

Newer Strategies for the Synthesis of Mono- and Bicyclic Oxygen Heterocycles

*A Dissertation Submitted to the
Indian Institute of Technology Guwahati
As Partial Fulfillment of the Requirements
for the Degree of*

Doctor of Philosophy



Submitted by

Pipas Saha

**Department of Chemistry
Indian Institute of Technology Guwahati
Guwahati 781039
August 2012**

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Dedicated

To

My Family Members



INDIAN INSTITUTE OF TECHNOLOGY GUWAHATI

Department of Chemistry

STATEMENT

I hereby declare that the matter embodied in this thesis entitled “*Newer Strategies for the Synthesis of Mono- and Bicyclic Oxygen Heterocycles*” is the result of investigations carried out by me in the Department of Chemistry, Indian Institute of Technology Guwahati, India under the supervision of Prof. Anil K. Saikia.

In keeping with the general practice of reporting scientific observations, due acknowledgement has been made wherever the work described is based on the findings of other investigators.

August, 2012
IIT Guwahati

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Dr. Anil K. Saikia
Professor

CERTIFICATE

This is to certify that Mr. Pipas Saha has been working under my supervision since July 2008. I am forwarding his thesis entitled “*Newer Strategies for the Synthesis of Mono- and Bicyclic Oxygen Heterocycles*” being submitted for the Ph.D. degree of this institute. I certify that he has fulfilled all the requirements according to the rules of this institute, and regarding the investigations embodied in his thesis and this work has not been submitted elsewhere for a degree.

August, 2012
IIT Guwahati

Prof. Anil K. Saikia
Supervisor

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I would like to acknowledge my sincere gratitude to all my doctoral committee members, Prof. T. Punniyamurthy, Dr. Bhubaneswar Mandal and Dr. V. V. Dasu for their insightful advices and valuable suggestions. My honest regards to all the faculty of the department of chemistry for their motivation and encouragement.

I wish to acknowledge IIT Guwahati for my research fellowship and also for all the facilities that were made available to me. I also thank Central Instruments Facility (CIF), IIT Guwahati for providing the instruments facility and DST-FIST for providing the X-ray facility.

I owe my sincere thanks to my lab mates Dr. U. C. Reddy, Mr. Paramartha Gogoi, Ms. Anupa Majumdar, Mr. Anup Bhunia, Ms. Priya Ghosh, Ms. Sabera Sultana and Mr. Manashjyoti Deka for their help, suggestions and encouragement during my research period. I extend my sincere thanks to other friends in the department Dr. Prasenjit Saha, Dr. W. Marjit Singh, Dr. Subash Chandra Sahoo, Dr. Shahzad Ali, Mr. Murali M. Guru, Mr. S. Sakthivel, Mr. Sandeep Kumar Dey and Mr. Pankaj Kumar.

I also take this opportunity to thank all of my Ph.D. batchmates (July, 08), the other research scholars in the chemistry department and all my IITG friends, who have shared their thoughts and views with me.

Finally, my deepest gratitude goes to my family for their unflagging love and support throughout my life. I feel deeply indebted to them for whatever I have achieved so far.

Pipas Saha

LIST OF ABBREVIATIONS

Ac	acetyl	NMR	nuclear magnetic resonance
AIBN	azobisisobutyronitrile	NOESY	nuclear overhauser enhancement spectroscopy
Bn	benzyl	ORTEP	oak ridge thermal ellipsoid plot
BOX	bisoxazoline	Ph	phenyl
Bz	benzoyl	PMA	phosphomolybdic acid
CCDC	cambridge crystallographic data centre	Pr	propyl
COSY	correlation spectroscopy	RCM	ring-closing metathesis
DBMP	2,6-di- <i>tert</i> -butyl-4-methylpyridine	r.t.	room temperature
DCE	1,1-dichloroethane	SEM	trimethylsilyl ethoxymethyl
DCM	dichloromethane	TBDMS	<i>tert</i> -butyldimethylsilyl
DTBMP	2,6-di- <i>tert</i> -butyl-4-methylpyridine	TBDPS	<i>tert</i> -butyldiphenylsilyl
dr	diastereomeric ratio	THF	tetrahydrofuran
ee	enantiomeric excess	TLC	thin layer chromatography
HRMS	high resolution mass spectrometry	TFA	trifluoroacetic acid
IR	infrared	TMS	trimethylsilyl
LDA	lithium diisopropylamide	Tf	trifluoromethanesulfonyl
MEM	methoxyethoxymethyl	<i>p</i> -TsOH	<i>p</i> -toluenesulfonic acid
MOM	methoxymethyl	Ts	<i>p</i> -toluenesulfonyl
M.P.	melting point	XRD	x-ray diffraction
MS	molecular sieves	m/z	mass to charge ratio

Abbreviations for intensities of ¹H-NMR signals

s	singlet	td	triplet of doublets
d	doublet	m	multiplet
dd	doublet of doublets	brs	broad singlet
ddd	doublet of doublet of doublets	Hz	hertz
t	triplet	MHz	mega-hertz

Abstract

The thesis contains six chapters. The first chapter provides a brief introduction about mono- and bicyclic oxygen heterocycles and oxonium-ene cyclization reactions. Special emphasis has been given to six membered oxygen heterocycles and oxabicyclic compounds. The second chapter gives a description about the novel synthesis of oxabicyclo[3.3.1]nonanone via (3,5)-oxonium-ene reaction. The third chapter describes about the synthesis of oxabicyclo[3.3.1]nonenes and substituted tetrahydropyrans via (3,5)-oxonium-ene reaction. In the fourth chapter there is a description about the stereoselective synthesis of 6-oxabicyclo[3.2.1]octene via (3,5)-oxonium-ene reaction. The fifth chapter presents the synthesis of 2,3,5,6-tetrasubstituted tetrahydropyrans via (3,5)-oxonium-ene reaction. The sixth chapter describes about the diastereoselective synthesis of substituted dihydropyrans via oxonium-ene cyclization reaction.

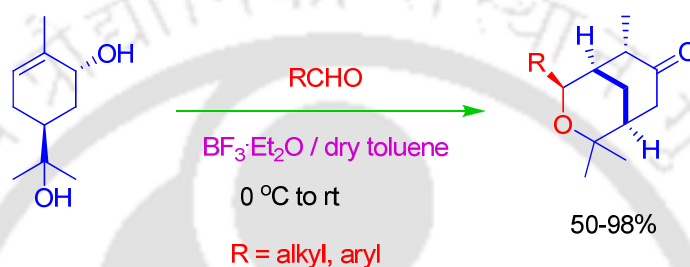
Chapter 1. Introduction to Mono- and Bicyclic Oxygen Heterocycles and Oxonium-Ene Cyclization Reactions

Heterocyclic compounds are widely distributed in nature and are essential to life. Almost all the drugs, most vitamins and many other natural products are heterocycles. They are key components of pharmaceutical chemistry and therefore, they have been a fruitful source of inspiration for the design of structural analogues to be used as pharmacological tools as well as new drugs. The six membered ring heterocycles of oxygen *i.e.*, dihydropyrans (DHPs) and tetrahydropyrans (THPs) are structural motifs of many biologically active natural products such as polyether antibiotics, marine toxins and pheromones. On the other hand oxabicyclic compounds are important due to their potential biological properties. Oxabicyclo[3.3.1]nonenes and selected derivatives are known to behave as estrogen receptor ligands.

The reaction of an alkene with an allylic hydrogen atom (the ene) with another alkene (enophile) to produce a new alkene is known as ene reaction. Ene cyclization that involves the oxonium ion as the enophile component, is called oxonium-ene cyclization reaction. There are three different types of oxonium-ene cyclization reactions, namely, (1,5), (2,5) and (3,5). Although (1,5) and (2,5) oxonium-ene-type cyclizations have been well studied, the (3,5)-oxonium-ene cyclization has not been investigated extensively.

Chapter 2: A Novel Synthesis of Oxabicyclo[3.3.1]nonanone via (3,5)-Oxonium-Ene Reaction

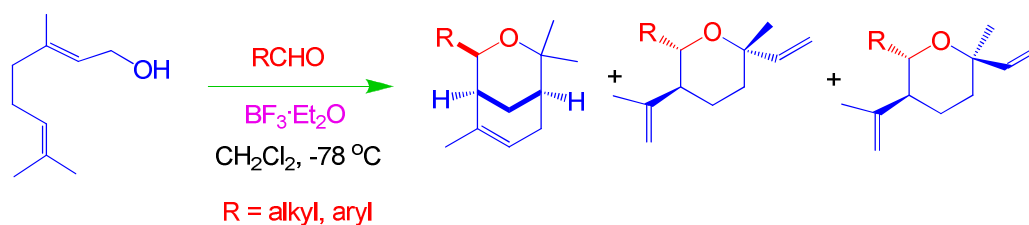
This chapter describes an efficient methodology for the synthesis of oxabicyclo[3.3.1]nonanone via (3,5)-oxonium-ene reaction from the reaction of aldehydes or epoxides with commercially available *trans-p*-menth-6-ene-2,8-diol, mediated by boron trifluoride etherate (Scheme 1). The reaction is diastereoselective and a series of oxabicyclo[3.3.1]nonanones have been synthesized in good yields.



Scheme 1

Chapter 3: Synthesis of Oxabicyclo[3.3.1]nonenes and Substituted Tetrahydropyrans via (3,5)-Oxonium-Ene Reaction

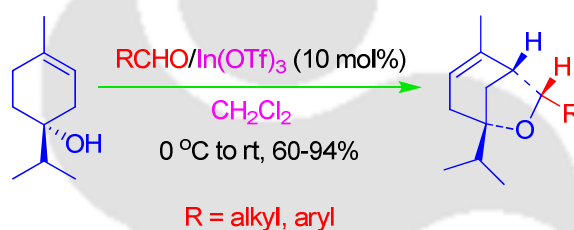
Oxabicyclo[3.3.1]nonenes and selected derivatives are known to behave as estrogen receptor ligands. On the other hand, tetrahydropyrans are found in many biologically active natural products and pharmaceuticals. Alkyl substituted tetrahydropyrans are used as aroma and flavouring substances for pharmaceuticals, cosmetics and food stuff. In this chapter an efficient methodology for the synthesis of oxabicyclo[3.3.1]nonenes and substituted tetrahydropyrans from the reaction of geraniol with aldehydes and epoxides promoted by boron trifluoride etherate is presented. The reaction can be generalized as shown in Scheme 2.



Scheme 2

Chapter 4: Stereoselective Synthesis of 6-Oxabicyclo[3.2.1]octene via (3,5)-Oxonium-Ene Reaction

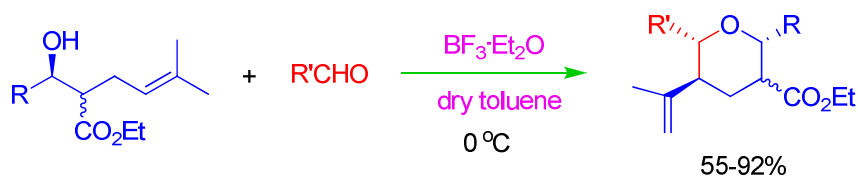
Oxabicyclo[3.2.1]octanes/octenes are important intermediate in organic synthesis for numerous organic transformations. Two different types of oxabicyclo[3.2.1]octane/octene are well known, one is 6-oxabicyclo[3.2.1]octane/octene and another one is 8-oxabicyclo[3.2.1]octane/octene. Although several strategies are developed for the synthesis of 8-oxabicyclo[3.2.1]octane/octene, methodology for the synthesis of 6-oxabicyclo[3.2.1]octane/octene are very limited in the literature. This chapter provides a stereoselective method for the synthesis of 6-oxabicyclo[3.2.1]octene via (3,5)-oxonium-ene reaction from the reaction of (-)-terpinen-4-ol with aldehyde or epoxide catalyzed by indium triflate ($\text{In}(\text{OTf})_3$) in good yields (Scheme 3).



Scheme 3

Chapter 5: Synthesis of 2,3,5,6-Tetrasubstituted Tetrahydropyrans via (3,5)-Oxonium-Ene Reaction

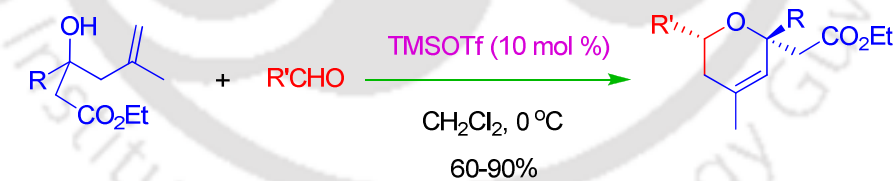
The tetrahydropyrans are important structural features in synthetic organic chemistry as they are core units of many biologically active molecules and natural products. The polyether antibiotics, marine toxins and pheromones belong to such biologically active natural products. The 2,3,5,6-tetrasubstituted tetrahydropyran units are found in antibiotics such as FR-901464 and X-206. The presence of ester group in the tetrahydropyran ring is important because a diverse functionalization could be achieved in the tetrahydropyran molecule. In this chapter an efficient methodology is presented for the synthesis of 2,3,5,6-tetrasubstituted tetrahydropyrans from the reaction of aldehydes and ethyl 2(1-hydroxyalkyl/hydroxy(phenyl)methyl)-5-methylhex-4-enoate using (3,5)-oxonium-ene reaction promoted by boron trifluoride etherate under mild conditions. The reaction is general and a series of substituted tetrahydropyrans have been synthesized in good yields (Scheme 4).



Scheme 4

Chapter 6: Diastereoselective Synthesis of Substituted Dihydropyrans via Oxonium-Ene Cyclization Reaction

Dihydropyrans are important intermediates for the synthesis of biologically active natural products such as ambruticin, phorbaxozoles, leucascandrolide, kendomycin, neopeltolide, clavisolides and diospongins. The olefin functionality of dihydropyrans can be manipulated for the synthesis of polysubstituted tetrahydropyrans. Substituted dihydropyrans are also used as a flavoring or aroma material for food and other products. The 4-methyl substituted dihydropyran unit is present in macrolide natural products laulimalide and okadaic acid. This chapter describes a methodology for the diastereoselective synthesis of substituted dihydropyrans from the reaction of aldehydes or epoxides and ethyl 3-alkyl-3-hydroxy-5-methylhex-5-enoate using oxonium-ene reaction catalyzed by trimethylsilyl trifluoromethanesulfonate (TMSOTf) under mild conditions. The reaction can be generalized as shown in Scheme 5.



Scheme 5

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CHAPTER 1

Introduction to Mono- and Bicyclic Oxygen Heterocycles and Oxonium-Ene Cyclization Reactions

1.1 Introduction to Mono- and Bicyclic Oxygen Heterocycles

Heterocyclic compounds are widely distributed in nature and are essential to life. Almost all the drugs, most vitamins and many other natural products are heterocycles. They are key components of pharmaceutical chemistry and therefore, they have been a fruitful source of inspiration for the design of structural analogues to be used as pharmacological tools as well as new drugs. The six membered ring heterocycles of oxygen *i.e.*, dihydropyrans (DHPs) and tetrahydropyrans (THPs) are structural motifs of many biologically active natural products such as polyether antibiotics, marine toxins and pheromones.¹ Consequently a huge amount of effort has been directed towards developing ever more efficient method to synthesize them. For example, over the last ten years thousands of tetrahydropyran containing compounds have been entered into preclinical and clinical trials.² Alkyl substituted tetrahydropyrans such as **1** (Figure 1.1.1) are used as aroma and flavouring substances for pharmaceuticals, cosmetics and food stuff.³ The macrocycle, (-)-Kendomycin **2** has a diverse and fascinating pharmacological profile. It was isolated in 1996 from

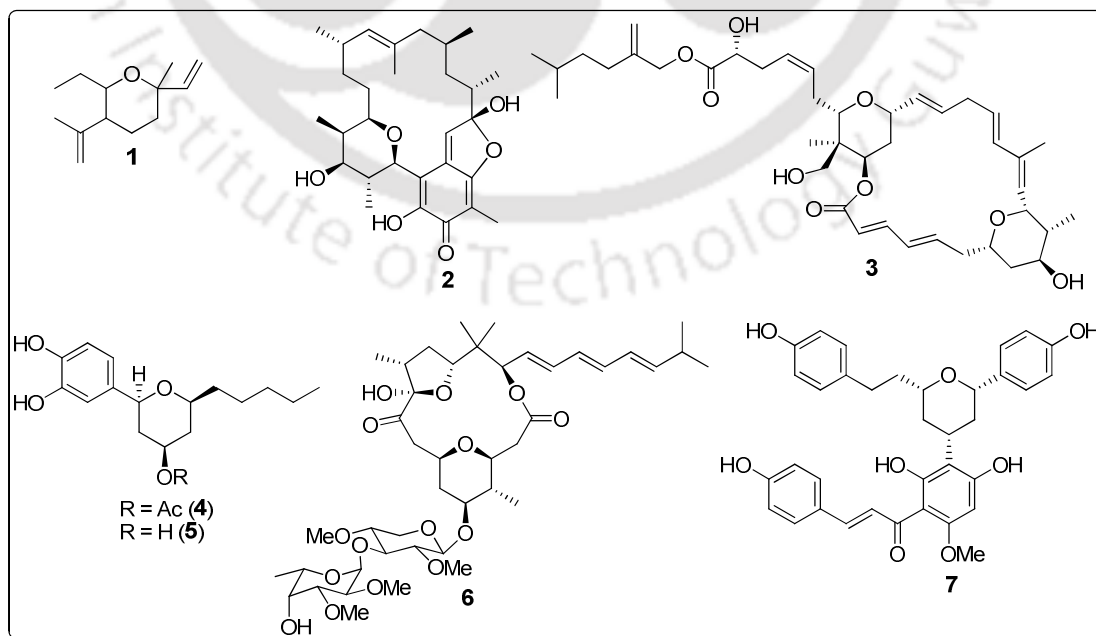


Figure 1.1.1. Some bioactive molecules containing tetrahydropyran ring

Streptomyces bacteria and exhibits potent antagonism of the endothelin receptor agonism.⁴ Lasonolide A, **3** isolated from a shallow water Caribbean sponge, species *Forcepia* shows potent activity against A-549 human lung carcinoma.⁵ Catechols **4** and **5** isolated from extracts of *Plectranthus sylvestris* (labiateae), a plant found in the woody hills in East Africa are potent antioxidants and posses anti-inflammatory properties. Polycavernoside A, **6** a potent marine toxin⁶ isolated from the red algae *Polycavernosa tsudai* also has a tetrahydropyran moiety. Calyxin, a new family of natural products isolated from *Alpinia blepharocalyx* seeds, are used for the treatment of stomach disorders in Chinese medicine.^{2h} Several of these natural products show interesting antiproliferative activity against carcinoma cells. Epicalyxin F, **7** is the most potent member of the class and showed ca. 1 μ M activity against human HT-1080 fibrosarcoma and murine 26-L5 carcinoma. Oxabicyclic compounds are important due to their potential biological properties. Oxabicyclo[3.3.1]nonenes and selected derivatives (Figure 1.1.2, **8** & **9**) are known to behave as estrogen receptor ligands.⁷ Oxabicyclo[3.2.1]octenes are important intermediate in organic synthesis for numerous organic transformations.⁸ The 6- and 8-oxabicyclo[3.2.1]octane/octene skeleton are present in biologically active natural products

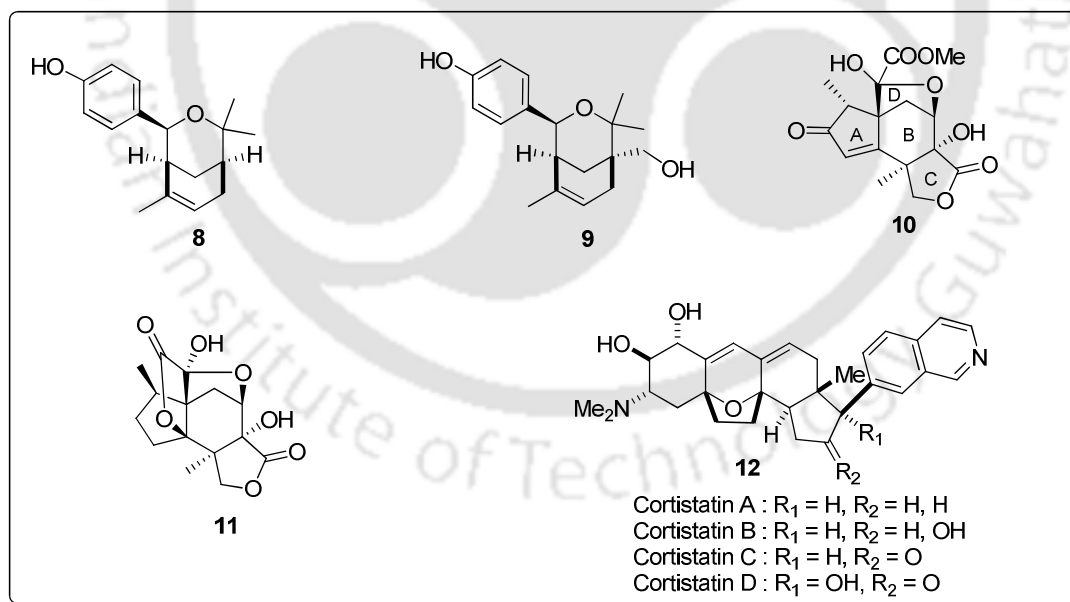


Figure 1.1.2. Biologically active oxabicyclic compounds

like Jiadifenin **10**, Jiadifenolide **11** and Cortistatin **12**.⁹ Although there are various natural and synthetic oxabicyclic compounds, this thesis will discuss efficient methodologies for the synthesis of oxabicyclo[3.3.1]nonanones, oxabicyclo[3.3.1]nonenes and 6-oxabicyclo[3.2.1]octenes.

1.2 General Approaches for the Synthesis of Dihydro- and Tetrahydropyrans

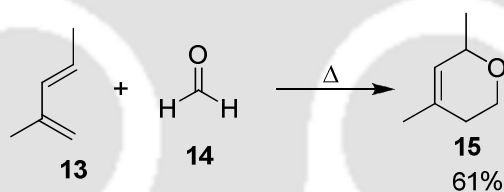
Dihydro- and Tetrahydropyrans are generally prepared by

- i) Hetero Diels-Alder reaction
- ii) Manipulation of carbohydrates
- iii) Intramolecular Michael reactions
- iv) Prins cyclization

A brief summary of different applications involving these four types of reactions are discussed below.

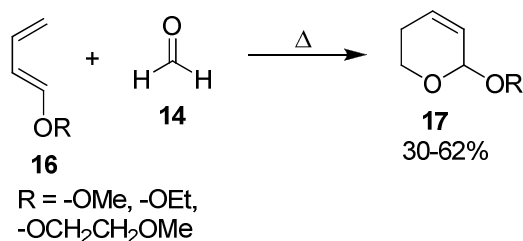
1.2.1 Hetero Diels-Alder Reaction

Hetero Diels-Alder reaction was first reported by Gresham and Steadman in 1951.¹⁰ In their report they have prepared 2,4-dimethyl-5,6-dihydro-1,2-pyran **15** by reacting methylpentadiene **13** with formaldehyde **14** (Scheme 1.2.1.1). This was an unusual reaction as in this case formaldehyde was acting as a dienophile.



Scheme 1.2.1.1.

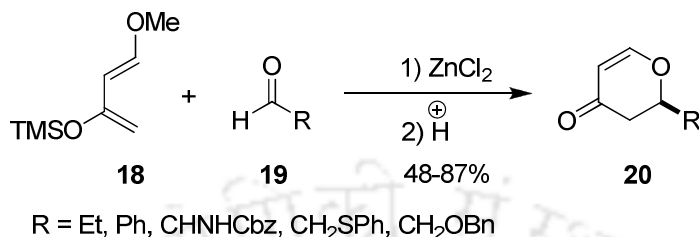
In 1962 Kubler reported that an alkoxy group on the 1-position of a 1,3-alkadiene provides sufficient activation to permit the Diels-Alder reaction with paraformaldehyde.¹¹ The reaction products are 2-alkoxy-5,6-dihydro-2H-pyrans **17** (Scheme 1.2.1.2). Later on it has been established by others¹² that electron donating group on diene increase the reactivity and regioselectivity of hetero Diels-Alder reaction to produce substituted dihydropyrans



Scheme 1.2.1.2.

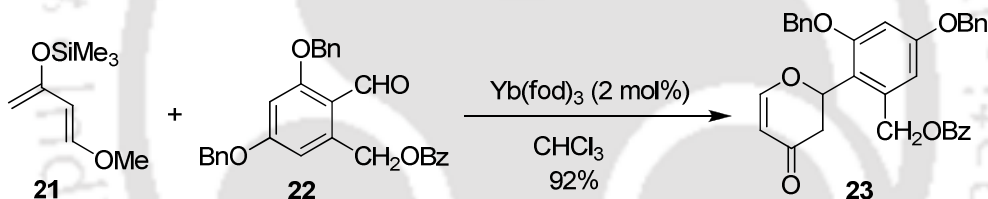
(DHPs) in good yields.

In 1982 Danishefsky reported the Lewis acid catalyzed hetero Diels-Alder reaction for the synthesis of substituted dihydropyrans (DHPs) in good yields (Scheme 1.2.1.3).¹³



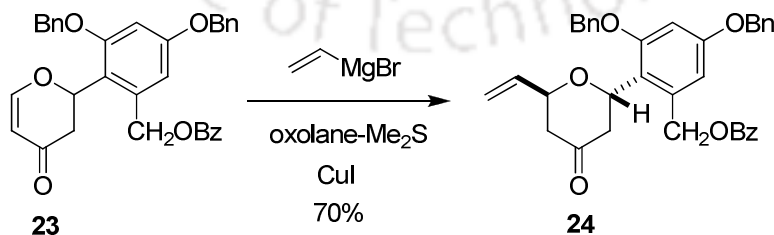
Scheme 1.2.1.3.

Lewis acid catalyzed hetero Diels-Alder reaction of (*E*)-1-methoxy-3-trimethylsilyloxy-1,3-butadiene **21** with 6-benzoyloxymethyl-2,4-dibenzyloxy benzaldehyde **22** afforded 2-(6-benzoyloxymethyl-2,4-dibenzyloxyphenyl)-2,3-dihydro-4*H*-pyran-4-one **23** in good yields (Scheme 1.2.1.4). The product **23** on treatment with vinylmagnesium bromide in oxolane-



Scheme 1.2.1.4.

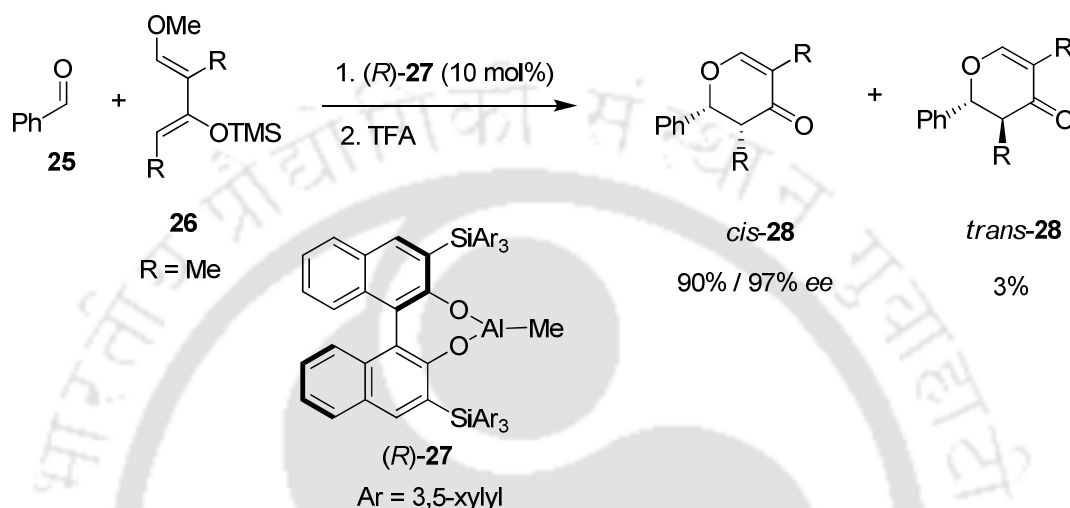
dimethylsulfide with cuprous iodide produces tetrahydropyrene **24** in 70% yield (Scheme 1.2.1.5). The presence of vinyl group at the C-6 position of the tetrahydropyrene **24** increases its synthetic utility as the vinyl group is a precursor of hydroxymethyl function (*i.e.*, C-6 of the C-1 arylated D-glycosyl residue).¹⁴



Scheme 1.2.1.5.

The first reliable and highly efficient chiral aluminium(III) catalyst for HDA reactions of aldehydes was reported by Yamamoto and co-workers.¹⁵ The use of the chiral BINOL-

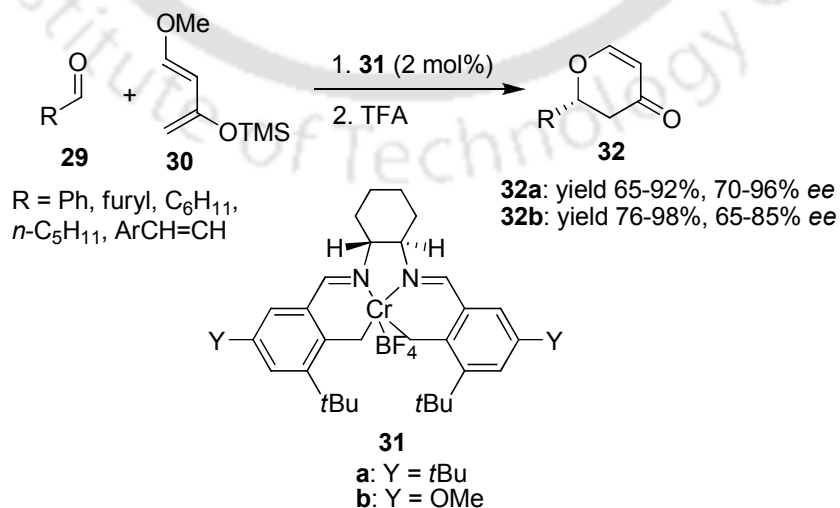
AlMe complex (*R*)-**27** was found to be highly effective for the HDA reaction of various aldehydes with activated dienes. For example, the reaction of benzaldehyde **25** with *trans*-1-methoxy-2-methyl-3-(trimethylsilyloxy)-1,3-pentadiene **26**, affords *cis*-dihydropyrones *cis*-**28** as the major product in high yield with up to 97% *ee* (Scheme 1.2.1.6). The reaction proceeds well in nonpolar solvents such as toluene.



Scheme 1.2.1.6.

Different chiral transition- and lanthanide metal complexes can catalyze the HDA reaction of unactivated and activated aldehydes with especially activated dienes.¹⁶

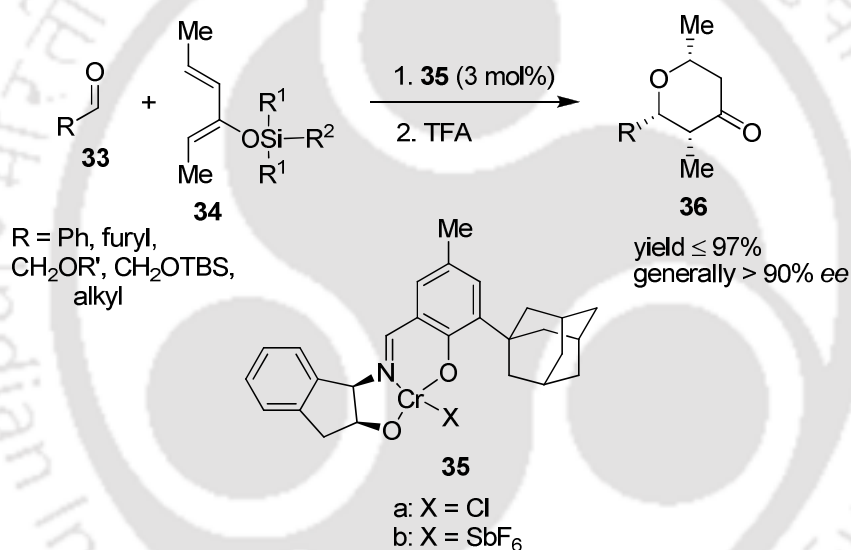
Jacobsen and co-workers¹⁷ have shown that the chiral salen-chromium(III) complexes **31a** and **31b** can catalyze the HDA reaction of different aldehydes **29** containing aromatic, aliphatic and conjugated substituents with diene **30** (Scheme 1.2.1.7). The reaction proceeds



Scheme 1.2.1.7.

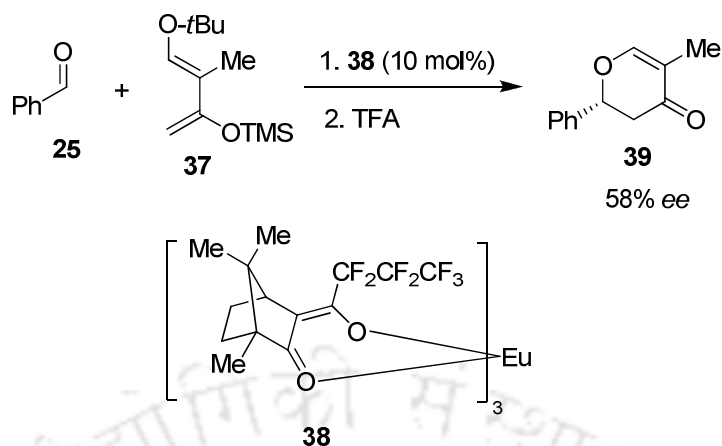
in good yield (up to 98%) and 62-93% *ee*. It was found that the presence of oven-dried powdered 4 Å molecular sieves (MS) led to an increased yield and enantioselectivity.

A more general catalytic enantioselective HDA reaction has also been achieved by Jacobsen and co-workers by the introduction of chiral tridentate Schiff base chromium(III) complexes **35** (Scheme 1.2.1.8).¹⁸ These complexes, which are highly diastereo- and enantioselective catalysts for the reaction of unactivated aldehydes, can catalyze the reaction of less nucleophilic dienes bearing fewer than two oxygen substituents. The adamantly-substituted catalysts **35a** and **35b** gave the best results, and both aliphatic and aromatic aldehydes underwent HDA reactions. It was found that use of the hexafluoroantimonate chromium catalyst **35b** resulted in a faster and more enantioselective reaction and that the reaction can proceed without solvent.



Scheme 1.2.1.8.

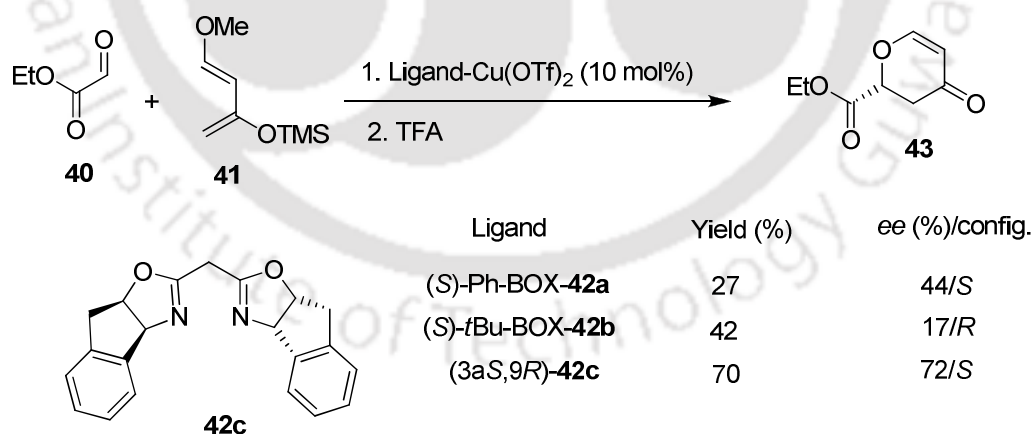
Danishefsky and co-workers were probably the first to observe that lanthanide complexes can catalyze the HDA reaction of aldehydes with activated dienes.¹⁹ The reaction of benzaldehyde **25** with activated conjugated dienes such as **37** was found to be catalyzed by [Eu(hfc)₃] **38** (hfc = 3-(heptafluoropropylhydroxymethylene)camphorate) giving up to 58% *ee* (Scheme 1.2.1.9).



Scheme 1.2.1.9.

In 1995, it was demonstrated that chiral C_2 -symmetric bisoxazoline-copper(II) complexes²⁰ are efficient catalysts for the HDA reactions.

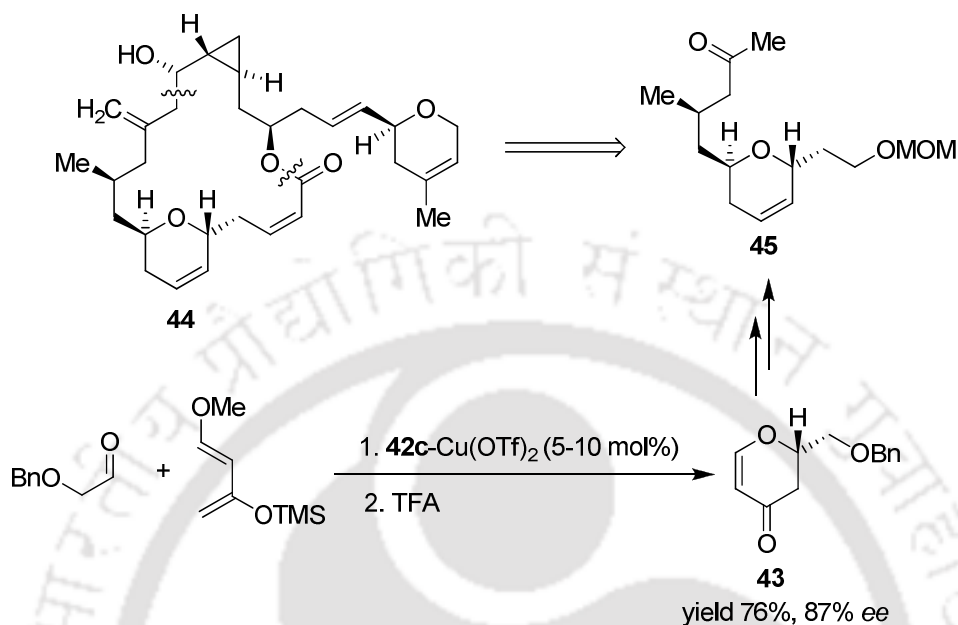
Ghosh *et al.* have investigated the HDA reaction catalyzed by chiral BOX-copper(II) complexes between ethyl glyoxylate **40** and Danishefsky's diene **41** by applying catalyst systems derived from $\text{Cu}(\text{OTf})_2$ and ligands (*S*)-Ph-BOX (*S*)-**42a**, (*S*)-*t*Bu-BOX (*S*)-**42b**, and the conformationally constrained BOX ligand **42c** in order to compare the properties of the latter ligand with the two others (Scheme 1.2.1.10).²¹ The HDA adduct **43** was obtained with a significant improvement compared with the two other catalysts.



Scheme 1.2.1.10.

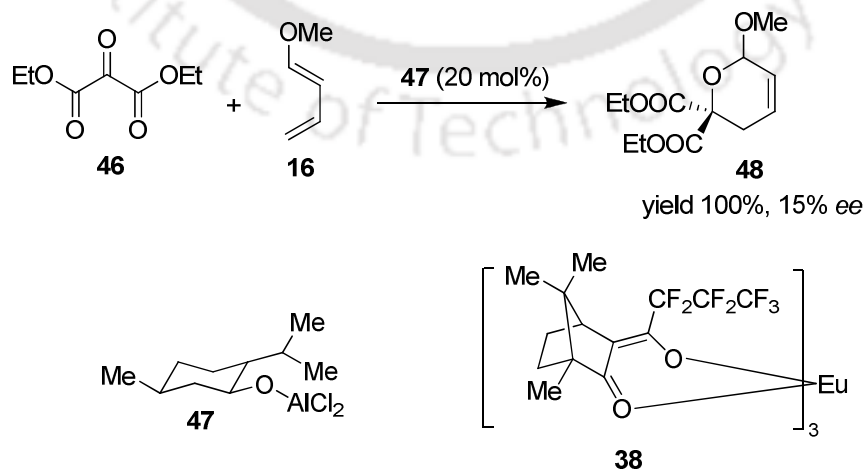
The methodology in Scheme 1.2.1.10 has been used for the synthesis of the C₃-C₁₄ segment **45** of the antitumor agent laulimalide **44** (Scheme 1.2.1.11).²² The constrained chiral BOX ligand **42c** in combination with $\text{Cu}(\text{OTf})_2$ afforded dihydropyrene **43** by a HDA reaction in

good yield and high enantiomeric excess, which was converted to the C₃-C₁₄ segment **45** by a Ferrier type of rearrangement in several steps.



Scheme 1.2.1.11.

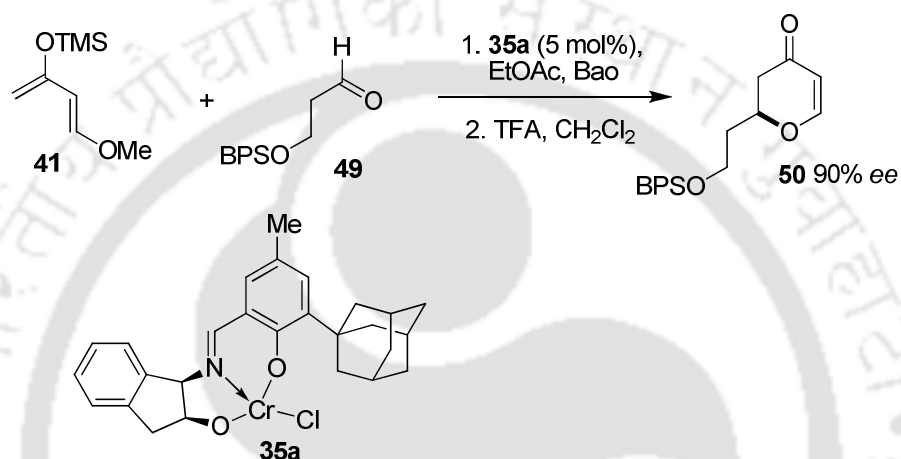
Ketones are generally less reactive than aldehydes. Therefore, the HDA reaction of ketones should be expected to be more difficult to achieve. This is well reflected in the few reported catalytic enantioselective HDA reactions of ketones compared with the many successful examples on the enantioselective reaction of aldehydes. Jankowski *et al.* have reported the reaction of Ketomalonate **46** with 1-methoxy-1,3-butadiene **16** using the menthoxyaluminium catalyst **47** and chiral lanthanide catalysts **38**. The highest enantio-



Scheme 1.2.1.12.

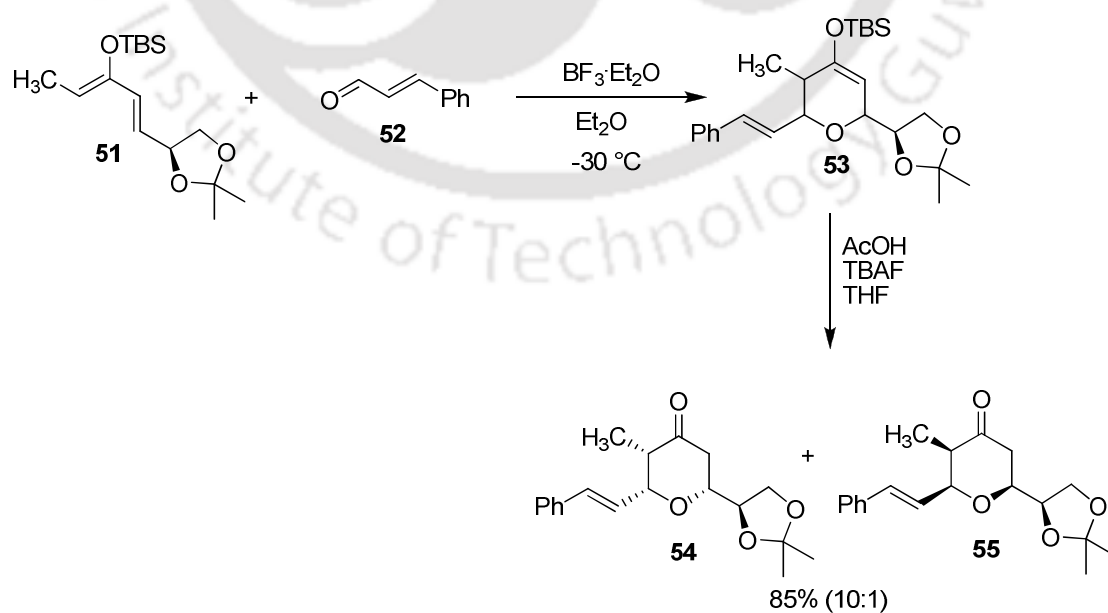
selectivity of 15% *ee* for the HDA adduct **48** was achieved by using **47** as the catalyst (Scheme 1.2.1.12).²³

Paterson *et al.* have reported the synthesis 2,3-dihydro- γ -pyrone (**50**) via a HDA reaction between Danishefsky's diene **41** and aldehyde **49** using Jacobsen's HDA catalyst **35a** (Scheme 1.2.1.13).²⁴ The product was obtained in 89% yield with 90% *ee*. The 2,3-dihydro- γ -pyrone **50** was recognized as being a suitable building block for the synthesis of C(1–14) fragment of Phorboxazoles A and B.²⁴



Scheme 1.2.1.13.

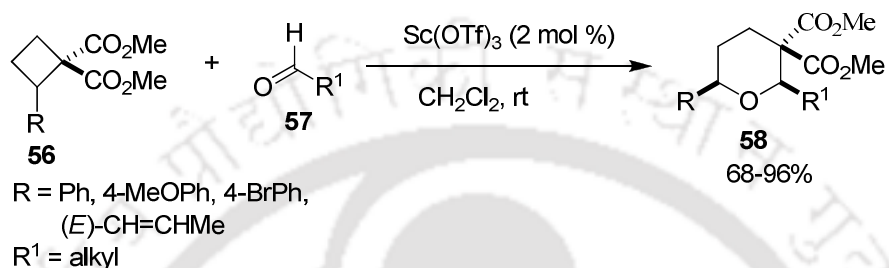
Wessjohann and co-workers have reported the synthesis of all *cis* 2,3,6-trisubstituted tetrahydro- γ -pyrones using a HDA reaction between enantiomerically pure chiral dienes and



Scheme 1.2.1.14.

α,β -unsaturated aldehydes (Scheme 1.2.1.14).²⁵

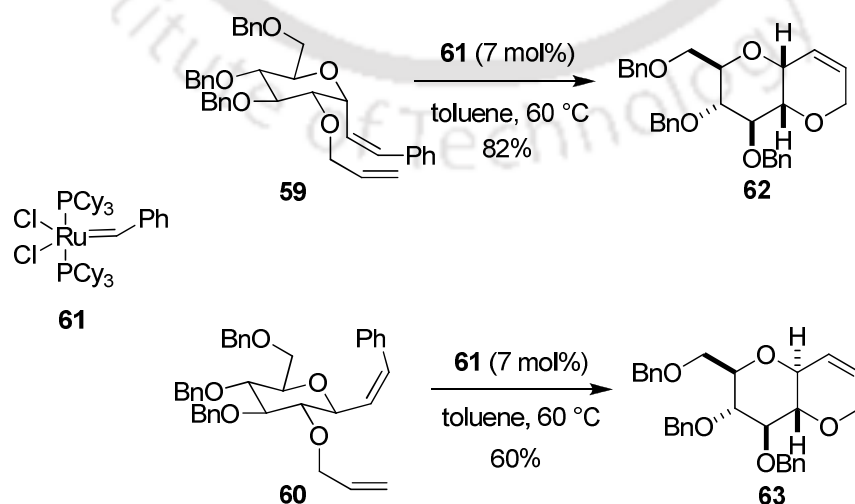
Recently Johnson and co-workers have reported the synthesis of *cis*-2,6-disubstituted tetrahydropyran derivatives via a HDA reaction between malonate derived cyclobutanes and aldehydes catalyzed by Lewis acid (Scheme 1.2.1.15).²⁶ The resulting THP products are of interest because of their prevalence in biologically relevant and structurally interesting molecules.²⁷



Scheme 1.2.1.15.

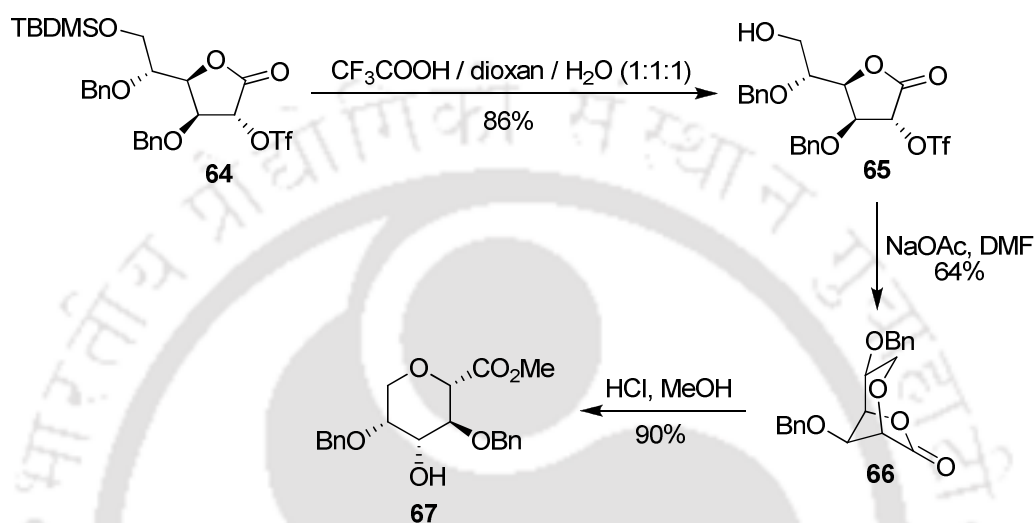
1.2.2 Manipulation of Carbohydrates

Carbohydrates are widely recognized as versatile building blocks in synthetic organic chemistry due to the wealth of functional, conformational and stereochemical information. The accessibility of carbohydrates in the acyclic (open chains) or cyclic form (five- or six membered rings) not only allows specific manipulations of the individual asymmetric centers, but also at either of the extremities (*i.e.* the reducing or non-reducing end). For example, van Boom and co-workers have reported the synthesis of *cis*- and *trans*-fused pyranopyrans **62** and **63** by performing RCM on the corresponding *cis*- and *trans*-1-vinyl-2-*O*-allyl systems **59** and **60** respectively (Scheme 1.2.2.1).²⁷

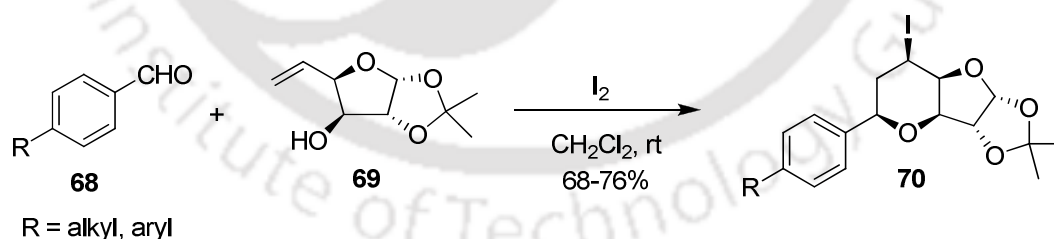


Scheme 1.2.2.1.

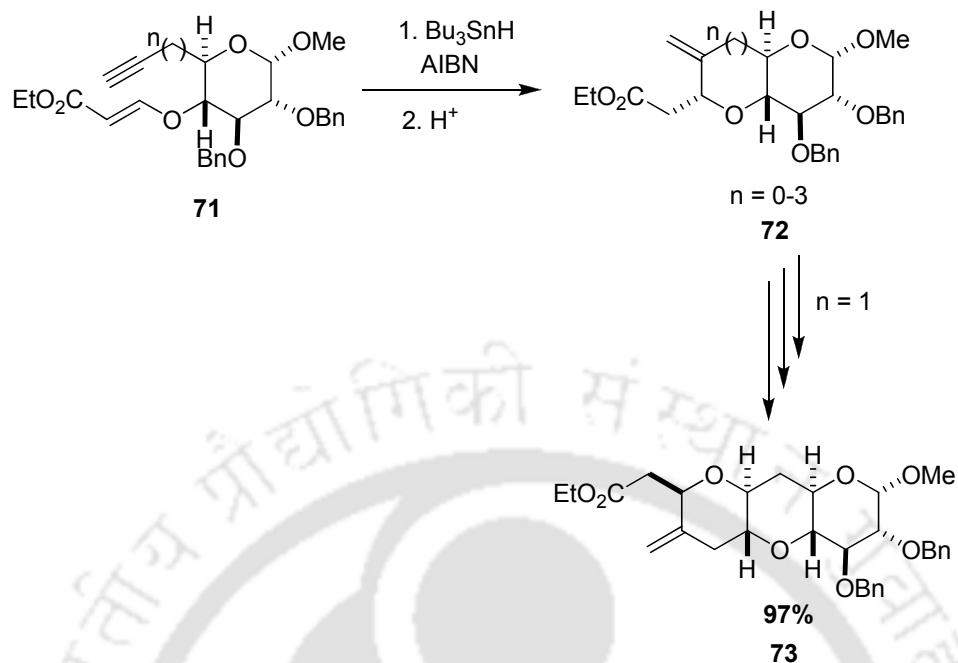
Estevez *et al.* have reported the synthesis of polysubstituted tetrahydropyran **67** starting from gluco triflate **64**. Treatment of gluco triflate **64** with CF_3COOH in dioxan produces gluco triflate **65**, with a free OH group at C-6. The triflate **65** on treatment with NaOAc in DMF gives bicyclic tetrahydropyran **66**. The bicyclic material **66** undergoes ring opening with 1% HCl in MeOH to give methyl ester **67** in 90% yield (Scheme 1.2.2.2).²⁸



Yadav *et al.* have reported the synthesis of 7-iodofurano[3,2-*b*]pyrans **70** from the reaction of secondary homoallylic alcohol derived from D-glucose **69** with aldehydes **68** in the presence of molecular iodine (Scheme 1.2.2.3).²⁹



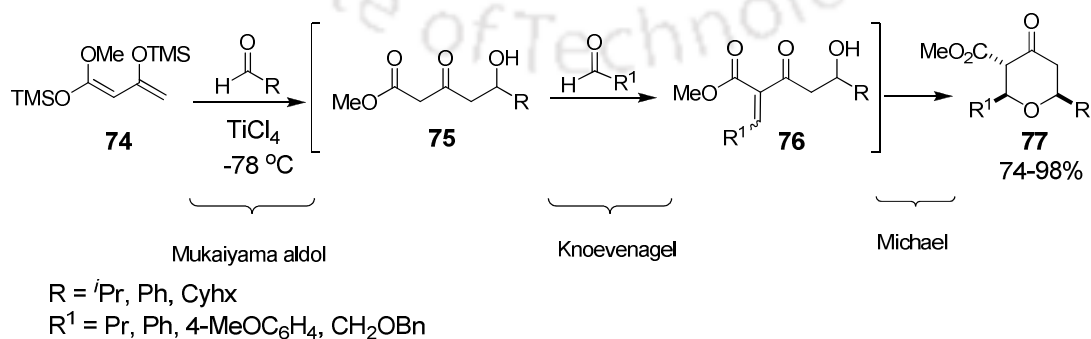
van Boom and co-workers have reported tributyltin radical mediated cyclization of carbohydrate-derived β -(alkynyloxy)acrylates **71** to highly functionalized *cis*- and *trans*-fused bicyclic ethers **73** of various ring sizes (Scheme 1.2.2.4).³⁰



Scheme 1.2.2.4.

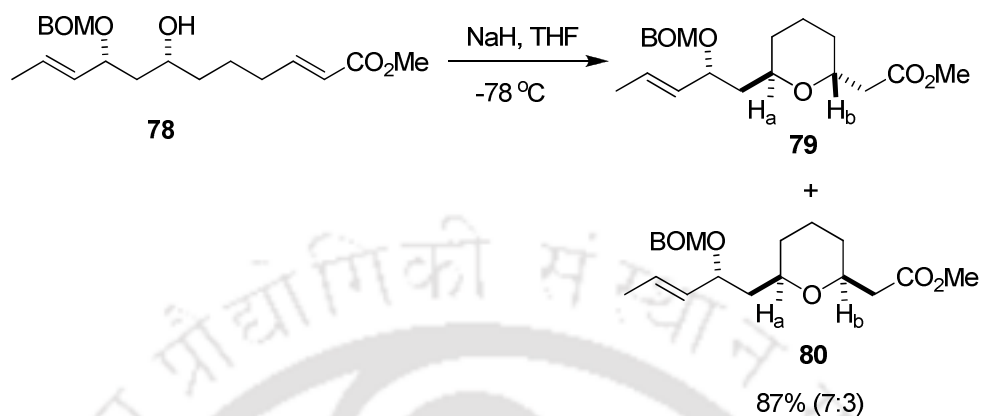
1.2.3 Intramolecular Michael Reaction

The intramolecular Michael addition is a facile method to construct a heterocyclic ring and has recently been used for the synthesis of tetrahydropyran-4-ones. For example, Clarke *et al.* have reported the synthesis of highly substituted tetrahydropyran-4-ones **77** starting from bis-trimethylsilyl enol ether **74** of methyl acetoacetate. Treatment of **74** with Lewis acid in the presence of an aldehyde generates aldol product **75**, which undergoes Lewis acid catalyzed Knoevenagel condensation with a second equivalent of a different aldehyde or ketone to furnish **76**. Finally Lewis acid catalyzed intramolecular oxy-Michael reaction produces tetrahydropyran-4-ones **77** (Scheme 1.2.3.1).³¹



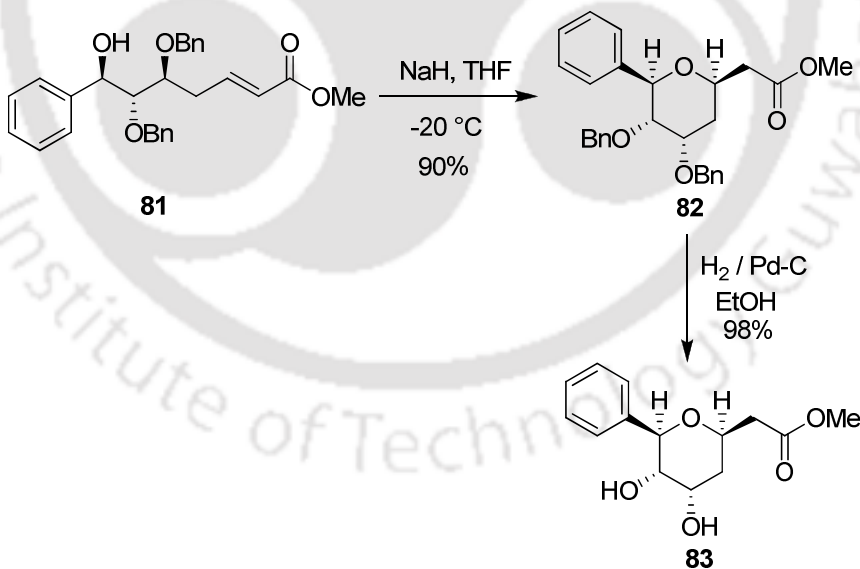
Scheme 1.2.3.1.

Bates *et al.* have reported the synthesis of tetrahydropyrans **79** and **80** in good yields starting from Michael precursor **78** (Scheme 1.2.3.2).³²



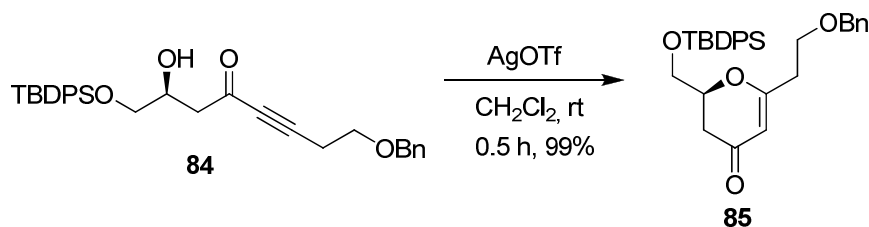
Scheme 1.2.3.2.

Fadnavis and co-workers have reported the synthesis of Goniotaldesdiol A, **83** a natural product containing 2,3,4,6-tetrasubstituted tetrahydropyran ring from precursor **81** by using base-catalyzed intramolecular oxa-Michael addition reaction followed by debenzoylation with Pd-C (Scheme 1.2.3.3).³³



Scheme 1.2.3.3.

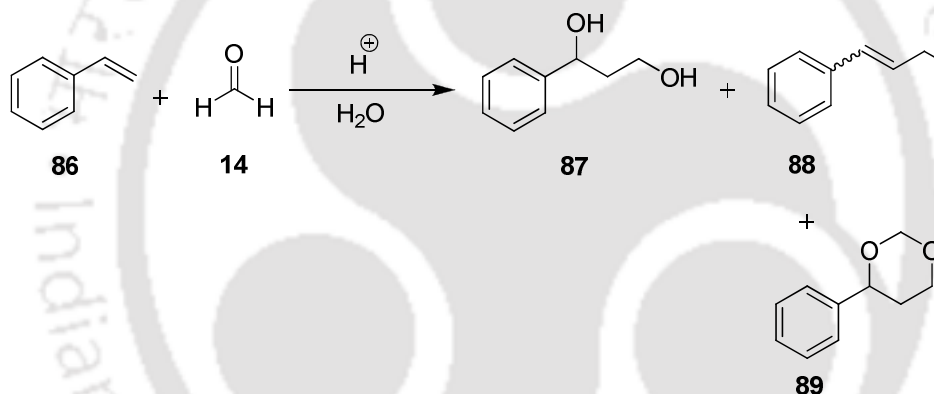
Reddy *et al.* have reported the synthesis of pyranone **85** via an intramolecular oxa-Michael addition of β -hydroxyynone **84** (Scheme 1.2.3.4). Here the pyranone **85** is an important intermediate for the synthesis of natural products (-)-Dactylolide and (-)-Zampanolide.³⁴



Scheme 1.2.3.4.

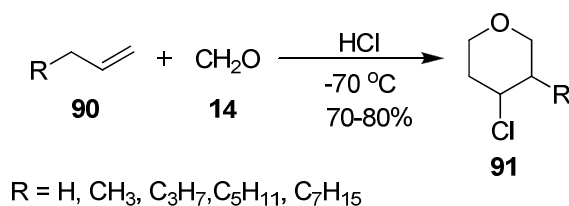
1.2.4 Prins Cyclization

Prins reaction was first reported by H. J. Prins in the year 1919.³⁵ In his initial study by using water as a solvent, he performed the reaction of styrene **86** with formaldehyde **14**. A mixture of 1,3-butanediol **87**, unsaturated alcohol **88** and 1,3-dioxane **89** were obtained as products (Scheme 1.2.4.1).



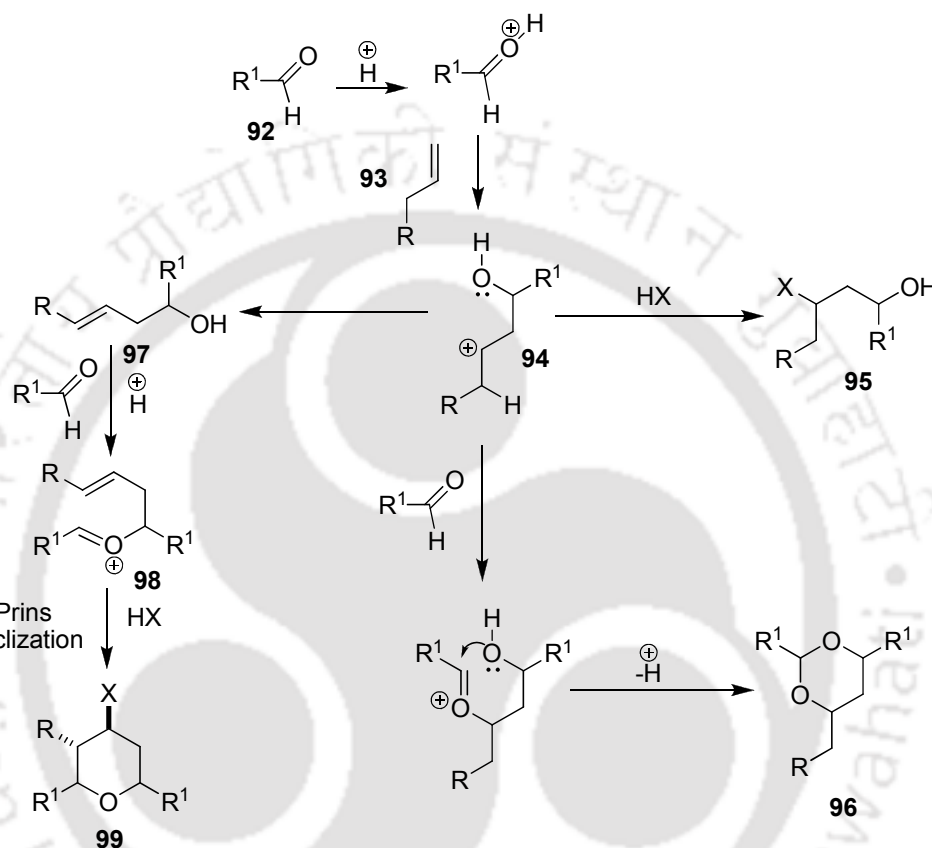
Scheme 1.2.4.1.

In the late 1960s the reaction was developed by Stapp to form tetrahydropyran derivatives.³⁶ This reaction proceeded in anhydrous media and involved the condensation of aliphatic terminal olefins **90** with paraformaldehyde **14** and hydrochloric acid to give 3-alkyl-4-chlorotetrahydropyrans **91** (Scheme 1.2.4.2). Furthermore, he demonstrated that homoallyl alcohols are the intermediates of this reaction, which induced the development of Prins cyclization that provides a powerful access to tetrahydropyran derivatives.



Scheme 1.2.4.2.

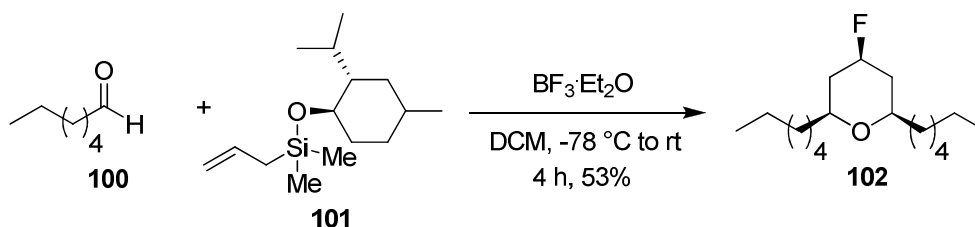
The general mechanism is shown in (Scheme 1.2.4.3). The alkene **93** reacts with carbonyl compounds **92** in the presence of Lewis or Bronsted acid and generates β -hydroxy carbocation **94** as a key intermediate, which either can react with a nucleophile such as chloride, water or acetate to give **95**, or adds to a second molecule of aldehyde to give **96**



Scheme 1.2.4.3.

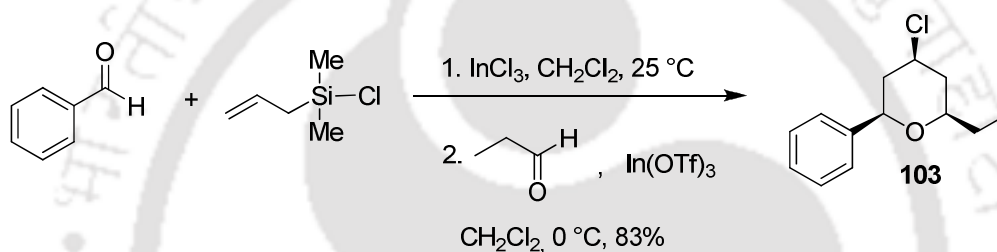
or loses a proton to give homoallylic alcohol **97**. Thus, the in situ generated homoallylic alcohol **97** reacts with another molecule of aldehyde in the presence of acidic medium to afford oxocarbenium ion **98**, which after Prins cyclization and subsequent addition of nucleophile gives the 2,4,6-trisubstituted-tetrahydropyran **99**.³⁷

Chan and co-workers have reported the synthesis of symmetrical *cis* 4-fluorotetrahydropyrans **102** from the reaction of two equivalents of aldehyde **100** and alkoxyallylsilane **101** (Scheme 1.2.4.4).³⁸



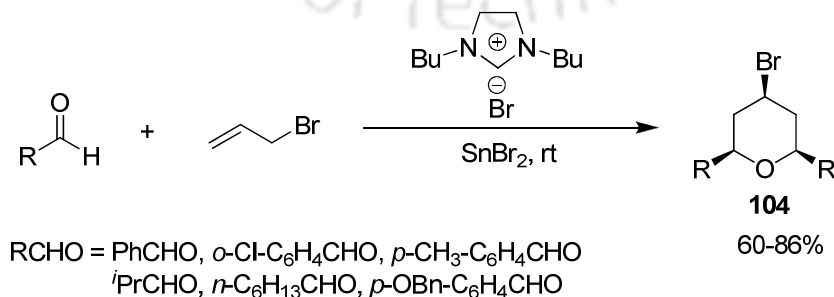
Scheme 1.2.4.4.

Loh and co-workers demonstrated the use of mild Lewis acid for the step-wise formation of homoallylic alcohol, which enable the coupling with another molecule of aldehyde in the presence of strong Lewis acid to form unsymmetrical 2,4,6-trisubstituted tetrahydropyran **103**, with high yields and selectivity (Scheme 1.2.4.5).³⁹



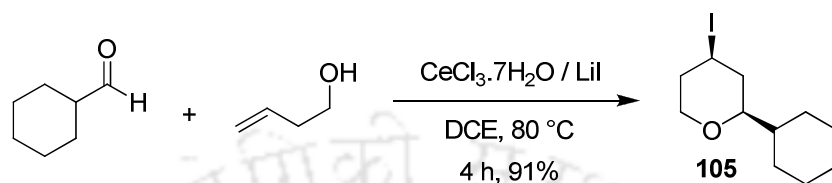
Scheme 1.2.4.5.

An interesting combination of Barbier reaction with Prins cyclization into a one-pot tandem reaction system for direct formation of tetrahydropyran was reported. Tetrahydropyran compounds can be directly synthesized from allyl bromide and carbonyl compounds by means of one-pot Barbier-Prins cyclization promoted by (N-benzyl pyridine halides) $BPyX/SnX_2$ or (Imidazole bromide salt) $BBIMBr/SnBr_2$ complex as ionic liquid under solvent-free conditions.⁴⁰ 2,6-Disubstituted-bromo-tetrahydropyrans **104** were prepared, with yields ranging from 60-86% (Scheme 1.2.4.6).



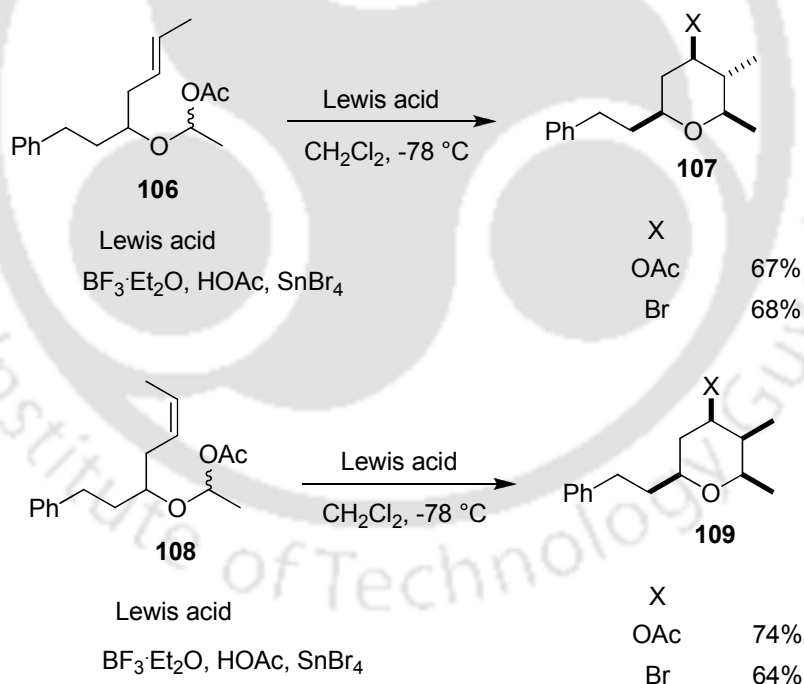
Scheme 1.2.4.6.

Homoallylic alcohols undergo smooth coupling with a variety of aldehydes in the presence of $\text{CeCl}_3 \cdot 7\text{H}_2\text{O} \cdot \text{LiI}$ reagent system in refluxing dichloroethane under neutral conditions to produce the corresponding 4-iodotetrahydropyran derivatives **105** in high yields (Scheme 1.2.4.7).⁴¹



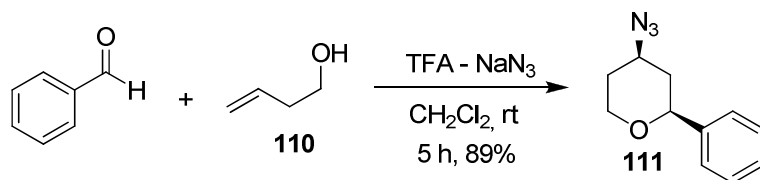
Scheme 1.2.4.7.

Rychnovsky and co-workers have reported the synthesis of highly substituted tetrahydropyrans by treatment of α -acetoxy ether, with Lewis acid. Here alkene geometries dictate the product configurations, with *E*-alkenes **106** leading to equatorial substituents **107** and *Z*-alkenes **108** leading to axial substituents **109** (Scheme 1.2.4.8).⁴²



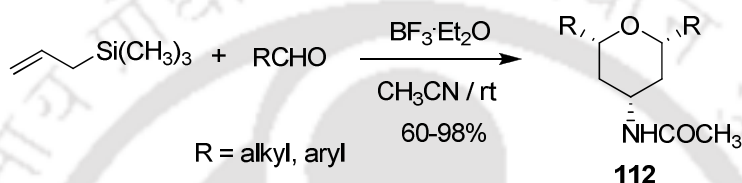
Scheme 1.2.4.8.

A three component coupling of aldehydes, homoallylic alcohols **110** and sodium azide in the presence of trifluoroacetic acid in dichloromethane had been developed to produce 4-azidotetrahydropyran **111** derivatives in high yields with all *cis*-selectivity (Scheme 1.2.4.9).⁴³



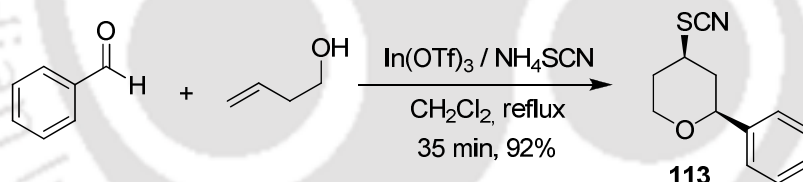
Scheme 1.2.4.9.

Saikia and co-workers have reported the synthesis of 4-amidotetrahydropyrans **112** from the reaction of aldehydes with allylsilane in acetonitrile mediated by boron trifluoride etherate in good yields and high stereoselectivity (Scheme 1.2.4.10).⁴⁴



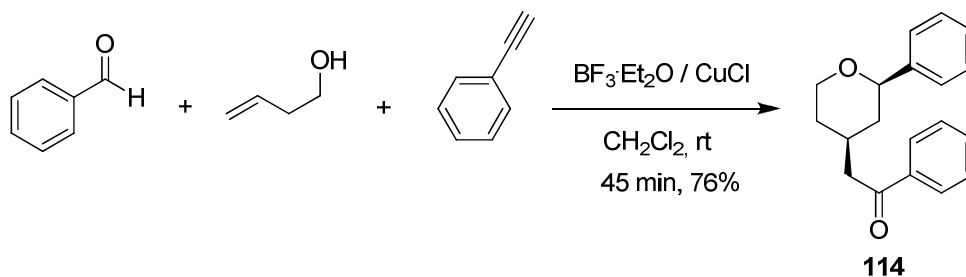
Scheme 1.2.4.10.

Jadav *et al.* have reported the synthesis of 4-thiocyanotetrahydropyrans **113** in good yields from the reaction of aldehydes, homoallylic alcohols and ammonium thiocyanate in the presence of 10 mol% In(OTf)₃ in refluxing dichloromethane (Scheme 1.2.4.11).⁴⁵



Scheme 1.2.4.11.

