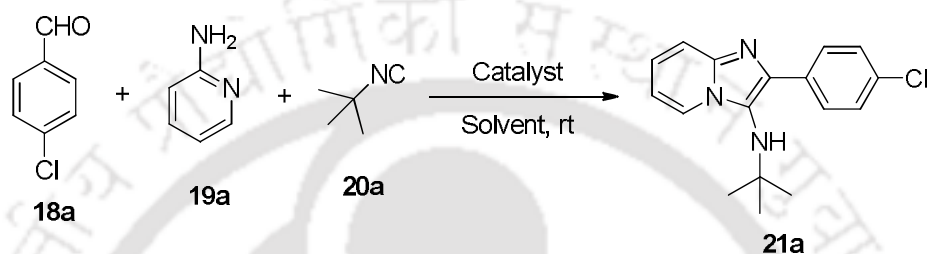
□ **Scheme 70.** Synthesis of imidazo[1,2-*a*]pyridine derivatives

At the outset, the reaction was performed with *p*-chlorobenzaldehyde (**18a**), amidine (**19a**) and *tert*-butyl isocyanide (**20a**) as a model substrate to find out the optimal reaction condition and the obtained results are summarized in Table 9. Initial screening of Lewis acid catalyst ceric ammonium nitrate (CAN) and Cu(OTf)₂ provided lower yield of the product **21a** (Table 9, entry 2 and 3). Moreover, various trial reactions were investigated and we observed that 5 mol% BDMS at room temperature gave the best result in terms of yield and reaction time. However, a systematic evaluation of the various solvent(s) system revealed that acetonitrile was a suitable solvent for the synthesis of imidazo[1,2-*a*]pyridine **21a**.

The optimized protocol was further examined with 4-bromobenzaldehyde, 2-aminopyridine and *tert*-butyl isocyanide in presence of 5 mol% BDMS under similar reaction condition and the corresponding product **21b** were obtained in 85% yield. Similarly, on reaction with 3-hydroxybenzaldehyde, 2-aminopyridine and *tert*-butyl isocyanide gave the required product **21c** with 90% yield. Encouraged by these results, a

variety of substituents on the aromatic ring of the aldehyde were treated with 2-amino-5-methylpyridine and *tert*-butyl isocyanide under identical reaction conditions which provided the resulting product **21d-k** in good to excellent yields. It is noteworthy to mention that electron withdrawing substituents on the aldehyde took lesser reaction time as compared to the electron donating group.

Table 9. Optimization of the reaction conditions^a



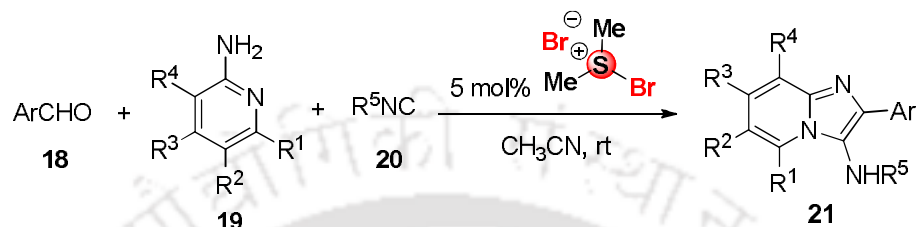
Entry	Catalyst	Solvent	Time (h)	Yield (%) ^b
1	–	CH ₃ CN	14	26
2	10 mol% CAN	CH ₃ CN	12	64
3	10 mol% Cu(OTf) ₂	CH ₃ CN	9	55
4	5 mol % BDMS	MeOH/DCM	3	75
5	5mol % BDMS	CH ₃ CN/MeOH	2	85
6	5mol % BDMS	CH ₃ CN/DCM	2.5	80
7	5mol % BDMS	THF	3.5	77
8	5mol % BDMS	1,4 dioxane	3	72
9	5mol % BDMS	DMF	3.5	68
10	5mol % BDMS	CH ₃ CN	0.5	95

^aThe reaction was performed using 1 mmol scale of *p*-chlorobenzaldehyde, 2-amino-5-methylpyridine and *tert*-butyl isocyanide at room temperature. ^bIsolated yield.

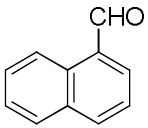
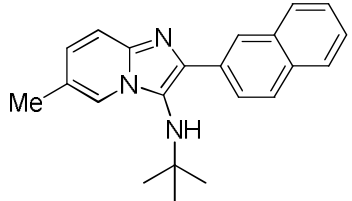
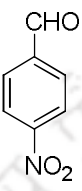
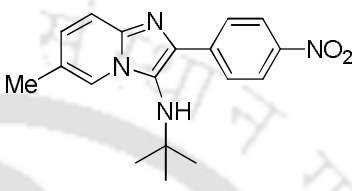
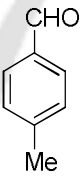
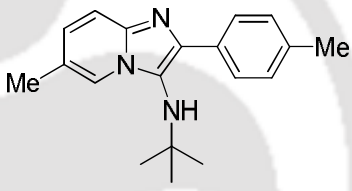
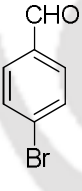
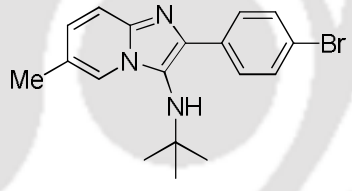
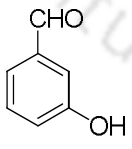
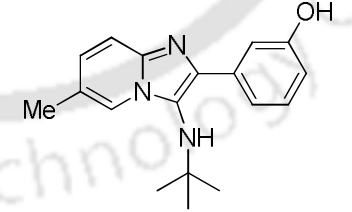
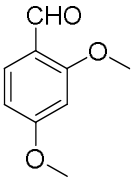
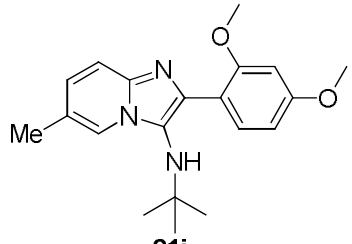
To our delight, next we set to explore the scope of the reaction with 2-amino-3-methylpyridine, 4-chlorobenzaldehyde and cyclohexyl isocyanide with a catalytic amount of BDMS afforded the product **21i** in 90% yield. Subsequently, the steric factor on the substrate was also examined on reaction with sterically hindered 2-amino-5-bromo-4,6-dimethylpyridine, aromatic aldehydes and isocyanides also provided the corresponding product **21m** and **21n** in 87% and 96% yield, respectively. From these successful results, we were motivated to explore the present protocol compatibility with aliphatic aldehyde. Upon reaction with valeraldehyde, 2-amino-5-methylpyridine and 2-

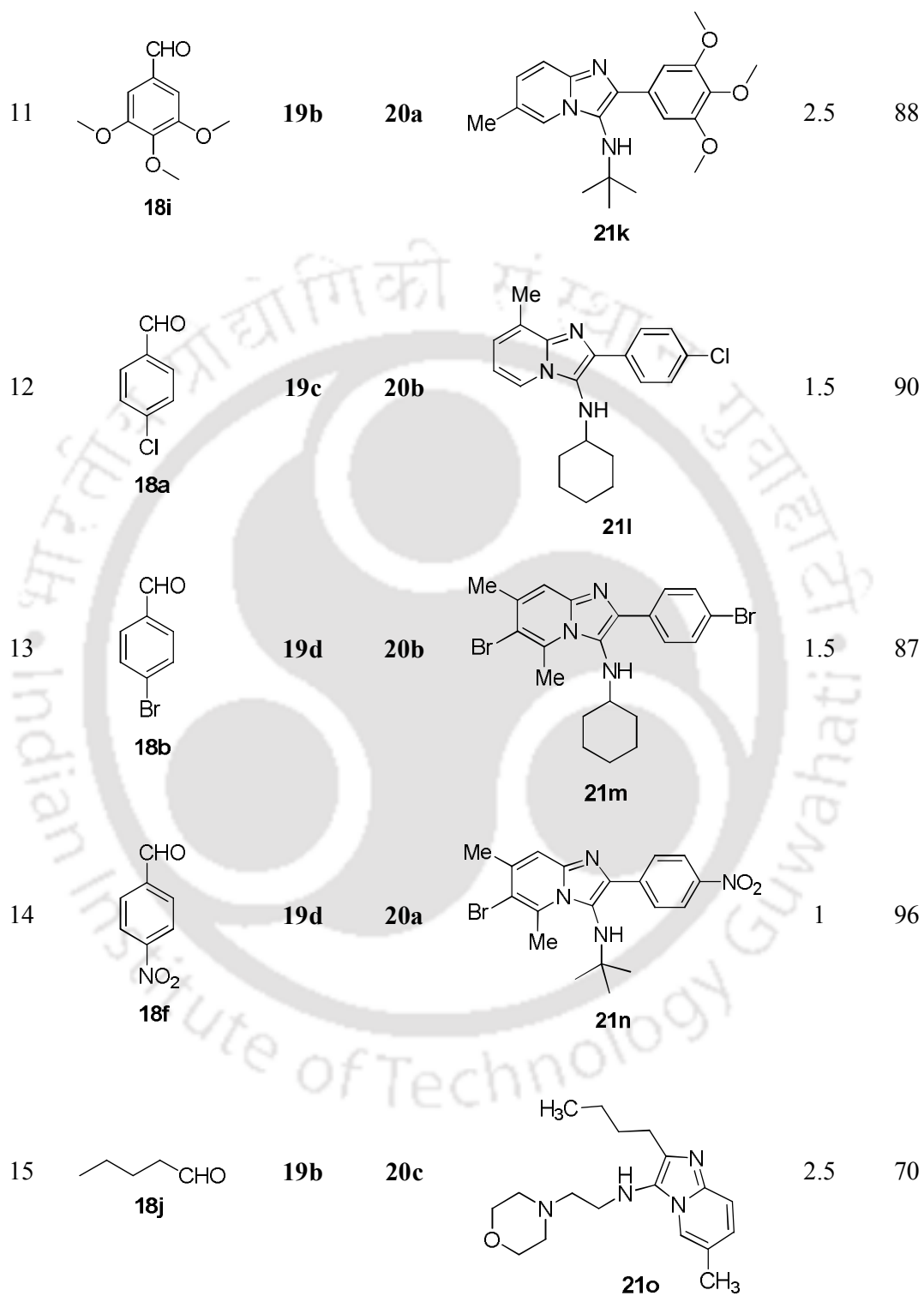
morpholinoethyl isocyanide under similar reaction conditions afforded the resulting product **21o** in good yield. The output of the results indicated that the present protocol work wells with a variety of amidines and aldehydes, which are summarized in Table 10.

Table 10. Synthesis of imidazo[1,2-*a*]pyridines



Entry	Aldehyde (18)	Amidine (19)	R ⁵ NC (20)	Product (21) ^a	Time (h)	Yield (%) ^b
1		19a	20a		2	95
2		19a	20a		2	85
3		19a	20a		2	90
4		19b	20a		1.5	96

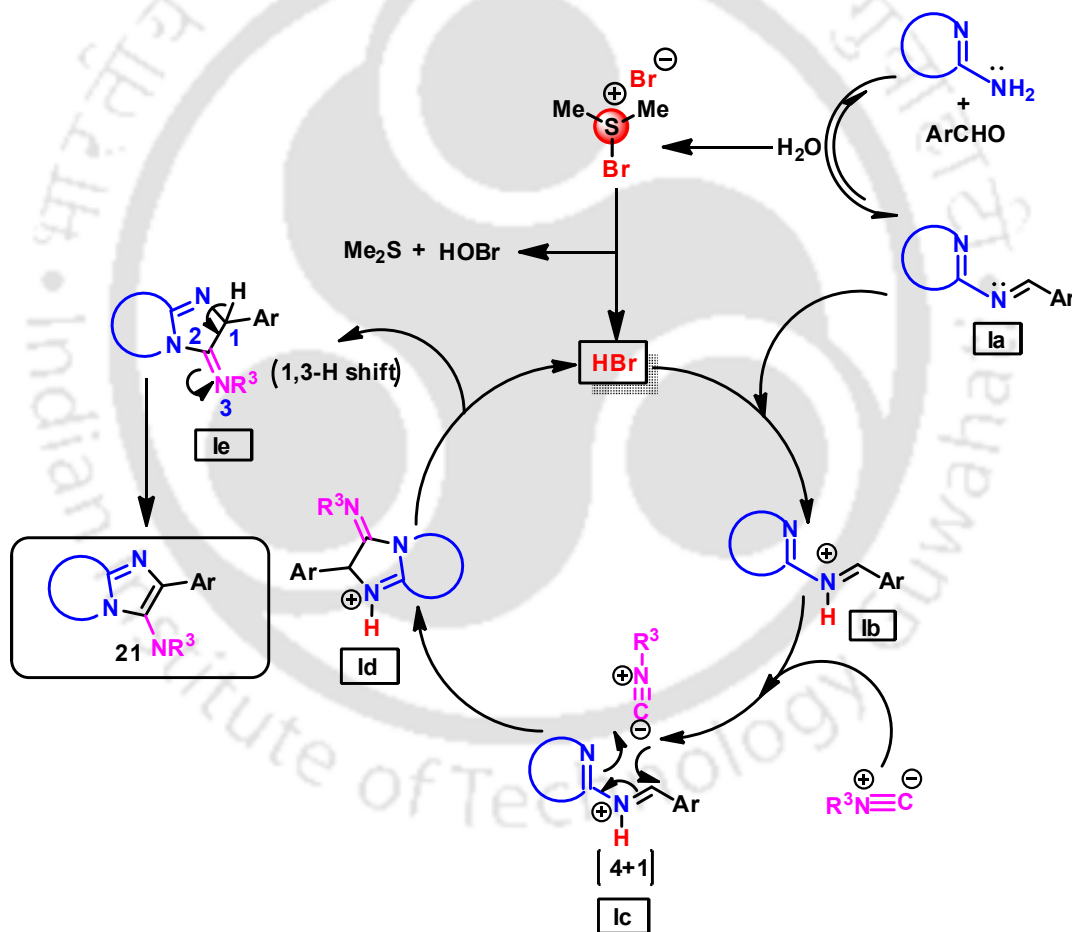
5	 18e	19b	20a	 21e	1	93
6	 18f	19b	20a	 21f	0.5	94
7	 18g	19b	20a	 21g	1	95
8	 18b	19b	20a	 21h	2.5	89
9	 18c	19b	20a	 21i	1	92
10	 18h	19b	20a	 21j	2	85



^aAll the reaction was performed using (1mmol) aldehyde, (1mmol) amidine and (1mmol) isocyanide. ^bIsolated yield.

To compare the efficiency of BDMS with aqueous HBr, a reaction was performed with 4-chlorobenzaldehyde, 2-aminopyridine and *tert*-butyl isocyanide using 48% aqueous HBr (0.1 mL) at room temperature and the time taken for the completion of the reaction was 6 h and the product **21a** obtained with lower yield of 65% as compare to BDMS catalyzed reaction. The comparative analysis of the result shows superiority and efficiency of BDMS for the synthesis of imidazo[1,2-*a*]pyridine derivatives.

All the obtained products were characterized by IR, NMR and elemental analyses. Furthermore, the structure of the molecule **21m** was confirmed through single-XRD data which is shown in Figure 10.



□ **Scheme 71.** Plausible mechanism for the formation of imidazo[1,2-*a*]pyridines

The mechanism of imidazo[1,2-*a*]pyridine can be described as follows: Amidine reacts with aromatic aldehyde to form imine **1a** and water. The liberated water molecule reacts with bromodimethylsulfonium bromide (BDMS) to form dimethyl sulfide, HOBr and

HBr. Then, the released HBr undergoes protonation with imine **1a** to form **1b**, which further reacts with isocyanide via [4+1] cycloaddition which leads to **1d**. Subsequently, 1,3-hydrogen shift from the intermediate **1e** gave the final product of imidazo[1,2-*a*]pyridines **21**, as shown in Scheme 71.

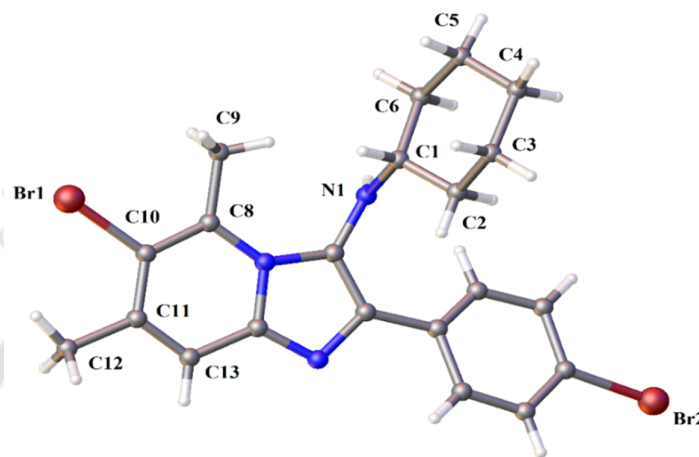


Figure 10. X-ray crystal structure of **21m**

In conclusion, we have explored a mild and facile methodology for the synthesis of imidazo[1,2-*a*]pyridine derivatives by employing aromatic aldehyde, aromatic amidine and isocyanide using BDMS as a catalyst. The merits of the protocol are simple, general, non-aqueous work up, inexpensive, high yield and compatible with wide range of substrates. Furthermore, the synthesized imidazo-heterocyclic scaffold might exist as an interesting pharmacophore in near future.

Experimental Section

Chapter V

*Synthesis of Imidazo[1,2-a]
pyridines in One-pot three
component reaction using
Bromodimethylsulfonium
bromide (BDMS) as a Catalyst*

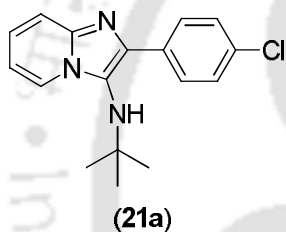
Experimental Section

General procedure for the synthesis of imidazo[1,2-a]pyridine

To a solution of aromatic aldehyde (1 mmol), amidine (1 mmol) and isocyanide (1 mmol) in 2 mL acetonitrile was added 5 mol% bromodimethylsulfonium bromide (0.011 g) at room temperature and allowed to stir for appropriate time as shown in Table 10. After completion of reaction as monitored by TLC, the solid product was filtered off and washed twice with acetonitrile and finally dried under reduced pressure.

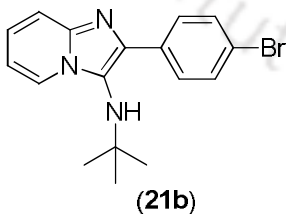
Spectral data of Compounds

N-tert-butyl-2-(4-chlorophenyl)*H*-imidazo[1,2-*a*]pyridin-3-amine (**21a**)

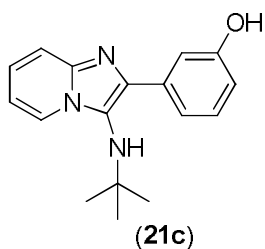


Brown solid, mp 146 °C; **IR** (KBr): 2961, 2925, 2854 cm⁻¹; **¹H NMR** (400 MHz, CDCl₃): δ 8.11 (d, *J* = 6.8 Hz, 1H), 7.87 (d, *J* = 8.4 Hz, 2H), 7.47 (d, *J* = 8.8 Hz, 1H), 7.32 (d, *J* = 8.4 Hz, 2H), 7.07 (t, *J* = 7.2 Hz, 1H), 6.70 (t, *J* = 6.8 Hz, 1H), 3.0 (bs, NH, 1H), 0.99 (s, 9H); **¹³C NMR** (100 MHz, CDCl₃): δ 142.1, 138.4, 133.9, 133.1, 129.4, 128.5, 124.3, 123.6, 123.5, 117.4, 111.5, 56.5, 30.5. **Anal. Calcd** for C₁₇H₁₈ClN₃ (299.79): C, 68.11; H, 6.05; N, 14.02. Found: C, 67.95; H, 5.99; N, 13.91.

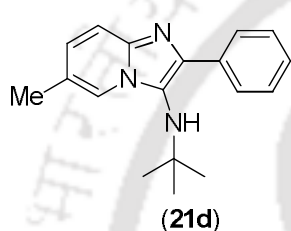
2-(4-bromophenyl)-*N*-tert-butyl*H*-imidazo[1,2-*a*]pyridin-3-amine (**21b**)



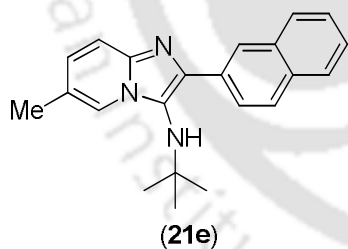
Pale brown solid, mp 146 °C; **IR** (KBr): 3280, 2963, 2923, 1632 cm⁻¹; **¹H NMR** (400 MHz, CDCl₃): δ 8.03 (d, *J* = 6 Hz, 1H), 7.74 (d, *J* = 8.0 Hz, 2H), 7.39-7.41 (m, 3H), 7.0 (t, *J* = 8.8 Hz, 1H), 6.61-6.64 (m, 1H), 2.95 (bs, NH, 1H), 0.91 (s, 9H); **¹³C NMR** (100 MHz, CDCl₃): δ 142.1, 138.3, 134.2, 131.3, 129.7, 124.4, 123.6, 123.4, 121.4, 117.3, 111.5, 56.4, 30.4. **Anal. Calcd** for C₁₇H₁₈BrN₃ (344.24): C, 59.31; H, 5.27; N, 12.21. Found: C, 59.05; H, 5.19; N, 12.11.

3-(3-(*tert*-butylamino)*H*-imidazo[1,2-*a*]pyridin-2-yl)phenol (**21c**)

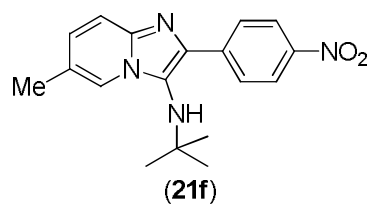
White solid, mp 126 °C; **IR** (KBr): 3429, 2252, 2130, 1638 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 8.23 (dd, $J = 8$ Hz, $J = 6.8$ Hz, 1H), 7.43-7.49 (m, 1H), 7.31-7.36 (m, 2H), 7.09-7.15 (m, 2H), 6.68-6.77 (m, 2H), 4.50 (bs, OH, 1H), 3.60 (bs, NH, 1H), 0.94 (s, 9H); **^{13}C NMR** (100 MHz, CDCl_3): δ 156.1, 140.3, 137.3, 135.0, 128.0, 123.4, 122.9, 122.8, 118.1, 115.4, 114.0, 113.5, 110.3, 55.0, 29.1. **Anal. Calcd** for $\text{C}_{17}\text{H}_{19}\text{N}_3\text{O}$ (281.35): C, 72.57; H, 6.81; N, 14.94. Found: C, 72.30; H, 6.73; N, 14.83.

N-*tert*-butyl-6-methyl-2-phenyl*H*-imidazo[1,2-*a*]pyridin-3-amine (**21d**)

White solid, mp 215-217 °C; **IR** (KBr): 3289, 2968, 2917, 1604 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 7.91 (s, 1H), 7.82 (d, $J = 7.6$ Hz, 2H), 7.31-7.39 (m, 3H), 7.22 (t, $J = 7.2$ Hz, 1H), 6.90 (d, $J = 9.2$ Hz, 1H), 3.0 (bs, NH, 1H), 2.25 (s, 3H), 0.95 (s, 9H); **^{13}C NMR** (100 MHz, CDCl_3): δ 141.3, 139.5, 135.6, 128.3, 128.2, 127.3, 123.4, 121.3, 121.0, 116.8, 56.5, 30.4, 18.5. **Anal. Calcd** for $\text{C}_{18}\text{H}_{21}\text{N}_3$ (279.37): C, 77.38; H, 7.58; N, 15.04. Found: C, 77.13; H, 7.50; N, 14.96.

N-*tert*-butyl-6-methyl-2-(naphthalen-2-yl)*H*-imidazo[1,2-*a*]pyridin-3-amine (**21e**)

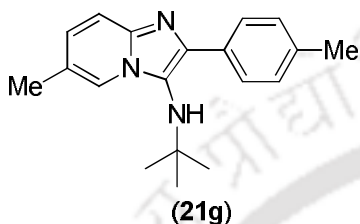
White solid, mp 211 °C; **IR** (KBr): 3252, 2972, 2923, 1627 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 8.40 (s, 1H), 8.32 (dd, $J = 8.4$ Hz, $J = 1.6$ Hz, 1H), 7.90 (s, 1H), 7.83-7.85 (m, 1H), 7.73-7.79 (m, 2H), 7.38-7.42 (m, 3H), 6.90 (dd, $J = 8.8$ Hz, $J = 1.2$ Hz, 1H), 3.23 (bs, NH, 1H), 2.24 (s, 3H), 0.96 (s, 9H); **^{13}C NMR** (100 MHz, CDCl_3): δ 141.1, 138.8, 133.6, 132.8, 132.6, 128.4, 127.7, 127.0, 126.3, 126.1, 125.9, 123.8, 121.2, 116.5, 56.5, 30.5, 18.5. **Anal. Calcd** for $\text{C}_{22}\text{H}_{23}\text{N}_3$ (329.43): C, 80.21; H, 7.04; N, 12.76. Found: C, 79.97; H, 6.97; N, 12.68.

N-*tert*-butyl-6-methyl-2-(4-nitrophenyl)*H*-imidazo[1,2-*a*]pyridin-3-amine (**21f**)

Yellow solid, mp 231-232 °C; **IR** (KBr): 3280, 2969, 1629, 1598 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 8.19-8.24 (m, 4H), 7.93 (s, 1H), 7.45 (d, $J = 8.8$ Hz, 1H),

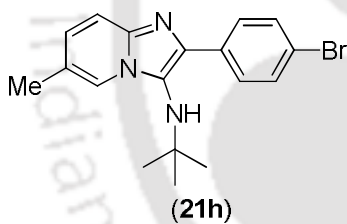
7.04 (dd, $J = 8$ Hz, $J = 1.2$ Hz, 1H), 3.15 (bs, 1H, NH), 2.34 (s, 3H), 1.06 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3): δ 146.7, 141.7, 141.5, 136.5, 128.8, 128.4, 124.8, 123.6, 122.2, 121.2, 117.0, 57.0, 30.6, 18.6. **Anal. Calcd** for $\text{C}_{18}\text{H}_{20}\text{N}_4\text{O}_2$ (324.37): C, 66.65; H, 6.21; N, 17.27. Found: C, 66.46; H, 6.13; N, 17.15.

N-tert-butyl-6-methyl-2-*p*-tolylH-imidazo[1,2-*a*]pyridin-3-amine (**21g**)



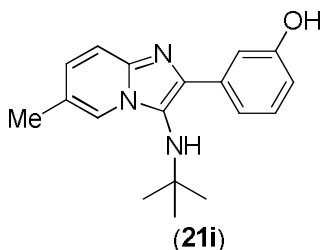
White solid, mp 209-211 °C; **IR** (KBr): 3283, 2967, 2921, 1639, 1557 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.90 (s, 1H), 7.72 (d, $J = 7.2$ Hz, 2H), 7.36 (d, $J = 9.2$ Hz, 1H), 7.14 (d, $J = 7.2$ Hz, 2H), 6.88 (d, $J = 9.2$ Hz, 1H), 3.04 (bs, NH, 1H), 2.30 (s, 3H), 2.24 (s, 3H), 0.96 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3): δ 141.1, 139.4, 137.0, 132.5, 129.0, 128.0, 127.1, 123.1, 121.2, 120.8, 116.6, 56.5, 30.4, 21.4, 18.5. **Anal. Calcd** for $\text{C}_{19}\text{H}_{23}\text{N}_3$ (293.40): C, 77.78; H, 7.90; N, 14.32. Found: C, 77.58; H, 7.82; N, 14.20.

2-(4-bromophenyl)-*N*-tert-butyl-6-methylH-imidazo[1,2-*a*]pyridin-3-amine (**21h**)

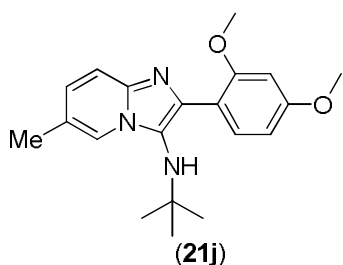


Pale yellow solid, mp 203 °C; **IR** (KBr): 2961, 2925, 2851, 1671 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.95 (s, 1H), 7.84 (dd, $J = 8.4$ Hz, $J = 6.4$ Hz, 2H), 7.54 (d, $J = 8.4$ Hz, 2H), 7.44 (d, $J = 8.8$ Hz, 1H), 6.99 (dd, $J = 9.2$ Hz, $J = 1.6$ Hz, 1H), 2.34 (s, 3H), 1.22 (bs, NH, 1H), 1.05 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3): 139.7, 136.4, 132.5, 130.3, 128.6, 127.1, 122.4, 120.6, 120.4, 120.1, 115.2, 55.4, 29.4, 17.4. **Anal. Calcd** for $\text{C}_{18}\text{H}_{20}\text{BrN}_3$ (358.27): C, 60.34; H, 5.63; N, 11.73. Found: C, 60.11; H, 5.56; N, 11.62.

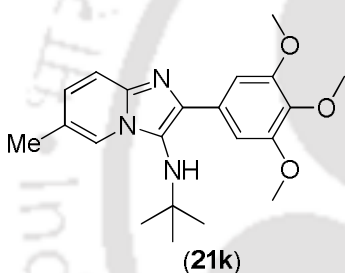
3-(3-(tert-butylamino)-6-methylH-imidazo[1,2-*a*]pyridin-2-yl)phenol (**21i**)



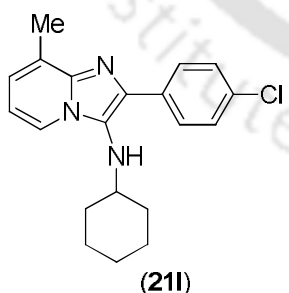
White solid, mp 293 °C; **IR** (KBr): 3445, 2968, 2930, 1596 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.68 (s, 1H), 7.0-7.05 (m, 3H), 6.81 (t, $J = 8$ Hz, 1H), 6.64 (d, $J = 8.8$ Hz, 1H), 6.37 (dd, $J = 2.4$ Hz, $J = 2.4$ Hz, 1H), 3.2 (bs, 1H, NH), 1.95 (s, 3H), 0.64 (s, 9H); ^{13}C NMR (100 MHz, CDCl_3): δ 156.5, 139.6, 137.4, 135.1, 128.5, 127.3, 123.0, 120.8, 120.6, 118.5, 115.1, 114.5, 114.1, 55.6, 29.6, 17.7. **Anal. Calcd** for $\text{C}_{18}\text{H}_{21}\text{N}_3\text{O}$ (295.37): C, 73.19; H, 7.17; N, 14.23. Found: C, 73.09; H, 7.10; N, 14.13.

N-*tert*-butyl-2-(2,4-dimethoxyphenyl)-6-methyl*H*-imidazo[1,2-*a*]pyridin-3-amine (**21j**)

Yellow semisolid; **IR** (KBr): 2963, 2928, 1611, 1577 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 8.11 (s, 1H), 7.77 (d, $J = 8.8$ Hz, 1H), 7.69 (d, $J = 9.2$ Hz, 1H), 7.11 (d, $J = 9.2$ Hz, 1H), 6.66 (dd, $J = 8.8$ Hz, $J = 2.8$ Hz, 1H), 6.52 (d, $J = 2$ Hz, 1H), 3.84 (d, $J = 7.2$ Hz, 6H), 2.35 (s, 3H), 1.22 (bs, NH, 1H), 0.93 (s, 9H); **^{13}C NMR** (100 MHz, CDCl_3): δ 160.3, 156.6, 141.0, 135.7, 131.9, 126.1, 124.8, 120.7, 120.0, 117.6, 115.9, 105.4, 98.6, 55.4, 54.9, 29.6, 29.3, 18.0. **Anal. Calcd** for $\text{C}_{20}\text{H}_{25}\text{N}_3\text{O}_2$ (339.43): C, 70.77; H, 7.42; N, 12.38. Found: C, 70.50; H, 7.35; N, 12.28.

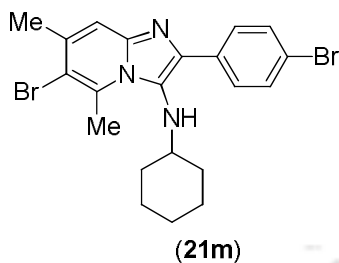
N-*tert*-butyl-6-methyl-2-(3,4,5-trimethoxyphenyl)*H*-imidazo[1,2-*a*]pyridin-3-amine (**21k**)

Yellow semisolid; **IR** (KBr): 2961, 2927, 1586 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 7.96 (s, 1H), 7.46 (d, $J = 8.8$ Hz, 2H), 7.0 (d, $J = 9.2$ Hz, 2H), 3.94 (s, 6H), 3.88 (s, 3H), 2.34 (s, 3H), 1.25 (bs, NH, 1H), 1.09 (s, 9H); **^{13}C NMR** (100 MHz, CDCl_3): δ 153.2, 140.6, 138.4, 137.8, 130.2, 128.0, 123.1, 121.6, 121.1, 116.2, 105.6, 61.0, 56.5, 30.6, 29.8, 18.5. **Anal. Calcd** for $\text{C}_{21}\text{H}_{27}\text{N}_3\text{O}_3$ (369.45): C, 68.27; H, 7.37; N, 11.37. Found: C, 68.10; H, 7.28; N, 11.26.

2-(4-chlorophenyl)-*N*-cyclohexyl-8-methyl*H*-imidazo[1,2-*a*]pyridin-3-amine (**21l**)

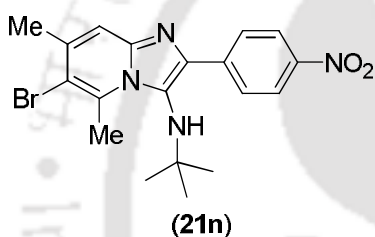
Pale brown solid, mp 108 $^{\circ}\text{C}$; **IR** (KBr): 2923, 2851, 1634 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 8.01 (dd, $J = 6.8$ Hz, $J = 1.6$ Hz, 2H), 7.94 (d, $J = 6.8$ Hz, 1H), 7.40 (dd, $J = 6.8$ Hz, $J = 2.0$ Hz, 2H), 6.92-6.94 (m, 1H), 6.70 (t, $J = 6.8$ Hz, 1H), 3.0 (bs, NH, 1H), 2.90-2.92 (m, 1H), 2.61 (s, 3H), 1.57-1.87 (m, 5H), 1.12-1.25 (m, 5H); **^{13}C NMR** (100 MHz, CDCl_3): δ 142.1, 135.4, 133.5, 132.9, 128.7, 128.6, 127.4, 125.4, 123.0, 120.7, 111.9, 57.0, 34.3, 25.8, 25.0, 16.7. **Anal. Calcd** for $\text{C}_{20}\text{H}_{22}\text{ClN}_3$ (339.86): C, 70.68; H, 6.52; N, 12.36. Found: C, 70.42; H, 6.44; N, 12.29.

6-bromo-2-(4-bromophenyl)-*N*-cyclohexyl-5,7-dimethyl-*H*-imidazo[1,2-*a*]pyridin-3-amine (**21m**)



White solid, mp 191 °C; **IR** (KBr): 2928, 2851, 1633 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 7.77 (d, $J = 8.8$ Hz, 3H), 7.52 (d, $J = 8.4$ Hz, 2H), 3.45 (s, 3H), 3.14 (s, 3H), 2.68 (bs, NH, 1H), 2.37-2.44 (m, 1H), 1.01-1.72 (m, 10H); **^{13}C NMR** (100 MHz, CDCl_3): δ 142.3, 138.4, 135.5, 134.8, 133.7, 131.5, 129.3, 127.0, 121.5, 114.8, 114.7, 58.6, 33.2, 25.8, 24.9, 24.1, 18.6. **Anal. Calcd** for $\text{C}_{21}\text{H}_{23}\text{Br}_2\text{N}_3$ (477.23): C, 52.85; H, 4.86; N, 8.80. Found: C, 52.68; H, 4.79; N, 8.70.

6-brom-*N*-*tert*-butyl-5,7-dimethyl-2-(4-nitrophenyl)-*H*-imidazo[1,2-*a*]pyridin-3-amine (**21n**)



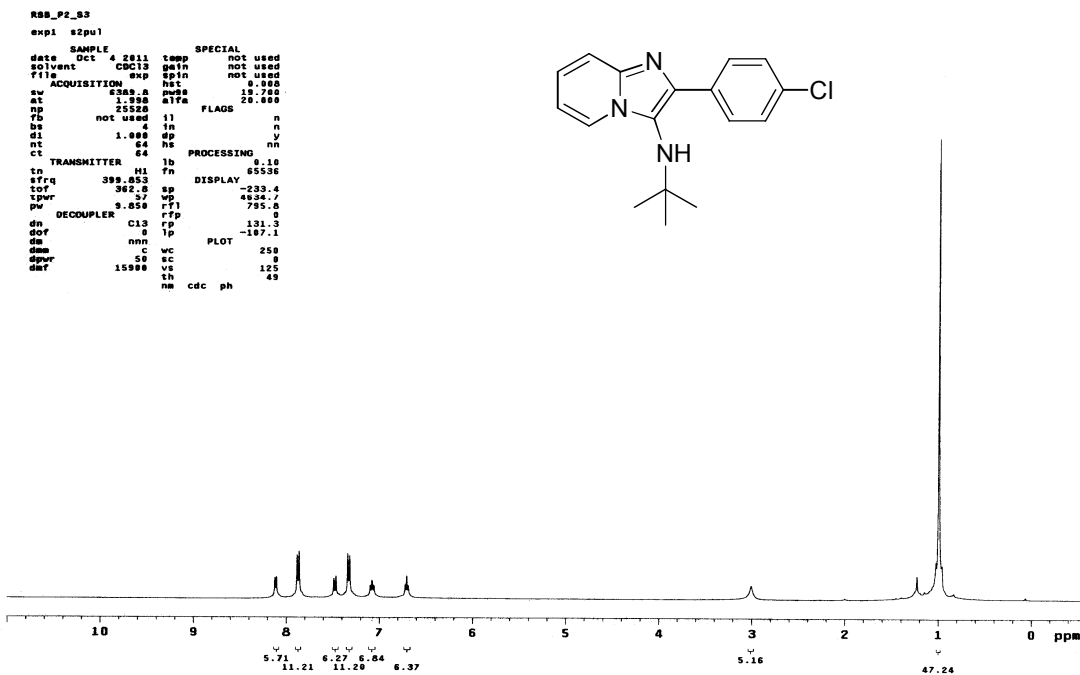
Yellow solid, mp 204-206 °C; **IR** (KBr): 3307, 2963, 1632, 1598 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 8.22 (d, $J = 8.4$ Hz, 2H), 7.99 (d, $J = 8.8$ Hz, 3H), 3.18 (bs, NH, 1H), 3.15 (s, 3H), 2.42 (s, 3H), 0.84 (s, 9H); **^{13}C NMR** (100 MHz, CDCl_3): δ 147.0, 142.8, 142.2, 139.8, 136.7, 135.6, 129.2, 127.3, 123.7, 115.8, 115.0, 57.0, 29.7, 24.1, 19.6. **Anal. Calcd** for $\text{C}_{19}\text{H}_{21}\text{BrN}_4\text{O}_2$ (417.29): C, 54.69; H, 5.07; N, 13.43. Found: C, 54.45; H, 4.99; N, 13.35.

Complete crystallographic data of **21m** for the structural analyses have been deposited with the Cambridge Crystallographic Data Centre, CCDC No. 848594. Copies of this information may be obtained free of charge from the Director, Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK, (fax: +44-1223-336033, e-mail: deposit@ccdc.cam.ac.uk or via: www.ccdc.cam.ac.uk).

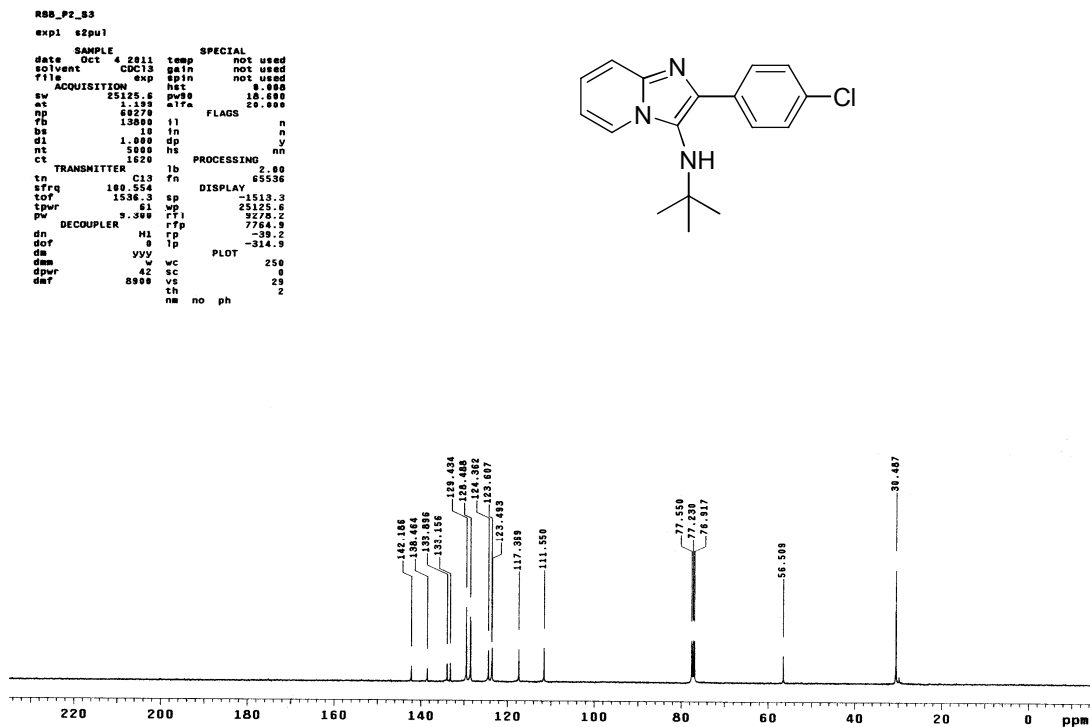
Table 11. Crystal data and structures refinement for the compound **21m**, for atomic coordinates, equivalent isotropic displacement parameters and bond angles, please check the CIF.