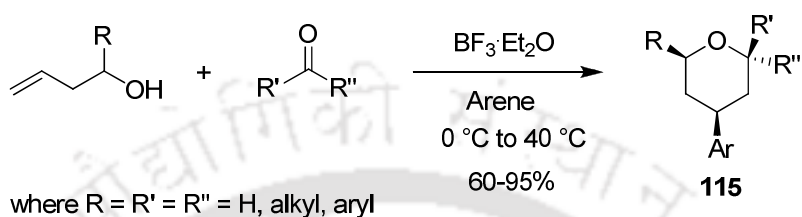
Aldehyde, homoallylic alcohol and alkyne undergo smooth Prins-type cyclization in the presence of BF₃·Et₂O / CuCl (10 mol% each) in dichloromethane under mild reaction conditions to afford 4-phenacyl tetrahydropyran derivatives **114** in good yields (Scheme



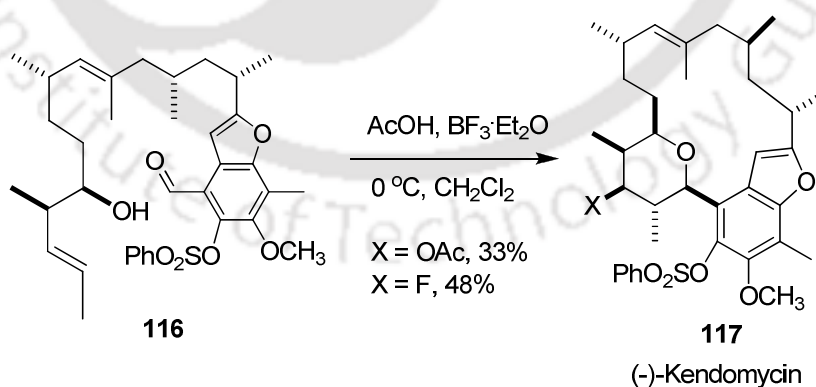
Scheme 1.2.4.12.

1.2.4.12).⁴⁶

Saikia and co-workers have reported a diastereoselective one-pot, three-component Prins-Friedel-Crafts reaction for the synthesis of 4-aryltetrahydropyran derivatives **115** from the reaction of carbonyl compounds with homoallylic alcohol in the presence of arene promoted by boron trifluoride etherate (Scheme 1.2.4.13).⁴⁷

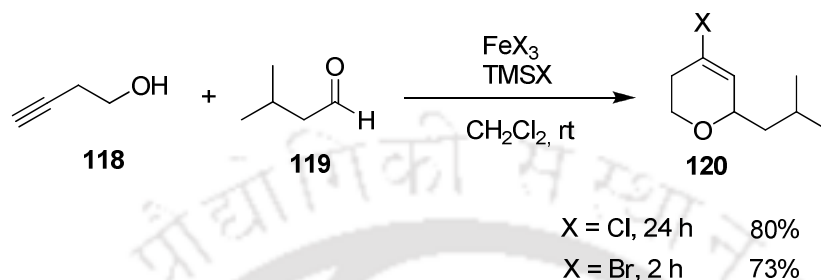
**Scheme 1.2.4.13.**

Kendomycin is an antitumor macrolide antibiotic isolated from the bacteria *Streptomyces violaceoruber*. It has potent activity as an endothelin receptor antagonist and anti-osteoporosis agent.⁴⁸ As shown in Scheme 1.2.4.14, intramolecular macro Prins cyclization using an electron-rich benzaldehyde and a homoallylic alcohol **116** efficiently delivered the fully substituted C-aryl tetrahydropyranoside Kendomycin **117**. The selective generation of three new stereocenters in the Prins cyclization facilitated the short and highly convergent assembly of the Kendomycin.⁴⁹ The reaction was mediated by $\text{BF}_3 \cdot \text{Et}_2\text{O}$ in a heterogeneous non-polar medium. For the electron rich benzaldehyde, the use of acetic acid as a trapping agent is necessary to suppress the side reactions.

**Scheme 1.2.4.14.**

More recently, Martin has also explored the synthetic potential of cyclization involving homopropargylic alcohols.⁵⁰ Reactions of homopropargylic alcohol **118** and aldehyde **119** mediated by anhydrous FeCl_3 or FeBr_3 led to 4-halo-2-alkyl-5,6-dihydro-2H-pyrans **120** in

yields ranging from 30 to 98%. Optimization of the experimental conditions has established that the best source of halide anion was TMSX in the presence of 7 mol% of the iron catalyst FeX_3 (Scheme 1.2.4.15). The cyclization could also be mediated with InCl_3 and InBr_3 , but the times of reaction were much longer and the yields were slightly lower.

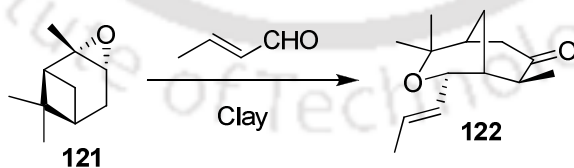


Scheme 1.2.4.15.

1.3 General Approaches for the Synthesis of Oxabicyclic Compounds

1.3.1 Method for the Synthesis of Oxabicyclo[3.3.1]nonanone

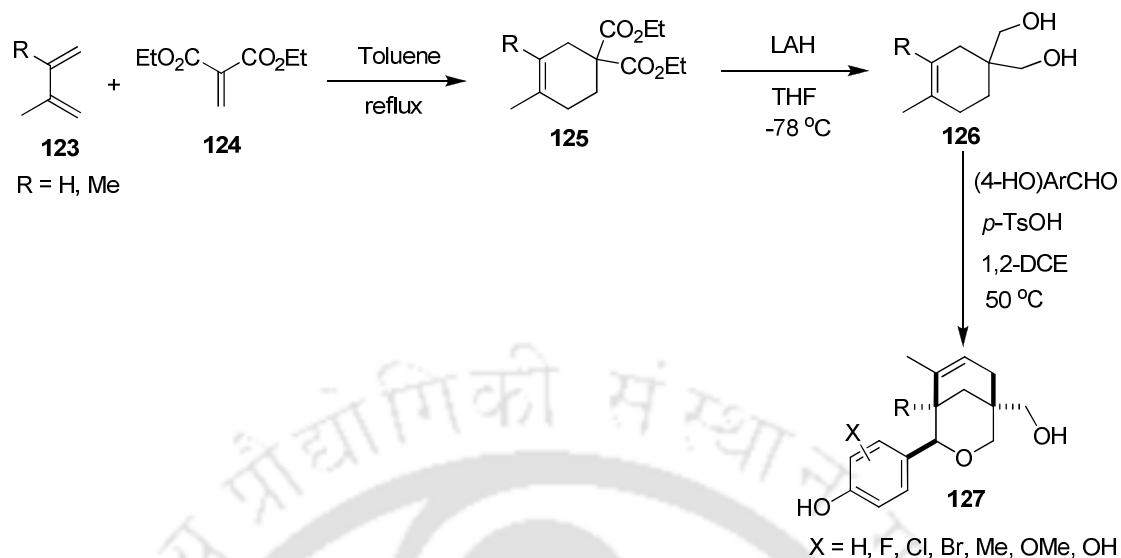
Oxygen-containing monoterpenes from the pinane series are commercially available compounds of high optical purity, and may serve as precursors of biologically active compounds⁵¹ and intermediates in asymmetric synthesis.⁵² In acidic media, they generally easily undergo numerous transformations, affording complex mixtures of products.⁵³ For example, Il'ina and co-workers have reported the synthesis of oxabicyclo[3.3.1]nonanone from α -pinene epoxide **121**.⁵⁴ The treatment of α -pinene epoxide **121** with crotonaldehyde in the presence of clay afforded oxabicyclo[3.3.1]nonanone **122**. But the yield of the product was less due to the formation of several side products (Scheme 1.3.1.1).



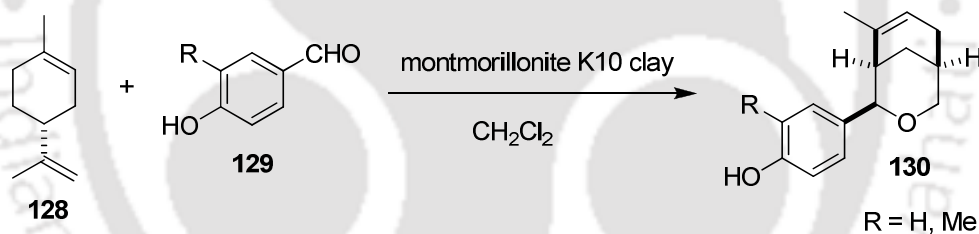
Scheme 1.3.1.1.

1.3.2 Method for the Synthesis of Oxabicyclo[3.3.1]nonene

Oxabicyclo[3.3.1]nonenes **127** are prepared by the reaction of aromatic aldehydes with olefin-containing compounds **126** in acidic media (Scheme 1.3.2.1).⁵⁵ The low yield and multistep reactions are the major drawback of these methods.

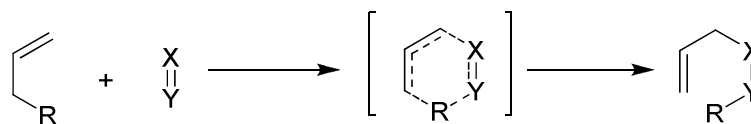


Again Hamann *et al.* have reported the synthesis of oxabicyclo[3.3.1]nonenes **130** from the reaction of Limonene **128** with aromatic aldehydes **129** in acidic medium (Scheme 1.3.2.2).⁵⁵



1.4 Oxonium-Ene Reaction

The reaction of an alkene with an allylic hydrogen atom (the ene) with another alkene (enophile) to produce a new alkene is known as ene reaction (Scheme 1.4.1). The ene (alkene) component of the ene reaction can react with a quite broad range of enophiles, which include alkenes, alkynes, allenes, carbonyls, imines and aza double bonds. The ene reaction is faster when the enophile contains one or more heteroatoms as in case of carbonyl



R = H or metal

X=Y: C=C, C≡C, C=N, C=O, C=S, N=N, N=O

Scheme 1.4.1. Ene reaction

or a nitroso group compared to simple alkene as enophile, in both of its intra- and intermolecular versions.⁵⁶ Ene cyclization that involves the oxonium ion as the enophile component, is called oxonium-ene cyclization reaction. Since, the ene and enophile can be connected in many ways, the ambiguity is inevitable. To make it unambiguous, Mikami classified the oxonium-ene reaction into three main categories namely (1,5), (2,5) and (3,5) depending upon the connectivity between enophile to the ene (Figure 1.4.1).⁵⁷ Although (1,5) and (2,5) oxonium-ene-type cyclizations have been well studied, the (3,5)-oxonium-ene cyclization has not been investigated extensively.

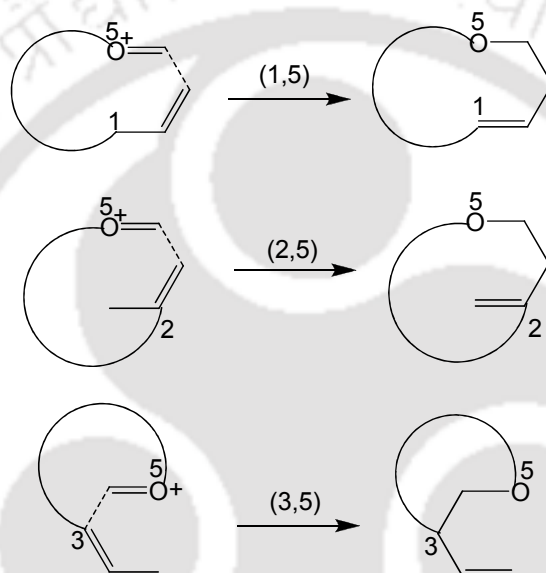
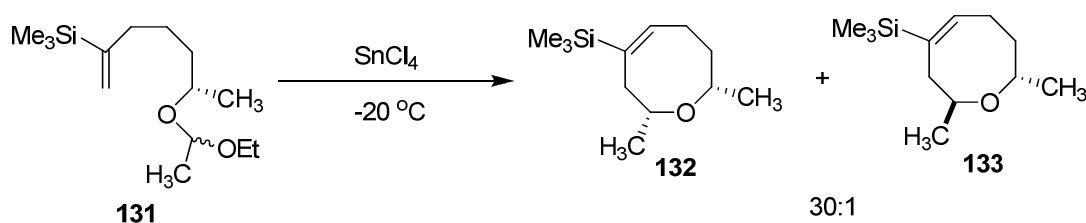


Figure 1.4.1. Classification of oxonium-ene cyclization reaction

1.4.1 (1,5)-Oxonium-Ene Reaction

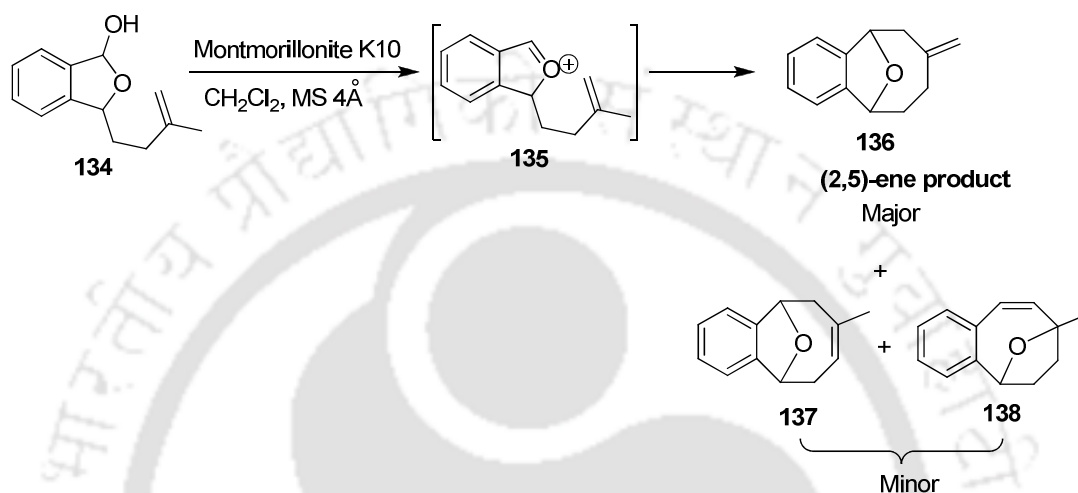
(1,5)-Oxonium-ene reaction have been well studied by Overman and co-workers. They have utilized this protocol for the synthesis of higher membered oxacycles. For example, treatment of unsaturated acetal **131** with SnCl_4 generates an oxocarbenium ion, which then undergoes cyclization via (1,5)-oxonium-ene reaction to afford mixture of 8-membered cyclic ethers **132** and **133** with excellent diastereoselectivity (Scheme 1.4.1.1).⁵⁸



Scheme 1.4.1.1.

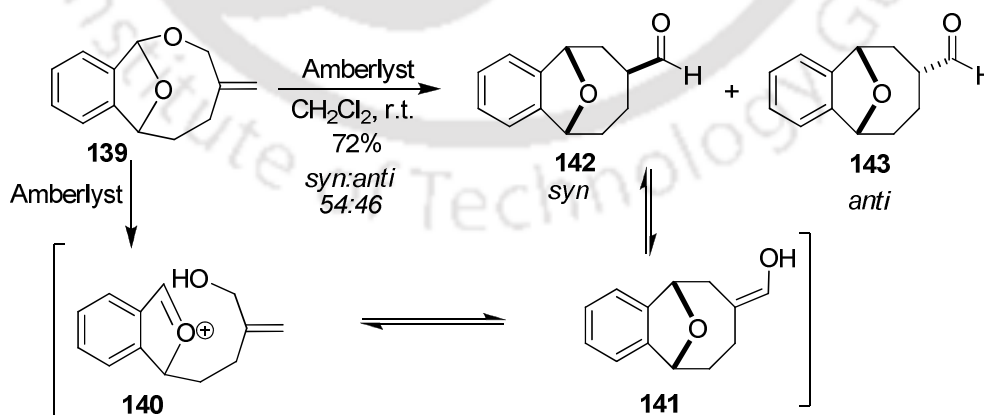
1.4.2 (2,5)-Oxonium-Ene Reaction

Mikami and co-workers have studied the (2,5)-oxonium-ene reaction. Lactol **134** on treatment with mesoporous solid acids such as montmorillonite K-10 produces tricyclic compound with exocyclic double bond **136** as a major product and its endo cyclic regioisomers **137** and **138** as minor products (Scheme 1.4.2.1).⁵⁹



Scheme 1.4.2.1.

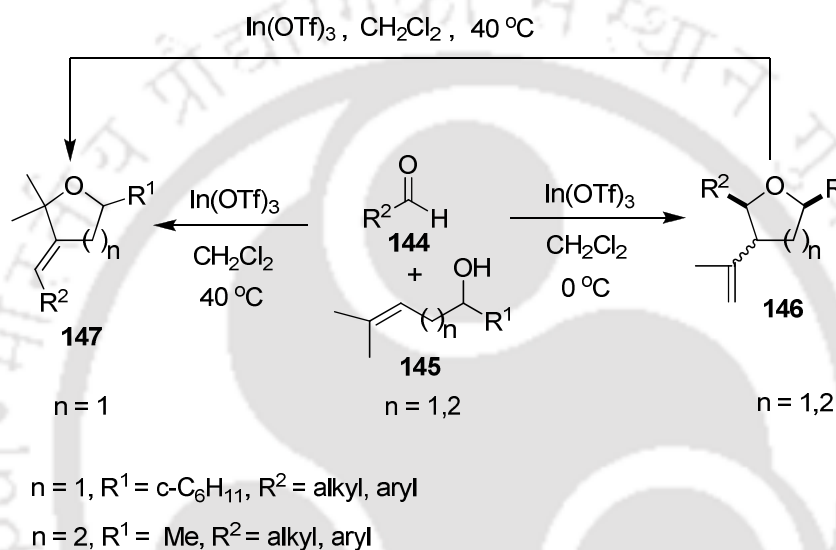
In addition to (2,5)-oxonium-ene reaction, Mikami investigated the mechanism of the reaction by trapping the (2,5) ene product in its enol form **141**, which immediately transformed to corresponding aldehydic forms **142** and **143** to restrict the regioisomerization to its endo cyclic isomers (Scheme 1.4.2.2).⁵⁹



Scheme 1.4.2.2.

1.4.3 (3,5)-Oxonium-Ene Reaction

Loh *et al.* have developed a facile method for the synthesis of tetrahydro-furans and pyrans **146** from the reaction of aldehyde **144** and unsaturated alkenes having γ and δ hydroxyl groups **145** promoted by $\text{In}(\text{OTf})_3$ via intramolecular (3,5)-oxonium-ene reaction at 0°C (Scheme 1.4.3.1). By increasing the reaction temperature from 0°C to 40°C , they observed the formation of exocyclic isomer **147** as a major product. However, this rearrangement is limited to the tetrahydrofurans skeletons only, but not to the tetrahydropyran counterparts.⁶⁰



Scheme 1.4.3.1.

1.5 References

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CHAPTER 2

A Novel Synthesis of Oxabicyclo[3.3.1]nonanone via (3,5)-Oxonium-Ene Reaction

2.1 Importance and Applications

Oxabicyclic compounds are important due to their potential biological properties. Oxabicyclo[3.3.1]nonene and its selected derivatives (Figure 2.1.1, **B**) are known to behave as estrogen receptor ligands.¹ The estrogen receptor (ER) is a member of the nuclear hormone receptor superfamily of ligand-dependent transcription factors, which mediate a broad range of physiological processes in response to binding of the endogenous ligand 17 β -estradiol (Figure 2.1.1, **A**).^{1b}

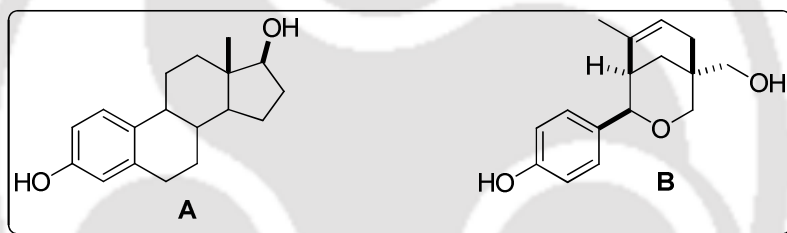
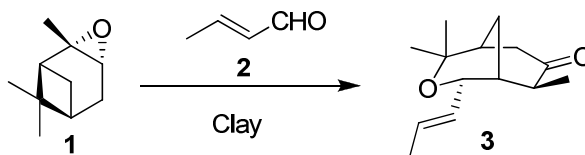


Figure 2.1.1. Estrogen receptor ligand

2.2 An Overview of Relevant Synthetic Methods

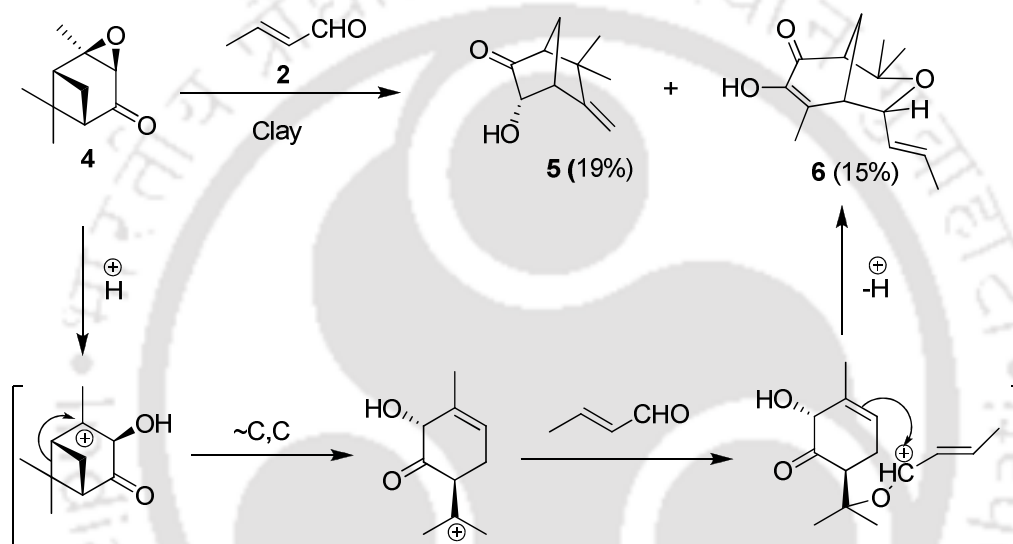
Review of literature reveals that there is only one strategy for the synthesis of oxabicyclo[3.3.1]nonanone and that has come from Il'ina's group. They have prepared oxabicyclo[3.3.1]nonanone from α -pinene epoxide and verbenone epoxide in acidic medium.²

Treatment of α -pinene epoxide **1** with aliphatic aldehydes such as crotonaldehyde **2** in the presence of clay produces oxabicyclo[3.3.1]nonanone **3** in low yield due to the formation of several side products (Scheme 2.2.1).²



Scheme 2.2.1.

On the other hand verbenone epoxide **4** on treatment with aliphatic and aromatic aldehydes in the presence of askanite-bentonite clay undergoes intermolecular cyclization. Thus, reaction of **4** with crotonaldehyde in the presence of clay afforded, besides the isomerization product **5**, the heterocyclic 1,2-diketone **6** (Scheme 2.2.2).² This compound was isolated in its enol form, which turned to be stable even upon heating to 40 °C. The mechanism includes protonation and cleavage of the epoxide ring, skeletal rearrangement into a cation with a *para*-menthane framework and further reaction of the cation with crotonaldehyde (Scheme 2.2.2).



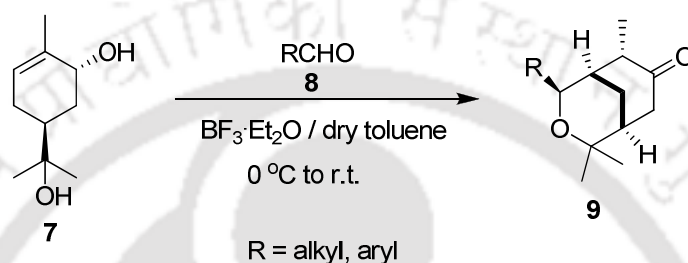
Scheme 2.2.2.

2.3 Present Work

An important objective of contemporary synthesis endeavors is the development of new transformations that rapidly evolve molecular complexity in a stereo controlled fashion. One approach towards this goal is to combine two or more distinct reactions into a single transformation, producing a process often referred to as sequential, tandem, cascade or domino reaction. Oxonium-ene reactions are powerful tool for construction of various cyclic ethers.^{3,4} There are three different types of oxonium-ene cyclization reactions, namely, (1,5), (2,5) and (3,5).^{3e,5} In this chapter, we discuss the Lewis acid catalyzed (3,5)-oxonium-ene reaction^{3e,6} for the synthesis of oxabicyclo[3.3.1]nonanone.

We have developed an efficient methodology for the synthesis of oxabicyclo[3.3.1]nonanone via (3,5)-oxonium-ene reaction from the reaction of aldehydes or epoxides with commercially available *trans*-*p*-menth-6-ene-2,8-diol, mediated by boron

trifluoride etherate (Scheme 2.3.1). *trans-p*-Menth-6-ene-2,8-diol has long been used as starting material for many organic transformations.⁷ We envisioned that treatment of *trans-p*-menth-6-ene-2,8-diol with boron trifluoride etherate would provide carbocation at the side chain, which after nucleophilic attack by aldehyde and subsequent (3,5)-oxonium-ene cyclization will give bicyclic product. Thus, the treatment of *trans-p*-menth-6-ene-2,8-diol with benzaldehyde and boron trifluoride etherate in dry toluene gave 2,2,6-trimethyl-4-phenyl-3-oxabicyclo[3.3.1]nonan-7-one in 80% yield.



Scheme 2.3.1. Synthesis of oxabicyclo[3.3.1]nonanone

The scope of the reaction was investigated by using different types of aliphatic and aromatic aldehydes (Table 2.3.1). It was observed that the aromatic aldehydes, except benzaldehyde, gave low yield, irrespective of electron withdrawing or donating groups on the ring, than the aliphatic aldehydes. The reason might be due to the formation of some highly non polar side products, which were unable to characterize by spectroscopic methods. The reaction was diastereoselective and in all the cases the substituents at 4 and 6 are in *trans* position. The structure and stereochemistry of the compound was determined by ¹H NMR, NOE experiment (Figure 2.3.1) and X-ray crystallographic analysis (Figure 2.3.2).⁸ In the ¹H NMR spectrum of compound **9h** the coupling constant between H_{5e} (2.30 ppm) and H_{9e} (2.22 ppm) is 3.2 Hz which indicates that they are in equatorial position. Again the coupling constant between H_{8a} (2.48 ppm) and H_{1e} (1.70-1.90 ppm) is 4.8 Hz which indicates that

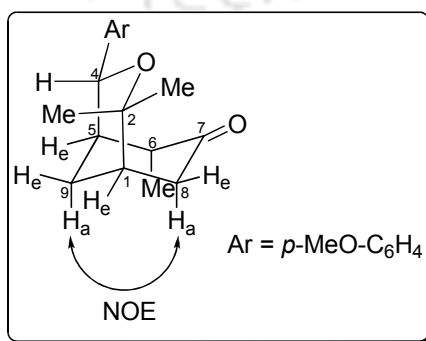
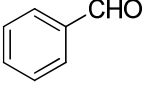
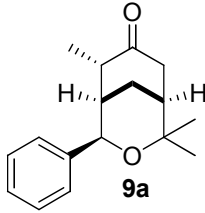
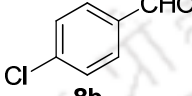
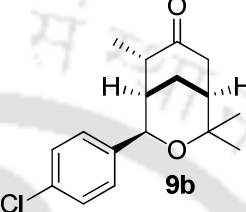
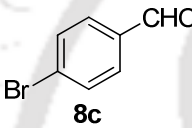
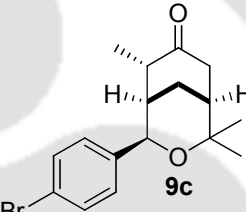
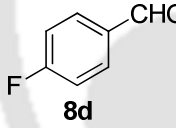
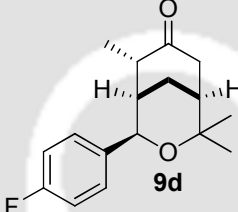
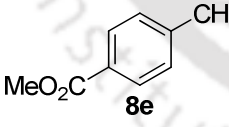
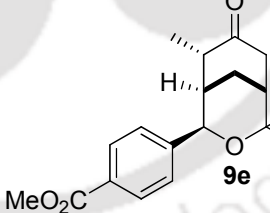
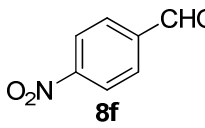
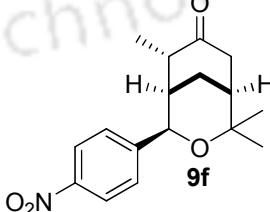


Figure 2.3.1. NOE diagram of compound **9h**

Table 2.3.1. Synthesis of oxabicyclo[3.3.1]nonanone

Entry	Aldehyde 8	Time / h	Product 9	Yield ^a (%)
1	 8a	8	 9a	80
2	 8b	10	 9b	50
3	 8c	10	 9c	51
4	 8d	10	 9d	62
5	 8e	12	 9e	52
6	 8f	12	 9f	50

continue....

Entry	Aldehyde 8	Time / h	Product 9	Yield ^a (%)
7		10		68
8		10		56
9		12		62
10		5		90
11		6		82
12		6		98
13		5		80

^aYields refer to isolated yield. The compounds are characterized by IR, ¹H, ¹³C, ¹⁹F NMR and Mass spectroscopy.

one proton is in an axial position and the other in an equatorial position. The presence of a strong NOE between H_{8a} and H_{9a} (1.93-1.95 ppm) confirms that they are in a 1,3-diaxial position. Therefore H_{1e} is in an equatorial position. Thus H_{1e} and H_{5e} are in equatorial positions and hence the fusion between two rings is *cis* fusion (Figure 2.3.1).

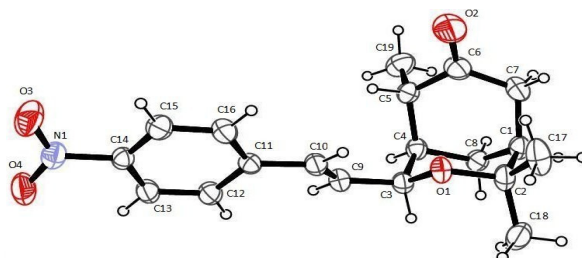
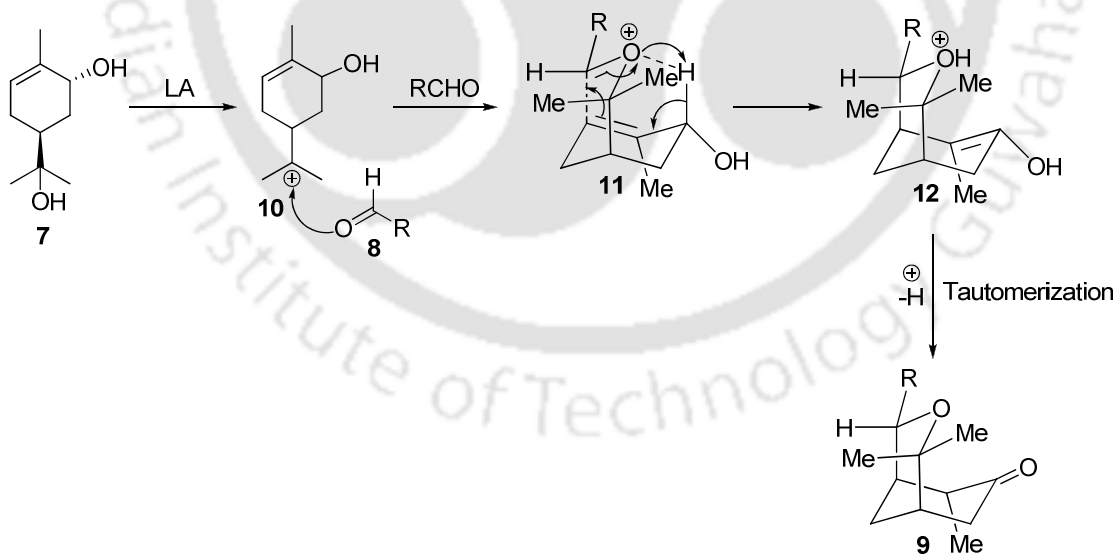


Figure 2.3.2. ORTEP diagram of 2,2,6-trimethyl-4-[2-(4-nitrophenyl-vinyl)]-3-oxabicyclo[3.3.1]nonan-7-one (**9i**)

The mechanism of the reaction can be explained as follows. The reaction of compound **7** with Lewis acid generates carbocation **10**, which after nucleophilic attack by aldehyde provides oxocarbenium ion **11**, which forms a stable six membered transition state. The oxocarbenium ion **11** undergoes (3,5)-oxonium-ene cyclization reaction to give enol **12**, which after tautomerization gives **9** (Scheme 2.3.2).



Scheme 2.3.2. Proposed mechanism of the reaction

The same oxabicyclic compounds can also be prepared by using epoxide as aldehyde equivalent as shown in Table 2.3.2. Here also the scope of the reaction was investigated by using various mono- and disubstituted epoxides. It was observed that monosubstituted

terminal epoxides are non reactive whereas the 2,2-disubstituted and styrene oxides are reactive and gives good to moderate yields. This is attributed to the lower stability of the carbocation **14**, obtained from monosubstituted epoxides, compared to 2,2-disubstituted

Table 2.3.2. Synthesis of oxabicyclo[3.3.1]nonanone from epoxides

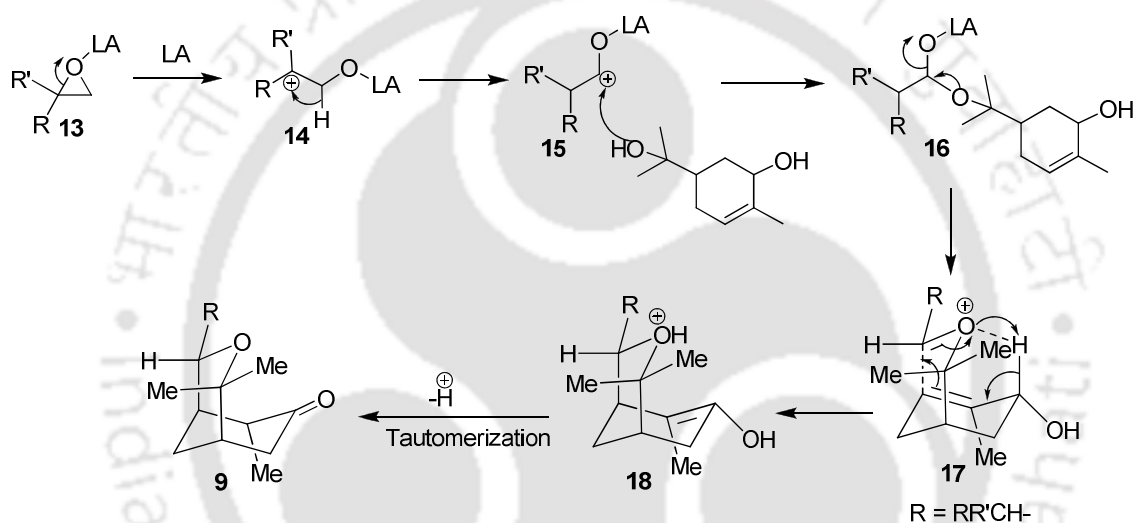
Entry	Epoxide 13	Time / h	Product 9	Yield ^a (%)
1		24		0
2		9		78
3		10		45
4		10		44
5		9		75

^aYields refer to isolated yield. The compounds are characterized by IR, NMR and Mass spectroscopy.

^bMixture of two isomers with a ratio of 1:1.

epoxides and styrene oxides, where carbocation **14** is better stabilized due to tertiary and benzylic centers, respectively.^{9a} However, the low yield obtained in case of styrene epoxides is also due to the formation of similar type of nonpolar side products. Epoxide **13d** gave two separable isomers whereas **13e** gave two inseparable isomers.

The mechanism of the reaction with epoxide can be explained by considering the already reported fact that epoxide after ring opening with Lewis acid rearranges to aldehyde.⁹ Here the intermediate **15** is attacked by alcohol to give acetal **16**, which after (3,5)-oxonium-ene cyclization gives enol **18**. The enol **18** tautomerizes to give oxabicyclic compound **9** (Scheme 2.3.3).



Scheme 2.3.3. Mechanism of the reaction with epoxide

Conclusions:

In conclusion, we have demonstrated an efficient methodology for the synthesis of oxabicyclo[3.3.1]nonanone in moderate to good yields. The bicyclic unit can further be transformed by manipulating carbonyl functionalities and therefore, can be used as a precursor for the synthesis of complex molecules.

2.4 Experimental Section

2.4.1 Instrumentation and Characterization

All the reagents were of reagent grade (AR grade) and were used as purchased without further purification. The solvents were of commercial grade and purified according to established procedures. Organic extracts were dried with anhydrous sodium sulfate. Solvents were removed in a rotary evaporator under reduced pressure. Silica gel (60-120

mesh size) was used for column chromatography. Reactions were monitored by TLC on silica gel GF₂₅₄ (0.25 mm). Melting points were recorded with a Büchi B-540 melting point apparatus. Mass spectra were recorded using Waters LC-MS/MS system (Q-TOF–Premier). HRMS (ESCI-TOF) were calibrated with sodium formate solution and leucine enkephalin (SIGMA) was used as an external standard. Fourier transform-infra red (FT-IR) spectra were recorded on Nicolet Impact-410 instrument either as neat liquid or KBr pellets. NMR spectra were recorded in CDCl₃ with tetramethylsilane as the internal standard for ¹H (400 MHz) or ¹³C (100 MHz). Crystal Data were collected with Bruker Smart Apex-II CCD diffractometer using graphite monochromatic MoK_α radiation ($\lambda = 0.71073 \text{ \AA}$) at 298 K. Cell parameters were retrieved using SMART software and refined with SAINT on all observed reflections. Data reduction was performed with the SAINT software and corrected for Lorentz and polarization effects. Absorption corrections were applied with the program SADABS. The structure was solved by direct methods implemented in SHELX-97 program and refined by full-matrix least-squares methods on F^2 . All non-hydrogen atomic positions were located in difference Fourier maps and refined anisotropically. The hydrogen atoms were placed in their geometrically generated positions.

2.4.2 General Procedure for the Synthesis of Oxabicyclic Compounds from Aldehydes (9a-9m, Table 2.3.1): To a solution of *trans-p*-menth-6-ene-2,8-diol (0.59 mmol, 1 equiv.) and aldehyde (1 equiv.) in 4 mL dry toluene at 0 °C was added boron trifluoride etherate (0.71 mmol, 1.2 equiv.) drop by drop over 5 minutes. The temperature was slowly brought to room temperature. The reaction mixture was stirred at room temperature for a specified time. The progress of the reaction was monitored by TLC with ethyl acetate and hexane as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with ethyl acetate and then washed with brine and water. The organic layer was dried (Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give the title compounds.

Synthesis of 2,2,6-Trimethyl-4-phenyl-3-oxabicyclo[3.3.1]nonan-7-one (9a, Table 2.3.1): To a solution of *trans-p*-menth-6-ene-2,8-diol (100 mg, 0.59 mmol) in dry toluene (2 mL) at 0 °C, was added boron trifluoride etherate (100 mg, 0.71 mmol). To this solution benzaldehyde (62 mg, 0.59 mmol) in dry toluene (2 mL) was added drop by drop over 5 min. The temperature was slowly brought to room temperature. The reaction mixture was stirred at room temperature for 8 h. The progress of the reaction was monitored by TLC

with ethyl acetate and hexane (1:4) as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with ethyl acetate and then washed with brine and water. The organic layer was dried (Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give **9a** (121.0 mg, 80%) as a crystalline solid; M.P. 92-96 °C.

2.4.3 General Procedure for the Synthesis of Oxabicyclic Compounds from Epoxides (**9n-9r**, Table 2.3.2):

To a solution of *trans-p*-menth-6-ene-2,8-diol (0.59 mmol, 1 equiv.) in dry toluene (2 mL) at 0 °C was added boron trifluoride etherate (0.7 mmol, 1.2 equiv.). To this solution epoxide (1.18 mmol, 2 equiv.) in dry toluene (2 mL) was added drop by drop over 5 minutes. The temperature was slowly brought to room temperature in 15 minutes. The reaction mixture was stirred at room temperature for a specified time. The progress of the reaction was monitored by TLC with ethyl acetate and hexane as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The crude product was extracted with ethyl acetate, then washed with brine and water. The organic layer was dried (Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give the title compounds.

Synthesis of 4-Isopropyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (**9o**, Table 2.3.2):

To a solution of *trans-p*-menth-6-ene-2,8-diol (100 mg, 0.59 mmol) in dry toluene (2 mL) at 0 °C was added boron trifluoride etherate (100 mg, 0.7 mmol). To this solution 1,2-epoxy-2-methyl propane (85 mg, 1.18 mmol) in dry toluene (2 mL) was added drop by drop over 5 minutes. The temperature was slowly brought to room temperature in 15 minutes. The reaction mixture was stirred at room temperature for 9 h. The progress of the reaction was monitored by TLC with ethyl acetate and hexane (1:9) as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The crude product was extracted with ethyl acetate, and then washed with brine and water. The organic layer was dried (Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give **9o** (109 mg, 78%) as a liquid.

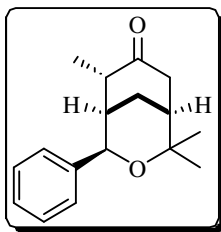
2.5 References and Notes

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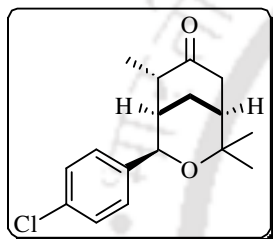
2.6 Spectral Data

2,2,6-Trimethyl-4-phenyl-3-oxabicyclo[3.3.1]nonan-7-one (9a):



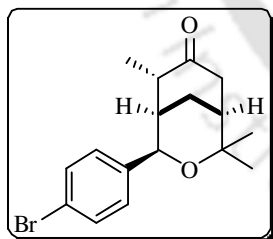
Solid (122 mg, 80%); **M.P.:** 92-96 °C; $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ 1.07 (d, $J = 7.2$ Hz, 3 H), 1.30 (s, 3 H), 1.41 (s, 3 H), 1.90-2.0 (m, 2 H), 2.25 (dt, $J = 14.0$ and 3.2 Hz, 1 H), 2.29-2.37 (m, 2 H), 2.48 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.57-2.64 (m, 1 H), 4.92 (brs, 1 H), 7.20-7.26 (m, 3 H), 7.30-7.35 (m, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 18.0, 23.4, 25.0, 27.8, 39.4, 40.6, 42.7, 43.5, 75.4, 75.5, 125.9, 126.9, 128.2, 141.1, 213.8; **IR** (KBr) 2964, 1699, 1450, 1380, 1231, 1065, 1044, 739, 697 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{22}\text{O}_2$ ($\text{M}+\text{H}$) $^+$ requires 259.1698; found 259.1691.

4-(4-Chlorophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9b):



Solid (86 mg, 50%); **M.P.:** 116-120 °C; $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ 1.08 (d, $J = 7.6$ Hz, 3 H), 1.30 (s, 3 H), 1.40 (s, 3 H), 1.88-2.0 (m, 2 H), 2.18-2.36 (m, 3 H), 2.48 (dd, $J = 15.6$ and 5.2 Hz, 1 H), 2.56-2.63 (m, 1 H), 4.89 (brs, 1 H), 7.18 (d, $J = 8.4$ Hz, 2 H), 7.29 (d, $J = 8.4$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 18.1, 23.6, 25.1, 27.9, 39.5, 40.7, 42.9, 43.6, 75.2, 75.7, 127.5, 128.5, 132.8, 139.8, 213.5; **IR** (KBr) 2983, 2940, 1702, 1491, 1382, 1234, 1083, 1070, 1048, 795 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{ClO}_2$ ($\text{M}+\text{H}$) $^+$ requires 293.1308; found 293.1315 (^{35}Cl).

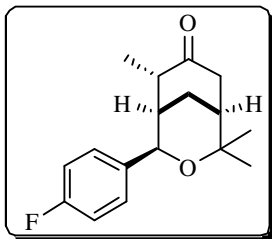
4-(4-Bromophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9c):



Solid (101 mg, 51%); **M.P.:** 128-130 °C; $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ 1.08 (d, $J = 7.6$ Hz, 3 H), 1.30 (s, 3 H), 1.41 (s, 3 H), 1.88-2.02 (m, 2 H), 2.17-2.36 (m, 3 H), 2.49 (dd, $J = 15.6$ and 4.4 Hz, 1 H), 2.56-2.64 (m, 1 H), 4.89 (brs, 1 H), 7.13 (d, $J = 8.4$ Hz, 2 H), 7.45 (d, $J = 8.4$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 18.1, 23.6, 25.1, 27.9, 39.5, 40.7, 42.9, 43.5, 75.2, 75.8, 120.9, 127.8, 131.4, 140.3, 213.5; **IR** (KBr) 2972, 2918, 1706, 1484, 1460, 1369, 1232, 1069, 1050, 793 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{BrO}_2$ ($\text{M}+\text{H}$) $^+$ requires 337.0803; found 337.0804 (^{79}Br).

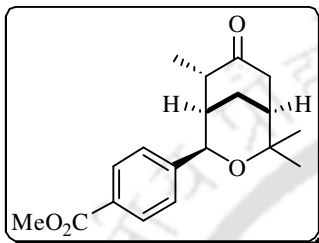
4-(4-Fluorophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9d):

Solid (101 mg, 62%); **M.P.:** 112-115 °C; $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ 1.08 (d, $J = 7.6$ Hz, 3 H), 1.30 (s, 3 H), 1.41 (s, 3 H), 1.88-2.01 (m, 2 H), 2.20-2.36 (m, 3 H), 2.49 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.56-2.64 (m, 1 H), 4.90 (brs, 1 H), 7.02 (t, $J = 8.8$ Hz, 2 H), 7.21 (t, $J = 8.0$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 18.1, 23.6, 25.1, 27.9, 39.5, 40.7, 42.8,



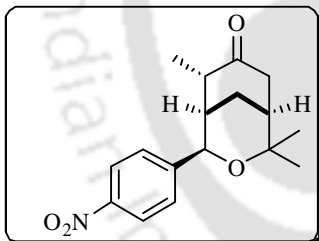
43.7, 75.2, 75.7, 115.1 (d, $J = 20.6$ Hz), 127.6 (d, $J = 7.7$ Hz), 136.9, 161.9 (d, $J = 243.3$ Hz), 213.6; ^{19}F NMR (376 MHz, $\text{CDCl}_3/\text{C}_6\text{F}_6$): δ -220.72-(-219.73) (m, 1 F); IR (KBr) 2966, 2941, 1702, 1510, 1465, 1380, 1221, 1066, 1044, 808, 784 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{FO}_2$ ($\text{M}+\text{H}$) $^+$ requires 277.1604; found 277.1610.

Methyl 4-(4,4,8-trimethyl-7-oxo-3-oxabicyclo[3.3.1]nonan-2-yl)benzoate (9e):



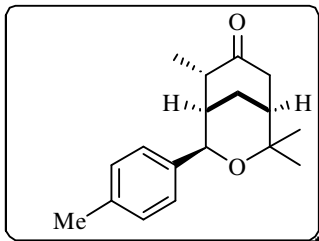
Solid (97 mg, 52%); M.P.: 115-120 $^{\circ}\text{C}$; ^1H NMR (CDCl_3 , 400 MHz): δ 1.06 (d, $J = 7.6$ Hz, 3 H), 1.32 (s, 3 H), 1.42 (s, 3 H), 1.96-2.02 (m, 2 H), 2.15 (q, $J = 7.2$ Hz, 1 H), 2.26 (dt, $J = 13.6$ and 2.8 Hz, 1 H), 2.32-2.38 (m, 1 H), 2.49 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.58-2.65 (m, 1 H), 3.90 (s, 3 H), 4.97 (brs, 1 H), 7.33 (d, $J = 8.0$ Hz, 2 H), 7.99 (d, $J = 8.4$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 17.9, 23.4, 24.9, 27.8, 39.3, 40.5, 42.9, 43.3, 52.1, 75.2, 75.7, 125.9, 128.7, 129.5, 146.5, 167.1, 213.4; IR (KBr) 2984, 2952, 1717, 1703, 1435, 1384, 1276, 1111, 1098, 1070, 754, 706 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{19}\text{H}_{24}\text{O}_4$ ($\text{M}+\text{H}$) $^+$ requires 317.1753; found 317.1753.

4-(4-Nitrophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9f):



Semisolid (89 mg, 50%); ^1H NMR (CDCl_3 , 400 MHz): δ 1.00 (d, $J = 7.2$ Hz, 3 H), 1.25 (s, 3 H), 1.35 (s, 3 H), 1.90-2.02 (m, 3 H), 2.21 (dt, $J = 14.4$ and 2.4 Hz, 1 H), 2.26-2.33 (m, 1 H), 2.44 (dd, $J = 15.2$ and 4.4 Hz, 1 H), 2.48-2.57 (m, 1 H), 4.93 (brs, 1 H), 7.35 (d, $J = 8.8$ Hz, 2 H), 8.10 (d, $J = 8.8$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 18.0, 23.5, 25.0, 27.8, 39.4, 40.6, 43.1, 43.4, 75.1, 76.1, 123.6, 126.9, 147.1, 148.8, 213.1; IR (KBr) 2967, 2923, 1703, 1517, 1344, 1229, 1094, 1067, 1048, 861, 740, 699 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{NO}_4$ ($\text{M}+\text{H}$) $^+$ requires 304.1549; found 304.1551.

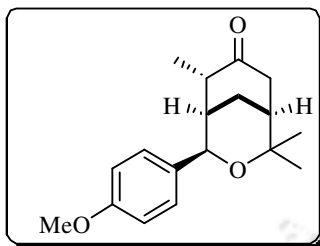
2,2,6-Trimethyl-4-(*p*-tolyl)-3-oxabicyclo[3.3.1]nonan-7-one (9g):



Solid (109 mg, 68%); M.P.: 100-104 $^{\circ}\text{C}$; ^1H NMR (CDCl_3 , 400 MHz): δ 1.07 (d, $J = 7.2$ Hz, 3 H), 1.30 (s, 3 H), 1.40 (s, 3 H), 1.85-1.98 (m, 2 H), 2.20-2.31 (m, 3 H), 2.32 (s, 3 H), 2.48 (dd, $J = 15.6$ and 4.4 Hz, 1 H), 2.55-2.62 (m, 1 H), 4.90 (brs, 1 H), 7.13 (brs, 4 H); ^{13}C NMR (100 MHz, CDCl_3): δ 18.1, 21.2,

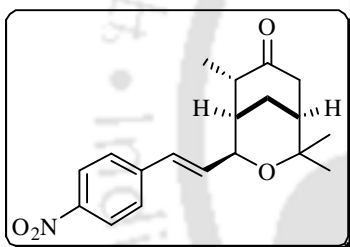
23.6, 25.1, 27.9, 39.5, 40.7, 42.7, 43.6, 75.4, 75.5, 125.9, 128.9, 136.5, 138.2, 213.9; **IR** (KBr) 2926, 2878, 1698, 1455, 1380, 1230, 1066, 1046, 799, 772 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{24}\text{O}_2$ ($\text{M}+\text{H}$)⁺ requires 273.1855; found 273.1853.

4-(4-Methoxyphenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9h):



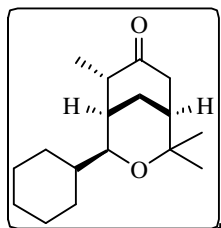
Solid (95 mg, 56%); **M.P.:** 97-101 °C; **¹H NMR** (CDCl_3 , 400 MHz): δ 1.07 (d, $J = 7.2$ Hz, 3 H), 1.29 (s, 3 H), 1.40 (s, 3 H), 1.70-1.90 (m, 1 H), 1.93-1.95 (m, 1 H), 2.22 (dt, $J = 13.6$ and 3.2 Hz, 1 H), 2.27-2.31 (m, 1 H), 2.32-2.36 (m, 1 H), 2.48 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.56-2.63 (m, 1 H), 3.78 (s, 3 H), 4.88 (brs, 1 H), 6.86 (d, $J = 8.8$ Hz, 2 H), 7.16 (d, $J = 8.4$ Hz, 2 H); **¹³C NMR** (100 MHz, CDCl_3): δ 18.1, 23.6, 25.1, 28.0, 39.6, 40.7, 42.7, 43.7, 55.4, 75.4, 75.5, 113.7, 127.2, 133.4, 158.6, 214.0; **IR** (KBr) 2929, 2846, 1699, 1513, 1461, 1376, 1249, 1042, 1035, 842, 803 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{24}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 289.1803; found 289.1808.

4-(4-Nitrostyryl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9i):

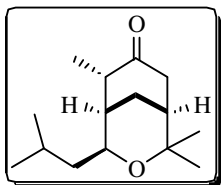


Solid (120 mg, 62%); **M.P.:** 135-138 °C; **¹H NMR** (CDCl_3 , 400 MHz): δ 1.22 (d, $J = 7.2$ Hz, 3 H), 1.28 (s, 3 H), 1.39 (s, 3 H), 1.64-1.68 (m, 1 H), 1.87-1.90 (m, 1 H), 1.93-1.98 (m, 1 H), 2.20-2.24 (m, 1 H), 2.48 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.56-2.65 (m, 2 H), 4.54 (brd, $J = 4.4$ Hz, 1 H), 6.32 (dd, $J = 16.0$ and 4.4 Hz, 1 H), 6.66 (d, $J = 16.0$ Hz, 1 H), 7.50 (d, $J = 8.8$ Hz, 2 H), 8.16 (d, $J = 8.8$ Hz, 2 H); **¹³C NMR** (100 MHz, CDCl_3): δ 18.0, 23.4, 24.6, 27.7, 39.3, 40.4, 41.9, 43.8, 73.8, 75.5, 123.9, 126.9, 128.4, 134.1, 143.5, 146.7, 213.5; **IR** (KBr) 2975, 2898, 1700, 1594, 1508, 1345, 1228, 1109, 1074, 974, 869 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{19}\text{H}_{23}\text{NO}_4$ ($\text{M}+\text{H}$)⁺ requires 330.1705; found 330.1714.

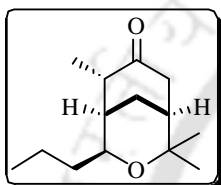
4-Cyclohexyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9j):



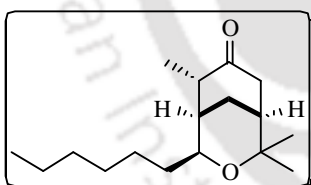
Solid (140 mg, 90%); **M.P.:** 73-77 °C; **¹H NMR** (CDCl_3 , 400 MHz): δ 0.69-0.84 (m, 3 H), 1.14 (s, 3 H), 1.20 (d, $J = 7.6$ Hz, 3 H), 1.25 (s, 3 H), 1.37-1.47 (m, 3 H), 1.60-1.73 (m, 3 H), 1.81-1.89 (m, 2 H), 1.98-2.16 (m, 4 H), 2.41 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.49-2.60 (m, 2 H), 3.24 (d, $J = 9.2$ Hz, 1 H); **¹³C NMR** (100 MHz, CDCl_3): δ 18.0, 23.5, 25.1, 25.6, 25.9, 26.6, 27.8, 27.9, 30.6, 38.1, 38.2, 40.0, 40.5, 43.3, 74.3, 78.5, 214.2; **IR** (KBr) 2927, 2852, 1702, 1450, 1382, 1234, 1099, 1040, 907 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{28}\text{O}_2$ ($\text{M}+\text{H}$)⁺ requires 265.2167; found 265.2171.

4-Isobutyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9k):

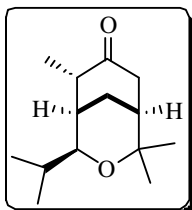
Liquid (115 mg, 82%); $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ 0.86 (d, $J = 6.4$ Hz, 3 H), 0.87 (d, $J = 6.4$ Hz, 3 H), 1.15 (s, 3 H), 1.22 (d, $J = 7.6$ Hz, 3 H), 1.30 (s, 3 H), 1.43-1.51 (m, 2 H), 1.58-1.70 (m, 2 H), 1.84-1.89 (m, 1 H), 2.00-2.18 (m, 2 H), 2.43 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.49-2.54 (m, 1 H), 2.63 (q, $J = 7.2$ Hz, 1 H), 3.75-3.79 (m, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 18.1, 22.5, 23.0, 23.5, 24.6, 25.2, 27.8, 39.8, 40.5, 41.5, 41.8, 43.4, 71.7, 74.4, 213.9; **IR** (Neat) 2955, 2928, 1710, 1468, 1381, 1231, 1096, 1028, 912 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{15}\text{H}_{26}\text{O}_2$ ($\text{M}+\text{H}$) $^+$ requires 239.2011; found 239.2009.

4-Propyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9l):

Liquid (130 mg, 98%); $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ 0.89 (t, $J = 7.2$ Hz, 3 H), 1.16 (s, 3 H), 1.22 (d, $J = 7.6$ Hz, 3 H), 1.25-1.35 (m, 2 H), 1.30 (s, 3 H), 1.42-1.56 (m, 2 H), 1.64-1.70 (m, 1 H), 1.83-1.89 (m, 1 H), 2.04-2.17 (m, 2 H), 2.43 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.49-2.54 (m, 1 H), 2.62 (q, $J = 7.6$ Hz, 1 H), 3.68 (t, $J = 6.4$ Hz, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.0, 17.9, 19.0, 23.5, 25.0, 27.8, 34.8, 39.8, 40.4, 41.0, 43.1, 73.5, 74.4, 213.8; **IR** (Neat) 2960, 2873, 1710, 1458, 1381, 1229, 1096, 1041, 913 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{14}\text{H}_{24}\text{O}_2$ ($\text{M}+\text{H}$) $^+$ requires 225.1854; found 225.1844.

4-Hexyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9m):

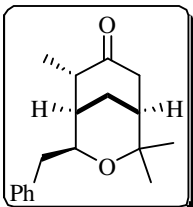
Liquid (126 mg, 80%); $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ 0.87 (t, $J = 6.0$ Hz, 3 H), 1.16 (s, 3 H), 1.23 (d, $J = 7.6$ Hz, 3 H), 1.25-1.28 (m, 6 H), 1.30 (s, 3 H), 1.32-1.36 (m, 2 H), 1.47-1.54 (m, 2 H), 1.65-1.70 (m, 1 H), 1.84-1.90 (m, 1 H), 2.04-2.16 (m, 2 H), 2.43 (dd, $J = 15.2$ and 4.4 Hz, 1 H), 2.49-2.56 (m, 1 H), 2.62 (q, $J = 7.2$ Hz, 1 H), 3.67 (t, $J = 5.6$ Hz, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 18.1, 22.6, 23.6, 25.1, 25.9, 27.9, 29.3, 31.8, 32.8, 39.9, 40.5, 41.0, 43.2, 73.9, 74.5, 214.0; **IR** (Neat) 2931, 2858, 1710, 1465, 1381, 1234, 1066, 1040, 910 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{30}\text{O}_2$ ($\text{M}+\text{H}$) $^+$ requires 267.2324; found 267.2328.

4-Isopropyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9o):

Liquid (103 mg, 78%); $^1\text{H NMR}$ (CDCl_3 , 400 MHz): δ 0.75 (dd, $J = 6.4$ and 1.2 Hz, 3 H), 0.92 (dd, $J = 6.4$ and 1.2 Hz, 3 H), 1.15 (s, 3 H), 1.22 (dd, $J = 7.6$ Hz and 1.2 Hz, 3 H), 1.27 (s, 3 H), 1.67 (octet, $J = 6.4$ Hz, 1 H), 1.86 (brs, 2 H), 1.98-2.18 (m, 2 H), 2.42 (dd, $J = 15.6$ and 4.8 Hz, 1 H),

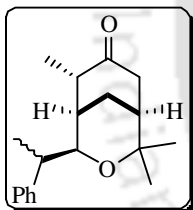
2.49 (brs, 1 H), 2.56 (q, $J = 7.6$ Hz, 1 H), 3.14 (d, $J = 9.6$ Hz, 1 H); ^{13}C NMR (100 MHz, CDCl_3): δ 17.9, 18.1, 20.4, 23.6, 25.2, 27.8, 29.2, 38.7, 39.8, 40.6, 43.3, 74.5, 79.9, 214.1; **IR** (Neat) 2964, 2935, 1710, 1470, 1381, 1232, 1096, 1042 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{14}\text{H}_{24}\text{O}_2$ ($\text{M}+\text{H}$) $^+$ requires 225.1855; found 225.1852.

4-Benzyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9p):



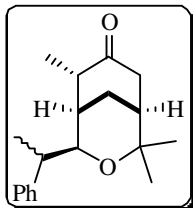
Liquid (72 mg, 45%); ^1H NMR (CDCl_3 , 400 MHz): δ 1.19 (s, 3 H), 1.20 (d, $J = 7.2$ Hz, 3 H), 1.29 (s, 3 H), 1.57-1.59 (m, 1 H), 1.84-1.90 (m, 1 H), 1.97-2.0 (m, 1 H), 2.10 (dt, $J = 13.6$ and 2.8 Hz, 1 H), 2.44 (dd, $J = 15.6$ and 5.2 Hz, 1 H), 2.53-2.58 (m, 1 H), 2.66 (dd, $J = 13.6$ and 7.6 Hz, 1 H), 2.80-2.89 (m, 2 H), 3.94 (dt, $J = 7.6$ and 1.6 Hz, 1 H), 7.16-7.22 (m, 3 H), 7.25-7.30 (m, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 18.0, 23.7, 25.2, 27.9, 39.1, 39.7, 39.8, 40.7, 43.3, 75.2, 75.3, 126.3, 128.5, 129.2, 138.8, 214.1; **IR** (Neat) 2970, 2930, 1708, 1454, 1381, 1235, 1089, 1041, 752, 700 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{24}\text{O}_2$ ($\text{M}+\text{H}$) $^+$ requires 273.1855; found 273.1860.

2,2,6-Trimethyl-4-(1-phenyl-ethyl)-3-oxabicyclo[3.3.1]nonan-7-one (Lower fraction, 9qL):



Solid (40 mg, 24%); M.P. 116-120 $^\circ\text{C}$; ^1H NMR (CDCl_3 , 400 MHz): δ 0.97 (s, 3 H), 1.03 (s, 3 H), 1.08 (d, $J = 7.2$ Hz, 3 H), 1.28 (d, $J = 7.6$ Hz, 3 H), 1.80-1.86 (m, 1 H), 1.97-2.14 (m, 2 H), 2.19 (dt, $J = 14.0$ and 2.8 Hz, 1 H), 2.43 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.48-2.56 (m, 1 H), 2.71 (q, $J = 7.6$ Hz, 1 H), 2.77-2.86 (m, 1 H), 3.69 (d, $J = 10.0$ Hz, 1 H), 7.13-7.19 (m, 3 H), 7.22-7.27 (m, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 17.6, 18.3, 23.1, 25.5, 27.6, 38.9, 39.9, 40.7, 41.2, 43.3, 74.6, 78.3, 125.9, 127.9, 128.0, 145.7, 213.9; **IR** (KBr) 2965, 2928, 1697, 1454, 1382, 1233, 1070, 1045, 698 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{19}\text{H}_{26}\text{O}_2$ ($\text{M}+\text{H}$) $^+$ requires 287.2011; found 287.2003.

2,2,6-Trimethyl-4-(1-phenyl-ethyl)-3-oxabicyclo[3.3.1]nonan-7-one (Upper fraction, 9qU):

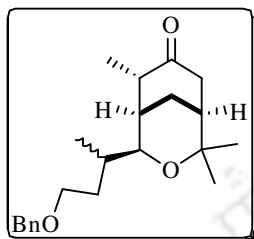


Solid (34 mg, 20%); M.P. 150-153 $^\circ\text{C}$; ^1H NMR (CDCl_3 , 400 MHz): δ 0.98 (d, $J = 7.6$ Hz, 3 H), 1.20 (s, 3 H), 1.27 (d, $J = 6.8$ Hz, 3 H), 1.35 (s, 3 H), 1.83-1.86 (m, 1 H), 1.91-2.04 (m, 3 H), 2.37 (dd, $J = 15.6$ and 4.8 Hz, 1 H), 2.53 (m, 1 H), 2.70 (q, $J = 7.6$ Hz, 1 H), 2.79-2.87 (m, 1 H), 3.65 (d, $J = 10.0$ Hz, 1 H), 7.13-7.21 (m, 3 H), 7.25-7.31 (m, 2 H); ^{13}C NMR (100 MHz,

CDCl₃): δ 17.9, 20.0, 23.9, 25.5, 27.9, 38.6, 39.8, 40.9, 41.8, 43.5, 75.2, 79.4, 126.6, 127.8, 128.8, 144.1, 214.5; **IR** (KBr) 2971, 2923, 1704, 1455, 1379, 1238, 1095, 1017, 711 cm⁻¹. **HRMS** (APCI) cald. for C₁₉H₂₆O₂ (M+H)⁺ requires 287.2011; found 287.2022.

4-(3-Benzyloxy-1-methyl-propyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one:

{(9r, mixture of two isomers (1:1)):



Liquid (152 mg, 75%); **¹H NMR** (CDCl₃, 400 MHz): δ 0.76 (d, J = 6.8 Hz, 1.5 H), 0.90 (d, J = 6.4 Hz, 1.5 H), 1.12 (s, 1.5 H), 1.15 (s, 1.5 H), 1.21 (d, J = 6.8 Hz, 1.5 H), 1.22 (d, J = 6.8 Hz, 1.5 H), 1.23 (s, 1.5 H), 1.26 (s, 1.5 H), 1.56-1.75 (m, 2 H), 1.82-1.92 (m, 2 H), 1.95-2.08 (m, 2 H), 2.10-2.18 (m, 1 H), 2.38-2.58 (m, 2 H), 3.22-3.28 (m, 2 H), 3.42-3.56 (m, 3 H), 4.48 (s, 2 H), 7.25-7.40 (m, 5 H); **¹³C NMR** (100 MHz, CDCl₃): δ 15.3, 16.7, 18.2, 23.5, 23.7, 25.3 (2C), 27.9, 29.9, 31.6, 31.7, 31.9, 33.9, 38.7, 39.9, 40.0, 40.6, 40.7, 42.8, 43.2, 43.4, 43.6, 68.5, 69.3, 73.0, 73.3, 74.7, 74.8, 78.4, 78.8, 127.6, 127.7, 127.8, 128.0, 128.5, 128.7, 138.5, 138.7, 214.4 (2C); **IR** (Neat) 2927, 2855, 1709, 1454, 1234, 1096, 1042, 737, 698 cm⁻¹. **HRMS** (APCI) cald. for C₂₂H₃₂O₃(M+H)⁺ requires 345.2430; found 345.2423.

2.7 Selected Spectra of Oxabicyclo[3.3.1]nonanone

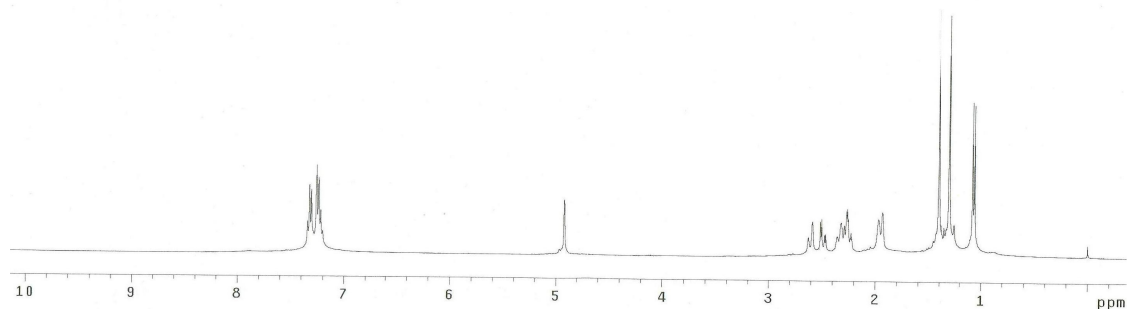
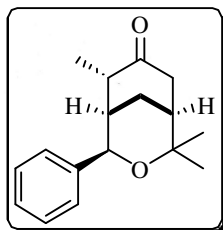
2,2,6-Trimethyl-4-phenyl-3-oxabicyclo[3.3.1]nonan-7-one (9a):

 ^1H NMR (400 MHz, CDCl_3)

```

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at 1.398 a1fa 20.000
np 25528 FLAGS
fb not used i1 n
bs 4 in n
d1 1.000 dp Y
nt 32 hs nn
ct 32
TRANSMITTER
tn H1 fb 0.10
sfrq 399.853 fn 65536
tof 362.8 sp -163.6
tpwr 57 wp 4273.6
pw 9.850 rfl 733.5
DECOUPLER
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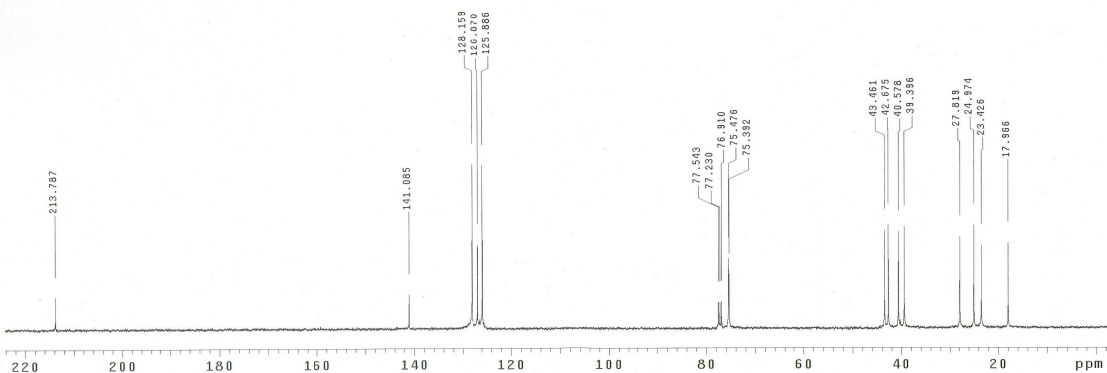
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 ^{13}C NMR (100 MHz, CDCl_3)

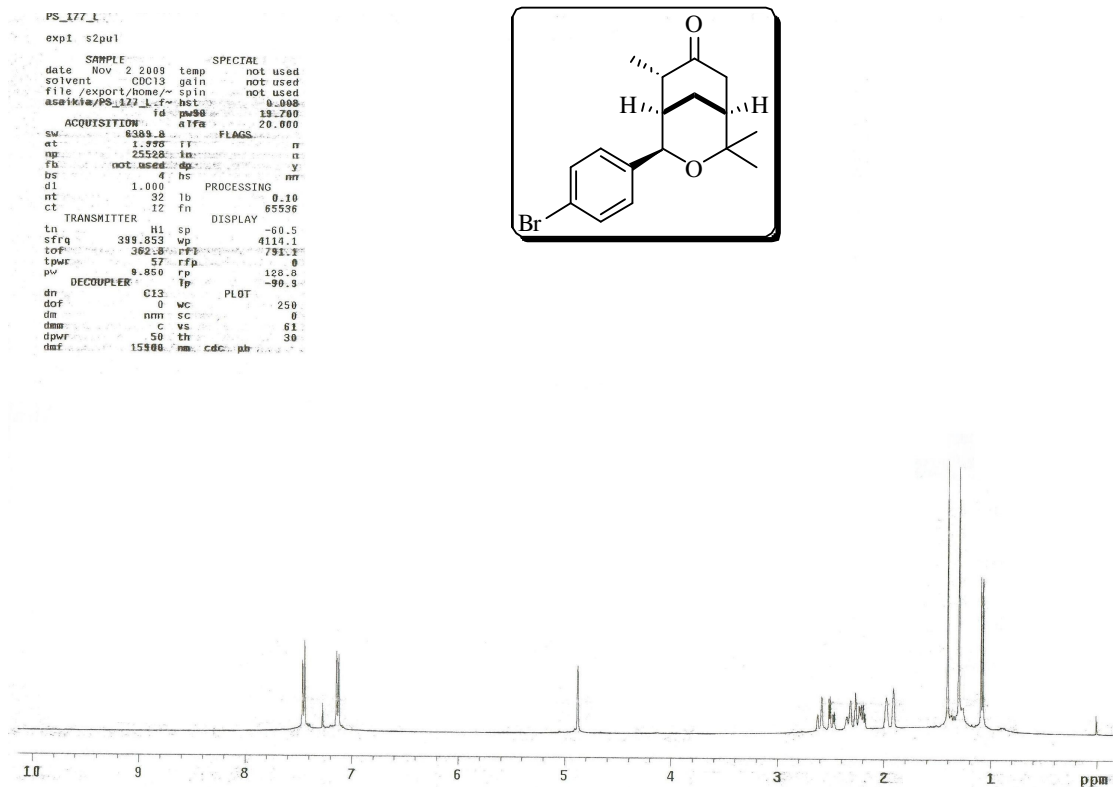
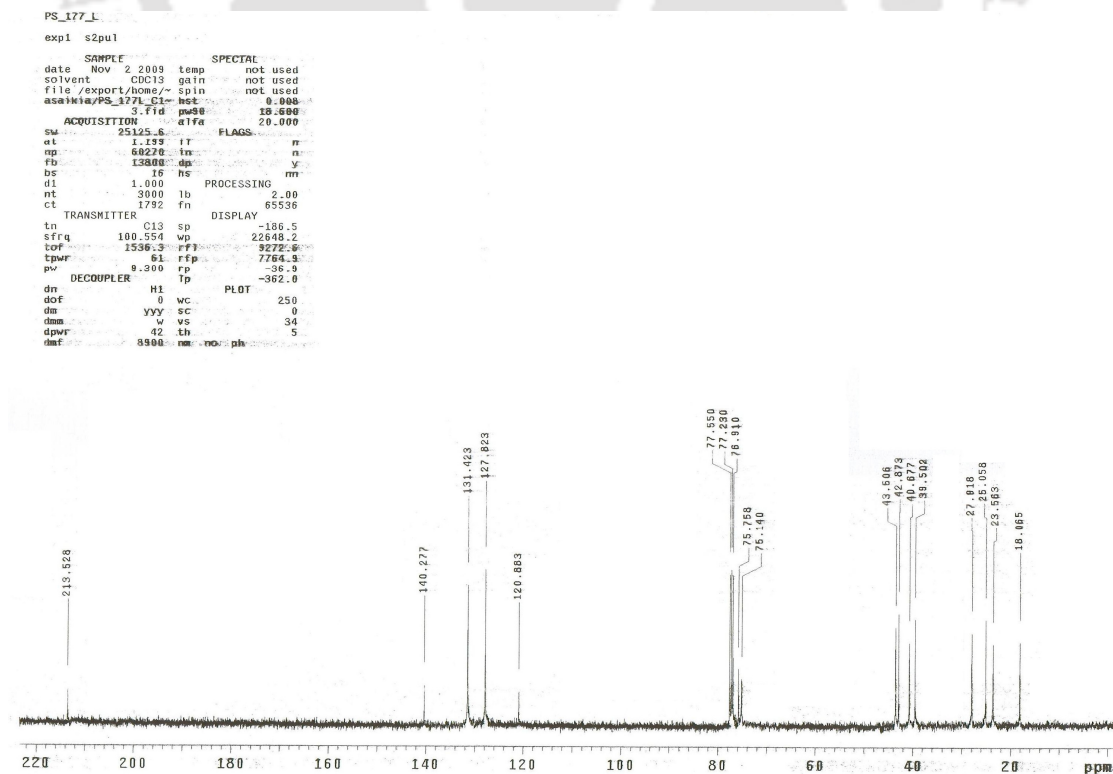
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solvent CDC13 gain not used
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ACQUISITION
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at 1.189 i1 n
np 60270 in n
fb 13800 dp Y
bs 16 hs nn
d1 1.000 lb 2.00
nt 4000 fb 65536
ct 592 fn
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sfrq 100.554 wp 22780.1
tof 1536.3 rfl 9289.5
tpwr 61 rfp 7764.9
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dpr 42 rh
dof 8900 nm no ph

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4-(4-Bromophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9c):

 ^1H NMR (400 MHz, CDCl_3) ^{13}C NMR (100 MHz, CDCl_3)

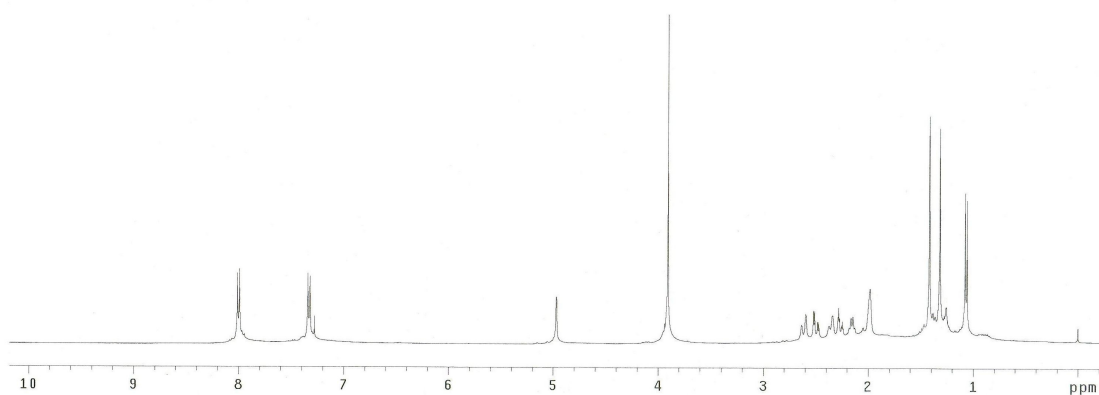
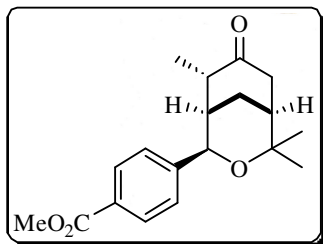
4-(4,4,8-Trimethyl-7-oxo-3-oxabicyclo[3.3.1]non-2-yl)-benzoic acid methyl ester (9e):

 ^1H NMR (400 MHz, CDCl_3)

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file exp spin not used
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at 1.998 a1fa 20.000
np 25528 FLAGS
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dl 1.000 dp y
nt 32 hs nm
ct
TRANSMITTER lb PROCESSING 0.10
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sfrq 399.853 DISPLAY
tof 382.8 sp -31.5
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pw 9.850 rfl 786.4
DECOUPLER rfp 0
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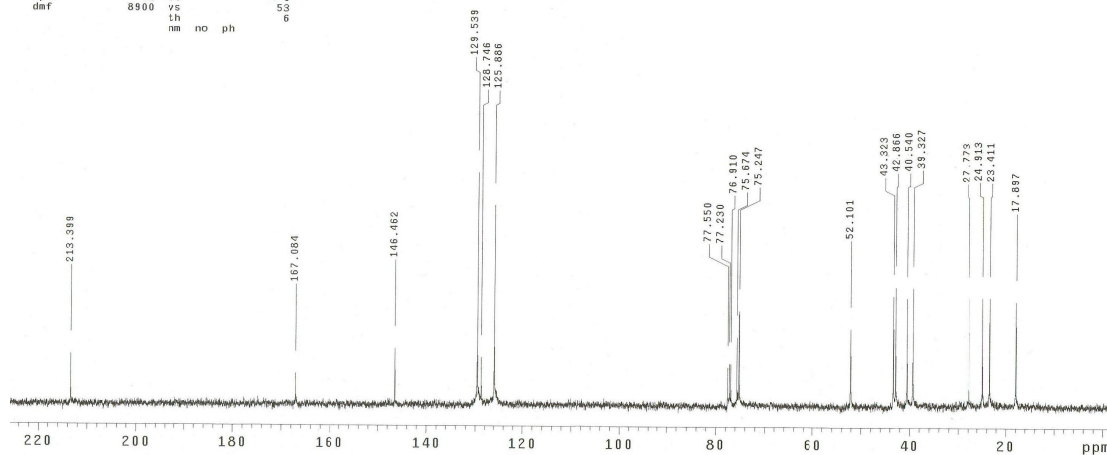
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 ^{13}C NMR (100 MHz, CDCl_3)

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file exp spin not used
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dl 1.000 dp y
nt 2000 hs nm
ct
TRANSMITTER lb PROCESSING 2.00
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sfrq 100.554 DISPLAY
tof 1536.3 sp -263.2
tpwr 61 wp 23044.6
pw 9.300 rfl 9283.3
DECOUPLER rfp 7764.9
dn H1 rp -53.2
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th th 6
nm no ph

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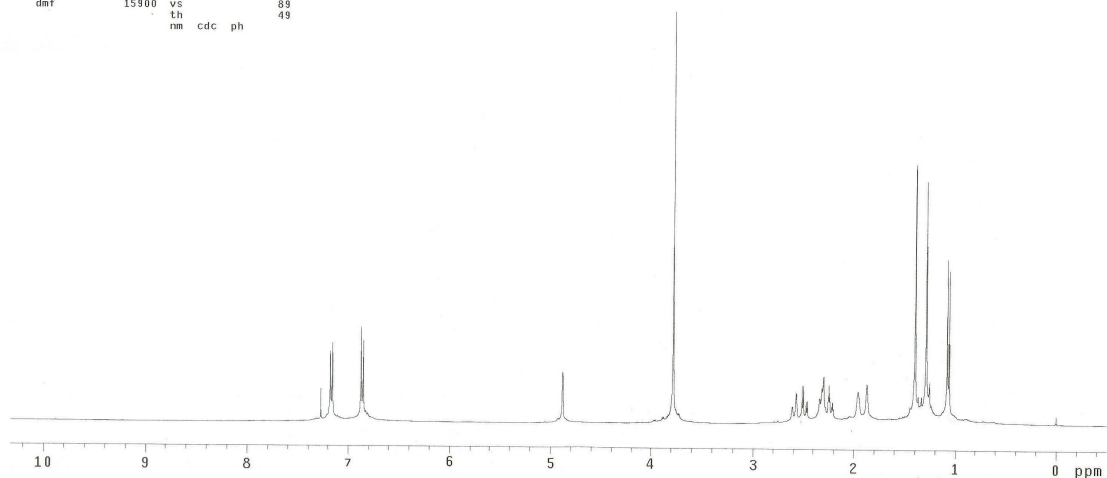
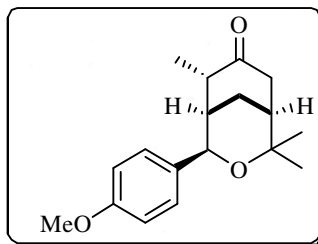
4-(4-Methoxyphenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]nonan-7-one (9h):

 ^1H NMR (400 MHz, CDCl_3)

```

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solvent CDC13 gain not used
file exp spin not used
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np 25528
fb not used il
bs 4 in
d1 1.000 dp
nt 32 hs
ct 32
TRANSMITTER lb fn
tn H1 fn
sfrq 399.853
tof 392.8 sp
tpwr 57 wp
pw 9.850 rfl
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dn 0 lp
dof 0 lp
dm nmm
dmm c wc
dpwr 50 sc
dmf 15900 vs
nm cdc ph
SPECIAL
not used
not used
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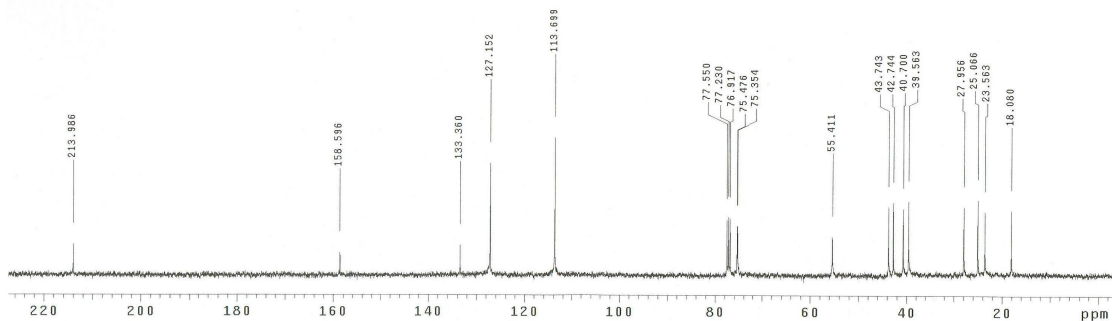
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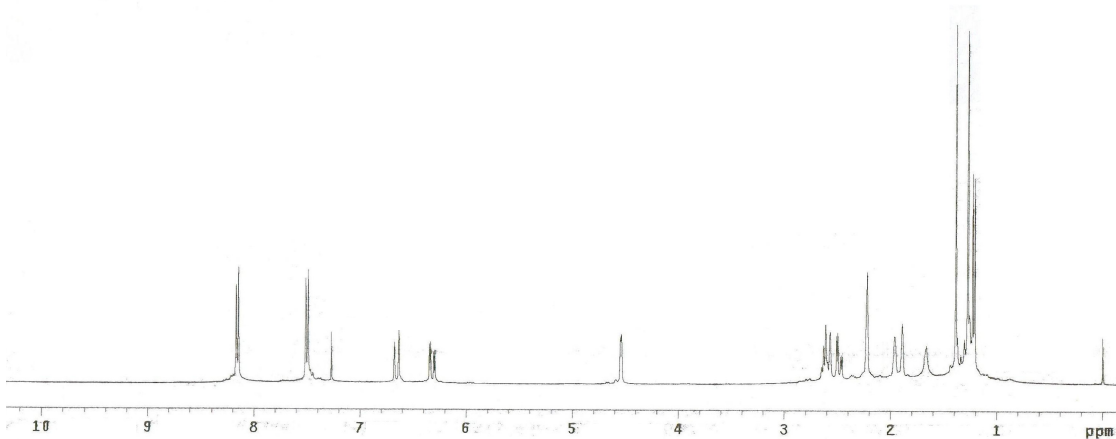
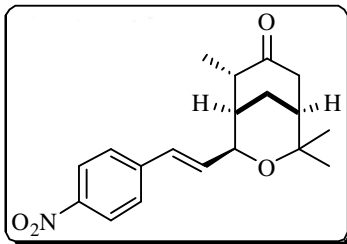


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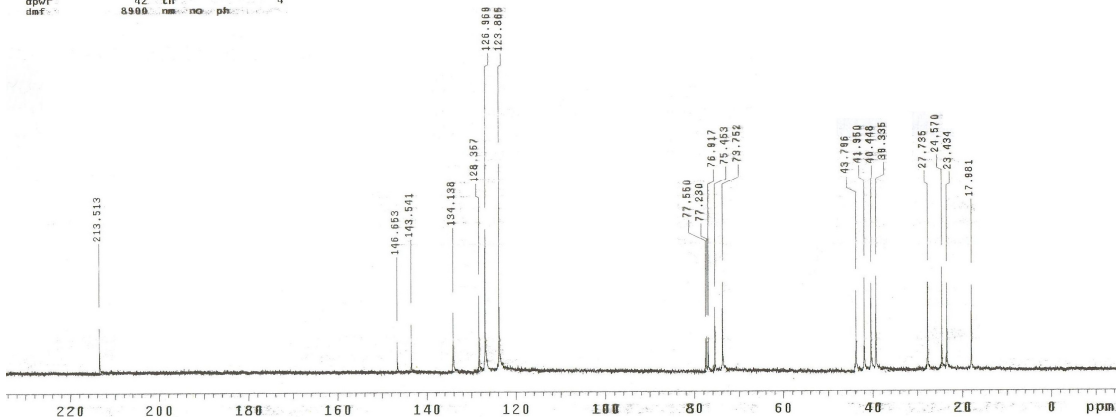
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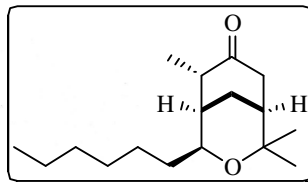
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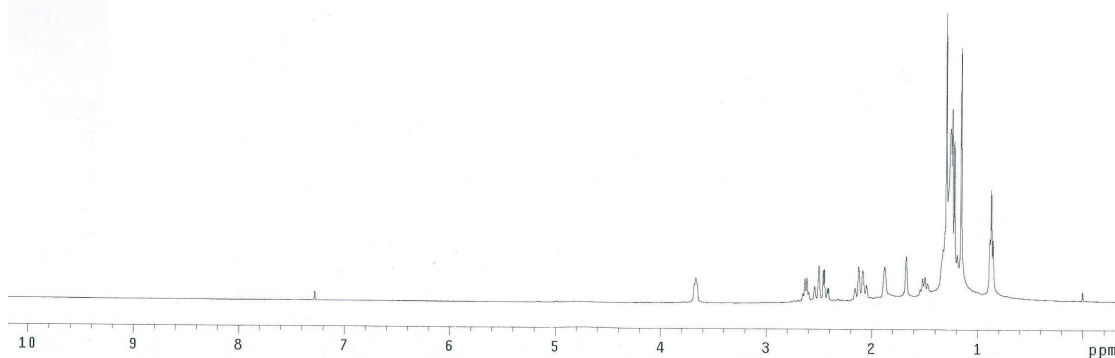
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 ^1H NMR (400 MHz, CDCl_3)

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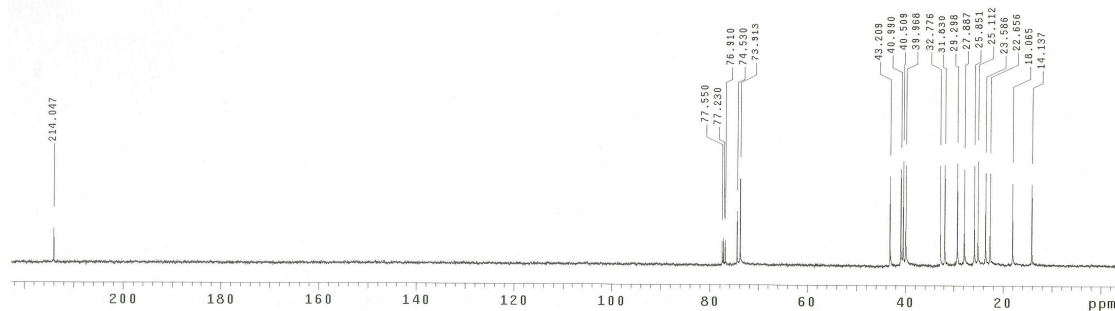
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nm no ph

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2.8 The Crystal Parameters of Compound 9i

Crystal Parameters	9i-CCDC 760090
Formula	C ₁₉ H ₂₃ NO ₄
Formula weight	329.38
<i>T</i> /K	296(2)
Crystal system	Monoclinic
Space group	P2(1)/c
<i>a</i> /Å	18.1890(4)
<i>b</i> /Å	10.0277(2)
<i>c</i> /Å	9.4674(2)
<i>α</i> /°	90.00
<i>β</i> /°	93.858 (2)
<i>γ</i> /°	90.00
<i>V</i> /Å ³	1722.88(6)
<i>Z</i>	4
Abs. Coeff./mm ⁻¹	0.089
Abs. Correction	Multi-Scan
GOF on <i>F</i> ²	0.993
Final <i>R</i> indices	<i>RI</i> = 0.0507
[<i>I</i> > 2σ(<i>I</i>)]	<i>wRI</i> = 0.1505
<i>R</i> indices [all data]	<i>RI</i> = 0.0880
	<i>wRI</i> = 0.1660

CHAPTER 3

Synthesis of Oxabicyclo[3.3.1]nonenes and Substituted Tetrahydropyrans via (3,5)-Oxonium-Ene Reaction

3.1 Importance and Applications

Oxabicyclo[3.3.1]nonenes and selected derivatives (Figure 3.1.1, A-B) are known to behave as estrogen receptor ligands.¹ Estrogen receptors (ER) have quickly emerged as attractive targets for therapeutic invention in a wide variety of diseases, including osteoporosis² and cancer.³ On the other hand, tetrahydropyrans are found in many biologically active natural products and pharmaceuticals.⁴ Alkyl substituted tetrahydropyrans (Figure 3.1.1, C) are used as aroma and flavouring substances for pharmaceuticals, cosmetics and food stuff.⁵

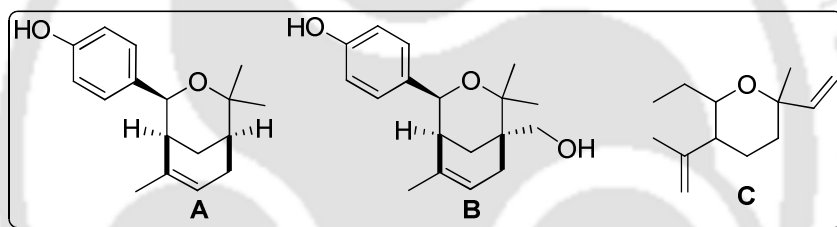
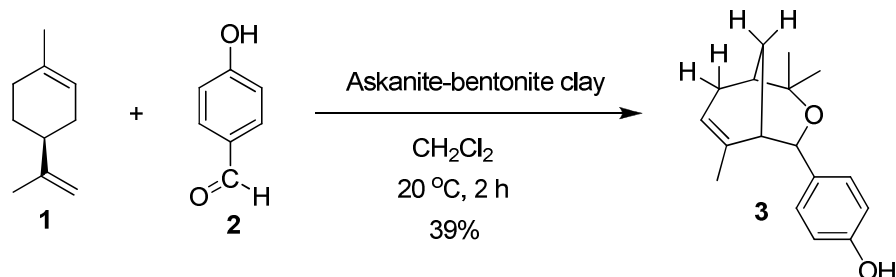


Figure 3.1.1. Biologically active cyclic ethers

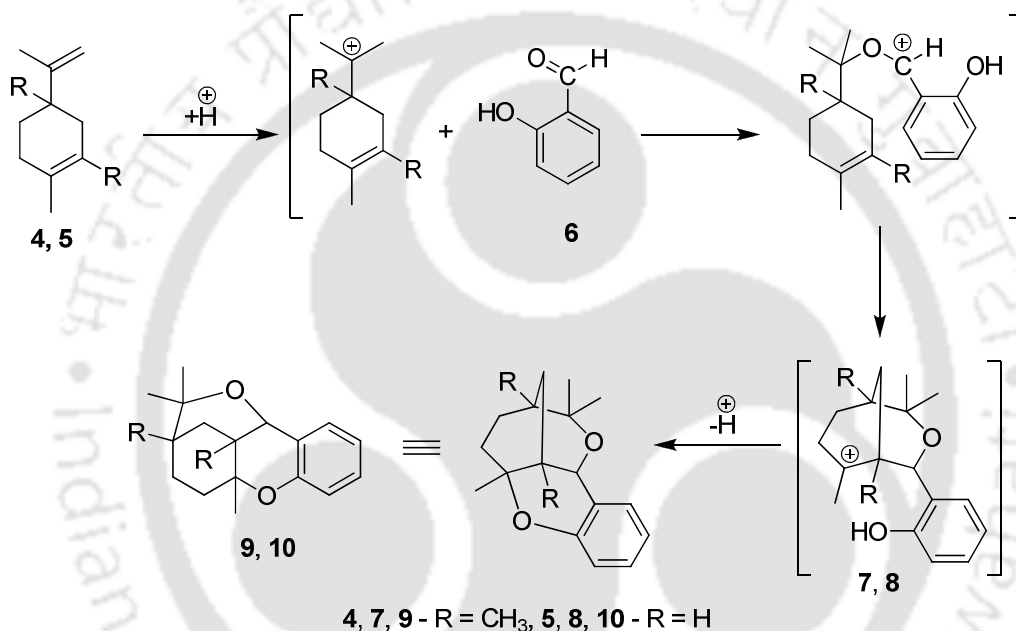
3.2 An Overview of Relevant Synthetic Methods

Review of literature reveals that oxabicyclo[3.3.1]nonenes are prepared by the reaction of aromatic aldehydes with naturally occurring terpenoids or olefin-containing compounds in acidic media.^{1,6} The low yield and multistep reactions are the major draw back of those methods.^{1,6} For example, Barkhash and co-workers reported the synthesis of 2,2,6-trimethyl-4-(4-hydroxyphenyl)-3-oxabicyclo[3.3.1]non-6-ene **3** from the reaction of *p*-hydroxybenzaldehyde **2** with R-(+)-limonene **1** in the presence of askanite-bentonite clay.



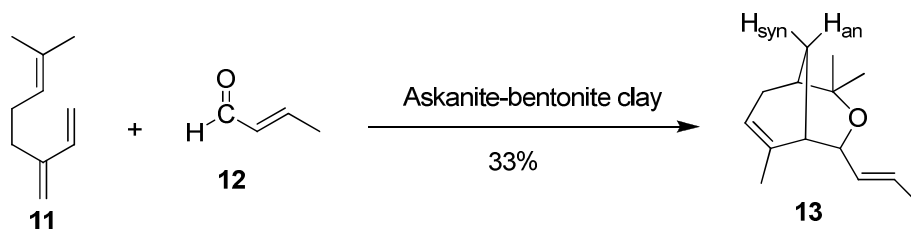
Scheme 3.2.1.

The oxabicyclic compound was obtained only in 39% yield (Scheme 3.2.1).^{6a} A very careful observation from the same group also revealed that in the case of the interaction of dienes **4** or **5** with salicylaldehyde **6** (in the presence of askanite-bentonite clay, 20 °C, CH₂Cl₂, 6-9 hr) the process does not stop as in the previous cases, at the stage of the formation of bicyclic ethers. The cationic center in the presumed intermediates **7** and **8** could be trapped by the phenolic hydroxy group to yield 1,2,2,8,12-pentamethyl-3,7-dioxa-5,6-benzotricyclo[6.2.2.0^{4,12}]dodecane **9** (yield 75%) and 2,2,8-trimethyl-3,7-dioxa-5,6-benzotricyclo[6.2.2.0^{4,12}]dodecane **10** (yield 29%) (Scheme 3.2.2).^{6a}



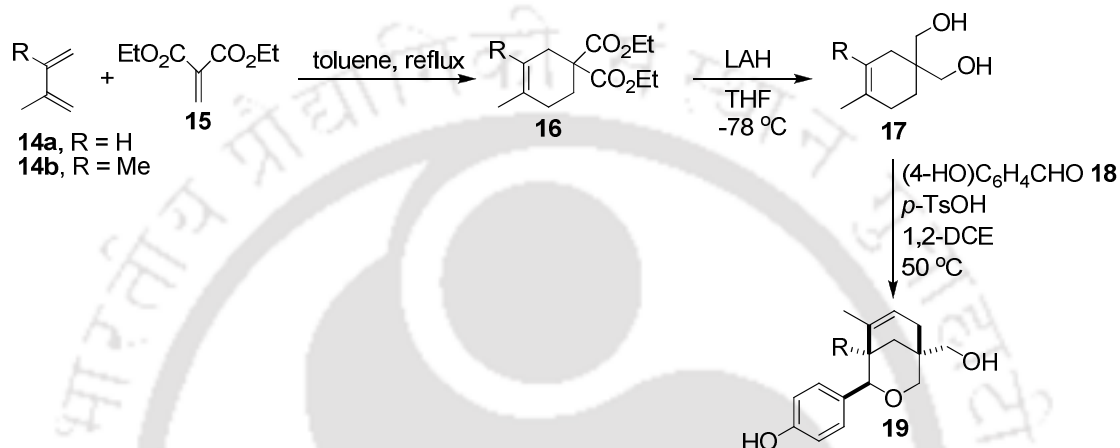
Scheme 3.2.2.

Salakhutdinov *et al.* reported the synthesis of 2,2,6-trimethyl-4-(prop-1-enyl)-3-oxabicyclo[3.3.1]non-6-ene **13** (Scheme 3.2.3) from the reaction of triene **11** with crotonaldehyde **12** in the presence of askanite-bentonite clay.^{6b} According to them this was an example of homo-Diels-Alder reaction as in the course of formation of bicyclic ether **11**, three π -bonds transform into three σ -bonds, producing two new rings.



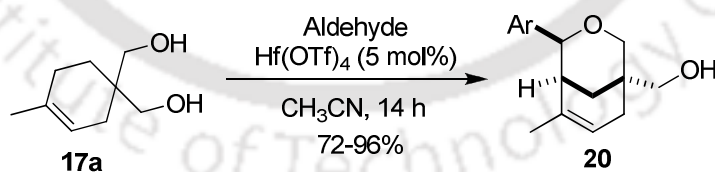
Scheme 3.2.3.

Hamann *et al.* reported the synthesis of oxabicyclo[3.3.1]nonene analogues **19** in good to excellent yields from the reaction of diols **17** with 4-hydroxybenzaldehyde **18** mediated by *p*-Toluenesulfonic acid (Scheme 3.2.4).^{1b} The starting diols **17** were synthesized by using Diels-Alder cycloaddition of isoprene or 2,3-dimethyl-1,3-butadiene **14a,b** with diethylmethylene malonate **15** to produce dicarboethoxycyclohexenes **16**, which were then reduced to the corresponding diols **17** with lithium aluminium hydride (Scheme 3.2.4).



Scheme 3.2.4.

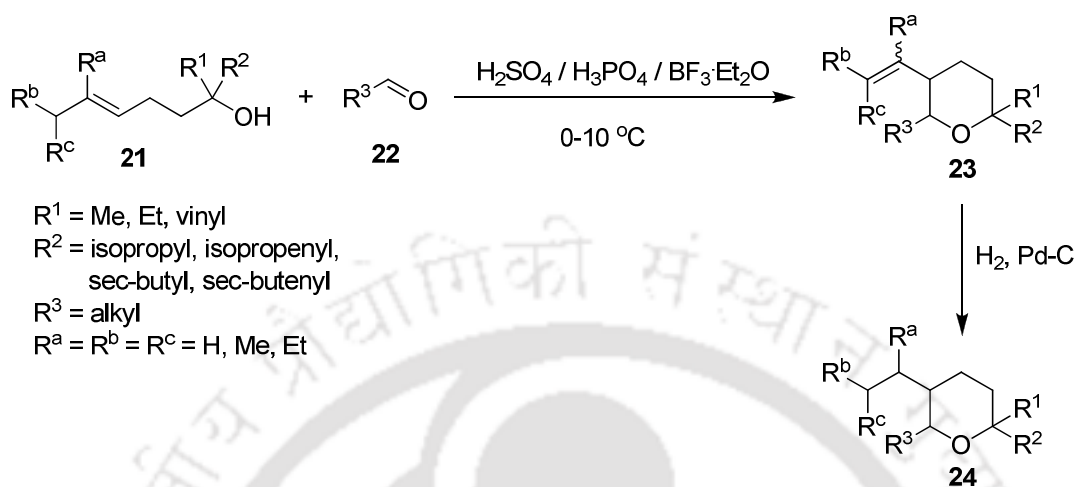
Nakamura *et al.* reported a versatile method for the synthesis of 4-substituted 6-methyl-3-oxabicyclo[3.3.1]non-6-ene-1-methanol derivatives **20** by using a Prins type cyclization reaction between diol **17a** and various aldehydes, including substituted benzaldehydes and heteroaromatic carbaldehydes catalyzed by hafnium triflate in good yields (Scheme 3.2.5).^{1d}



Scheme 3.2.5.

On the other hand method for the synthesis of alkyl substituted tetrahydropyrans are limited. Only one report is available in the literature and that has come from Oertling group. They have prepared alkyl substituted tetrahydropyrans from the reaction of alcohol component **21** with aldehyde **22** in the presence of acid catalysts like sulfuric acid, phosphoric acid and boron trifluoride etherate to the corresponding cyclized tetrahydropyran compound **23**

which on catalytic hydrogenation gives saturated tetrahydropyran compound **24** (Scheme 3.2.6).⁵



Scheme 3.2.6.

3.3 Present Work

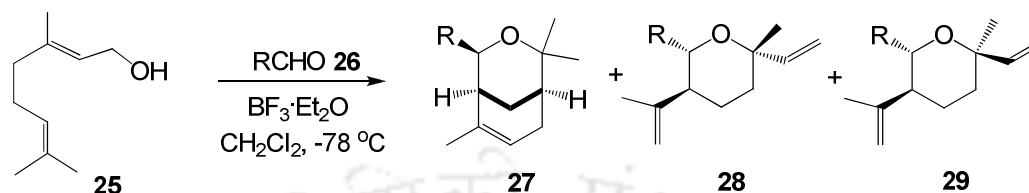
In the previous chapter we have described a novel methodology for the synthesis of oxabicyclo[3.3.1]nonanone by using (3,5)-oxonium-ene reaction. In this chapter an efficient methodology for the synthesis of oxabicyclo[3.3.1]nonenes and substituted tetrahydropyrans from the reaction of geraniol with aldehydes and epoxides promoted by boron trifluoride etherate is presented.

Taking clues from our previous work⁷ and the literature,⁶ and the fact that geraniol gives α -terpineol under acidic conditions; we envisioned that treatment of geraniol **25** with boron trifluoride etherate would provide carbocation **30**, which after cyclization and subsequent nucleophilic attack by aldehyde would give oxocarbenium ion **32**. The oxocarbenium ion **32** after (3,5)-oxonium-ene cyclization would give the oxabicyclic compound **27**. Accordingly, geraniol was reacted with benzaldehyde in dry dichloromethane using boron trifluoride etherate as Lewis acid and 2,2,6-trimethyl-4-phenyl-3-oxabicyclo[3.3.1]non-6-ene **27a** was obtained in 52% yield along with an inseparable mixture of two substituted tetrahydropyrans **28a** and **29a** with 25% overall yield. The reaction is generalized as shown in Table 3.3.1.

To prove its general applicability different types of aromatic and aliphatic aldehydes were subjected to react with geraniol and it was observed that aromatic aldehydes and cyclohexanecarboxaldehyde give oxabicyclo[3.3.1]nonene as the major product (Table 3.3.1), whereas simple aliphatic aldehydes give tetrahydropyrans as major product (Scheme

3.3.2). This might be due to the better nucleophilicity and smaller size of the aliphatic aldehydes compared to the aromatic aldehydes, which makes it an easy approach to the

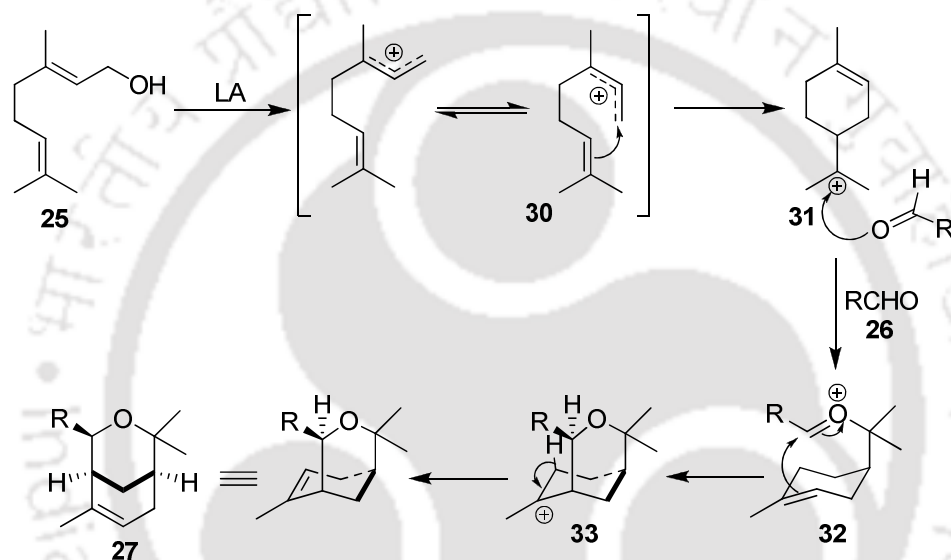
Table 3.3.1. Reaction of geraniol with aldehydes



Entry	Aldehyde R =	Time/h	Product ratio 27:28:29 ^a	Product	
				27 Yield ^b (%)	28+29 Yield ^c (%)
a	C ₆ H ₅	2.5	4:2:1	52	25
b	<i>p</i> -Cl-C ₆ H ₄	2.5	1:0:0	70	0
c	<i>p</i> -Br-C ₆ H ₄	2.5	1:0:0	70	0
d	<i>p</i> -F-C ₆ H ₄	2.5	1:0:0	68	0
e	<i>p</i> -NO ₂ -C ₆ H ₄	3	2:1:0	50	15
f	<i>o</i> -NO ₂ -C ₆ H ₄	3	1:0:0	54	0
g	C ₁₀ H ₇ (2-Naphthyl)	2.5	1:0:0	55	0
h	C ₆ H ₄ -CH=CH	2	1:0:0	62	0
i	<i>p</i> -NO ₂ -C ₆ H ₄ -CH=CH	2	1:0:0	55	0
j	<i>p</i> -MeO ₂ C-C ₆ H ₄	3	1:0:0	57	0
k	<i>p</i> -OH-C ₆ H ₄	3	1:0:0	60	0
l	<i>p</i> -CH ₃ -C ₆ H ₄	2.5	1:0:0	68	0
m	Cyclo-C ₆ H ₁₁	2	1:0:0	62	0
n	C ₂ H ₅	2	1:3:1	8	65
o	C ₃ H ₇	2	1:3:1	8	65
p	(CH ₃) ₂ CHCH ₂	2	1:3:1	10	67
q	CH ₃ (CH ₂) ₅	2	1:3:1	15	60
r	CH ₃ (CH ₂) ₁₄	2	1:3:1	10	62

^aRatios are determined by ¹HNMR of crude product. ^bYields refer to isolated yield. The compounds are characterized by IR, ¹H, ¹³C NMR and Mass spectroscopy. ^cTotal isolated yield of **28** and **29**.

carbocation **30** rather than allowing it to transform into **31** (Scheme 3.3.1). Among the aromatic aldehydes, simple benzaldehyde and 4-nitro benzaldehyde gave a small amount of tetrahydropyrans. The formation of some tetrahydropyrans in the case of benzaldehyde and *p*-nitrobenzaldehyde might be due to the electron-withdrawing effect of phenyl and *p*-nitrophenyl ring, which destabilizes the oxocarbenium ion. On the other hand, the electron-donating groups on aromatic ring (entry c-d, k-l) stabilize the oxocarbenium ion.⁸ However, the formation of only one tetrahydropyran in case of 4-nitrobenzaldehyde is quite uncommon. The structure and stereochemistry of bicyclic compounds and substituted



Scheme 3.3.1. Mechanism for the formation of oxabicyclic compounds

tetrahydropyrans were determined by ¹H NMR and NOE experiments (Figure 3.3.1). The coupling constant of H_{4e} (4.88 ppm) proton of **27a** is 2.0 Hz, which is typical for vicinal diequatorial coupling and therefore, H_{4e} and H_{5e} (2.19-2.21 ppm) are in equatorial position. This is in accordance with the Dreiding model where H_{5e} proton cannot be axial. Again the coupling constant between H_{1e} (1.56 ppm) and H_{8a} (2.32-2.43 ppm) is 3.2 Hz which indicates that one proton is in an axial position and the other in an equatorial position. The presence of a strong NOE between H_{8a} and H_{9syn} (1.73 ppm) confirms that they are in a 1,3-diaxial position. Therefore, H_{1e} is in an equatorial position. This is also in accordance with the Dreiding model. Thus H_{5e} and H_{1e} are in equatorial positions and hence the fusion between two rings is *cis* fusion (**27a**, Figure 3.3.1). Similarly, the stereochemistry of substituted tetrahydropyran **28e** was determined as follows. The coupling constant of H_{6a} (4.56 ppm) is 10.0 Hz and it indicates that H_{5a} (2.11-2.19 ppm) and H_{6a} are in axial

positions. The presence of NOE between H_{6a} and C-7 methyl protons (1.31 ppm) and between H_{5a} and H_{3a} (1.80-1.92) further confirms that the protons H_{5a} and H_{6a} are in axial position and are *trans* to each other, whereas H_{6a} and C-7 methyl protons are *cis* (**28e**, Figure 3.3.1).

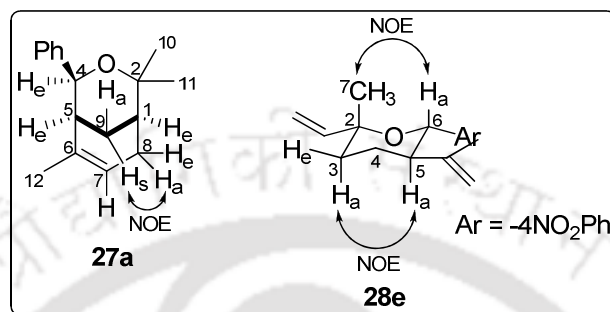
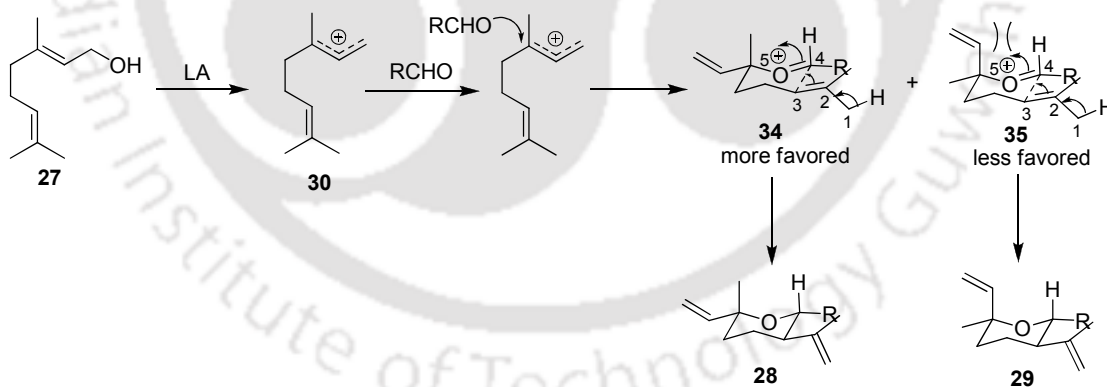


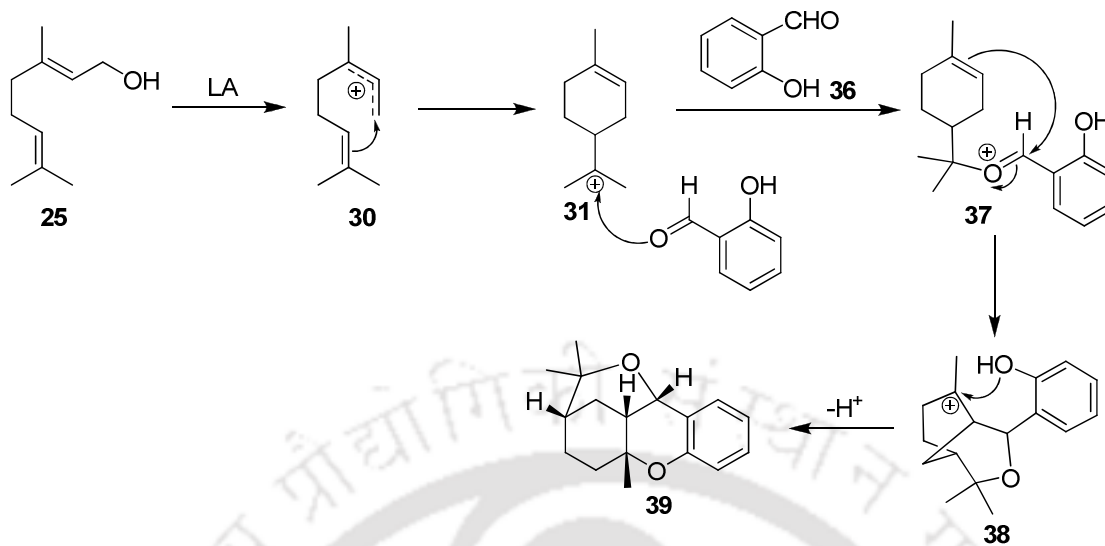
Figure 3.3.1. NOE of **27a** and **28e**

The mechanism of the formation of tetrahydropyrans can be represented as shown in Scheme 3.3.2. Here the smaller aliphatic aldehydes attack the carbocation **30** to form more favored transition state **34**, which after cyclization gives the more stable isomer **28** as a major product and a minor product **29** from the less favored transition state **35** (Scheme 3.3.2).⁹



Scheme 3.3.2.

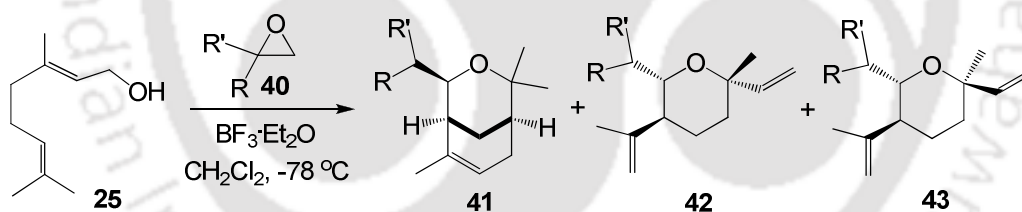
In the case of salicylaldehyde **36** a tetracyclic compound **39** was obtained in 40% yield. The formation of **39** can be explained as shown in Scheme 3.3.3. Here carbocation **31** is attacked by phenolic group to give tetracyclic compound **39**. The structure and stereochemistry of the compound was determined from ^1H NMR, COSY and NOE experiments.

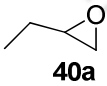

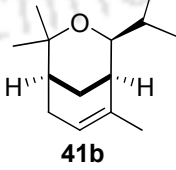
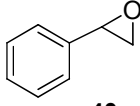
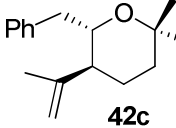
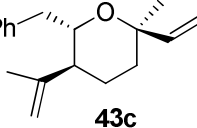


Scheme 3.3.3. Reaction with salicylaldehyde

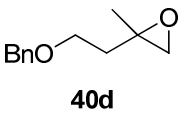
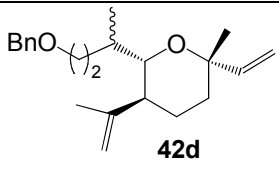
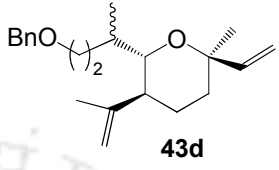
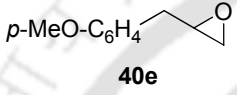
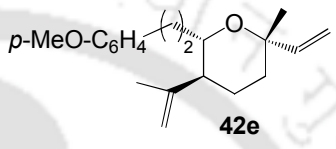
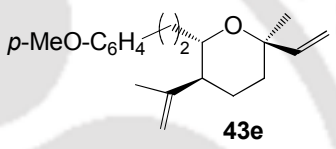
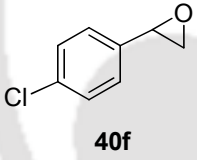
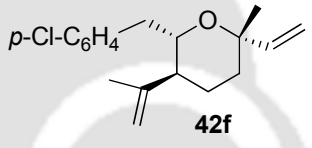
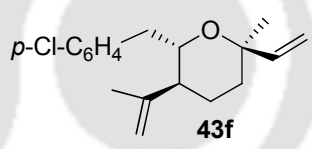
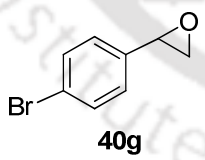
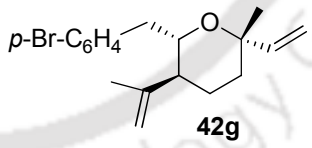
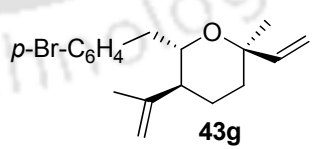
The reaction of geraniol with epoxides was also performed and the results are summarized in Table 3.3.2. It was observed that monosubstituted terminal epoxides are non-reactive, whereas the 2,2-disubstituted and styrene oxides are active and give good to moderate

Table 3.3.2. Reaction of geraniol with epoxides



Entry	Epoxide	Time/h	Products	Yield ^a (%)
1	 40a	12	No reaction	0
2	 40b	3	 41b	45
3	 40c	3	 42c  43c 42c:43c = 3:2 ^b	52

continue...

Entry	Epoxide	Time/h	Products	Yield ^a (%)
4	 40d	3	 42d  43d 42d:43d = 2:1 ^b	48
5	 40e	3	 42e  43e 42e:43e = 2:1 ^b	45
6	 40f	3	 42f  43f 42f:43f = 3:2 ^b	48
7	 40g	3	 42g  43g 42g:43g = 3:2 ^b	48

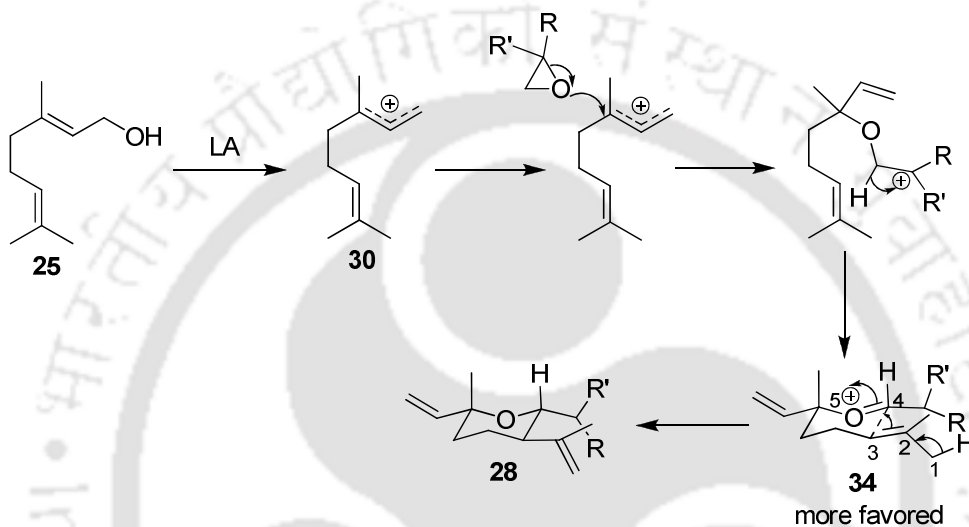
^aYields refer to isolated yield. The compounds are characterized by IR, ¹H, ¹³C NMR and Mass spectroscopy.

^bRatios are determined by ¹H NMR of crude products.

yields.⁷ Interestingly monosubstituted benzylic terminal epoxide (entry 5) was also found to give corresponding tetrahydropyrans. The formation of substituted tetrahydropyrans **42c-42g** and **43c-43g** from epoxides **40c-40g** are in accordance with the facts already discussed; but the formation of bicyclic compound **41b** in case of 2,2-dimethyl oxirane **40b** is not clear

(Table 3.3.2).

The mechanism of the formation of tetrahydropyrans can be explained as follows. The carbocation **30** is attacked by epoxide to give oxocarbenium intermediate **34**, which after oxonium-ene cyclization gives tetrahydropyran **28** (Scheme 3.3.4). In the case of 2,2-dimethyl oxirane the carbocation **30** rearranges to **31** and then attacked by epoxide to give the oxabicyclic compound **41b**.



Scheme 3.3.4. Mechanism of the reaction with epoxide

Conclusions:

In summary, an efficient methodology for the synthesis of oxabicyclo[3.3.1]nonenes and substituted tetrahydropyrans has been developed from readily available starting material in moderate to good yields. The bicyclic unit can further be transformed by manipulating olefinic functionality and therefore, can be used as a precursor for the synthesis of complex organic molecules.

3.4 Experimental Section

3.4.1 Instrumentation and Characterization

As described in Chapter 2, Section 2.4.1.

3.4.2 General Procedure for the Synthesis of Oxabicyclic and Substituted Tetrahydropyran from Aldehyde:

To a solution of geraniol (1 mmol, 1 equiv.) in dry CH_2Cl_2 (2 mL) at $-78\text{ }^\circ\text{C}$ was added boron trifluoride etherate (1.2 mmol, 1.2 equiv.). To this solution aldehyde (1 mmol, 1 equiv.) in dry CH_2Cl_2 (2 mL) was added drop by drop

over 5 min. The reaction mixture was stirred at that temperature for a specified time. The progress of the reaction was monitored by TLC with hexane as eluent. After completion of the reaction, the solvent was removed by rotary evaporator. The resultant residue was quenched with saturated solution of NaHCO_3 . The product was extracted with ethyl acetate and then washed with brine and water. The organic layer was dried (Na_2SO_4) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give the title compounds. All compounds were obtained as colourless oil or semisolid and therefore boiling points were not determined.

Synthesis of 2,2,6-Trimethyl-4-phenyl-3-oxabicyclo[3.3.1]non-6-ene (27a,

Table 3.3.1): To a solution of geraniol (154 mg, 1 mmol) in dry CH_2Cl_2 (2 mL) at $-78\text{ }^\circ\text{C}$, was added boron trifluoride etherate (170 mg, 1.2 mmol). To this solution benzaldehyde (106 mg, 1.0 mmol) in dry CH_2Cl_2 (2 mL) was added drop by drop over 5 minutes. The reaction mixture was stirred at that temperature for 2.5 h. The progress of the reaction was monitored by TLC with hexane as eluent. After completion of the reaction, the solvent was removed by rotary evaporator. The resultant residue was quenched with saturated NaHCO_3 solution. The product was extracted with ethyl acetate and then washed with brine and water. The organic layer was dried (Na_2SO_4) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give **27a** as a colourless oil (126 mg, 52%).

3.4.3 General Procedure for the Synthesis of Oxabicyclic and Substituted Tetrahydropyran from Epoxide:

To a solution of geraniol (1 mmol, 1 equiv.) in dry CH_2Cl_2 (2 mL) at $-78\text{ }^\circ\text{C}$, was added boron trifluoride etherate (1.2 mmol, 1.2 equiv.). To this solution epoxide (2 mmol, 2 equiv) in dry CH_2Cl_2 (2 mL) was added drop by drop over 5 minutes. Then the temperature of the reaction was brought to room temperature. The reaction mixture was stirred at that temperature for a specified time. The progress of the reaction was monitored by TLC with hexane as eluent. After completion of the reaction, the solvent was removed by rotary evaporator. The resultant residue was quenched with saturated solution of NaHCO_3 . The product was extracted with ethyl acetate and then washed with brine and water. The organic layer was dried (Na_2SO_4) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give the title compounds.

Synthesis of 4-Isopropyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene

(41b, Table 3.3.2): To a solution of geraniol (154 mg, 1 mmol) in dry CH₂Cl₂ (2 mL) at -78 °C, was added boron trifluoride etherate (170 mg, 1.2 mmol). To this solution 1,2-epoxy-2-methylpropane (144 mg, 2.0 mmol) in dry CH₂Cl₂ (2 mL) was added drop by drop over 5 min. The temperature was slowly brought to room temperature. The reaction mixture was stirred at that temperature for 3 h. The progress of the reaction was monitored by TLC with hexane as eluent. After completion of the reaction, the solvent was removed by rotary evaporator. The resultant residue was quenched with saturated NaHCO₃ solution. The product was extracted with ethyl acetate and then washed with brine and water. The organic layer was dried (Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give **41b** as a colourless oil (94 mg, 45%).

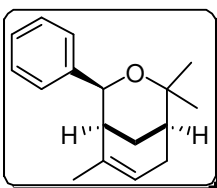
3.5 References

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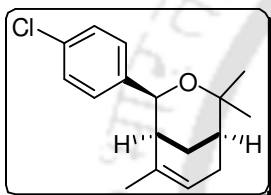
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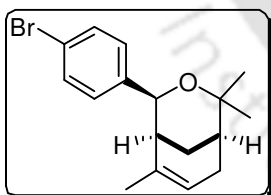
3.6 Spectral Data

2,2,6-Trimethyl-4-phenyl-3-oxabicyclo[3.3.1]non-6-ene (27a):

Colourless oil (126 mg, 52%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.86 (m, 3 H), 1.34 (s, 3 H), 1.40 (s, 3 H), 1.56 (ddd, $J = 6.4, 3.2$ and 3.2 Hz, 1 H), 1.73 (ddd, $J = 12.4, 3.2$ and 3.2 Hz, 1 H), 2.08 (dddq, $J = 18.8, 6.3, 3.2$ and 2.4 Hz, 1 H), 2.19-2.21 (m, 1 H), 2.32-2.43 (m, 2 H), 4.88 (d, $J = 2.0$ Hz, 1 H), 5.40-5.44 (m, 1 H), 7.17-7.20 (m, 1 H), 7.25-7.28 (m, 2 H), 7.30-7.33 (m, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.2, 24.3, 27.9, 28.5, 28.9, 34.2, 41.8, 74.4, 75.5, 123.5, 125.8, 126.7, 128.0, 133.4, 143.1; **IR** (Neat): 2935, 1450, 1379, 1207, 1084, 1057, 755, 700 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{22}\text{O}$ (M^+) requires 242.1671; found 242.1663.

4-(4-Chlorophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27b):

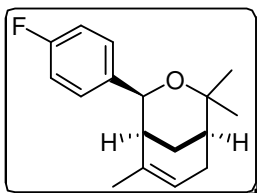
Colourless oil (194 mg, 70%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.88 (m, 3 H), 1.32 (s, 3 H), 1.38 (s, 3 H), 1.56 (ddd, $J = 6.0, 3.2$ and 3.2 Hz, 1 H), 1.73 (ddd, $J = 12.4, 3.2$ and 3.2 Hz, 1 H), 2.08 (dddq, $J = 19.2, 6.4, 3.2$ and 2.4 Hz, 1 H), 2.15-2.18 (m, 1 H), 2.30-2.42 (m, 2 H), 4.84 (d, $J = 2.0$ Hz, 1 H), 5.44 (brs, 1 H), 7.22-7.27 (m, 4 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.2, 24.4, 27.9, 28.4, 28.8, 34.1, 41.6, 73.9, 75.7, 123.9, 127.4, 128.2, 132.4, 132.9, 141.7; **IR** (Neat): 2933, 1635, 1464, 1380, 1226, 1075, 1053, 801, 677 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{ClO}$ (M^+) requires 276.1281; found 276.1269.

4-(4-Bromophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27c):

Colourless oil (225 mg, 70%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.88 (m, 3 H), 1.32 (s, 3 H), 1.37 (s, 3 H), 1.56 (ddd, $J = 6.0, 3.2$ and 2.8 Hz, 1 H), 1.73 (ddd, $J = 12.4, 3.2$ and 2.8 Hz, 1 H), 2.08 (dddq, $J = 18.8, 6.0, 3.2$ and 2.8 Hz, 1 H), 2.15-2.17 (m, 1 H), 2.30-2.42 (m, 2 H), 4.81 (d, $J = 2.0$ Hz, 1 H), 5.40-5.45 (m, 1 H), 7.19 (d, $J = 8.8$ Hz, 2 H), 7.39 (d, $J = 8.8$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.1, 24.5, 27.8, 28.4, 28.8, 34.0, 41.6, 73.9, 75.7, 120.5, 123.9, 127.7, 131.1, 132.9, 142.2; **IR** (Neat): 2932, 1461, 1369, 1204, 1072, 1056, 806, 801 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{BrO}$ (M^+) requires 320.0776; found 320.0789.

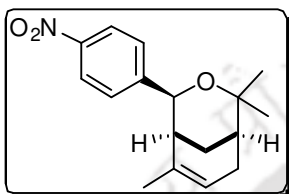
4-(4-Fluorophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27d):

Colourless oil (177 mg, 68%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.87 (s, 3 H), 1.33 (s, 3 H), 1.39 (s, 3 H), 1.54-1.62 (m, 1 H), 1.72 (ddd, $J = 12.4, 3.2$ and 3.2 Hz, 1 H), 2.05-2.16 (m, 2 H), 2.31-2.42 (m, 2 H), 4.85 (s, 1 H), 5.40-5.45 (m, 1 H), 6.96 (t, $J = 8.4$ Hz, 2 H), 7.28 (t,



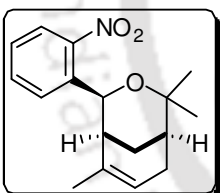
$J = 7.6\text{ Hz}$, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.2, 24.3, 27.9, 28.4, 28.9, 34.1, 41.7, 73.9, 75.7, 114.8 (d, $J = 21.3$ Hz), 123.8, 127.4 (d, $J = 7.7$ Hz), 133.0, 138.9, 162.0 (d, $J = 242.5$ Hz); $^{19}\text{F NMR}$ (376 MHz, $\text{CDCl}_3/\text{C}_6\text{F}_6$): δ -220.69-(-220.77) (m, 1 F); **IR** (Neat): 2932, 1626, 1509, 1377, 1227, 1074, 1056, 824, 798 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{FO}$ (M^+) requires 260.1576; found 260.1565.

4-(4-Nitrophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27e):



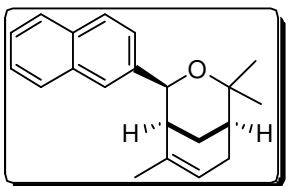
Colourless oil (144 mg, 50%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.82 (m, 3 H), 1.35 (s, 3 H), 1.40 (s, 3 H), 1.59 (ddd, $J = 6.0$, 3.2 and 2.8 Hz, 1 H), 1.77 (ddd, $J = 12.8$, 3.6 and 3.2 Hz, 1 H), 2.10 (dddq, $J = 18.4$, 6.4, 3.2 and 2.8 Hz, 1 H), 2.20-2.28 (m, 1 H), 2.35-2.43 (m, 2 H), 4.95 (s, 1 H), 5.47 (brs, 1 H), 7.50 (d, $J = 8.4$ Hz, 2 H), 8.14 (d, $J = 8.8$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.1, 24.4, 27.8, 28.4, 28.7, 33.9, 41.6, 74.0, 76.0, 123.3, 124.6, 126.8, 132.1, 147.0, 150.9; **IR** (Neat): 2934, 1600, 1520, 1346, 1229, 1077, 1057, 851, 707 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{NO}_3$ ($\text{M}+\text{H}^+$) requires 288.1599; found 288.1609.

4-(2-Nitrophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27f):



Colourless oil (155 mg, 54%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.85 (m, 3 H), 1.33 (s, 3 H), 1.40 (s, 3 H), 1.59 (ddd, $J = 6.4$, 3.2, and 3.2 Hz, 1 H), 1.72 (ddd, $J = 12.4$, 3.2 and 2.8 Hz, 1 H), 2.11 (dddq, $J = 18.8$, 6.0, 3.2 and 2.8 Hz, 1 H), 2.36-2.50 (m, 3 H), 5.45-5.47 (m, 1 H), 5.49-5.52 (m, 1 H), 7.36 (t, $J = 8.4$ Hz, 1 H), 7.55 (t, $J = 8.0$ Hz, 1 H), 7.80 (d, $J = 8.0$ Hz, 1 H), 7.93 (d, $J = 8.4$ Hz, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.2, 24.3, 27.8, 28.2, 28.8, 33.9, 39.1, 70.1, 76.2, 124.2, 124.5, 127.6, 129.5, 132.4, 133.3, 138.3, 147.5; **IR** (Neat): 2933, 1525, 1345, 1228, 1067, 1055, 746, 709 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{NO}_3$ (M^+) requires 287.1521; found 287.1528.

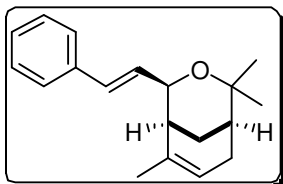
2,2,6-Trimethyl-4-naphthyl-3-oxabicyclo[3.3.1]non-6-ene (27g):



Colourless oil (161 mg, 55%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.79 (m, 3 H), 1.40 (s, 3 H), 1.42 (s, 3 H), 1.59 (ddd, $J = 6.4$, 3.6 and 2.8 Hz, 1 H), 1.76 (ddd, $J = 12.0$, 3.6 and 3.2 Hz, 1 H), 2.10 (dddq, $J = 18.4$, 6.4, 3.2 and 2.8 Hz, 1 H), 2.32-2.48 (m, 3 H), 5.03 (d, $J = 1.2$ Hz, 1 H), 5.46 (brs, 1 H), 7.38-7.45 (m, 3 H), 7.73-7.82 (m, 4 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.3, 24.4, 27.9, 28.7, 28.9, 34.2, 41.6, 74.6, 75.7, 123.6, 124.1, 124.7, 125.4, 125.8, 127.5, 127.7, 128.3, 132.8, 133.4, 133.5, 140.6; **IR** (Neat): 2931, 1461,

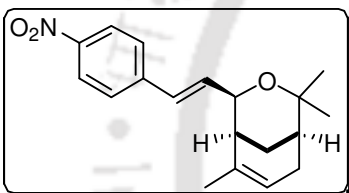
1368, 1284, 1126, 1097, 1078, 818, 776, 739 cm^{-1} . **HRMS** (APCI) cald. For $\text{C}_{21}\text{H}_{24}\text{O}$ (M^+) requires 292.1827; found 292.1818.

2,2,6-Trimethyl-4-styryl-3-oxabicyclo[3.3.1]non-6-ene (27h):



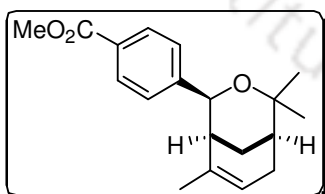
Colourless oil (166 mg, 62%); **^1H NMR** (400 MHz, CDCl_3): δ 1.29 (s, 3 H), 1.37 (s, 3 H), 1.64-1.68 (m, 1 H), 1.69 (s, 3 H), 1.70 (ddd, $J = 12.0, 3.6$ and 3.2 Hz, 1 H), 2.06-2.10 (m, 1 H), 2.11-2.16 (m, 1 H), 2.22-2.29 (m, 1 H), 2.35-2.44 (m, 1 H), 4.47 (d, $J = 5.6$ Hz, 1 H), 5.55-5.60 (m, 1 H), 6.16 (dd, $J = 16.0$ and 6.4 Hz, 1 H), 6.56 (d, $J = 16.4$ Hz, 1 H), 7.18-7.21 (m, 1 H), 7.27-7.32 (m, 2 H), 7.34-7.38 (m, 2 H); **^{13}C NMR** (100 MHz, CDCl_3): δ 24.3, 25.3, 27.9, 28.2, 28.8, 34.2, 40.7, 74.7, 75.6, 124.0, 126.6, 127.4, 128.6, 129.0, 131.4, 133.5, 137.4; **IR** (Neat): 2930, 1646, 1456, 1374, 1226, 1083, 1061, 968, 746, 694 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{19}\text{H}_{24}\text{O}$ (M^+) requires 268.1827; found 268.1838.

2,2,6-Trimethyl-4-[2-(4-nitrophenyl)-vinyl]-3-oxabicyclo[3.3.1]non-6-ene (27i):



Colourless oil (172 mg, 55%); **^1H NMR** (400 MHz, CDCl_3): δ 1.30 (s, 3 H), 1.37 (s, 3 H), 1.56 (ddd, $J = 5.6, 2.8$ and 2.8 Hz, 1 H), 1.64 (s, 3 H), 1.73 (ddd, $J = 12.4, 3.2$ and 2.8 Hz, 1 H), 2.08-2.17 (m, 2 H), 2.23-2.29 (m, 1 H), 2.34-2.42 (m, 1 H), 4.51 (d, $J = 5.2$ Hz, 1 H), 5.56 (brs, 1 H), 6.37 (dd, $J = 16.0$ and 5.6 Hz, 1 H), 6.64 (d, $J = 16.0$ Hz, 1 H), 7.48 (d, $J = 8.8$ Hz, 2 H), 8.15 (d, $J = 8.8$ Hz, 2 H); **^{13}C NMR** (100 MHz, CDCl_3): δ 24.1, 25.0, 27.8, 28.1, 28.6, 34.0, 40.2, 73.7, 75.7, 124.0, 124.3, 126.5, 126.8, 132.7, 136.5, 143.9, 146.7; **IR** (Neat): 2930, 1596, 1518, 1341, 1083, 1050, 973, 858, 746 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{19}\text{H}_{23}\text{NO}_3$ (M^+) requires 313.1678; found 313.1686.

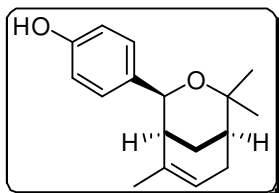
4-(4,4,8-Trimethyl-3-oxabicyclo[3.3.1]non-7-en-2-yl)-benzoic acid methyl ester (27j):



Colourless oil (171 mg, 57%); **^1H NMR** (400 MHz, CDCl_3): δ 0.82 (m, 3 H), 1.34 (s, 3 H), 1.40 (s, 3 H), 1.57 (ddd, $J = 6.4, 3.6$ and 2.8 Hz, 1 H), 1.74 (ddd, $J = 12.8, 3.2$ and 2.8 Hz, 1 H), 2.08 (dddq, $J = 18.8, 6.0, 3.2$ and 2.8 Hz, 1 H), 2.22-2.29 (m, 1 H), 2.32-2.43 (m, 2 H), 3.89 (s, 3 H), 4.92 (s, 1 H), 5.44 (brs, 1 H), 7.40 (d, $J = 8.0$ Hz, 2 H), 7.95 (d, $J = 8.4$ Hz, 2 H); **^{13}C NMR** (100 MHz, CDCl_3): δ 24.1, 24.3, 27.8, 28.5, 28.8, 34.0, 41.6, 52.2, 74.3, 75.7, 124.0, 125.9, 128.6, 129.4, 132.8, 148.5, 167.4; **IR** (Neat): 2952, 1720, 1436, 1278, 1107, 1079, 1017, 768, 713 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{19}\text{H}_{24}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 301.1803; found 301.1808.

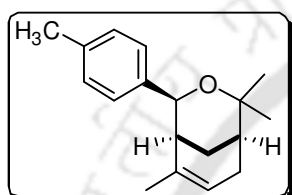
4-(4,4,8-Trimethyl-3-oxabicyclo[3.3.1]non-7-en-2-yl)-phenol (27k):

Colourless oil (156 mg, 60%); **^1H NMR** (400 MHz, CDCl_3): δ 0.91 (m, 3 H), 1.32 (s, 3 H),



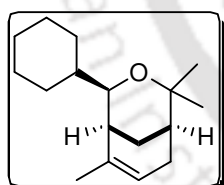
1.38 (s, 3 H), 1.54 (ddd, $J = 6.0, 3.2$ and 2.8 Hz, 1 H), 1.71 (ddd, $J = 12.4, 3.2$ and 2.8 Hz, 1 H), 2.08 (dddq, $J = 18.8, 6.0, 3.2$ and 2.8 Hz, 1 H), 2.12-2.17 (m, 1 H), 2.30-2.42 (m, 2 H), 4.91 (brs, 1 H), 5.27 (brs, 1 H), 5.44 (brs, 1 H), 6.73 (d, $J = 8.4$ Hz, 2 H), 7.17 (d, $J = 8.0$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.1, 24.4, 27.8, 28.3, 28.8, 34.2, 41.7, 74.2, 75.7, 115.0, 123.4, 127.0, 133.4, 134.9, 154.6; **IR** (Neat): 3372, 2929, 1611, 1457, 1377, 1227, 1168, 1075, 1053, 828 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{22}\text{O}_2$ ($\text{M}+\text{H}$)⁺ requires 259.1698; found 259.1701.

2,2,6-Trimethyl-4-*p*-tolyl-3-oxabicyclo[3.3.1]non-6-ene (27l):



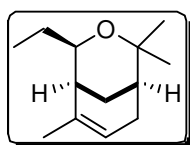
Colourless oil (174 mg, 68%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.88 (m, 3 H), 1.32 (s, 3 H), 1.38 (s, 3 H), 1.54 (ddd, $J = 6.0, 3.2$ and 2.8 Hz, 1 H), 1.71 (ddd, $J = 12.4, 3.2$ and 3.2 Hz, 1 H), 2.08 (dddq, $J = 18.4, 6.4, 3.6$ and 2.8 Hz, 1 H), 2.14-2.19 (m, 1 H), 2.30 (s, 3 H), 2.30-2.42 (m, 2 H), 4.84 (brs, 1 H), 5.43 (brs, 1 H), 7.06 (d, $J = 8.0$ Hz, 2 H), 7.20 (d, $J = 8.0$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 21.2, 24.1, 24.4, 27.8, 28.4, 28.8, 34.0, 41.6, 74.2, 75.3, 123.2, 125.6, 128.5, 133.4, 135.9, 140.0; **IR** (Neat): 2967, 2930, 1610, 1445, 1375, 1285, 1109, 1075, 1020, 806, 754 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{24}\text{O}$ (M^+) requires 256.1827; found 256.1827.

4-Cyclohexyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27m):



Colourless oil (154 mg, 62%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.70-1.00 (m, 4 H), 1.09-1.19 (m, 1 H), 1.20 (s, 3 H), 1.25 (s, 3 H), 1.46 (ddd, $J = 6.0, 3.2$ and 2.8 Hz, 1 H), 1.58-1.70 (m, 4 H), 1.74 (s, 3 H), 1.84-1.92 (m, 2 H), 2.00-2.19 (m, 4 H), 2.30-2.40 (m, 1 H), 3.25 (d, $J = 10.0$ Hz, 1 H), 5.53 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.2, 25.3, 25.8, 26.0, 26.9, 28.1, 28.7, 29.0, 29.5, 30.6, 34.4, 35.0, 40.1, 75.0, 76.9, 124.4, 133.7; **IR** (Neat): 2923, 2851, 1641, 1449, 1378, 1228, 1063, 1045, 880, 801 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{28}\text{O}$ (M^+) requires 248.2140; found 248.2144.

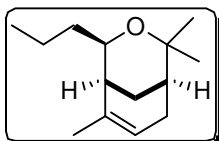
2,2,6-Trimethyl-4-ethyl-3-oxabicyclo[3.3.1]non-6-ene (27n):



Colourless oil (16 mg, 8%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.93 (t, $J = 7.6$ Hz, 3 H), 1.21 (s, 3 H), 1.28, (s, 3 H), 1.40 (quin, $J = 7.2$ Hz, 2 H), 1.48 (ddd, $J = 5.6, 3.2$ and 2.8 Hz, 1 H), 1.63 (ddd, $J = 12.4, 2.8$ and 2.8 Hz, 1 H), 1.73 (s, 3 H), 1.92-1.96 (m, 1 H), 2.06 (dddq, $J = 18.8, 6.4, 2.8$ and 2.8 Hz, 1 H), 2.09-2.14 (m, 1 H), 2.28-2.38 (m, 1 H), 3.56 (t, $J = 6.8$ Hz, 1 H), 5.51 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 11.1, 24.2, 25.4, 27.8, 28.0, 28.6, 28.9, 34.5, 38.1, 75.0, 75.6, 123.9,

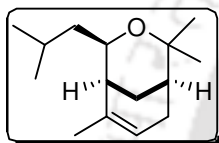
133.7; **IR** (Neat): 2959, 2931, 1648, 1458, 1377, 1108, 1068, 1042, 866, 743 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{13}\text{H}_{22}\text{O}$ (M^+) requires 194.1671; found 194.1677.

4-Propyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27o):



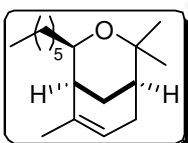
Colourless oil (17 mg, 8%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.90 (t, $J = 6.8$ Hz, 3 H), 1.20 (s, 3 H), 1.24-1.44 (m, 4 H), 1.29, (s, 3 H), 1.47 (ddd, $J = 6.0, 3.6$ and 2.8 Hz, 1 H), 1.62 (ddd, $J = 12.4, 3.2$ and 3.2 Hz, 1 H), 1.73 (s, 3 H), 1.88-1.92 (m, 1 H), 2.05 (dddq, $J = 18.8, 6.0, 3.2$ and 2.8 Hz, 1 H), 2.10-2.16 (m, 1 H), 2.30-2.38 (m, 1 H), 3.68 (ddd, $J = 7.6, 4.8$ and 1.6 Hz, 1 H), 5.52 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 19.7, 24.2, 25.4, 28.0, 28.7, 28.9, 34.5, 37.1, 38.5, 73.5, 74.9, 123.9, 133.7; **IR** (Neat): 2957, 2931, 1645, 1458, 1378, 1111, 1077, 886, 806 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{14}\text{H}_{24}\text{O}$ (M^+) requires 208.1827; found 208.1824.

4-Isobutyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27p):



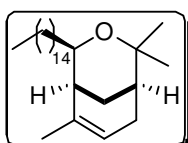
Colourless oil (22 mg, 10%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.87 (d, $J = 6.4$ Hz, 3 H), 0.89 (d, $J = 6.8$ Hz, 3 H), 1.08-1.16 (m, 1 H), 1.20 (s, 3 H), 1.30 (s, 3 H), 1.34-1.41 (m, 1 H), 1.48 (ddd, $J = 6.4, 3.2$ and 3.2 Hz, 1 H), 1.63 (ddd, $J = 12.0, 3.6$ and 3.2 Hz, 1 H), 1.74 (s, 3 H), 1.75-1.82 (m, 1 H), 1.83-1.88 (m, 1 H), 2.05 (dddq, $J = 18.8, 6.4, 3.2$ and 2.4 Hz, 1 H), 2.10-2.17 (m, 1 H), 2.29-2.37 (m, 1 H), 3.78 (ddd, $J = 9.2, 4.4,$ and 2.8 Hz, 1 H), 5.51 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 22.0, 23.5, 24.1, 24.6, 25.5, 28.0, 28.8, 29.9, 34.4, 38.9, 43.9, 71.4, 74.9, 123.9, 133.8; **IR** (Neat): 2954, 2933, 1463, 1378, 1227, 1120, 1077, 1015, 892, 800 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{15}\text{H}_{26}\text{O}$ (M^+) requires 222.1984; found 222.1986.

4-Hexyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27q):



Colourless oil (38 mg, 15%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.87 (t, $J = 6.8$ Hz, 3 H), 1.20 (s, 3 H), 1.22-1.32 (m, 10 H), 1.28 (s, 3 H), 1.47 (ddd, $J = 6.4, 3.2$ and 3.2 Hz, 1 H), 1.62 (ddd, $J = 12.4, 3.2$ and 2.8 Hz, 1 H), 1.73 (s, 3 H), 1.85-1.92 (m, 1 H), 2.05 (dddq, $J = 19.2, 6.0, 3.2$ and 2.8 Hz, 1 H), 2.10-2.14 (m, 1 H), 2.28-2.40 (m, 1 H), 3.65 (t, $J = 5.6$ Hz, 1 H), 5.50 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.3, 22.8, 24.2, 25.4, 26.6, 28.1, 28.7, 28.9, 29.5, 32.1, 34.5, 35.1, 38.5, 73.9, 74.9, 123.9, 133.7; **IR** (Neat): 2929, 2868, 1647, 1458, 1377, 1077, 1017, 886, 724 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{30}\text{O}$ (M^+) requires 250.2297; found 250.2298.

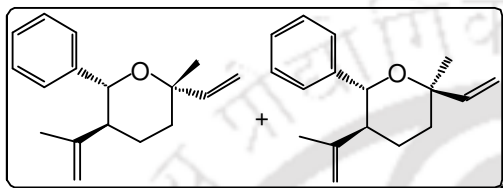
4-Pentadecyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27r):



Semisolid (38 mg, 10%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.88 (t, $J = 6.8$ Hz, 3 H), 1.20 (s, 3 H), 1.21-1.28 (m, 20 H), 1.28 (s, 3 H), 1.30-1.49 (m, 5

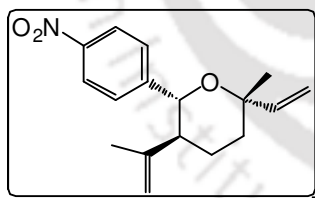
H), 1.62 (ddd, $J = 12.4, 3.2$ and 3.2 Hz, 1 H), 1.73 (s, 3 H), 1.90-1.93 (m, 2 H), 2.00-2.15 (m, 4 H), 2.26-2.37 (m, 2 H), 3.65 (t, $J = 5.6$ Hz, 1 H), 5.51 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.3, 22.9, 24.2, 25.4, 26.6, 28.0, 28.7, 28.9, 29.85 (5C), 29.9 (5C), 32.1, 34.5, 35.0, 38.4, 73.9, 74.9, 123.8, 133.7; **IR** (KBr): 2924, 2854, 1644, 1459, 1378, 1226, 1076, 1067, 801, 721 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{26}\text{H}_{48}\text{O}$ (M^+) requires 376.3705; found 376.3696.

5-Isopropenyl-2-methyl-6-phenyl-2-vinyltetrahydropyran (28a/29a) (2:1):



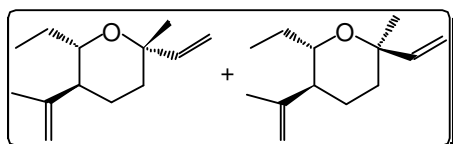
Colourless oil (61 mg, 25%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.25 (s, 1 H), 1.29 (s, 2 H), 1.39 (s, 2 H), 1.42 (s, 1 H), 1.62-1.74 (m, 2 H), 1.79-1.90 (m, 1 H), 1.96-2.02 (m, 1 H), 2.20-2.28 (m, 1 H), 4.44 (d, $J = 10.4$ Hz, 1 H), 4.52 (brs, 0.33 H), 4.55 (brs, 0.67 H), 4.59 (brs, 0.67 H), 4.65 (brs, 0.33 H), 4.99 (d, $J = 10.8$ Hz, 0.33 H), 5.07 (d, $J = 14.4$ Hz, 0.33 H), 5.22 (dd, $J = 18.0$ and 1.2 Hz, 0.67 H), 5.29 (dd, $J = 11.2$ and 1.2 Hz, 0.67 H), 5.89 (dd, $J = 18.0$ and 12.0 Hz, 0.67 H), 5.99 (dd, $J = 17.2$ and 10.8 Hz, 0.33 H), 7.22-7.26 (m, 2 H), 7.28-7.31 (m, 3 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 21.6, 27.1, 31.4, 34.6, 50.8, 75.8, 78.5, 112.1, 115.3, 127.7, 128.1, 128.2, 141.8, 143.1, 146.7 (major, **28a**); **IR** (Neat): 2967, 2930, 1643, 1450, 1370, 1116, 1050, 889, 759, 698 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{22}\text{O}$ ($\text{M}+\text{H}^+$) requires 243.1749; found 243.1760.