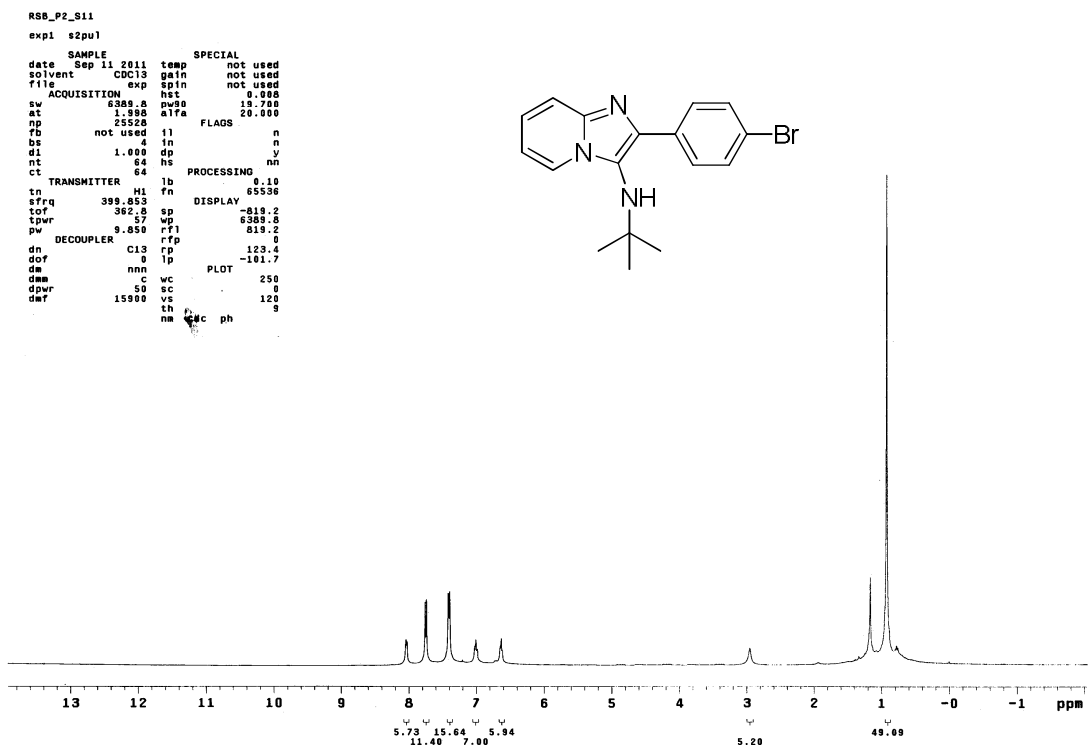
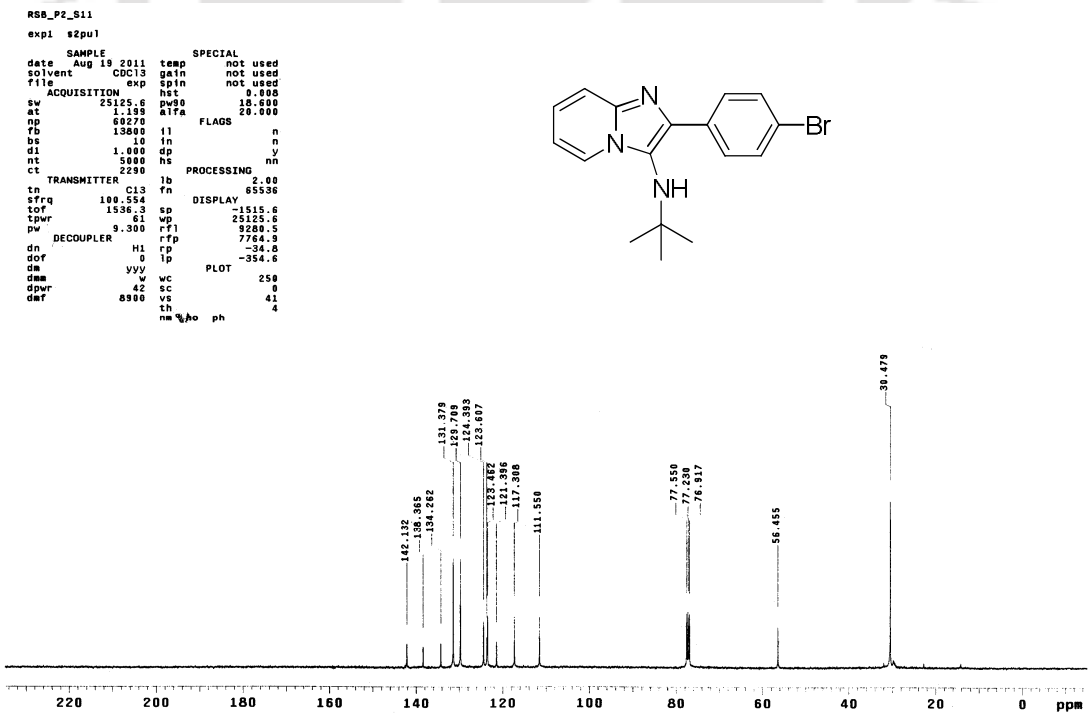
Parameters	Compound 21m
Empirical Formula	'C ₂₁ H ₂₃ Br ₂ N ₃ '
Formula weight	477.22
Temperature	296 K
CCDC no	848594
Wavelength (Å)	0.71073 Å
Crystal system	Monoclinic
Space group	P 21/c
Radiation type	Mo K α
Radiation source	'fine-focus sealed tube'
<i>a</i> (Å)	6.3991 (5) Å
<i>b</i> (Å)	32.852 (3) Å
<i>c</i> (Å)	9.1812 (7) Å
α (°)	90.00°
β (°)	90.00°
γ (°)	90.00°
Cell Volume	1930.1 (3) Å ³
<i>Z</i>	4
Density	1.639 g/cm ³
<i>F</i> (0 0 0)	956
Index ranges	-6 \leq h \leq 7, -39 \leq k \leq 39, -11 \leq l \leq 10
Number of unique reflections	3480
Number of parameters	237
Number of restraints	0
Goodness-of-fit (GOF) ^a on <i>F</i> ²	4.788
Refinement method	Full-matrix least-squares on <i>F</i> ²

^1H NMR (400 MHz, CDCl_3): *N*-*tert*-butyl-2-(4-chlorophenyl)*H*-imidazo[1,2-*a*]pyridin-3-amine (21a)

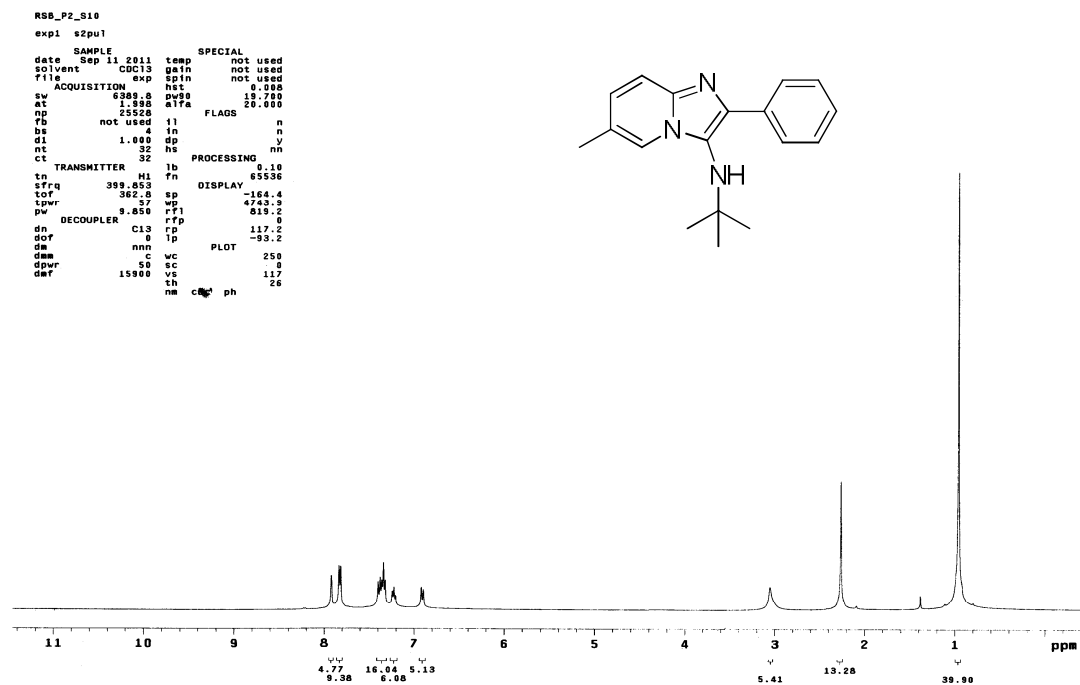


^{13}C NMR (100 MHz, CDCl_3): *N*-*tert*-butyl-2-(4-chlorophenyl)*H*-imidazo[1,2-*a*]pyridin-3-amine (21a)

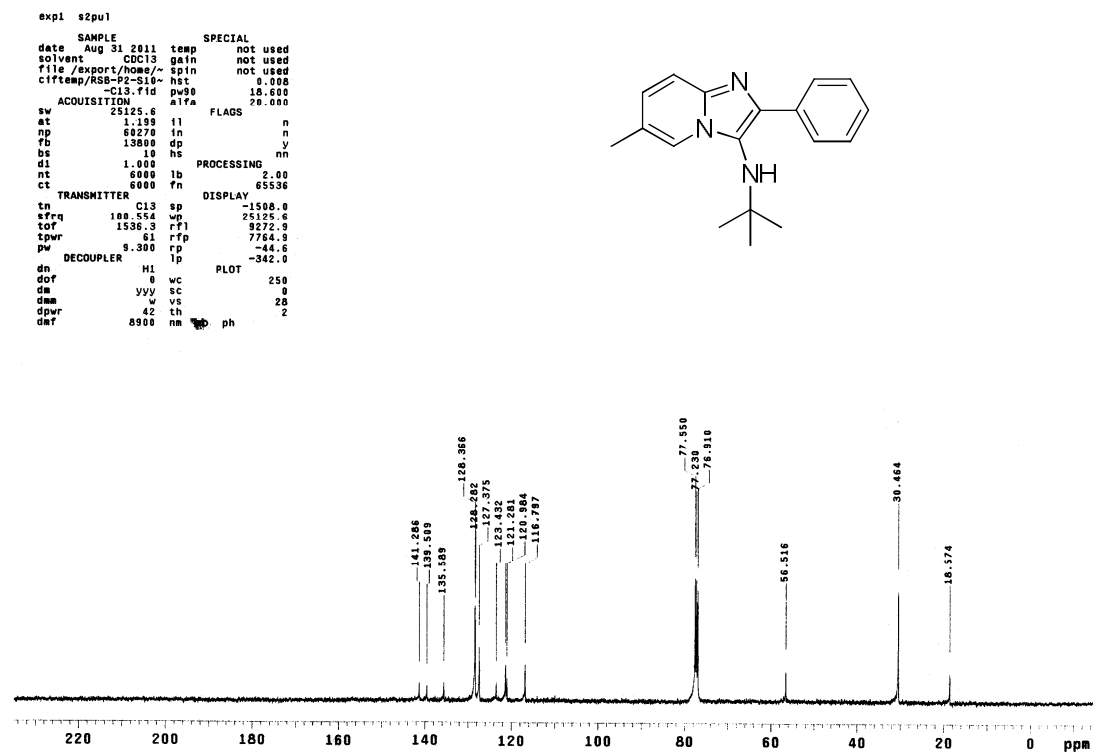


¹H NMR (400 MHz, CDCl₃): 2-(4-bromophenyl)-N-tert-butylimidazo[1,2-a]pyridin-3-amine (21b)**¹³C NMR (100 MHz, CDCl₃): 2-(4-bromophenyl)-N-tert-butylimidazo[1,2-a]pyridin-3-amine (21b)**

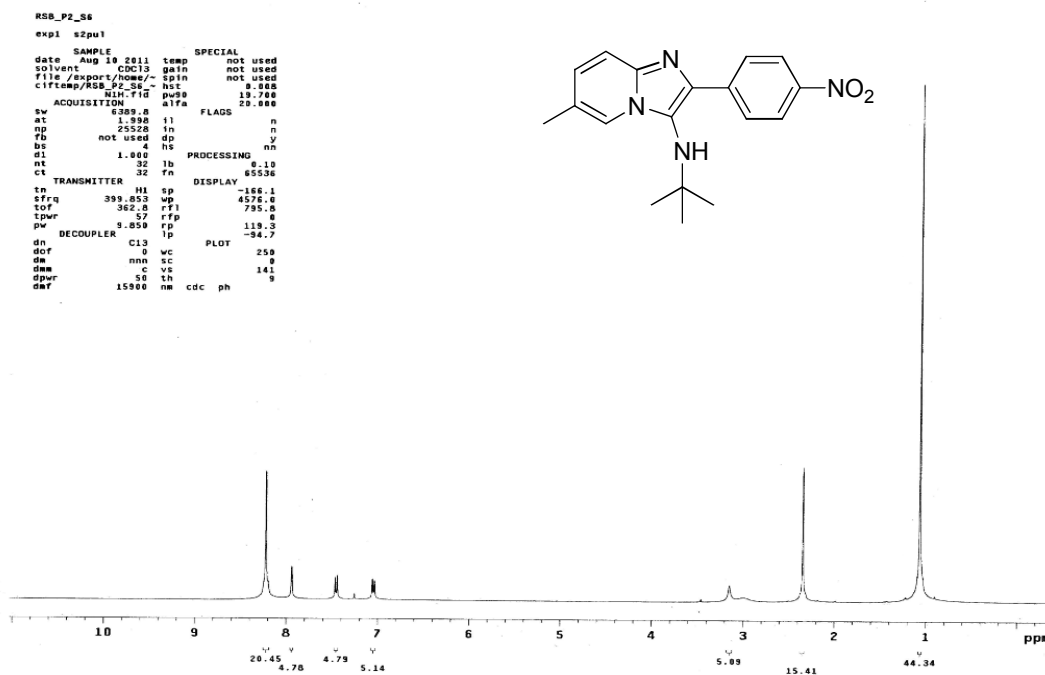
^1H NMR (400 MHz, CDCl_3): *N*-*tert*-butyl-6-methyl-2-phenylH-imidazo[1,2-*a*]pyridin-3-amine (21d)



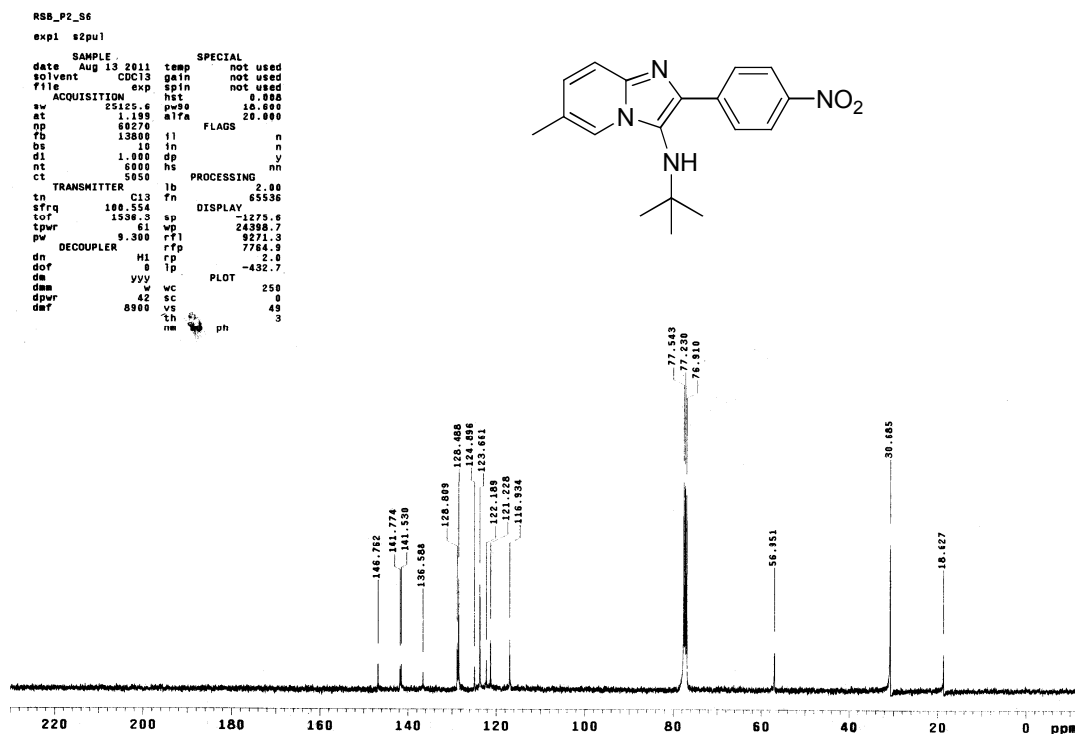
^{13}C NMR (100 MHz, CDCl_3): *N*-*tert*-butyl-6-methyl-2-phenylH-imidazo[1,2-*a*]pyridin-3-amine (21d)



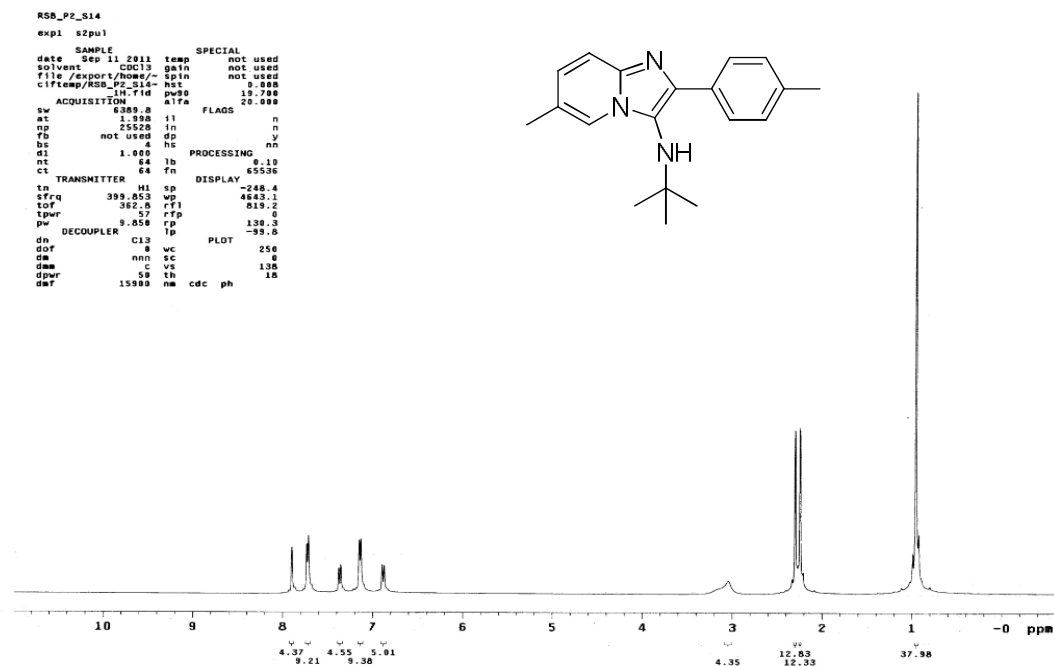
¹H NMR (400 MHz, CDCl₃): *N*-*tert*-butyl-6-methyl-2-(4-nitrophenyl)H-imidazo[1,2-*a*]pyridin-3-amine (21f)



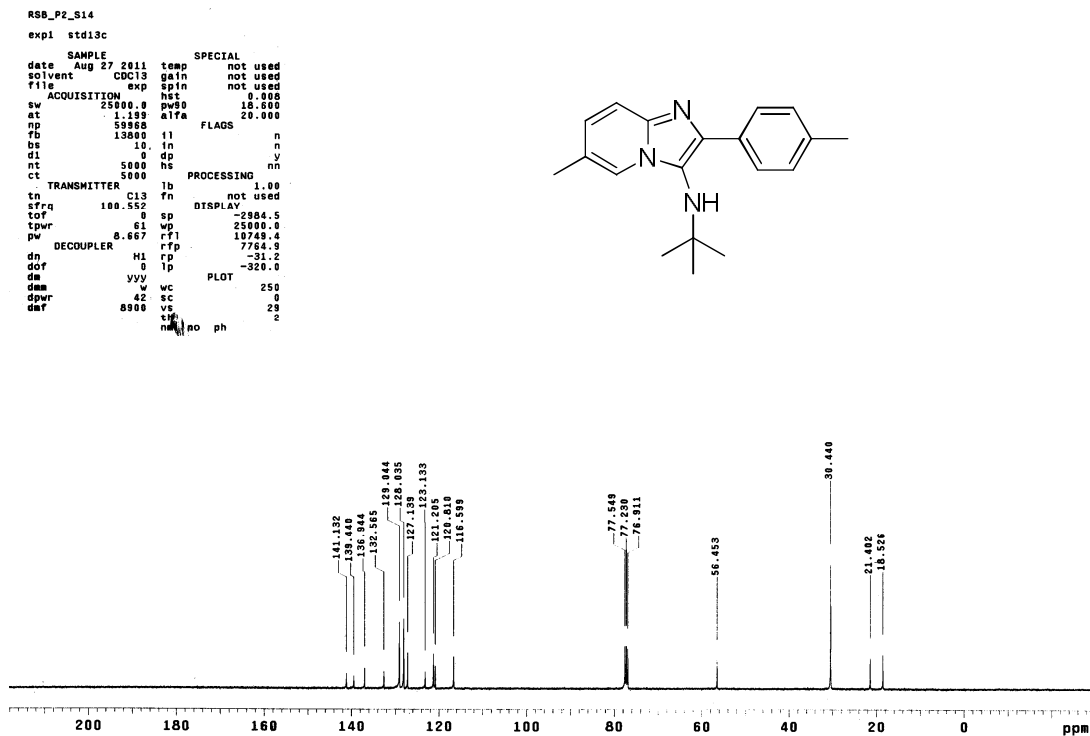
¹³C NMR (100 MHz, CDCl₃): *N*-*tert*-butyl-6-methyl-2-(4-nitrophenyl)H-imidazo[1,2-*a*]pyridin-3-amine (21f)



^1H NMR (400 MHz, CDCl_3): *N*-*tert*-butyl-6-methyl-2-*p*-tolylH-imidazo[1,2-*a*]pyridin-3-amine (21g)



^{13}C NMR (100 MHz, CDCl_3): *N*-*tert*-butyl-6-methyl-2-*p*-tolylH-imidazo[1,2-*a*]pyridin-3-amine (21g)



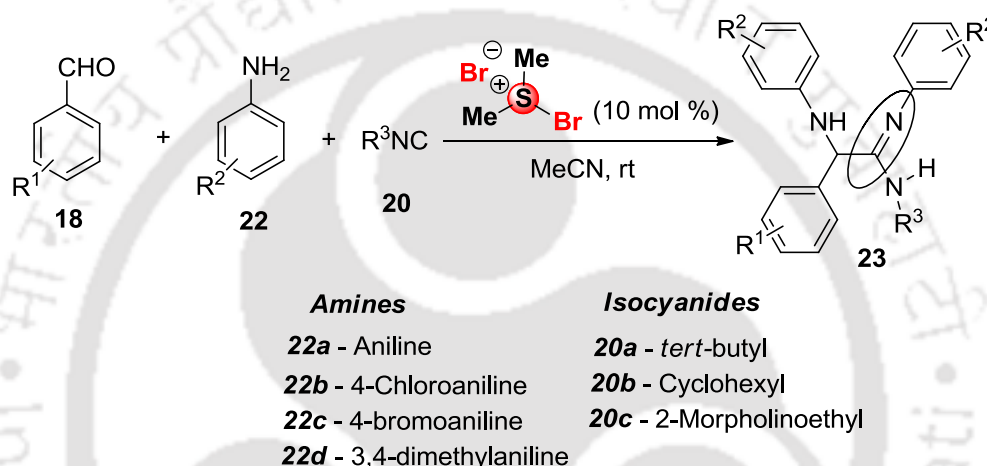
Chapter VI

AB²C type UGI reaction for the Synthesis of α -amino Amidines in presence of Bromodimethylsulfonium bromide (BDMS)

Results and Discussion

● Results and Discussion

Brief literature survey of aminoamidine importance and its synthetic paths have been discussed in Chapter IV. To our delight, continuous effort on exploration of BDMS in organic synthesis as a catalyst which access new methodologies to various heterocyclic scaffolds. In this Chapter a new AB²C Ugi reaction is demonstrated for the synthesis of α -amino amidines using aromatic aldehydes, aromatic amines and isocyanides in one-pot three-component reaction catalyzed by bromodimethylsulfonium bromide (BDMS), which is shown in Scheme 72.



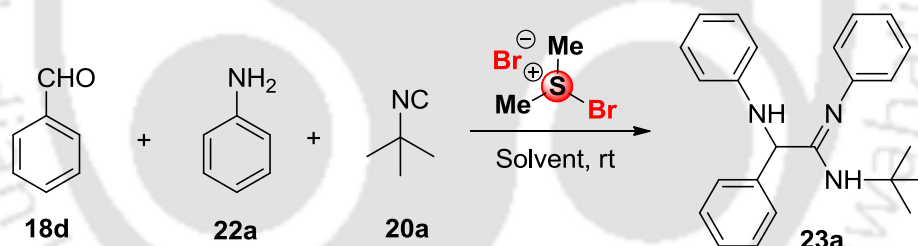
□ **Scheme 72.** Synthesis of α -amino amidine using three-component reaction

Bromodimethylsulfonium bromide (BDMS) is found to be a cheap and readily available catalyst used for various organic transformations. Our initial effort began from the preparation of the catalyst BDMS by known literature procedure followed by optimization of the reaction condition by using benzaldehyde (**18d**), aniline (**22a**) and *tert*-butyl isocyanide (**20a**) as a model substrate. Various reactions were scrutinized with different mol % of the catalyst and with different solvent such as CH₂Cl₂, CH₃CN and MeOH. Initial attempts were carried out on reaction with 2 mol %, 4 mol %, and 5 mol % of BDMS which provided 57%, 61% and 68% yield, respectively (Table 12, entries 4-6). Meanwhile, in case of 10 mol% of BDMS, the reaction underwent smoothly and it also enhances the yield to 92% (Table 12, entry 7). We also observe that the reaction did not take place in absence of catalyst. From this point, it was clear that BDMS plays a major role for the synthesis of α -amino amidine. However, it was noted that excess loading of the catalyst BDMS decreased the yield (Table 12, entry 8). The output of the

optimized results indicates that 10 mol% of BDMS is sufficient for the synthesis of α -amino amidine.

We sought to study the solvent role by executing the same reaction with different solvent such as dichloromethane, acetonitrile and methanol. However, acetonitrile was found to be the best solvent for this synthesis. Eventually, the isolated product **23a** was characterized by ^1H NMR which indicated a singlet for tertiary butyl proton at 1.46, $-\text{CH}$ proton as singlet at 4.92, $-\text{NH}$ proton as two broad peak at 3.71, 5.90 and aromatic proton at 6.48-7.30 ppm respectively. On the other hand, the peak at 154 ppm in ^{13}C NMR corresponds to $\text{C}=\text{N}$ of α -amino amidine **23a**. Moreover, the structure of α -amino amidine derivative **23a** was confirmed through single-X-ray crystallographic data as shown in Figure 11. Each unit cell contained four molecule of **23a** embedded in the crystal lattice in $P2_1/c$ space group. The analysis of the crystal structure specify the position of the $\text{C}=\text{N}$ ($\text{C7}-\text{N1} = 1.27 \text{ \AA}$) and the protonated nitrogen attached to the alkyl group ($\text{C7}-\text{N3} = 1.36 \text{ \AA}$).

Table 12. Optimization of reaction condition for the synthesis of α -amino amidine **23a**^a



Entry	BDMS used	Solvent	Time (h)	Yield (%) ^b
1	-	CH_2Cl_2	22	00
2	2 mol%	CH_2Cl_2	7	52
3	2 mol%	CH_3OH	6	45
4	2 mol%	CH_3CN	5	57
5	4 mol%	CH_3CN	4	61
6	5 mol%	CH_3CN	3	68
7	10 mol%	CH_3CN	1	92
8	15 mol%	CH_3CN	2	91

^aThe reactions were performed using 1 mmol of benzaldehyde (**18d**), 2 mmol of aniline (**22a**), 1 mmol of *tert*-butyl isocyanide (**20a**) in 5 mL of solvent at room temperature.

^bIsolated yield.

The optimized protocol was further explored with various aromatic aldehydes, substituted aromatic amines and isocyanides which afford the resulting product of α -amino amidine derivatives in good yields as depicted in Table 13. We notice that **23q** offer a low yield of the product (Table 13, entry 17) due to less reactivity of 2-morpholinoethyl isocyanide. The respective products of all α -amino amidine derivatives were characterized by $^1\text{H-NMR}$, $^{13}\text{C-NMR}$ and elemental analysis. To our delight, next the reaction were performed with benzaldehyde (**18d**), aniline (**22a**) and *tert*-butyl isocyanide (**20a**) using 48% aqueous HBr at room temperature in order to compare the effectiveness of BDMS. However it resulted in a low yield of **23a** with 40% after 4h.

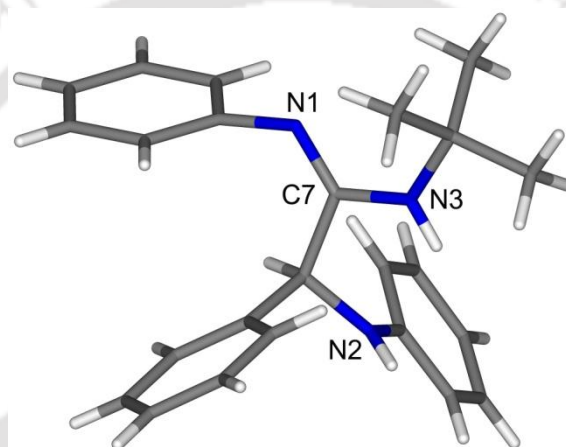
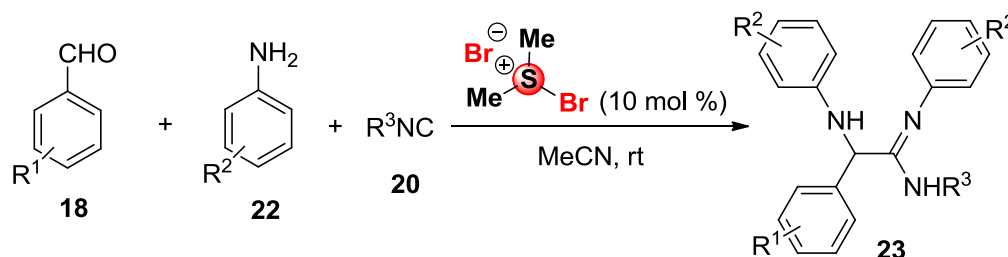
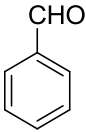
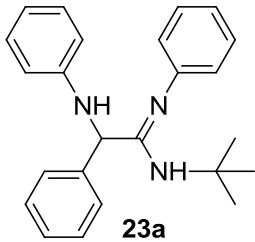
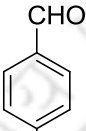
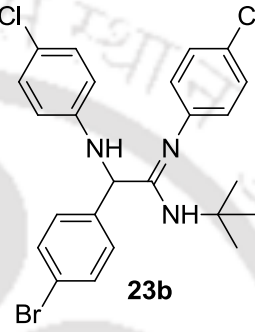
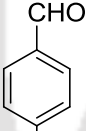
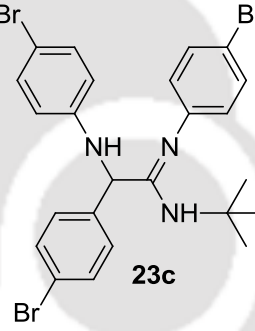
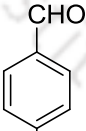
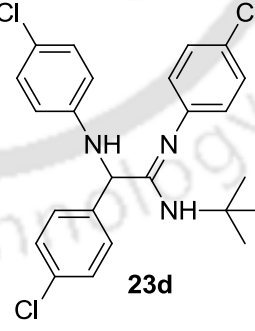
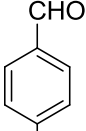
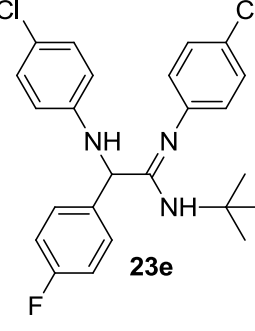


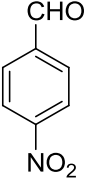
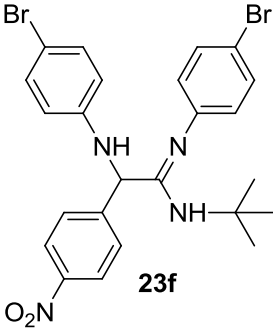
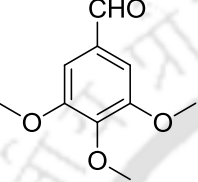
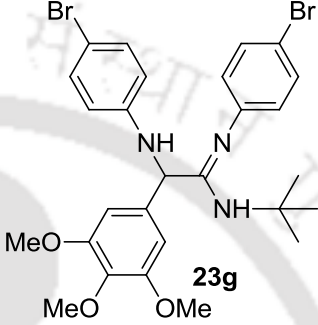
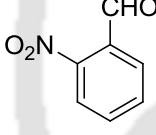
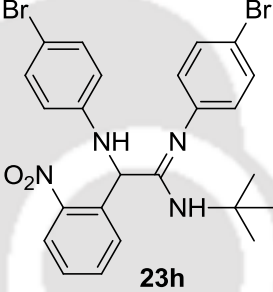
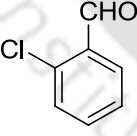
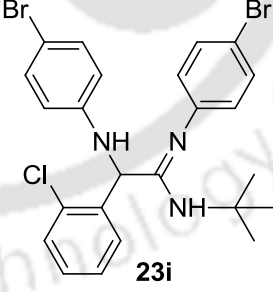
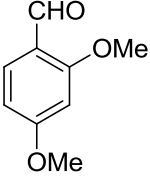
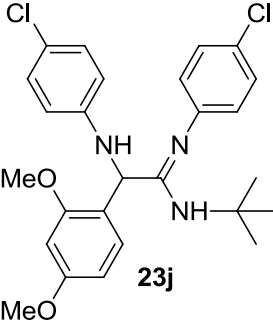
Figure 11. Crystal structures of **23a**

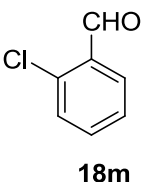
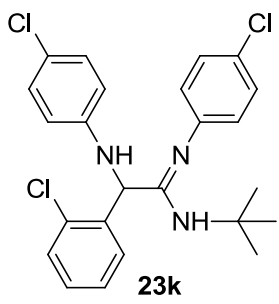
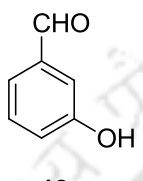
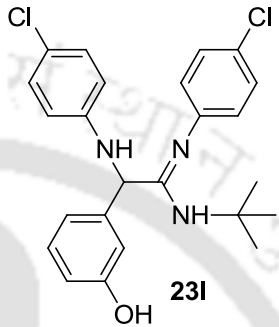
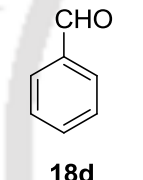
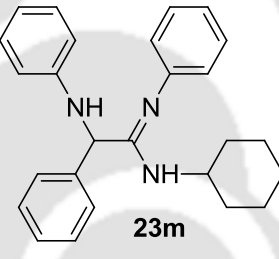
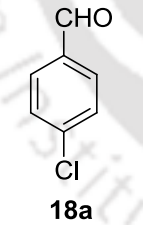
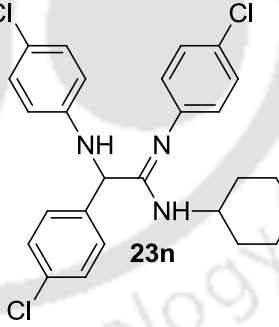
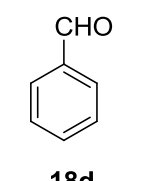
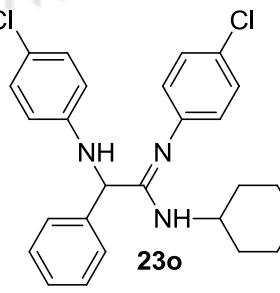
In addition, the structure of the molecule **23b** and **23n** was also determined by single-crystal X-ray crystallography data. The compounds **23b** and **23n** are isostructural to **23a** and the protonated nitrogen attached to the alkyl group (C25–N1 = 1.35 Å in **23b** and C14–N3 = 1.34 Å in **23n**) is shown in Figure 12a and 12b.

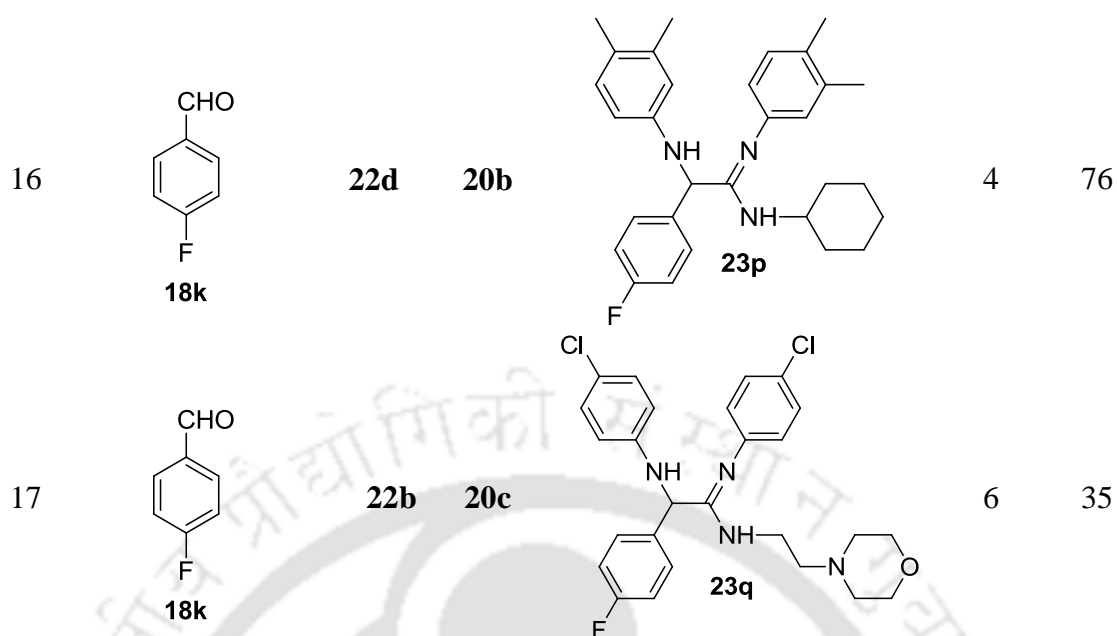
Table 13. BDMS-catalyzed synthesis of α -amino amidines^a



S.No	Aldehyde (18)	Amine (22)	R ³ NC (20)	Product (23)	Time (h)	Yield (%) ^b
1	 18d	22a	20a	 23a	1	92
2	 18b	22b	20a	 23b	1	85
3	 18b	22c	20a	 23c	2.5	77
4	 18a	22b	20a	 23d	0.5	93
5	 18k	22b	20a	 23e	1.5	89

6	 18f	22c	20a	 23f	2.5	76
7	 18i	22c	20a	 23g	3	82
8	 18l	22c	20a	 23h	2	75
9	 18m	22c	20a	 23i	3	65
10	 18h	22b	20a	 23j	3.5	78

11	 18m	22b	20a	 23k	1	87
12	 18c	22b	20a	 23l	2	82
13	 18d	22a	20b	 23m	1.5	85
14	 18a	22b	20b	 23n	0.5	90
15	 18d	22b	20b	 23o	3.5	80



^aAll the reactions were carried out using 1 mmol of aldehyde, 2 mmol of amine and 1 mmol of isocyanide in 5 mL acetonitrile at room temperature. ^bIsolated yield.

Interestingly, the inter-molecular hydrogen bonding interactions in **23n** through N–H···N bonds ($H\cdots N = 2.24 \text{ \AA}$, $N\cdots N = 3.074 \text{ \AA}$, $\angle N-H\cdots N = 164^\circ$) propagated along *b*-axis which exhibited a one-dimension zigzag structure as depicted in Figure 13.

The plausible mechanism of α -amino amidine can be elucidated as follows: In the first step the aromatic amine reacts with aromatic aldehyde to form imine **1a** and water. The liberated water molecule reacts with BDMS which leads to the formation of dimethyl sulfide, HOBr and HBr. Then the imine **1a** undergoes protonation with the liberated HBr to form **1b**, which reacts with isocyanide to form intermediate **1c**.