5-Isopropenyl-2-methyl-6-(4-nitrophenyl)-2-vinyl-tetrahydropyran (28e):



Colourless oil (43 mg, 15%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.31 (s, 3 H), 1.42 (s, 3 H), 1.54-1.78 (m, 2 H), 1.80-1.92 (m, 1 H), 2.00-2.05 (m, 1 H), 2.11-2.19 (m, 1 H), 4.53 (brs, 1 H), 4.56 (d, $J = 10.0$ Hz, 1 H), 4.63 (brs, 1 H), 5.22 (d, $J = 17.6$ Hz, 1 H), 5.32 (d, $J = 11.2$ Hz, 1 H), 5.86 (dd, $J = 18.0$ and 11.2 Hz, 1 H), 7.47 (d, $J = 8.4$ Hz, 2 H), 8.15 (d, $J = 8.0$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 21.6, 26.8, 31.2, 34.4, 51.4, 76.1, 77.4, 113.1, 115.8, 123.4, 128.4, 142.4, 145.5, 147.6, 149.3; **IR** (Neat): 2928, 2857, 1606, 1520, 1346, 1262, 1108, 1055, 894, 751, 696 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{NO}_3$ ($\text{M}+\text{H}^+$) requires 288.1599; found 288.1605.

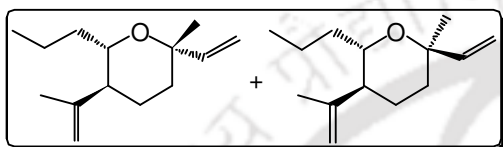
6-Ethyl-5-isopropenyl-2-methyl-2-vinyltetrahydropyran (28n/29n) (3:1):



Colourless oil (126 mg, 65%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.91 (t, $J = 7.2$ Hz, 0.75 H), 0.94 (t, $J = 7.2$ Hz, 2.25 H), 1.23 (s, 3 H), 1.24-1.30 (m, 2 H), 1.47-1.57 (m, 2 H), 1.62 (s, 3 H), 1.61-1.65 (m, 1 H), 1.66-1.71 (m, 1 H), 1.85-1.94 (m, 1

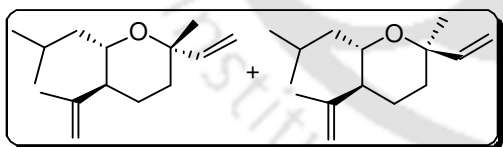
H), 3.36-3.42 (m, 0.75 H), 3.45-3.51 (m, 0.25 H), 4.70 (brs, 1.50 H), 4.75 (brs, 0.50 H), 4.99 (dd, $J = 11.2$ and 1.6 Hz, 0.25 H), 5.13 (dd, $J = 18.0$ and 1.6 Hz, 0.75 H), 5.19 (dd, $J = 10.8$ and 1.2 Hz, 0.75 H), 5.22 (dd, $J = 17.6$ and 1.6 Hz, 0.25 H), 5.76 (dd, $J = 17.6$ and 10.8 Hz, 0.75 H), 5.93 (dd, $J = 17.6$ and 11.2 Hz, 0.25 H); ^{13}C NMR (100 MHz, CDCl_3): δ 10.2, 20.3, 26.6, 27.0, 31.4, 34.6, 49.7, 74.8, 75.0, 111.7, 114.8, 143.5, 147.6 (major, **28n**); IR (Neat): 2928, 2873, 1641, 1458, 1377, 1116, 995, 888, 740 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{13}\text{H}_{22}\text{O}$ ($\text{M}+\text{H}$) $^+$ requires 195.1749; found 195.1747.

6-Propyl-5-isopropenyl-2-methyl-2-vinyltetrahydropyran (**28o/29o**) (3:1):

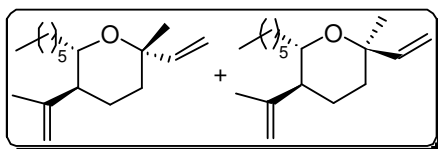


Colourless oil (135 mg, 65%); ^1H NMR (400 MHz, CDCl_3): δ 0.87 (t, $J = 7.6$ Hz, 0.75 H), 0.88 (t, $J = 7.2$ Hz, 2.25 H), 1.20 (s, 0.75 H), 1.21 (s, 2.25 H), 1.23-1.35 (m, 4 H), 1.36-1.43 (m, 1 H), 1.45-1.60 (m, 2 H), 1.62 (s, 2.25 H), 1.64 (s, 0.75 H), 1.67-1.71 (m, 1 H), 1.85-1.92 (m, 1 H), 3.46 (dt, $J = 9.2$ and 2.0 Hz, 0.75 H), 3.56 (dt, $J = 10.0$ and 2.4 Hz, 0.25 H), 4.70 (brs, 1.50 H), 4.75 (brs, 0.50 H), 4.98 (dd, $J = 10.8$ and 1.6 Hz, 0.25 H), 5.12 (dd, $J = 17.6$ and 1.6 Hz, 0.75 H), 5.19 (dd, $J = 11.2$ and 1.2 Hz, 0.75 H), 5.20 (dd, $J = 16.4$ and 1.6 Hz, 0.25 H), 5.76 (dd, $J = 18.0$ and 12.2 Hz, 0.75 H), 5.92 (dd, $J = 17.2$ and 10.4 Hz, 0.25 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.2, 18.8, 20.3, 27.0, 31.4, 34.5, 35.9, 50.1, 73.4, 74.8, 111.7, 114.7, 143.5, 147.6 (major, **28o**); IR (Neat): 2957, 2931, 1643, 1452, 1377, 1114, 1067, 1026, 916, 889, 746 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{14}\text{H}_{24}\text{O}$ ($\text{M}+\text{H}$) $^+$ requires 209.1905; found 209.1914.

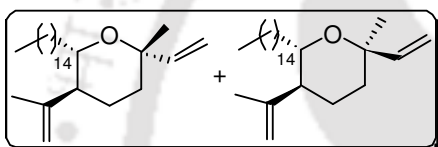
6-Isobutyl-5-isopropenyl-2-methyl-2-vinyltetrahydropyran (**28p/29p**) (3:1):



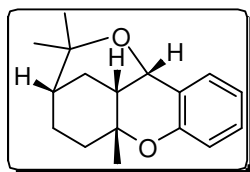
Colourless oil (149 mg, 67%); ^1H NMR (400 MHz, CDCl_3): δ 0.83 (d, $J = 6.8$ Hz, 3 H), 0.89 (d, $J = 6.8$ Hz, 3 H), 1.11-1.18 (m, 2 H), 1.21 (s, 3 H), 1.23-1.31 (m, 2 H), 1.48-1.58 (m, 1 H), 1.61 (s, 3 H), 1.65-1.71 (m, 1 H), 1.83-1.92 (m, 2 H), 3.50 (dt, $J = 10.4$ and 2.0 Hz, 0.75 H), 3.62 (dt, $J = 10.0$ and 2.0 Hz, 0.25 H), 4.69 (brs, 1.50 H), 4.73 (brs, 0.50 H), 4.98 (dd, $J = 10.8$ and 2.0 Hz, 0.25 H), 5.12 (dd, $J = 17.6$ and 1.2 Hz, 0.75 H), 5.18 (dd, $J = 8.0$ and 1.6 Hz, 0.25 H), 5.21 (dd, $J = 11.2$ and 1.2 Hz, 0.75 H), 5.79 (dd, $J = 18.0$ and 11.2 Hz, 0.75 H), 5.92 (dd, $J = 17.2$ and 10.8 Hz, 0.25 H); ^{13}C NMR (100 MHz, CDCl_3): δ 20.1, 21.5, 23.9, 24.2, 27.0, 31.4, 34.2, 43.1, 50.6, 71.7, 74.7, 111.8, 114.8, 143.6, 147.6 (major, **28p**); IR (Neat): 2950, 2932, 1643, 1450, 1380, 1264, 1118, 1068, 997, 918, 890, 735 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{15}\text{H}_{26}\text{O}$ ($\text{M}+\text{H}$) $^+$ requires 223.2062.; found 223.2063.

6-Hexyl-5-isopropenyl-2-methyl-2-vinyltetrahydropyran (28q/29q) (3:1):

Colourless oil (150 mg, 60%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.86 (t, $J = 6.8$ Hz, 3 H), 1.22 (s, 3 H), 1.23-1.34 (m, 10 H), 1.40-1.57 (m, 2 H), 1.62 (s, 3 H), 1.64-1.71 (m, 1 H), 1.85-1.91 (m, 2 H), 3.45 (t, $J = 9.6$ Hz, 0.75 H), 3.54 (t, $J = 8.0$ Hz, 0.25 H), 4.70 (brs, 1.50 H), 4.73 (brs, 0.50 H), 4.98 (dd, $J = 10.8$ and 1.2 Hz, 0.25 H), 5.12 (dd, $J = 17.6$ and 1.2 Hz, 0.75 H), 5.19 (dd, $J = 11.2$ and 1.6 Hz, 0.75 H), 5.20 (dd, $J = 14.0$ and 1.2 Hz, 0.25 H), 5.76 (dd, $J = 18.0$ and 11.2 Hz, 0.75 H), 5.93 (dd, $J = 17.6$ and 10.8 Hz, 0.25 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.3, 20.3, 22.9, 25.8, 27.1, 29.6, 31.4, 32.2, 33.9, 34.6, 50.1, 73.7, 74.8, 111.7, 114.7, 143.6, 147.7 (major, **28q**); **IR** (Neat): 2929, 2858, 1643, 1454, 1378, 1117, 1074, 916, 889 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{30}\text{O}$ ($\text{M}+\text{H}$)⁺ requires 251.2375; found 251.2380.

5-Isopropenyl-2-methyl-6-pentadecyl-2-vinyltetrahydropyran (28r/29r) (3:1):

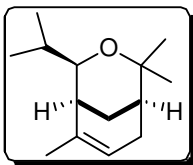
Colourless oil (233 mg, 62%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.88 (t, $J = 7.2$ Hz, 3 H), 1.21 (s, 3 H), 1.22-1.30 (m, 24 H), 1.40-1.57 (m, 5 H), 1.62 (s, 2.25 H), 1.64 (s, 0.75 H), 1.66-1.71 (m, 2 H), 1.84-1.92 (m, 2 H), 3.42-3.47 (m, 0.75 H), 3.50-3.54 (m, 0.25 H), 4.70 (brs, 1.50 H), 4.74 (brs, 0.50 H), 4.98 (dd, $J = 10.4$ and 1.6 Hz, 0.25 H), 5.12 (dd, $J = 18.0$ and 1.2 Hz, 0.75 H), 5.19 (dd, $J = 11.2$ and 1.2 Hz, 0.75 H), 5.21 (dd, $J = 16.0$ and 1.6 Hz, 0.25 H), 5.76 (dd, $J = 18.0$ and 11.2 Hz, 0.75 H), 5.93 (dd, $J = 17.6$ and 10.8 Hz, 0.25 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.3, 20.3, 22.9, 25.8, 27.1, 29.6, 29.9 (9C), 31.4, 32.2, 33.9, 34.6, 50.2, 73.7, 74.8, 111.7, 114.7, 143.6, 147.7 (major, **28r**); **IR** (Neat): 2923, 2855, 1643, 1455, 1377, 1118, 1077, 889, 721, 683 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{26}\text{H}_{48}\text{O}$ (M^+) requires 376.3705; found 376.3705.

Tetracyclic compound (39):

Colourless oil (103 mg, 40%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.22 (s, 3 H), 1.39 (dddd, $J = 4.4, 3.6, 3.6$ and 3.6 Hz, 1 H), 1.42 (s, 3 H), 1.52 (d, $J = 1.2$ Hz, 3 H), 1.53-1.59 (m, 2 H), 1.80-1.84 (m, 1 H), 1.87 (ddd, $J = 13.6, 3.2$ and 2.8 Hz, 1 H), 2.06 (dddd, $J = 13.0, 5.2, 3.6, 3.2$ and 2.8 Hz, 1 H), 2.33 (dddd, $J = 13.6, 3.2, 3.2$ and 2.8 Hz, 1 H), 2.63 (brddd, $J = 14.8, 7.6, 5.6$ and 3.6 Hz, 1 H), 4.62 (d, $J = 2.4$ Hz, 1 H), 6.80 (d, $J = 8.0$ Hz, 1 H), 6.85 (td, $J = 7.2$ and 1.2 Hz, 1 H), 7.15-7.21 (m, $J = 7.6$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 24.4, 26.2, 26.7, 26.9, 28.5, 32.4, 34.8, 36.7, 66.7, 74.9, 77.7, 117.0, 119.9, 122.2, 129.8,

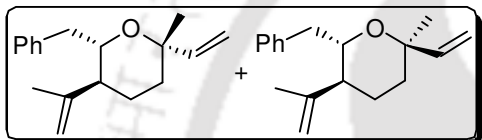
130.6, 153.4; **IR** (Neat): 2961, 2934, 1611, 1484, 1366, 1247, 1107, 1094, 1049, 933, 753 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{22}\text{O}_2$ (M^+) requires 258.1620; found 258.1621.

4-Isopropyl-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (41b):



Colourless oil (94 mg, 45%); **^1H NMR** (400 MHz, CDCl_3): δ 0.92 (d, $J = 6.4$ Hz, 3 H), 0.94 (d, $J = 6.4$ Hz, 3 H), 1.21 (s, 3 H), 1.26 (s, 3 H), 1.46 (ddd, $J = 6.4, 3.2$ and 3.2 Hz, 1 H), 1.49-1.60 (m, 1 H), 1.65 (ddd, $J = 12.4, 3.2$ and 3.2 Hz, 1 H), 1.75 (s, 3 H), 2.00-2.12 (m, 2 H), 2.14-2.20 (m, 1 H), 2.28-2.38 (m, 1 H), 3.14 (d, $J = 10.0$ Hz, 1 H), 5.52 (brs, 1 H); **^{13}C NMR** (100 MHz, CDCl_3): δ 19.5, 20.7, 24.2, 25.4, 28.2, 28.7, 29.1, 30.8, 34.3, 35.6, 75.1, 80.2, 124.4, 133.7; **IR** (Neat) 2956, 2931, 1642, 1458, 1377, 1226, 1067, 1050, 886, 801 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{14}\text{H}_{24}\text{O}$ (M^+) requires 208.1827; found 208.1832.

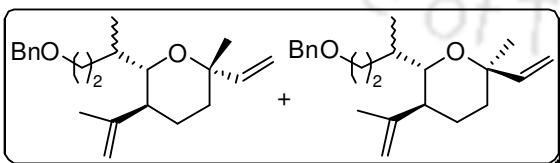
6-Benzyl-5-isopropyl-2-methyl-2-vinyltetrahydropyran (42c/43c) (3:2):



Colourless oil (133 mg, 52%); **^1H NMR** (400 MHz, CDCl_3): δ 1.01 (s, 1.8 H), 1.18 (s, 1.2 H), 1.27 (s, 1.8 H), 1.32 (s, 1.2 H), 1.37-1.47 (m, 2 H), 1.59-1.78 (m, 2 H), 1.90-2.00 (m, 1 H), 2.76-2.84 (m, 1 H), 3.02 (dd, $J = 16$ and 5.6 Hz, 1 H), 3.87-3.99 (m, 1 H), 4.98-5.26 (m, 4 H), 5.83 (dd, $J = 18.0$ and 10.8 Hz, 0.6 H), 5.97 (dd, $J = 17.6$ and 10.8 Hz, 0.4 H), 7.03-7.11 (m, 3 H), 7.13-7.19 (m, 1 H), 7.30-7.34 (m, 1 H); **^{13}C NMR** (100 MHz, CDCl_3): δ 20.5, 20.6, 27.7, 27.8, 28.5, 31.3, 35.4, 37.9, 38.2, 38.4, 47.2, 47.4, 66.9, 67.9, 73.0, 74.5, 111.4, 114.9, 125.8, 125.9, 126.3, 126.7, 128.4, 128.9, 129.2, 129.3, 133.6, 133.7, 143.0, 145.8, 145.9, 146.4; **IR** (Neat): 2967, 2934, 1643, 1448, 1382, 1117, 1076, 1037, 921, 759, 729 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{24}\text{O}$ ($\text{M}+\text{H}^+$) requires 257.1905; found 257.1914.

6-(3-Benzyloxy-1-methylpropyl)-5-isopropenyl-2-methyl-2-vinyltetrahydropyran

(42d/43d) (2:1):

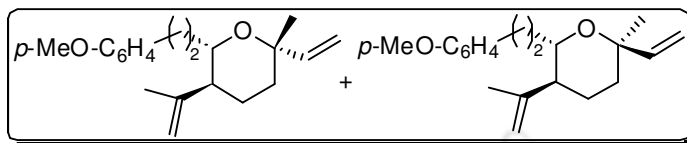


Colourless oil (157 mg, 48%); **^1H NMR** (400 MHz, CDCl_3): δ 0.85 (d, $J = 6.4$ Hz, 2 H), 0.97 (d, $J = 6.4$ Hz, 1 H), 1.15 (s, 2 H), 1.20 (s, 1 H), 1.23 (s, 1 H), 1.26 (s, 2 H), 1.33-1.53 (m, 2 H), 1.62-1.76 (m, 2 H), 1.80-1.88 (m, 1 H), 2.00-2.16 (m, 1 H), 3.38-3.57 (m, 3 H), 4.48 (s, 0.66 H), 4.56 (s, 1.34 H), 4.70-4.75 (m, 2 H), 4.91-5.20 (m, 4 H), 5.67-5.76 (m, 0.67 H), 5.87 (dd, $J = 17.6$ and 10.8 Hz, 0.33 H), 7.32-7.37 (m, 5 H); **^{13}C NMR** (100 MHz, CDCl_3): δ 12.6, 19.8, 27.1, 29.9, 31.1, 31.2, 34.4, 46.8, 69.1, 72.3, 73.1, 75.2, 111.8, 114.7, 127.8, 128.0, 128.5, 128.6, 138.5, 143.8 (major, **42d**); **IR** (Neat): 2928, 2862,

1639, 1448, 1264, 1080, 1050, 1027, 795, 738 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{22}\text{H}_{32}\text{O}_2$ ($\text{M}+\text{H}$)⁺ requires 329.2480; found 329.2491.

5-Isopropenyl-6-[2-(4-methoxyphenyl)-ethyl]-2-methyl-2-vinyltetrahydropyran

(42e/43e) (2:1):



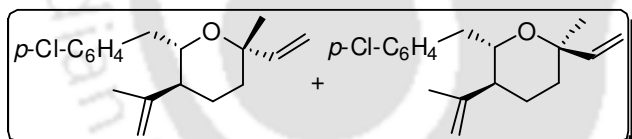
Colourless oil (135 mg, 45%);

¹H NMR (400 MHz, CDCl_3): δ

1.24 (s, 2 H), 1.25 (s, 1 H), 1.55

(brs, 3 H), 1.48-1.80 (m, 5 H), 1.86-1.96 (m, 2 H), 2.47-2.60 (m, 1 H), 2.73-2.87 (m, 1 H), 3.41-3.52 (m, 1 H), 3.78 (brs, 3 H), 4.56-4.71 (m, 2 H), 5.00 (dd, $J = 10.8$ and 1.6 Hz, 0.33 H), 5.11 (dd, $J = 18.0$ and 1.2 Hz, 0.67 H), 5.17 (dd, $J = 11.2$ and 1.2 Hz, 0.67 H), 5.24 (dd, $J = 17.6$ and 1.6 Hz, 0.33 H), 5.76 (dd, $J = 18.0$ and 11.2 Hz, 0.67 H), 5.96 (dd, $J = 17.6$ and 10.8 Hz, 0.33 H), 6.80 (d, $J = 8.8$ Hz, 2 H), 7.10 (d, $J = 8.8$ Hz, 2 H); ¹³C NMR (100 MHz, CDCl_3): δ 20.2, 20.9, 22.5, 22.9, 26.3, 26.9, 29.6, 29.9, 30.8, 31.0, 31.3, 34.5, 34.9, 35.8, 40.0, 50.1, 55.5, 71.1, 73.0, 75.0, 110.4, 111.0, 111.9, 113.8, 114.0, 114.8, 124.6, 126.7, 128.4, 129.6, 129.7, 133.6, 135.1, 143.5, 146.7, 147.5; **IR** (Neat): 2929, 2857, 1633, 1454, 1246, 1078, 1039, 888, 827 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{20}\text{H}_{28}\text{O}_2$ ($\text{M}+\text{H}$)⁺ requires 301.2167; found 301.2168.

6-(4-Chlorobenzyl)-5-isopropenyl-2-methyl-2-vinyl-tetrahydropyran (42f/43f) (3:2):



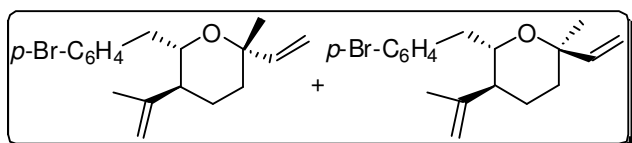
Colourless oil (139 mg, 48%); ¹H

NMR (400 MHz, CDCl_3): δ 1.07 (s,

1.80 H), 1.17 (s, 1.20 H), 1.27 (s,

1.80 H), 1.33 (s, 1.20 H), 1.35-1.43 (m 1 H), 1.54-1.77 (m, 3 H), 1.96-2.04 (m, 1 H), 2.61-2.80 (m, 1 H), 2.98 (dd, $J = 16.0$ and 5.6 Hz, 1 H), 3.83-3.96 (m, 1 H), 4.91-5.10 (m, 1 H), 5.14-5.26 (m, 3 H), 5.83 (dd, $J = 17.6$ and 11.2 Hz, 0.60 H), 5.96 (dd, $J = 17.6$ and 10.8 Hz, 0.40 H), 6.96-7.0 (m, 1 H), 7.04-7.08 (m, 1 H), 7.23-7.32 (m, 2 H); ¹³C NMR (100 MHz, CDCl_3): δ 20.5, 20.6, 27.7, 28.4, 31.2, 35.3, 35.4, 37.3, 37.6, 38.7, 46.9, 47.1, 66.7, 67.8, 73.2, 74.6, 74.8, 111.6, 115.0, 124.5, 126.1, 126.2, 126.8, 128.3, 130.6, 130.7, 132.0, 132.2, 143.0, 146.3, 147.9; **IR** (Neat): 2968, 2929, 2872, 1633, 1467, 1369, 1227, 1116, 1077, 922, 881, 814, 802 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{23}\text{ClO}$ ($\text{M}+\text{H}$)⁺ requires 291.1515; found 291.1519.

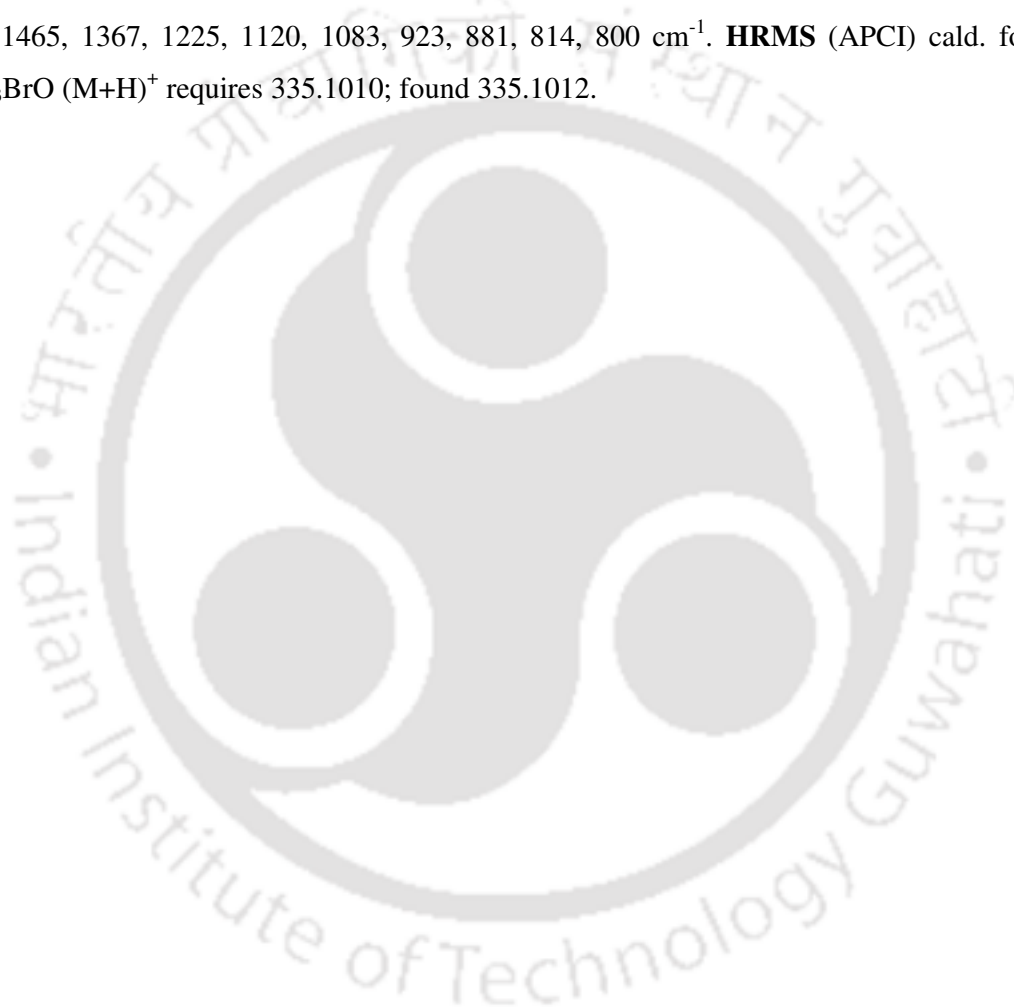
6-(4-Bromobenzyl)-5-isopropenyl-2-methyl-2-vinyltetrahydropyran (42g/43g) (3:2):



Colourless oil (160 mg, 48%); ¹H

NMR (400 MHz, CDCl_3): δ 1.07 (s,

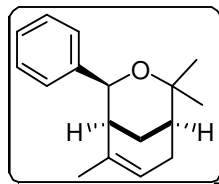
1.80 H), 1.17 (s, 1.20 H), 1.27 (s, 1.80 H), 1.33 (s, 1.20 H), 1.35-1.42 (m, 1 H), 1.56-1.76 (m, 3 H), 1.94-2.0 (m, 1 H), 2.60-2.78 (m, 1 H), 2.98 (dd, $J = 16.0$ and 5.6 Hz, 1 H), 3.80-3.98 (m, 1 H), 5.04 (dd, $J = 10.8$ and 1.2 Hz, 1H), 5.14-5.28 (m, 3 H), 5.83 (dd, $J = 17.6$ and 11.2 Hz, 0.60 H), 5.96 (dd, $J = 17.6$ and 10.8 Hz, 0.40 H), 6.90-6.95 (m, 1 H), 7.18-7.23 (m, 1 H), 7.43 (dd, $J = 8.4$ and 2.0 Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 20.4, 20.5, 27.7, 28.4, 31.2, 35.3, 35.4, 37.3, 37.6, 38.7, 46.8, 47.0, 66.6, 67.7, 73.1, 74.6, 111.6, 115.0, 120.0, 129.0, 129.8, 130.9, 131.0, 132.7, 143.0, 146.2, 148.3; IR (Neat): 2969, 2932, 2872, 1632, 1465, 1367, 1225, 1120, 1083, 923, 881, 814, 800 cm^{-1} . HRMS (APCI) calcd. for $\text{C}_{18}\text{H}_{23}\text{BrO}$ (M+H) $^+$ requires 335.1010; found 335.1012.



3.7 Selected Spectra of Oxabicyclo[3.3.1]nonenes and Substituted Tetrahydropyrans

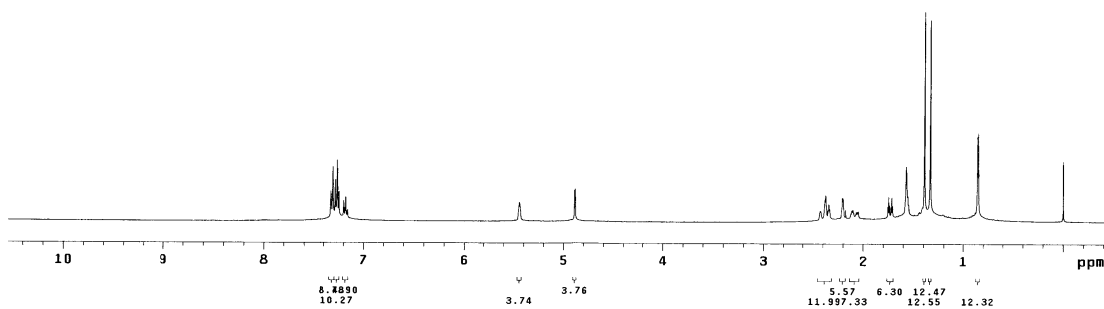
2,2,6-Trimethyl-4-phenyl-3-oxabicyclo[3.3.1]non-6-ene (27a):

^1H NMR (400 MHz, CDCl_3)



```

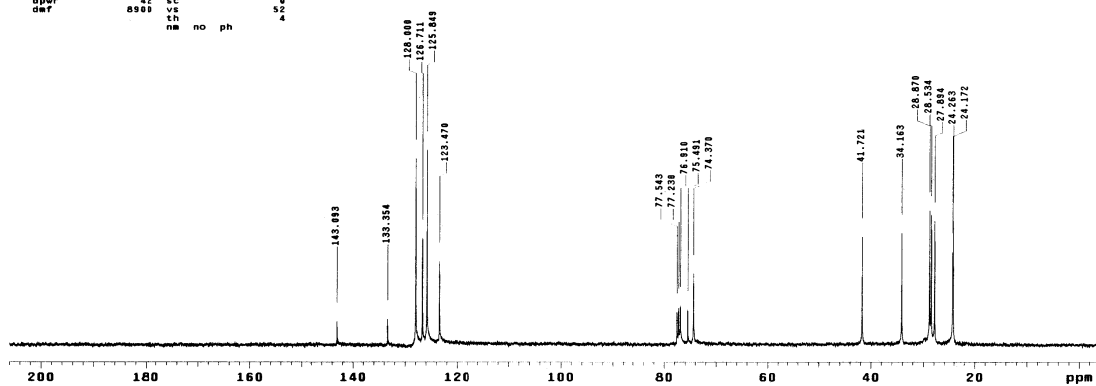
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file exp sp1n not used
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sw 6389.8 pw90 19.700
at 1.998 a1fa 20.000
np 25528 FLAGS
fb not used i1 n
bs 8 in n
d1 1.000 dp y
nt 200 hs nn
ct 176 PROCESSING
tn TRANSMITTER lb 0.10
fn fn 65536
sfrq 399.853 DISPLAY
tof 362.8 sp -183.3
tpr 57 wp 4399.7
pw 9.850 rfl 796.2
DECOUPLER rfp 0
dn C13 rp 113.0
dof 0 lp -96.4
dm nnn PLOT
dmm c wc 250
dpwr 50 sc 0
dnt 15900 vs 49
nm cdc ph 14
  
```



^{13}C NMR (100 MHz, CDCl_3)

```

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date Sep 18 2010 temp SPECIAL
solvent CDCl3 gain not used
file exp sp1n not used
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sw 25125.8 pw50 10.000
at 8927.0 a1fa 20.000
np 13800 FLAGS
fb 16 in n
bs 16 in n
d1 1.000 dp y
nt 500 hs nn
ct 1520 PROCESSING
tn TRANSMITTER lb 2.00
fn C13 fn 65536
sfrq 100.624 DISPLAY
tof 1536.3 sp -353.9
tpr 61 wp 21983.9
pw 9.308 rfl 3274.4
DECOUPLER rfp 7764.9
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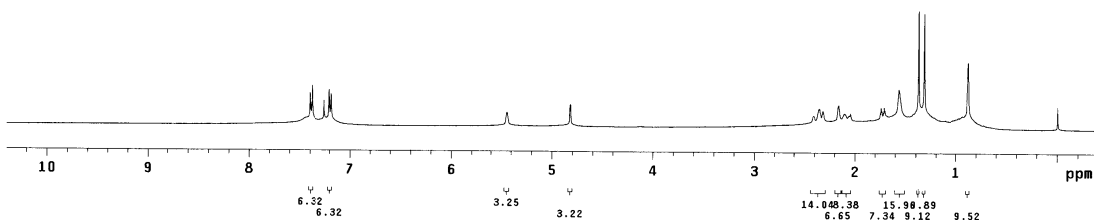
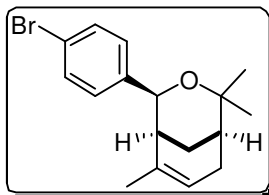
4-(4-Bromophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]-non-6-ene (27c):

 ^1H NMR (400 MHz, CDCl_3)

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PS_167_M
exp1 s2pu1
SAMPLE
date Mar 7 2011 temp SPECIAL
solvent CDC13 gain not used
file exp spin not used
ACQUISITION hst 0.008
sw 6389.8 pw90 19.700
at 1.398 alfa 20.000
np 25528
fb not used i1 FLAGS
bs 8 in n
dl 1.800 dp y
nt 200 hs nn
ct 120
TRANSMITTER i1 b 0.10
tn H1 fn 65536
sfrq 399.853 DISPLAY
tof 362.8 sp -157.7
tpwr 1.57 wp 4315.7
pw 9.850 rfp 735.8
DECOUPLER C13 rp 0
dn C13 ip 129.7
dof 0 lp -89.9
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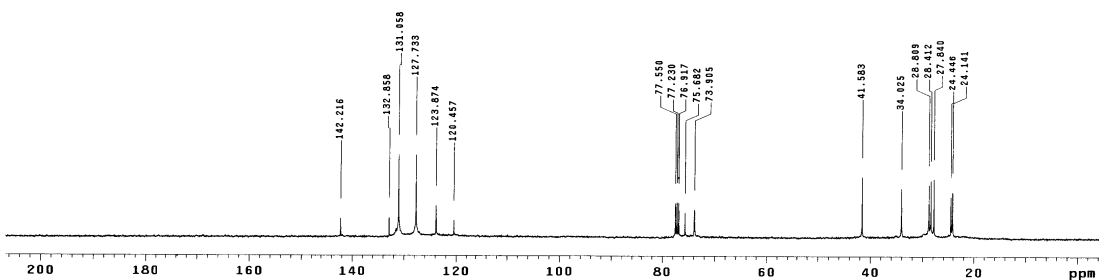
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nt 8000 hs nn
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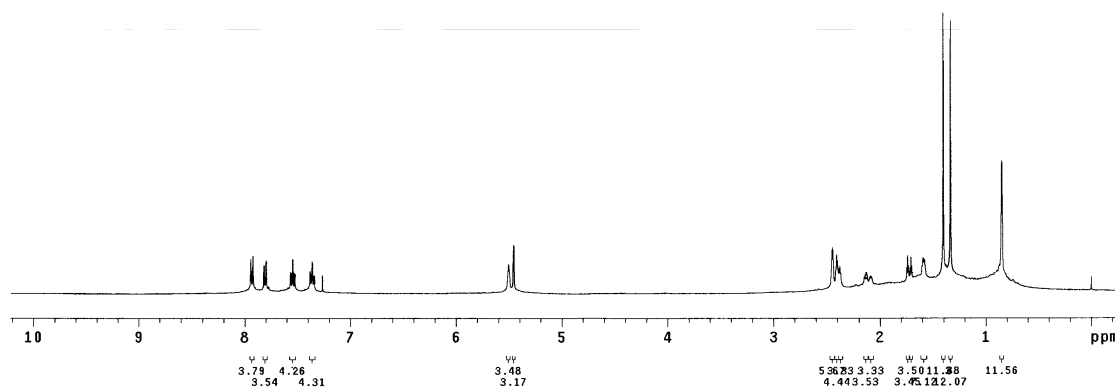
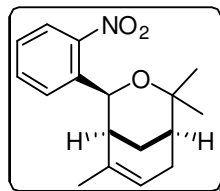
4-(2-Nitrophenyl)-2,2,6-trimethyl-3-oxabicyclo[3.3.1]non-6-ene (27f):

 ^1H NMR (400 MHz, CDCl_3)

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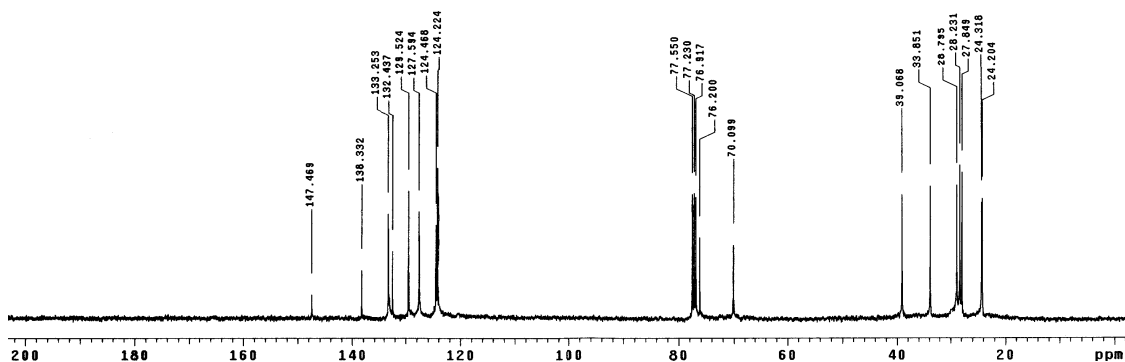
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 ^{13}C NMR (100 MHz, CDCl_3)

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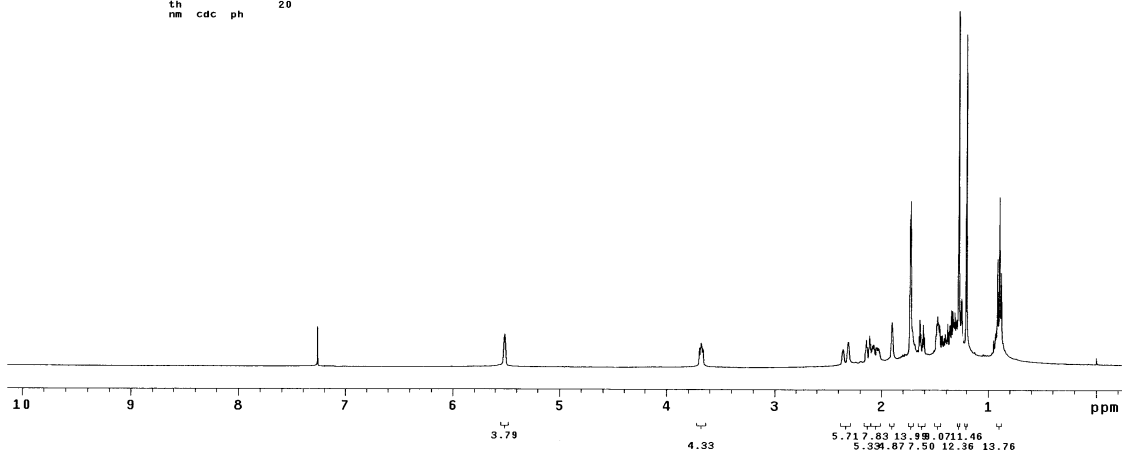
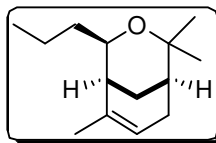
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ct 32
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tof 362.8 sp -109.2
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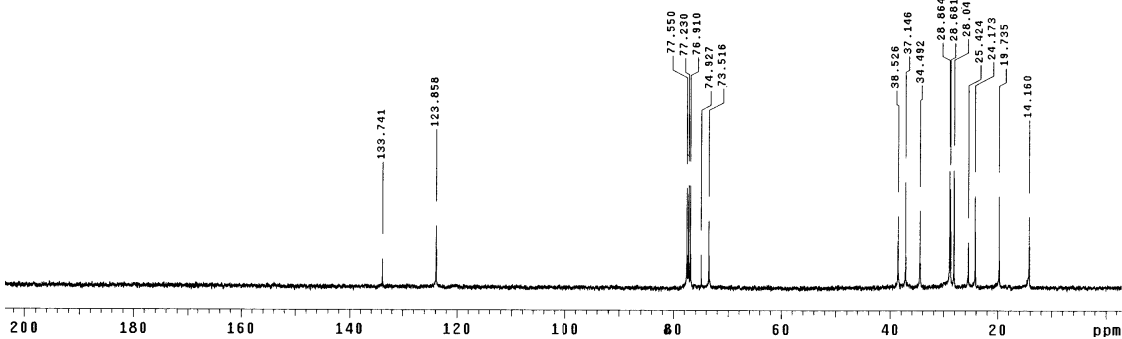
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 ^{13}C NMR (100 MHz, CDCl_3)

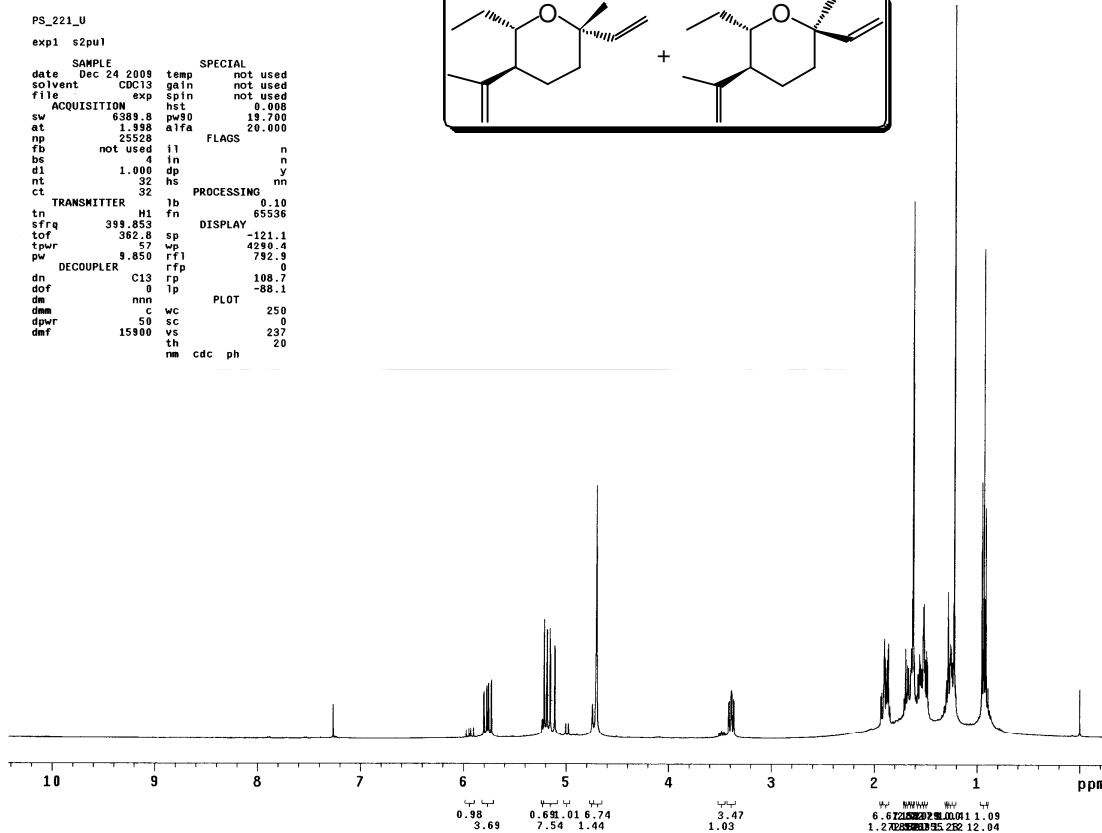
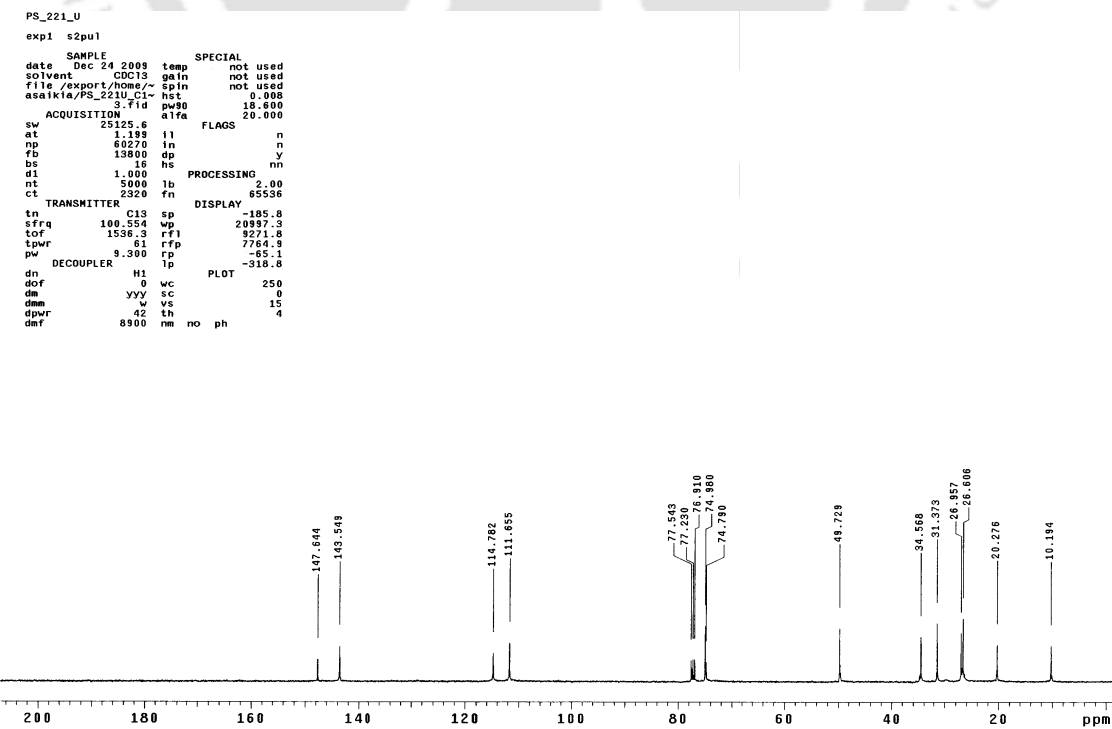
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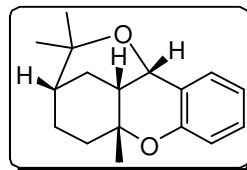
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6-Ethyl-5-isopropenyl-2-methyl-2-vinyltetrahydropyran (28n/29n) (3:1):

 ^1H NMR (400 MHz, CDCl_3) ^{13}C NMR (100 MHz, CDCl_3)

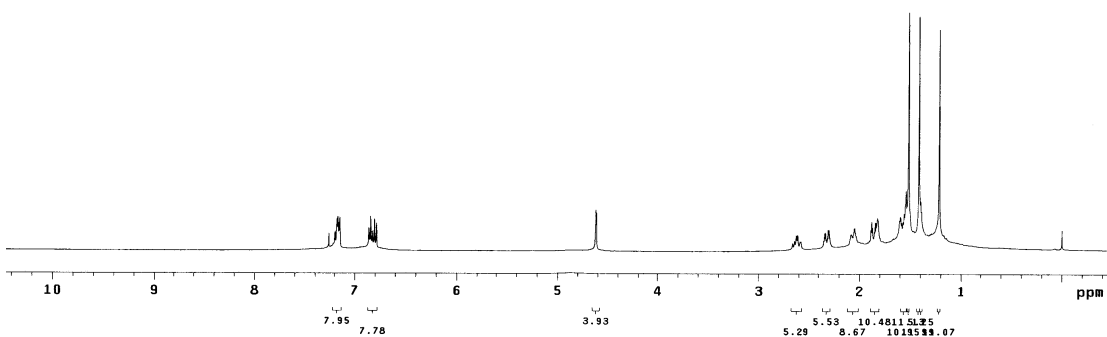
Tetracyclic compound (39):

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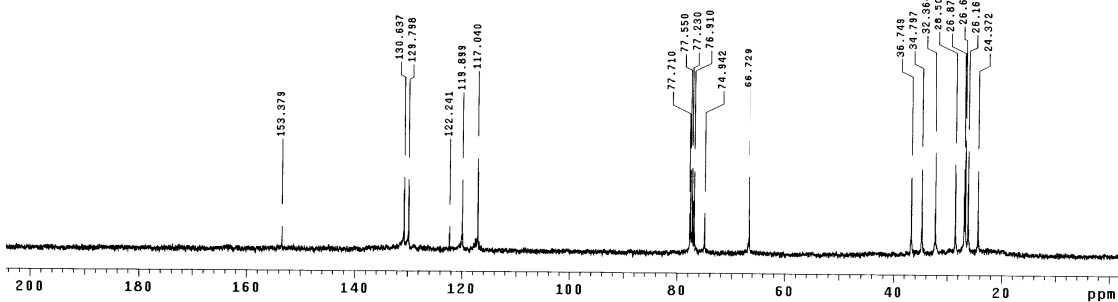
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¹³C NMR (100 MHz, CDCl₃)

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tof 1536.3 rfl 9276.4
tpr 61 rfp 7764.9
pw 9.300 rp -62.0
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dof 0 wc 250
dm vvy sc 0
dmw w vs 26
dpr 42 th 4
dmf 8900 nm no ph

```



CHAPTER 4

Stereoselective Synthesis of 6-Oxabicyclo[3.2.1]octene via (3,5)-Oxonium-Ene Reaction

4.1 Importance and Applications

Oxabicyclo[3.2.1]octanes/octenes are important intermediates in organic synthesis for numerous organic transformations.¹ Two different types of oxabicyclo[3.2.1]octane/octene are well known, one is 6-oxabicyclo[3.2.1]octane/octene (Figure 4.1.1, **A**) and another one is 8-oxabicyclo[3.2.1]octane/octene (Figure 4.1.1, **B**). The 6-oxabicyclo[3.2.1]octane



Figure 4.1.1. Different types of oxabicyclo[3.2.1]octane/octene

moiety is present in natural products like jiadifenin, jiadifenolide (Figure 4.1.2, **C-D**) which are potent promoter of neurite outgrowth.² On the other hand 8-oxabicyclo[3.2.1]octane moiety is present in aryloxatropanes (Figure 4.1.2, **E-F**) which are particularly potent inhibitors of the dopamine transporter³. Recently several 8-oxabicyclo[3.2.1]octanes have been found to possess moderate anti-HIV activity.⁴

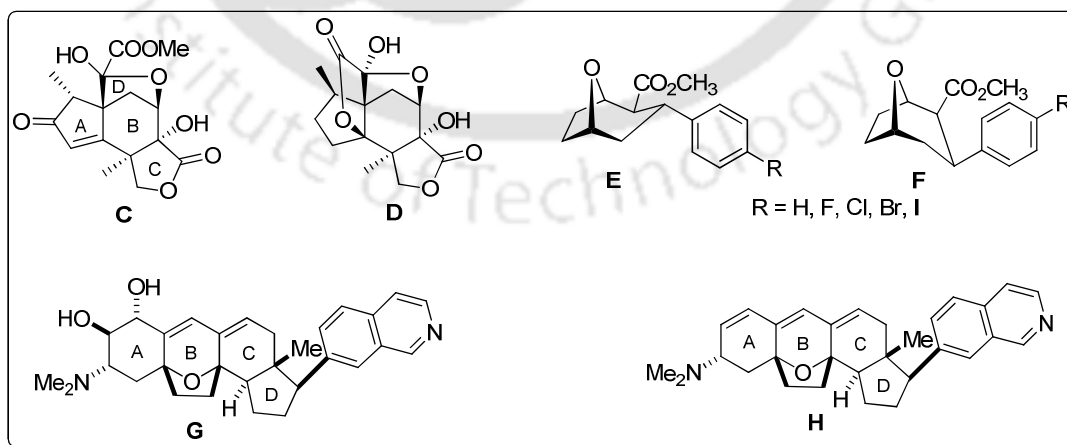


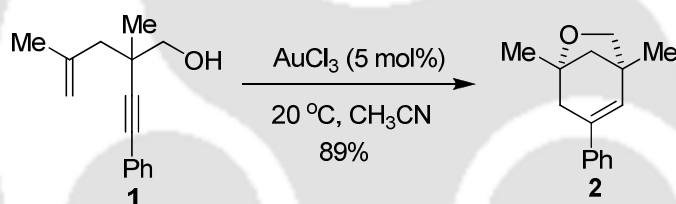
Figure 4.1.2. Some biologically important molecules containing oxabicyclo[3.2.1]octane/octene as core unit

8-Oxabicyclo[3.2.1]octene system is present in steroidal alkaloids cortistatin A and J (Figure 4.1.2, **G-H**) which are the most potent congener, exhibited strong inhibition against the proliferation of human umbilical vein endothelial cells (HUVECs: $IC_{50} = 1.8$ nM) with extreme selectivity among cell lines.⁵

4.2 An Overview of Relevant Synthetic Methods

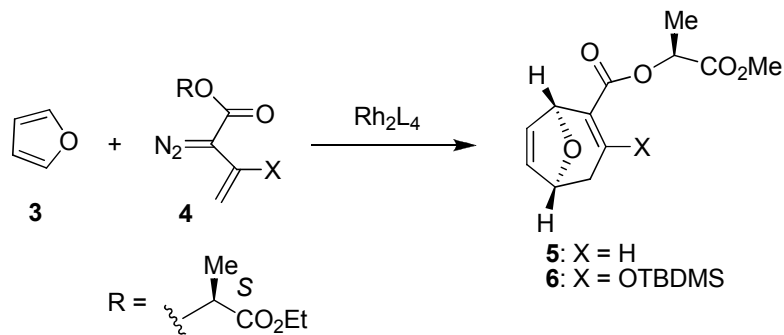
Review of literature reveals that several strategies are developed for the synthesis of 8-oxabicyclo[3.2.1]octane/octene. However only one methodology for the synthesis of 6-oxabicyclo[3.2.1]octene has been reported in the literature.

Kozmin and co-workers have reported a gold-catalyzed cycloisomerization of 1,5-enynes **1** for the synthesis of 6-oxabicyclo[3.2.1]octene.⁶ Treatment of enyne **1** with either a Au(I) or Au(III) catalytic promoter afforded 6-oxabicyclo[3.2.1]octene **2** (Scheme 4.2.1). The optimized protocol entailed a brief exposure of alcohol **1** to a catalytic amount of $AuCl_3$ in CH_3CN , which furnished alkene **2** in 89% yield.



Scheme 4.2.1.

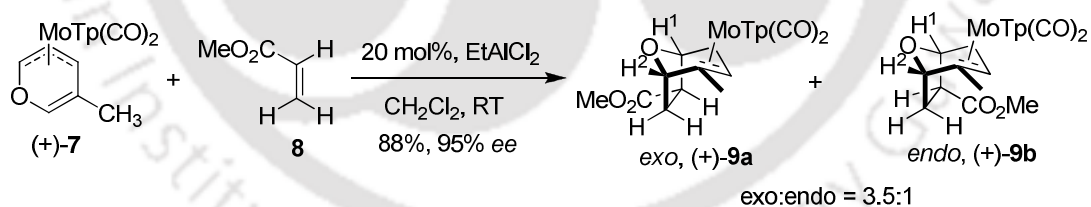
Davies *et al.* have reported a practical asymmetric synthesis of 8-oxabicyclo[3.2.1]octadienes by using the chemistry of rhodium-stabilized vinylcarbenoids. Treatment of vinylcarbenoids **4** having α -hydroxy ester auxiliaries with furan **3** in the presence of rhodium catalysts resulted in the formation of 8-oxabicyclo[3.2.1]octadienes **5** and **6** (Scheme 4.2.2).^{1c}



Rh_2L_4	5, de, %	6, de, %
$Rh_2(OOct)_4$	57	79
$Rh_2(S-TBSP)_4$	68	80
$Rh_2(R-TBSP)_4$	0	53

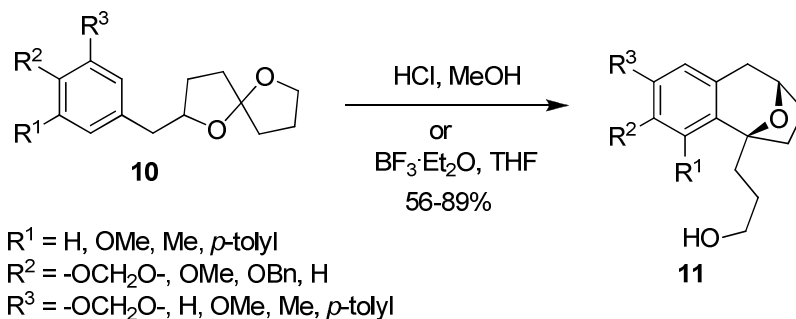
Scheme 4.2.2.

Liebeskind and co-workers have reported an enantiocontrolled [5+2] cycloaddition reaction for the synthesis of 8-oxabicyclo[3.2.1]octene with high enantiopurity. The reaction of η^3 -3-methylpyranylmolybdenum complex (+)-7 with electron deficient alkene 8 in the presence of catalytic amount of $EtAlCl_2$ in CH_2Cl_2 produce 8-oxabicyclo[3.2.1]octenes 9 in good yields with high enantioselectivity (Scheme 4.2.3).⁷



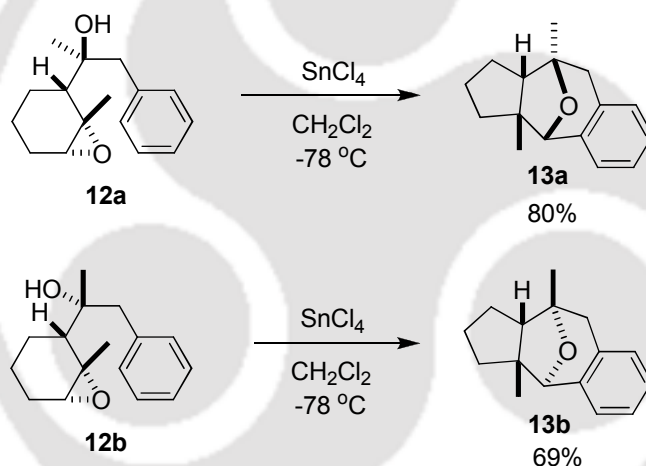
Scheme 4.2.3.

Fan *et al.* have reported an acid catalyzed intramolecular Friedel-Crafts alkylation (cyclization) using a spiroketal as electrophile leading to benzene-fused 8-oxabicyclo[3.2.1]octane ring system. Treatment of spiroketal 10 with either HCl in MeOH or $BF_3 \cdot Et_2O$ in THF produces benzene-fused 8-oxabicyclo[3.2.1]octane 11 in good yields (Scheme 4.2.4).⁸



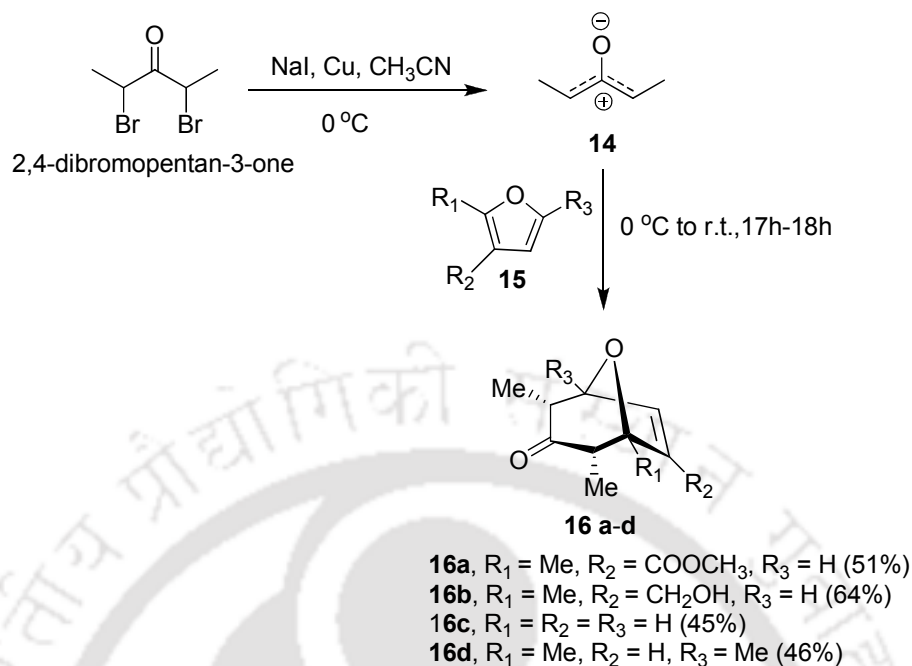
Scheme 4.2.4.

Marson *et al.* have reported a [3+4] cycloaddition reaction for the synthesis of fused 8-oxabicyclo[3.2.1]octane systems **13a** and **13b** starting from 3,4-epoxy alcohols **12a** and **12b** (Scheme 4.2.5).⁹ This reaction sequence has led to the development of lactol cyclizations and hence a general route to bridged bicyclo ethers.



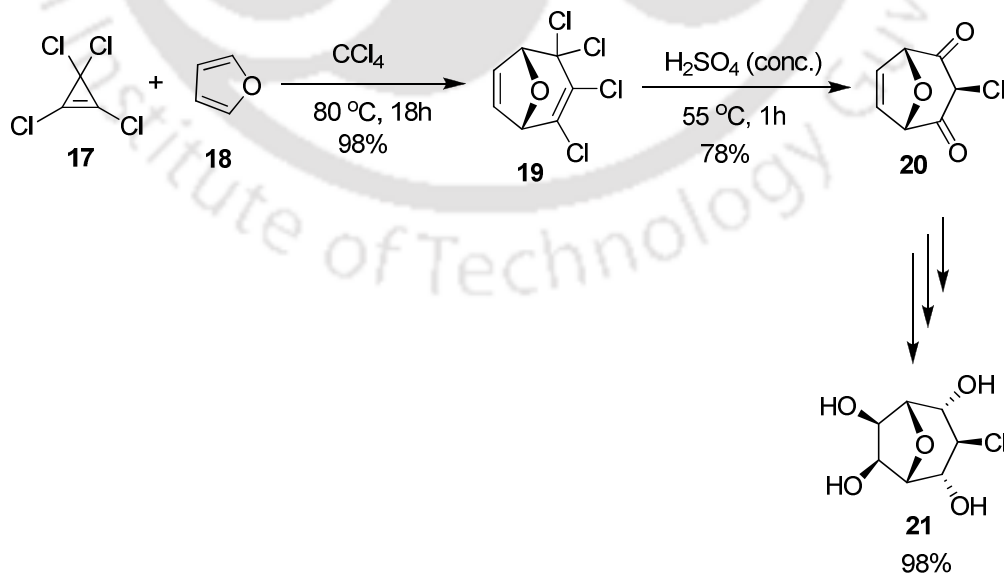
Scheme 4.2.5.

Barbosa *et al.* have reported a [4+3] cycloaddition reaction between oxyallyl carbocation **14** generated from 2,4-dibromopentan-3-one, and dienes **15** for the synthesis of 8-oxabicyclo[3.2.1]octene derivatives **16a-d** (Scheme 4.2.6).¹⁰ The important thing is that most of the oxabicyclic compounds are biologically active.¹¹



Scheme 4.2.6.

Recently Khlevin *et al.* have reported a methodology for the synthesis of polyhydroxylated 8-oxabicyclo[3.2.1]octane **21** with good yield and high stereoselectivity. They have prepared this bicyclic compound from a 1,3-dione **20** and several steps are involved in this conversion. However the initial 1,3-dione **20** have been prepared by using a [4+3] cyclocondensation between tetrachlorocyclopropene **17** and furan **18** to produce **19**. Direct



Scheme 4.2.7.

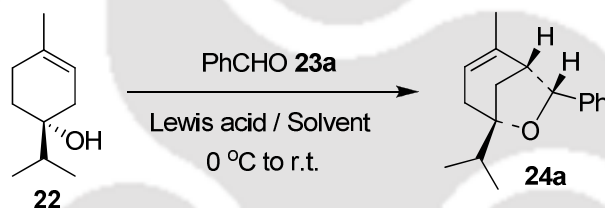
hydrolysis of **19** under strongly acidic conditions and elevated temperatures delivers the 1,3-dione **20** (Scheme 4.2.7).⁴

4.3 Present Work

In the previous chapter we have described an efficient methodology for the synthesis of oxabicyclo[3.3.1]nonenes and substituted tetrahydropyrans. In this chapter we report a stereoselective method for the synthesis of 6-oxabicyclo[3.2.1]octene via (3,5)-oxonium-ene reaction from the reaction of (-)-terpinen-4-ol with aldehyde or epoxide catalyzed by indium triflate, In(OTf)₃.

Initially (-)-terpinen-4-ol (1 equiv.), **22** was reacted with benzaldehyde (1 equiv), **23a** in dry toluene under boron trifluoride etherate (1 equiv.) at 0 °C and the desired cyclization product **24a** was obtained in 32% yield along with some decomposed products. Further

Table 4.3.1. Reaction with different Lewis acids and solvents



Entry	Lewis acid (equiv.)	Time/h	Solvent	Yield ^a (%)
1	BF ₃ ·OEt ₂ (1)	0.5	CH ₂ Cl ₂	32
2	Sc(OTf) ₃ (0.1)	24	CH ₂ Cl ₂	0
3	In(OTf) ₃ (0.1)	24	Toluene	45
4	In(OTf) ₃ (0.1)	6	CH ₂ Cl ₂	75
5	In(OTf) ₃ (0.1)	24	CH ₃ CN	0
6	In(OTf) ₃ (0.05)	24	CH ₂ Cl ₂	40
7	Bi(OTf) ₃ (0.1)	24	CH ₂ Cl ₂	52
8	Bi(OTf) ₃ (0.1)	24	Toluene	38
9	InCl ₃ (0.1)	24	CH ₂ Cl ₂	0
10	InCl ₃ (0.2)	24	CH ₂ Cl ₂	0

^aYields refer to isolated yield. The compound is characterized by IR, ¹H, ¹³C NMR and Mass spectroscopy.

lowering the temperature to $-30\text{ }^{\circ}\text{C}$ resulted in the same yield. The reaction was also performed in the presence of other Lewis acids and the results are summarised in Table 4.3.1. These results showed that indium triflate ($\text{In}(\text{OTf})_3$) was the efficient Lewis acid, which acted as a catalyst for this transformation. No reaction was observed when scandium triflate or indium trichloride was used. The reaction was also studied in different solvents. Among the solvents studied, dichloromethane was found to be the best solvent, whereas acetonitrile was found to be ineffective. Therefore, $\text{In}(\text{OTf})_3$ in dry CH_2Cl_2 was considered as the best combination for this reaction and the scope of the reaction was investigated using various aldehydes (Table 4.3.2). In all of the cases studied, 6-oxabicyclo[3.2.1]octenes **24a-n** could be obtained in high purity without any side products as determined from ^1H to ^{13}C NMR spectrum of the crude product. Both aliphatic and aromatic aldehydes gave the corresponding products in good yields. The substituents on the aromatic ring have a promising effect on this reaction. The aromatic aldehydes having electron-donating groups on the aromatic ring gave higher yields compared to the electron-withdrawing groups on the ring. On the other hand, aliphatic aldehydes are better substrates than the aromatic aldehydes. This might be due to the better stability imparted to the oxocarbenium ion **26** (Scheme 4.3.1) by the aliphatic, and aromatic aldehydes having electron-donating groups on the aromatic ring, which in turn is attacked by the double bond efficiently. On the other hand, aromatic aldehydes having electron-withdrawing groups on the ring destabilize the oxocarbenium ion **26**. The reaction is stereoselective, and in all the cases the fusion between two rings is *cis*-fusion. The structure and stereochemistry of the compound was determined by NOE experiment and X-ray crystallographic analysis (Figure 4.3.1).¹² There is a strong NOE between H_1 (δ 2.29 ppm) and H_7 (δ 3.97 ppm) protons of **24j**, which confirms that they are *cis* to each other (**24j**, Figure 4.3.2). This structure is further confirmed by the fact that the olefinic methyl protons at C-2 position are resonating

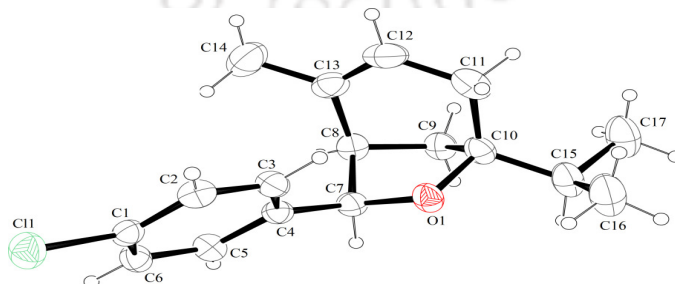
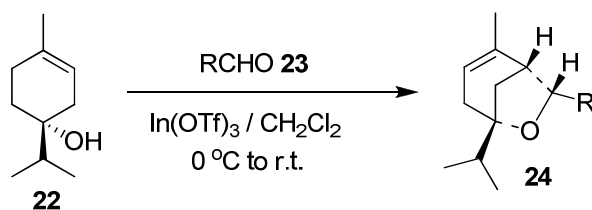
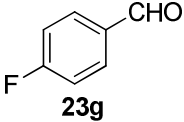
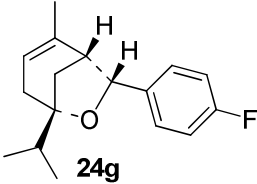
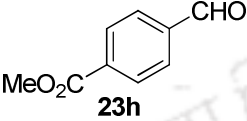
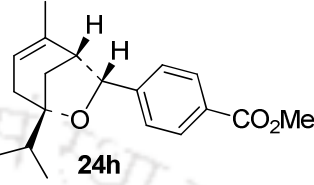
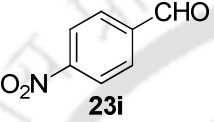
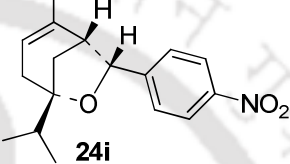
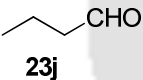
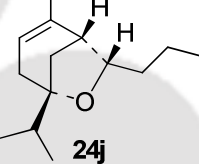
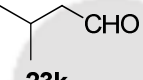
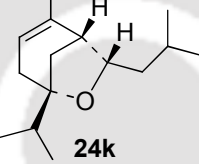
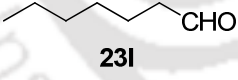
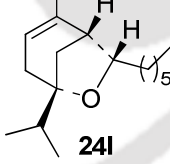
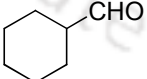
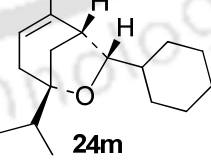
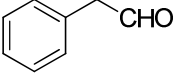
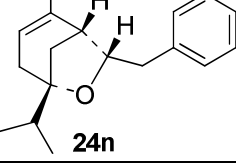


Figure 4.3.1. ORTEP diagram of (1*R*,5*R*,7*R*)-7-(4-chlorophenyl)-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene (**24d**).

Table 4.3.2. Synthesis of 6-oxabicyclo[3.2.1]octene

Entry	RCHO 23	Time (h)	Product 24	Yield ^a (%)
1		6		75
2		5		82
3		6		80
4		7		68
5		7		65
6		8		64

continue...

Entry	RCHO 23	Time (h)	Product 24	Yield ^a (%)
7	 23g	6	 24g	66
8	 23h	7	 24h	60
9	 23i	7	 24i	62
10	 23j	5	 24j	92
11	 23k	4	 24k	88
12	 23l	5	 24l	85
13	 23m	7	 24m	84
14	 23n	7	 24n	82

^aYields refer to isolated yield. The compounds are characterized by IR, ¹H, ¹³C, ¹⁹F NMR and Mass spectroscopy.

at higher field in the case of *endo* aromatic substituents at C-7 position than the corresponding aliphatic substituents. This indicates that the methyl protons are in the shielding region of the aromatic ring.

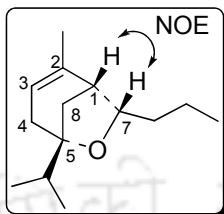
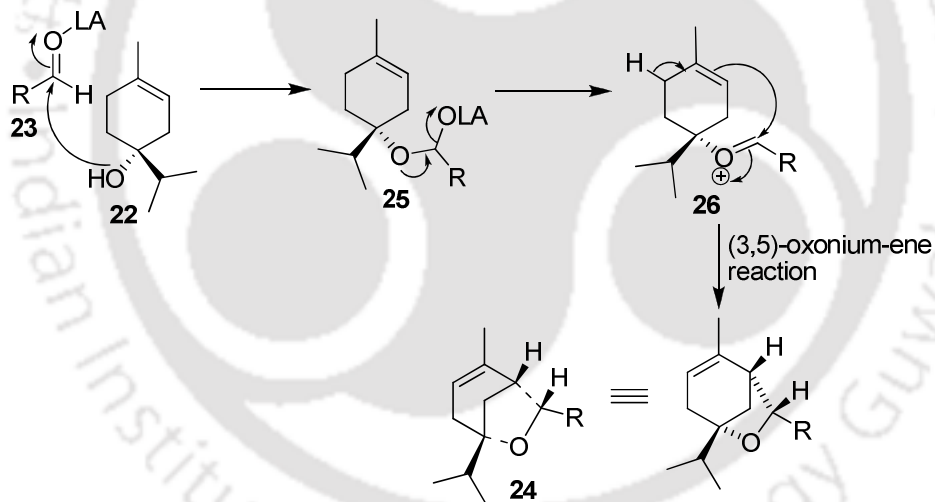


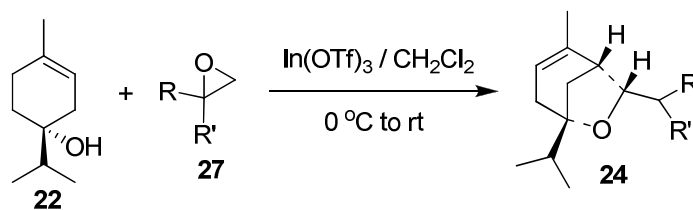
Figure 4.3.2. NOE diagram of compound **24j**

The mechanism of the reaction can be explained as follows. Lewis acid activates the aldehyde **23** for the nucleophilic attack by alcohol **22** to give acetal **25**, which after decomposition forms oxocarbenium ion **26**. The oxocarbenium ion **26** undergoes (3,5)-oxonium-ene cyclization reaction to give final product **24** (Scheme 4.3.1).



Scheme 4.3.1. Mechanism of the reaction

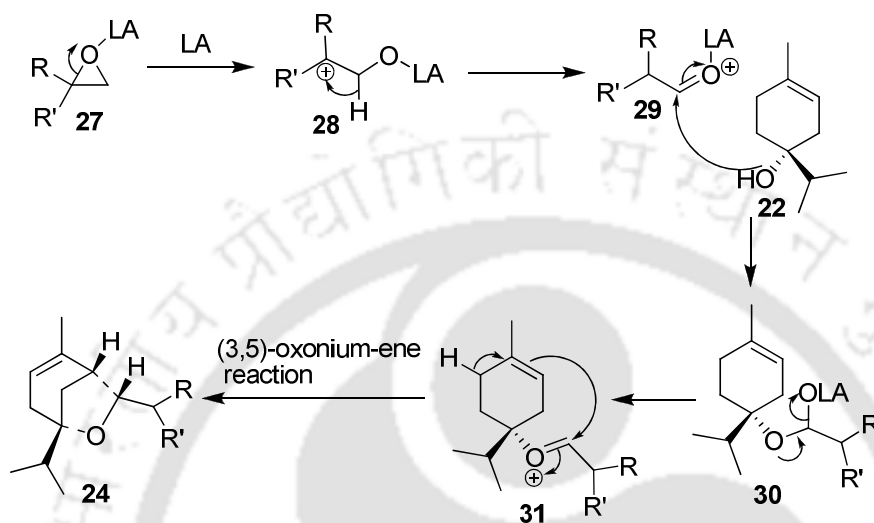
The reaction can also be extended to epoxides (Table 4.3.3). The epoxides under Lewis acid conditions rearrange to aldehyde equivalent.¹³ The results in Table 4.3.3 showed that monosubstituted terminal epoxides (entry 1) are unreactive, whereas the 2,2-disubstituted and styrene oxides give the corresponding products in good to moderate yields. This is attributed to the lower stability of the carbocation **28**, formed from monosubstituted epoxides, compared to 2,2-disubstituted epoxides and styrene oxides, where carbocation **28** is better stabilized due to the tertiary and benzylic centres, respectively.