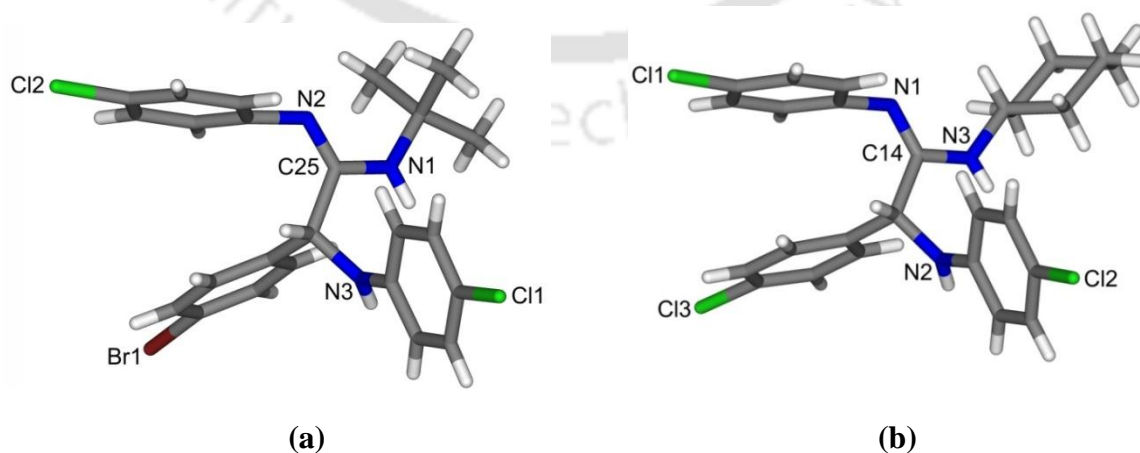


Figure 12. Molecular structure of (a) **23b** and (b) **23n**

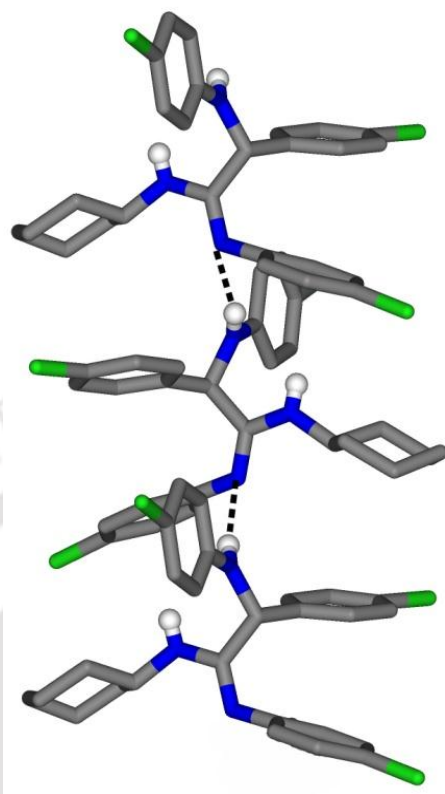
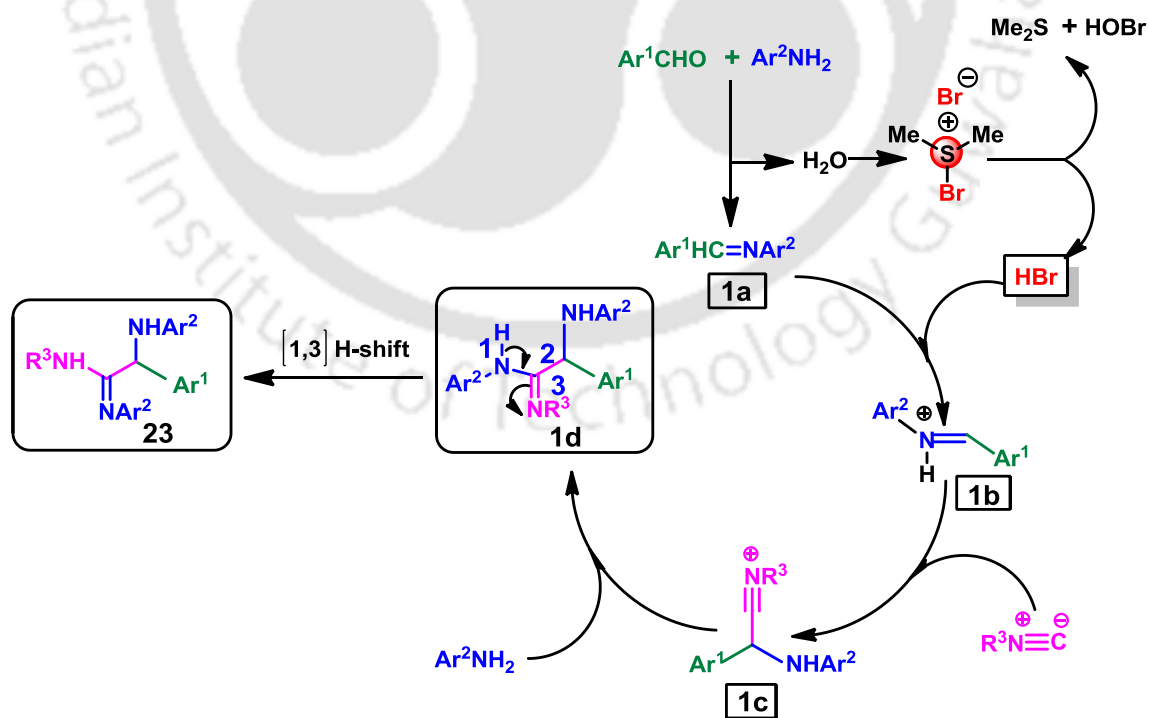


Figure 13. Zigzag hydrogen bonded chain of **23n** propagated along *b*-axis



Scheme 73. Plausible Mechanism for the formation of α -aminoamidine

The intermediate **1c** further reacts with another molecule of amine to give α -amino amidines **1d** which undergoes [1,3] hydrogen shift to afford the product of α -amino amidines **23** as shown in Scheme 73.

In conclusion, we have developed a new one-pot three-component AB²C type UGI reaction for the synthesis of α -amino amidines using aromatic aldehydes, aromatic amines and isocyanides in presence of bromodimethylsulfonium bromide (BDMS) as a catalyst. The important aspects of the present protocol are metal-free catalyst, good yield, shorter reaction time, and variety of substrate scope compatibility, mild reaction conditions, easy separation and purification process. In addition to that, the molecule **23n** forms fascinating one-dimension zigzag structure via inter-molecular hydrogen bonding interactions.



Experimental Section

*AB²C type UGI reaction for the
Synthesis of α -amino Amidines in
presence of Bromodimethylsulfonium
bromide (BDMS)*

Chapter VI

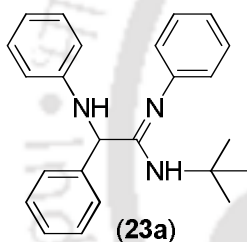
Experimental Section

General procedure for the synthesis of α -amino amidine

To a stirred mixture of aromatic aldehyde (1 mmol), aromatic amine (2 mmol) and isocyanide (1 mmol) in 5 mL acetonitrile, a catalytic amount of bromodimethyl sulfonium bromide (0.022 g) was added at room temperature and allowed to stir till the completion of reaction as indicated by TLC. Then, the reaction mixture was concentrate in a rotary evaporator and the obtained residue was treated with 5 mL methanol, a crystalline product separate out, which was filtered off and washed with cold methanol, diethyl ether and finally dried under reduced pressure.

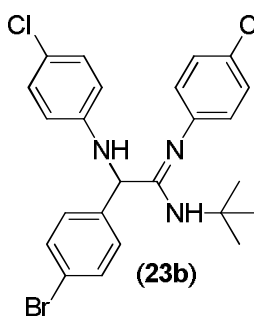
Spectral data of Compounds

N-tert-Butyl-*N'*,2-diphenyl-2-(phenyl amino)acetamidine (**23a**)



White solid, mp 134 °C; **IR** (KBr): 3389, 3027, 2963, 2906, 1634, 1601, 1591, 1485, 1310, 1253, 1220, 1185, 1166, 1070 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 7.30-7.22 (m, 5H), 7.14 (dd, $J = 8$ Hz, $J = 2$ Hz, 2H), 7.03 (t, $J = 8$ Hz, 2H), 6.86-6.81 (m, 2H), 6.70 (d, $J = 7.6$ Hz, 2H), 6.48 (d, $J = 7.2$ Hz, 2H), 5.90 (bs, 1H, NH), 4.92 (s, 1H), 3.71 (bs, 1H, NH), 1.46 (s, 9H); **^{13}C NMR** (100 MHz, CDCl_3): δ 154.0, 150.7, 147.3, 140.0, 129.4, 128.9, 128.6, 128.4, 128.2, 122.4, 121.3, 119.5, 113.9, 60.0, 51.0, 28.5. **Anal. Calcd** for $\text{C}_{24}\text{H}_{27}\text{N}_3$ (357.49): C, 80.63; H, 7.61; N, 11.75. Found: C, 80.54; H, 7.53; N, 11.66.

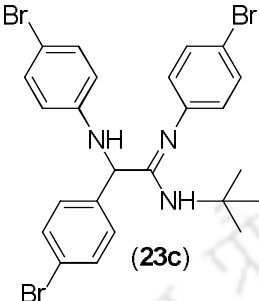
2-(4-Chlorophenylamino)-*N*-tert-butyl-2-(4-bromophenyl)-*N'*-(4-chlorophenyl)acetamidine (**23b**)



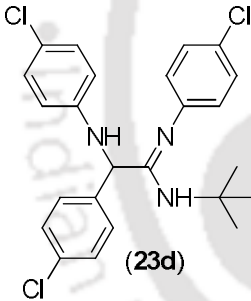
Pale yellow solid, mp 190 °C; **IR** (KBr): 3382, 2963, 1640, 1485, 1259, 1084 cm^{-1} ; **^1H NMR** (400 MHz, CDCl_3): δ 7.44 (d, $J = 8.4$ Hz, 2H), 7.19 (d, $J = 8.8$ Hz, 2H), 7.03 (dd, $J = 8.8$ Hz, $J = 4.0$ Hz, 4H), 6.57 (d, $J = 8.8$ Hz, 2H), 6.38 (d, $J = 8.4$ Hz, 2H), 5.79 (bs, 1H, NH), 4.79 (s, 1H), 3.68 (bs, 1H, NH), 1.43 (s, 9H); **^{13}C NMR** (100 MHz, CDCl_3): δ 153.5, 149.1, 145.3, 138.3, 135.1, 132.4, 129.8, 129.5, 128.8, 126.7, 124.8, 123.5, 115.0, 59.4, 51.4, 28.5. **Anal. Calcd** for $\text{C}_{24}\text{H}_{24}\text{BrCl}_2\text{N}_3$ (505.28): C, 57.05; H, 4.79; N, 8.32. Found: C,

56.92; H, 4.70; N, 8.21. **HRMS** (ESI) calcd for $C_{24}H_{24}BrCl_2N_3$ [M+H] 506.0581 found 506.0600.

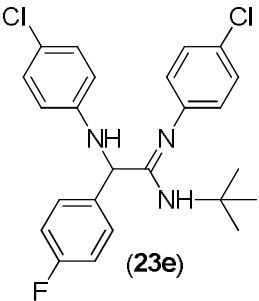
2-(4-Bromophenylamino)-N-tert-butyl-N',2-bis(4-bromophenyl)acetamidine (23c)

 Pale yellow solid, mp 182 °C; **IR** (KBr): 3381, 2965, 2901, 1641, 1591, 1581, 1506, 1488, 1388, 1253, 1180, 1164, 1070 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 7.43 (d, J = 8.4 Hz, 2H), 7.32 (d, J = 8.4 Hz, 2H), 7.16 (d, J = 8.0 Hz, 2H), 7.02 (d, J = 8.0 Hz, 2H), 6.52 (d, J = 8.8 Hz, 2H), 6.33 (d, J = 8.4 Hz, 2H), 5.77 (bs, 1H, NH), 4.78 (s, 1H), 3.71 (bs, 1H, NH), 1.42 (s, 9H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 153.4, 149.5, 145.7, 138.2, 132.4, 131.7, 129.8, 124.0, 122.9, 115.4, 114.3, 111.9, 59.3, 51.4, 28.4. **Anal. Calcd** for $C_{24}H_{24}Br_3N_3$ (594.18): C, 48.51; H, 4.07; N, 7.07. Found: C, 48.42; H, 3.98; N, 6.97.

2-(4-Chlorophenylamino)-N-tert-butyl-N',2-bis(4-chlorophenyl)acetamidine (23d)

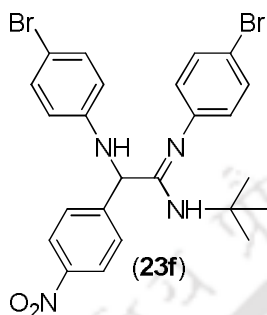
 White solid, mp 180 °C; **IR** (KBr): 3384, 2967, 2901, 1641, 1491, 1482, 1284, 1253, 1180, 1088 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 7.28 (d, J = 8.4 Hz, 2H), 7.19 (d, J = 8.8 Hz, 2H), 7.08 (d, J = 8.4 Hz, 2H), 7.03 (d, J = 8.4 Hz, 2H), 6.57 (d, J = 8.8 Hz, 2H), 6.38 (d, J = 8.4 Hz, 2H), 5.80 (bs, 1H, NH), 4.80 (s, 1H), 3.68 (bs, 1H, NH), 1.43 (s, 9H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 153.6, 149.1, 145.4, 137.8, 134.8, 129.6, 129.5, 129.4, 128.8, 126.7, 124.8, 123.6, 115.0, 59.4, 51.4, 28.5. **Anal. Calcd** for $C_{24}H_{24}Cl_3N_3$ (460.83): C, 62.55; H, 5.25; N, 9.12. Found: C, 62.44; H, 5.17; N, 9.02.

2-(4-Chlorophenylamino)-N-tert-butyl-N'-(4-chlorophenyl)-2-(4-fluorophenyl)acetamidine (23e)

 White solid, mp 153 °C; **IR** (KBr): 3408, 3386, 2964, 2924, 1634, 1601, 1509, 1494, 1483, 1230, 1181, 1157, 1086 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 7.19 (d, J = 9.2 Hz, 2H), 7.25-7.10 (m, 2H), 7.03-6.97 (m, 4H), 6.58 (d, J = 8.8 Hz, 2H), 6.37 (d, J = 8.8 Hz, 2H), 5.83 (bs, 1H, NH), 4.80 (d, J = 1.6 Hz, 1H), 3.69 (bs, 1H, NH), 1.43 (s, 9H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 153.9, 149.1, 145.5, 135.1, 130.0, 129.5, 128.7, 126.6, 124.7, 123.5, 116.2, 116.0, 115.0, 59.4, 51.3,

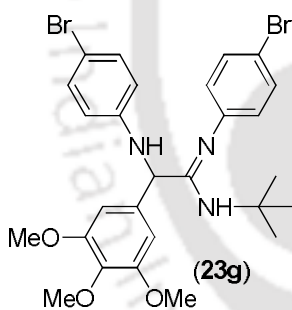
28.5. **Anal. Calcd** for $C_{24}H_{24}Cl_2FN_3$ (444.37): C, 64.87; H, 5.44; N, 9.46. Found: C, 64.76; H, 5.38; N, 9.34.

2-(4-Bromophenylamino)-N-tert-butyl-N'-(4-bromophenyl)-2-(4-nitrophenyl)acetamide (23f)



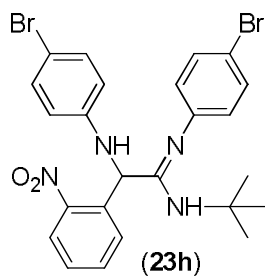
Yellow solid, mp 218 °C; **IR** (KBr): 3384, 2960, 1634, 1522, 1492, 1479, 1347, 1182 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 8.17 (d, $J = 8.8$ Hz, 2H), 7.37-7.32 (m, 4H), 7.18 (d, $J = 8.4$ Hz, 2H), 6.55 (d, $J = 8.8$ Hz, 2H), 6.33 (d, $J = 8.8$ Hz, 2H), 5.77 (bs, 1H, NH), 4.99 (d, $J = 1.6$ Hz, 1H), 3.73 (bs, 1H, NH), 1.44 (s, 9H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 152.7, 149.3, 148.0, 145.9, 145.4, 132.5, 131.9, 129.3, 124.4, 123.8, 115.6, 114.6, 112.4, 59.2, 51.6, 28.4. **Anal. Calcd** for $C_{24}H_{24}Br_2N_4O_2$ (560.28): C, 51.45; H, 4.32; N, 10.00. Found: C, 51.32; H, 4.26; N, 9.90.

2-(4-Bromophenylamino)-N-tert-butyl-N'-(4-bromophenyl)-2-(3,4,5-trimethoxyphenyl)acetamide (23g)



White solid, mp 215 °C; **IR** (KBr): 3382, 3362, 2958, 2935, 1637, 1595, 1508, 1481, 1329, 1131 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 7.34 (d, $J = 8.4$ Hz, 2H), 7.17 (d, $J = 8.8$ Hz, 2H), 6.54 (d, $J = 8.8$ Hz, 2H), 6.36 (d, $J = 8.4$ Hz, 2H), 6.32 (s, 2H), 5.89 (bs, 1H, NH), 4.71 (s, 1H), 3.85 (s, 3H), 3.76 (s, 7H), 1.44 (s, 9H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 153.9, 153.6, 149.9, 146.1, 138.3, 134.6, 132.4, 131.6, 124.3, 115.5, 114.1, 111.8, 105.1, 61.2, 60.5, 56.3, 51.3, 28.6. **Anal. Calcd** for $C_{27}H_{31}Br_2N_3O_3$ (605.36): C, 53.57; H, 5.16; N, 6.94. Found: C, 53.42; H, 5.08; N, 6.85.

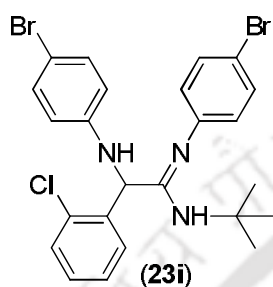
2-(4-Bromophenylamino)-N-tert-butyl-N'-(4-bromophenyl)-2-(2-nitrophenyl)acetamide (23h)



Light yellow solid, mp 198 °C; **IR** (KBr): 3357, 3298, 2969, 2956, 1615, 1580, 1520, 1480, 1347, 1258, 1181, 1071 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 7.75 (d, $J = 8.8$ Hz, 1H), 7.61-7.60 (m, 2H), 7.50-7.45 (m, 1H), 7.34 (d, $J = 8.8$ Hz, 2H), 7.06 (d, $J = 8.8$ Hz, 2H), 6.55 (d, $J = 8.8$ Hz, 2H), 6.16 (d, $J = 8.8$ Hz, 2H), 5.98 (bs, 1H, NH), 5.34 (d, $J = 3.2$ Hz, 1H), 3.94 (bs, 1H, NH), 1.45 (s, 9H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 153.4, 149.3, 149.0, 145.8, 133.7, 133.0,

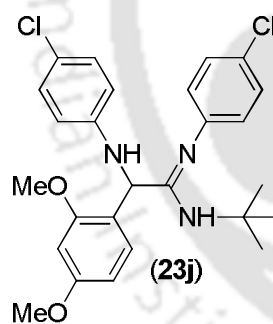
132.4, 131.7, 130.0, 129.8, 125.0, 123.4, 115.7, 114.4, 112.3, 56.1, 51.6, 28.4. **Anal. Calcd** for $C_{24}H_{24}Br_2N_4O_2$ (560.28): C, 51.45; H, 4.32; N, 10.00. Found: C, 51.34; H, 4.25; N, 9.90.

N'-(4-Bromophenyl)-2-(4-bromophenylamino)-*N*-tert-butyl-2-(2-chlorophenyl)acetamide (**23i**)



White solid, mp 166 °C; **IR** (KBr): 3373, 2966, 1719, 1669, 1633, 1588, 1492, 1373, 1212, 1127, 1071, 1033 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 7.57 (d, J = 6.4 Hz, 1H), 7.34-7.29 (m, 5H), 7.11 (d, J = 8.8 Hz, 2H), 6.54 (d, J = 8.8 Hz, 2H), 6.27 (d, J = 8.8 Hz, 2H), 5.89 (bs, 1H, NH), 5.16 (s, 1H), 3.65 (bs, 1H, NH), 1.47 (s, 9H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 153.7, 149.6, 146.3, 136.7, 134.9, 132.3, 131.8, 131.6, 130.2, 129.1, 127.4, 123.6, 115.4, 114.2, 111.9, 57.1, 51.4, 28.5. **Anal. Calcd** for $C_{24}H_{24}Br_2ClN_3$ (549.73): C, 52.44; H, 4.40; N, 7.64. Found: C, 52.38; H, 4.28; N, 7.52.

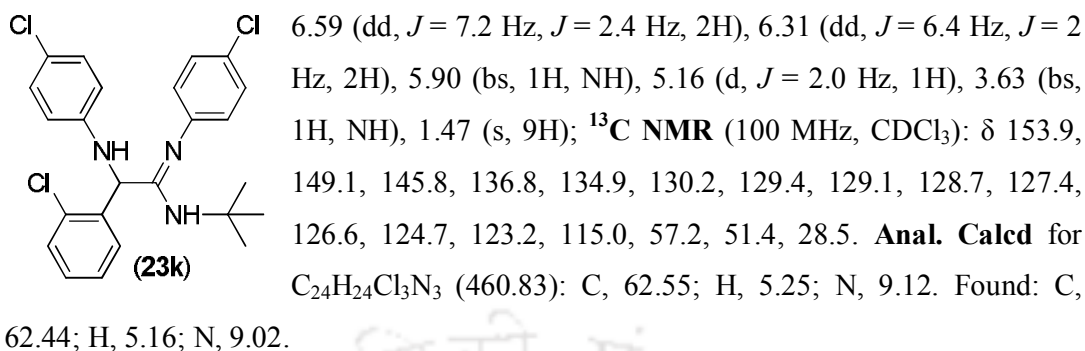
2-(4-Chlorophenylamino)-*N*-tert-butyl-*N'*-(4-chlorophenyl)-2-(2,4-dimethoxyphenyl)acetamide (**23j**)



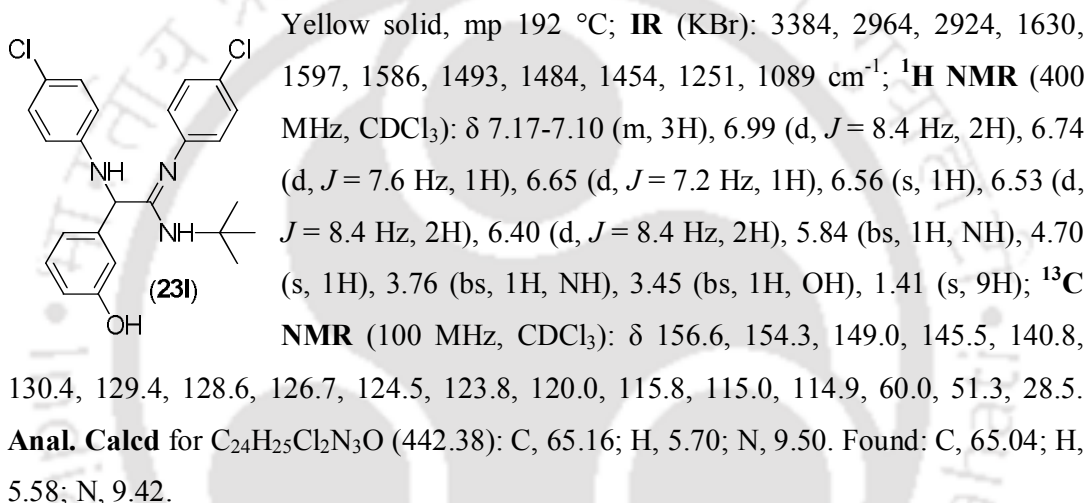
White solid, mp 140 °C; **IR** (KBr): 3358, 2962, 2838, 1614, 1587, 1509, 1493, 1482, 1290, 1207, 1171, 1098, 1032 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 7.30 (d, J = 8.4 Hz, 1H), 7.17 (d, J = 8.8 Hz, 2H), 6.96 (d, J = 8.8 Hz, 2H), 6.56 (d, J = 9.2 Hz, 2H), 6.47 (dd, J = 8.4 Hz, J = 2.4 Hz, 1H), 6.33 (d, J = 8.4 Hz, 2H), 6.28 (d, J = 2 Hz, 1H), 5.92 (bs, 1H, NH), 5.05 (s, 1H), 3.82 (s, 3H), 3.64 (bs, 1H, NH), 3.42 (s, 3H), 1.44 (s, 9H); **^{13}C NMR** (100 MHz, $CDCl_3$): δ 161.3, 158.0, 155.3, 149.8, 146.3, 129.2, 129.0, 128.3, 126.0, 124.0, 123.6, 120.1, 115.0, 104.6, 98.5, 55.5, 55.4, 53.4, 51.1, 28.5. **Anal. Calcd** for $C_{26}H_{29}Cl_2N_3O_2$ (486.43): C, 64.20; H, 6.01; N, 8.64. Found: C, 64.08; H, 5.93; N, 8.56.

2-(4-Chlorophenylamino)-*N*-tert-butyl-2-(2-chlorophenyl)-*N'*-(4-chlorophenyl)acetamide (**23k**)

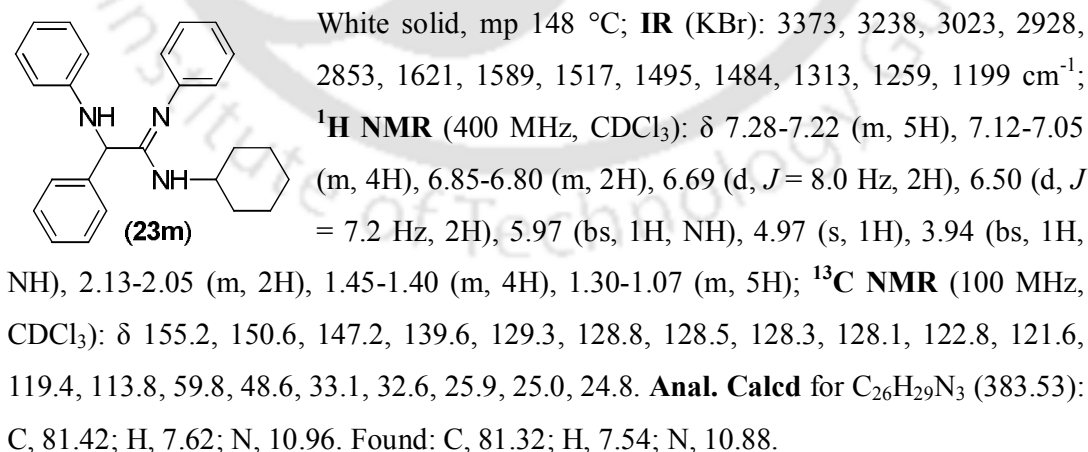
White solid, mp 188 °C; **IR** (KBr): 3401, 3371, 3045, 2961, 1634, 1599, 1513, 1494, 1482, 1182, 1089 cm^{-1} ; **1H NMR** (400 MHz, $CDCl_3$): δ 7.58 (d, J = 6.4 Hz, 1H), 7.31-7.27 (m, 3H), 7.19 (dd, J = 6.8 Hz, J = 2 Hz, 2H), 6.97 (dd, J = 6.8 Hz, J = 2.4 Hz, 2H),

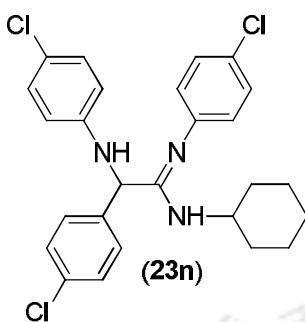


N-tert-Butyl-N'-(4-chlorophenyl)-2-(4-chlorophenylamino)2-(3-hydroxyphenyl)acetamide (23l)

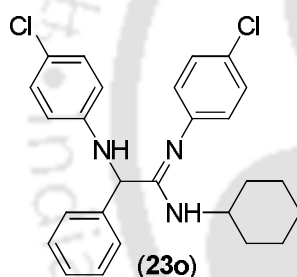


N-Cyclohexyl-N',2-diphenyl-2-(phenylamino)acetamide (23m)

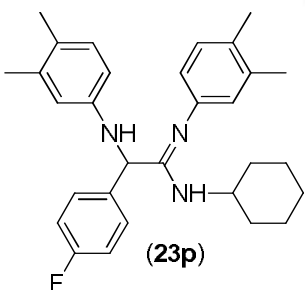


N',2-bis(4-Chlorophenyl)-2-(4-chlorophenylamino)-*N*-cyclohexylacetamidine (**23n**)

Pale yellow, mp 148 °C; **IR** (KBr): 3255, 2934, 2855, 1640, 1600, 1492, 1256, 1091, 1013 cm^{-1} ; **^1H NMR** (400 MHz, $\text{CDCl}_3/\text{DMSO}$): δ 7.25 (d, $J = 7.2$ Hz, 2H), 7.18 (m, 4H), 6.99 (d, $J = 6.8$ Hz, 2H), 6.70 (d, $J = 6.8$ Hz, 2H), 6.56 (d, $J = 7.6$ Hz, 2H), 4.97 (s, 1H), 3.79 (bs, 1H, NH), 3.20 (bs, 1H, NH), 1.98 (bs, 1H), 1.74-1.53 (m, 5H), 1.36-1.15 (m, 5H); **^{13}C NMR** (100 MHz, CDCl_3): δ 161.5, 142.8, 132.8, 132.5, 131.8, 131.4, 128.1, 128.0, 127.5, 127.4, 126.7, 121.4, 113.5, 56.2, 50.7, 29.2, 23.0, 22.6 **Anal.** **Calcd** for $\text{C}_{26}\text{H}_{26}\text{Cl}_3\text{N}_3$ (486.86): C, 64.14; H, 5.38; N, 8.63 Found: C, 64.05; H, 5.29; N, 8.57.

N'-(4-chlorophenyl)-2-((4-chlorophenyl)amino)-*N*-cyclohexyl-2-phenylacetimidamide (**23o**)

White solid, mp 178 °C; **^1H NMR** (400 MHz, CDCl_3): δ 7.31-7.26 (m, 3H), 7.16 (d, $J = 8.4$ Hz, 2H), 7.10 (d, $J = 6.8$ Hz, 2H), 6.99 (d, $J = 8$ Hz, 2H), 6.55 (d, $J = 8.4$ Hz, 2H), 6.38 (d, $J = 8.4$ Hz, 2H), 5.93 (d, $J = 7.6$ Hz, 1H), 4.83 (bs, NH, 1H), 3.85 (bs, NH, 1H), 2.11-1.97 (m, 2H), 1.72-1.61 (m, 3H), 1.45-1.03 (m, 6H); **^{13}C NMR** (100 MHz, CDCl_3): δ 155.3, 148.9, 145.5, 138.8, 129.3, 129.1, 128.8, 128.6, 128.0, 126.9, 124.4, 124.0, 114.9, 59.9, 48.8, 33.1, 32.6, 25.8, 25.0, 24.9. **Anal.** **Calcd** for $\text{C}_{26}\text{H}_{27}\text{Cl}_2\text{N}_3$ (452.41): C, 69.02; H, 6.02; N, 9.29 Found: C, 68.94; H, 5.92; N, 9.21.

N-cyclohexyl-*N'*-(3,4-dimethylphenyl)-2-((3,4-dimethylphenyl)amino)-2-(4-fluorophenyl)acetimidamide (**23p**)

Pale yellow solid, mp 175 °C; **^1H NMR** (400 MHz, CDCl_3): δ 7.10-7.06 (m, 2H), 6.98-6.91 (m, 2H), 6.80 (d, $J = 7.6$ Hz, 1H), 6.50 (d, $J = 2$ Hz, 1H), 6.43 (dd, $J = 8$ Hz, $J = 2.4$, 1H), 6.25-6.22 (m, 2H), 5.93 (d, $J = 8.4$ Hz, 1H), 4.88 (bs, NH, 1H), 3.93-3.91 (m, 1H), 3.53 (bs, NH, 1H), 2.22 (s, 3H), 2.17 (s, 3H), 2.12 (s, 3H), 2.02 (s, 3H), 1.72-1.69 (m, 1H), 1.65-1.57 (m, 3H), 1.46-1.34 (m, 2H), 1.26-1.04 (m, 5H); **^{13}C NMR** (100 MHz, CDCl_3): δ 155.6, 148.1, 145.4, 137.5, 136.4, 136.1, 130.4, 130.0, 129.7, 129.4, 127.6, 124.2, 119.8,

115.7, 115.5, 111.4, 59.5, 48.6, 33.3, 32.7, 26.0, 25.2, 25.0, 20.1, 19.8, 19.1, 18.9. **Anal. Calcd** for $C_{30}H_{36}FN_3$ (457.62): C, 78.74; H, 7.93; N, 9.18 Found: C, 78.63; H, 7.82; N, 9.05.

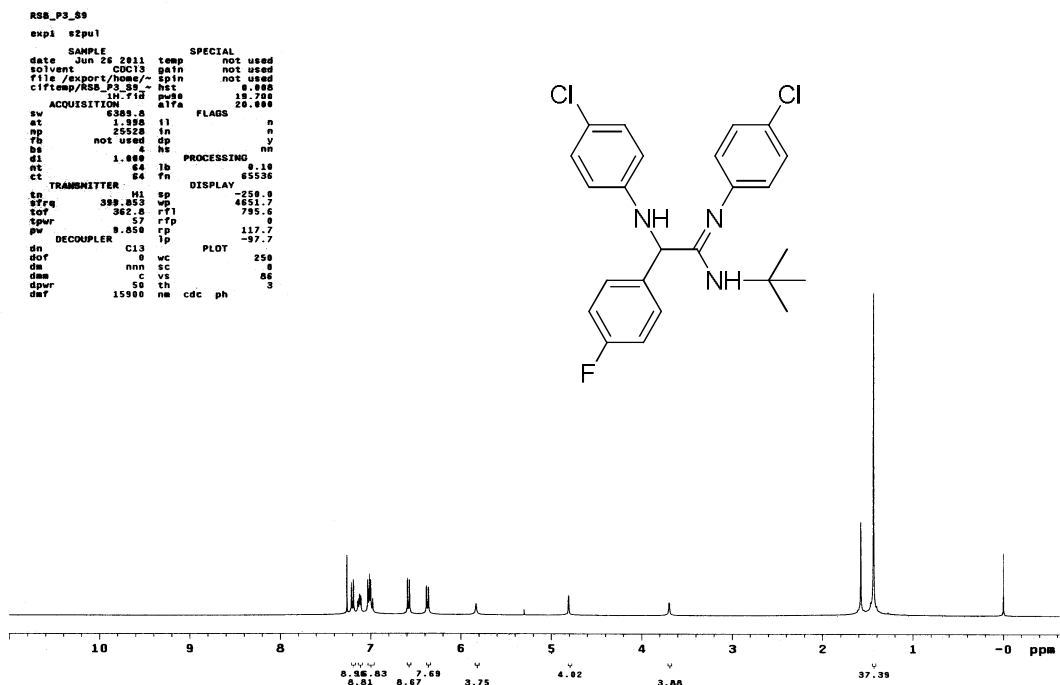
Complete crystallographic data of **23a**, **23b** and **23n** for the structural analysis have been deposited with the Cambridge Crystallographic Data Centre, CCDC No. 855248, 838468 and 837583 respectively. Copies of this information may be obtained free of charge from the Director, Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK, (fax: +44-1223-336033, e-mail: deposit@ccdc.cam.ac.uk or via: www.ccdc.cam.ac.uk).

Table 14. Crystal data and structures refinement for the compound **23a**, **23b** and **23n**, for atomic coordinates, equivalent isotropic displacement parameters and bond angles, please check the CIF.

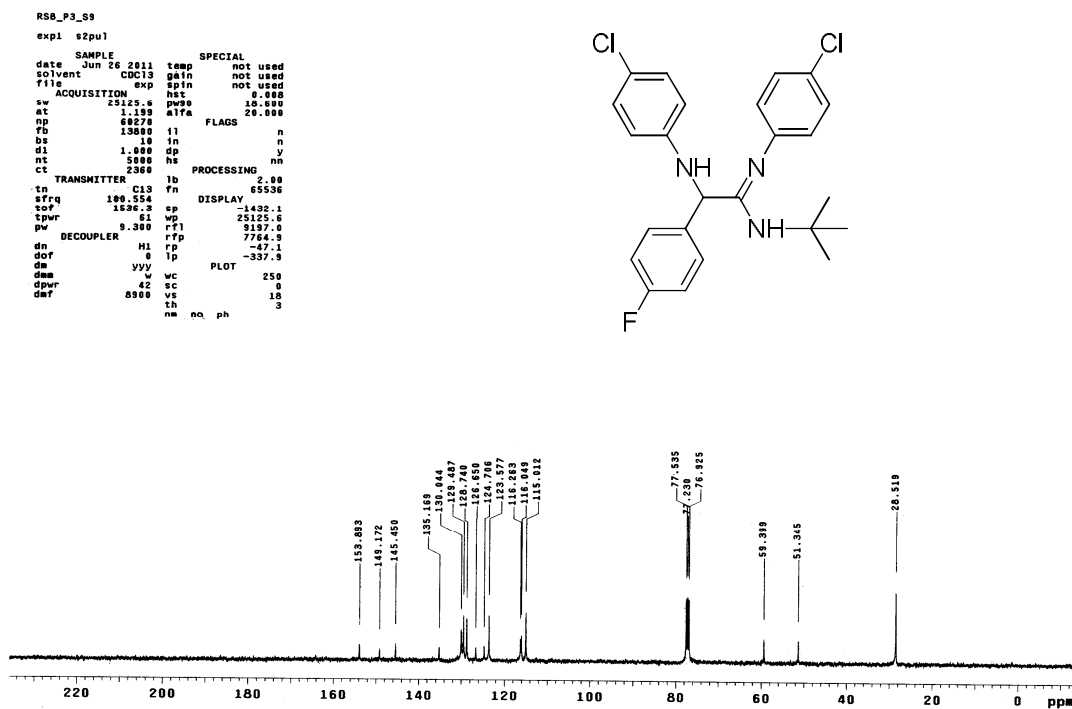
Parameters	Compound 23a	Compound 23b	Compound 23n
Empirical Formula	' $C_{24}H_{27}N_3$ '	' $C_{24}H_{24}BrCl_2N_3$ '	' $C_{26}H_{26}Cl_3N_3$ '
Formula weight	357.49	505.27	486.85
Temperature	296 K	296 K	296 K
CCDC no	855248	838468	837583
Wavelength (Å)	0.71073 Å	0.71073 Å	0.71073 Å
Crystal system	Monoclinic	Orthorhombic	Monoclinic
Space group	P 21/c	Pna2(1)	P2(1)/n
Radiation type	Mo K α	Mo K α	Mo K α
Radiation source	'fine-focus sealed tube'	'fine-focus sealed tube'	'fine-focus sealed tube'
<i>a</i> (Å)	11.6045 (7) Å	13.3588 (10) Å	12.9999 (9) Å
<i>b</i> (Å)	11.6974 (7) Å	10.7992 (8) Å	11.9658 (10) Å
<i>c</i> (Å)	15.6896 (10) Å	16.6412 (12) Å	16.7478 (12) Å
α (°)	90.00°	90.00°	90.00°
β (°)	102.044 (3)°	90.00°	92.120 (4)°
γ (°)	90.00°	90.00°	90.00°
Cell Volume	2082.9 (2) Å ³	2400.7 (3) Å ³	2603.4 (3) Å ³
<i>Z</i>	4	4	4

Density	1.140 g/cm ³	1.398 g/cm ³	1.242 g/cm ³
<i>F</i> (0 0 0)	768	1032	1016
Theta range for data collection	1.79 to 25.50°	2.25 to 28.44°	1.95 to 26.49°
Index ranges	-10 ≤ <i>h</i> ≤ 14, -13 ≤ <i>k</i> ≤ 13, -18 ≤ <i>l</i> ≤ 18	-17 ≤ <i>h</i> ≤ 17, -14 ≤ <i>k</i> ≤ 14, -22 ≤ <i>l</i> ≤ 22	-13 ≤ <i>h</i> ≤ 15, -13 ≤ <i>k</i> ≤ 12, -20 ≤ <i>l</i> ≤ 20
Reflections collected	20224	29277	21584
Independent reflections	3784 [<i>R</i> (int) = 0.1111]	5976 [<i>R</i> (int) = 0.0367]	4970 [<i>R</i> (int) = 0.0450]
Completeness to theta	25.50° 97.7 %	28.44° 99.6 %	26.49° 92.4 %
Number of parameters	247	271	289
Number of restraints	0	1	0
Goodness-of-fit (GOF) on <i>F</i> ²	0.973	0.993	1.031
Refinement method	Full-matrix least-squares on <i>F</i> ²	Full-matrix least-squares on <i>F</i> ²	Full-matrix least-squares on <i>F</i> ²

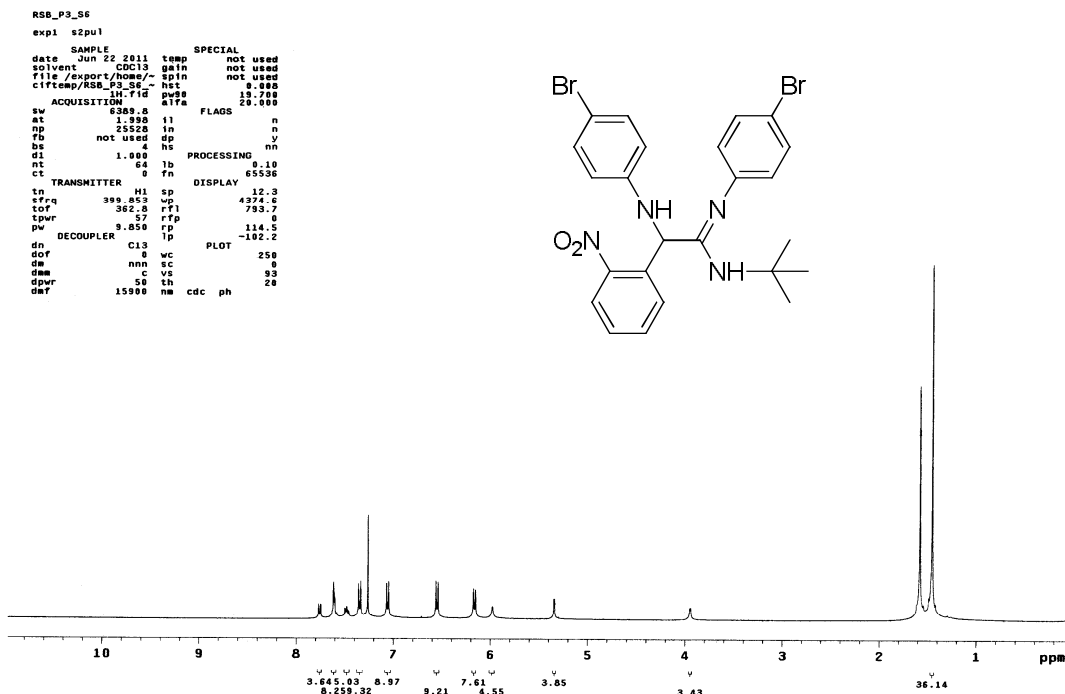
¹H NMR (400 MHz, CDCl₃): 2-(4-Chlorophenylamino)-N-tert-butyl-N'-(4-chlorophenyl)-2-(4-fluorophenyl)acetamidine (23e)