Table 4.3.3. Synthesis of 6-oxabicyclo[3.2.1]octene from epoxide

Entry	Epoxide 27	Time/h	Product 24	Yield ^a (%)
1		24		0
2		6		68
3		7		62
4		8		55
5		8		58
6		7		60

^aYields refer to isolated yield. The compounds are characterized by IR, ^1H , ^{13}C NMR and Mass spectroscopy.

The mechanism of the reaction with epoxides is shown in Scheme 4.3.2. The epoxide **27** in the presence of Lewis acid rearranges to aldehyde equivalent **29**, which after nucleophilic attack by alcohol **22** gives acetal **30**. The acetal **30** decomposes to give oxocarbenium ion **31**, which after (3,5)-oxonium-ene cyclization reaction gives final product **24**.



Scheme 4.3.2. Mechanism of the reaction with epoxide

Conclusions:

In summary, we have developed an efficient methodology for the synthesis of 6-oxabicyclo[3.2.1]octene catalyzed by indium triflate in good to high yields. To the best of our knowledge, this is the first report for the synthesis of 6-oxabicyclo[3.2.1]octene starting from a readily available naturally occurring terpenoid.

4.4 Experimental Section

4.4.1 Instrumentation and Characterization

As described in Chapter 2, Section 2.4.1.

4.4.2 General Procedure for the Synthesis of Oxabicyclic Compounds

from Aldehydes: To a suspension of $\text{In}(\text{OTf})_3$ (0.1 mmol, 0.1 equiv.) in dry CH_2Cl_2 (2 ml) at 0 °C was added aldehyde (1 mmol, 1 equiv.). To this solution (-)-terpinen-4-ol (1 mmol, 1 equiv.) in dry CH_2Cl_2 (2 ml) was added drop by drop over 5 min. The temperature was slowly brought to room temperature. The reaction mixture was stirred at room temperature for a specified time. The progress of the reaction was monitored by TLC with

ethyl acetate and hexane as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with CH₂Cl₂ and then washed with brine and water. The organic layer was dried (Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give the title compounds.

Synthesis of (1R,5R,7R)-5-Isopropyl-2-methyl-7-phenyl-6-oxabicyclo[3.2.1]oct-2-ene (24a, Table 4.3.2): To a suspension of In(OTf)₃ (56 mg, 0.1 mmol) in dry CH₂Cl₂ (2 ml) at 0 °C was added benzaldehyde (106 mg, 1 mmol). To this solution (-)-terpinen-4-ol (154 mg, 1 mmol) in dry CH₂Cl₂ (2 ml) was added dropwise over 5 min. The temperature was slowly brought to room temperature. The reaction mixture was stirred at room temperature for 6 h. The progress of the reaction was monitored by TLC with hexane as eluent. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with CH₂Cl₂ and then washed with brine and water. The organic layer was dried (Na₂SO₄) and evaporated to give the crude product, which was purified by short column chromatography over silica gel (100% Hexane) to give **24a** (182 mg, 75%) as a colourless oil.

4.4.3 General Procedure for the Synthesis of Oxabicyclic Compounds from Epoxides: To a suspension of In(OTf)₃ (0.1 mmol, 0.1 equiv.) in dry CH₂Cl₂ (2 ml) at 0 °C was added epoxide (1.5 mmol, 1.5 equiv.). To this solution (-)-terpinen-4-ol (1 mmol, 1 equiv.) in dry CH₂Cl₂ (2 ml) was added drop by drop over 5 min. The temperature was slowly brought to room temperature. The reaction mixture was stirred at room temperature for a specified time. The progress of the reaction was monitored by TLC with hexane as eluent. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with CH₂Cl₂ and then washed with brine and water. The organic layer was dried (Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give the title compounds.

Synthesis of (1R,5R,7S)-5,7-Diisopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene (24p, Table 4.3.3): To a suspension of In(OTf)₃ (56 mg, 0.1 mmol) in dry CH₂Cl₂ (2 ml) at 0 °C was added 1,2-epoxy-2-methyl propane (108 mg, 1.5 mmol). To this solution (-)-terpinen-4-ol (154 mg, 1 mmol) in dry CH₂Cl₂ (2 ml) was added dropwise over 5 min. The temperature was slowly brought to room temperature. The reaction mixture was stirred at room temperature for 6 h. The progress of the reaction was monitored by TLC with

hexane as eluent. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with CH₂Cl₂ and then washed with brine and water. The organic layer was dried (Na₂SO₄) and evaporated to give the crude product, which was purified by short column chromatography over silica gel (100% Hexane) to give **24p** (142 mg, 68%) as a colourless oil.

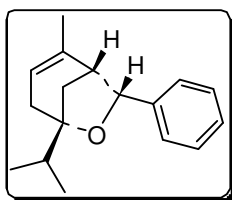
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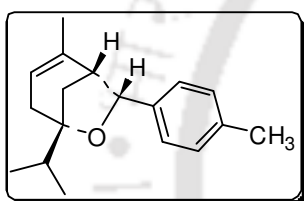
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13. (a) Ranu, B. C.; Jana, U. *J. Org. Chem.* **1998**, *63*, 8212-8216; (b) Li, J.; Li, C.-J. *Tetrahedron Lett.* **2001**, *42*, 793-796.



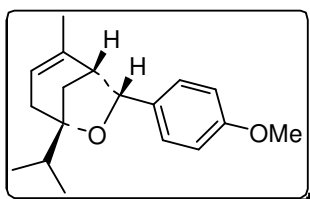
4.6 Spectral Data

(1R,5R,7R)-5-Isopropyl-2-methyl-7-phenyl-6-oxabicyclo[3.2.1]oct-2-ene (24a):

Colourless oil (182 mg, 75%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.00 (d, $J = 6.8$ Hz, 3 H), 1.01 (s, 3 H), 1.03 (d, $J = 6.8$ Hz, 3 H), 1.83-1.88 (m, 1 H), 1.95 (septet, $J = 6.8$ Hz, 1 H), 2.07 (dd, $J = 10.4$ and 4.0 Hz, 1 H), 2.13-2.21 (m, 1 H), 2.32-2.40 (m, 1 H), 2.55 (t, $J = 3.6$ Hz, 1 H), 5.22 (d, $J = 3.2$ Hz, 1 H), 5.27 (brs, 1 H), 7.17-7.21 (m, 1 H), 7.23-7.28 (m, 3 H), 7.29-7.33 (m, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 17.8, 18.0, 23.1, 36.0, 36.7, 38.5, 48.4, 85.8, 87.4, 120.0, 126.1, 126.9, 127.7, 136.6, 140.5; **IR** (Neat) 2962, 2934, 1602, 1449, 1380, 1286, 1053, 1017, 751, 702 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{22}\text{O}$ ($\text{M}+\text{H}$) $^+$ requires 243.1749; found 243.1747. Specific rotation $[\alpha]_{\text{D}}^{25} +130$ (c 1.07, CHCl_3).

(1R,5R,7R)-5-Isopropyl-2-methyl-7-*p*-tolyl-6-oxabicyclo[3.2.1]oct-2-ene (24b):

Colourless solid; M.P.: 58-60 °C (210 mg, 82%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.99 (d, $J = 7.2$ Hz, 3 H), 1.01 (d, $J = 6.8$ Hz, 3 H), 1.02 (s, 3 H), 1.81-1.86 (m, 1 H), 1.94 (septet, $J = 7.2$ Hz, 1 H), 2.05 (dd, $J = 10.4$ and 3.6 Hz, 1 H), 2.12-2.18 (m, 1 H), 2.30 (s, 3 H), 2.32-2.41 (m, 1 H), 2.51 (t, $J = 3.6$ Hz, 1 H), 5.17 (d, $J = 3.2$ Hz, 1 H), 5.25 (brs, 1 H), 7.05 (d, $J = 7.6$ Hz, 2 H), 7.17 (d, $J = 7.6$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 17.8, 18.0, 21.3, 23.3, 36.0, 36.8, 38.4, 48.4, 85.7, 87.4, 119.9, 126.1, 128.4, 136.3, 136.7, 137.5; **IR** (KBr) 2960, 2928, 1639, 1447, 1380, 1055, 1020, 800, 779 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{24}\text{O}$ (M^+) requires 256.1827; found 256.1822. Specific rotation $[\alpha]_{\text{D}}^{25} +133$ (c 0.24, CHCl_3).

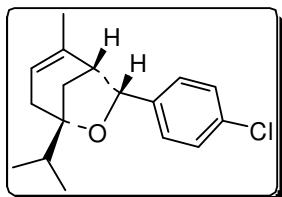
(1R,5R,7R)-5-Isopropyl-7-(4-methoxyphenyl)-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene**(24c):**

Colourless oil (218 mg, 80%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.99 (d, $J = 7.2$ Hz, 3 H), 1.01 (d, $J = 6.8$ Hz, 3 H), 1.04 (t, $J = 1.2$ Hz, 3 H), 1.81-1.86 (m, 1 H), 1.93 (septet, $J = 6.8$ Hz, 1 H), 2.04 (dd, $J = 10.4$ and 3.6 Hz, 1 H), 2.12-2.19 (m, 1 H), 2.32-2.39 (m, 1 H), 2.48 (t, $J = 4.0$ Hz, 1 H), 3.78 (s, 3 H), 5.16 (d, $J = 3.2$ Hz, 1 H), 5.27 (brs, 1 H), 6.80 (d, $J = 8.8$ Hz, 2 H), 7.21 (d, $J = 8.4$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 17.8, 18.0, 23.2, 36.0, 36.8, 38.4, 48.4, 55.4, 85.7, 87.1, 113.2, 119.9, 127.2, 132.6, 136.8, 158.7; **IR** (Neat) 2960, 2928, 1613, 1511, 1463, 1363, 1246, 1055, 1037, 845, 803 cm^{-1} .

HRMS (APCI) cald. for $C_{18}H_{24}O_2$ ($M+H$)⁺ requires 273.1854; found 273.1841. Specific rotation $[\alpha]_D^{25} +100$ (c 0.32, $CHCl_3$).

(1R,5R,7R)-7-(4-Chlorophenyl)-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene

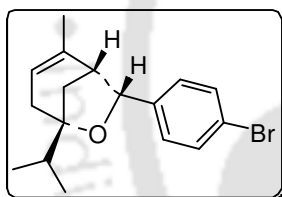
(24d):



Colourless solid; M.P.: 78-80 °C (188 mg, 68%); **¹H NMR** (400 MHz, $CDCl_3$): δ 0.98 (d, $J = 6.8$ Hz, 3 H), 1.01 (d, $J = 7.2$ Hz, 3 H), 1.03 (s, 3 H), 1.84-1.89 (m, 1 H), 1.93 (septet, $J = 6.8$ Hz, 1 H), 2.06 (dd, $J = 10.4$ and 3.6 Hz, 1 H), 2.11-2.18 (m, 1 H), 2.31-2.39 (m, 1 H), 2.52 (t, $J = 3.6$ Hz, 1 H), 5.17 (d, $J = 3.6$ Hz, 1 H), 5.27 (brs, 1 H), 7.18-7.28 (m, 4 H); **¹³C NMR** (100 MHz, $CDCl_3$): δ 17.8, 17.9, 23.2, 36.0, 36.7, 38.5, 48.4, 86.1, 86.7, 120.2, 127.6, 127.9, 132.5, 136.3, 139.2; **IR** (KBr) 2965, 2928, 1594, 1465, 1369, 1289, 1102, 1091, 849, 760 cm^{-1} . **HRMS** (APCI) cald. for $C_{17}H_{21}ClO$ ($M+H$)⁺ requires 277.1359; found 277.1348 (^{35}Cl). Specific rotation $[\alpha]_D^{25} +109$ (c 0.55, $CHCl_3$).

(1R,5R,7R)-7-(4-Bromophenyl)-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene

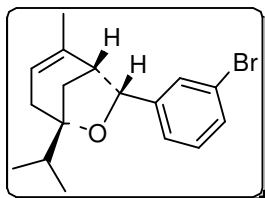
(24e):



Colourless solid; M.P.: 68-70 °C (209 mg, 65%); **¹H NMR** (400 MHz, $CDCl_3$): δ 0.98 (d, $J = 6.8$ Hz, 3 H), 1.01 (d, $J = 6.8$ Hz, 3 H), 1.03 (t, $J = 2.0$ Hz, 3 H), 1.83-1.88 (m, 1 H), 1.93 (septet, $J = 6.8$ Hz, 1 H), 2.05 (dd, $J = 10.8$ and 4.0 Hz, 1 H), 2.12-2.18 (m, 1 H), 2.31-2.39 (m, 1 H), 2.53 (t, $J = 3.6$ Hz, 1 H), 5.15 (d, $J = 3.2$ Hz, 1 H), 5.26 (brs, 1 H), 7.17 (d, $J = 8.4$ Hz, 2 H), 7.38 (d, $J = 8.4$ Hz, 2 H); **¹³C NMR** (100 MHz, $CDCl_3$): δ 17.7, 17.9, 23.2, 35.9, 36.6, 38.5, 48.2, 86.0, 86.6, 120.1, 120.6, 127.9, 130.7, 136.2, 139.7; **IR** (KBr) 2959, 2929, 1595, 1465, 1375, 1281, 1055, 1010, 847, 802 cm^{-1} . **HRMS** (APCI) cald. for $C_{17}H_{21}BrO$ ($M+H$)⁺ requires 321.0854; found 321.0852 (^{79}Br). Specific rotation $[\alpha]_D^{25} +103$ (c 1.84, $CHCl_3$).

(1R,5R,7R)-7-(3-Bromophenyl)-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene

(24f):

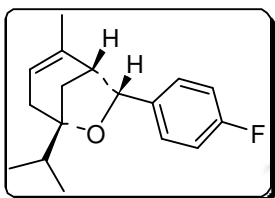


Colourless oil (206 mg, 64%); **¹H NMR** (400 MHz, $CDCl_3$): δ 0.99 (d, $J = 6.8$ Hz, 3 H), 1.02 (d, $J = 6.8$ Hz, 3 H), 1.04 (dd, $J = 2.0$ and 1.6 Hz, 3 H), 1.84-1.88 (m, 1 H), 1.93 (septet, $J = 6.8$ Hz, 1 H), 2.05 (dd, $J = 10.8$ and 4.0 Hz, 1 H), 2.12-2.20 (m, 1 H), 2.31-2.40 (m, 1 H), 2.55 (t, $J = 3.6$ Hz, 1 H), 5.15 (d, $J = 3.2$ Hz, 1 H), 5.29 (brs, 1 H), 7.12 (t, $J = 7.6$ Hz, 1 H), 7.18-7.22 (m, 1 H), 7.31-7.34 (m, 1 H), 7.43 (s, 1 H); **¹³C NMR** (100 MHz,

CDCl₃): δ 17.8, 17.9, 23.2, 35.9, 36.7, 38.5, 48.3, 86.2, 86.6, 120.2, 122.1, 124.9, 129.2, 129.3, 130.0, 136.1, 143.2; **IR** (Neat) 2965, 2934, 1593, 1467, 1421, 1369, 1294, 1260, 1050, 781, 747 cm⁻¹. **HRMS** (APCI) cald. for C₁₇H₂₁BrO (M+H)⁺ requires 321.0854; found 321.0848 (⁷⁹Br). Specific rotation [α]_D²⁵+105 (*c* 0.63, CHCl₃).

(1R,5R,7R)-7-(4-Fluorophenyl)-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene

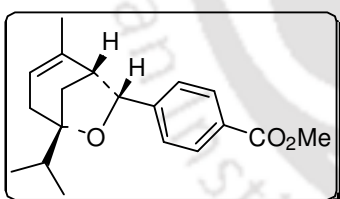
(24g):



Colourless oil (172 mg, 66%); **¹H NMR** (400 MHz, CDCl₃): δ 0.99 (d, *J* = 6.8 Hz, 3 H), 1.01 (d, *J* = 6.8 Hz, 3 H), 1.02 (s, 3 H), 1.82-1.88 (m, 1 H), 1.94 (septet, *J* = 6.8 Hz, 1 H), 2.05 (dd, *J* = 10.8 and 4.0 Hz, 1 H), 2.12-2.19 (m, 1 H), 2.32-2.40 (m, 1 H), 2.51 (t, *J* = 3.6 Hz, 1 H), 5.18 (d, *J* = 3.6 Hz, 1 H), 5.27 (brs, 1 H), 6.95 (t, *J* = 8.8 Hz, 2 H), 7.22-7.28 (m, 2 H); **¹³C NMR** (100 MHz, CDCl₃): δ 17.8, 18.0, 23.2, 36.0, 36.8, 38.5, 48.5, 86.0, 86.8, 114.6 (d, *J* = 21.3 Hz), 120.1 (d, *J* = 6.9 Hz), 127.7 (d, *J* = 7.6 Hz), 136.2, 136.4, 162.1 (d, *J* = 243.3 Hz); **¹⁹F NMR** (376 MHz, CDCl₃/C₆F₆): δ -220.51(-220.43) (m, 1 F); **IR** (Neat) 2961, 2933, 1606, 1508, 1465, 1380, 1221, 1056, 1020, 825, 803 cm⁻¹. **HRMS** (APCI) cald. for C₁₇H₂₁FO (M⁺) requires 260.1576; found 260.1568. Specific rotation [α]_D²⁵+131 (*c* 0.26, CHCl₃).

Methyl 4-((1R,5R,7R)-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-en-7-yl)benzoate

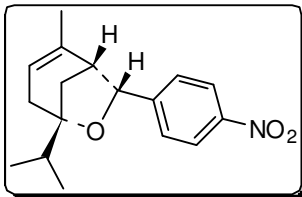
(24h):



Colourless oil (180 mg, 60%); **¹H NMR** (400 MHz, CDCl₃): δ 0.97 (t, *J* = 1.6 Hz, 3 H), 1.00 (d, *J* = 6.8 Hz, 3 H), 1.03 (d, *J* = 6.8 Hz, 3 H), 1.86-1.90 (m, 1 H), 1.95 (septet, *J* = 6.8 Hz, 1 H), 2.08 (dd, *J* = 10.4 and 3.6 Hz, 1 H), 2.15-2.22 (m, 1 H), 2.33-2.43 (m, 1 H), 2.59 (t, *J* = 3.2 Hz, 1 H), 3.90 (s, 3 H), 5.24-5.27 (m, 2 H), 7.36 (d, *J* = 8.4 Hz, 2 H), 7.94 (d, *J* = 8.4 Hz, 2 H); **¹³C NMR** (100 MHz, CDCl₃): δ 17.7, 17.9, 23.1, 35.9, 36.6, 38.5, 48.4, 52.0, 86.2, 87.0, 120.2, 126.1, 128.7, 129.1, 136.1, 146.2, 167.3; **IR** (Neat) 2961, 1722, 1611, 1435, 1277, 1111, 1056, 1017, 764, 713 cm⁻¹. **HRMS** (APCI) cald. for C₁₉H₂₄O₃ (M+H)⁺ requires 301.1803; found 301.1795. Specific rotation [α]_D²⁵+116 (*c* 1.77, CHCl₃).

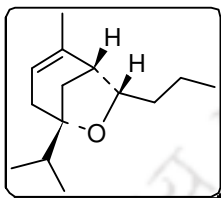
(1R,5R,7R)-5-Isopropyl-2-methyl-7-(4-nitrophenyl)-6-oxabicyclo[3.2.1]oct-2-ene (24i):

Colourless oil (178 mg, 62%); **¹H NMR** (400 MHz, CDCl₃): δ 0.99 (s, 3 H), 1.00 (d, *J* = 6.8 Hz, 3 H), 1.03 (d, *J* = 6.8 Hz, 3 H), 1.87-1.92 (m, 1 H), 1.96 (septet, *J* = 6.8 Hz, 1 H), 2.10 (dd, *J* = 10.8 and 4.0 Hz, 1 H), 2.16-2.23 (m, 1 H), 2.33-2.43 (m, 1 H), 2.64 (t, *J* = 3.6 Hz, 1



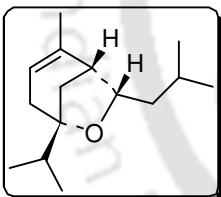
H), 5.27-5.30 (m, 2 H), 7.46 (d, $J = 8.4$ Hz, 2 H), 8.14 (d, $J = 8.8$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 17.7, 17.8, 23.1, 35.9, 36.5, 38.7, 48.4, 86.5, 86.7, 120.6, 123.0, 127.0, 135.7, 147.1, 148.7; **IR** (Neat) 2966, 2928, 1606, 1525, 1348, 1288, 1104, 1051, 1011, 853, 720 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{NO}_3$ ($\text{M}+\text{H}$)⁺ requires 288.1599; found 288.1587. Specific rotation $[\alpha]_{\text{D}}^{25} +113$ (c 0.30, CHCl_3).

(1R,5R,7S)-5-Isopropyl-2-methyl-7-propyl-6-oxabicyclo[3.2.1]oct-2-ene (24j):



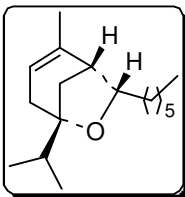
Colourless oil (192 mg, 92%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.92 (d, $J = 6.8$ Hz, 3 H), 0.93 (d, $J = 7.6$ Hz, 3 H), 0.94 (d, $J = 6.8$ Hz, 3 H), 1.30-1.35 (m, 1 H), 1.36-1.46 (m, 2 H), 1.51-1.58 (m, 1 H), 1.65-1.69 (m, 1 H), 1.70-1.73 (m, 3 H), 1.77-1.85 (m, 2 H), 1.86-1.94 (m, 1 H), 2.17-2.25 (m, 1 H), 2.29 (t, $J = 3.2$ Hz, 1 H), 3.97 (ddd, $J = 9.2, 6.0$ and 3.2 Hz, 1 H), 5.31 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.5, 17.8, 17.9, 20.2, 24.0, 34.4, 36.0, 36.8, 37.9, 44.5, 84.3, 86.6, 120.2, 136.7; **IR** (Neat) 2959, 2933, 2875, 1638, 1465, 1383, 1072, 1054, 968 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{14}\text{H}_{24}\text{O}$ ($\text{M}+\text{H}$)⁺ requires 209.1905; found 209.1896. Specific rotation $[\alpha]_{\text{D}}^{25} +45$ (c 0.56, CHCl_3).

(1R,5R,7S)-7-Isobutyl-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene (24k):



Colourless oil (196 mg, 88%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.90 (d, $J = 6.8$ Hz, 3 H), 0.91 (d, $J = 7.2$ Hz, 3 H), 0.92 (d, $J = 6.4$ Hz, 3 H), 0.94 (d, $J = 7.2$ Hz, 3 H), 1.29 (septet, $J = 7.2$ Hz, 1 H), 1.42 (septet, $J = 6.8$ Hz, 1 H), 1.61-1.69 (m, 2 H), 1.70-1.73 (m, 3 H), 1.77-1.85 (m, 2 H), 1.86-1.93 (m, 1 H), 2.17-2.24 (m, 1 H), 2.26 (t, $J = 4.0$ Hz, 1 H), 4.07 (ddd, $J = 10.0, 6.4$ and 3.2 Hz, 1 H), 5.30 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 17.7, 18.0, 23.2, 23.3, 24.1, 25.9, 36.0, 37.0, 37.9, 41.1, 44.8, 84.3, 84.8, 120.2, 136.7; **IR** (Neat) 2957, 2931, 2873, 1638, 1466, 1382, 1077, 1048, 979 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{15}\text{H}_{26}\text{O}$ ($\text{M}+\text{H}$)⁺ requires 223.1972; found 223.1968. Specific rotation $[\alpha]_{\text{D}}^{25} +41$ (c 0.61, CHCl_3).

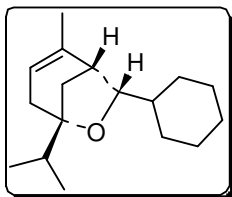
(1R,5R,7S)-7-Hexyl-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene (24l):



Colourless oil (213 mg, 85%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.87 (t, $J = 6.8$ Hz, 3 H), 0.92 (d, $J = 6.8$ Hz, 3 H), 0.94 (d, $J = 7.2$ Hz, 3 H), 1.20-1.35 (m, 8 H), 1.38-1.45 (m, 1 H), 1.53-1.61 (m, 1 H), 1.65-1.70 (m, 1 H), 1.71-1.73 (m, 3 H), 1.77-1.85 (m, 2 H), 1.86-1.94 (m, 1 H), 2.17-2.25 (m, 1 H), 2.30 (t, $J = 3.6$ Hz, 1 H), 3.96 (ddd, $J = 10.0, 6.8$ and 3.2 Hz, 1 H), 5.31 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.2, 17.8, 18.0, 22.8, 24.0, 27.0, 29.8, 32.0, 32.2, 36.0, 36.8,

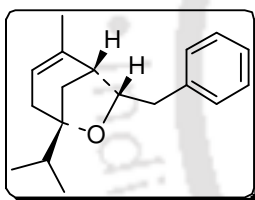
38.0, 44.5, 84.4, 87.0, 120.2, 136.7; **IR** (Neat) 2958, 2930, 2873, 1639, 1466, 1383, 1075, 1054, 875 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{30}\text{O}$ ($\text{M}+\text{H}$)⁺ requires 251.2375; found 251.2368. Specific rotation $[\alpha]_{\text{D}}^{25} +46$ (*c* 0.48, CHCl_3).

(1R,5R,7S)-7-Cyclohexyl-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene (24m):



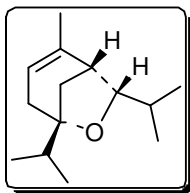
Colourless oil (209 mg, 84%); **¹H NMR** (400 MHz, CDCl_3): δ 0.80-0.91 (m, 4 H), 0.94 (d, *J* = 7.2 Hz, 3 H), 0.96 (d, *J* = 7.2 Hz, 3 H), 1.07-1.11 (m, 1 H), 1.12-1.17 (m, 2 H), 1.18-1.24 (m, 2 H), 1.25-1.29 (m, 1 H), 1.54-1.58 (m, 1 H), 1.60-1.66 (m, 1 H), 1.69 (dd, *J* = 2.0 and 1.6 Hz, 3 H), 1.71-1.76 (m, 1 H), 1.81 (septet, *J* = 6.8 Hz, 1 H), 1.97-2.05 (m, 1 H), 2.20 (t, *J* = 2.6 Hz, 1 H), 2.25-2.29 (m, 1 H), 3.51 (d, *J* = 9.6 Hz, 1 H), 5.18 (brs, 1 H); **¹³C NMR** (100 MHz, CDCl_3): δ 18.1, 18.2, 22.1, 26.1(2C), 26.7, 29.7, 30.0, 34.4, 35.8, 36.2, 43.0, 43.1, 84.8, 92.7, 118.4, 140.9; **IR** (Neat) 2950, 2924, 2851, 1641, 1448, 1380, 1083, 1055, 861 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{28}\text{O}$ ($\text{M}+\text{H}$)⁺ requires 249.2218; found 249.2212. Specific rotation $[\alpha]_{\text{D}}^{25} +42$ (*c* 0.57, CHCl_3).

(1R,5R,7S)-7-Benzyl-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene (24n):



Colourless oil (210 mg, 82%); **¹H NMR** (400 MHz, CDCl_3): δ 0.92 (d, *J* = 6.8 Hz, 3 H), 0.95 (d, *J* = 6.8 Hz, 3 H), 1.68-1.76 (m, 4 H), 1.79-1.87 (m, 2 H), 1.97-2.04 (m, 1 H), 2.21-2.28 (m, 1 H), 2.34 (t, *J* = 3.6 Hz, 1 H), 2.72 (dd, *J* = 14.4 and 6.4 Hz, 1 H), 2.95 (dd, *J* = 14.4 and 7.2 Hz, 1 H), 4.33 (ddd, *J* = 10.0, 6.4 and 3.2 Hz, 1 H), 5.39 (brs, 1 H), 7.16-7.30 (m, 5 H); **¹³C NMR** (100 MHz, CDCl_3): δ 17.9(2C), 24.1, 36.1, 36.8, 38.2, 38.4, 44.8, 84.8, 86.6, 120.7, 126.0, 128.3, 129.2, 136.6, 140.1; **IR** (Neat) 2960, 2928, 1638, 1453, 1382, 1064, 1055, 751, 700 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{24}\text{O}$ ($\text{M}+\text{H}$)⁺ requires 257.1905; found 257.1894. Specific rotation $[\alpha]_{\text{D}}^{25} +20$ (*c* 0.30, CHCl_3).

(1R,5R,7S)-5,7-diisopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene (24p):

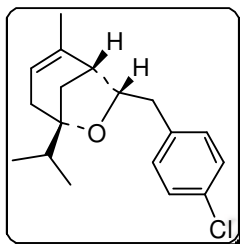


Colourless oil (142 mg, 68%); **¹H NMR** (400 MHz, CDCl_3): δ 0.86 (d, *J* = 6.4 Hz, 3 H), 0.90 (d, *J* = 6.8 Hz, 3 H), 0.93 (d, *J* = 6.8 Hz, 3 H), 0.98 (d, *J* = 6.4 Hz, 3 H), 1.56-1.64 (m, 1 H), 1.65-1.69 (m, 1 H), 1.74-1.78 (m, 3 H), 1.79-1.84 (m, 2 H), 1.85-1.93 (m, 1 H), 2.15-2.22 (m, 1 H), 2.40 (t, *J* = 2.8 Hz, 1 H), 3.50 (dd, *J* = 10.0 and 2.8 Hz, 1 H), 5.30 (brs, 1 H); **¹³C NMR** (100 MHz, CDCl_3): δ 17.8, 17.9, 20.1, 20.7, 24.2, 30.2, 36.2, 36.7, 38.5, 43.7, 84.1, 93.2, 120.6, 136.9; **IR** (Neat) 2958, 2928, 1640, 1467, 1386, 1278, 1056, 1020, 858 cm^{-1} . **HRMS**

(APCI) cald. for $C_{14}H_{24}O$ ($M+H$)⁺ requires 209.1905; found 209.1898. Specific rotation $[\alpha]_D^{25} +30$ (*c* 0.20, $CHCl_3$).

(1*R*,5*R*,7*S*)-7-(4-Chlorobenzyl)-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene

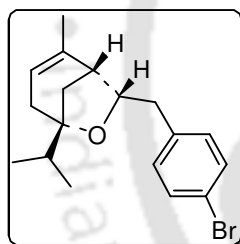
(24q):



Colourless oil (160 mg, 55%); ¹H NMR (400 MHz, $CDCl_3$): δ 0.91 (d, *J* = 6.8 Hz, 3 H), 0.94 (d, *J* = 6.8 Hz, 3 H), 1.69-1.76 (m, 4 H), 1.77-1.85 (m, 2 H), 1.94-2.03 (m, 1 H), 2.20-2.28 (m, 1 H), 2.32 (t, *J* = 3.6 Hz, 1 H), 2.68 (dd, *J* = 14.4 and 5.2 Hz, 1 H), 2.87 (dd, *J* = 14.4 and 8.0 Hz, 1 H), 4.25 (ddd, *J* = 10.4, 6.0 and 3.2 Hz, 1 H), 5.39 (brs, 1 H), 7.17-7.26 (m, 4 H); ¹³C NMR (100 MHz, $CDCl_3$): δ 17.9(2C), 24.0, 36.1, 36.7, 38.0, 38.2, 45.0, 85.0, 86.2, 120.8, 128.4, 130.6, 131.8, 136.5, 138.8; IR (Neat) 2961, 2928, 1637, 1467, 1380, 1078, 1055, 877, 831, 805 cm^{-1} . HRMS (APCI) cald. for $C_{18}H_{23}ClO$ ($M+H$)⁺ requires 291.1515; found 291.1504 (³⁵Cl). Specific rotation $[\alpha]_D^{25} +20$ (*c* 0.44, $CHCl_3$).

(1*R*,5*R*,7*S*)-7-(4-Bromobenzyl)-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene

(24r):



Colourless oil (194 mg, 58%); ¹H NMR (400 MHz, $CDCl_3$): δ 0.90 (d, *J* = 7.2 Hz, 3 H), 0.94 (d, *J* = 6.8 Hz, 3 H), 1.69-1.76 (m, 4 H), 1.78-1.85 (m, 2 H), 1.94-2.02 (m, 1 H), 2.19-2.27 (m, 1 H), 2.32 (t, *J* = 3.6 Hz, 1 H), 2.66 (dd, *J* = 14.4 and 5.2 Hz, 1 H), 2.84 (dd, *J* = 14.8 and 8.0 Hz, 1 H), 4.24 (ddd, *J* = 10.4, 6.4 and 3.6 Hz, 1 H), 5.39 (brs, 1 H), 7.13 (d, *J* = 8.0 Hz, 2 H), 7.37 (d, *J* = 8.4 Hz, 2 H); ¹³C NMR (100 MHz, $CDCl_3$): δ 17.9(2C), 24.1, 36.1, 36.7, 38.0, 38.2, 45.0, 85.0, 86.2, 119.8, 120.8, 131.0, 131.3, 136.5, 139.3; IR (Neat) 2960, 2928, 1641, 1467, 1382, 1071, 1053, 878, 834, 802 cm^{-1} . HRMS (APCI) cald. for $C_{18}H_{23}BrO$ ($M+H$)⁺ requires 335.1010; found 335.1019 (⁷⁹Br). Specific rotation $[\alpha]_D^{25} +32$ (*c* 0.40, $CHCl_3$).

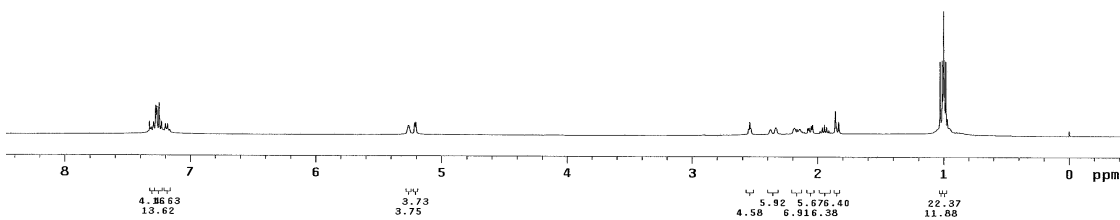
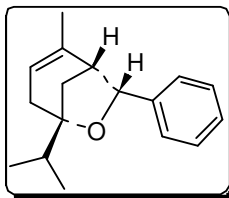
4.7 Selected Spectra of 6-Oxabicyclo[3.2.1]octenes

(1*R*,5*R*,7*R*)-5-Isopropyl-2-methyl-7-phenyl-6-oxabicyclo[3.2.1]oct-2-ene (24a):¹H NMR (400 MHz, CDCl₃)

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ACQUISITION aifa 20.000
sw 6389.8 FLAGS
at 1.938 i1 n
np 25520 fn n
Tb not used dp y
bs 8 hs nn
dl 1.000 PROCESSING
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ct 200 fn 65536
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scf 362.6 rF1 790.4
tpwr 57 rfp 0
pw 9.850 rp 118.7
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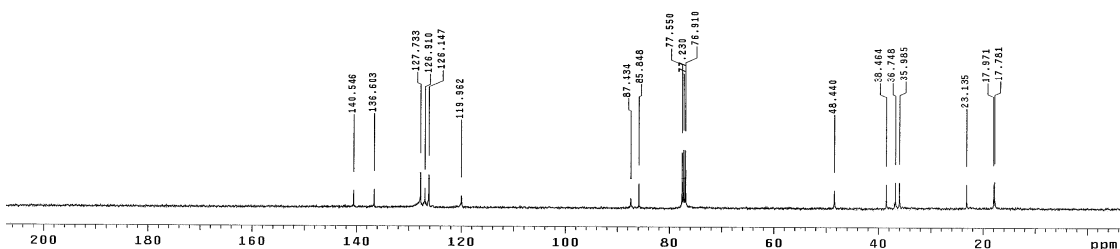
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¹³C NMR (100 MHz, CDCl₃)

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sw 25125.6 FLAGS
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pw 9.300 rp -75.3
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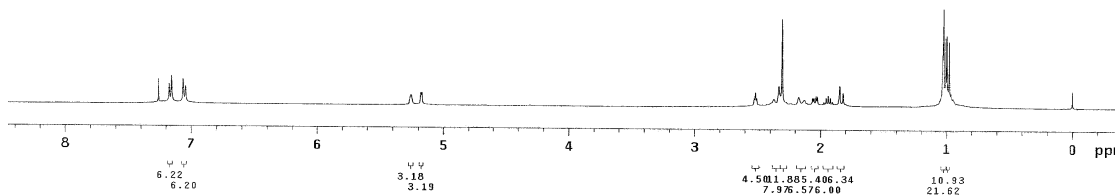
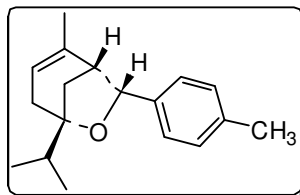


(1R,5R,7R)-5-Isopropyl-2-methyl-7-p-tolyl-6-oxabicyclo[3.2.1]oct-2-ene (24b):¹H NMR (400 MHz, CDCl₃)

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dl 1.000 dp y
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ct 200 PROCESSING
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tof 362.8 sp -173.9
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pw 9.850 rfd 735.2
dn DECOUPLER C13 rfd 0
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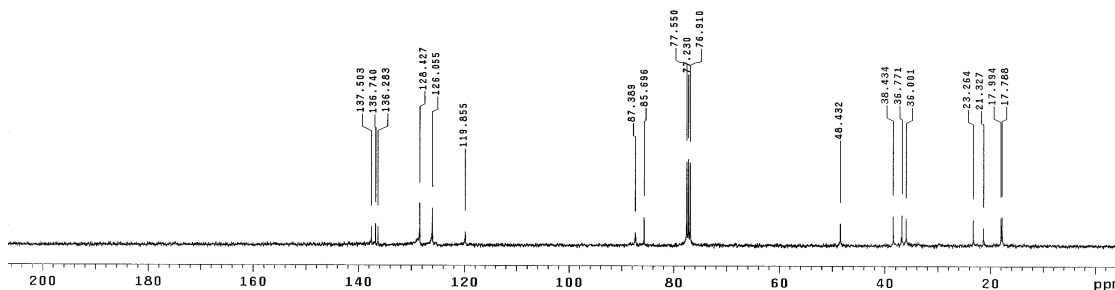
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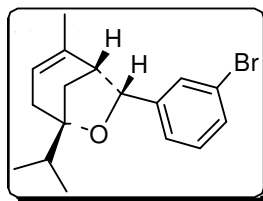
¹³C NMR (100 MHz, CDCl₃)

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alpha 20.000
sw 29123.5 FLAGS
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fb 13800 dp y
bs 16 hs nn
dl 1.000 PROCESSING
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pw 9.300 rp -71.6
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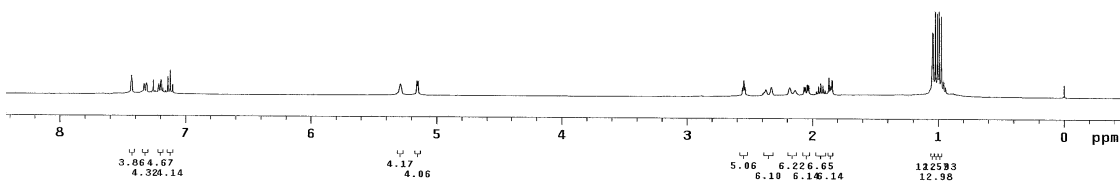


(1R,5R,7R)-7-(3-Bromophenyl)-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene**(24f):**¹H NMR (400 MHz, CDCl₃)

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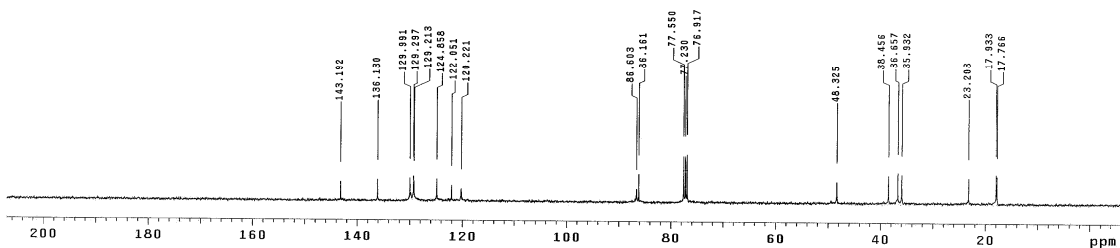
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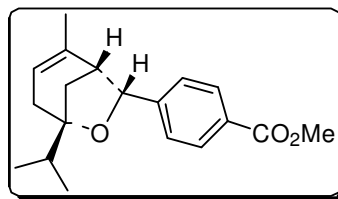
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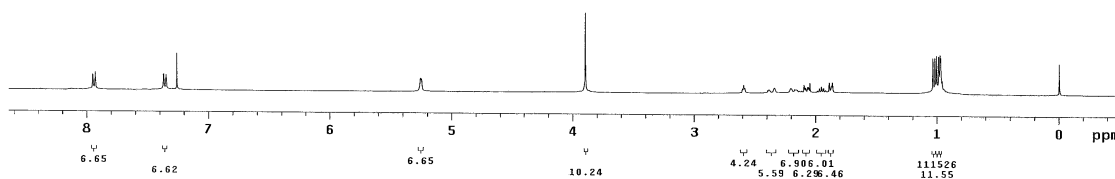
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(24h):**

^1H NMR (400 MHz, CDCl_3)



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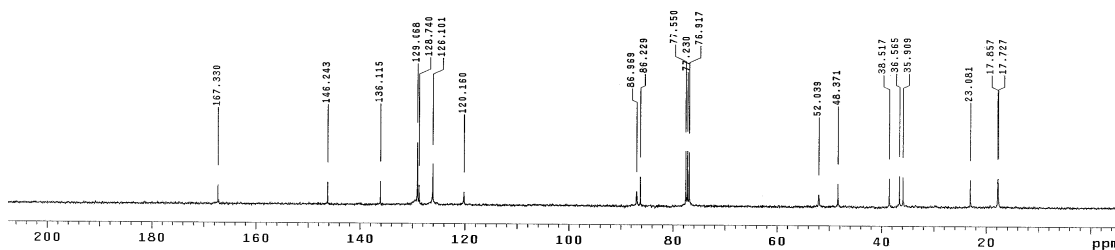
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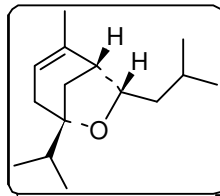


^{13}C NMR (100 MHz, CDCl_3)

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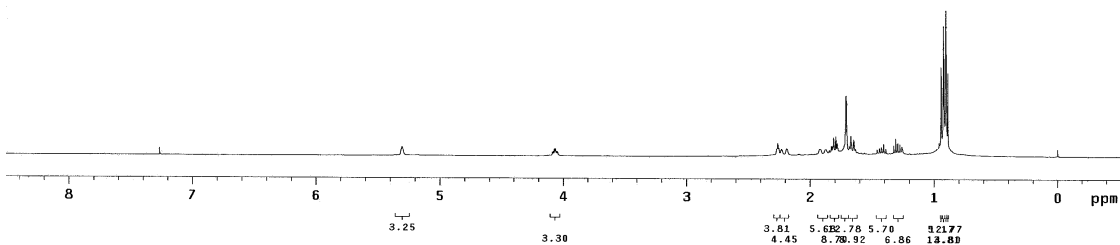


(1R,5R,7S)-7-Isobutyl-5-isopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene (24k):¹H NMR (400 MHz, CDCl₃)

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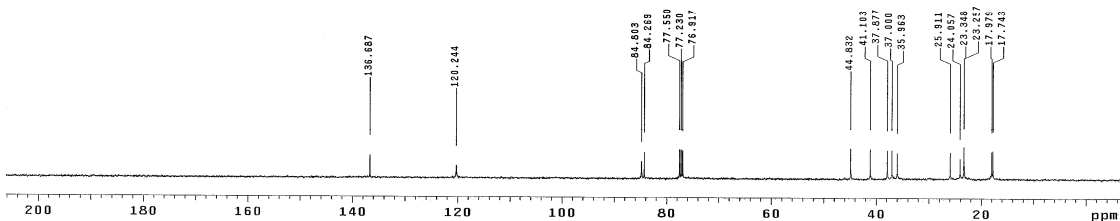
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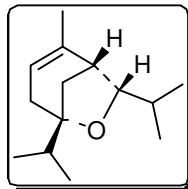
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(1R,5R,7S)-5,7-diisopropyl-2-methyl-6-oxabicyclo[3.2.1]oct-2-ene (24p):¹H NMR (400 MHz, CDCl₃)

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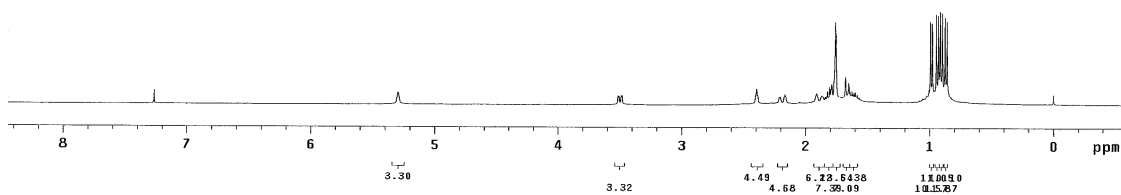
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daf 15900 nm cdc ph

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¹³C NMR (100 MHz, CDCl₃)

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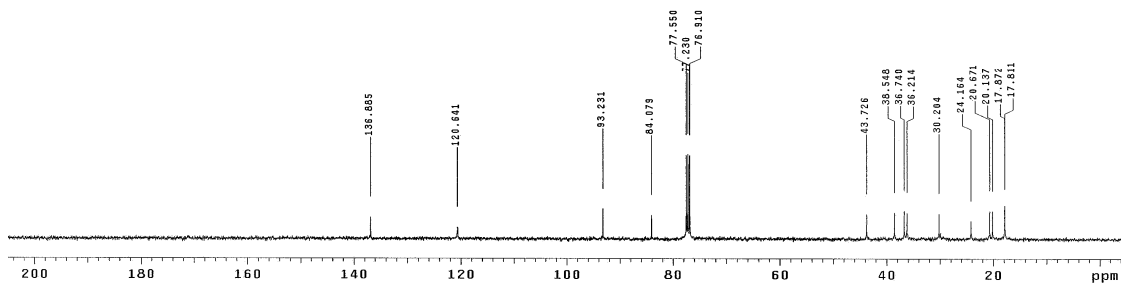
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daf 8900 nm no ph

```



4.8 The Crystal Parameters of Compound 24d

Crystal Parameters	24d-CCDC 840604
Formula	C ₁₇ H ₂₁ ClO
Formula weight	276.62
<i>T</i> /K	293(2)
Crystal system	Monoclinic
Space group	P 21
<i>a</i> /Å	7.251(2)
<i>b</i> /Å	7.497(2)
<i>c</i> /Å	13.967(4)
α /°	90.00(17)
β /°	92.81(2)
γ /°	90.00(18)
<i>V</i> /Å ³	758.3(4)
<i>Z</i>	2
Abs. Coeff./mm ⁻¹	0.242
Abs. Correction	None
GOF on <i>F</i> ²	0.993
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>RI</i> = 0.0384 <i>wR2</i> = 0.0932
<i>R</i> indices [all data]	<i>RI</i> = 0.0484 <i>wR2</i> = 0.1014

CHAPTER 5

Synthesis of 2,3,5,6-Tetrasubstituted Tetrahydropyrans via (3,5)-Oxonium-Ene Reaction

5.1 Importance and Applications

The tetrahydropyrans are important structural features in synthetic organic chemistry as they are core units of many biologically active molecules and natural products.¹ The polyether antibiotics, marine toxins and pheromones belong to such biologically active natural products.² The 2,3,5,6-tetrasubstituted tetrahydropyran units are found in antibiotics such as FR-901464 and X-206 (Figure 5.1.1, A-B).³ The presence of ester group in the tetrahydropyran ring is important because a diverse functionalization could be achieved in the tetrahydropyran molecule.

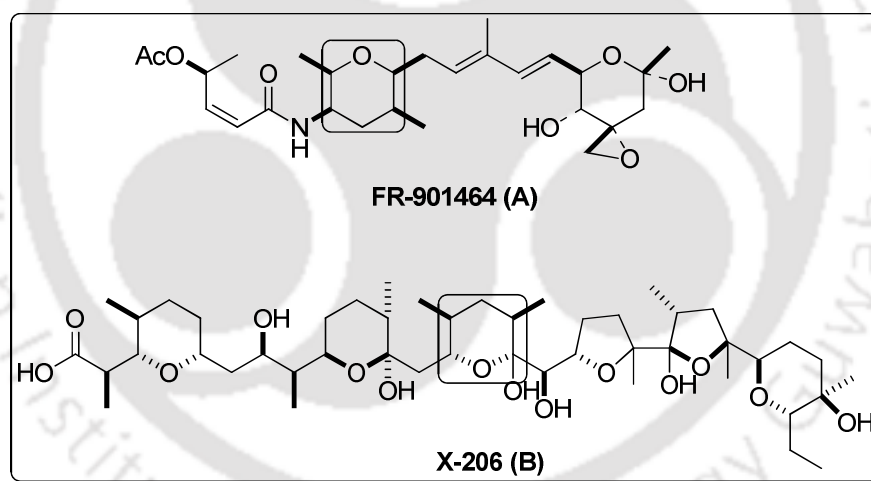
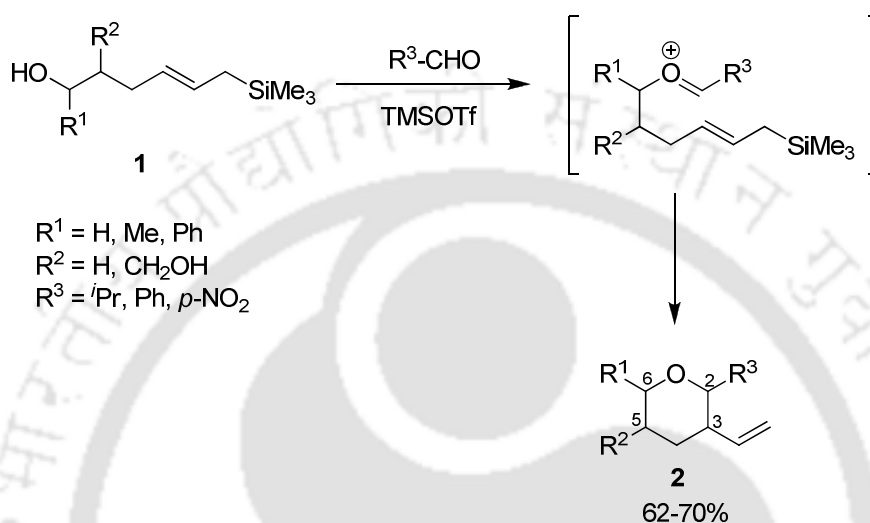


Figure 5.1.1. Some biologically important molecules containing tetrahydropyran as core unit

5.2 An Overview of Relevant Synthetic Methods

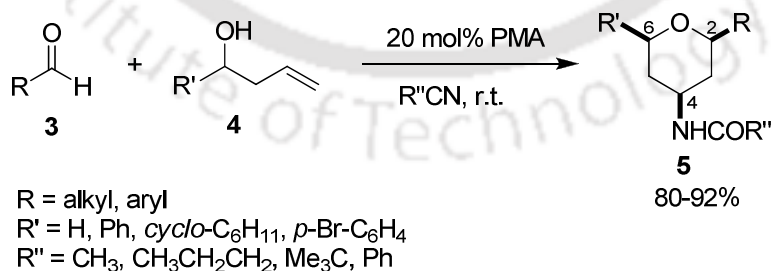
Review of literature reveals that several strategies are developed for the synthesis of 2,4-, 2,6-, 2,3,6-, 2,4,5-, 2,4,6-, 2,3,4,6- and 2,4,5,6-substituted tetrahydropyrans. However synthetic procedure for the synthesis of 2,3,5,6-tetrasubstituted tetrahydropyrans are limited.⁴ Tetrahydropyrans are generally prepared by hetero Diels-Alder reactions, manipulation of carbohydrates, intramolecular Michael reactions and Prins cyclization. These are already discussed in the introductory part (Chapter 1).

Szabó and co-workers have reported a methodology for the synthesis of 2,3,5,6-substituted tetrahydropyrans from the reaction of functionalized allylsilanes **1** and aldehydes **2** mediated by Lewis acid with excellent stereoselectivity (Scheme 5.2.1).⁴ According to DFT calculations the high stereoselectivity arises from electronically induced steric effects occurring in the key-intermediate of the cyclization.



Scheme 5.2.1.

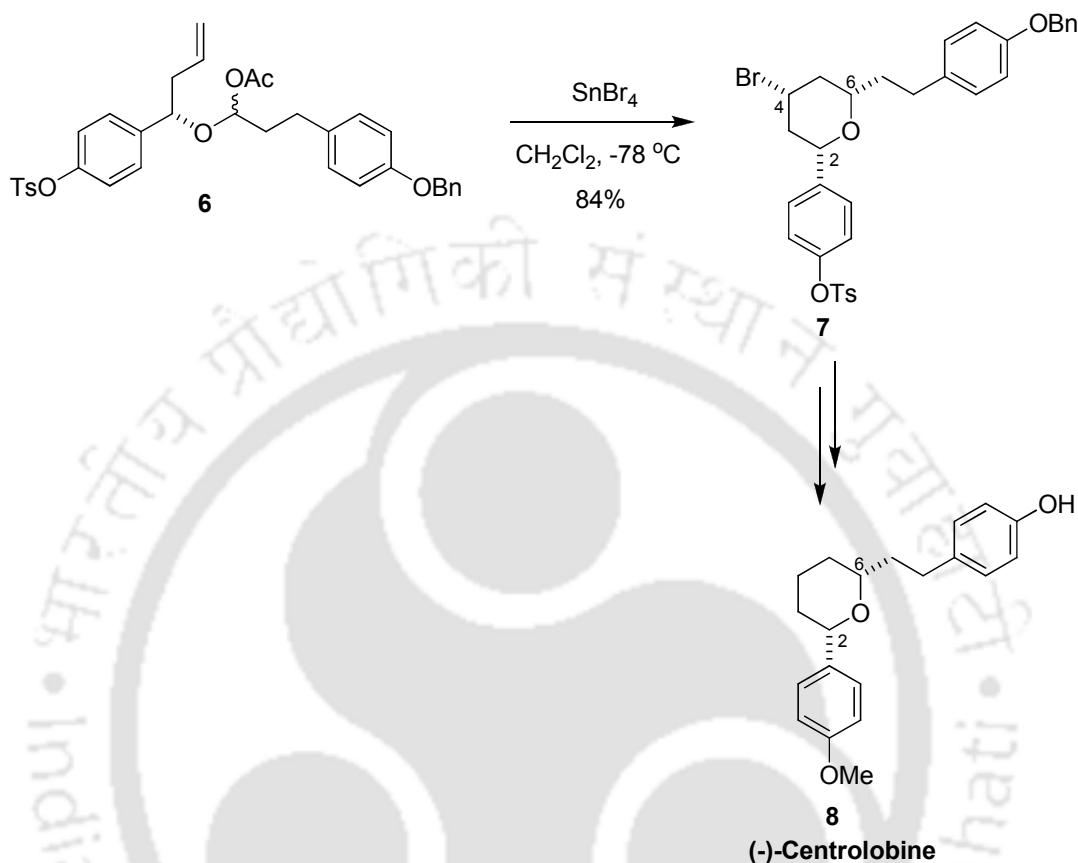
Yadav *et al.* have reported an efficient Prins-Ritter reaction to produce 2,4- and 2,4,6-substituted 4-amidotetrahydropyrans via a three component coupling of carbonyl compounds **3**, homoallylic alcohols **4** and nitriles in the presence of 20 mol % of phosphomolybdic acid (Scheme 5.2.2).⁵ The reaction is diastereoselective and the tetrahydropyrans are obtained in high yields with all *cis* selectivity.



Scheme 5.2.2.

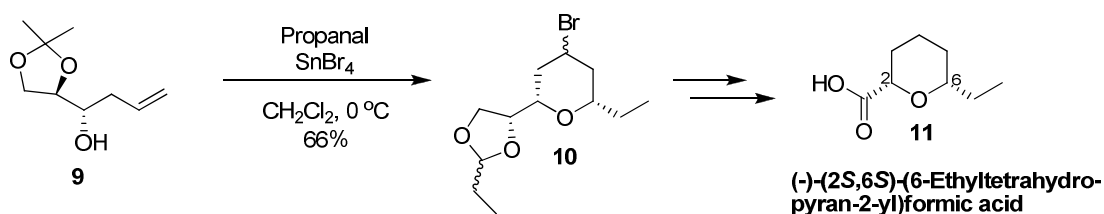
Rychnovsky and co-workers have reported the synthesis of (-)-centrolobine **8**, an antibiotic isolated from the heartwood of *Centrolobium robustum*, via the Prins cyclization of enantiopure α -acetoxy ether **6** in the presence of SnBr_4 . The initially formed 4-bromo-

tetrahydropyran **7** has been converted to (-)-centrolobine **8** via consecutive steps (Scheme 5.2.3).⁶



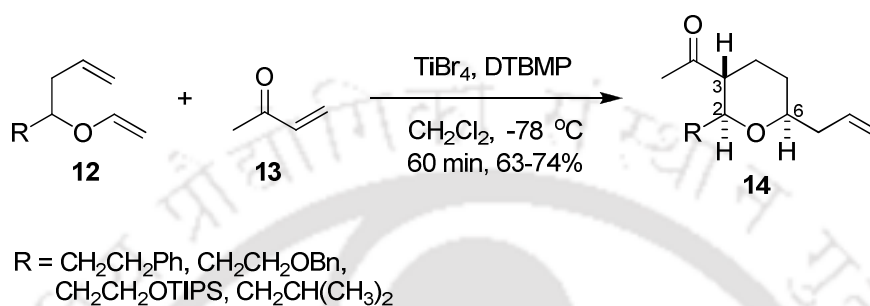
Scheme 5.2.3.

(±)-*cis*-(6-Ethyltetrahydropyran-2-yl)formic acid **11**, is a novel class of non-steroidal analgesic compound. Vasconcellos and co-workers described the enantioselective synthesis of (-)-(2*S*,6*S*)-(6-ethyl-tetrahydropyran-2-yl)-formic acid **11** in five steps (30% overall yield, 87% *ee*), from the commercial chiral template (*R*)-2,3-isopropylidene-glyceraldehyde. An efficient Prins cyclization between **9** with propanal mediated by SnBr₄ resulted in a mixture of four diastereomers **10** (Scheme 5.2.4).⁷ Reduction of the latter, followed by oxidative cleavage of diol, led to the levo enantiomer of **11** in 87% *ee*.



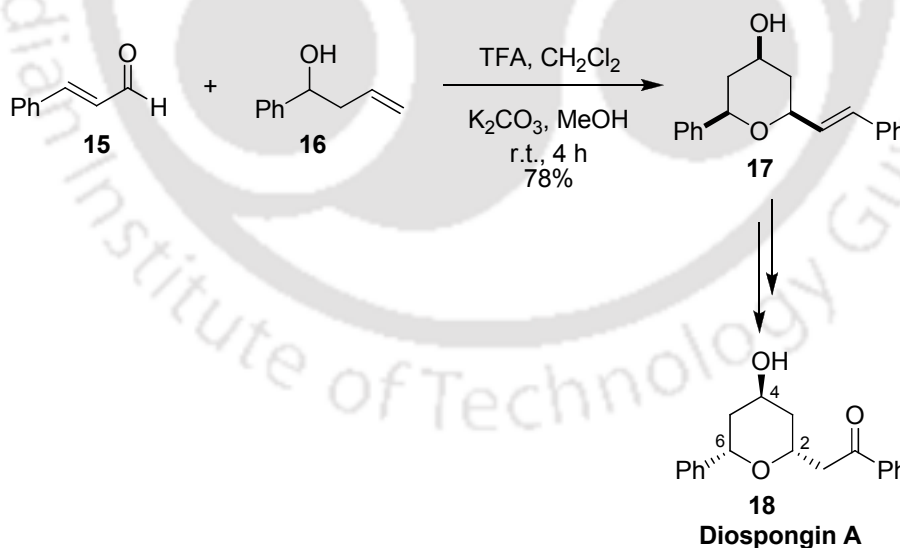
Scheme 5.2.4.

Rychnovsky and co-workers have reported the synthesis of 2,3,6-substituted tetrahydropyrans **14** from a Mukaiyama-Michael cascade reaction of homoallylic enol ethers **12** and 3-butene-2-one **13** promoted by TiBr_4 and 2,6-di-*tert*-butyl-4-methylpyridine (DTBMP) (Scheme 5.2.5).⁸ The resulting tetrahydropyrans **14** are obtained in high yields and as a single diastereomer.



Scheme 5.2.5.

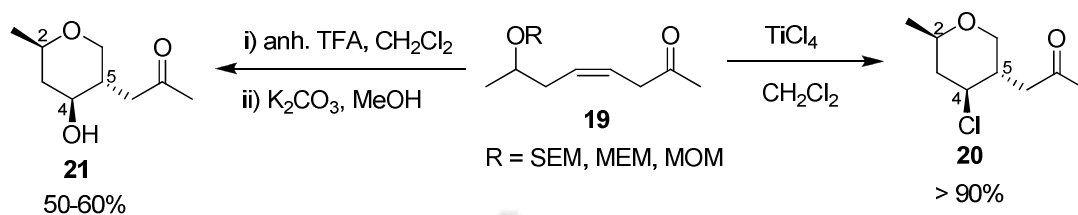
Yadav *et al.* developed a concise and efficient total synthesis of diospongin A using Prins cyclization and enzymatic kinetic resolution as key steps. Diospongin A is active against osteoporosis and 2,4,6-*cis*-trisubstituted tetrahydropyran ring was achieved via a Prins reaction of cinnamaldehyde and homoallylic alcohol (Scheme 5.2.6).⁹



Scheme 5.2.6.

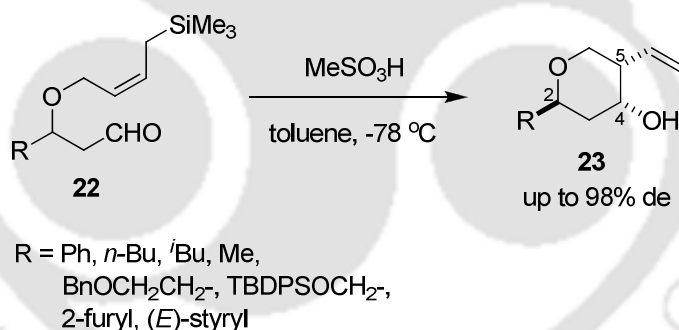
Willis and co-workers have reported a Prins type cyclization of homoallylic acetals for the synthesis of 2,4,5-trisubstituted tetrahydropyrans (Scheme 5.2.7).¹⁰ The reaction is versatile and by varying the acid and nucleophile a halide (bromide, chloride or fluoride),

oxygenated group (hydroxyl, ether or ester) or a nitrogen containing substituent may be introduced at C-4 of the tetrahydropyran.



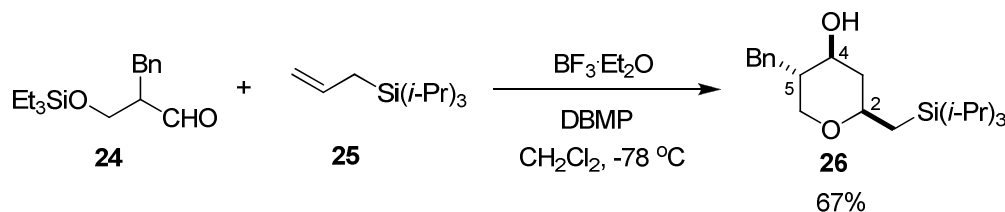
Scheme 5.2.7.

Cox and co-workers have reported a highly stereoselective route for the synthesis of 2,4,5-trisubstituted tetrahydropyrans **23** (Scheme 5.2.8).¹¹ The key step employs an intramolecular allylation of a (*Z*)-allylsilane onto an aldehyde under Brønsted acid activation. Complete 1,4-stereoselection accounts for the formation of only two out of the possible four THP products. The level of 1,3-stereoselection is optimal when the reaction is carried out in an apolar solvent, which is in accord with electrostatics being key to controlling this aspect of the stereoselectivity.



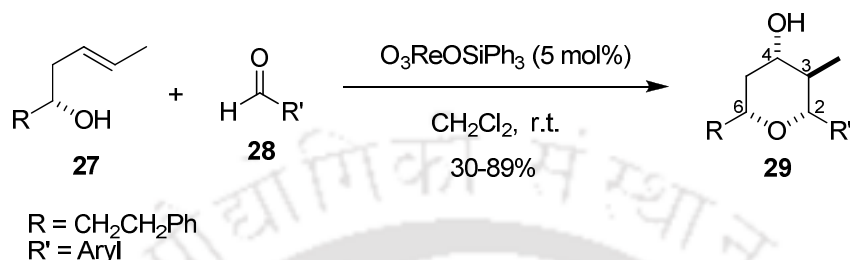
Scheme 5.2.8.

Angle *et al.* have reported stereoselective synthesis of 2,4,5-trisubstituted tetrahydropyrans via a formal [4+2]-cycloaddition of allylsilane.¹² The reaction of triethylsilyl-protected β -hydroxy aldehyde **24** with allyltriisopropylsilane **25** in the presence of BF₃·Et₂O produces 2,4,5-trisubstituted tetrahydropyran **26** in good yields (Scheme 5.2.9).



Scheme 5.2.9.

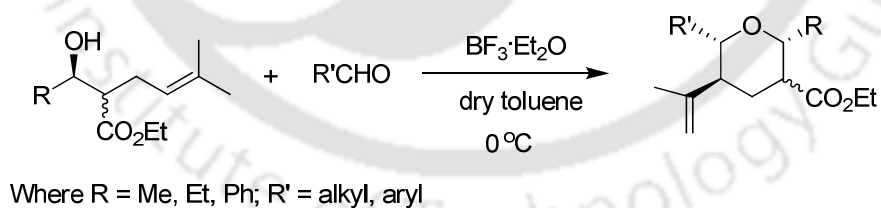
Rychnovsky and co-workers have reported the synthesis of 2,3,4,6-substituted tetrahydropyrans via Rhenium(VII) complex ($O_3ReOSiPh_3$) catalyzed Prins cyclization reaction between aromatic and α,β -unsaturated aldehydes with an enantiomerically pure homoallyl alcohol (Scheme 5.2.10).¹³



Scheme 5.2.10.

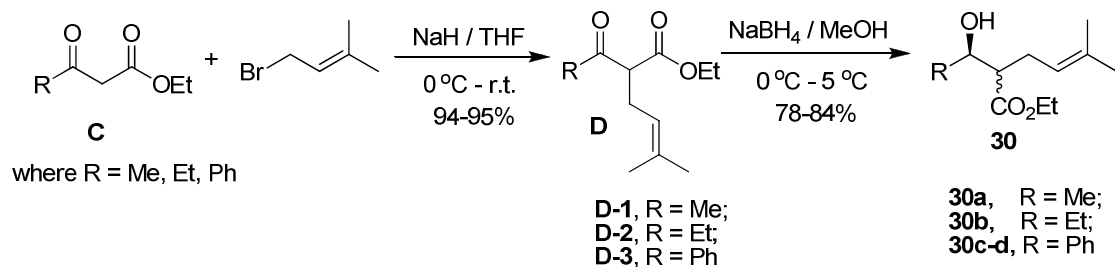
5.3 Present Work

In the previous chapters we have described efficient methodologies for the synthesis of oxabicyclo[3.3.1]nonanones, oxabicyclo[3.3.1]nonenes and substituted tetrahydropyrans, 6-oxabicyclo[3.2.1]octenes by using (3,5)-oxonium-ene reaction. In this chapter we are going to present an efficient methodology for the synthesis of 2,3,5,6-tetrasubstituted tetrahydropyrans from the reaction of aldehydes and ethyl 2-(1-hydroxyalkyl/hydroxy(phenyl)methyl)-5-methylhex-4-enoate using (3,5)-oxonium-ene reaction promoted by boron trifluoride etherate in good yields under mild conditions. The reaction can be generalized as shown in Scheme 5.3.1.



Scheme 5.3.1.

The ethyl 2-(1-hydroxyalkyl/hydroxy(phenyl)methyl)-5-methylhex-4-enoate **30** was synthesized starting from β -keto ester **C** and 1-bromo-3-methylbut-2-ene as shown in Scheme 5.3.2.¹⁴ Thus the reaction of β -keto ester **C** with 1-bromo-3-methylbut-2-ene in presence of sodium hydride in THF afforded α -substituted β -keto ester **D**, which after reduction with sodium borohydride in methanol gives alcohol **30** in 78-84% yields.



Scheme 5.3.2. Synthesis of ethyl 2-(1-hydroxyalkyl/hydroxy(phenyl)methyl)-5-methylhex-4-enoate

The methyl and ethyl substituted β -keto esters **D-1** and **D-2** afforded two inseparable diastereomers¹⁵ **30a,b** whereas phenyl substituted β -ketoester **D-3** gave two separable *anti*- and *syn*-diastereomers **30c** and **30d**, respectively. The structures of all compounds are determined from IR, ¹H, ¹³C NMR and mass spectroscopy. The stereochemistry of compounds **30c** and **30d** are determined from coupling constants values.^{15b,16}

The reaction of alcohols ethyl 2-(1-hydroxyethyl)-5-methylhex-4-enoate **30a** (1:1 diastereomeric mixture) and ethyl 2-(1-hydroxypropyl)-5-methylhex-4-enoate **30b** (1:1 diastereomeric mixture) with different aldehydes in dry toluene in the presence of boron trifluoride etherate afforded a mixture of diastereomeric products **32** and **33** with an epimeric center on C-3 bearing an ester group (Table 5.3.1). In all the cases the

Table 5.3.1. Synthesis of 2,3,5,6-tetrasubstituted tetrahydropyrans

Entry	Alcohol 30a , R = Me; 30b , R = Et	Aldehyde R' =	Time (min)	32+33 Yield ^{a,b} (%)
a	Me	C ₆ H ₅	60	78
b	Me	<i>p</i> -Me-C ₆ H ₄	60	72
c	Me	<i>p</i> -Br-C ₆ H ₄	60	66
d	Me	<i>p</i> -Cl-C ₆ H ₄	60	62
e	Me	<i>p</i> -F-C ₆ H ₄	60	64
f	Me	<i>p</i> -NO ₂ -C ₆ H ₄	60	62
g	Me	<i>p</i> -MeO ₂ C-C ₆ H ₄	60	68

continue...

Entry	Alcohol 30a , R = Me; 30b , R= Et	Aldehyde R' =	Time (min)	32+33 Yield ^{a,b} (%)
h	Me	C ₃ H ₇	30	86
i	Me	C ₆ H ₁₃	30	82
j	Me	(CH ₃) ₂ CHCH ₂	30	84
k	Et	C ₆ H ₅	90	74
l	Et	<i>p</i> -Me-C ₆ H ₄	90	70
m	Et	<i>p</i> -MeO ₂ C-C ₆ H ₄	90	68
n	Et	C ₃ H ₇	50	88
o	Et	C ₆ H ₁₃	50	80
p	Et	(CH ₃) ₂ CHCH ₂	50	84

^aThe ratio of formation of **32** and **33** is (1:1) and is determined by ¹H NMR of crude products. ^bTotal isolated yield of **32** and **33**. The compounds are characterized by IR, ¹H, ¹³C, ¹⁹F NMR and mass spectroscopy.

diastereomeric ratio was 1:1. The diastereomers **32** and **33** could be separated by column chromatography. The stereochemistry of **32** and **33** was determined by X-ray crystallographic analysis (Figure 5.3.1).¹⁸ In the ¹H NMR spectrum the axial proton at position C-4 (H_{4a}) of **32a** shows a ddd with coupling constants 12.8, 12.4 and 12.4 Hz at δ 1.99 ppm, which indicates that the proton is coupled with two neighbouring axial protons H_{3a} and H_{5a}. On the other hand same proton (H_{4a}) of **33a** shows a ddd with coupling constants 13.6, 12.8 and 5.2 Hz at δ 1.90 ppm, which is indicative of coupling with one axial H_{5a} and one equatorial proton H_{3e}. This clearly indicates that **32a** and **33a** are epimeric at C-3. A comparison of other ¹H NMR signals of **32a** and **33a** is shown in Table 5.3.2.

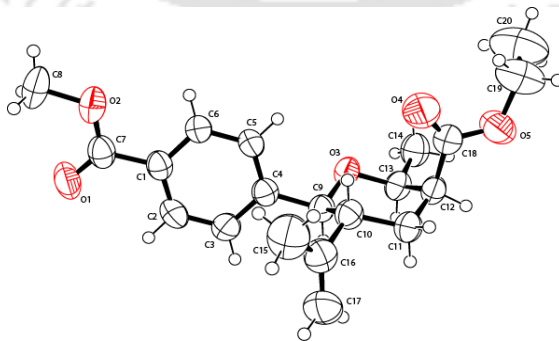
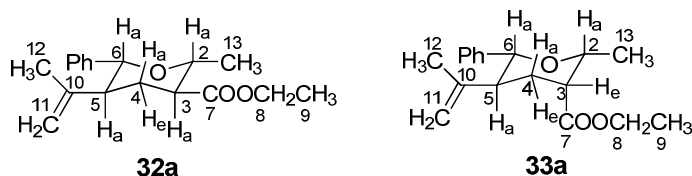
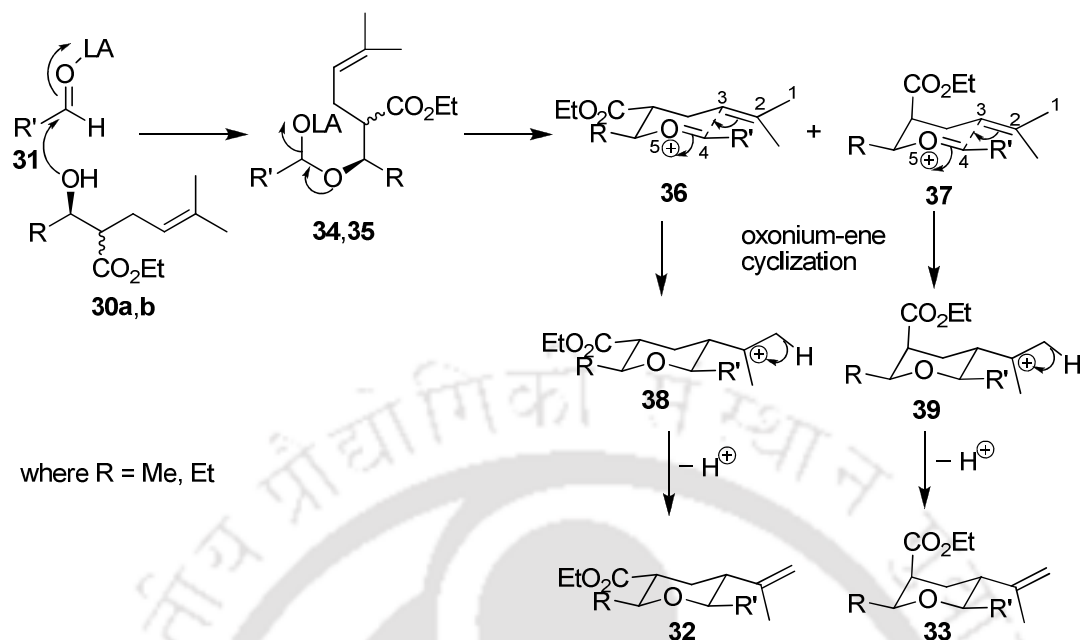


Figure 5.3.1. ORTEP diagram of (2*R**,3*R**,5*S**,6*R**)-ethyl 6-(4-(methoxycarbonyl)phenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (**33g**).

Table 5.3.2. Comparison of ^1H NMR data of **32a** and **33a**

Assignment	δ (ppm)	Multiplicity	Coupling constants (Hz)
13-CH ₃	1.27 (32a)	d	7.2
	1.32 (33a)	d	6.4
9-CH ₃	1.28 (32a)	t	7.2
	1.33 (33a)	t	7.2
12-CH ₃	1.43 (32a)	s	-
	1.42 (33a)	s	-
H _{4a}	1.99 (32a)	ddd	12.8, 12.4, 12.4
	1.90 (33a)	ddd	13.6, 12.8, 5.2
H _{4e}	2.09 (32a)	ddd	12.8, 4.0, 4.0
	2.14 (33a)	ddd	13.6, 4.0, 4.0
H _{3a}	2.33 (32a)	ddd	12.4, 10.4, 4.0
	2.60-2.67 (33a)	m	-
H _{3e}	2.46 (32a)	ddd	12.4, 10.0, 4.0
	3.01 (33a)	ddd	12.8, 12.4, 3.6
H _{2a}	3.76-3.85 (32a)	m	-
	3.82-3.89 (33a)	m	-
8-CH ₂	4.17 (32a)	q	7.2
	4.14-4.28 (33a)	m	-
H _{6a}	4.29 (32a)	d	10.0
	4.14-4.28 (33a)	m	-
11H	4.62 (32a)	brs	-
	4.63 (33a)	brs	-
11H	4.66 (32a)	brs	-
	4.67 (33a)	brs	-
Ar-H	7.22-7.32 (32a)	m	-
	7.22-7.25, 7.26-7.31	m	-
	7.32-7.36 (33a)	m	-

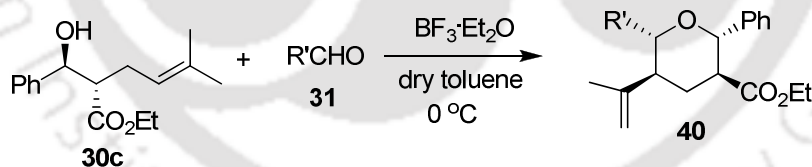
The mechanism for the formation of two diastereomers **32** and **33** from alcohols **30a** and **30b** can be explained as follows (Scheme 5.3.3). Here, Lewis acid activates the aldehyde **31** for the nucleophilic attack by alcohols **30a** and **30b** to give acetals **34** and **35**, which after decomposition generate oxocarbenium ions **36** and **37**. The species **36** and **37** after (3,5)-oxonium-ene reaction form carbocations **38** and **39**, which after deprotonation give **32** and **33** in equal amounts.



Scheme 5.3.3. Mechanism for the formation of tetrahydropyrans

The reaction was also performed with *anti*- and *syn*-alcohols **30c** and **30d**. The reaction of aldehyde **31** with *anti*-ethyl 2-(hydroxy(phenyl)methyl)-5-methylhex-4-enoate **30c** gave single diastereomer **40** in good yields (Table 5.3.3). The stereochemistry of the compounds was determined by X-ray crystallographic analysis (Figure 5.3.2).¹⁸

Table 5.3.3. Synthesis of 2,3,5,6-tetrasubstituted tetrahydropyrans from *anti*-alcohol **30c**



Entry	Aldehyde R' =	Time (min)	Product	Yield ^a (%)
1	C ₆ H ₅ (a)	60	40a	80
2	<i>p</i> -Me-C ₆ H ₄ (b)	60	40b	70
3	<i>p</i> -MeO-C ₆ H ₄ (c)	60	40c	74
4	<i>p</i> -F-C ₆ H ₄ (d)	60	40d	60
5	<i>p</i> -NO ₂ -C ₆ H ₄ (e)	60	40e	60
6	<i>p</i> -MeO ₂ C-C ₆ H ₄ (f)	60	40f	65
7	C ₂ H ₅ (g)	30	40g	92
8	(CH ₃) ₂ CHCH ₂ (h)	30	40h	87

^aYields refer to isolated yield. The compounds are characterized by IR, ¹H, ¹³C, ¹⁹F NMR and mass spectroscopy.

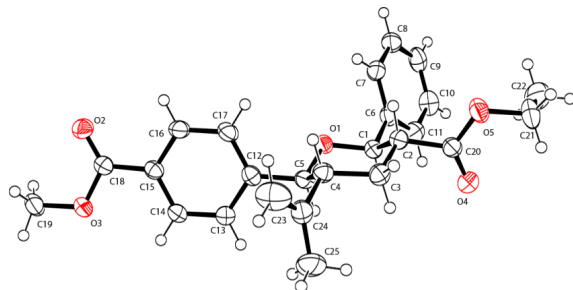
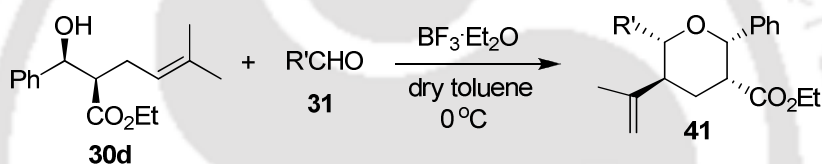


Figure 5.3.2. ORTEP diagram of (2*S**,3*S**,5*S**,6*R**)-ethyl 6-(4-(methoxycarbonyl)phenyl)-tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (**40f**).

Similarly, the reaction of *syn*-ethyl 2-(hydroxy(phenyl)methyl)-5-methylhex-4-enoate **30d** gave single diastereomer **41** in good yields (Table 5.3.4). The stereochemistry of the compounds was determined by NOE experiment (Figure 5.3.3). In the ¹H NMR spectrum of

Table 5.3.4. Synthesis of 2,3,5,6-tetrasubstituted tetrahydropyrans from *syn*-alcohol **30d**



Entry	Aldehyde R' =	Time (min)	Product	Yield ^a (%)
1	<i>p</i> -Me-C ₆ H ₄ (a)	60	41a	72
2	<i>p</i> -Br-C ₆ H ₄ (b)	60	41b	58
3	<i>p</i> -Cl-C ₆ H ₄ (c)	60	41c	55
4	<i>p</i> -MeO ₂ C-C ₆ H ₄ (d)	60	41d	62
5	C ₃ H ₇ (e)	30	41e	90
6	<i>cyclo</i> -C ₆ H ₁₁ (f)	30	41f	80

^aYields refer to isolated yield. The compounds are characterized by IR, ¹H, ¹³C NMR and mass spectroscopy.

41e the coupling constant between H_{2a} (4.69 ppm) and H_{3e} (2.94-2.98 ppm) is 3.2 Hz which indicates that one proton is in an axial position and other in an equatorial position. The

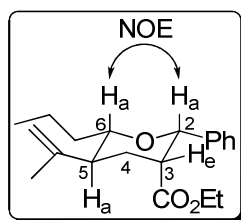


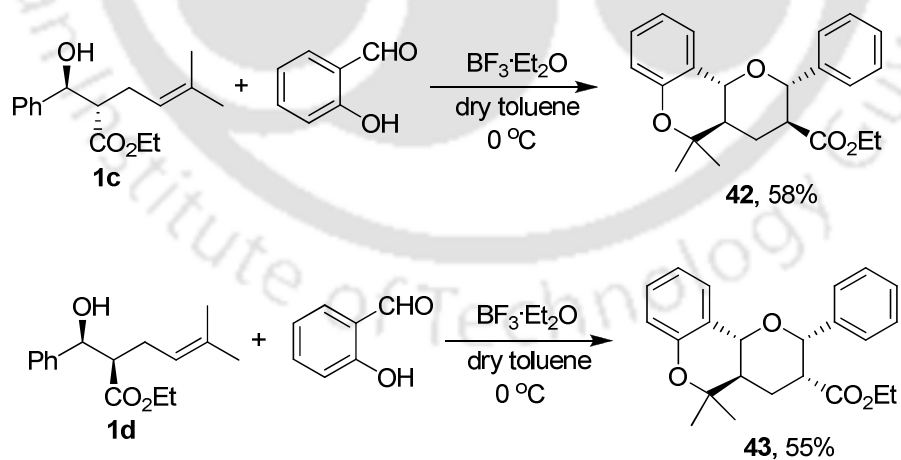
Figure 5.3.3. NOE diagram of compound **41e**

presence of a strong NOE between H_{2a} and H_{6a} (3.47 ppm) confirms that they are in a 1,3-diaxial position. Therefore H_{3e} proton is in an equatorial position. Again the coupling constant between H_{5a} (2.79 ppm) and H_{6a} proton is 10.4 Hz which indicates that they are in diaxial positions.

It was observed from Tables 5.3.1, 5.3.3 and 5.3.4 that aliphatic aldehydes gave better yields compared to the aromatic aldehydes. The aromatic aldehydes having electron-withdrawing groups gave lower yield than the aromatic aldehydes having electron-donating groups. This might be due to the better stability imparted to the oxocarbenium ions **36** and **37** (Scheme 5.3.3) by the aliphatic and aromatic aldehydes having electron donating groups on the aromatic ring, which in turn is attacked by the double bond efficiently. On the other hand aromatic aldehydes having electron withdrawing groups on the ring destabilize the oxocarbenium ions **36** and **37**.

The mechanisms for the formation of 2,3,5,6-substituted tetrahydropyrans **40** and **41** is similar to one shown in Scheme 5.3.3.

The reaction of salicylaldehyde with *anti/syn*-ethyl 2-(hydroxy(phenyl)methyl)-5-methylhex-4-enoate, **30c** and **30d** gave tricyclic chromene compounds **42** and **43** with 58% and 55% yields, respectively (Scheme 5.3.4). Here in both the cases 30% unreacted starting materials were recovered. The structure and stereochemistry of the compounds was determined by X-ray crystallographic analysis.¹⁸



Scheme 5.3.4. Reaction with salicylaldehyde

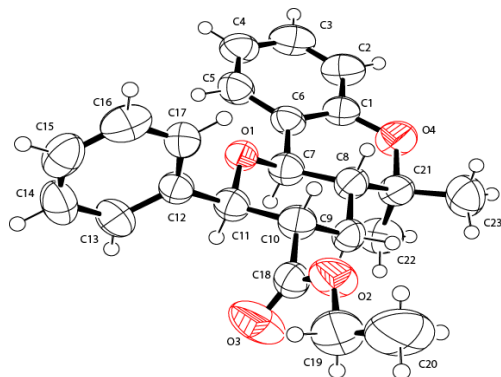


Figure 5.3.4. ORTEP diagram of (2*S**,3*S**,4*aR**,10*bR**)-ethyl 2,3,4,4*a*,5,10*b*-hexahydro-5,5-dimethyl-2-phenylpyrano[3,2-*c*]chromene-3-carboxylate (**42**)

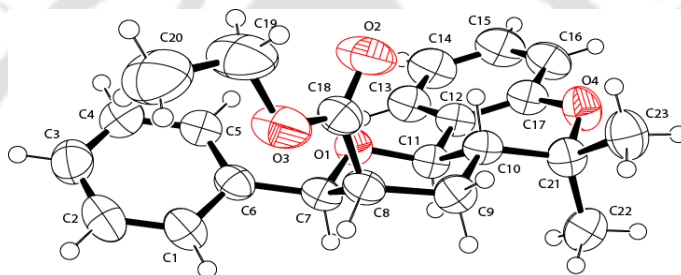
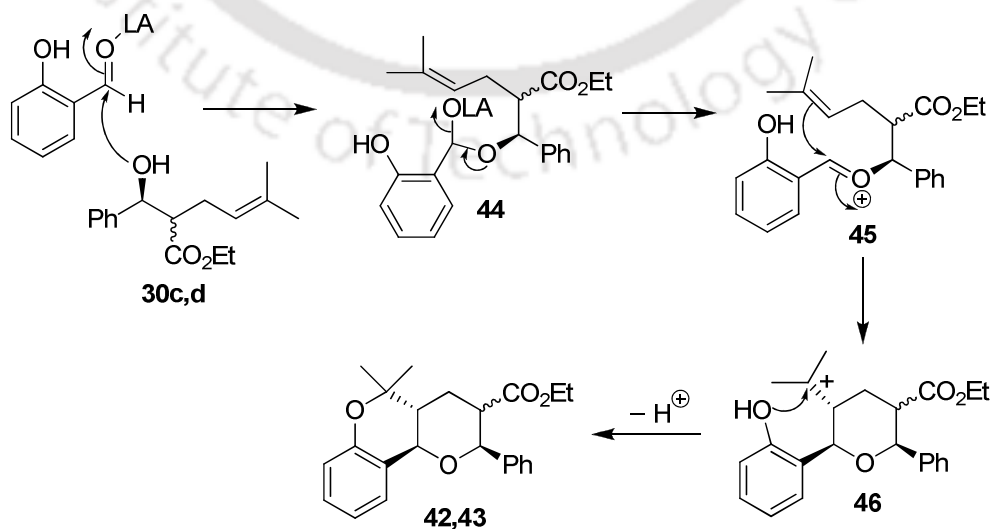


Figure 5.3.5. ORTEP diagram of (2*S**,3*R**,4*aR**,10*bR**)-ethyl 2,3,4,4*a*,5,10*b*-hexahydro-5,5-dimethyl-2-phenylpyrano[3,2-*c*]chromene-3-carboxylate (**43**)

The mechanism for the formation of chromenes is shown in Schemes 5.3.5. Here the carbocation **46** formed after oxonium-ene cyclization of **45** is attacked by nucleophilic phenolic group to form the tricyclic chromene compounds **42** and **43**. The mechanism is in accordance with the mechanism reported in the literature.^{17a,b}



Scheme 5.3.5. Mechanism for the formation of chromenes

Conclusions:

In conclusion we have developed an efficient methodology for the synthesis of 2,3,5,6-tetrasubstituted tetrahydropyrans from aldehydes and δ -unsaturated alcohols using (3,5)-oxonium-ene reaction mediated by boron trifluoride etherate in good yields under mild conditions. Synthesis of highly substituted tetrahydropyrans is of great importance in organic synthesis as they are the key structural features in many biologically active natural products and pharmaceuticals. Therefore, the present method would be beneficial to synthetic chemistry community.

5.4 Experimental Section

5.4.1 Instrumentation and Characterization

As described in Chapter 2, Section 2.4.1.

5.4.2 General Procedure for the Synthesis of α -Alkyl- β -keto esters (D-1-3):

To a suspension of sodium hydride (15.69 mmol, 1 equiv.) in THF (15 mL) at 0 °C was added β -keto ester (15.69 mmol, 1 equiv.) dropwise via syringe. After 20 min, a solution of 3,3-dimethylallyl bromide (17.26 mmol, 1.1 equiv.) in THF (5 mL) was added and the mixture was left at room temperature overnight. The solvent was removed under reduced pressure, and the residue was dissolved in Et₂O (10 mL) and washed with brine (25 mL). The organic layer was dried and filtered, and the filtrate was concentrated under reduced pressure. The residue was purified by column chromatography over silica gel to afford the title compounds as colourless oil.

Synthesis of Ethyl 2-acetyl-5-methylhex-4-enoate (D-1): To a suspension of sodium hydride (0.378 g, 15.69 mmol) in THF (15 mL) at 0 °C was added ethyl acetoacetate (2 mL, 15.69 mmol) dropwise via syringe. After 20 min, a solution of 3,3-dimethylallyl bromide (2 mL, 17.26 mmol) in THF (5 mL) was added and the mixture was left at room temperature overnight. The solvent was removed under reduced pressure, and the residue was dissolved in Et₂O (10 mL) and washed with brine (25 mL). The organic layer was dried (Na₂SO₄) and filtered, and the filtrate was concentrated under reduced pressure. The residue was purified by column chromatography over silica gel (10% EtOAc/hexane) to afford ethyl 2-acetyl-5-methylhex-4-enoate **D-1** (2.97 g, 95%) as a colourless oil.

5.4.3 General Procedure for the Synthesis of Ethyl 2-(1-hydroxyalkyl/hydroxy(phenyl)methyl)-5-methylhex-4-enoate (30a-d): To a

solution of α -alkyl- β -keto esters (14.98 mmol, 1 equiv.) in dry MeOH (15 mL) at 0 °C, was added sodium borohydride (39.25 mmol, 2.62 equiv.) in small portions. The reaction mixture was stirred in between 0 °C to 5 °C for 1.5 h. The progress of the reaction was monitored by TLC with ethyl acetate and hexane as eluents. After completion of the reaction, the product was extracted with ethyl acetate (30 mL) and then washed with water (15 mL) and brine (15 mL). The organic layer was dried (Na_2SO_4) and evaporated to leave the crude product, which was purified by column chromatography over silica gel to give the title compounds.

Synthesis of Ethyl 2-(1-hydroxyethyl)-5-methylhex-4-enoate (30a): To a solution of ethyl 2-acetyl-5-methylhex-4-enoate (2.97 g, 14.98 mmol) in dry MeOH (15 mL) at 0 °C, was added sodium borohydride (1.485 g, 39.25 mmol) in small portions. The reaction mixture was stirred in between 0 °C to 5 °C for 1.5 h. The progress of the reaction was monitored by TLC with ethyl acetate and hexane (3:22) as eluents. After completion of the reaction, the product was extracted with ethyl acetate (30 mL) and then washed with water (15 mL) and brine (15 mL). The organic layer was dried (Na_2SO_4) and evaporated to leave the crude products, which were purified by column chromatography over silica gel to give an inseparable mixture of two diastereomers **30a** (2.40 g, 80% overall yield) as a colourless oil.

5.4.4 General Procedure for the Synthesis of 2,3,5,6-Tetrasubstituted Tetrahydropyrans and Chromenes: To a solution of aldehyde (1 mmol, 1 equiv.) in dry toluene (2 mL) at 0 °C was added boron trifluoride etherate (1.1 mmol, 1.1 equiv.). To this solution ethyl 2-(1-hydroxyalkyl/hydroxy(phenyl)methyl)-5-methylhex-4-enoate (1 mmol, 1 equiv.) in dry toluene (2 mL) was added drop by drop over 5 min. The reaction mixture was stirred at that temperature for a specified time. The progress of the reaction was monitored by TLC with ethyl acetate and hexane as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO_3 . The product was extracted with ethyl acetate (30 mL) and then washed with water (15 mL) and brine (15 mL). The organic layer was dried (Na_2SO_4) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give the title compounds.

Synthesis of (2R*,3S*,5S*,6R*)/(2R*,3R*,5S*,6R*)-Ethyl tetrahydro-2-methyl-6-phenyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (32a & 33a):

To a solution of benzaldehyde (106 mg, 1.00 mmol) in dry toluene (2 mL) at 0 °C was added boron trifluoride etherate (156 mg, 1.10 mmol). To this solution ethyl 2-(1-

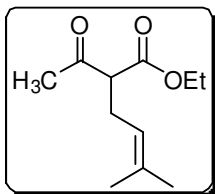
hydroxyethyl)-5-methylhex-4-enoate (200 mg, 1 mmol) in dry toluene (2 mL) was added drop by drop over 5 min. The reaction mixture was stirred at that temperature for another 55 min. The progress of the reaction was monitored by TLC with ethyl acetate and hexane (1:24) as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with ethyl acetate (30 mL) and then washed with water (15 mL) and brine (15 mL). The organic layer was dried (Na₂SO₄) and evaporated to leave the crude products, which were separated by column chromatography over silica gel to give **32a** (115 mg, 40%) and **33a** (110 mg, 38%) as colourless oil.

5.5 References and Notes

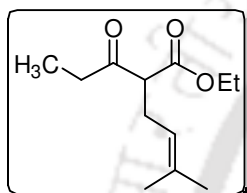
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18. The crystallographic data for compounds **33g**, **40f**, **42** and **43** have been deposited with the Cambridge Crystallographic Data Centre as Supplementary publication numbers CCDC 819363, 819365, 819364 and 819366.

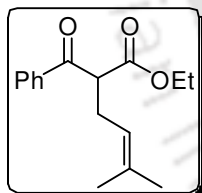
5.6 Spectral Data

Ethyl 2-acetyl-5-methylhex-4-enoate (D-1):

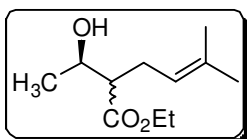
Colourless oil (2.97 g, 95%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.27 (t, J = 7.2 Hz, 3 H), 1.63 (s, 3 H), 1.68 (s, 3 H), 2.22 (s, 3 H), 2.54 (t, J = 7.2 Hz, 2 H), 3.43 (t, J = 7.6 Hz, 1 H), 4.19 (q, J = 7.2 Hz, 2 H), 5.03 (dt, J = 7.6 and 1.6 Hz, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 17.8, 25.8, 27.0, 29.1, 59.8, 61.3, 119.9, 134.7, 169.7, 203.2; **IR** (Neat): 2981, 2930, 1739, 1718, 1205, 1150 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{11}\text{H}_{18}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 199.1334; found 199.1337.

Ethyl 5-methyl-2-propionylhex-4-enoate (D-2):

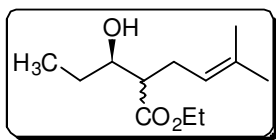
Colourless oil (3.13 g, 94%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.06 (t, J = 7.2 Hz, 3 H), 1.26 (t, J = 7.2, 3 H), 1.62 (s, 3 H), 1.67 (s, 3 H), 2.47-2.60 (m, 4 H), 3.45 (t, J = 7.2 Hz, 1 H), 4.17 (q, J = 7.2 Hz, 2 H), 5.02 (dt, J = 7.6 and 1.6 Hz, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 7.8, 14.2, 17.9, 25.9, 27.2, 35.6, 58.9, 61.4, 120.1, 134.8, 169.9, 206.0; **IR** (Neat): 2980, 2937, 1741, 1716, 1198, 1156 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{12}\text{H}_{20}\text{O}_3$ (M^+) requires 212.1412; found 212.1408.

Ethyl 2-benzoyl-5-methylhex-4-enoate (D-3):

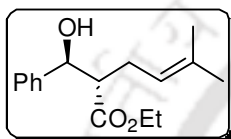
Colourless oil (3.88 g, 95%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.67 (t, J = 7.2 Hz, 3 H), 1.62 (s, 3 H), 1.65 (s, 3 H), 2.61-2.78 (m, 2 H), 4.14 (q, J = 7.2 Hz, 2 H), 4.30 (t, J = 7.2 Hz, 1 H), 5.11 (dt, J = 7.2 and 1.6 Hz, 1 H), 7.44-7.50 (m, 2 H), 7.55-7.61 (m, 1 H), 7.97-8.00 (m, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 17.9, 25.8, 27.8, 54.6, 61.4, 120.3, 128.7, 128.8, 133.5, 134.7, 136.4, 169.9, 195.2; **IR** (Neat): 2979, 2929, 1737, 1688, 1448, 1379, 1241, 1153 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{16}\text{H}_{20}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 261.1490; found 261.1494.

Ethyl 2-(1-hydroxyethyl)-5-methylhex-4-enoate (30a):

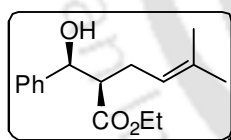
Colourless oil (2.40 g, 80% overall yield); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.19-1.29 (m, 6 H), 1.62 (s, 3 H), 1.69 (s, 3 H), 2.29-2.47 (m, 3 H), 2.51 (brs 0.5 H), 2.69 (brs, 0.5 H), 3.86-3.96 (m, 0.5 H), 3.98-4.06 (m, 0.5 H), 4.08-4.24 (m, 2 H), 5.04-5.14 (m, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1(2C), 17.5(2C), 20.5, 21.1, 25.6(2C), 26.7, 27.7, 53.0, 53.1, 60.2, 60.3, 67.8, 67.9, 120.5, 121.1, 133.3, 133.6, 174.7, 175.0; **IR** (Neat): 3441, 2972, 2929, 1732, 1640, 1182, 1155 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{11}\text{H}_{20}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 201.1490; found 201.1498.

Ethyl 2-(1-hydroxypropyl)-5-methylhex-4-enoate (30b):

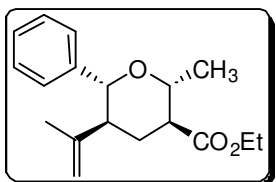
Colourless oil (2.50 g, 78%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.98 (t, $J = 7.6$ Hz, 3 H), 1.25 (t, $J = 7.2$ Hz, 1.5 H), 1.26 (t, $J = 7.2$ Hz, 1.5 H), 1.45-1.54 (m, 2 H), 1.61 (s, 1.5 H), 1.62 (s, 1.5 H), 1.69 (s, 3 H), 2.28-2.50 (m, 3 H), 2.67 (brs, 1 H), 3.56-3.64 (m, 0.5 H), 3.70-3.78 (m, 0.5 H), 4.08-4.22 (m, 2 H), 5.04-5.16 (m, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 10.2, 10.3, 14.3(2C), 17.8(2C), 25.8, 26.1, 27.4(2C), 28.3, 28.6, 50.7, 51.2, 60.5(2C), 73.3, 73.4, 120.7, 121.3, 133.6, 134.1, 175.3, 175.5; **IR** (Neat): 3456, 2968, 2932, 1731, 1642, 1183, 1159 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{12}\text{H}_{22}\text{O}_3$ ($\text{M}+\text{H}^+$) requires 215.1647; found 215.1653.

anti-Ethyl 2-(hydroxy(phenyl)methyl)-5-methylhex-4-enoate (30c):

Colourless oil (1.73 g, 44%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.18 (t, $J = 7.2$ Hz, 3 H), 1.50 (s, 3 H), 1.65 (s, 3 H), 2.05 (ddd, $J = 13.2, 7.6$ and 5.6 Hz, 1 H), 2.25 (ddd, $J = 14.8, 8.0$ and 6.8 Hz, 1 H), 2.75 (ddd, $J = 12.8, 8.8, 5.2$ Hz, 1 H), 3.16 (d, $J = 5.2$ Hz, 1 H), 4.11 (q, $J = 7.2$ Hz, 2 H), 4.80 (dd, $J = 6.4$ and 4.4 Hz, 1 H), 5.03 (dt, $J = 6.8$ and 1.2 Hz, 1 H), 7.26-7.37 (m, 5 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.3, 17.8, 25.9, 28.4, 53.3, 60.7, 74.8, 120.2, 126.6, 128.0, 128.6, 134.3, 142.2, 175.2; **IR** (Neat): 3461, 2978, 2929, 1729, 1452, 1377, 1180, 1037, 766, 702 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{16}\text{H}_{22}\text{O}_3$ (M^+) requires 262.1569; found 262.1573.

syn-Ethyl 2-(hydroxy(phenyl)methyl)-5-methylhex-4-enoate (30d):

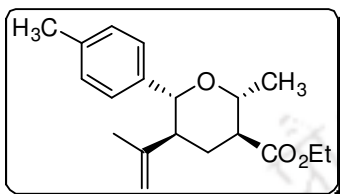
Colourless oil (1.58 g, 40%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.11 (t, $J = 7.2$ Hz, 3 H), 1.54 (s, 3 H), 1.64 (s, 3 H), 2.29 (ddd, $J = 14.4, 8.8$ and 5.8 Hz, 1 H), 2.46 (ddd, $J = 14.4, 8.8$ and 5.6 Hz, 1 H), 2.72 (ddd, $J = 10.4, 5.6$ and 4.4 Hz, 1 H), 3.03 (brs, 1H), 4.02 (q, $J = 7.2$ Hz, 2 H), 4.95 (d, $J = 5.6$ Hz, 1 H), 5.04 (t, $J = 6.8$ Hz, 1 H), 7.24-7.39 (m, 5 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.2, 17.8, 25.9, 26.1, 53.3, 60.7, 74.2, 121.1, 126.4, 127.8, 128.4, 134.0, 141.7, 175.0; **IR** (Neat): 3460, 2979, 2930, 1728, 1453, 1375, 1180, 1026, 767, 701 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{16}\text{H}_{22}\text{O}_3$ ($\text{M}+\text{H}^+$) requires 263.1647; found 263.1652.

(2R*,3S*,5S*,6R*)-Ethyl tetrahydro-2-methyl-6-phenyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (32a):

Colourless oil (115 mg, 40%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.27 (d, $J = 6.4$ Hz, 3 H), 1.28 (t, $J = 7.2$ Hz, 3 H), 1.43 (s, 3 H), 1.99 (ddd, $J = 12.8, 12.4$ and 12.4 Hz, 1 H), 2.09 (ddd, $J = 12.8, 4.0$ and 4.0 Hz, 1 H), 2.33 (ddd, $J = 12.4, 10.4$ and 4.0 Hz, 1 H), 2.46 (ddd, $J = 12.4, 10.0$ and 4.0 Hz, 1 H), 3.76-3.85 (m, 1 H), 4.17 (q, $J = 7.2$ Hz, 2 H), 4.29 (d, $J =$

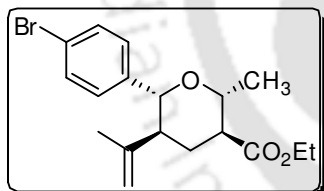
10.0 Hz, 1 H), 4.62 (brs, 1 H), 4.66 (brs, 1 H), 7.22-7.32(m, 5 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.4, 20.3, 21.5, 34.0, 49.5, 49.8, 60.6, 75.1, 84.1, 113.0, 127.5, 128.0, 128.3, 140.7, 145.1, 173.9; **IR** (Neat): 2978, 2933, 1730, 1644, 1452, 1375, 1175, 1102, 1064, 755, 699 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{24}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 289.1803; found 289.1803.

(2*R,3*S**,5*S**,6*R**)-Ethyl tetrahydro-2-methyl-5-(prop-1-en-2-yl)-6-*p*-tolyl-2*H*-pyran-3-carboxylate (32b):**



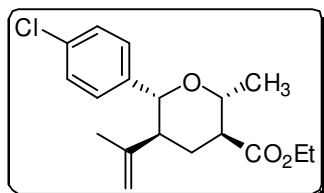
Colourless oil (115 mg, 38%); ^1H NMR (400 MHz, CDCl_3): δ 1.24 (d, $J = 6.4$ Hz, 3 H), 1.28 (t, $J = 7.2$ Hz, 3 H), 1.45 (s, 3 H), 1.97 (ddd, $J = 12.8, 12.6$ and 12.4 Hz, 1 H), 2.10 (ddd, $J = 12.8, 4.0$ and 3.6 Hz, 1 H), 2.30 (s, 3 H), 2.34 (ddd, $J = 12.6, 12.4$ and 3.6 Hz, 1 H), 2.44 (ddd, $J = 12.4, 10.4$ and 4.0 Hz, 1 H), 3.75-3.84 (m, 1 H), 4.16 (q, $J = 7.2$ Hz, 2 H), 4.26 (d, $J = 10.4$ Hz, 1 H), 4.63 (brs, 1 H), 4.65 (brs, 1 H), 7.09 (d, $J = 7.6$ Hz, 2 H), 7.17 (d, $J = 8.0$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.4, 20.3, 21.3, 21.5, 34.1, 49.3, 49.8, 60.6, 75.1, 83.8, 112.9, 127.5, 129.0, 137.6, 137.7, 145.3, 174.0; **IR** (Neat): 2977, 2929, 2857, 1731, 1375, 1101, 1065, 812 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{19}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 303.1960; found 303.1952.

(2*R,3*S**,5*S**,6*R**)-Ethyl 6-(4-bromophenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (32c):**



Colourless oil (132 mg, 36%); ^1H NMR (400 MHz, CDCl_3): δ 1.24 (d, $J = 6.0$ Hz, 3 H), 1.28 (t, $J = 7.2$ Hz, 3 H), 1.45 (s, 3 H), 1.98 (ddd, $J = 12.8, 12.4$ and 12.4 Hz, 1 H), 2.09 (ddd, $J = 12.8, 4.0$ and 3.6 Hz, 1 H), 2.25 (ddd, $J = 12.8, 11.6$ and 3.6 Hz, 1 H), 2.44 (ddd, $J = 12.4, 10.0$ and 4.0 Hz, 1 H), 3.75-3.83 (m, 1 H), 4.17 (q, $J = 7.2$ Hz, 2 H), 4.26 (d, $J = 10.0$ Hz, 1 H), 4.62 (brs, 1 H), 4.68 (brs, 1 H), 7.17 (d, $J = 8.4$ Hz, 2 H), 7.42 (d, $J = 8.4$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.4, 20.3, 21.6, 33.9, 49.7 (2C), 60.7, 75.2, 83.3, 113.4, 121.8, 129.2, 131.4, 139.9, 144.8, 173.9; **IR** (Neat): 2977, 2931, 2866, 1729, 1644, 1375, 1175, 1102, 1069, 894, 810 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{23}\text{BrO}_3$ ($\text{M}+\text{H}$) $^+$ requires 367.0909; found 367.0905 (^{79}Br).

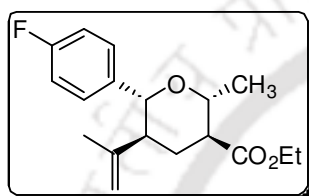
(2*R,3*S**,5*S**,6*R**)-Ethyl 6-(4-chlorophenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (32d):**



Colourless oil (106 mg, 33%); ^1H NMR (400 MHz, CDCl_3): δ 1.25 (d, $J = 6.0$ Hz, 3 H), 1.28 (t, $J = 7.2$ Hz, 3 H), 1.45 (s, 3 H), 1.97 (ddd, $J = 12.8, 12.8$ and 12.4 Hz, 1 H), 2.09 (ddd, $J =$

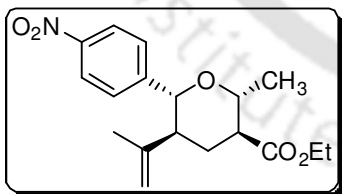
12.8, 4.0 and 3.6 Hz, 1 H), 2.25 (ddd, $J = 12.8, 12.8$ and 3.6 Hz, 1 H), 2.44 (ddd, $J = 12.4, 10.0$ and 4.0 Hz, 1 H), 3.76-3.84 (m, 1 H), 4.17 (q, $J = 7.2$ Hz, 2 H), 4.27 (d, $J = 10.0$ Hz, 1 H), 4.62 (brs, 1 H), 4.68 (brs, 1 H), 7.23 (d, $J = 8.4$ Hz, 2 H), 7.26 (d, $J = 8.8$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.4, 20.3, 21.6, 33.9, 49.7 (2C), 60.7, 75.2, 83.2, 113.4, 128.5, 128.8, 133.6, 139.3, 144.7, 173.8; IR (Neat): 2978, 2932, 2866, 1729, 1645, 1449, 1376, 1175, 1099, 1067, 894, 824 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{18}\text{H}_{23}\text{ClO}_3$ ($\text{M}+\text{H}$) $^+$ requires 323.1414; found 323.1407 (^{35}Cl).

(2R*,3S*,5S*,6R*)-Ethyl 6-(4-fluorophenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (32e):



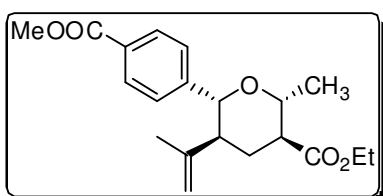
Colourless oil (89 mg, 29%); ^1H NMR (400 MHz, CDCl_3): δ 1.25 (d, $J = 6.0$ Hz, 3 H), 1.28 (t, $J = 7.2$ Hz, 3 H), 1.44 (s, 3 H), 1.97 (ddd, $J = 14.0, 12.8$ and 12.0 Hz, 1 H), 2.09 (ddd, $J = 14.0, 4.0$ and 4.0 Hz, 1 H), 2.27 (ddd, $J = 12.8, 10.4$ and 4.0 Hz, 1 H), 2.44 (ddd, $J = 12.0, 10.0$ and 4.0 Hz, 1 H), 3.76-3.84 (m, 1 H), 4.17 (q, $J = 7.2$ Hz, 2 H), 4.28 (d, $J = 10.0$ Hz, 1 H), 4.62 (brs, 1 H), 4.67 (brs, 1 H), 6.98 (t, $J = 8.4$ Hz, 2 H), 7.24-7.28 (m, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.4, 20.3, 21.5, 33.9, 49.7, 49.8, 60.7, 75.2, 83.2, 113.2, 115.0 (d, $J = 21.3$ Hz), 129.1 (d, $J = 8.4$ Hz), 136.7, 144.9, 162.5 (d, $J = 244.1$ Hz), 173.8; ^{19}F NMR (376 MHz, $\text{CDCl}_3/\text{C}_6\text{F}_6$): δ -218.70-(-218.62) (m, 1F); IR (Neat): 2979, 2934, 2867, 1729, 1644, 1450, 1376, 1176, 1102, 1065, 832, 788 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{18}\text{H}_{23}\text{FO}_3$ ($\text{M}+\text{H}$) $^+$ requires 307.1709; found 307.1715.

(2R*,3S*,5S*,6R*)-Ethyl tetrahydro-2-methyl-6-(4-nitrophenyl)-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (32f):



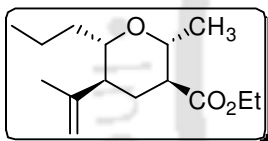
Colourless oil (117 mg, 35%); ^1H NMR (400 MHz, CDCl_3): δ 1.27 (d, $J = 6.0$ Hz, 3 H), 1.29 (t, $J = 7.2$ Hz, 3 H), 1.47 (s, 3 H), 1.99 (ddd, $J = 13.2, 12.5$ and 12.4 Hz, 1 H), 2.12 (ddd, $J = 13.2, 3.6$ and 3.6 Hz, 1 H), 2.24 (ddd, $J = 12.8, 12.5$ and 3.6 Hz, 1 H), 2.47 (ddd, $J = 12.4, 10.0$ and 3.6 Hz, 1 H), 3.78-3.87 (m, 1 H), 4.18 (q, $J = 7.2$ Hz, 2 H), 4.41 (d, $J = 10.0$ Hz, 1 H), 4.61 (brs, 1 H), 4.70 (brs, 1 H), 7.46 (d, $J = 8.8$ Hz, 2 H), 8.16 (d, $J = 8.8$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.4, 20.2, 21.6, 33.7, 49.5, 50.2, 60.9, 75.3, 82.8, 114.0, 123.5, 128.3, 144.1, 147.7, 148.2, 173.6; IR (Neat): 2975, 2935, 2871, 1726, 1643, 1519, 1372, 1344, 1170, 1101, 1060, 849 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{18}\text{H}_{23}\text{NO}_5$ ($\text{M}+\text{H}$) $^+$ requires 334.1654; found 334.1652.

(2R*,3S*,5S*,6R*)-Ethyl 6-(4-(methoxycarbonyl)phenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (32g):



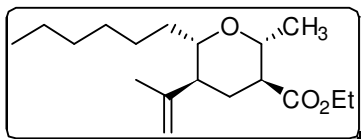
Colourless oil (125 mg, 36%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.27 (d, $J = 6.0$ Hz, 3 H), 1.29 (t, $J = 7.2$ Hz, 3 H), 1.43 (s, 3 H), 1.99 (ddd, $J = 12.8, 12.8$ and 12.4 Hz, 1 H), 2.10 (ddd, $J = 12.8, 4.0$ and 3.6 Hz, 1 H), 2.28 (ddd, $J = 12.8, 12.0$ and 4.0 Hz, 1 H), 2.47 (ddd, $J = 12.4, 10.0$ and 3.6 Hz, 1 H), 3.78-3.86 (m, 1 H), 3.90 (s, 3 H), 4.17 (q, $J = 7.2$ Hz, 2 H), 4.35 (d, $J = 10.0$ Hz, 1 H), 4.60 (brs, 1 H), 4.66 (brs, 1 H), 7.36 (d, $J = 8.4$ Hz, 2 H), 7.97 (d, $J = 8.4$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.3, 20.1, 21.5, 33.7, 49.5, 49.8, 52.1, 60.6, 75.0, 83.4, 113.3, 127.4, 129.5, 130.2, 144.4, 145.8, 167.0, 173.6; **IR** (Neat): 2979, 2936, 2867, 1725, 1645, 1437, 1376, 1279, 1102, 1066, 896, 772, 706 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{20}\text{H}_{26}\text{O}_5$ ($\text{M}+\text{H}$) $^+$ requires 347.1858; found 347.1865.

(2R*,3S*,5S*,6S*)-Ethyl tetrahydro-2-methyl-5-(prop-1-en-2-yl)-6-propyl-2H-pyran-3-carboxylate (32h):



Colourless oil (115 mg, 45%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.88 (t, $J = 7.2$ Hz, 3 H), 1.19 (d, $J = 6.4$ Hz, 3 H), 1.26 (t, $J = 7.2$ Hz, 3 H), 1.27-1.35 (m, 2 H), 1.42-1.56 (m, 2 H), 1.67 (s, 3 H), 1.80 (ddd, $J = 12.8, 12.4$ and 12.0 Hz, 1 H), 1.91 (ddd, $J = 12.8, 4.0$ and 3.6 Hz, 1 H), 1.99 (ddd, $J = 12.4, 10.0$ and 3.6 Hz, 1 H), 2.27 (ddd, $J = 12.0, 10.0$ and 4.0 Hz, 1 H), 3.31 (dt, $J = 9.2$ and 7.2 Hz, 1 H), 3.54-3.63 (m, 1 H), 4.13 (q, $J = 7.2$ Hz, 2 H), 4.76 (brs, 1 H), 4.77 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 14.3, 18.8, 20.2(2C), 33.8, 35.5, 48.8, 49.8, 60.5, 74.5, 79.5, 112.7, 146.1, 174.1; **IR** (Neat): 2960, 2937, 2872, 1732, 1644, 1452, 1376, 1150, 1113, 1037, 893, 750 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{15}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 255.1960; found 255.1964.

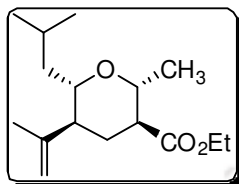
(2R*,3S*,5S*,6S*)-Ethyl 6-hexyl-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (32i):



Colourless oil (127 mg, 43%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.87 (t, $J = 6.8$ Hz, 3 H), 1.20 (d, $J = 6.0$ Hz, 3 H), 1.24-1.34 (m, 8 H), 1.26 (t, $J = 7.2$ Hz, 3 H), 1.47-1.52 (m, 2 H), 1.67 (s, 3 H), 1.80 (ddd, $J = 12.8, 12.4$ and 12.0 Hz, 1 H), 1.91 (ddd, $J = 12.8, 4.0$ and 3.6 Hz, 1 H), 1.97 (ddd, $J = 12.4, 10.4$ and 3.6 Hz, 1 H), 2.28 (ddd, $J = 12.0, 10.4$ and 4.0 Hz, 1 H), 3.29 (dt, $J = 8.8$ and 8.4 Hz, 1 H), 3.55-3.64 (m, 1 H), 4.13 (q, $J = 7.2$ Hz, 2 H), 4.75 (brs, 1 H), 4.77 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.3, 14.4, 20.2(2C),

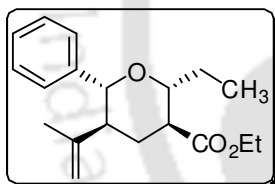
22.9, 25.7, 29.5, 32.1, 33.5, 33.9, 48.8, 49.8, 60.6, 74.6, 79.9, 112.8, 146.1, 174.2; **IR** (Neat): 2929, 2858, 1732, 1643, 1455, 1376, 1174, 1115, 1035, 892 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{32}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 297.2429; found 297.2424.

(2R*,3S*,5S*,6S*)-Ethyl tetrahydro-6-isobutyl-2-methyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (32j):



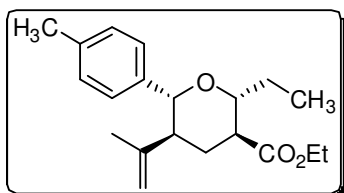
Colourless oil (110 mg, 41%); **¹H NMR** (400 MHz, CDCl_3): δ 0.85 (d, $J = 6.8$ Hz, 3 H), 0.89 (d, $J = 6.8$ Hz, 3 H), 1.19 (d, $J = 6.0$ Hz, 3 H), 1.24-1.30 (m, 2 H), 1.26 (t, $J = 7.2$ Hz, 3 H), 1.66 (s, 3 H), 1.76-1.99 (m, 4 H), 2.28 (ddd, $J = 10.0, 10.0$ and 4.0 Hz, 1 H), 3.37 (dt, $J = 10.0$ and 7.6 Hz, 1 H), 3.54-3.62 (m, 1 H), 4.12 (q, $J = 7.2$ Hz, 2 H), 4.75 (brs, 1 H), 4.77 (brs, 1 H); **¹³C NMR** (100 MHz, CDCl_3): δ 14.4, 20.1, 20.2, 21.6, 24.0, 24.2, 33.9, 42.4, 49.3, 49.8, 60.5, 74.6, 77.9, 112.8, 146.1, 174.1; **IR** (Neat): 2955, 2928, 2870, 1732, 1644, 1451, 1374, 1176, 1115, 1068, 1035, 892 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{16}\text{H}_{28}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 269.2116; found 269.2122.

(2R*,3S*,5S*,6R*)-Ethyl 2-ethyl-tetrahydro-6-phenyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (32k):



Colourless oil (118 mg, 39%); **¹H NMR** (400 MHz, CDCl_3): δ 0.96 (t, $J = 7.2$ Hz, 3 H), 1.28 (t, $J = 7.2$ Hz, 3 H), 1.43 (s, 3 H), 1.46-1.57 (m, 1 H), 1.58-1.68 (m, 1 H), 2.01 (ddd, $J = 12.8, 12.4$ and 12.0 Hz, 1 H), 2.08 (ddd, $J = 12.8, 4.4$ and 4.0 Hz, 1 H), 2.28 (ddd, $J = 12.4, 10.0$ and 4.0 Hz, 1 H), 2.56 (ddd, $J = 12.0, 10.0$ and 4.4 Hz, 1 H), 3.64 (ddd, $J = 10.0, 7.2, 3.2$ Hz, 1 H), 4.16 (q, $J = 7.2$ Hz, 2 H), 4.28 (d, $J = 10.0$ Hz, 1 H), 4.62 (brs, 1 H), 4.66 (brs, 1 H), 7.24-7.30 (m, 5 H); **¹³C NMR** (100 MHz, CDCl_3): δ 9.7, 14.4, 21.6, 27.2, 34.2, 47.5, 49.8, 60.6, 79.7, 83.9, 113.0, 127.5, 127.8, 128.2, 141.1, 145.3, 174.2; **IR** (Neat): 2963, 2929, 2855, 1724, 1644, 1447, 1379, 1177, 1102, 1061, 1027, 747, 697 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{19}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 303.1960; found 303.1962.

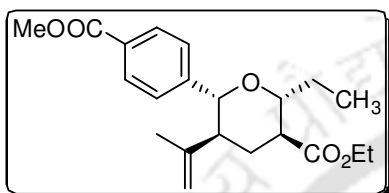
(2R*,3S*,5S*,6R*)-Ethyl 2-ethyl-tetrahydro-5-(prop-1-en-2-yl)-6-p-tolyl-2H-pyran-3-carboxylate (32l):



Colourless oil (108 mg, 34%); **¹H NMR** (400 MHz, CDCl_3): δ 0.94 (t, $J = 7.2$ Hz, 3 H), 1.28 (t, $J = 7.2$ Hz, 3 H), 1.45 (s, 3 H), 1.47-1.55 (m, 1 H), 1.56-1.65 (m, 1 H), 2.00 (ddd, $J = 12.8, 12.0$ and 12.0 Hz, 1 H), 2.07 (ddd, $J = 12.8, 4.0$ and 4.0 Hz, 1 H), 2.31 (s, 3 H), 2.32 (ddd, $J = 12.0, 10.4$ and 4.0 Hz, 1 H), 2.54 (ddd, $J = 12.0, 10.0$ and 4.0 Hz, 1 H), 3.62 (ddd, $J = 10.4, 7.2$ and 3.2 Hz, 1 H), 4.15 (q, $J = 7.2$ Hz, 2 H), 4.25

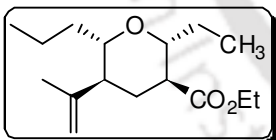
(d, $J = 10.0$ Hz, 1 H), 4.63 (brs, 1 H), 4.66 (brs, 1 H), 7.09 (d, $J = 7.6$ Hz, 2 H), 7.18 (d, $J = 8.0$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 9.7, 14.4, 21.4, 21.5, 27.1, 34.3, 47.5, 49.5, 60.6, 79.7, 83.6, 112.8, 127.4, 128.9, 137.4, 138.1, 145.5, 174.2; IR (Neat): 2965, 2935, 2862, 1730, 1645, 1451, 1375, 1173, 1108, 1067, 967, 891, 815, 768 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{20}\text{H}_{28}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 317.2116; found 317.2128.

(2*R,3*S**,5*S**,6*R**)-Ethyl 6-(4-(methoxycarbonyl)phenyl)-2-ethyl-tetrahydro-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (32m):**



Colourless oil (133 mg, 37%); ^1H NMR (400 MHz, CDCl_3): δ 0.96 (t, $J = 7.2$ Hz, 3 H), 1.28 (t, $J = 7.2$ Hz, 3 H), 1.43 (s, 3 H), 1.48-1.58 (m, 1 H), 1.60-1.69 (m, 1 H), 2.02 (ddd, $J = 13.2, 12.8$ and 12.0 Hz, 1 H), 2.09 (ddd, $J = 13.2, 4.0$ and 4.0 Hz, 1 H), 2.24 (ddd, $J = 12.8, 10.4$ and 4.0 Hz, 1 H), 2.57 (ddd, $J = 12.0, 10.0$ and 4.0 Hz, 1 H), 3.65 (ddd, $J = 10.4, 7.2$ and 2.8 Hz, 1 H), 3.90 (s, 3 H), 4.16 (q, $J = 7.2$ Hz, 2 H), 4.33 (d, $J = 10.0$ Hz, 1 H), 4.60 (brs, 1 H), 4.67 (brs, 1 H), 7.36 (d, $J = 8.4$ Hz, 2 H), 7.97 (d, $J = 8.8$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 9.6, 14.4, 21.7, 27.1, 34.0, 47.4, 50.0, 52.2, 60.7, 76.7, 83.3, 113.4, 127.4, 129.5(2C), 144.7, 146.2, 167.2, 174.0; IR (Neat): 2935, 2857, 1726, 1645, 1437, 1375, 1278, 1176, 1108, 1069, 895, 760 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{21}\text{H}_{28}\text{O}_5$ ($\text{M}+\text{H}$)⁺ requires 361.2015; found 361.2023.

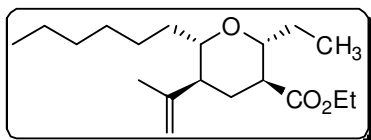
(2*R,3*S**,5*S**,6*S**)-Ethyl 2-ethyl-tetrahydro-5-(prop-1-en-2-yl)-6-propyl-2*H*-pyran-3-carboxylate (32n):**



Colourless oil (113 mg, 42%); ^1H NMR (400 MHz, CDCl_3): δ 0.89 (t, $J = 6.8$ Hz, 3 H), 0.98 (t, $J = 7.2$ Hz, 3 H), 1.25 (t, $J = 7.2$ Hz, 3 H), 1.28-1.58 (m, 6 H), 1.67 (s, 3 H), 1.82 (ddd, $J = 12.8, 12.4$ and 12.0 Hz, 1 H), 1.92 (ddd, $J = 12.8, 4.0$ and 3.6 Hz, 1 H), 1.98 (ddd, $J = 12.4, 10.0$ and 3.6 Hz, 1 H), 2.34 (ddd, $J = 12.0, 10.0$ and 4.0 Hz, 1 H), 3.28 (dt, $J = 10.0$ and 7.2 Hz, 1 H), 3.36 (dt, $J = 10.0$ and 6.8 Hz, 1 H), 4.12 (q, $J = 7.2$ Hz, 2 H), 4.75 (brs, 1 H), 4.77 (brs, 1 H); ^{13}C NMR (100 MHz, CDCl_3): δ 10.2, 14.1, 14.4, 19.0, 20.2, 27.5, 34.0, 35.6, 48.3, 49.1, 60.5, 79.4, 79.6, 112.7, 146.1, 174.3; IR (Neat): 2960, 2937, 2873, 1732, 1644, 1454, 1375, 1175, 1116, 1038, 1011, 893, 859 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{16}\text{H}_{28}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 269.2116; found 269.2108.

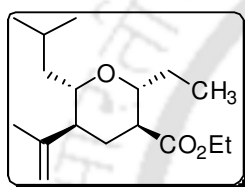
(2*R,3*S**,5*S**,6*S**)-Ethyl 2-ethyl-6-hexyl-tetrahydro-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (32o):**

Colourless oil (127 mg, 41%); ^1H NMR (400 MHz, CDCl_3): δ 0.87 (t, $J = 6.4$ Hz, 3 H), 0.98 (t, $J = 7.2$ Hz, 3 H), 1.25 (t, $J = 7.2$ Hz, 3 H), 1.25-1.35 (m, 9 H), 1.37-1.46 (m, 1 H),



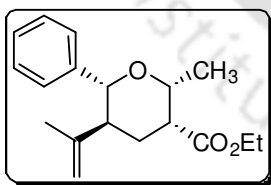
1.47-1.57 (m, 2 H), 1.66 (s, 3 H), 1.82 (ddd, $J = 12.4, 12.4$ and 12.4 Hz, 1 H), 1.92 (ddd, $J = 12.4, 4.0$ and 3.6 Hz, 1 H), 1.98 (ddd, $J = 12.4, 10.0$ and 3.6 Hz, 1 H), 2.33 (ddd, $J = 12.4, 10.0$ and 4.0 Hz, 1 H), 3.27 (dt, $J = 10.0$ and 8.8 Hz, 1 H), 3.35 (dt, $J = 10.0$ and 6.8 Hz, 1 H), 4.12 (q, $J = 7.2$ Hz, 2 H), 4.75 (brs, 1 H), 4.77 (brs, 1 H); ^{13}C NMR (100 MHz, CDCl_3): δ 10.2, 14.3, 14.4, 20.2, 22.8, 25.7, 27.4, 29.3, 32.1, 33.5, 34.0, 48.3, 49.1, 60.5, 79.6, 79.7, 112.7, 146.1, 174.3; IR (Neat): 2930, 2857, 1732, 1640, 1454, 1375, 1173, 1117, 1038, 892 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{19}\text{H}_{34}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 311.2586; found 311.2587.

(2*R,3*S**,5*S**,6*S**)-Ethyl 2-ethyl-tetrahydro-6-isobutyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (32p):**



Colourless oil (124 mg, 44%); ^1H NMR (400 MHz, CDCl_3): δ 0.86 (d, $J = 6.8$ Hz, 3 H), 0.90 (d, $J = 6.8$ Hz, 3 H), 0.97 (t, $J = 7.6$ Hz, 3 H), 1.17-1.58 (m, 4 H), 1.25 (t, $J = 7.2$ Hz, 3 H), 1.66 (s, 3 H), 1.78-2.00 (m, 4 H), 2.33 (ddd, $J = 12.0, 10.0$ and 4.0 Hz, 1 H), 3.32-3.38 (m, 2 H), 4.12 (q, $J = 7.2$ Hz, 2 H), 4.75 (brs, 1 H), 4.76 (brs, 1 H); ^{13}C NMR (100 MHz, CDCl_3): δ 10.4, 14.4, 20.1, 21.4, 24.1, 24.3, 27.4, 34.1, 42.5, 48.4, 49.4, 60.5, 77.8, 79.8, 112.8, 146.1, 174.3; IR (Neat): 2956, 2928, 2870, 1732, 1644, 1462, 1374, 1175, 1153, 1117, 1070, 1016, $892, 856\text{ cm}^{-1}$. HRMS (APCI) cald. for $\text{C}_{17}\text{H}_{30}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 283.2273; found 283.2275.

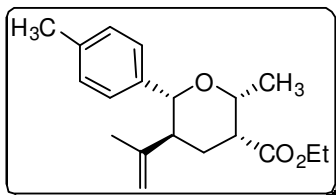
(2*R,3*R**,5*S**,6*R**)-Ethyl tetrahydro-2-methyl-6-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (33a):**



Colourless oil (110 mg, 38%); ^1H NMR (400 MHz, CDCl_3): δ 1.32 (d, $J = 6.4$ Hz, 3 H), 1.33 (t, $J = 7.2$ Hz, 3 H), 1.42 (s, 3 H), 1.90 (ddd, $J = 13.6, 12.8$ and 5.2 Hz, 1 H), 2.14 (ddd, $J = 13.6, 4.0$ and 4.0 Hz, 1 H), 2.60-2.67 (m, 1 H), 3.01 (ddd, $J = 12.8, 12.4$ and 4.0 Hz, 1 H), 3.82-3.89 (m, 1 H), 4.14-4.28 (m, 3 H), 4.63 (brs, 1 H), 4.67 (brs, 1 H), 7.22-7.25 (m, 1 H), 7.26-7.31 (m, 2 H), 7.32-7.36 (m, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.6, 20.0, 21.8, 32.7, 44.4, 45.2, 60.3, 74.3, 85.0, 112.6, 127.7, 127.8, 128.2, 141.3, 145.8, 173.0; IR (Neat): 2979, 2934, 1732, 1644, 1449, 1383, 1174, 1160, 1073, 758, 700 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{18}\text{H}_{24}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 289.1803; found 289.1799.

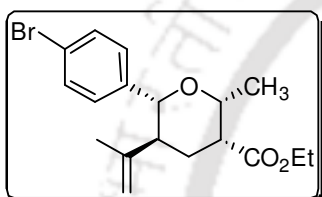
(2*R,3*R**,5*S**,6*R**)-Ethyl tetrahydro-2-methyl-5-(prop-1-en-2-yl)-6-*p*-tolyl-2*H*-pyran-3-carboxylate (33b):**

Colourless oil (103 mg, 34%); ^1H NMR (400 MHz, CDCl_3): δ 1.32 (d, $J = 6.8$ Hz, 3 H),



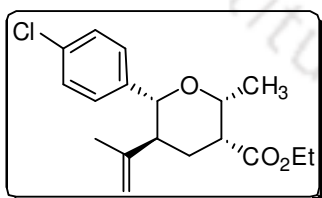
1.33 (t, $J = 7.2$ Hz, 3 H), 1.44 (s, 3 H), 1.90 (ddd, $J = 13.6$, 13.2 and 4.8 Hz, 1 H), 2.14 (ddd, $J = 13.6$, 4.0 and 4.0 Hz, 1 H), 2.30 (s, 3 H), 2.59-2.66 (m, 1 H), 3.02 (ddd, $J = 13.2$, 10.4 and 4.0 Hz, 1 H), 3.80-3.89 (m, 1 H), 4.15-4.28 (m, 3 H), 4.64 (brs, 1 H), 4.67 (brs, 1 H), 7.09 (d, $J = 7.6$ Hz, 2 H), 7.23 (d, $J = 8.0$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.6, 20.1, 21.4, 21.7, 32.8, 44.5, 45.0, 60.3, 74.3, 84.8, 112.5, 127.6, 129.0, 137.4, 138.4, 146.1, 173.1; IR (Neat): 2978, 2927, 2855, 1733, 1644, 1380, 1174, 1157, 1075, 810 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{19}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 303.1960; found 303.1958.

(2*R,3*R**,5*S**,6*R**)-Ethyl 6-(4-bromophenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (33c):**



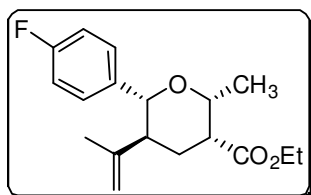
Colourless oil (110 mg, 30%); ^1H NMR (400 MHz, CDCl_3): δ 1.31 (d, $J = 5.6$ Hz, 3 H), 1.33 (t, $J = 7.2$ Hz, 3 H), 1.44 (s, 3 H), 1.90 (ddd, $J = 13.6$, 12.8 and 5.2 Hz, 1 H), 2.13 (ddd, $J = 13.6$, 4.0 and 3.6 Hz, 1 H), 2.60-2.67 (m, 1 H), 2.95 (ddd, $J = 12.8$, 12.8 and 4.0 Hz, 1 H), 3.81-3.88 (m, 1 H), 4.12-4.28 (m, 3 H), 4.62 (brs, 1 H), 4.68 (brs, 1 H), 7.22 (d, $J = 8.4$ Hz, 2 H), 7.41 (d, $J = 8.4$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.6, 20.0, 21.8, 32.6, 44.2, 45.2, 60.3, 74.3, 84.1, 113.0, 121.7, 129.4, 131.4, 140.5, 145.4, 172.9; IR (Neat): 2979, 2934, 2855, 1732, 1644, 1446, 1383, 1174, 1157, 1074, 921, 813 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{18}\text{H}_{23}\text{BrO}_3$ ($\text{M}+\text{H}$) $^+$ requires 367.0909; found 367.0911 (^{79}Br).

(2*R,3*R**,5*S**,6*R**)-Ethyl 6-(4-chlorophenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (33d):**



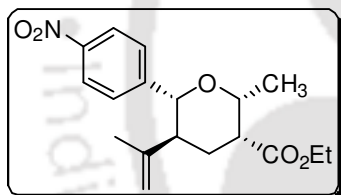
Colourless oil (94 mg, 29%); ^1H NMR (400 MHz, CDCl_3): δ 1.32 (d, $J = 6.0$ Hz, 3 H), 1.33 (t, $J = 7.2$ Hz, 3 H), 1.43 (s, 3 H), 1.89 (ddd, $J = 13.2$, 12.8 and 4.8 Hz, 1 H), 2.14 (ddd, $J = 13.2$, 4.0 and 3.6 Hz, 1 H), 2.60-2.68 (m, 1 H), 2.95 (ddd, $J = 12.8$, 12.8 and 3.6 Hz, 1 H), 3.80-3.89 (m, 1 H), 4.16-4.29 (m, 3 H), 4.62 (brs, 1 H), 4.69 (brs, 1 H), 7.26 (d, $J = 8.8$ Hz, 2 H), 7.41 (d, $J = 8.8$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.6, 20.0, 21.8, 32.6, 44.2, 45.3, 60.3, 74.3, 84.1, 113.0, 128.4, 129.0, 133.4, 140.0, 145.5, 172.9; IR (Neat): 2978, 2933, 2857, 1731, 1644, 1447, 1382, 1174, 1157, 1076, 894, 815 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{18}\text{H}_{23}\text{ClO}_3$ ($\text{M}+\text{H}$) $^+$ requires 323.1414; found 323.1405 (^{35}Cl).

(2R*,3R*,5S*,6R*)-Ethyl 6-(4-fluorophenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (33e):



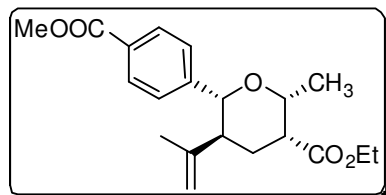
Colourless oil (107 mg, 35%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.32 (d, $J = 6.4$ Hz, 3 H), 1.33 (t, $J = 7.2$ Hz, 3 H), 1.43 (s, 3 H), 1.89 (ddd, $J = 13.6, 12.8$ and 4.6 Hz, 1 H), 2.14 (ddd $J = 13.6, 4.0$ and 3.6 Hz, 1 H), 2.60-2.67 (m, 1 H), 2.97 (ddd, $J = 12.8, 10.4$ and 4.0 Hz, 1 H), 3.82-3.89 (m, 1 H), 4.18-4.28 (m, 3 H), 4.63 (brs, 1 H), 4.69 (brs, 1 H), 6.98 (t, $J = 8.8$ Hz, 2 H), 7.30-7.34 (m, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.6, 20.0, 21.8, 32.6, 44.3, 45.4, 60.3, 74.3, 84.1, 112.8, 115.1 (d, $J = 20.6$ Hz), 129.3 (d, $J = 9.7$ Hz), 137.3, 145.7, 162.5 (d, $J = 244.1$ Hz), 173.0; $^{19}\text{F NMR}$ (376 MHz, $\text{CDCl}_3/\text{C}_6\text{F}_6$): δ -219.09-(-219.01) (m, 1F); **IR** (Neat): 2979, 2934, 2858, 1732, 1644, 1448, 1382, 1176, 1157, 1075, 832 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{23}\text{FO}_3$ ($\text{M}+\text{H}$) $^+$ requires 307.1709; found 307.1702.

(2R*,3R*,5S*,6R*)-Ethyl tetrahydro-2-methyl-6-(4-nitrophenyl)-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (33f):



Colourless oil (90 mg, 27%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.33 (d, $J = 5.6$ Hz, 3 H), 1.34 (t, $J = 7.2$ Hz, 3 H), 1.46 (s, 3 H), 1.89 (ddd, $J = 14.0, 12.8$ and 5.2 Hz, 1 H), 2.16 (ddd, $J = 14.0, 4.0$ and 2.0 Hz, 1 H), 2.63-2.70 (m, 1 H), 2.95 (ddd, $J = 12.8, 12.4$ and 4.0 Hz, 1 H), 3.84-3.92 (m, 1 H), 4.17-4.28 (m, 2 H), 4.37 (d, $J = 10.4$ Hz, 1 H), 4.62 (brs, 1 H), 4.72 (brs, 1 H), 7.51 (d, $J = 8.8$ Hz, 2 H), 8.16 (d, $J = 8.8$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.6, 19.9, 21.7, 32.4, 44.0, 45.6, 60.5, 74.3, 83.6, 113.6, 123.5, 128.5, 144.7, 147.6, 148.8, 172.8; **IR** (Neat): 2979, 2935, 2858, 1728, 1643, 1520, 1447, 1345, 1174, 1077, 851, 752 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{23}\text{NO}_5$ ($\text{M}+\text{H}$) $^+$ requires 334.1654; found 334.1666.

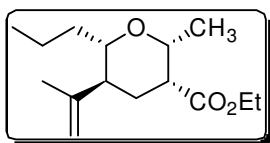
(2R*,3R*,5S*,6R*)-Ethyl 6-(4-(methoxycarbonyl)phenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (33g):



Colourless solid; M.P.: 100-102 $^{\circ}\text{C}$ (111 mg, 32%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.33 (t, $J = 7.2$ Hz, 3 H), 1.34 (d, $J = 6.8$ Hz, 3 H), 1.41 (s, 3 H), 1.92 (ddd, $J = 14.0, 12.8$ and 5.2 Hz, 1 H), 2.15 (ddd, $J = 14.0, 4.0$ and 2.0 Hz, 1 H), 2.63-2.70 (m, 1 H), 2.97 (ddd, $J = 12.8, 10.0$ and 4.0 Hz, 1 H), 3.81-3.88 (m, 1 H), 3.89 (s, 3 H), 4.16-4.28 (m, 2 H), 4.30 (d, $J = 10.0$ Hz, 1 H), 4.62 (brs, 1 H), 4.68 (brs, 1 H), 7.42 (d, $J = 8.4$ Hz, 2 H), 7.97 (d, $J = 8.4$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ

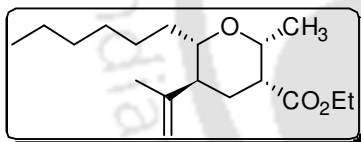
14.6, 19.9, 21.8, 32.5, 44.2, 45.4, 52.2, 60.4, 74.3, 84.4, 113.1, 127.6, 129.6 (2C), 145.2, 146.5, 167.2, 172.9; **IR** (KBr): 2955, 2928, 2873, 1726, 1643, 1449, 1277, 1173, 1158, 1076, 772, 705 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{20}\text{H}_{26}\text{O}_5$ ($\text{M}+\text{H}$)⁺ requires 347.1858; found 347.1868.

(2*R,3*R**,5*S**,6*S**)-Ethyl tetrahydro-2-methyl-5-(prop-1-en-2-yl)-6-propyl-2*H*-pyran-3-carboxylate (33h):**



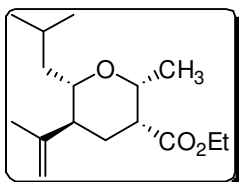
Colourless oil (104 mg, 41%); **¹H NMR** (400 MHz, CDCl_3): δ 0.97 (t, $J = 7.6$ Hz, 3 H), 1.27 (t, $J = 7.2$ Hz, 3 H), 1.28 (d, $J = 6.0$ Hz, 3 H), 1.30-1.42 (m, 2 H), 1.43-1.58 (m, 2 H), 1.67 (s, 3 H), 1.72 (ddd, $J = 13.6, 12.8$ and 5.2 Hz, 1 H), 1.98 (ddd, $J = 13.6, 4.0$ and 2.4 Hz, 1 H), 2.49-2.55 (m, 1 H), 2.63 (ddd, $J = 12.8, 10.0$ and 4.0 Hz, 1 H), 3.26 (dt, $J = 10.0$ and 6.0 Hz, 1 H), 3.62-3.70 (m, 1 H), 4.11-4.21 (m, 2 H), 4.77 (brs, 2 H); **¹³C NMR** (100 MHz, CDCl_3): δ 14.1, 14.4, 18.8, 19.8, 20.2, 32.6, 35.6, 44.2, 44.3, 60.1, 73.9, 80.7, 112.4, 146.7, 173.0; **IR** (Neat): 2960, 2936, 2872, 1735, 1648, 1382, 1173, 1115, 1084, 992, 892 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{15}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 255.1960; found 255.1965.

(2*R,3*R**,5*S**,6*S**)-Ethyl 6-hexyl-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (33i):**



Colourless oil (116 mg, 39%); **¹H NMR** (400 MHz, CDCl_3): δ 0.87 (t, $J = 7.2$ Hz, 3 H), 1.20-1.40 (m, 8 H), 1.27 (t, $J = 7.2$ Hz, 3 H), 1.28 (d, $J = 6.0$ Hz, 3 H), 1.45-1.53 (m, 2 H), 1.67 (s, 3 H), 1.72 (ddd, $J = 12.8, 12.8$ and 5.2 Hz, 1 H), 1.97 (ddd, $J = 12.8, 4.0$ and 2.0 Hz, 1 H), 2.50-2.56 (m, 1 H), 2.63 (ddd, $J = 12.8, 10.0$ and 4.0 Hz, 1 H), 3.25 (dt, $J = 10.0$ and 8.4 Hz, 1 H), 3.62-3.69 (m, 1 H), 4.11-4.21 (m, 2 H), 4.77 (brs, 2 H); **¹³C NMR** (100 MHz, CDCl_3): δ 14.3, 14.5, 19.9, 20.4, 22.9, 25.8, 29.5, 32.1, 32.6, 33.6, 44.3, 44.4, 60.2, 74.0, 81.1, 112.5, 146.8, 173.1; **IR** (Neat): 2927, 2856, 1735, 1640, 1445, 1380, 1173, 1113, 1083, 892 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{32}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 297.2429; found 297.2424.

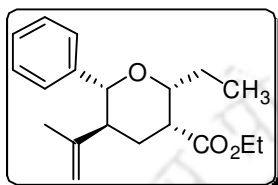
(2*R,3*R**,5*S**,6*S**)-Ethyl tetrahydro-6-isobutyl-2-methyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (33j):**



Colourless oil (115 mg, 43%); **¹H NMR** (400 MHz, CDCl_3): δ 0.84 (d, $J = 6.4$ Hz, 3 H), 0.88 (d, $J = 6.4$ Hz, 3 H), 1.19 (ddd, $J = 10.4, 8.0$ and 6.0 Hz, 1 H), 1.27 (t, $J = 6.8$ Hz, 3 H), 1.28 (d, $J = 6.4$ Hz, 3 H), 1.39 (ddd, $J = 10.4, 6.4$ and 4.0 Hz, 1 H), 1.67 (s, 3 H), 1.73 (ddd, $J = 13.6, 13.2$ and 5.2 Hz, 1 H), 1.83-1.91 (m, 1 H), 1.98 (ddd, $J = 13.6, 4.0$ and 2.0 Hz, 1 H),

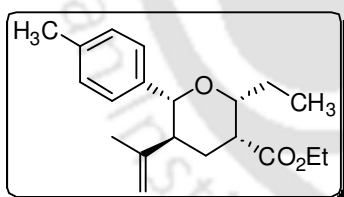
2.49-2.55 (m, 1 H), 2.60 (ddd, $J = 13.2, 10.8$ and 4.0 Hz, 1 H), 3.32 (dt, $J = 10.8$ and 8.4 Hz, 1 H), 3.62-3.69 (m, 1 H), 4.11-4.21 (m, 2 H), 4.77 (brs, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.5, 19.9, 20.2, 21.6, 24.0, 24.1, 32.7, 42.5, 44.3, 44.8, 60.2, 74.0, 79.0, 112.5, 146.8, 173.1; IR (Neat): 2954, 2928, 2868, 1736, 1644, 1448, 1383, 1173, 1116, 1082, 1026, 892 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{16}\text{H}_{28}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 269.2116; found 269.2112.

(2*R,3*R**,5*S**,6*R**)-Ethyl 2-ethyl-tetrahydro-6-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (33k):**



Colourless oil (106 mg, 35%); ^1H NMR (400 MHz, CDCl_3): δ 0.97 (t, $J = 7.6$ Hz, 3 H), 1.32 (t, $J = 7.2$ Hz, 3 H), 1.43 (s, 3 H), 1.57-1.74 (m, 2 H), 1.88 (ddd, $J = 13.6, 13.2$ and 5.6 Hz, 1 H), 2.16 (ddd, $J = 13.6, 4.0$ and 2.0 Hz, 1 H), 2.70-2.76 (m, 1 H), 3.03 (ddd, $J = 13.2, 10.4$ and 4.0 Hz, 1 H), 3.55 (dt, $J = 6.8$ and 2.8 Hz, 1 H), 4.16-4.26 (m, 3 H), 4.63 (brs, 1 H), 4.67 (brs, 1 H), 7.20-7.31 (m, 3 H), 7.33-7.35 (m, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 10.8, 14.6, 21.8, 27.1, 32.8, 42.5, 45.5, 60.3, 80.4, 85.1, 112.6, 127.7, 128.0, 128.2, 141.5, 146.0, 173.0; IR (Neat): 2965, 2935, 2856, 1733, 1644, 1453, 1376, 1173, 1071, 892, 756, 699 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{19}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 303.1960; found 303.1969.

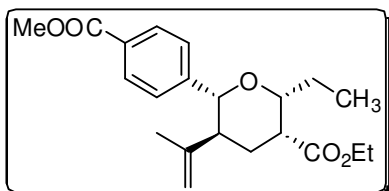
(2*R,3*R**,5*S**,6*R**)-Ethyl 2-ethyl-tetrahydro-5-(prop-1-en-2-yl)-6-*p*-tolyl-2*H*-pyran-3-carboxylate (33l):**



Colourless oil (114 mg, 36%); ^1H NMR (400 MHz, CDCl_3): δ 0.96 (t, $J = 7.6$ Hz, 3 H), 1.32 (t, $J = 7.2$ Hz, 3 H), 1.45 (s, 3 H), 1.57-1.72 (m, 2 H), 1.86 (ddd, $J = 13.6, 13.2$ and 5.2 Hz, 1 H), 2.15 (ddd, $J = 13.6, 4.0$ and 2.0 Hz, 1 H), 2.30 (s, 3 H), 2.68-2.74 (m, 1 H), 3.03 (ddd, $J = 13.2, 10.8$ and 4.0 Hz, 1 H), 3.54 (dt, $J = 7.2$ and 4.4 Hz, 1 H), 4.14-4.26 (m, 3 H), 4.64 (brs, 1 H), 4.67 (brs, 1 H), 7.09 (d, $J = 7.6$ Hz, 2 H), 7.22 (d, $J = 8.0$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 10.8, 14.6, 21.4, 21.7, 27.1, 32.9, 42.6, 45.3, 60.2, 80.4, 84.8, 112.5, 127.6, 128.9, 137.3, 138.5, 146.2, 173.1; IR (Neat): 2961, 2935, 2956, 1733, 1644, 1449, 1376, 1174, 1160, 1074, 890, 813 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{20}\text{H}_{28}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 317.2116; found 317.2122.

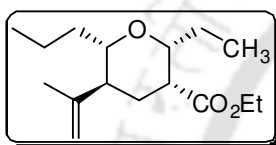
(2*R,3*R**,5*S**,6*R**)-Ethyl 6-(4-(methoxycarbonyl)phenyl)-2-ethyl-tetrahydro-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (33m):**

Colourless oil (112 mg, 31%); ^1H NMR (400 MHz, CDCl_3): δ 0.97 (t, $J = 7.6$ Hz, 3 H), 1.33 (t, $J = 7.2$ Hz, 3 H), 1.42 (s, 3 H), 1.57-1.74 (m, 2 H), 1.89 (ddd, $J = 12.8, 12.8$ and 5.2



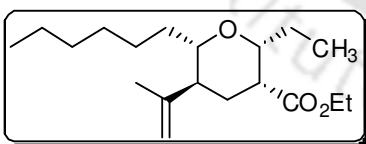
Hz, 1 H), 2.16 (ddd, $J = 12.8, 4.0$ and 2.0 Hz, 1 H), 2.71-2.77 (m, 1 H), 2.98 (ddd, $J = 12.8, 10.4$ and 4.0 Hz, 1 H), 3.56 (dt, $J = 9.6$ and 3.2 Hz, 1 H), 3.89 (s, 3 H), 4.17-4.30 (m, 2 H), 4.28 (d, $J = 10.4$ Hz, 1 H), 4.62 (brs, 1 H), 4.68 (brs, 1 H), 7.42 (d, $J = 8.0$ Hz, 2 H), 7.97 (d, $J = 8.4$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 10.7, 14.6, 21.8, 27.1, 32.7, 42.4, 45.7, 52.2, 60.3, 80.3, 84.5, 113.0, 127.7, 129.3, 129.5, 145.4, 146.7, 167.3, 172.9; **IR** (Neat): 2935, 2851, 1726, 1644, 1436, 1376, 1278, 1176, 1112, 1078, 859, 767 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{21}\text{H}_{28}\text{O}_5$ ($\text{M}+\text{H}$) $^+$ requires 361.2015; found 361.2012.

(2*R,3*R**,5*S**,6*S**)-Ethyl 2-ethyl-tetrahydro-5-(prop-1-en-2-yl)-6-propyl-2H-pyran-3-carboxylate (33n):**

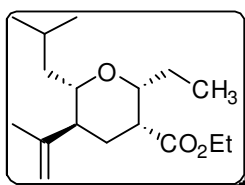


Colourless oil (123 mg, 46%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.88 (t, $J = 7.2$ Hz, 3 H), 0.99 (t, $J = 7.2$ Hz, 3 H), 1.26 (t, $J = 7.2$ Hz, 3 H), 1.29-1.51 (m, 4 H), 1.52-1.64 (m, 2 H), 1.67 (s, 3 H), 1.71 (ddd, $J = 14.0, 12.4$ and 5.2 Hz, 1 H), 1.98 (ddd, $J = 14.0, 4.0$ and 2.0 Hz, 1 H), 2.56-2.62 (m, 1 H), 2.64 (ddd, $J = 12.4, 10.0$ and 4.0 Hz, 1 H), 3.23 (dt, $J = 10.0$ and 6.8 Hz, 1 H), 3.35 (ddd, $J = 8.4, 5.6$ and 3.2 Hz, 1 H), 4.08-4.20 (m, 2 H), 4.77 (brs, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 11.0, 14.1, 14.5, 19.0, 20.4, 27.2, 32.7, 35.7, 43.0, 44.9, 60.1, 80.1, 80.9, 112.4, 146.9, 173.1; **IR** (Neat): 2960, 2936, 2874, 1736, 1644, 1456, 1379, 1174, 1115, 1084, 1033, 892 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{16}\text{H}_{28}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 269.2116; found 269.2125.

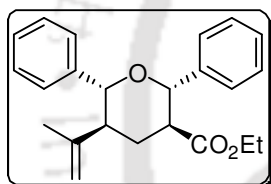
(2*R,3*R**,5*S**,6*S**)-Ethyl 2-ethyl-6-hexyl-tetrahydro-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (33o):**



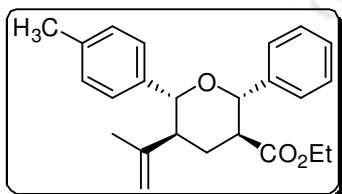
Colourless oil (121 mg, 39%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.87 (t, $J = 6.8$ Hz, 3 H), 0.99 (t, $J = 7.6$ Hz, 3 H), 1.24-1.40 (m, 10 H), 1.26 (t, $J = 7.2$ Hz, 3 H), 1.47-1.54 (m, 1 H), 1.55-1.64 (m, 1 H), 1.67 (s, 3 H), 1.71 (ddd, $J = 13.2, 12.8$ and 5.2 Hz, 1 H), 1.98 (ddd, $J = 13.2, 4.0$ and 1.6 Hz, 1 H), 2.55-2.61 (m, 1 H), 2.64 (ddd, $J = 12.8, 10.4$ and 4.0 Hz, 1 H), 3.22 (dt, $J = 10.4$ and 8.4 Hz, 1 H), 3.34 (ddd, $J = 8.4, 5.6$ and 3.2 Hz, 1 H), 4.09-4.20 (m, 2 H), 4.77 (brs, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 10.9, 14.2, 14.4, 20.3, 22.8, 25.8, 27.1, 29.4, 32.1, 32.7, 33.6, 42.9, 44.8, 60.0, 80.1, 81.1, 112.3, 146.8, 173.0; **IR** (Neat): 2930, 2857, 1736, 1644, 1457, 1378, 1173, 1157, 1085, 1032, 892 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{19}\text{H}_{34}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 311.2586; found 311.2588.

(2R*,3R*,5S*,6S*)-Ethyl 2-ethyl-tetrahydro-6-isobutyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (33p):

Colourless oil (113 mg, 40%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.85 (d, $J = 6.8$ Hz, 3 H), 0.88 (d, $J = 6.8$ Hz, 3 H), 0.99 (t, $J = 7.2$ Hz, 3 H), 1.19 (ddd, $J = 10.8, 8.8$ and 8.4 Hz, 1 H), 1.26 (t, $J = 7.2$ Hz, 3 H), 1.42 (ddd, $J = 10.4, 6.4$ and 4.0 Hz, 1 H), 1.50-1.64 (m, 2 H), 1.66 (s, 3 H), 1.73 (ddd, $J = 12.8, 12.8$ and 4.0 Hz, 1 H), 1.85-1.94 (m, 1 H), 1.98 (ddd, $J = 12.8, 4.0$ and 2.0 Hz, 1 H), 2.55-2.61 (m, 1 H), 2.62 (ddd, $J = 12.8, 10.0$ and 4.0 Hz, 1 H), 3.27-3.37 (m, 2 H), 4.08-4.20 (m, 2 H), 4.77 (brs, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 11.1, 14.5, 20.2, 21.4, 24.0, 24.1, 27.1, 32.7, 42.5, 43.0, 45.2, 60.1, 79.1, 80.2, 112.5, 146.7, 173.1; **IR** (Neat): 2956, 2934, 2870, 1735, 1643, 1465, 1382, 1173, 1116, 1084, 1015, 891, 853 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{17}\text{H}_{30}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 283.2273; found 283.2280.

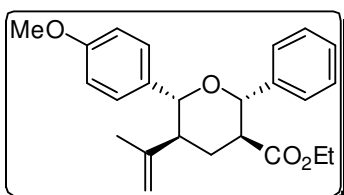
(2S*,3S*,5S*,6R*)-Ethyl tetrahydro-2,6-diphenyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (40a):

Colourless oil (280 mg, 80%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.01 (t, $J = 6.8$ Hz, 3 H), 1.48 (s, 3 H), 2.17-2.23 (m, 2 H), 2.48 (ddd, $J = 10.4, 10.0$ and 3.2 Hz, 1 H), 2.87 (ddd, $J = 10.0, 9.6$ and 2.4 Hz, 1 H), 3.94 (q, $J = 6.8$ Hz, 2 H), 4.49 (d, $J = 10.0$ Hz, 1 H), 4.69-4.73 (m, 3 H), 7.23-7.32 (m, 4 H), 7.33-7.40 (m, 6 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 21.7, 34.2, 49.7, 50.5, 60.4, 81.8, 84.6, 113.2, 127.2, 127.4, 127.9, 128.2(2C), 128.4, 140.2, 140.6, 144.9, 173.1; **IR** (Neat): 2978, 2934, 2865, 1728, 1644, 1452, 1373, 1174, 1089, 1064, 755, 699 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{23}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 351.1960; found 351.1962.

(2S*,3S*,5S*,6R*)-Ethyl tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-6-*p*-tolyl-2H-pyran-3-carboxylate (40b):

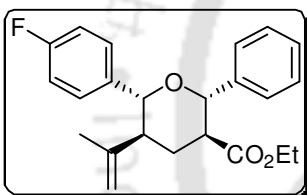
Colourless oil (255 mg, 70%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.00 (t, $J = 7.2$ Hz, 3 H), 1.50 (s, 3 H), 2.16-2.21 (m, 2 H), 2.29 (s, 3 H), 2.48 (ddd, $J = 9.6, 9.6$ and 2.4 Hz, 1 H), 2.84 (ddd, $J = 9.6, 9.6$ and 2.8 Hz, 1 H), 3.93 (q, $J = 7.2$ Hz, 2 H), 4.45 (d, $J = 9.6$ Hz, 1 H), 4.69-4.72 (m, 3 H), 7.08 (d, $J = 7.6$ Hz, 2 H), 7.22-7.30 (m, 5 H), 7.38 (d, $J = 7.6$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.0, 21.3, 21.6, 34.2, 49.5, 50.5, 60.4, 81.8, 84.3, 113.1, 127.2, 127.4, 128.1, 128.3, 128.9, 137.5, 137.6, 140.3, 145.1, 173.2; **IR** (Neat): 2978, 2930, 2865, 1728, 1644, 1452, 1373, 1173, 1062, 892, 768 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{24}\text{H}_{28}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 365.2116; found 365.2112.

(2*S,3*S**,5*S**,6*R**)-Ethyl tetrahydro-6-(4-methoxyphenyl)-2-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (40c):**



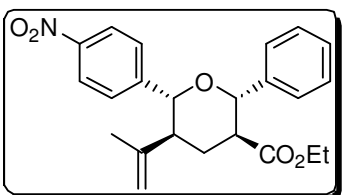
Colourless solid; M.P. 114-116 °C (281 mg, 74%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.00 (t, $J = 7.2$ Hz, 3 H), 1.49 (s, 3 H), 2.16-2.21 (m, 2 H), 2.47 (ddd, $J = 10.0, 9.6$ and 2.8 Hz, 1 H), 2.84 (ddd, $J = 10.0, 9.6$ and 3.2 Hz, 1 H), 3.75 (s, 3 H), 3.93 (q, $J = 7.2$ Hz, 2 H), 4.44 (d, $J = 10.0$ Hz, 1 H), 4.68-4.73 (m, 3 H), 6.82 (d, $J = 7.8$ Hz, 2 H), 7.20-7.31 (m, 5 H), 7.38 (d, $J = 7.6$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 21.7, 34.3, 49.7, 50.6, 55.3, 60.5, 81.9, 84.1, 113.1, 113.6, 127.3, 128.2, 128.4, 128.7, 133.0, 140.3, 145.2, 159.3, 173.2; **IR** (Neat): 2936, 2868, 1727, 1643, 1454, 1372, 1174, 1059, 1039, 910, 768 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{24}\text{H}_{28}\text{O}_4$ ($\text{M}+\text{H}$) $^+$ requires 381.2066; found 381.2062.

(2*S,3*S**,5*S**,6*R**)-Ethyl 6-(4-fluorophenyl)-tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (40d):**



Colourless oil (221 mg, 60%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.00 (t, $J = 7.2$ Hz, 3 H), 1.49 (s, 3 H), 2.17-2.21 (m, 2 H), 2.42 (ddd, $J = 10.4, 10.0$ and 2.4 Hz, 1 H), 2.85 (ddd, $J = 10.0, 9.6$ and 2.8 Hz, 1 H), 3.93 (q, $J = 7.2$ Hz, 2 H), 4.47 (d, $J = 10.4$ Hz, 1 H), 4.68-4.72 (m, 3 H), 6.98 (t, $J = 8.4$ Hz, 2 H), 7.23-7.34 (m, 5 H), 7.38 (d, $J = 7.6$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 21.7, 34.1, 50.0, 50.5, 60.5, 82.0, 83.8, 113.4, 115.0 (d, $J = 21.4$ Hz), 127.2, 128.4 (d, $J = 10.7$ Hz), 129.0, 129.1, 136.6, 140.1, 144.8, 162.5 (d, $J = 243.2$ Hz), 173.1; $^{19}\text{F NMR}$ (376 MHz, $\text{CDCl}_3/\text{C}_6\text{F}_6$): δ -219.01-(-218.94) (m, 1F); **IR** (Neat): 2929, 2857, 1728, 1640, 1453, 1374, 1174, 1157, 1086, 1064, 836, 769 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{23}\text{H}_{25}\text{FO}_3$ ($\text{M}+\text{H}$) $^+$ requires 369.1866; found 369.1873.

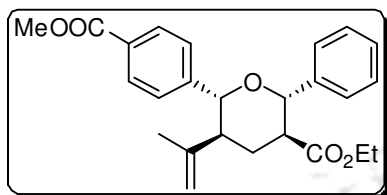
(2*S,3*S**,5*S**,6*R**)-Ethyl tetrahydro-6-(4-nitrophenyl)-2-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (40e):**



Colourless oil (237 mg, 60%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.01 (t, $J = 6.8$ Hz, 3 H), 1.52 (s, 3 H), 2.18-2.30 (m, 2 H), 2.40 (ddd, $J = 10.0, 10.0$ and 2.4 Hz, 1 H), 2.89 (ddd, $J = 10.4, 10.0$ and 2.4 Hz, 1 H), 3.95 (q, $J = 6.8$ Hz, 2 H), 4.61 (d, $J = 10.0$ Hz, 1 H), 4.68 (brs, 1 H), 4.74 (d, $J = 10.0$ Hz, 1 H), 4.76 (brs, 1 H), 7.26-7.41 (m, 5 H), 7.52 (d, $J = 8.8$ Hz, 2 H), 8.16 ($J = 8.8$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 21.6, 33.9, 50.1, 50.2, 60.7, 81.9, 83.3, 114.2, 123.5, 127.2, 128.2, 128.6 (2C), 139.6, 143.9, 147.6, 147.9, 172.9; **IR** (Neat): 2978, 2927, 2855, 1729, 1606, 1522, 1347, 1176,

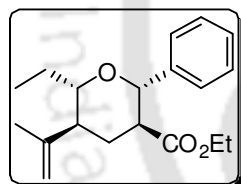
1070, 751, 699 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{23}\text{H}_{25}\text{NO}_5$ ($\text{M}+\text{H}$)⁺ requires 396.1811; found 396.1815.

(2*S,3*S**,5*S**,6*R**)-Ethyl 6-(4-(methoxycarbonyl)phenyl)-tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (40f):**



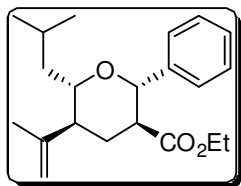
Colourless solid; M.P. 118-120 °C (266 mg, 65%); **¹H NMR** (400 MHz, CDCl_3): δ 1.01 (t, $J = 7.2$ Hz, 3 H), 1.48 (s, 3 H), 2.17-2.23 (m, 2 H), 2.44 (ddd, $J = 10.0, 9.6$ and 2.8 Hz, 1 H), 2.89 (ddd, $J = 10.0, 9.6$ and 3.2 Hz, 1 H), 3.89 (s, 3 H), 3.94 (q, $J = 7.2$ Hz, 2 H), 4.54 (d, $J = 10.0$ Hz, 1 H), 4.67 (brs, 1 H), 4.72 (brs, 1 H), 4.73 (d, $J = 9.6$ Hz, 1 H), 7.27-7.33 (m, 3 H), 7.39-7.43 (m, 4 H), 7.97 (d, $J = 8.4$ Hz, 2 H); **¹³C NMR** (100 MHz, CDCl_3): δ 14.0, 21.6, 33.9, 49.9, 50.3, 52.0, 60.4, 81.7, 83.9, 113.5, 127.1, 127.3, 128.2, 128.3, 129.4, 129.6, 139.9, 144.2, 145.6, 166.9, 172.8; **IR** (KBr): 2978, 2950, 2862, 1725, 1643, 1374, 1278, 1176, 1111, 1065, 897, 763 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{25}\text{H}_{28}\text{O}_5$ ($\text{M}+\text{H}$)⁺ requires 409.2015; found 409.2025.

(2*S,3*S**,5*S**,6*S**)-Ethyl 6-ethyl-tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (40g):**



Colourless oil (278 mg, 92%); **¹H NMR** (400 MHz, CDCl_3): δ 0.93 (t, $J = 7.2$ Hz, 3 H), 0.99 (t, $J = 7.2$ Hz, 3 H), 1.38-1.47 (m, 1 H), 1.61-1.71 (m, 1 H), 1.72 (s, 3 H), 1.99-2.05 (m, 2 H), 2.20 (ddd, $J = 10.4, 10.0$ and 5.6 Hz, 1 H), 2.67 (ddd, $J = 10.0, 10.0$ and 5.6 Hz, 1 H), 3.47 (ddd, $J = 10.4, 10.0$ and 7.6 Hz, 1 H), 3.91 (q, $J = 7.2$ Hz, 2 H), 4.51 (d, $J = 10.0$ Hz, 1 H), 4.82 (brs, 2 H), 7.24-7.36 (m, 5 H); **¹³C NMR** (100 MHz, CDCl_3): δ 9.8, 14.1, 20.2, 26.2, 34.0, 47.8, 50.4, 60.4, 81.2, 81.4, 112.9, 127.2, 128.1, 128.4, 140.8, 146.0, 173.5; **IR** (Neat): 2961, 2931, 2875, 1730, 1644, 1374, 1344, 1174, 1067, 894, 754, 699 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{19}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 303.1960; found 303.1954.

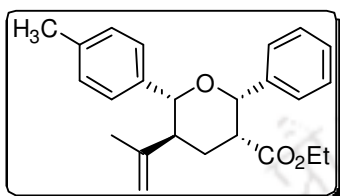
(2*S,3*S**,5*S**,6*S**)-Ethyl tetrahydro-6-isobutyl-2-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (40h):**



Colourless oil (288 mg, 87%); **¹H NMR** (400 MHz, CDCl_3): δ 0.84 (d, $J = 6.4$ Hz, 3 H), 0.85 (d, $J = 7.2$ Hz, 3 H), 1.00 (t, $J = 7.2$ Hz, 3 H), 1.26 (ddd, $J = 14.0, 10.0$ and 2.0 Hz, 1 H), 1.40 (ddd, $J = 14.0, 10.0$ and 4.0 Hz, 1 H), 1.71 (s, 3 H), 1.77-1.87 (m, 1 H), 1.99-2.05 (m, 2 H), 2.11 (ddd, $J = 9.6, 9.6$ and 5.2 Hz, 1 H), 2.68 (ddd, $J = 10.0, 10.0$ and 5.6 Hz, 1 H), 3.57 (dt, $J = 10.0$ and 7.6 Hz, 1 H), 3.92 (q, $J = 7.2$ Hz, 2 H), 4.51 (d, $J = 10.0$ Hz, 1 H), 4.81 (brs, 2 H), 7.25-7.29 (m, 2 H), 7.29-7.35 (m, 3 H); **¹³C NMR** (100 MHz, CDCl_3): δ

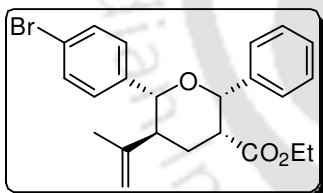
14.1, 20.1, 21.6, 24.0 (2C), 34.2, 42.5, 49.1, 50.1, 60.3, 78.5, 81.2, 113.0, 127.1, 128.0, 128.3, 140.8, 145.9, 173.5; **IR** (Neat): 2954, 2928, 2869, 1730, 1644, 1371, 1173, 1093, 1066, 849, 699 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{21}\text{H}_{30}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 331.2273; found 331.2262.

(2S*,3R*,5S*,6R*)-Ethyl tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-6-*p*-tolyl-2H-pyran-3-carboxylate (41a):



Colourless oil (262 mg, 72%); **¹H NMR** (400 MHz, CDCl_3): δ 0.89 (t, $J = 7.2$ Hz, 3 H), 1.47 (s, 3 H), 2.12 (ddd, $J = 14.0$, 12.8 and 5.2 Hz, 1 H), 2.23 (ddd, $J = 14.0$, 4.4 and 4.0 Hz, 1 H), 2.32 (s, 3 H), 3.01-3.03 (m, 1 H), 3.15 (ddd, $J = 12.8$, 10.4 and 4.0 Hz, 1 H), 3.83 (q, $J = 7.2$ Hz, 2 H), 4.38 (d, $J = 10.4$ Hz, 1 H), 4.70 (brs, 1 H), 4.72 (brs, 1 H), 4.86 (d, $J = 3.2$ Hz, 1 H), 7.12 (d, $J = 8.0$ Hz, 2 H), 7.18-7.21 (m, 1 H), 7.24-7.29 (m, 2 H), 7.32-7.37 (m, 4 H); **¹³C NMR** (100 MHz, CDCl_3): δ 14.0, 21.4, 21.9, 32.8, 45.1, 45.7, 60.0, 79.4, 84.9, 112.7, 125.7, 127.3, 127.6, 128.0, 128.9, 137.4, 138.4, 140.7, 145.9, 172.4; **IR** (Neat): 2926, 2854, 1731, 1644, 1451, 1373, 1174, 1067, 1028, 814, 748 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{24}\text{H}_{28}\text{O}_3$ ($\text{M}+\text{H}$)⁺ requires 365.2116; found 365.2108.

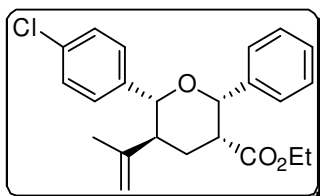
(2S*,3R*,5S*,6R*)-Ethyl 6-(4-bromophenyl)-tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (41b):



Colourless oil (249 mg, 58%); **¹H NMR** (400 MHz, CDCl_3): δ 0.88 (t, $J = 7.2$ Hz, 3 H), 1.48 (s, 3 H), 2.11 (ddd, $J = 13.6$, 12.8 and 4.8 Hz, 1 H), 2.23 (ddd, $J = 13.6$, 4.0 and 2.0 Hz, 1 H), 2.98-3.02 (m, 1 H), 3.10 (ddd, $J = 12.8$, 10.0 and 4.0 Hz, 1 H), 3.84 (q, $J = 7.2$ Hz, 2 H), 4.38 (d, $J = 10.0$ Hz, 1 H), 4.69 (brs, 1 H), 4.75 (brs, 1 H), 4.86 (d, $J = 3.2$ Hz, 1 H), 7.09 (d, $J = 8.0$ Hz, 2 H), 7.20-7.36 (m, 5 H), 7.45 (d, $J = 8.4$ Hz, 2 H); **¹³C NMR** (100 MHz, CDCl_3): δ 14.0, 21.9, 32.6, 45.2, 45.5, 60.0, 79.3, 84.2, 113.2, 126.1, 127.4, 128.1, 129.4, 130.7, 131.3, 140.3, 140.5, 145.2, 172.3; **IR** (Neat): 2978, 2934, 2859, 1730, 1644, 1484, 1374, 1174, 1070, 1031, 817, 748 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{23}\text{H}_{25}\text{BrO}_3$ ($\text{M}+\text{H}$)⁺ requires 429.1065; found 429.1060 (^{79}Br).

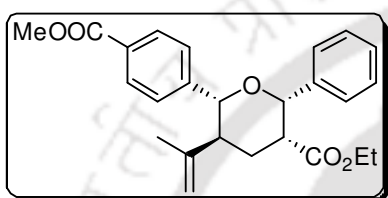
(2S*,3R*,5S*,6R*)-Ethyl 6-(4-chlorophenyl)-tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (41c):

Colourless oil (212 mg, 55%); **¹H NMR** (400 MHz, CDCl_3): δ 0.88 (t, $J = 7.2$ Hz, 3 H), 1.48 (s, 3 H), 2.12 (ddd, $J = 14.0$, 12.8 and 4.8 Hz, 1 H), 2.23 (ddd, $J = 14.0$, 4.0 and 2.0 Hz, 1 H), 2.99-3.03 (m, 1 H), 3.10 (ddd, $J = 12.8$, 10.4 and 4.0 Hz, 1 H), 3.84 (q, $J = 7.2$ Hz,



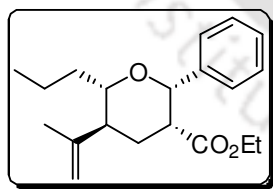
2 H), 4.40 (d, $J = 10.4$ Hz, 1 H), 4.69 (brs, 1 H), 4.75 (brs, 1 H), 4.86 (d, $J = 3.2$ Hz, 1 H), 7.20-7.31 (m, 5 H), 7.35(d, $J = 8.8$ Hz, 2 H), 7.39 ($J = 8.4$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 13.9, 21.9, 32.5, 45.2, 45.5, 60.0, 79.3, 84.2, 113.1, 125.6, 127.4, 128.1, 128.3, 129.0, 130.3, 140.0, 140.3, 145.2, 172.2; IR (Neat): 2978, 2933, 2857, 1730, 1644, 1452, 1374, 1174, 1088, 1069, 895, 748 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{23}\text{H}_{25}\text{ClO}_3$ (M+H) $^+$ requires 385.1570; found 385.1565 (^{35}Cl).

(2S*,3R*,5S*,6R*)-Ethyl 6-(4-(methoxycarbonyl)phenyl)-tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (41d):



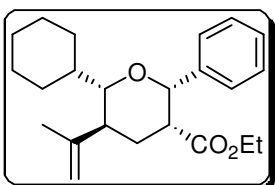
Colourless oil (253 mg, 62%); ^1H NMR (400 MHz, CDCl_3): δ 0.90 (t, $J = 7.2$ Hz, 3 H), 1.46 (s, 3 H), 2.13 (ddd, $J = 14.0, 12.8$ and 4.8 Hz, 1 H), 2.25 (ddd, $J = 14.0, 4.0$ and 2.4 Hz, 1 H), 3.02-3.06 (m, 1 H), 3.12 (ddd, $J = 12.8, 10.4$ and 4.0 Hz, 1 H), 3.86 (q, $J = 7.2$ Hz, 2 H), 3.91 (s, 3 H), 4.48 (d, $J = 10.4$ Hz, 1 H), 4.69 (brs, 1 H), 4.74 (brs, 1 H), 4.88 (d, $J = 3.2$ Hz, 1 H), 7.20-7.24 (m, 1 H), 7.26-7.32 (m, 2 H), 7.34-7.38 (m, 2 H), 7.53 (d, $J = 8.4$ Hz, 2 H), 8.01 (d, $J = 8.4$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 13.9, 21.9, 32.5, 45.3, 45.5, 52.1, 59.9, 79.3, 84.5, 113.3, 125.6, 127.4, 127.6, 128.1, 128.4, 129.5, 140.2, 145.0, 146.5, 167.2, 172.2; IR (Neat): 2983, 2951, 2857, 1725, 1644, 1372, 1276, 1177, 1100, 1070, 898, 758 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{25}\text{H}_{28}\text{O}_5$ (M+H) $^+$ requires 409.2015; found 409.2021.

(2S*,3R*,5S*,6S*)-Ethyl tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-6-propyl-2H-pyran-3-carboxylate (41e):



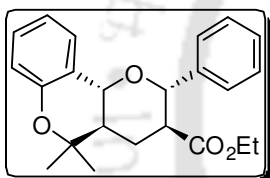
Colourless oil (285 mg, 90%); ^1H NMR (400 MHz, CDCl_3): δ 0.83 (t, $J = 7.2$ Hz, 3 H), 0.91 (t, $J = 6.8$ Hz, 3 H), 1.39-1.46 (m, 1 H), 1.49-1.68 (m, 3 H), 1.72 (s, 3 H), 1.94 (ddd, $J = 13.6, 13.2$ and 5.6 Hz, 1 H), 2.10 (ddd, $J = 13.6, 3.6$ and 2.0 Hz, 1 H), 2.79 (ddd, $J = 13.2, 10.4$ and 3.6 Hz, 1 H), 2.94-2.98 (m, 1 H), 3.47 (ddd, $J = 10.4, 7.2$ and 7.2 Hz, 1 H), 3.77 (q, $J = 7.2$ Hz, 2 H), 4.69 (d, $J = 3.2$ Hz, 1 H), 4.83 (brs, 1 H), 4.84 (brs, 1 H), 7.20-7.26 (m, 2 H), 7.28-7.36 (m, 3 H); ^{13}C NMR (100 MHz, CDCl_3): δ 13.8, 14.3, 18.7, 20.3, 32.6, 35.6, 43.9, 45.3, 59.7, 78.9, 80.8, 112.6, 125.5, 127.0, 128.0, 141.2, 146.5, 172.3; IR (Neat): 2958, 2934, 2871, 1732, 1643, 1378, 1175, 1098, 1047, 894, 746 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{20}\text{H}_{28}\text{O}_3$ (M+H) $^+$ requires 317.2116; found 317.2110.

(2S*,3R*,5S*,6S*)-Ethyl 6-cyclohexyl-tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-2H-pyran-3-carboxylate (41f):



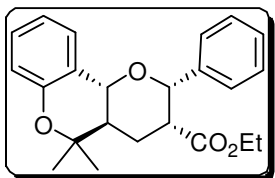
Colourless oil (285 mg, 80%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.84 (t, $J = 7.2$ Hz, 3 H), 1.15-1.28 (m, 4 H), 1.43-1.66 (m, 5 H), 1.72 (s, 3 H), 1.75-1.87 (m, 2 H), 1.94 (ddd, $J = 13.6, 12.4$ and 5.2 Hz, 1 H), 2.08 (ddd, $J = 13.6, 4.4$ and 2.0 Hz, 1 H), 2.90-2.93 (m, 1 H), 3.04 (ddd, $J = 12.4, 10.0$ and 4.0 Hz, 1 H), 3.32 (d, $J = 10.0$ Hz, 1 H), 3.77 (q, $J = 7.2$ Hz, 2 H), 4.64 (d, $J = 3.2$ Hz, 1 H), 4.83 (brs, 1 H), 4.84 (brs, 1 H), 7.20-7.40 (m, 5 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.0, 20.2, 25.7, 26.8, 26.9, 27.0, 31.3, 32.8, 39.7, 40.1, 45.6, 59.8, 78.9, 84.8, 112.8, 125.6, 127.0, 128.0, 141.5, 146.6, 172.3; **IR** (Neat): 2933, 2854, 1737, 1643, 1373, 1168, 1157, 1099, 1058, 890, 751 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{23}\text{H}_{32}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 357.2429; found 357.2435.

(2S*,3S*,4aR*,10bR*)-Ethyl 2,3,4,4a,5,10b-hexahydro-5,5-dimethyl-2-phenylpyrano[3,2-c]chromene-3-carboxylate (42):



Colourless solid; M.P.: 166-169 $^{\circ}\text{C}$ (213 mg, 58%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.00 (t, $J = 7.2$ Hz, 3 H), 1.27 (s, 3 H), 1.45 (s, 3 H), 1.82 (ddd, $J = 12.4, 12.0$ and 12.0 Hz, 1 H), 1.91 (ddd, $J = 12.4, 4.0$ and 3.2 Hz, 1 H), 2.21 (ddd, $J = 12.0, 10.0$ and 3.2 Hz, 1 H), 2.77 (ddd, $J = 12.0, 10.0$ and 4.0 Hz, 1 H), 3.94 (q, $J = 7.2$ Hz, 2 H), 4.53 (d, $J = 10.0$ Hz, 1 H), 4.77 (d, $J = 10.0$ Hz, 1 H), 6.79 (d, $J = 8.4$ Hz, 1 H), 6.84-6.88 (m, 1 H), 7.14-7.19 (m, 1 H), 7.28-7.43 (m, 6 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 20.7, 27.9, 29.6, 43.9, 50.4, 60.7, 73.8, 78.3, 81.5, 116.9, 120.1, 122.0, 126.6, 127.4, 128.4(2C), 129.2, 140.1, 152.8, 173.2; **IR** (KBr): 2978, 2934, 2855, 1731, 1612, 1458, 1371, 1167, 1122, 1065, 759, 704 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{23}\text{H}_{26}\text{O}_4$ ($\text{M}+\text{H}$) $^+$ requires 367.1909; found 367.1905.

(2S*,3R*,4aR*,10bR*)-Ethyl 2,3,4,4a,5,10b-hexahydro-5,5-dimethyl-2-phenylpyrano[3,2-c]chromene-3-carboxylate (43):



Colourless solid; M.P.: 150-152 $^{\circ}\text{C}$ (202 mg, 55%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.77 (t, $J = 7.2$ Hz, 3 H), 1.23 (s, 3 H), 1.44 (s, 3 H), 1.74 (ddd, $J = 13.2, 12.8$ and 5.2 Hz, 1 H), 2.27 (ddd, $J = 13.2, 3.8$ and 2.0 Hz, 1 H), 2.57(ddd, $J = 12.8, 10.8$ and 3.8 Hz, 1 H), 3.06-3.11 (m, 1 H), 3.67-3.79 (m, 2 H), 4.45 (d, $J = 10.8$ Hz, 1 H), 4.96 (d, $J = 3.2$ Hz, 1 H), 6.80 (d, $J = 8.0$ Hz, 1 H), 6.88-6.92 (m, 1 H), 7.15-7.19 (m, 1 H), 7.23-7.42 (m, 5 H), 7.52 (d, $J = 7.6$ Hz, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 13.8, 20.4, 27.8, 28.1, 39.6, 45.2,

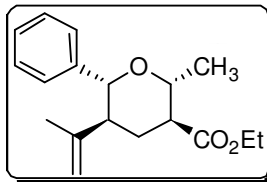
60.0, 74.5, 78.4, 79.2, 116.8, 120.0, 122.3, 125.7, 126.5, 127.3, 128.1, 129.0, 140.6, 153.0, 171.8; **IR** (KBr): 2979, 2942, 2862, 1731, 1607, 1459, 1371, 1172, 1100, 1076, 760, 748 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{23}\text{H}_{26}\text{O}_4$ ($\text{M}+\text{H}$)⁺ requires 367.1909; found 367.1914.



5.7 Selected Spectra of 2,3,5,6-Tetrasubstituted Tetrahydropyrans

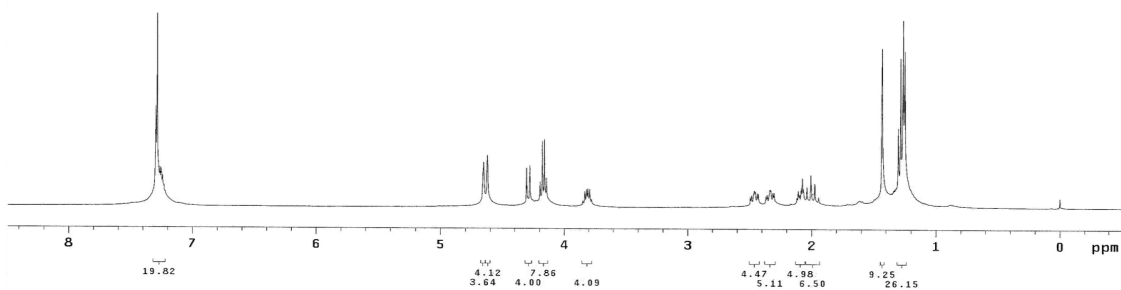
(2*R**,3*S**,5*S**,6*R**)-Ethyl tetrahydro-2-methyl-6-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (32a):

¹H NMR (400 MHz, CDCl₃)