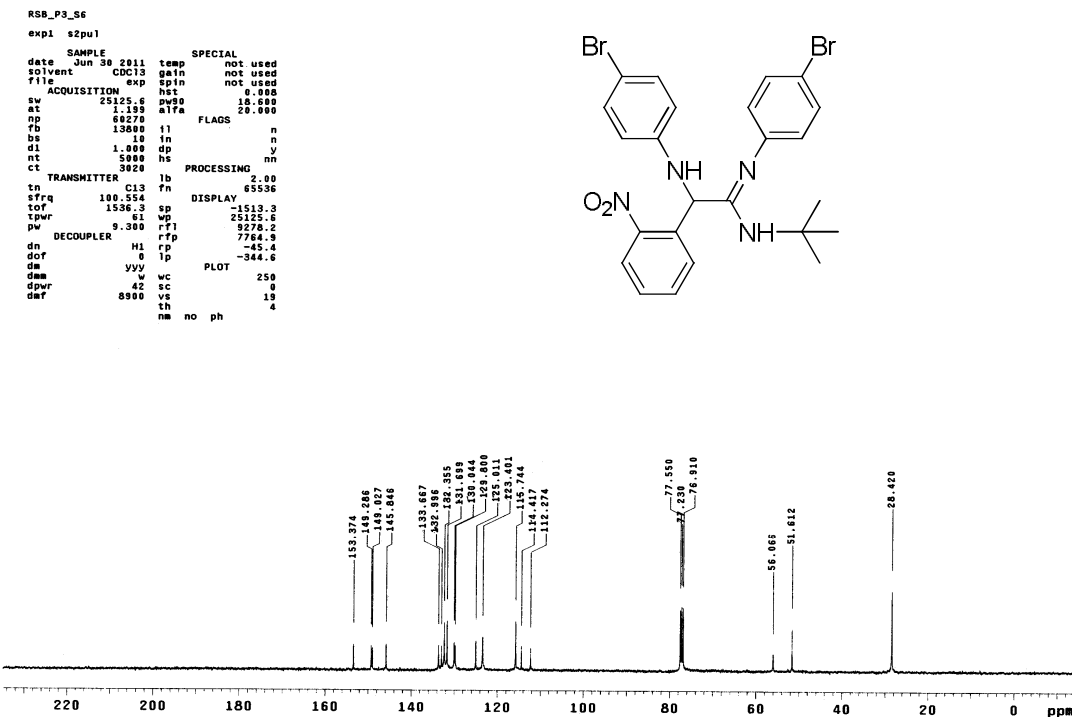
¹³C NMR (100 MHz, CDCl₃): 2-(4-Chlorophenylamino)-N-tert-butyl-N'-(4-chlorophenyl)-2-(4-fluorophenyl)acetamidine (23e)



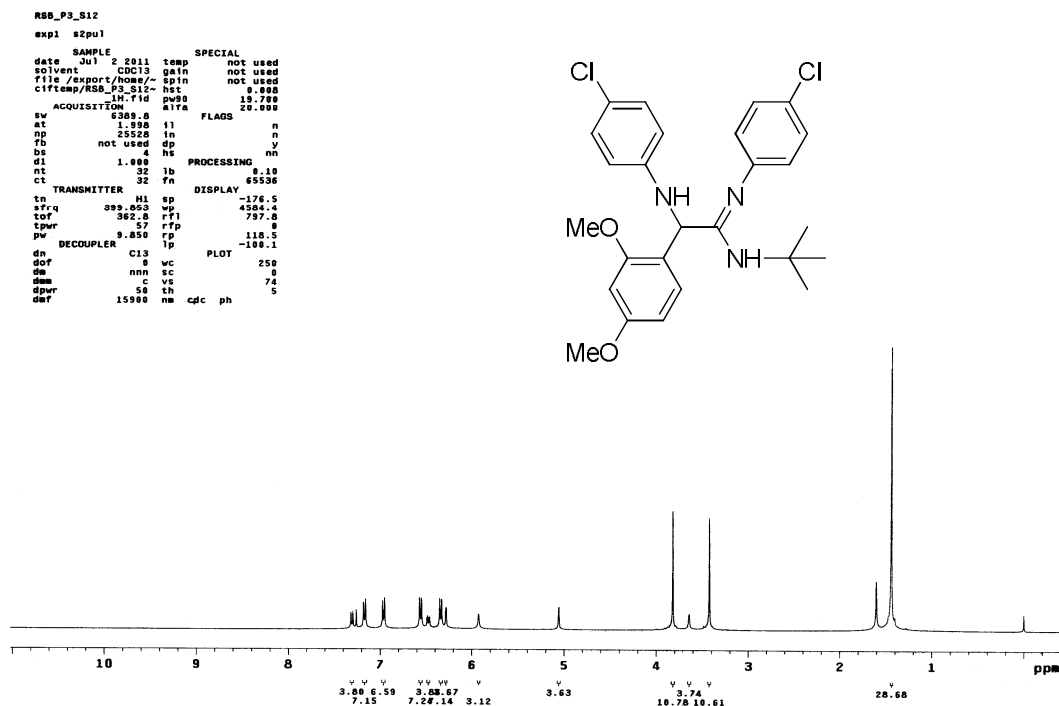
^1H NMR (400 MHz, CDCl_3): 2-(4-Bromophenylamino)-*N*-*tert*-butyl-*N'*-(4-bromophenyl)-2-(2-nitrophenyl)acetamide (23h)



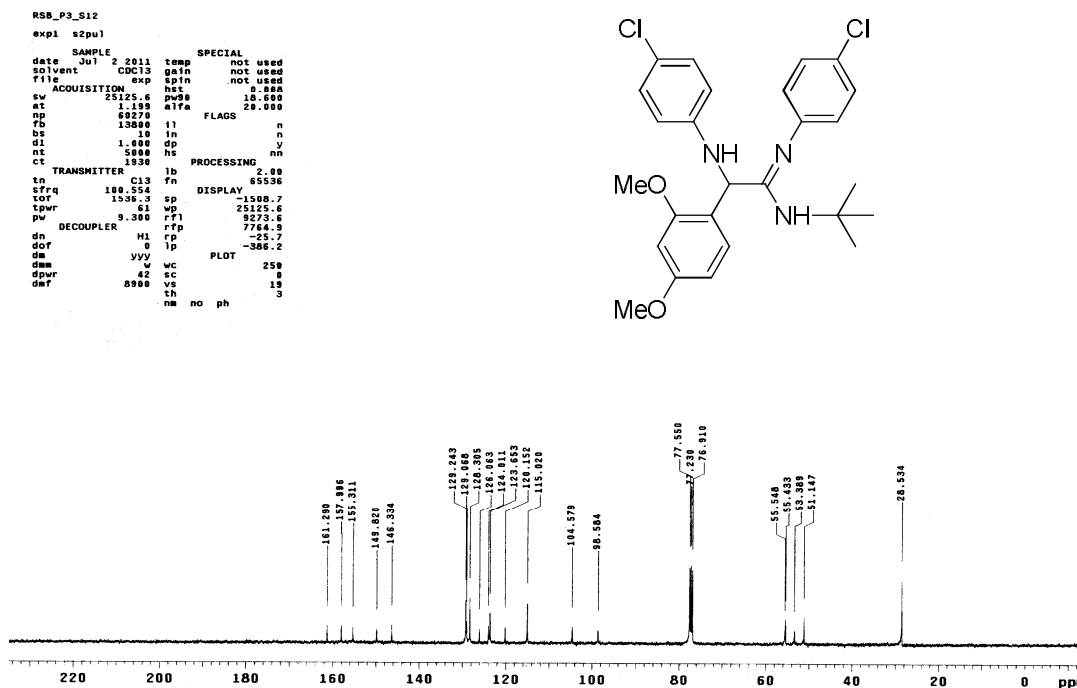
^{13}C NMR (100 MHz, CDCl_3): 2-(4-Bromophenylamino)-*N*-*tert*-butyl-*N'*-(4-bromophenyl)-2-(2-nitrophenyl)acetamide (23h)



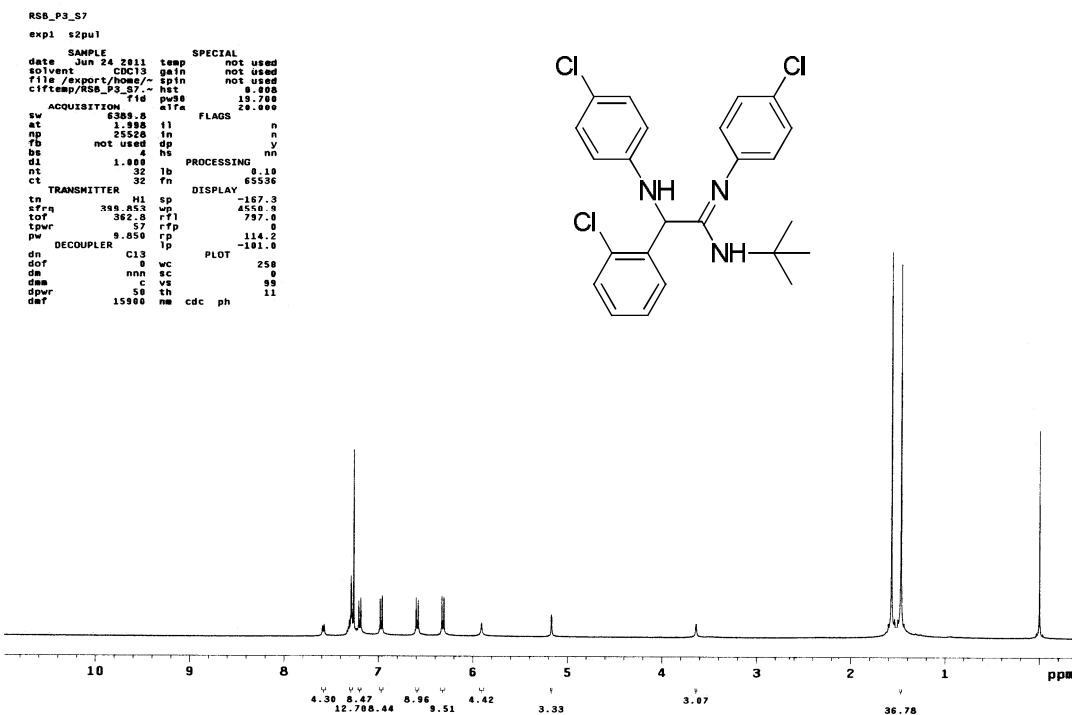
¹H NMR (400 MHz, CDCl₃): 2-(4-Chlorophenylamino)-N-tert-butyl-N'-(4-chlorophenyl)-2-(2,4-dimethoxyphenyl)acetamide (23j)



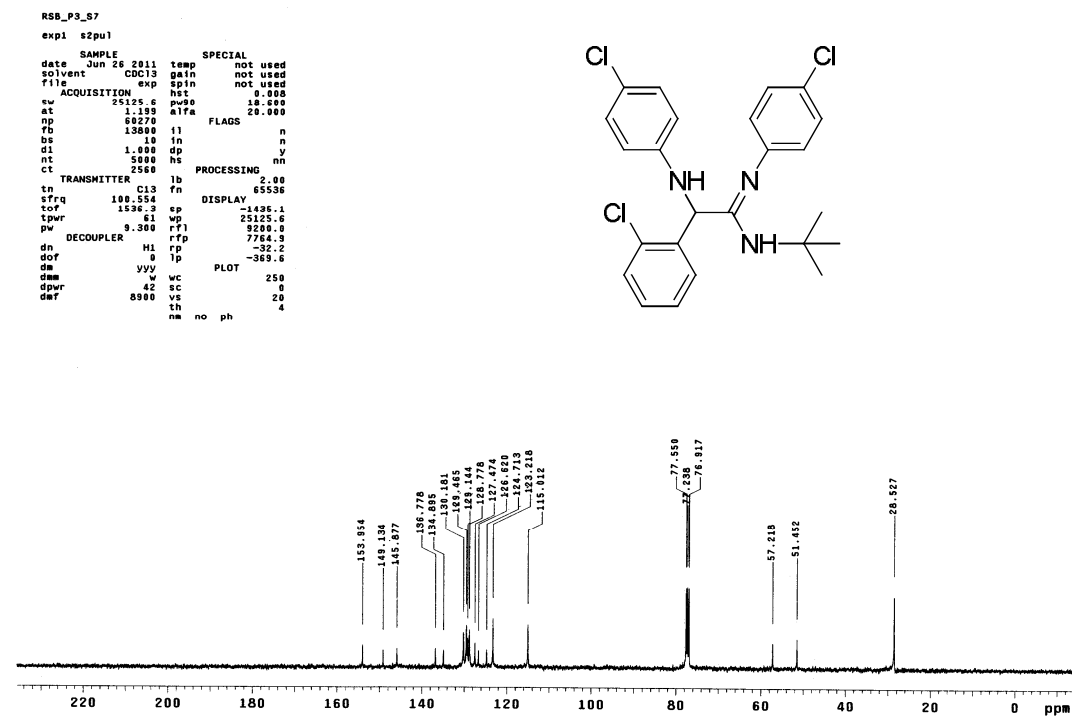
¹³C NMR (100 MHz, CDCl₃): 2-(4-Chlorophenylamino)-N-tert-butyl-N'-(4-chlorophenyl)-2-(2,4-dimethoxyphenyl)acetamide (23j)



$^1\text{H NMR}$ (400 MHz, CDCl_3): 2-(4-Chlorophenylamino)-N-tert-butyl-2-(2-chlorophenyl)-N'-(4-chlorophenyl)acetamidine (23k)



$^{13}\text{C NMR}$ (100 MHz, CDCl_3): 2-(4-Chlorophenylamino)-N-tert-butyl-2-(2-chlorophenyl)-N'-(4-chlorophenyl)acetamidine (23k)





References

References

1. Jiménez-González, C.; Constable, D. J. C. *Green Chemistry and Engineering*, John Wiley & Sons, Inc.: New Jersey, USA, 2011.
2. (a) Zhu, J.; Bienaymé, H. *Multicomponent Reactions*; 1st ed.; Wiley-VCH: Weinheim, Germany, 2005. (b) Orru, R. V. A.; Ruijter, E. In *Synthesis of Heterocycles via Multicomponent Reaction I & II*, ed. Maes, B. U. W.; Springer: Heidelberg, Germany, 2010.
3. Dömling, A.; Ugi, I. *Angew. Chem. Int. Ed.* **2000**, *39*, 3168-3210.
4. Pellissier, H. *Chem. Rev.* **2013**, *113*, 442-524.
5. Armstrong, R. W.; Combs, A. P.; Tempest, P. A.; Brown, S. D.; Keating, T. A. *Acc. Chem. Res.* **1996**, *29*, 123-131.
6. (a) Ruijter, E.; Scheffelaar, R.; Orru, R. V. A. *Angew. Chem. Int. Ed.* **2011**, *50*, 6234-6246. (b) Toure', B. B.; Hall, D. G. *Chem. Rev.* **2009**, *109*, 4439-4486.
7. Dömling, A.; Wang, W.; Wang, K. *Chem. Rev.* **2012**, *112*, 3083-3135.
8. Lipinski, C. A.; Lombardo, F.; Dominy, B. W.; Feeney, P. J. *Adv. Drug Delivery Rev.* **2001**, *46*, 3-26.
9. Strecker, A. *Liebigs Ann. Chem.* **1850**, *75*, 27-45.
10. (a) Hantzsch, A. *Justus Liebigs Ann. Chem.* **1882**, *215*, 1-82. (b) Hantzsch, A. *Ber. Dtsch. Chem. Ges.* **1890**, *23*, 1474-1476.
11. (a) Biginelli, P. *Ber. Dtsch. Chem. Ges.* **1891**, *24*, 1317-1319. (b) Biginelli, P. *Ber. Dtsch. Chem. Ges.* **1891**, *24*, 2962-2967. (c) Biginelli, P. *Ber. Dtsch. Chem. Ges.* **1893**, *26*, 447-450.
12. Mannich, C.; Krosche, W. *Arch. Pharm.* **1912**, *250*, 647-667.
13. (a) Passerini, M. *Gazz. Chim. Ital.* **1921**, *51*, 126-129. (b) Passerini, M. *Gazz. Chim. Ital.* **1921**, *51*, 181-189.
14. (a) Ugi, I.; Meyr, R.; Fetzer, U.; Steinbrückner, C. *Angew. Chem. Int. Ed.* **1959**, *71*, 386-388. (b) Ugi, I.; Steinbrückner, C. *Angew. Chem. Int. Ed.* **1960**, *72*, 267-268. (c) Ugi, I.; Steinbrückner, C. *Chem. Ber.* **1961**, *94*, 734-742.
15. Brauch, S.; Gabriela, L.; Westermann, B. *Chem. Commun.* **2010**, *46*, 3387-3389.

16. Elders, N.; Born, D. V. D.; Hendrickx, L. J. D.; Timmer, B. J. J.; Krause, A.; Janssen, E.; de Kanter, F. J. J.; Ruijter, E.; Orru, R. V. A. *Angew. Chem. Int. Ed.* **2009**, *48*, 5856-5859.
17. (a) Domling, A. *Chem. Rev.* **2006**, *106*, 17-89. (b) Sağırli, A.; Dürüst, Y.; Kariuki, B.; Knight, D. W. *Tetrahedron* **2013**, *69*, 69-72.
18. Schmidt, U.; Weinbrenner, S. *J. Chem. Soc., Chem. Commun.* **1994**, 1003-1004.
19. Chapman, T. M.; Davies, L. G.; Gu, B.; Block, T. M.; Scopes, D. I. C.; Hay, P. A.; Courtney, S. M.; McNeill, L. A.; Schofield, C. J.; Davis, B. G. *J. Am. Chem. Soc.* **2005**, *127*, 506-507.
20. Domling, A.; Beck, B.; Eichelberger, U.; Sakamuri, S.; Menon, S.; Chen, Q.-Z.; Lu, Y.; Wessjohann, L. A. *Angew. Chem. Int. Ed.* **2006**, *45*, 7235-7239.
21. Yue, T.; Wang, M-X.; Wang, D-X.; Zhu, J. *Angew. Chem. Int. Ed.* **2008**, *47*, 9454-9457.
22. Znabet, A.; Polak, M. M.; Janssen, E.; de Kanter, F. J. J.; Turner, N. J.; Orru, R. V. A.; Ruijter, E. *Chem. Commun.* **2010**, *46*, 7918-7920.
23. Semple, J. E.; Wang, P. C.; Lysenko, Z.; Joullié, M. M. *J. Am. Chem. Soc.* **1980**, *102*, 7505-7510.
24. De Laszlo, S. E.; Williard, P. G. *J. Am. Chem. Soc.* **1985**, *107*, 199-203.
25. Rivera, D. G.; Wessjohann, L. A. *J. Am. Chem. Soc.* **2009**, *131*, 3721-3732.
26. Enguehard-Gueiffier, C.; Gueiffier, A. *Mini-Rev. Med. Chem.* **2007**, *7*, 888-899.
27. Hanson, S. M.; Morlock, E. V.; Satyshur, K. A.; Czajkowski, C. *J. Med. Chem.* **2008**, *51*, 7243-7252.
28. Wiegand, M. H. *Drugs* **2008**, *68*, 2411-2417.
29. Rival, Y.; Grassy, G.; Michel, G. *Chem. Pharm. Bull.* **1992**, *40*, 1170-1176.
30. (a) Elhakmaoui, A.; Gueiffier, A.; Milhavet, J.-C.; Blache, Y.; Chapat, J.-P.; Chavignon, O.; Teulade, J.-C.; Snoeck, R.; Andrei, G.; De Clercq, E. *Bioorg. Med. Chem. Lett.* **1994**, *4*, 1937-1940. (b) Mavel, S.; Renou, J. L.; Galtier, C.; Snoeck, R.; Andrei, G.; Balzarini, J.; De Clercq, E.; Gueiffier, A. *Arzneim Forsch.* **2001**, *51*, 304-309.
31. Lacerda, R. B.; de Lima, C. K. F.; da Silva, L. L.; Romeiro, N. C.; Miranda, A. L. P.; Barreiro, E. J.; Fraga, C. A. M. *Bioorg. Med. Chem.* **2009**, *17*, 74-84.

32. Baskaran, S.; Maung, J.; Neitzel, M. L.; Rai, R.; Slododov, I.; Tai, V. W.-F. US 0204265, 2010.
33. Trapani, G.; Latrofa, A.; Franco, M.; Carrieri, A.; Cellamare, S.; Serra, M.; Sanna, E.; Biggio, G.; Liso, G. *Eur. J. Pharm. Sci.* **2003**, *18*, 231-240.
34. Almirante, L.; Polo, L.; Mugnaini, A.; Provinciali, E.; Rugarli, P.; Biancotti, A.; Gamba, A.; Murmann, W. *J. Med. Chem.* **1965**, *8*, 305-312.
35. Kaminski, J. J.; Bristol, J. A.; Puchalski, C.; Lovey, R. G.; Elliott, A. J.; Guzik, H.; Solomon, D. M.; Conn, D. J.; Domalski, M. S.; Wong, S. C.; Gold, E. H.; Long, J. F.; Chiu, P. J. S.; Steinberg, M.; Mc Phail, A. T. *J. Med. Chem.* **1985**, *28*, 876-892.
36. (a) Lopez-Martinez, M.; Salgado-Zamora, H.; San-Juan, E. R.; Zamudio, S.; Picazo, O.; Campos, M. E.; Naranjo-Rodriguez, E. B. *Drug Dev. Res.* **2010**, *71*, 371-381. (b) Baviskar, A. T.; Madaan, C.; Preet, R.; Mohapatra, P.; Jain, V.; Agarwal, A.; Guchhait, S. K.; Kundu, C. N.; Banerjee, U. C.; Bharatam, P. V. *J. Med. Chem.* **2011**, *54*, 5013-5030.
37. (a) Kalpan, J.-P.; George, P. US Patent 4460592, 1984. (b) Davey, D.; Erhardt, P. W.; Lumma, W. C., Jr.; Wiggins, J.; Sullivan, M.; Pang, D.; Cantor, E. *J. Med. Chem.* **1987**, *30*, 1337-1342.
38. Sanfilippo, P. J.; Urbanski, M.; Press, J. B.; Dubinsky, B.; Moore, J. B., Jr. *J. Med. Chem.* **1988**, *31*, 2221-2227.
39. (a) Byth, K. F.; Culshaw, J. D.; Green, S.; Oakes, S. E.; Thomas, A. P. *Bioorg. Med. Chem.* **2004**, *14*, 2245-2248. (b) Cai, D.; Byth, K. F.; Shapiro, G. I. *Cancer Res.* **2006**, *66*, 435-444.
40. Abe, Y.; Kayakiri, H.; Satoh, S.; Inoue, T.; Sawada, Y.; Imai, K.; Inamura, N.; Asano, M.; Hatori, C.; Katayama, A.; Oku, T.; Tanaka, H. *J. Med. Chem.* **1998**, *41*, 564-578.
41. Higaki, J.; Murphy, G. M., Jr.; Cordell, B. *J. Neurochem.* **1997**, *68*, 333-336.
42. (a) Howard, A. S. In *Comprehensive Heterocyclic Chemistry*; Katritzky, A. R.; Rees, C. W.; Sciven, E. V. F., Eds.; Pergamon Press: London, 1996; Vol. 8, p 262. (b) Kayakiri, H.; Takase, S.; Shibata, T.; Okamoto, M.; Terano, H.; Hashimoto, M. *J. Org. Chem.* **1989**, *54*, 4015-4016. (c) Hering, K. W.; Karaveg, K.; Moremen, K. W.; Pearson, W. H. *J. Org. Chem.* **2005**, *70*, 9892-9904.

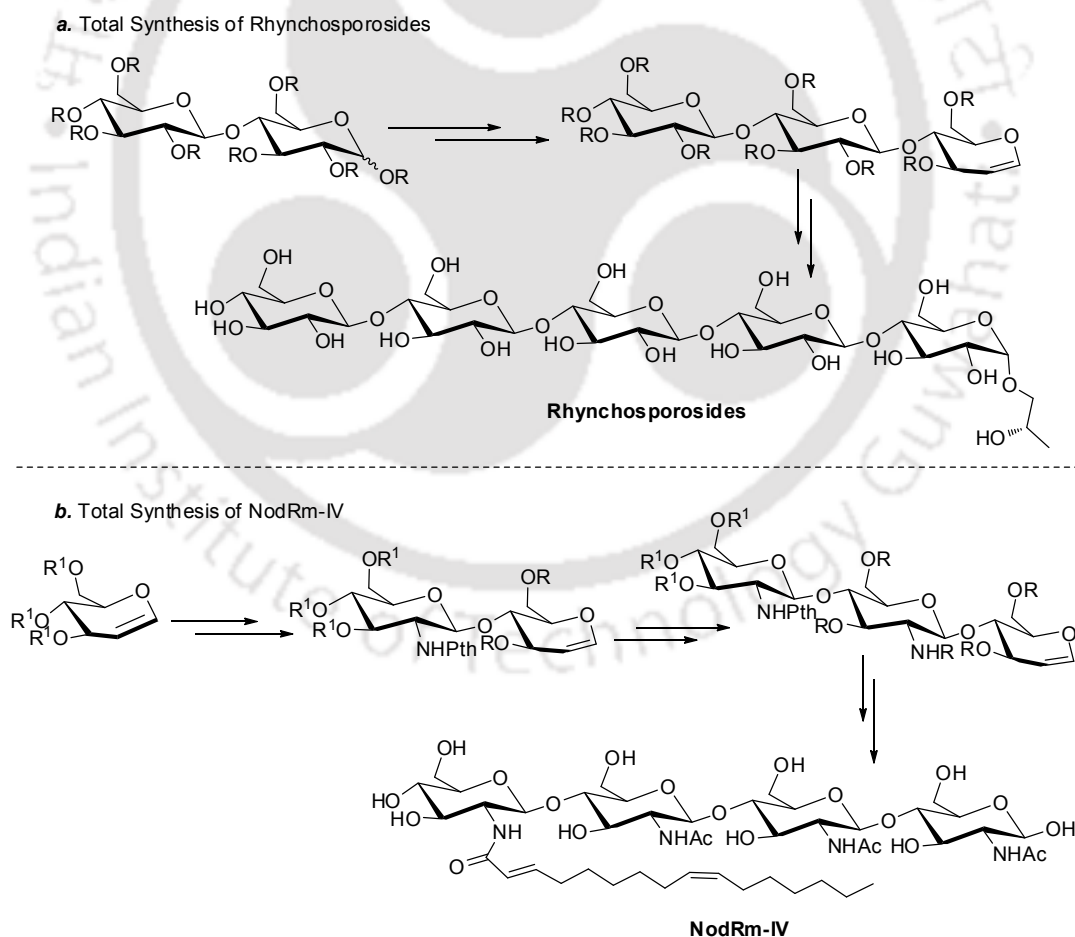
43. Melis, M. R.; Succu, S.; Sanna, F.; Melis, T.; Mascia, M. S.; Enguehard-Gueiffier, C.; Hubner, H.; Gmeiner, P.; Gueiffier, A.; Argiolas, A. *Eur. J. Neurosci.* **2006**, *24*, 2021-2030.
44. Ohta, M.; Suzuki, T.; Koide, T.; Matsuhisa, A.; Furuya, T.; Miyata, K.; Yanagisawa, I. *Chem. Pharm. Bull.* **1996**, *44*, 991-999.
45. Parrino, L.; Smerieri, A.; Giglia, F.; Milioli, G.; De Paolis, F.; Terzano, M. G. *Clin Neuropharmacol.* **2008**, *31*, 40-50.
46. Langer, S. Z.; Arbilla, S.; Benavides, J.; Scatton, B. *Adv. Biochem. Psychopharmacol.* **1990**, *46*, 61-72.
47. Adachi, H.; Kamata, S.; Kodama, K.; Nagakura, T. *Eur. J. Pharmacol.* **2000**, *396*, 43-47.
48. (a) Nagarajan, N.; Velmurugan, G.; Prakash, A.; Shakti, N.; Katiyar, M.; Venuvanalingam, P.; Renganathan, R. *Chem. Asian J.* **2014**, *9*, 294-304. (b) Mutai, T.; Tomoda, H.; Ohkawa, T.; Yabe, Y.; Araki, K. *Angew. Chem. Int. Ed.* **2008**, *47*, 9522-9524.
49. Laquintana, V.; Denora, N.; Lopodota, A.; Suzuki, H.; Sawada, M.; Serra, M.; Biggio, G.; Latrofa, A.; Trapani, G.; Liso, G. *Bioconjugate Chem.* **2007**, *18*, 1397-1407.
50. Starrett, J. E., Jr.; Montzka, T. A.; Crosswell, A. R.; Cavanagh, R. L. *J. Med. Chem.* **1989**, *32*, 2204-2210.
51. Gueiffier, A.; Mavel, S.; Lhassani, M.; Elhakmaoui, A.; Snoeck, R.; Andrei, G.; Chavignon, O.; Teulade, J. C.; Witvrouw, M.; Balzarini, J.; Clercq, E. D.; Chapat, J. P. *J. Med. Chem.* **1998**, *41*, 5108-5112.
52. Kercher, T.; Rao, C.; Bencsik, J. R.; Josey, J. A. *J. Comb. Chem.* **2007**, *9*, 1177-1187.
53. DiMauro, E. F.; Kennedy, J. M. *J. Org. Chem.* **2007**, *72*, 1013-1016.
54. Guchhait, S. K.; Madaan, C. *Org. Biomol. Chem.* **2010**, *8*, 3631-3634.
55. Adib, M.; Mohamadi, A.; Sheikhi, E.; Ansari, S.; Bijanzadeh, H. R. *Synlett* **2010**, 1606-1608.
56. Burchak, O. N.; Mughlerli, L.; Ostuni, M.; Lacapère, J. J.; Balakirev, M. Y. *J. Am. Chem. Soc.* **2011**, *133*, 10058-10061.

57. Shao, N.; Pang, G. -X.; Yan, C. -X.; Shi, G. -F.; Cheng, Y. *J. Org. Chem.* **2011**, *76*, 7458-7465.
58. Wu, Z.; Pan, Y.; Zhou, X. *Synthesis* **2011**, 2255-2260.
59. Chunavala, K. C.; Joshi, G.; Suresh, E.; Adimurthy, S. *Synthesis* **2011**, 635-641.
60. Husinec, S.; Markovic, R.; Petkovic, M.; Nasufovic, V.; Savic, V. *Org. Lett.* **2011**, *13*, 2286-2289.
61. Ireland, S. M.; Tye, H.; Whittaker, M. *Tetrahedron Lett.* **2003**, *44*, 4369-4371.
62. Rousseau, A. L.; Matlaba, P.; Parkinson, C. J. *Tetrahedron Lett.* **2007**, *48*, 4079-4082.
63. Varma, R. S.; Kumar, D. *Tetrahedron Lett.* **1999**, *40*, 7665-7669.
64. Shaabani, A.; Soleimani, E.; Maleki, A. *Tetrahedron Lett.* **2006**, *47*, 3031-3034.
65. Lyon, M. A.; Kercher, T. S. *Org. Lett.* **2004**, *6*, 4989-4992.
66. Chen, J. J.; Golebiowski, A.; McClenaghan, J.; Klopfenstein, S. R.; West, L. *Tetrahedron Lett.* **2001**, *42*, 2269-2271.
67. Adib, M.; Mahdavi, M.; Noghani, M. A.; Mirzaei, P. *Tetrahedron Lett.* **2007**, *48*, 7263-7265.
68. Greenhill, J. V.; Lue, P. *Prog. Med. Chem.* **1993**, *30*, 203-326.
69. (a) Gupta, S.; Krasnoff, S. B.; Roberts, D. W.; Renwick, J. A. A. *J. Am. Chem. Soc.* **1991**, *113*, 707-709. (b) Aoyama, T.; Kojima, F.; Imada, C.; Muraoka, Y.; Naganawa, H.; Okami, Y.; Takexichi, T.; Aoyagi, T. *J. Enzyme Inhibition.* **1995**, *8*, 223-232. (c) Fredenhagen, A.; Molleyres, L-P.; Bohlendorf, B.; Laue, G. *J. Antibiot.* **2006**, *59*, 267-280.
70. (a) Carle, J. S.; Christophersen, C. *J. Am. Chem. Soc.* **1979**, *101*, 4012-4013. (b) Carle, J. S.; Christophersen, C. *J. Org. Chem.* **1981**, *46*, 3440-3443. (c) Lindel, T.; Brauchle, L.; Golz, G.; Bohrer, P. *Org. Lett.* **2007**, *9*, 283-286. (d) Verbitski, S. M.; Mayne, C. L.; Davis, R. A.; Concepcion, G. P.; Ireland, C. M. *J. Org. Chem.* **2002**, *67*, 7124-7126. (e) Fuchs, J. R.; Funk, R. L. *J. Am. Chem. Soc.* **2004**, *126*, 5068-5069.
71. Soli's, P. N.; Ravelo, A. G.; Palenzuela, J. A.; Gupta, M. P.; González, A.; Phillipson, J. D. *Phytochemistry*, **1997**, *44*, 963-969.

72. (a) Yamada, T.; Suegane, K.; Kuwata, S.; Watanabe, H. *Bull. Chem. Soc. Jpn.* **1977**, *50*, 1088-1093. (b) Yamada, T.; Suegane, K.; Miyazawa, S.; Kuwata, S.; Watanabe, H. *Bull. Chem. Soc. Jpn.* **1978**, *51*, 878-883.
73. Umezuma, H.; Takita, T.; Sugiura, Y.; Otsuka, M.; Kobayashi, S.; Ohno, M. *Tetrahedron* **1984**, *40*, 501-509.
74. Goto, T.; Isobe, M.; Coviello, D. A.; Kishi, Y. *Tetrahedron* **1973**, *29*, 2035-2039.
75. Kaufmann, H. P.; Budwig, J.; Mohnke, K. *Chem. Ber.* **1942**, *75*, 1585-1600.
76. Mengelberg, M. *Chem. Ber.* **1956**, *89*, 1185-1195.
77. McFarland, J. W. *J. Org. Chem.* **1963**, *28*, 2179-2181.
78. Barluenga, J.; Aznar, F.; Liz, R. *J. Chem. Soc. Chem. Commun.* **1981**, 1181-1182.
79. Garigipati, R. S. *Tetrahedron Lett.* **1990**, *31*, 1969-1972.
80. Norton, J. R.; Tunge, J. A.; Czerwinski, C. J.; Gately, D. A. *Organometallics* **2001**, *20*, 254-260.
81. Katritzky, A. R.; Button, M. A. C.; Busont, S. *J. Org. Chem.* **2001**, *66*, 2865-2868.
82. Keung, W.; Bakir, F.; Patron, A. P.; Rogers, D.; Priest, C. D.; Darmohusodo, V. *Tetrahedron Lett.* **2004**, *45*, 733-737.

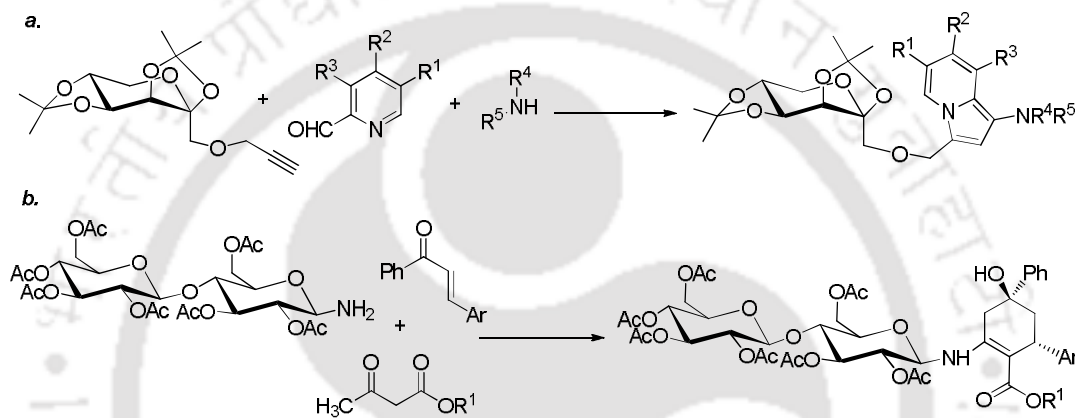
➤ Conclusion and Future Perspectives

Over all, the thesis describes a new methodology for the synthesis of 2-deoxy-2-iodo-*O*-glycosylation using *in situ* generated iodonium ion catalyzed by oxobis(acetylacetonato) vanadium. Furthermore, we have explored BDMS as a catalyst, in carbohydrate chemistry and in multicomponent reaction to deliver a new methodology for the synthesis of 2,3-unsaturated-*O*-glycoside, imidazo[1,2-*a*]pyridine and α -aminoamidine. Besides that, we believe 2-deoxy-2-iodo-*O*-glycoside approach can be extensively utilized for target-oriented synthesis of natural products in modern organic synthesis. Indeed, 2,3-unsaturated-*O*-glycoside were further used as a synthetic intermediate for targeted complex natural products and as asymmetric catalyst in near future. Moreover, the synthesized imidazo[1,2-*a*]pyridine and α -aminoamidine can be explored for biological studies.



Scheme 1

The knowledge gathered from the output of 2-iodo-*O*-glycosylation synthesis spurred us to work on target-oriented synthesis. In future, we will try to achieve the total synthesis of Rhynchosporosides from cellobiose through *O*-glycosylation. In addition to that, we will try to achieve the target NodRm-IV from glucal and their retrosynthetic representation is shown in Scheme 1. Synthetic chemists are keen to generate sugar-based library of molecules for drug design in near future. In this regard the future endeavor would be to achieve the sugar based multicomponent reaction as shown in Scheme 2.



Scheme 2

● *List of Publications*

1. '*VO(acac)₃/H₂O₂/NaI: a mild and efficient combination for the cleavage of dithio acetal derivatives of sugars*' Abu T. Khan*, Shahzad Ali, **R. Sidick Basha**, Md. Musawwer Khan and Mohan Lal, *Carbohydr. Res.* **2011**, 346, 2629-2632.
2. '*Bromodimethylsulfonium bromide (BDMS) catalyzed synthesis of imidazo[1,2-a]pyridine derivatives and their fluorescence properties*' Abu T. Khan*, **R. Sidick Basha** and Mohan Lal, *Tetrahedron Lett.* **2012**, 53, 2211-2217.
3. '*Formation of unexpected α -amino amidine through three-component 'UGI condensation reaction*' Abu T. Khan*, **R. Sidick Basha**, Mohan Lal and Mohammad Hedayetullah Mir, *RSC Adv.*, **2012**, 2, 5506-5509.
4. '*Synthesis of tetra-substituted pyrroles, a potential phosphodiesterase 4B inhibitor, through nickel(II) chloride hexahydrate catalyzed one-pot four-component reaction*' Abu T. Khan*, Mohan Lal, Prasanta Ray Bagdi, **R. Sidick Basha**, Parameswaran Saravanan and Sanjukta Patra, *Tetrahedron Lett.* **2012**, 53, 4145-4150. (Recognized as one of the Top 25 most downloaded article July'12-Sep'12).
5. '*Bromodimethylsulfonium bromide (BDMS) catalyzed synthesis of 2,3-unsaturated-O-glycosides via Ferrier rearrangement*' Abu T. Khan*, **R. Sidick Basha** and Mohan Lal, *ARKIVOC* **2013** (ii) 201-212. (Special Issue in honor of Prof Richard R. Schmidt).
6. '*Synthesis of Trisubstituted 1H-Pyrazole-4-carbodithioates in a One-Pot Three-Component Reaction Catalyzed by Ferric Sulfate*' Abu T. Khan*, Arindam Ghosh, **R. Sidick Basha** and Mohammad Hedayetullah Mir, *Asian J. Org. Chem.* **2013**, 2, 126-129. (Featured as editor choice spotlight article).

-
7. 'Regio- and diastereoselective synthesis of trans-2,3-dihydrofuran derivatives in an Aqueous medium' Abu T. Khan*, Mohan Lal and **R. Sidick Basha**, *Synthesis* **2013**, *45*, 406-412.
 8. 'Cobalt triflate catalyzed one-pot Synthesis of fluorophore 1,4-Dihydropyridine derivatives via Hantzsch Reaction' Prasanta Ray Bagdi, **R. Sidick Basha**, Mohan Lal, Rajkumar Misra and Abu T. Khan,* *J. Indian Chem. Soc.*, **2013**, *90*, 1589-1598. (Special Issue in honor of Prof. Sunil Kumar Talapatra).
 9. 'Bromodimethylsulfonium bromide (BDMS)-catalyzed synthesis of substituted pyrroles through a one-pot four-component reaction' Prasanta Ray Bagdi, **R. Sidick Basha**, Mohan Lal, and Abu T. Khan,* *Chem. Lett.* **2013**, *42*, 939-941.
 10. '2,6-Pyridinedicarboxylic acid as organocatalyst for the synthesis of 1,5-benzodiazepines through one-pot reaction' Mohan Lal, **R. Sidick Basha**, Satavisha Sarkar and Abu T. Khan,* *Tetrahedron Lett.* **2013**, *54*, 4264-4272.
 11. 'Copper oxide nanoparticle mediated 'click chemistry' for the synthesis of mono-bis- and tris-triazole derivatives from 10,10-dipropargyl-9-anthrone as a key building block' Prasanta Ray Bagdi, **R. Sidick Basha**, Pranjal Kumar Baruah and Abu T. Khan,* *RSC Adv.*, **2014**, *4*, 10652-10659.
 12. 'Aqueous Biphasic Mediated Stereoselective Synthesis of 2-deoxy-2-iodo-O-glycoside through in situ generated Iodonium ion Catalyzed by Oxobis(acetyl acetonato)vanadium' **R. Sidick Basha** and Abu T. Khan* (Manuscript under preparation).
 13. 'Copper oxide nanoparticle catalyzed synthesis of 3-substituted-2-(1-alkyl-1,2,3-triazol-4-yl)-imidazo[1,2-a]pyridine derivatives in one-pot three component reaction' Prasanta Ray Bagdi, **R. Sidick Basha** and Abu T. Khan* (Communicated).

14. 'Synthesis of 1,4-di-substituted bis-triazole derivatives in one-pot three component reaction catalyzed by Copper oxide nanoparticle' Prasanta Ray Bagdi, **R. Sidick Basha** and Abu T. Khan* (Manuscript under preparation).

15. 'Synthesis and characterization of Symmetrical and Unsymmetrical Vanadium salen complexes' **R. Sidick Basha**, Prasanta Ray Bagdi and Abu T. Khan* (Manuscript under preparation).