```

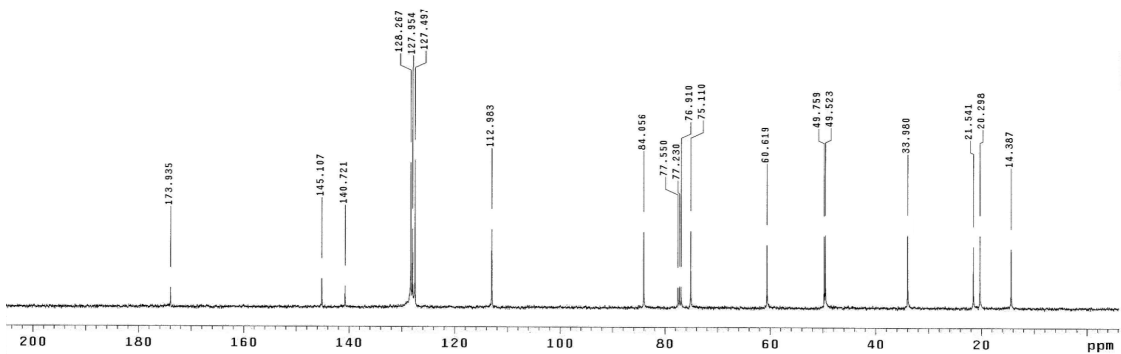
PS_392_U
exp1 s2pu1
SAMPLE
date Nov 26 2010 temp not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftmp/AKS_PS_392- hst 0.008
U_C13_T1d pu90 15.000
ACQUISITION alfa 20.000
SPECIAL
sw 6385.8 alfa FLAGS
at 1.998 f1 n
np 25528 in n
fb not used dp y
bs 4 hs PROCESSING nn
d1 1.000 lb 0.10
nt 32 fb DISPLAY 65536
ct 32 fn
TRANSMITTER H1 sp -298.5
tn 399.853 wp 3602.0
tof 362.8 rf1 796.2
tpwr 57 rfp 0
pw 9.850 rp 106.1
DECOUPLER C13 lp -85.7
dn H1 PLOT
dof 0 wc 250
dm nnn sc 0
dmm c vs 48
dpwr 50 th cdc ph 39
dmf 15900 nm
  
```



¹³C NMR (100 MHz, CDCl₃)

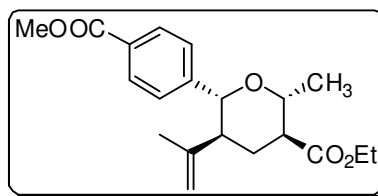
```

PS_392_U
exp1 s2pu1
SAMPLE
date Nov 26 2010 temp not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftmp/AKS_PS_392- hst 0.008
U_C13_T1d pu90 15.000
ACQUISITION alfa 20.000
SPECIAL
sw 25125.8 alfa FLAGS
at 1.199 f1 n
np 60270 in n
fb 13800 dp y
bs 16 hs PROCESSING nn
d1 1.000 lb 2.00
nt 3000 fb DISPLAY 65536
ct 528 fn
TRANSMITTER C13 sp -620.8
tn 180.554 wp 21269.5
tof 1536.3 rf1 9276.7
tpwr 61 rfp 7764.9
pw 9.380 rp -52.1
DECOUPLER H1 lp -305.8
dn H1 PLOT
dof 0 wc 250
dm yyy sc 0
dmm w vs 36
dpwr 42 th no ph 3
dmf 8900 nm
  
```



(2*R,3*S**,5*S**,6*R**)-Ethyl 6-(4-(methoxycarbonyl)phenyl)-tetrahydro-2-methyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (32g):**

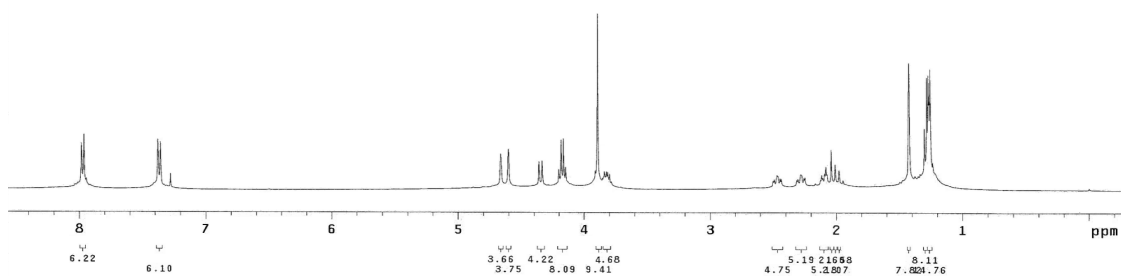
¹H NMR (400 MHz, CDCl₃)



```

PS_338_U
exp1 s2pu1
SAMPLE
date Aug 6 2010 temp SPECIAL
solvent CDCl3 gain not used
file /export/home/~ sp1n not used
quest123/AKS_PS_33~ hst 0.008
8_U.Fid pw90 19.700
ACQUISITION alfa 20.000
sw 6389.8 FLAGS
at 1.998 f1 n
np 25228 f2 n
fb not used dp y
ds 4 hs PROCESSING nn
d1 1.000 PROCESSING
nt 32 lb 0.10
ct 32 f1 DISPLAY 65538
TRANSMITTER DISPLAY
tn H1 sp -108.0
sfrq 399.853 wp 3531.4
tof 362.8 r f1 786.9
tpwr 57 r f2 0
pw 9.850 rp 125.8
DECOUPLER C13 1p PLOT -81.9
dn C13 wc 250
dof 0 nnn sc 0
dmm c vc 42
dpwr 50 th 50
dmf 15900 nm cdc ph

```

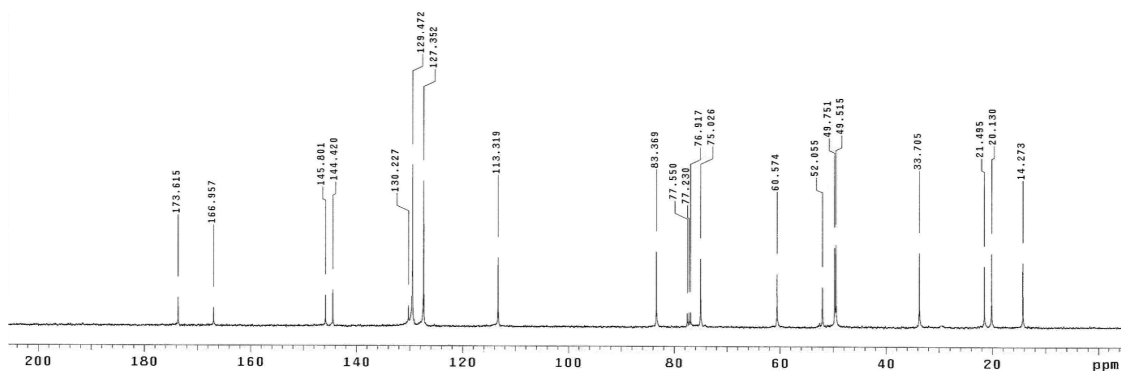


¹³C NMR (100 MHz, CDCl₃)

```

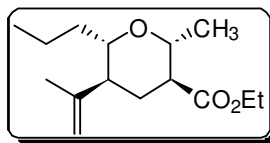
PS_338_U
exp1 s2pu1
SAMPLE
date Aug 6 2010 temp SPECIAL
solvent CDCl3 gain not used
file /export/home/~ sp1n not used
quest123/AKS_PS_33~ hst 0.008
8_U.C13.Fid pw90 18.600
ACQUISITION alfa 20.000
sw 25125.6 FLAGS
at 1.199 f1 n
np 60270 f2 n
fb 13800 dp y
ds 16 hs PROCESSING
d1 1.000 PROCESSING
nt 300 lb 2.00
ct 892 f1 DISPLAY 65538
TRANSMITTER DISPLAY
tn C13 sp -492.7
sfrq 100.554 wp 21163.6
tof 1536.3 r f1 9280.5
tpwr 61 r f2 754.9
pw 9.300 rp -45.6
DECOUPLER H1 1p PLOT -314.3
dn 0 wc 250
dof 0 yyy sc 0
dmm w vs 39
dpwr 42 th 3
dmf 8900 nm no ph

```



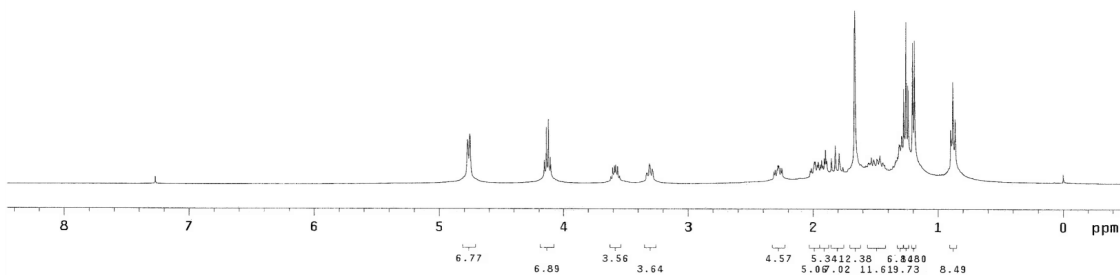
(2*R**,3*S**,5*S**,6*S**)-Ethyl tetrahydro-2-methyl-5-(prop-1-en-2-yl)-6-propyl-2*H*-pyran-3-carboxylate (32h):

¹H NMR (400 MHz, CDCl₃)



```

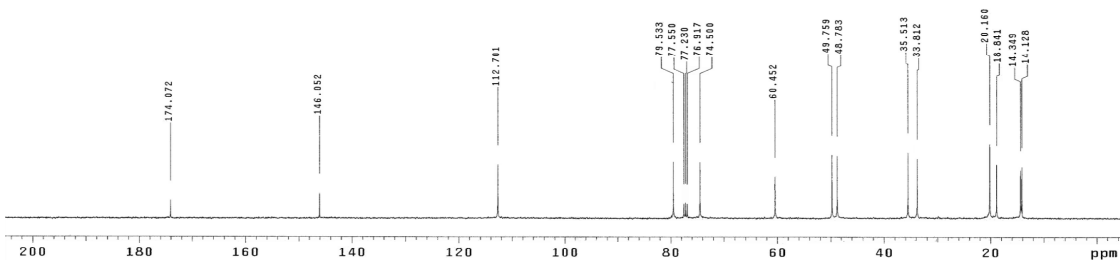
PS_468_U
exp1 s2pu1
SAMPLE          SPECIAL
date Mar 23 2011 temp not used
solvent CDCl3 gain not used
file          exp spin not used
ACQUISITION   hst 0.008
sw 6389.8 pw90 19.700
at 1.998 alfa 20.000
np 25528
Tb not used il n
bs 8 fn n
d1 1.000 dp y
nt 100 hs nn
ct 100
TRANSMITTER   lb 8.10
tn H1 fn 65536
sfrq 399.853 DISPLAY
tof 352.8 sp -194.2
tpwr 57 wp 3576.8
pw 9.850 rfl 790.4
DECOUPLER     C13 rfp 0
dn C13 rfp 105.1
dof 0 ip -55.9
dm n PLOT
dmm c wc 250
dpwr 50 sc 0
daf 15900 vs 40
nm cdc ph 20
  
```

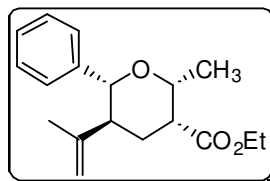


¹³C NMR (100 MHz, CDCl₃)

```

PS_468_U
exp1 s2pu1
SAMPLE          SPECIAL
date Mar 24 2011 temp not used
solvent CDCl3 gain not used
file          exp spin not used
ACQUISITION   hst 0.008
sw 25125.6 pw90 18.600
at 1.998 alfa 20.000
np 60270
Tb 13800 il n
bs 16 fn n
d1 1.000 dp y
nt 200 hs nn
ct 1008
TRANSMITTER   lb 2.00
tn C13 fn 65536
sfrq 100.626 DISPLAY
tof 1536.3 sp -482.8
tpwr 61 wp 21190.6
pw 9.390 rfl 9276.6
DECOUPLER     H1 rfp 7764.9
dn H1 rfp -47.6
dof 0 ip -338.3
dm v/v wc 250
dmm c sc 0
dpwr 8900 vs 19
daf nm no ph 2
  
```

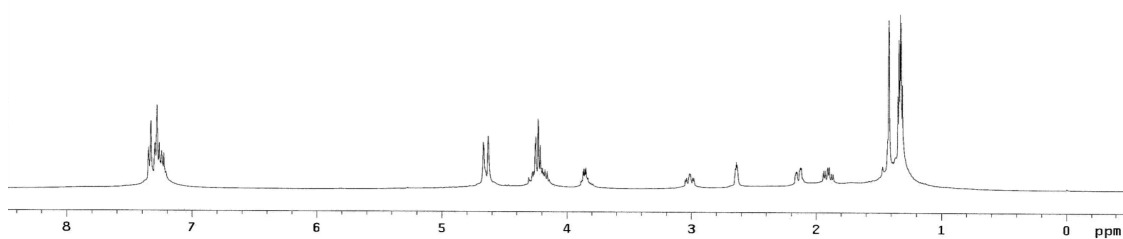


(2*R,3*R**,5*S**,6*R**)-Ethyl tetrahydro-2-methyl-6-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (33a):**¹H NMR (400 MHz, CDCl₃)

```

PS_425_M
expt s2pu1
SAMPLE
date Jan 7 2011 temp SPECIAL
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_425~ hst 0.008
M_C13-71d pw90 19.700
ACQUISITION M_C13-71d alpha 20.000
sw 6389.8 FLAGS
at 1.998 f1 n
np 25528 in n
fb not used dp y
bs 8 hs nn
d1 1.000 PROCESSING 0.10
nt 100 lb 65536
ct 100 fn
TRANSMITTER H1 sp DISPLAY -193.8
tn 399.853 wp 3585.2
sfrq 382.8 rfl 798.4
tpvr 57 rfp 0
pw 9.850 rp 118.5
DECOUPLER lp -98.5
dn C13 PLOT 250
dof 0 wc 0
da nnn sc 0
dmm c vs 43
dpcr 50 sh 31
daf 15900 nm cdc ph

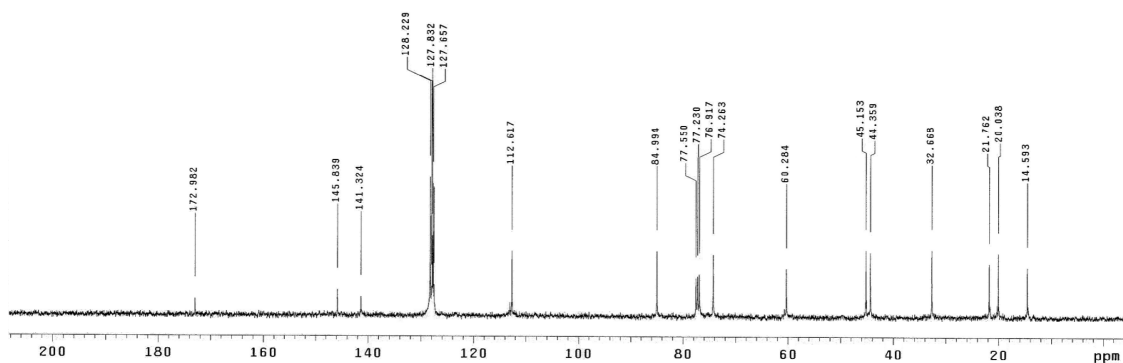
```

¹³C NMR (100 MHz, CDCl₃)

```

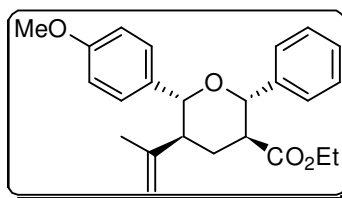
PS_425_M
expt s2pu1
SAMPLE
date Jan 7 2011 temp SPECIAL
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_425~ hst 0.008
M_C13-71d pw90 19.700
ACQUISITION M_C13-71d alpha 20.000
sw 25125.6 FLAGS
at 1.199 f1 n
np 60270 in n
fb 13000 dp y
bs 16 hs nn
d1 1.000 PROCESSING 2.00
nt 400 lb 65536
ct 1152 fn
TRANSMITTER C13 sp DISPLAY -420.7
tn 100.554 wp 21361.4
sfrq 1836.3 rfl 3274.4
tpwr 61 rfp 7764.9
pw 9.300 rp -48.1
DECOUPLER lp -335.6
dn H1 PLOT 250
dof 0 wc 0
da yyy sc 0
dmm w vs 33
dpcr 42 sh 3
daf 8900 nm no ph

```



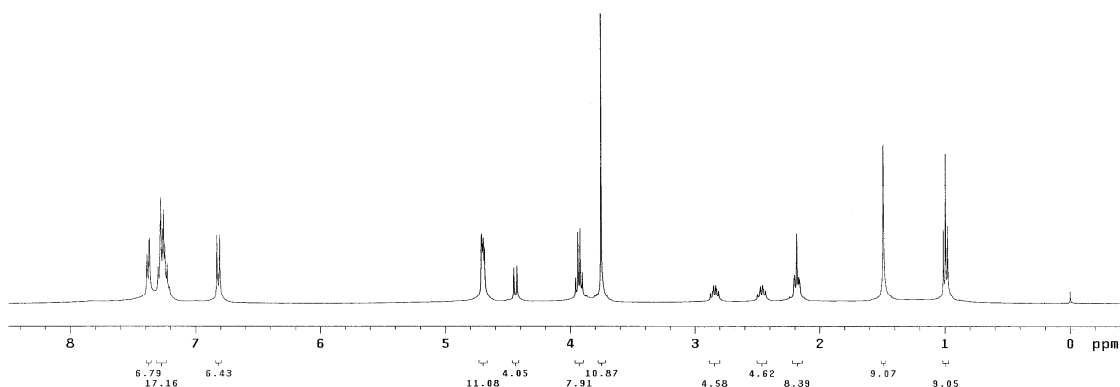
(2*S**,3*S**,5*S**,6*R**)-Ethyl tetrahydro-6-(4-methoxyphenyl)-2-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (40c):

¹H NMR (400 MHz, CDCl₃)



```

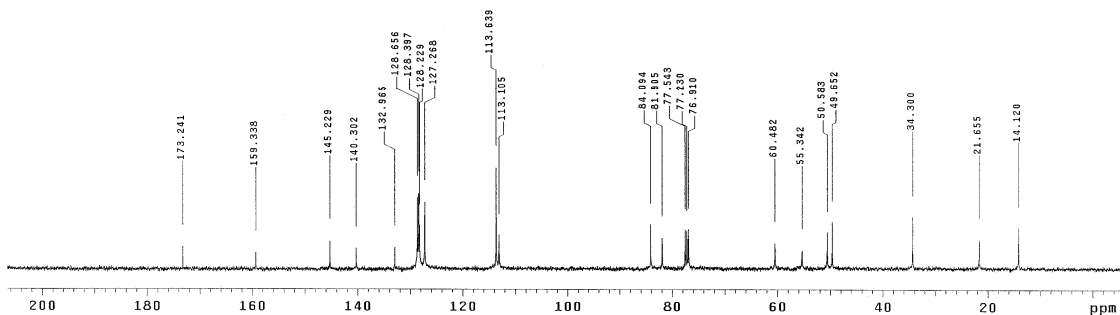
exp1 s2pu1
SAMPLE
date Jun 9 2011 temp not used
solvent CDCl3 gain not used
file /export/home/ spin not used
cifttemp/AKS_PS_494- hct 0.008
M_pdt.fid pw90 18.700
ACQUISITION
sw 6389.8 fl 20.000
at 1.988 fl 0
np 25528 ln n
fb not used dp n
bs 4 hs nn
d1 1.000 PROCESSING
nt 32 lb 9.10
ct 32 fn 65536
TRANSMITTER H1 sp -171.8
tn 399.853 wd 3568.5
tof 362.9 rfl 801.5
tpwr 57 rfp 0
pw 9.850 rp 119.8
DECOUPLER C13 lp -91.4
dn dof 0 wc 250
dm nnn sc 0
dmm c vs 69
dpwr 50 th 42
dmt 15900 nm cdc ph
  
```



¹³C NMR (100 MHz, CDCl₃)

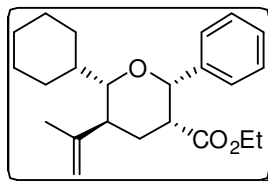
```

PS_494_M
exp1 s2pu1
SAMPLE
date Jun 9 2011 temp not used
solvent CDCl3 gain not used
file /export/home/ spin not used
ACQUISITION
sw 25125.6 pw90 18.600
at 1.199 fl 20.000
np 60270 fl 0
fb 13800 ln n
bs 18 ln n
d1 1.000 dp y
nt 4000 hs nn
ct 1552 PROCESSING
TRANSMITTER lb 2.00
tn 100.554 fn 65536
tof 1536.3 sp -551.8
tpwr 51 wp 21328.5
pw 9.300 rfl 9273.6
DECOUPLER rfp 7764.9
dn dof 0 lp -349.1
dm yvy wc 250
dmm w sc 0
dpwr 42 sc 0
dmt 8900 vs 24
nm no ph
  
```



(2*S**,3*R**,5*S**,6*S**)-Ethyl 6-cyclohexyl-tetrahydro-2-phenyl-5-(prop-1-en-2-yl)-2*H*-pyran-3-carboxylate (41f):

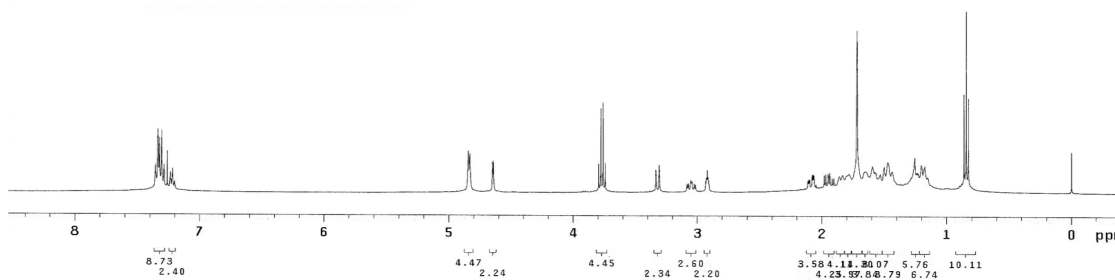
¹H NMR (400 MHz, CDCl₃)



```

PS_452_M
expl s2pu1
SAMPLE
date Feb 23 2011 temp not used
solvent CDCl3 gain not used
file /export/home/~ hst not used
ciftemp/AKS_PS_452~ hst 0.008
ACQUISITION M-71d pw90 19.700
al7a 20.000
sw 6389.8 FLAGS
at 1.998 f1 n
np 2528 in n
fb not used dp y
bs 0 hs nn
d1 1.000 PROCESSING
nt 100 lb 0.10
ct 100 fn 65536
TRANSMITTER DISPLAY
tn H1 sp -181.4
sfrq 399.853 wp 3933.6
tof 362.8 rf1 794.3
tpwr 57 rfp 0
pw 9.850 rp 114.1
DECOUPLER C13 lp -31.7
dn d1
dof 0 wc PLOT 250
dm nnn sc 0
dmm c vs 43
dpwr 50 th 20
def 15900 nm cdc ph

```

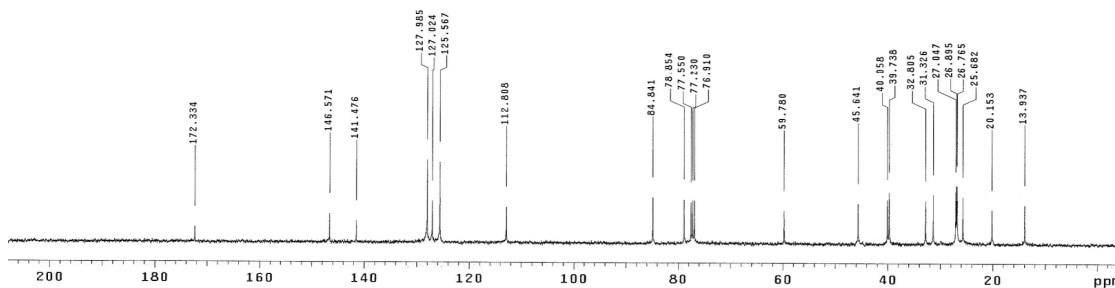


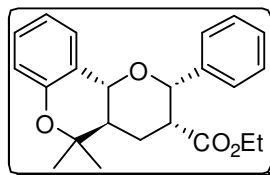
¹³C NMR (100 MHz, CDCl₃)

```

PS_452_M
expl s2pu1
SAMPLE
date Feb 23 2011 temp not used
solvent CDCl3 gain not used
file /export/home/~ hst not used
ciftemp/AKS_PS_452~ hst 0.008
ACQUISITION M-515-71d pw90 19.600
al7a 20.000
sw 25125.6 FLAGS
at 1.199 f1 n
np 60270 in n
fb 13600 dp y
bs 16 hs nn
d1 1.000 PROCESSING
nt 5000 lb 2.00
ct 2080 fn 65536
TRANSMITTER DISPLAY
tn C13 sp -549.5
sfrq 100.624 wp 21460.3
tof 1536.3 rf1 9271.3
tpwr 61 rfp 7764.9
pw 0.390 rp 51.1
DECOUPLER H1 lp -302.7
dn d1
dof 0 wc PLOT 250
dm yyy sc 0
dmm w vs 21
dpwr 42 th 3
def 8900 nm no ph

```

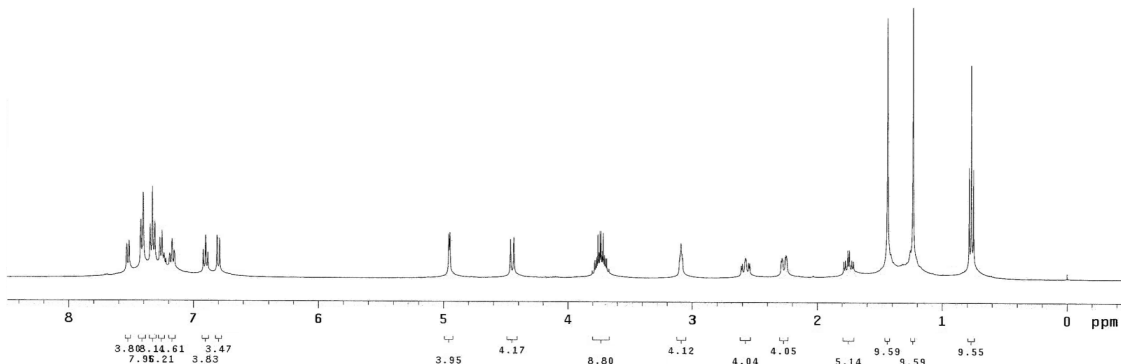


(2*S,3*R**,4*aR**,10*bR**)-Ethyl****2,3,4,4*a*,5,10*b*-hexahydro-5,5-dimethyl-2-****phenylpyrano[3,2-*c*]chromene-3-carboxylate (43):**¹H NMR (400 MHz, CDCl₃)

```

PS_433_L
exp1 s2pul
SAMPLE
date Feb 14 2011 temp SPECIAL not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_433- hst 0.008
l1_G13_rfid pw90 19.700
ACQUISITION alfa 20.000
sw 6389.8 FLAGS
at 1.998 f1 n
np 25528 f2 n
fb not used dp y
bs 4 hs
d1 1.000 PROCESSING
nt 32 lb 0.19
ct 32 fn 65536
TRANSMITTER H1 sp DISPLAY
tn 399.853 wp 3575.3
tof 362.8 rf1 810.4
tpwr 57 rfp 0
pw 9.850 rp 129.6
DECOUPLER lp -97.3
dn C13 PLOT
dof 0 wc 250
dm nnn sc 0
dmm c vs 82
dpwr 50 th cdc ph 20
dmf 15900 nm

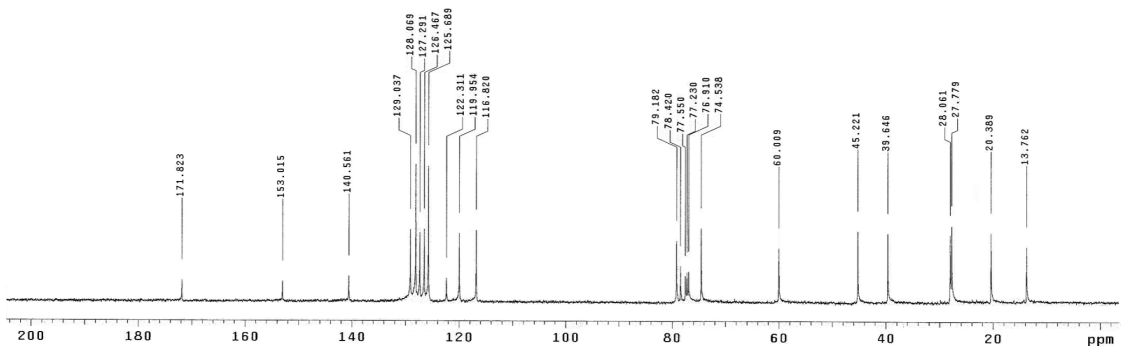
```

¹³C NMR (100 MHz, CDCl₃)

```

PS_433_L1
exp1 s2pul
SAMPLE
date Feb 21 2011 temp SPECIAL not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_433- hst 0.008
l1_G13_rfid pw90 19.400
ACQUISITION alfa 20.000
sw 25125.6 FLAGS
at 1.199 f1 n
np 60270 f2 n
fb 13800 dp y
bs 16 hs
d1 1.000 PROCESSING
nt 3000 lb 2.00
ct 3000 fn 65536
TRANSMITTER H1 sp DISPLAY
tn 100.534 wp 20832.0
tof 1536.3 rf1 9275.0
tpwr 61 rfp 7764.9
pw 9.300 rp -77.5
DECOUPLER lp -269.1
dn H1 PLOT
dof 0 wc 250
dm yyy sc 0
dmm w vs 84
dpwr 42 th no ph 4
dmf 8900 nm

```



5.8 Crystal Parameters

The crystal parameters of compound **33g**

Parameters	33g -CCDC 819363
Formula	C ₂₀ H ₂₆ O ₅
Formula weight	346.41
<i>T</i> /K	296(2)
Crystal system	Orthorhombic
Space group	P n a 21
<i>a</i> /Å	19.8370(10)
<i>b</i> /Å	15.2026(8)
<i>c</i> /Å	6.3744(4)
<i>α</i> °	90.00
<i>β</i> °	90.00
<i>γ</i> °	90.00
<i>V</i> /Å ³	1922.35(18)
<i>Z</i>	4
Abs. Coeff./mm ⁻¹	0.085
Abs. Correction	multi-scan
GOF on <i>F</i> ²	0.943
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> 1 = 0.0677 <i>wR</i> 2 = 0.1644
<i>R</i> indices [all data]	<i>R</i> 1 = 0.0856 <i>wR</i> 2 = 0.1735

The crystal parameters of compound **40f**

Parameters	40f -CCDC 819365
Formula	C ₂₅ H ₂₈ O ₅
Formula weight	408.47
<i>T</i> /K	296(2)
Crystal system	Monoclinic
Space group	P 21/c
<i>a</i> /Å	12.0752(3)
<i>b</i> /Å	15.3970(4)
<i>c</i> /Å	12.4647(3)
α ^o	90.00
β ^o	104.3910(10)
γ ^o	90.00
<i>V</i> /Å ³	2244.74(10)
<i>Z</i>	4
Abs. Coeff./mm ⁻¹	0.083
Abs. Correction	multi-scan
GOF on <i>F</i> ²	1.070
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> 1 = 0.0615 <i>wR</i> 2 = 0.1806
<i>R</i> indices [all data]	<i>R</i> 1 = 0.0759 <i>wR</i> 2 = 0.2203

The crystal parameters of compound **42**

Parameters	42-CCDC 819364
Formula	C ₂₃ H ₂₆ O ₄
Formula weight	366.44
<i>T</i> /K	296(2)
Crystal system	Monoclinic
Space group	P 21/n
<i>a</i> /Å	8.9913(7)
<i>b</i> /Å	19.6654(14)
<i>c</i> /Å	11.4452(9)
<i>α</i> [°]	90.00
<i>β</i> [°]	103.239(4)
<i>γ</i> [°]	90.00
<i>V</i> /Å ³	1969.9(3)
<i>Z</i>	4
Abs. Coeff./mm ⁻¹	0.084
Abs. Correction	Multi-scan
GOF on <i>F</i> ²	1.064
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> 1 = 0.0667 <i>wR</i> 2 = 0.1815
<i>R</i> indices [all data]	<i>R</i> 1 = 0.0923 <i>wR</i> 2 = 0.1973

The crystal parameters of compound **43**

Parameters	43 -CCDC 819366
Formula	C ₂₃ H ₂₆ O ₄
Formula weight	366.44
<i>T</i> /K	296(2)
Crystal system	Monoclinic
Space group	P 21/c
<i>a</i> /Å	12.5282(8)
<i>b</i> /Å	8.7884(6)
<i>c</i> /Å	18.2013(12)
α °	90.00
β °	94.582(2)
γ °	90.00
<i>V</i> /Å ³	1997.6(2)
<i>Z</i>	4
Abs. Coeff./mm ⁻¹	0.082
Abs. Correction	Multi-scan
GOF on <i>F</i> ²	1.373
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> 1 = 0.0613 <i>wR</i> 2 = 0.1695
<i>R</i> indices [all data]	<i>R</i> 1 = 0.0967 <i>wR</i> 2 = 0.2108

CHAPTER 6

Diastereoselective Synthesis of Substituted Dihydropyrans via Oxonium-Ene Cyclization Reaction**6.1 Importance and Applications**

Dihydropyrans are important intermediates for the synthesis of biologically active natural products such as ambruticin, phorbaxozoles, leucascandrolide, kendomycin, neopeltolide, clavisolides and diospongins.¹ The 4-methyl substituted dihydropyran unit is present in macrolide natural products laulimalide and okadaic acid (Figure 6.1.1).² The olefin functionality of dihydropyrans can be manipulated for the synthesis of polysubstituted tetrahydropyrans.³ Substituted dihydropyrans are also used as a flavoring or aroma material for food and other products.⁴

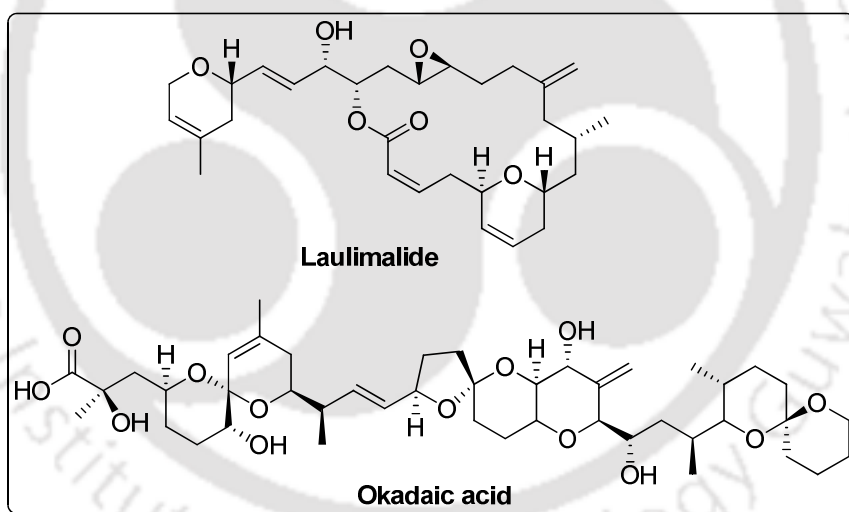


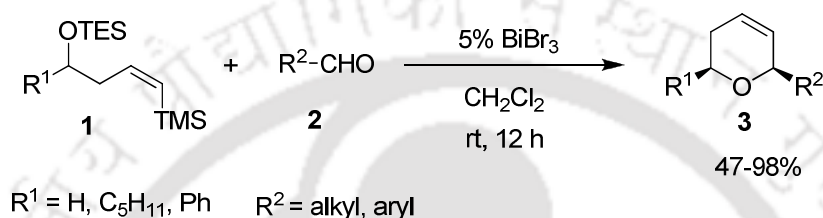
Figure 6.1.1. Biologically active natural products containing 4-methyl dihydropyran as core unit

6.2 An Overview of Relevant Synthetic Methods

Review of literature reveals that dihydropyrans are generally prepared by hetero Diels-Alder reactions, olefin metathesis, Prins cyclization or base-promoted cyclizations of sulfinyl dienols. Among these Prins cyclization is the most convenient because it provides the desired product in a single step. Usually Lewis acid is used for this reaction and in most of the cases, the product is a 4-halosubstituted dihydropyran, but the synthetic access to 4-alkyldihydropyrans is limited in the literature.

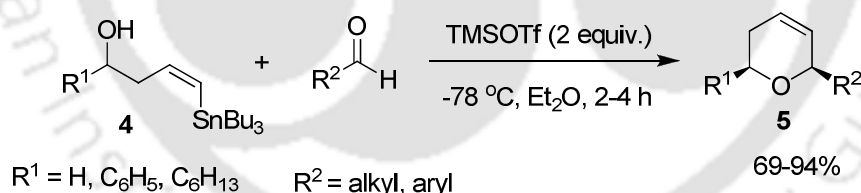
6.2.1 Prins Cyclization

Hinkle and co-workers have reported a tandem silyl-Prins reaction between δ -triethylsilyloxyvinyltrimethylsilanes and a variety of aldehydes to afford *cis*-2,6-disubstituted dihydropyrans (DHPs) using 5 mol % of BiBr₃ in CH₂Cl₂ (Scheme 6.2.1.1).⁵ The diastereoselectivities in the crude products were significantly affected by aldehyde substitution with electron-rich aldehydes, providing 2-3:1 (*cis:trans*) and neutral (or electron-poor) aldehydes affording dr \geq 19:1 (*cis:trans*).



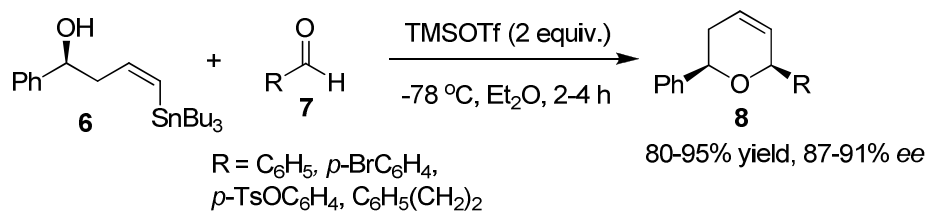
Scheme 6.2.1.1.

Furman and co-workers have reported a stereoselective synthesis of *cis*-2,6-disubstituted dihydropyrans (DHPs) via stannyl-Prins cyclization. The reaction of vinylstannanes **4** with aldehydes in the presence of trimethylsilyl trifluoromethanesulfonate (TMSOTf) afforded *cis*-2,6-disubstituted dihydropyrans **5** in good yields with excellent stereoselectivity (Scheme 6.2.1.2).⁶ Although the dihydropyrans are obtained in the racemic form but the use



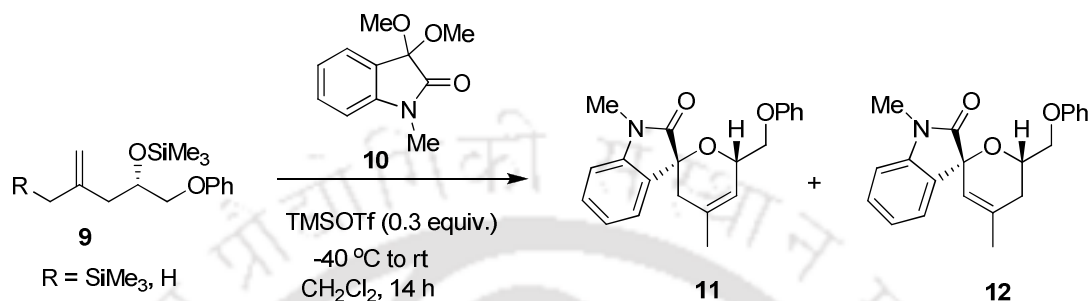
Scheme 6.2.1.2.

of optically pure vinylstannane afforded optically pure 2,6-disubstituted dihydropyrans (Scheme 6.2.1.3). Finally they applied this methodology for the synthesis of natural product (-)-centrolobine.⁶



Scheme 6.2.1.3.

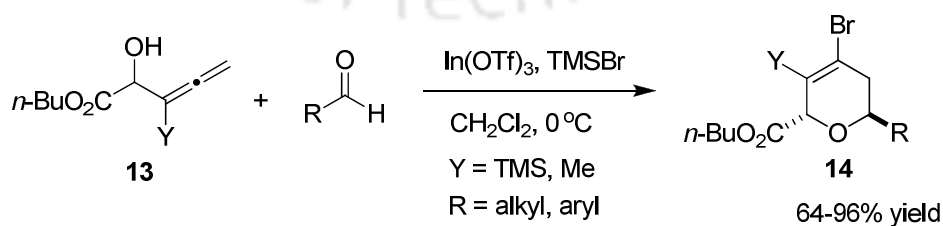
Porco, Jr. and co-workers have reported the synthesis of spirocyclic oxindole pyrans having 4-methyl substituted dihydropyran unit. These spirocyclic compounds are prepared from the reaction of homoallyl alcohols **9** with isatin ketals **10** in the presence of trimethylsilyl trifluoromethanesulfonate (Scheme 6.2.1.4).⁷



Entry	R	Additive	Yield
1	SiMe ₃	DBMP (0.1 equiv.)	11 = 38%, 12 = 19%
2	H	DBMP (0.1 equiv.)	11 = 46%, 12 = 20%
3	H	None	11 = 58%, 12 = 24%

Scheme 6.2.1.4.

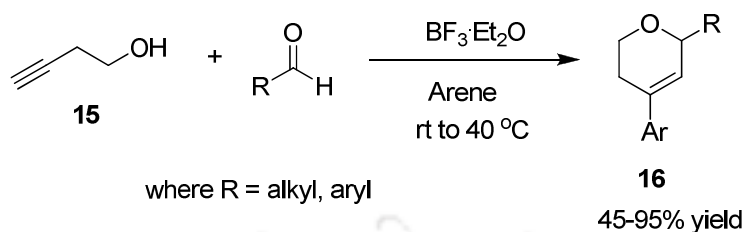
Loh and co-workers have reported the synthesis of 2,6-*trans* dihydropyrans from the reaction of allenic alcohols and aldehydes in the presence of indium salt catalyst in good yields.⁸ However a good diastereoselectivity was obtained by using bulky-silicon substituted allenic alcohols under the same condition (Scheme 6.2.1.5). The reason of high *trans* diastereoselectivity is due to the anomeric effect as well as lone pair stabilization of the δ^+ of the oxo-carbenium intermediate by the ester group.⁸



Scheme 6.2.1.5.

Saikia and co-workers have reported the synthesis of 4-aryl-5,6-dihydro-2*H*-pyrans from the reaction of carbonyl compounds or epoxides with homopropargyl alcohol and arenes

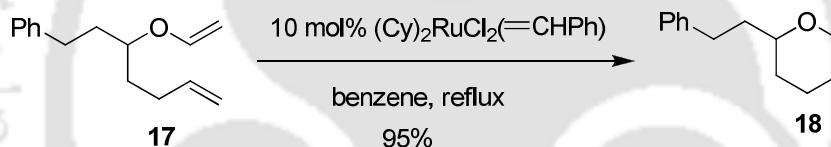
mediated by boron trifluoride etherate by using alkyne-Prins-Friedel-Crafts reaction (Scheme 6.2.1.6).⁹



Scheme 6.2.1.6.

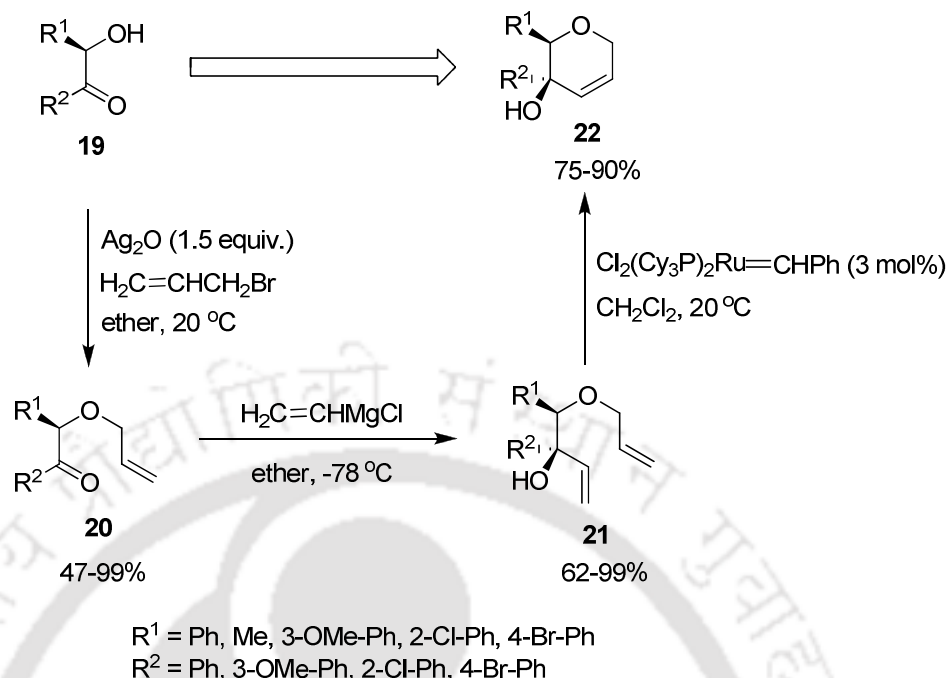
6.2.2 Olefin Metathesis

Sturino *et al.* have reported the synthesis of dihydropyrans (DHPs) by ring closing metathesis (RCM) of acyclic vinyl ethers with Grubb's catalyst (Scheme 6.2.2.1).¹⁰ They found that vinyl ethers are excellent substrates for the ring closing metathesis (RCM) with ruthenium based catalysts.



Scheme 6.2.2.1.

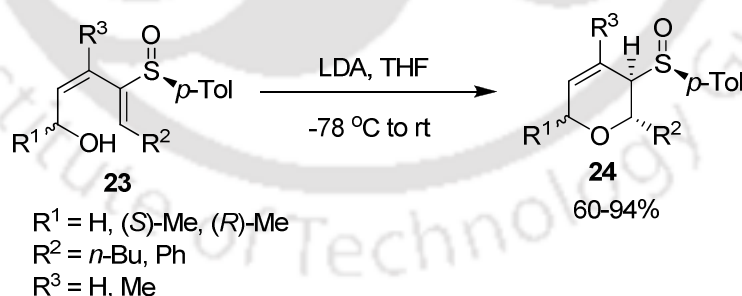
Schmidt and co-workers have reported a highly diastereoselective route to enantiomerically pure dihydropyrans.¹¹ The synthetic concept is based on the use of *R*-hydroxy ketones which are conveniently obtained in enantiomerically pure form, a highly diastereoselective vinylation relying on efficient chelate control and a ringclosing olefin metathesis step (Scheme 6.2.2.2).



Scheme 6.2.2.2.

6.2.3 Base-promoted Cyclization of Sulfinyl Dienols

de la Pradilla *et al.* have reported a base-promoted intramolecular cyclization of 2-sulfinyl dienols to afford sulfinyl dihydropyrans.¹² This strategy allows for creation of two asymmetric centers within a synthetically useful dihydropyran framework in an expedient manner (Scheme 6.2.3.1).

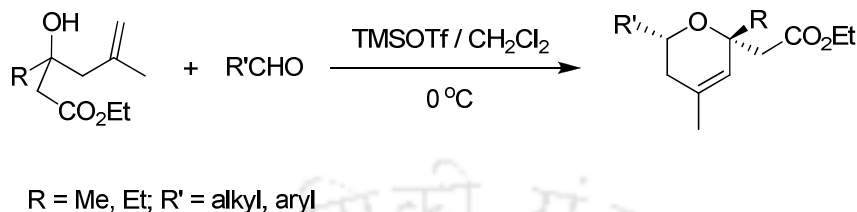


Scheme 6.2.3.1.

6.3 Present Work

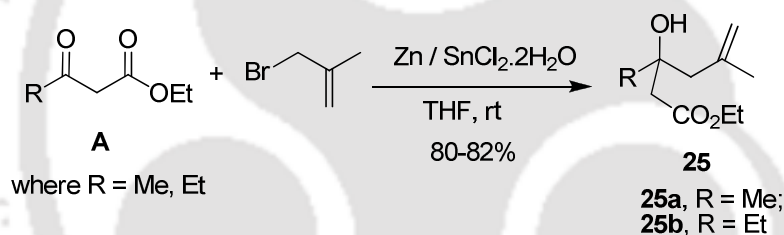
In the previous chapter we have described an efficient methodology for the synthesis of 2,3,5,6-tetrasubstituted tetrahydropyrans via (3,5)-oxonium-ene reaction. In this chapter we are going to present a diastereo- and regioselective synthesis of substituted dihydropyrans from the reaction of aldehydes or epoxides and ethyl 3-alkyl-3-hydroxy-5-methylhex-5-

enoate using oxonium-ene reaction catalyzed by trimethylsilyl trifluoromethanesulfonate (TMSOTf) under mild conditions. The reaction can be generalized as shown in Scheme 6.3.1.



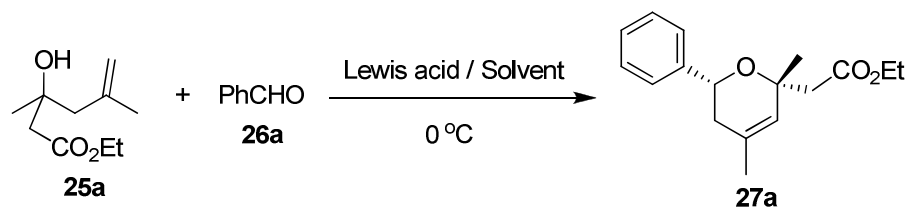
Scheme 6.3.1.

The ethyl 3-alkyl-3-hydroxy-5-methylhex-5-enoate **25** was synthesized starting from β -keto ester **A** and 3-bromo-2-methylpropene as shown below.¹³ The reaction of β -keto ester **A** with 3-bromo-2-methylpropene in presence of metallic Zn and SnCl₂·2H₂O in THF gives alcohol **25** in 80-82% yields (Scheme 6.3.2).



Scheme 6.3.2. Synthesis of ethyl 3-alkyl-3-hydroxy-5-methylhex-5-enoate

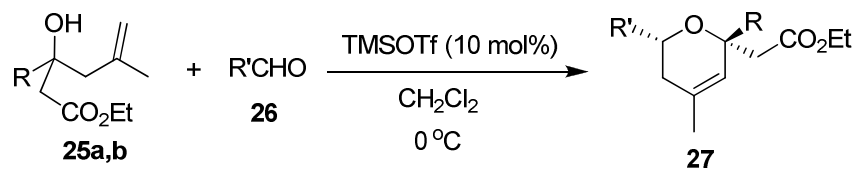
To determine the optimum reaction conditions, the reaction of benzaldehyde **26a** and ethyl 3-hydroxy-3,5-dimethylhex-5-enoate **25a** with a varied amount of different Lewis acids with variation of solvents have been studied. The results are summarized in Table 6.3.1. It was found that the reaction proceeds smoothly with one equivalent of boron trifluoride etherate in dry CH₂Cl₂ yielding 68% of desired product, whereas only 10 mol% of trimethylsilyl trifluoromethanesulfonate produces 78% of same product in the same solvent. A low yield was obtained with bismuth triflate (0.1 equiv.) in dry CH₂Cl₂ and no reaction was observed with indium triflate (In(OTf)₃) (0.1 equiv.) in both CH₂Cl₂ and toluene. Therefore, TMSOTf in dry CH₂Cl₂ was considered as the best combination for this reaction and the scope of the reaction was investigated by using different types of aliphatic and aromatic aldehydes (Table 6.3.2). Both aliphatic and aromatic aldehydes gave the corresponding products in good yields. The reaction is diastereoselective as well as

Table 6.3.1. Optimization of reaction conditions

Entry	Lewis acid (equiv.)	Time / h	Solvent	Yield ^a (%)
1	BF ₃ ·Et ₂ O (1)	1	CH ₂ Cl ₂	68
2	BF ₃ ·Et ₂ O (1)	1	Toluene	64
3	BF ₃ ·Et ₂ O (1)	24	CH ₃ CN	45
4	TMSOTf (0.1)	1	CH ₂ Cl ₂	78
5	TMSOTf (0.1)	1	Toluene	70
6	Bi(OTf) ₃ (0.1)	24	CH ₂ Cl ₂	40
7	Bi(OTf) ₃ (0.1)	24	Toluene	35
8	In(OTf) ₃ (0.1)	24	Toluene	0
9	In(OTf) ₃ (0.1)	24	CH ₂ Cl ₂	0

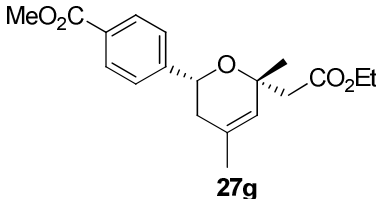
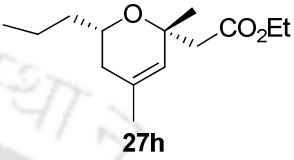
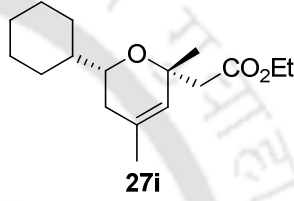
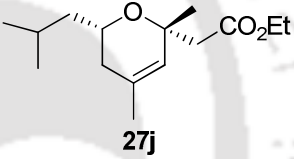
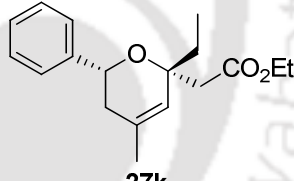
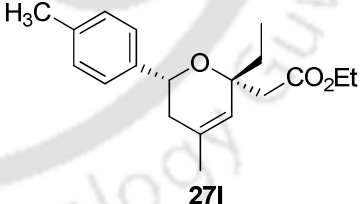
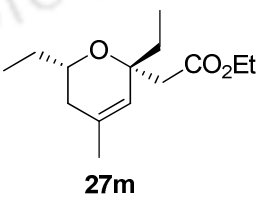
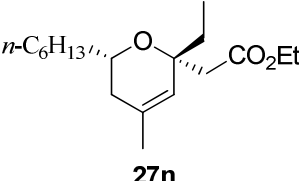
^aYields refer to isolated yield. The compound is characterized by IR, ¹H, ¹³C NMR and Mass spectroscopy.

regioselective and in all the cases studied, substituted dihydropyrans **27a-27n** could be obtained in high purity without any other diastereo- and regioisomeric products as determined from ¹H and ¹³C NMR spectrum of the crude product. The substituents on the aromatic ring have a promising effect on this reaction. The aromatic aldehydes having electron-donating groups on the aromatic ring gave higher yields compared to the electron-withdrawing groups on the ring. On the other hand, aliphatic aldehydes are better substrates than the aromatic aldehydes. This might be due to the better stability imparted to the oxocarbenium ion **29** (Scheme 6.3.3) by the aliphatic, and aromatic aldehydes having electron-donating groups on the aromatic ring, which in turn is attacked by the double bond efficiently. On the other hand, aromatic aldehydes having electron-withdrawing groups on the ring destabilize the oxocarbenium ion **29**. The structure and stereochemistry of the compound was determined by ¹H NMR and NOE experiment. In the ¹H NMR spectrum the proton at position C-6 of **27a** shows a dd with coupling constants 10.4 and 3.2 Hz at δ 4.71 ppm, which indicates that the proton is coupled with two neighbouring diastereotropic

Table 6.3.2. Synthesis of substituted dihydropyrans

Entry	Alcohol 25a , R = Me 25b , R = Et	Aldehyde R' =	Time / h	Product 27	Yield ^a (%)
a	Me	C ₆ H ₅ -	1		78
b	Me	<i>p</i> -Me-C ₆ H ₄ -	1		82
c	Me	<i>p</i> -Br-C ₆ H ₄ -	1		65
d	Me	<i>p</i> -Cl-C ₆ H ₄ -	1		64
e	Me	<i>p</i> -F-C ₆ H ₄ -	1		62
f	Me	<i>p</i> -NO ₂ -C ₆ H ₄ -	1		62

continue...

Entry	Alcohol 25a , R = Me 25b , R = Et	Aldehyde R' =	Time / h	Product 27	Yield ^a (%)
g	Me	<i>p</i> -MeO ₂ C-C ₆ H ₄ -	1		68
h	Me	C ₃ H ₇ -	0.5		90
i	Me	Cyclo-C ₆ H ₁₁ -	0.5		84
j	Me	(CH ₃) ₂ CHCH ₂ -	0.5		86
k	Et	C ₆ H ₅ -	1		74
l	Et	<i>p</i> -Me-C ₆ H ₄ -	1.5		80
m	Et	C ₂ H ₅ -	0.8		88
n	Et	<i>n</i> -C ₆ H ₁₃ -	0.8		82

^aYields refer to isolated yield. The compounds are characterized by IR, ¹H, ¹³C, ¹⁹F NMR and Mass spectroscopy.

methylene protons H_{5a} and H_{5b} . On the other hand proton H_{5a} shows a dd with coupling constants 16.8 and 3.2 Hz at δ 2.01 ppm. Similarly proton H_{5b} also shows a dd with coupling constants 16.8 and 10.4 Hz at δ 2.13 ppm. This indicates that these two protons are coupling each other and neighbouring H_6 proton. Thus the position of the double bond between C-3 and C-4 carbons has been confirmed. The presence of a strong NOE between H_6 and C-2 methyl protons of **27a** clearly indicates that they are *cis* to each other (Figure 6.3.1).

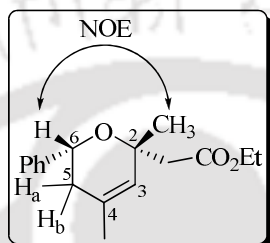
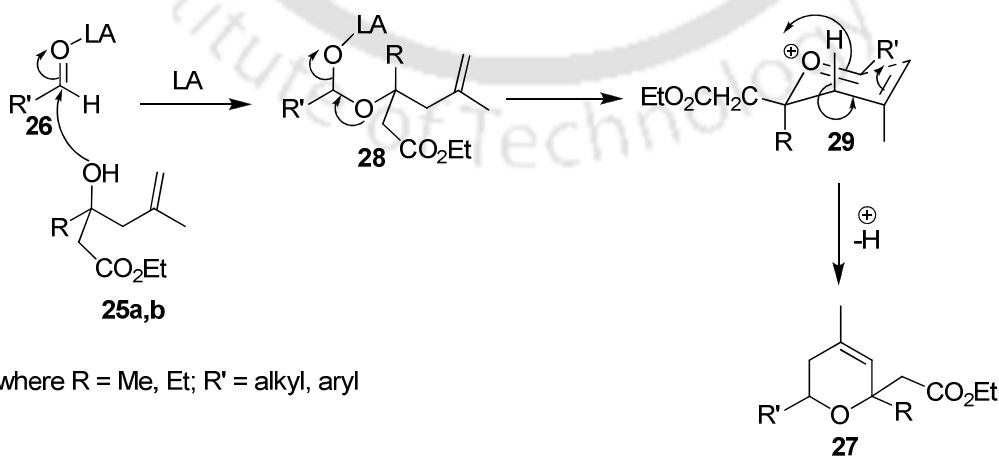


Figure 6.3.1. NOE diagram of compound **27a**

The mechanism of the reaction can be explained as follows. Lewis acid activates the aldehyde **26** for the nucleophilic attack by alcohol **25** to give acetal **28**, which after decomposition forms oxocarbenium ion **29**. The oxocarbenium ion **29** undergoes oxonium-ene cyclization reaction to give final product **27** (Scheme 6.3.3). The formation of single regioisomer supports the concerted mechanism. On the other hand formation of single diastereomer may be explained as follows. After decomposition of acetal **28**, it forms six membered chair like transition state **29**, in which the more bulkier methylene ester group occupies the equatorial position to minimize the 1,3-diaxial interaction.



Scheme 6.3.3. Mechanism of the reaction

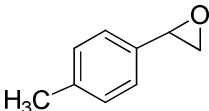
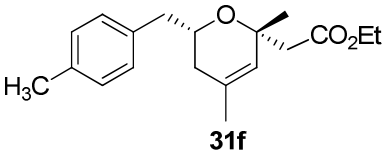

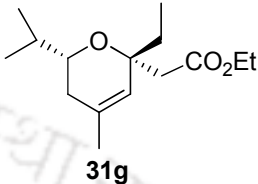
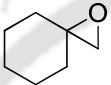
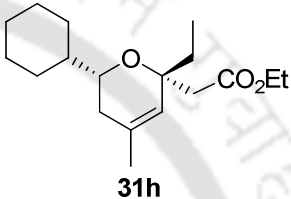
The same substituted dihydropyrans can also be prepared by using epoxides (Table 6.3.3). This is due to the fact that epoxides under Lewis acid conditions rearrange to aldehyde equivalent.¹⁴ Here monosubstituted terminal epoxides (entry a) fail to give substituted dihydropyran, whereas the 2,2-disubstituted and styrene oxides give the corresponding products in good to moderate yields. This is attributed to the lower stability of the carbocation **32**, obtained from monosubstituted epoxides, compared to 2,2-disubstituted epoxides and styrene oxides, where carbocation **32** is better stabilized due to tertiary and benzylic centers, respectively.^{14a}

Table 6.3.3. Synthesis of substituted dihydropyran from epoxide

Reaction scheme: Alcohol **25a,b** (with R and CO₂Et groups) reacts with epoxide **30** (with R' and R'' groups) in the presence of TMSOTf (10 mol%) in CH₂Cl₂ at 0°C to yield substituted dihydropyran **31**.

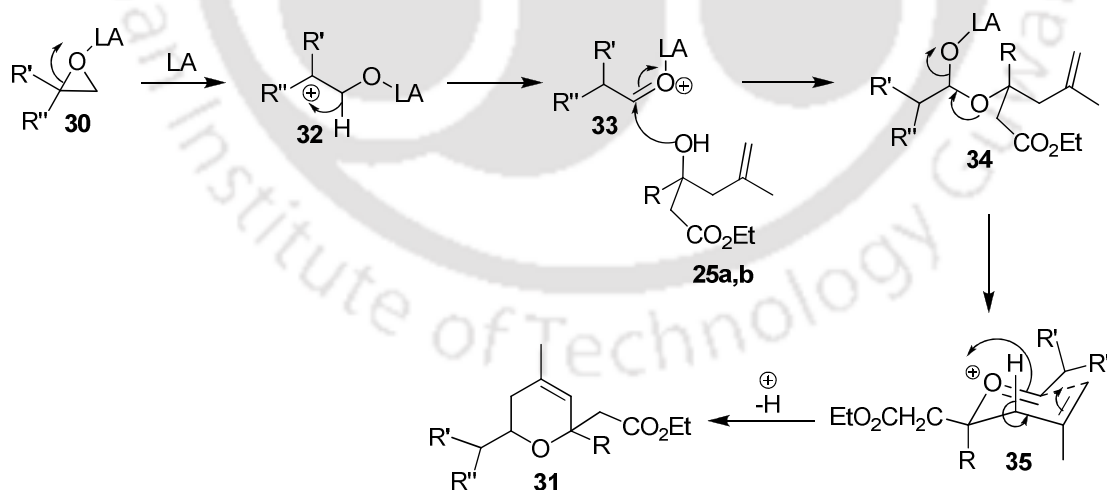
Entry	Alcohol 25a , R = Me 25b , R = Et	Epoxide	Time / h	Product 31	Yield ^a (%)
a	Me		24		0
b	Me		1.5		68
c	Me		1.5		64
d	Me		1.5		60
e	Me		1.5		62

continue...

Entry	Alcohol 25a , R = Me 25b , R = Et	Epoxide	Time / h	Product 31	Yield ^a (%)
f	Me		1.5		64
g	Et		2		65
h	Et		2		62

^aYields refer to isolated yield. The compounds are characterized by IR, ¹H, ¹³C NMR and Mass spectroscopy.

The mechanism of the reaction with epoxides is shown in Scheme 6.3.4. The epoxide **30** in the presence of Lewis acid rearranges to aldehyde equivalent **33**, which after nucleophilic attack by alcohol **25** gives acetal **34**. The acetal **34** decomposes to give oxocarbenium ion **35**, which after oxonium-ene cyclization reaction gives final product **31**.



Scheme 6.3.4. Mechanism of the reaction with epoxides

Conclusions:

In summary, we have developed a simple and efficient methodology for the synthesis of substituted dihydropyrans in good yields. The reaction is diastereoselective as well as

regioselective. The presence of methylene ester functionality in the C-2 position of the dihydropyran ring will increase its synthetic applicability as the ester group could further be converted to other groups.

6.4 Experimental Section

6.4.1 Instrumentation and Characterization

As described in Chapter 2, Section 2.4.1.

6.4.2 General Procedure for the Synthesis of Ethyl 3-alkyl-3-hydroxy-5-methylhex-5-enoate (25a-b):

A 250 mL round-bottom flask was charged with metallic Zn (20 mmol, 2 equiv.), SnCl₂·2H₂O (10 mmol, 1 equiv.) and 20 mL of THF. To this solution β -keto ester (10 mmol, 1 equiv.) and 3-bromo-2-methylpropene (15 mmol, 1.5 equiv.) were added sequentially and the mixture was stirred at room temperature for 24 h. The reaction mixture was then filtered and the filtrate was washed with ethyl acetate (2 X 20 mL). The combined filtrate was concentrated and purified by column chromatography over silica gel to afford the title compounds as colourless oil.

Synthesis of Ethyl 3-hydroxy-3,5-dimethylhex-5-enoate (25a): A 250 mL round-bottom flask was charged with metallic Zn (1.31 g, 20 mmol), SnCl₂·2H₂O (2.25 g, 10 mmol) and 20 mL of THF. To this solution ethyl acetoacetate (1.30 g, 10 mmol) and 3-bromo-2-methylpropene (2.02 g, 15 mmol) were added sequentially and the mixture was stirred at room temperature for 24 h. The reaction mixture was then filtered and the filtrate was washed with ethyl acetate (2 X 20 mL). The combined filtrate was concentrated and purified by column chromatography over silica gel (10% EtOAc/hexane) to afford ethyl 3-hydroxy-3,5-dimethylhex-5-enoate **25a** (1.53 g, 82%) as a colourless oil.

6.4.3 General Procedure for the Synthesis of Substituted Dihydropyrans

from Aldehydes: To a solution of aldehyde (1 mmol, 1 equiv.) in dry CH₂Cl₂ (2 mL) at 0 °C was added trimethylsilyl trifluoromethanesulfonate (0.1 mmol, 0.1 equiv.). To this solution ethyl 3-alkyl-3-hydroxy-5-methylhex-5-enoate (1 mmol, 1 equiv.) in dry CH₂Cl₂ (2 mL) was added drop by drop over 5 min. The reaction mixture was stirred at that temperature for a specified time. The progress of the reaction was monitored by TLC with ethyl acetate and hexane as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with CH₂Cl₂ (30 mL) and then washed with water (15 mL) and brine (15 mL). The organic layer was dried (Na₂SO₄) and

evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give the title compounds.

Synthesis of Ethyl 2-((2*R,6*R**)-5,6-dihydro-2,4-dimethyl-6-phenyl-2*H*-pyran-2-yl)acetate (27a):** To a solution of benzaldehyde (106 mg, 1.00 mmol) in dry CH₂Cl₂ (2 mL) at 0 °C was added trimethylsilyl trifluoromethanesulfonate (22 mg, 0.1 mmol). To this solution ethyl 3-hydroxy-3,5-dimethylhex-5-enoate (186 mg, 1 mmol) in dry CH₂Cl₂ (2 mL) was added drop by drop over 5 min. The reaction mixture was stirred at that temperature for 1 h. The progress of the reaction was monitored by TLC with ethyl acetate and hexane (1:24) as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with CH₂Cl₂ (30 mL) and then washed with water (15 mL) and brine (15 mL). The organic layer was dried (Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give **27a** (214 mg, 78 %) as a colourless oil.

6.4.4 General Procedure for the Synthesis of Substituted Dihydropyrans

from Epoxides: To a solution of epoxide (1.5 mmol, 1.5 equiv.) in dry CH₂Cl₂ (2 mL) at 0 °C was added trimethylsilyl trifluoromethanesulfonate (0.1 mmol, 0.1 equiv.). To this solution ethyl 3-alkyl-3-hydroxy-5-methylhex-5-enoate (1 mmol, 1 equiv.) in dry CH₂Cl₂ (2 mL) was added drop by drop over 5 min. The reaction mixture was stirred at that temperature for a specified time. The progress of the reaction was monitored by TLC with ethyl acetate and hexane as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with CH₂Cl₂ (30 mL) and then washed with water (15 mL) and brine (15 mL). The organic layer was dried (Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give the title compounds.

Synthesis of Ethyl 2-((2*R,6*R**)-5,6-dihydro-6-isopropyl-2,4-dimethyl-2*H*-pyran-2-yl)acetate (31b):** To a solution of 1,2-epoxy-2-methyl propane (108 mg, 1.5 mmol) in dry CH₂Cl₂ (2 mL) at 0 °C was added trimethylsilyl trifluoromethanesulfonate (22 mg, 0.1 mmol). To this solution ethyl 3-hydroxy-3,5-dimethylhex-5-enoate (186 mg, 1 mmol) in dry CH₂Cl₂ (2 mL) was added drop by drop over 5 min. The reaction mixture was stirred at that temperature for 1.5 h. The progress of the reaction was monitored by TLC with ethyl acetate and hexane (1:49) as eluents. After completion of the reaction, it was quenched with saturated solution of NaHCO₃. The product was extracted with CH₂Cl₂ (30 mL) and then washed with water (15 mL) and brine (15 mL). The organic layer was dried

(Na₂SO₄) and evaporated to leave the crude product, which was purified by short column chromatography over silica gel to give **31b** (163 mg, 68%) as a colourless oil.

6.5 References

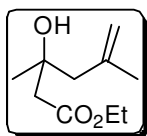
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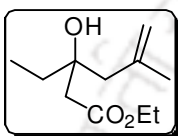
6.6 Spectral Data

Ethyl 3-hydroxy-3,5-dimethylhex-5-enoate (25a):



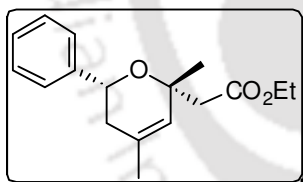
Colourless oil (1.53 g, 82%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.27 (s, 3 H), 1.29 (t, $J = 7.2$ Hz, 3 H), 1.85 (t, $J = 1.2$ Hz, 3 H), 2.27 (brs, 2 H), 2.43 (d, $J = 15.6$ Hz, 1 H), 2.56 (d, $J = 15.6$ Hz, 1 H), 3.68 (brs, 1 H), 4.19 (q, $J = 7.2$ Hz, 2 H), 4.70-4.74 (m, 1 H), 4.89-4.92 (m, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.1, 24.5, 27.3, 44.8, 49.6, 60.5, 71.1, 115.0, 142.4, 173.1; **IR** (Neat) 3482, 2980, 2935, 1716, 1642, 1374, 1190, 1095, 1031 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{10}\text{H}_{18}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 187.1334; found 187.1328.

Ethyl 3-ethyl-3-hydroxy-5-methylhex-5-enoate (25b):



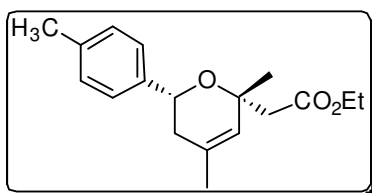
Colourless oil (1.6 g, 80%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.92 (t, $J = 7.2$ Hz, 3 H), 1.28 (t, $J = 7.2$ Hz, 3 H), 1.55-1.63 (m, 2 H), 1.85 (s, 3 H), 2.22 (d, $J = 13.6$ Hz, 1 H), 2.28 (d, $J = 13.6$ Hz, 1 H), 2.44 (d, $J = 15.6$ Hz, 1 H), 2.52 (d, $J = 15.6$ Hz, 1 H), 3.64 (brs, 1 H), 4.18 (q, $J = 7.2$ Hz, 2 H), 4.72 (brs, 1 H), 4.90 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 8.4, 14.3, 24.7, 32.8, 42.3, 46.8, 60.8, 73.6, 115.2, 142.7, 173.7; **IR** (Neat) 3504, 2977, 2940, 1715, 1642, 1372, 1186, 1095, 1029 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{11}\text{H}_{20}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 201.1490; found 201.1486.

Ethyl 2-((2*R**,6*R**)-5,6-dihydro-2,4-dimethyl-6-phenyl-2*H*-pyran-2-yl)acetate (27a):



Colourless oil (214 mg, 78 %); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.27 (t, $J = 7.2$ Hz, 3 H), 1.41 (s, 3 H), 1.74 (s, 3 H), 2.01 (dd, $J = 16.8$ and 3.2 Hz, 1 H), 2.13 (dd, $J = 16.8$ and 10.4 Hz, 1 H), 2.59 (d, $J = 13.6$ Hz, 1 H), 2.65 (d, $J = 13.6$ Hz, 1 H), 4.10-4.18 (m, 2 H), 4.71 (dd, $J = 10.4$ and 3.2 Hz, 1 H), 5.55 (brs, 1 H), 7.26-7.29 (m, 1 H), 7.32-7.36 (m, 2 H), 7.38-7.42 (m, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.4, 23.1, 25.2, 37.9, 47.6, 60.2, 70.8, 74.5, 126.0, 126.2, 127.4, 128.4, 131.7, 143.1, 170.7; **IR** (Neat) 2971, 2927, 1734, 1674, 1450, 1367, 1211, 1094, 1076, 752, 699 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{17}\text{H}_{22}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 275.1647; found 275.1642.

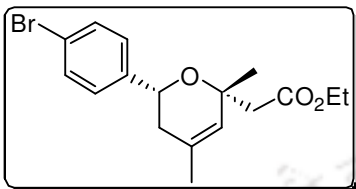
Ethyl 2-((2*R**,6*R**)-5,6-dihydro-2,4-dimethyl-6-*p*-tolyl-2*H*-pyran-2-yl)acetate (27b):



Colourless oil (236 mg, 82%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.27 (t, $J = 7.2$ Hz, 3 H), 1.40 (s, 3 H), 1.73 (s, 3 H), 1.97 (dd, $J = 16.8$ and 3.2 Hz, 1 H), 2.12 (dd, $J = 16.8$ and 10.8 Hz, 1 H), 2.33 (s, 3 H), 2.58 (d, $J = 13.6$ Hz, 1 H), 2.63 (d, $J = 13.6$ Hz, 1 H), 4.10-4.18 (m, 2 H), 4.67 (dd, $J = 10.8$ and 3.2 Hz, 1 H), 5.54 (brs, 1 H), 7.15 (d, $J = 8.0$ Hz, 2 H), 7.28 (d, $J = 8.0$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz,

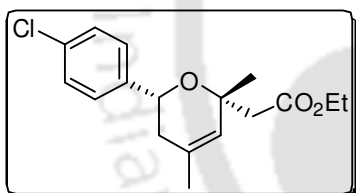
CDCl₃): δ 14.4, 21.2, 23.1, 25.1, 37.9, 47.6, 60.1, 70.7, 74.5, 126.0, 126.2, 129.0, 131.7, 136.9, 140.1, 170.7; **IR** (Neat) 2973, 2926, 1734, 1687, 1446, 1367, 1215, 1089, 1033, 808, 716 cm⁻¹. **HRMS** (APCI) cald. for C₁₈H₂₄O₃ (M+H)⁺ requires 289.1803; found 289.1790.

Ethyl 2-((2*R,6*R**)-6-(4-bromophenyl)-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (27c):**



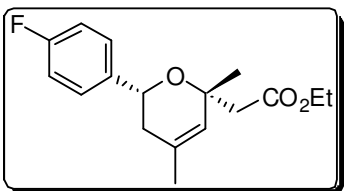
Colourless oil (230 mg, 65%); **¹H NMR** (400 MHz, CDCl₃): δ 1.26 (t, *J* = 7.2 Hz, 3 H), 1.39 (s, 3 H), 1.74 (s, 3 H), 1.98 (dd, *J* = 16.4 and 3.6 Hz, 1 H), 2.06 (dd, *J* = 16.4 and 10.4 Hz, 1 H), 2.56 (d, *J* = 13.6 Hz, 1 H), 2.63 (d, *J* = 13.6 Hz, 1 H), 4.09-4.18 (m, 2 H), 4.66 (dd, *J* = 10.8 and 3.2 Hz, 1 H), 5.54 (brs, 1 H), 7.27 (d, *J* = 8.4 Hz, 2 H), 7.46 (d, *J* = 8.4 Hz, 2 H); **¹³C NMR** (100 MHz, CDCl₃): δ 14.4, 23.1, 25.2, 37.8, 47.6, 60.3, 70.2, 74.6, 121.1, 126.3, 127.8, 131.5 (2C), 142.2, 170.6; **IR** (Neat) 2974, 2927, 1734, 1679, 1445, 1367, 1217, 1088, 1033, 815, 710 cm⁻¹. **HRMS** (APCI) cald. for C₁₇H₂₁BrO₃ (M+H)⁺ requires 353.0754; found 353.0742 (⁷⁹Br).

Ethyl 2-((2*R,6*R**)-6-(4-chlorophenyl)-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (27d):**



Colourless oil (198 mg, 64%); **¹H NMR** (400 MHz, CDCl₃): δ 1.26 (t, *J* = 7.2 Hz, 3 H), 1.39 (s, 3 H), 1.74 (s, 3 H), 1.97 (dd, *J* = 16.8 and 3.6 Hz, 1 H), 2.05 (dd, *J* = 16.4 and 10.4 Hz, 1 H), 2.58 (d, *J* = 13.6 Hz, 1 H), 2.63 (d, *J* = 13.6 Hz, 1 H), 4.08-4.18 (m, 2 H), 4.67 (dd, *J* = 10.4 and 3.6 Hz, 1 H), 5.54 (brs, 1 H), 7.28-7.34 (m, 4 H); **¹³C NMR** (100 MHz, CDCl₃): δ 14.4, 23.1, 25.2, 37.8, 47.6, 60.3, 70.2, 74.7, 126.3, 127.4, 128.5, 131.5, 133.0, 141.7, 170.7; **IR** (Neat) 2963, 2926, 1735, 1685, 1445, 1368, 1217, 1087, 1033, 817, 692 cm⁻¹. **HRMS** (APCI) cald. for C₁₇H₂₁ClO₃ (M+H)⁺ requires 309.1257; found 309.1247 (³⁵Cl).

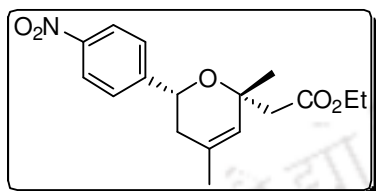
Ethyl 2-((2*R,6*R**)-6-(4-fluorophenyl)-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (27e):**



Colourless oil (181 mg, 62%); **¹H NMR** (400 MHz, CDCl₃): δ 1.26 (t, *J* = 7.2 Hz, 3 H), 1.40 (s, 3 H), 1.74 (s, 3 H), 1.98 (dd, *J* = 16.8 and 3.2 Hz, 1 H), 2.06-2.13 (dd, *J* = 16.8 and 10.8 Hz, 1 H), 2.58 (d, *J* = 13.6 Hz, 1 H), 2.64 (d, *J* = 13.6 Hz, 1 H), 4.09-4.18 (m, 2 H), 4.69 (dd, *J* = 10.8 and 3.6 Hz, 1 H), 5.54 (brs, 1 H), 7.02 (t, *J* = 8.8 Hz, 2 H), 7.33-7.37 (m, 2 H); **¹³C NMR** (100 MHz, CDCl₃): δ 14.5, 23.1, 25.3, 38.0, 47.7, 60.3, 70.4, 74.7, 115.2 (d, *J* = 21.3 Hz), 126.3, 127.7 (d, *J* = 8.4 Hz), 131.6, 139.1,

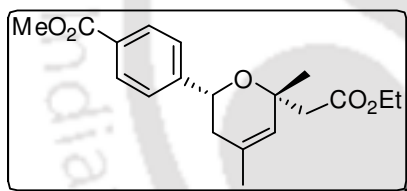
162.2 (d, $J = 243.3$ Hz), 170.7; ^{19}F NMR (376 MHz, $\text{CDCl}_3/\text{C}_6\text{F}_6$): δ -219.7-(-219.6) (m, 1 F); IR (Neat) 2977, 2930, 1734, 1685, 1511, 1444, 1368, 1223, 1086, 1033, 826, 774 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{FO}_3$ (M+H) $^+$ requires 293.1553; found 293.1545.

Ethyl 2-((2*R,6*R**)-5,6-dihydro-2,4-dimethyl-6-(4-nitrophenyl)-2*H*-pyran-2-yl)acetate (27f):**



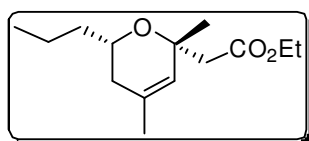
Colourless oil (198 mg, 62%); ^1H NMR (400 MHz, CDCl_3): δ 1.27 (t, $J = 7.2$ Hz, 3 H), 1.41 (s, 3 H), 1.76 (s, 3 H), 2.02-2.09 (m, 2 H), 2.60 (d, $J = 13.6$ Hz, 1 H), 2.66 (d, $J = 13.6$ Hz, 1 H), 4.10-4.19 (m, 2 H), 4.82 (t, $J = 7.2$ Hz, 1 H), 5.57 (brs, 1 H), 7.56 (d, $J = 8.8$ Hz, 2 H), 8.20 (d, $J = 8.8$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.4, 23.0, 25.1, 37.6, 47.4, 60.3, 70.0, 74.8, 123.6, 126.4, 126.6, 131.1, 147.2, 150.6, 170.5; IR (Neat) 2975, 2928, 1733, 1605, 1521, 1446, 1374, 1212, 1091, 1032, 855, 749 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{17}\text{H}_{21}\text{NO}_5$ (M+H) $^+$ requires 320.1498; found 320.1485.

Methyl 4-((2*R,6*R**)-6-((ethoxycarbonyl)methyl)-3,6-dihydro-4,6-dimethyl-2*H*-pyran-2-yl)benzoate (27g):**

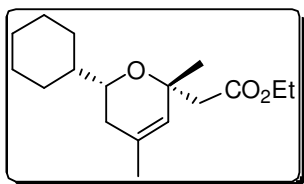


Colourless oil (226 mg, 68%); ^1H NMR (400 MHz, CDCl_3): δ 1.27 (t, $J = 7.2$ Hz, 3 H), 1.41 (s, 3 H), 1.75 (s, 3 H), 1.99-2.10 (m, 2 H), 2.60 (d, $J = 13.6$ Hz, 1 H), 2.66 (d, $J = 13.6$ Hz, 1 H), 3.91 (s, 3 H), 4.10-4.19 (m, 2 H), 4.77 (dd, $J = 10.0$ and 4.0 Hz, 1 H), 5.55 (brs, 1 H), 7.46 (d, $J = 8.4$ Hz, 2 H), 8.01 (d, $J = 8.4$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.4, 23.1, 25.1, 37.7, 47.5, 52.1, 60.3, 70.4, 74.6, 125.9, 126.3, 129.1, 129.8, 131.4, 148.4, 167.1, 170.6; IR (Neat) 2953, 2927, 1725, 1613, 1436, 1368, 1278, 1090, 1033, 768, 704 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{19}\text{H}_{24}\text{O}_5$ (M+H) $^+$ requires 333.1702; found 333.1692.

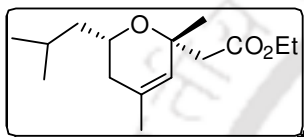
Ethyl 2-((2*R,6*S**)-5,6-dihydro-2,4-dimethyl-6-propyl-2*H*-pyran-2-yl)acetate (27h):**



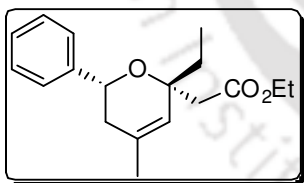
Colourless oil (216 mg, 90%); ^1H NMR (400 MHz, CDCl_3): δ 0.91 (t, $J = 7.2$ Hz, 3 H), 1.24 (t, $J = 7.2$ Hz, 3 H), 1.30 (s, 3 H), 1.35-1.43 (m, 2 H), 1.45-1.54 (m, 2 H), 1.62-1.72 (m, 4 H), 1.82 (dd, $J = 16.0$ and 10.8 Hz, 1 H), 2.49 (brs, 2 H), 3.61-3.66 (m, 1 H), 4.11 (q, $J = 7.2$ Hz, 2 H), 5.45 (brs, 1 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.2, 14.4, 18.7, 23.2, 24.6, 35.6, 38.0, 47.7, 60.1, 68.2, 73.6, 126.4, 131.7, 170.8; IR (Neat) 2961, 2930, 2873, 1736, 1678, 1456, 1370, 1216, 1081, 1034, 825 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{14}\text{H}_{24}\text{O}_3$ (M+H) $^+$ requires 241.1803; found 241.1794.

Ethyl 2-((2*R,6*R**)-6-cyclohexyl-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (27i):**

Colourless oil (236 mg, 84%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.93-1.04 (m, 2 H), 1.11-1.21 (m, 2 H), 1.24 (t, $J = 7.2$ Hz, 3 H), 1.28 (s, 3 H), 1.34-1.42 (m, 2 H), 1.65-1.76 (m, 7 H), 1.83-1.96 (m, 3 H), 2.47 (brs, 2 H), 3.33-3.38 (m, 1 H), 4.06-4.15 (m, 2 H), 5.43 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.4, 23.4, 24.4, 26.2, 26.3, 26.8, 28.5, 29.1, 32.6, 42.5, 47.7, 60.1, 72.6, 73.5, 126.5, 131.8, 170.9; **IR** (Neat) 2967, 2925, 2852, 1736, 1679, 1449, 1367, 1216, 1097, 1034, 833 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{17}\text{H}_{28}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 281.2116; found 281.2112.

Ethyl 2-((2*R,6*S**)-5,6-dihydro-6-isobutyl-2,4-dimethyl-2*H*-pyran-2-yl)acetate (27j):**

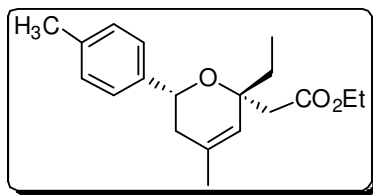
Colourless oil (219 mg, 86%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.89 (d, $J = 6.4$ Hz, 3 H), 0.90 (d, $J = 6.4$ Hz, 3 H), 1.24 (t, $J = 7.2$ Hz, 3 H), 1.31 (s, 3 H), 1.44-1.52 (m, 2 H), 1.65-1.72 (m, 4 H), 1.78-1.86 (m, 2 H), 2.48 (brs, 2 H), 3.67-3.74 (m, 1 H), 4.12 (q, $J = 7.2$ Hz, 2 H), 5.45 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.4, 22.4, 23.3, 23.4, 24.5 (2C), 36.2, 45.1, 47.7, 60.2, 66.6, 73.6, 126.5, 131.8, 171.0; **IR** (Neat) 2957, 2927, 2867, 1736, 1654, 1458, 1369, 1215, 1086, 1034, 971 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{15}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 255.1960; found 255.1954.

Ethyl 2-((2*R,6*R**)-2-ethyl-5,6-dihydro-4-methyl-6-phenyl-2*H*-pyran-2-yl)acetate (27k):**

Colourless oil (213 mg, 74%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.93 (t, $J = 7.2$ Hz, 3 H), 1.26 (t, $J = 7.2$ Hz, 3 H), 1.64-1.72 (m, 1 H), 1.76 (s, 3 H), 1.77-1.86 (m, 1 H), 2.01 (dd, $J = 16.8$ and 3.2 Hz, 1 H), 2.09-2.18 (dd, $J = 16.8$ and 10.8 Hz, 1 H), 2.56 (d, $J = 13.6$ Hz, 1 H), 2.63 (d, $J = 13.6$ Hz, 1 H), 4.06-4.17 (m, 2 H), 4.68 (dd, $J = 10.4$ and 3.2 Hz 1 H), 5.59 (brs, 1 H), 7.24-7.28 (m, 1 H), 7.32-7.36 (m, 2 H), 7.38-7.42 (m, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 8.3, 14.5, 23.3, 30.7, 37.7, 44.2, 60.2, 70.8, 76.8, 125.7, 126.1, 127.4, 128.4, 132.2, 143.3, 171.1; **IR** (Neat) 2970, 2927, 2854, 1734, 1676, 1450, 1369, 1207, 1095, 1031, 752, 698 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{18}\text{H}_{24}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 289.1803; found 289.1794.

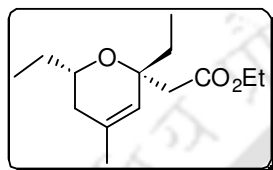
Ethyl 2-((2*R,6*R**)-2-ethyl-5,6-dihydro-4-methyl-6-*p*-tolyl-2*H*-pyran-2-yl)acetate (27l):**

Colourless oil (242 mg, 80%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.92 (t, $J = 7.2$ Hz, 3 H), 1.26 (t, $J = 7.2$ Hz, 3 H), 1.63-1.72 (m, 1 H), 1.75 (s, 3 H), 1.77-1.84 (m, 1 H), 1.98 (dd, $J = 16.8$ and 3.2 Hz, 1 H), 2.09-2.17 (dd, $J = 16.8$ and 10.4 Hz, 1 H), 2.34 (s, 3 H), 2.55 (d, $J =$



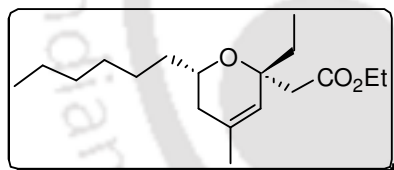
13.6 Hz, 1 H), 2.61 (d, $J = 13.6$ Hz, 1 H), 4.07-4.17 (m, 2 H), 4.64 (dd, $J = 10.4$ and 3.2 Hz, 1 H), 5.58 (brs, 1 H), 7.15 (d, $J = 7.6$ Hz, 2 H), 7.28 (d, $J = 8.0$ Hz, 2 H); ^{13}C NMR (100 MHz, CDCl_3): δ 8.3, 14.5, 21.3, 23.3, 30.7, 37.7, 44.2, 60.2, 70.7, 76.7, 125.7, 126.1, 129.1, 132.2, 137.0, 140.3, 171.1; IR (Neat) 2969, 2926, 2856, 1734, 1679, 1445, 1368, 1206, 1088, 1033, 809, 668 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{19}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 303.1960; found 303.1948.

Ethyl 2-((2R*,6S*)-2,6-diethyl-5,6-dihydro-4-methyl-2H-pyran-2-yl)acetate (27m):



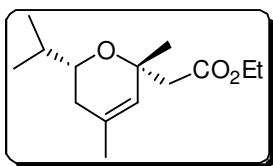
Colourless oil (212 mg, 88%); ^1H NMR (400 MHz, CDCl_3): δ 0.90 (t, $J = 7.6$ Hz, 3 H), 0.95 (t, $J = 7.6$ Hz, 3 H), 1.23 (t, $J = 7.2$ Hz, 3 H), 1.41-1.65 (m, 3 H), 1.69 (s, 3 H), 1.70-1.85 (m, 3 H), 2.48 (brs, 2 H), 3.48-3.54 (m, 1 H), 4.07-4.15 (m, 2 H), 5.50 (brs, 1 H); ^{13}C NMR (100 MHz, CDCl_3): δ 8.0, 10.2, 14.5, 23.4, 29.0, 29.8, 35.4, 43.9, 60.2, 69.8, 75.7, 126.0, 131.9, 171.2; IR (Neat) 2965, 2929, 2878, 1735, 1676, 1463, 1378, 1208, 1074, 1037, 941 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{14}\text{H}_{24}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 241.1803; found 241.1798.

Ethyl 2-((2R*,6S*)-2-ethyl-6-hexyl-5,6-dihydro-4-methyl-2H-pyran-2-yl)acetate (27n):

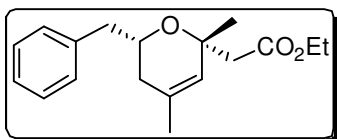


Colourless oil (243 mg, 82%); ^1H NMR (400 MHz, CDCl_3): δ 0.88 (t, $J = 6.8$ Hz, 3 H), 0.90 (t, $J = 7.6$ Hz, 3 H), 1.23 (t, $J = 7.2$ Hz, 3 H), 1.25-1.35 (m, 8 H), 1.37-1.47 (m, 2 H), 1.51-1.62 (m, 2 H), 1.69 (s, 3 H), 1.71-1.75 (m, 1 H), 1.78-1.85 (m, 1 H), 2.48 (brs, 2 H), 3.52-3.60 (m, 1 H), 4.10 (q, $J = 7.2$ Hz, 2 H), 5.50 (brs, 1 H); ^{13}C NMR (100 MHz, CDCl_3): δ 7.97, 14.3, 14.4, 22.8, 23.4, 25.7, 29.5, 29.7, 32.1, 35.8, 36.2, 43.8, 60.2, 68.4, 75.8, 126.0, 132.0, 171.2; IR (Neat) 2959, 2928, 2856, 1736, 1676, 1461, 1368, 1205, 1091, 1033, 869 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{18}\text{H}_{32}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 297.2429; found 297.2418.

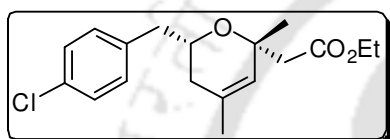
Ethyl 2-((2R*,6R*)-5,6-dihydro-6-isopropyl-2,4-dimethyl-2H-pyran-2-yl)acetate (31b):



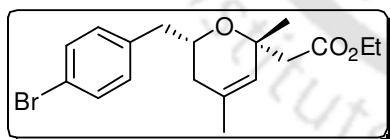
Colourless oil (163 mg, 68%); ^1H NMR (400 MHz, CDCl_3): δ 0.89 (d, $J = 6.8$ Hz, 3 H), 0.94 (d, $J = 6.8$ Hz, 3 H), 1.24 (t, $J = 7.2$ Hz, 3 H), 1.29 (s, 3 H), 1.64-1.72 (m, 5 H), 1.82-1.89 (m, 1 H), 2.48 (brs, 2 H), 3.30-3.36 (m, 1 H), 4.06-4.17 (m, 2 H), 5.44 (brs, 1 H); ^{13}C NMR (100 MHz, CDCl_3): δ 14.4, 18.2, 18.7, 23.4, 24.5, 32.4, 32.8, 47.7, 60.1, 73.4, 73.6, 126.5, 131.7, 171.0; IR (Neat) 2965, 2929, 1736, 1676, 1448, 1368, 1211, 1081, 1035, 841 cm^{-1} . HRMS (APCI) cald. for $\text{C}_{14}\text{H}_{24}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 241.1803; found 241.1794.

Ethyl 2-((2*R,6*R**)-6-benzyl-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (31c):**

Colourless oil (185 mg, 64%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.24 (t, $J = 7.2$ Hz, 3 H), 1.29 (s, 3 H), 1.60-1.67 (m, 4 H), 1.84-1.91 (m, 1 H), 2.51 (brs, 2 H), 2.69 (dd, $J = 13.6$ and 6.8 Hz, 1 H), 2.95 (dd, $J = 13.6$ and 6.8 Hz, 1 H), 3.81-3.87 (m, 1 H), 4.11 (q, $J = 7.2$ Hz, 2 H), 5.42 (brs, 1 H), 7.20-7.30 (m, 5 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.4, 23.2, 24.7, 35.0, 42.5, 47.6, 60.2, 69.6, 74.0, 126.2, 126.4, 128.2, 129.6, 131.4, 138.8, 170.8; **IR** (Neat) 2974, 2927, 1734, 1674, 1453, 1369, 1214, 1086, 1032, 751, 700 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{18}\text{H}_{24}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 289.1803; found 289.1797.

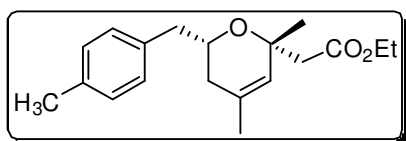
Ethyl 2-((2*R,6*R**)-6-(4-chlorobenzyl)-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (31d):**

Colourless oil (194 mg, 60%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.23 (t, $J = 7.2$ Hz, 3 H), 1.29 (s, 3 H), 1.61-1.68 (m, 4 H), 1.83-1.90 (m, 1 H), 2.51 (brs, 2 H), 2.68 (dd, $J = 14.0$ and 6.8 Hz, 1 H), 2.86 (dd, $J = 14.0$ and 6.8 Hz, 1 H), 3.81-3.88 (m, 1 H), 4.11 (q, $J = 7.2$ Hz, 2 H), 5.43 (brs, 1 H), 7.17 (d, $J = 8.4$, 2 H), 7.24 (d, $J = 8.8$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.4, 23.2, 24.6, 35.0, 41.7, 47.6, 60.3, 69.3, 74.0, 126.5, 128.3, 131.0, 131.3, 132.0, 137.4, 170.8; **IR** (Neat) 2972, 2927, 1734, 1676, 1492, 1369, 1214, 1091, 1033, 822, 798 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{18}\text{H}_{23}\text{ClO}_3$ ($\text{M}+\text{H}$) $^+$ requires 323.1414; found 323.1404 (^{35}Cl).

Ethyl 2-((2*R,6*R**)-6-(4-bromobenzyl)-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (31e):**

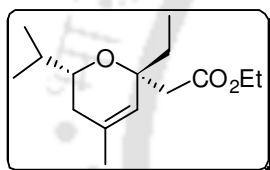
Colourless oil (228 mg, 62%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.23 (t, $J = 7.2$ Hz, 3 H), 1.25 (s, 3 H), 1.60-1.68 (m, 4 H), 1.83-1.90 (m, 1 H), 2.49 (brs, 2 H), 2.67 (dd, $J = 14.0$ and 6.8 Hz, 1 H), 2.85 (dd, $J = 14.0$ and 6.8 Hz, 1 H), 3.80-3.88 (m, 1 H), 4.11 (q, $J = 7.2$ Hz, 2 H), 5.43 (brs, 1 H), 7.12 (d, $J = 8.4$, 2 H), 7.39 (d, $J = 8.4$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.4, 23.2, 24.6, 35.0, 41.7, 47.6, 60.2, 69.2, 74.0, 120.0, 126.5, 128.1, 131.2, 131.4, 137.9, 170.7; **IR** (Neat) 2973, 2927, 1734, 1674, 1488, 1369, 1213, 1073, 1033, 863, 820 cm^{-1} . **HRMS** (APCI) calcd. for $\text{C}_{18}\text{H}_{23}\text{BrO}_3$ ($\text{M}+\text{H}$) $^+$ requires 367.0909; found 367.0892 (^{79}Br).

Ethyl 2-((2*R,6*R**)-6-(4-methylbenzyl)-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (31f):**



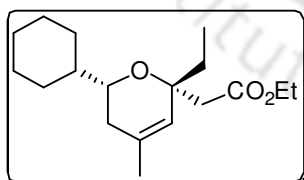
Colourless oil (194 mg, 64%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 1.24 (t, $J = 7.2$ Hz, 3 H), 1.29 (s, 3 H), 1.60-1.67 (m, 4 H), 1.82-1.89 (m, 1 H), 2.32 (s, 3 H), 2.51 (brs, 2 H), 2.65 (dd, $J = 13.6$ and 6.8 Hz, 1 H), 2.92 (dd, $J = 13.6$ and 6.8 Hz, 1 H), 3.82-3.89 (m, 1 H), 4.12 (q, $J = 7.2$ Hz, 2 H), 5.43 (brs, 1 H), 7.08 (d, $J = 8.0$, 2 H), 7.13 (d, $J = 8.0$ Hz, 2 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 14.4, 21.2, 23.2, 24.7, 35.0, 42.1, 47.7, 60.2, 69.7, 74.0, 126.4, 129.0, 129.5, 131.5, 135.6, 135.7, 170.9; **IR** (Neat) 2975, 2926, 1734, 1676, 1445, 1369, 1213, 1078, 1033, 807, 749 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{19}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 303.1960; found 303.1948.

Ethyl 2-((2*R,6*R**)-2-ethyl-5,6-dihydro-6-isopropyl-4-methyl-2*H*-pyran-2-yl)acetate (31g):**



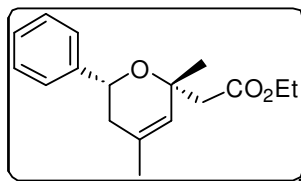
Colourless oil (165 mg, 65%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.89 (t, $J = 7.6$ Hz, 3 H), 0.90 (d, $J = 7.2$ Hz, 3 H), 0.96 (d, $J = 7.2$ Hz, 3 H), 1.23 (t, $J = 7.6$ Hz, 3 H), 1.50-1.61 (m, 2 H), 1.63-1.74 (m, 5 H), 1.80-1.87 (m, 1 H), 2.47 (brs, 2 H), 3.26-3.31 (m, 1 H), 4.05-4.13 (m, 2 H), 5.50 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 8.0, 14.4, 18.4, 18.9, 23.5, 29.7, 32.7, 33.1, 43.8, 60.1, 73.2, 75.7, 126.0, 131.9, 171.2; **IR** (Neat) 2963, 2928, 1735, 1685, 1449, 1368, 1207, 1073, 1033, 867 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{15}\text{H}_{26}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 255.1960; found 255.1952.

Ethyl 2-((2*R,6*R**)-6-cyclohexyl-2-ethyl-5,6-dihydro-4-methyl-2*H*-pyran-2-yl)acetate (31h):**



Colourless oil (183 mg, 62%); $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 0.89 (t, $J = 7.6$ Hz, 3 H), 0.94-1.03 (m, 2 H), 1.12-1.19 (m, 2 H), 1.23 (t, $J = 7.2$ Hz, 3 H), 1.31-1.40 (m, 2 H), 1.48-1.57 (m, 2 H), 1.62-1.76 (m, 7 H), 1.81-1.88 (m, 2 H), 1.98-2.02 (m, 1 H), 2.47 (brs, 2 H), 3.28-3.34 (m, 1 H), 4.05-4.12 (m, 2 H), 5.49 (brs, 1 H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3): δ 8.0, 14.5, 23.6, 26.3, 26.5, 26.9, 28.8, 29.4, 29.6, 32.9, 42.9, 43.8, 60.2, 72.5, 75.7, 126.1, 132.0, 171.3; **IR** (Neat) 2965, 2925, 1735, 1675, 1449, 1377, 1207, 1096, 1033, 844 cm^{-1} . **HRMS** (APCI) cald. for $\text{C}_{18}\text{H}_{30}\text{O}_3$ ($\text{M}+\text{H}$) $^+$ requires 295.2273; found 295.2265.

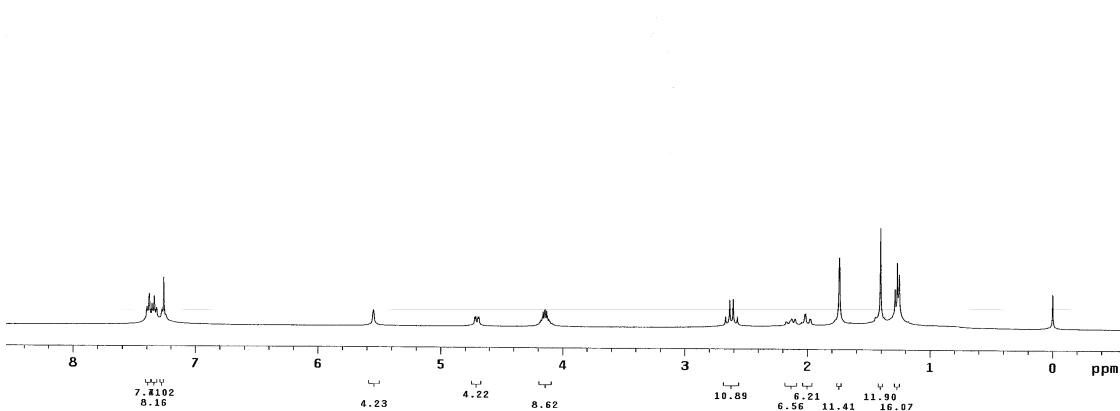
6.7 Selected Spectra of Substituted Dihydropyrans

Ethyl 2-((2*R**,6*R**)-5,6-dihydro-2,4-dimethyl-6-phenyl-2*H*-pyran-2-yl)acetate (**27a**):¹H NMR (400 MHz, CDCl₃)

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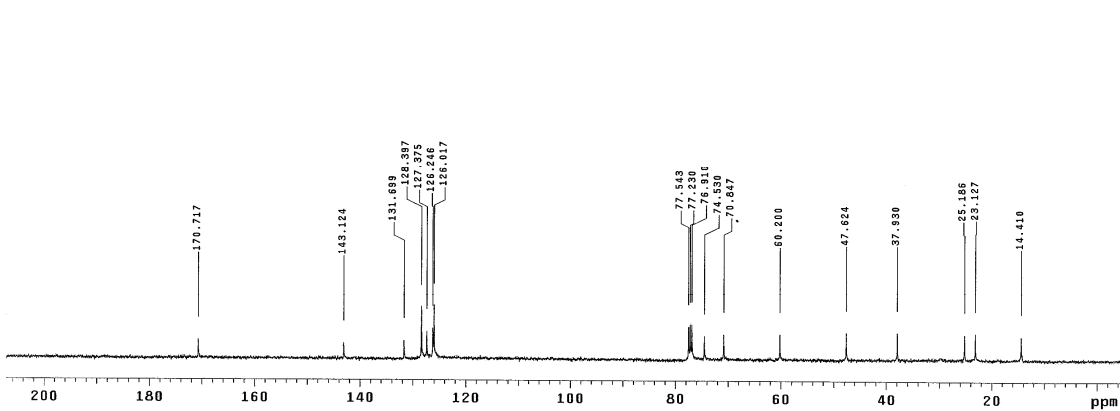
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¹³C NMR (100 MHz, CDCl₃)

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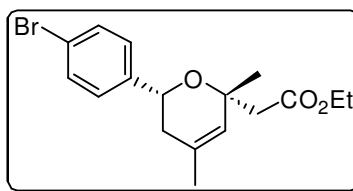
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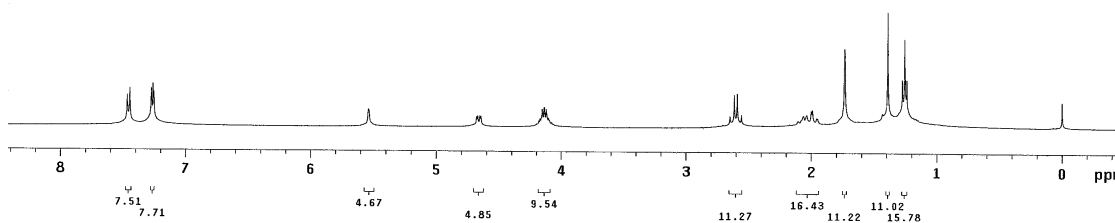
Ethyl 2-((2*R,6*R**)-6-(4-bromophenyl)-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (27c):**

¹H NMR (400 MHz, CDCl₃)



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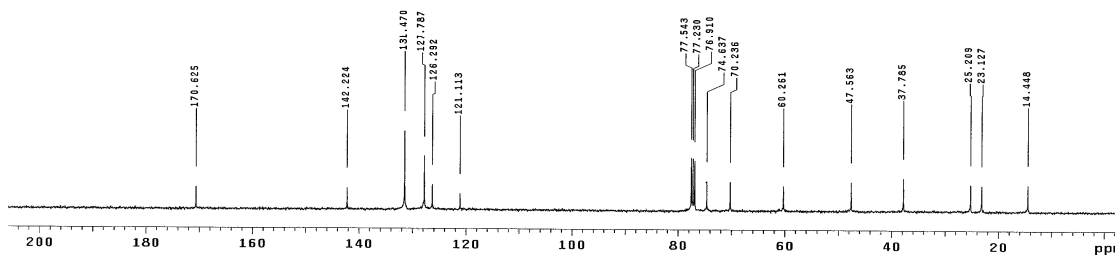
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solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_580~ nst 0.000
M_Pdt.Fid pw90 19.700
ACQUISITION alfa 20.000
sw 6389.8 FLAGS
at 1.998 f1 n
np 25528 in n
fb not used dp y
bs 8 hs n
d1 1.000 PROCESSING
nt 200 lb 0.10
ct 84 fn 65536
TRANSMITTER H1 sp DISPLAY
tn 399.853 wp -206.9
sfrq 362.8 rfl 794.7
tpwr 57 rfp 0
pw 9.850 rp 100.7
DECOUPLER l1 p -87.0
dn C13 PLOT
dof 0 vc 250
dm nnn sc 0
dmm c vs 25
dpwr 50 th 20
dmf 15900 nm cdc ph
  
```



¹³C NMR (100 MHz, CDCl₃)

```

PS_580_M
expl s2pu1
SAMPLE
date Jan 28 2012 temp not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_580~ nst 0.000
M_Pdt.Fid pw90 18.600
ACQUISITION alfa 20.000
sw 25125.6 FLAGS
at 1.199 f1 n
np 60270 in n
fb 13800 dp y
bs 16 hs n
d1 1.000 PROCESSING
nt 5000 lb 2.00
ct 2832 fn 65536
TRANSMITTER C13 sp DISPLAY
tn 100.554 wp 21130.6
sfrq 1536.3 rfl 9279.8
tpwr 61 rfp 7764.9
pw 9.300 rp -49.3
DECOUPLER H1 p -360.6
dn H1 PLOT
dof 0 vc 250
dm YVY sc 0
dmm w vs 20
dpwr 42 th 3
dmf 8900 nm no ph
  
```

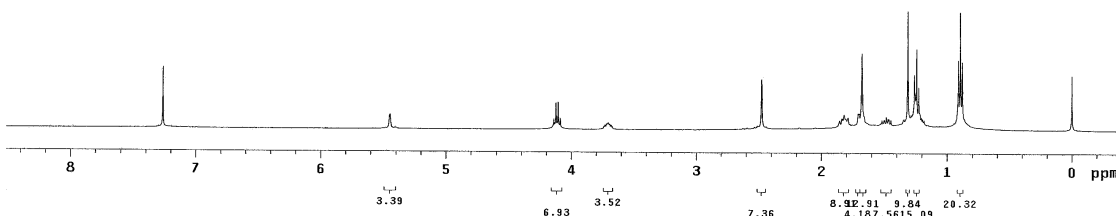
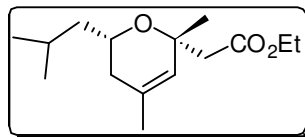


Ethyl 2-((2*R**,6*S**)-5,6-dihydro-6-isobutyl-2,4-dimethyl-2*H*-pyran-2-yl)acetate (27j):¹H NMR (400 MHz, CDCl₃)

```

PS_587_M
exp1 s2pu1
SAMPLE
date Feb 8 2012 temp not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_587- hst 0.008
M_Fid pw90 19.700
ACQUISITION a1fa 20.000
sw 6389.8 FLAGS n
at 1.998 i1 n
np 25528 in n
fb not used dp n
bs 8 hs y
d1 1.000 PROCESSING nn
nt 200 lb 0.10
ct 200 fn 65536
TRANSMITTER H1 sp -164.0
tn 399.853 wp 3568.4
tof 362.8 rfl 793.7
tpwr 57 rfp 0
pw 9.850 rp 110.6
DECOUPLER C13 lp -90.4
dn 0 wc PLOT 250
dof 0 sc 0
dm c vs 27
dmm 50 th 9
dmf 15900 nm cdc ph

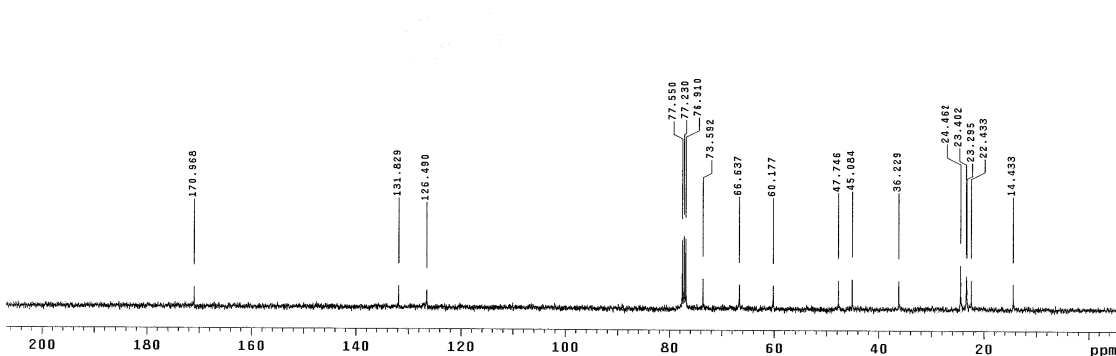
```

¹³C NMR (100 MHz, CDCl₃)

```

PS_587_M
exp1 s2pu1
SAMPLE
date Feb 8 2012 temp not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_587- hst 0.008
M_C13.fid pw90 18.600
ACQUISITION a1fa 20.000
sw 25125.6 FLAGS n
at 1.199 i1 n
np 60270 in n
fb 13800 dp y
bs 16 hs y
d1 1.000 PROCESSING nn
nt 4000 lb 2.00
ct 992 fn 65536
TRANSMITTER C13 sp -649.2
tn 100.594 wp 21493.3
tof 1536.3 rfl 9272.1
tpwr 61 rfp 7764.9
pw 9.300 rp -80.6
DECOUPLER H1 lp -271.4
dn 0 wc PLOT 250
dof 0 sc 0
dm yyy vs 16
dmm 42 th 4
dmf 8900 nm no ph

```

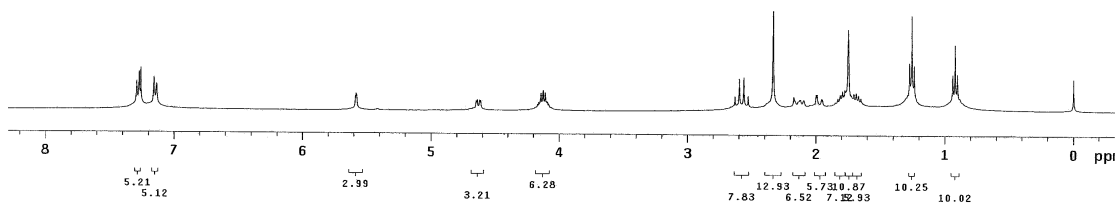
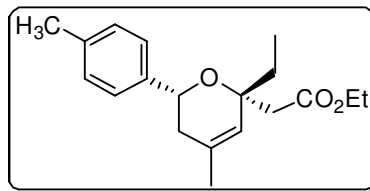


Ethyl 2-((2*R**,6*R**)-2-ethyl-5,6-dihydro-4-methyl-6-*p*-tolyl-2*H*-pyran-2-yl)acetate (27l):¹H NMR (400 MHz, CDCl₃)

```

PS_592_M
exp1 s2pu1
SAMPLE
date Feb 19 2012 temp not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_592~ hst 0.008
M.Fid pw90 19.700
ACQUISITION alfa 20.000
sw 6389.8 FLAGS
at 1.998 f1 n
np 25528 ln n
fb not used dp y
bs 8 hs nn
d1 1.000 PROCESSING
nt 200 lb 0.10
ct 200 fn 65536
TRANSMITTER H1 sp DISPLAY
tn 399.853 wp -166.0
tof 362.8 rff 3404.6
tpwr 57 rfp 795.6
pw 9.850 rp 109.2
DECOUPLER C13 lp -97.8
dn dof 0 wc PLOT 250
dm nnn sc 0
dmm c vs 23
dpwr 50 th 6
dmf 15900 nm cdc ph

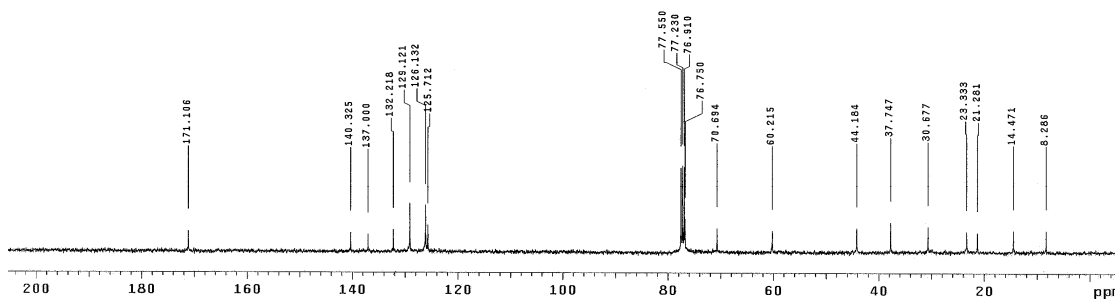
```

¹³C NMR (100 MHz, CDCl₃)

```

PS_592_M
exp1 s2pu1
SAMPLE
date Feb 19 2012 temp not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_592~ hst 0.008
M.C13.Fid pw90 18.600
ACQUISITION alfa 20.000
sv 2425.6 FLAGS
at 1.199 f1 n
np 68278 ln n
fb 13800 dp y
bs 16 hs nn
d1 1.000 PROCESSING
nt 8000 lb 2.00
ct 3200 fn 65536
TRANSMITTER C13 sp DISPLAY
tn 100.554 wp -685.2
tof 1536.3 rff 21361.4
tpwr 61 rfp 9275.2
pw 9.300 rp 7764.9
DECOUPLER H1 lp -349.0
dn dof 0 wc PLOT 250
dm yyy sc 0
dmm w vs 20
dpwr 42 th 3
dmf 8900 nm no ph

```



Ethyl 2-((2*R**,6*R**)-6-benzyl-5,6-dihydro-2,4-dimethyl-2*H*-pyran-2-yl)acetate (**31c**):¹H NMR (400 MHz, CDCl₃)

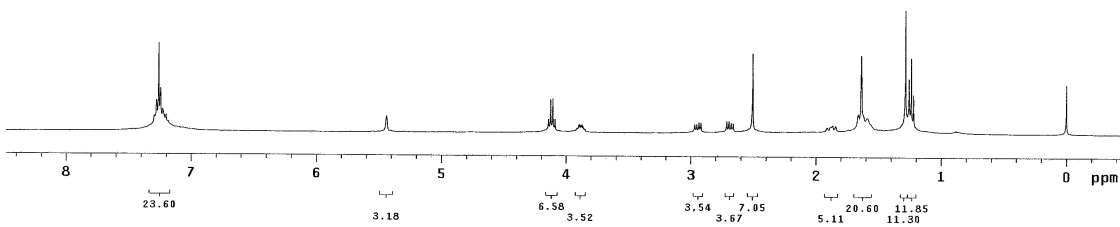
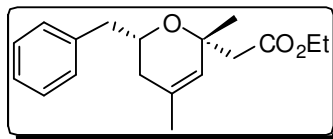
```

PS_595_M
expl s2pul

SAMPLE
date Feb 25 2012 temp not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_595- hst 0.008
M.Fid pw90 19.700
ACQUISITION
6389.8 a1fa 20.000
sw
at 1.998 i1
np 25528 in
fb not used dp
bs 8 hs
d1 1.000 PROCESSING
nt 300 lb 0.10
ct 264 fn 65536

TRANSMITTER H1 sp DISPLAY
tn H1 sp -181.9
sfrq 399.853 wp 3976.8
tof 362.8 rF1 794.8
tpwr 57 rfp 0
pw 9.850 rp 108.6
DECOUPLER C13 lp -88.8
dn
dof 0 wc PLOT 250
dm nnn sc 0
dmb c vs 27
dpwr 50 th 4
dmt 15900 nm cdc ph

```

¹³C NMR (100 MHz, CDCl₃)

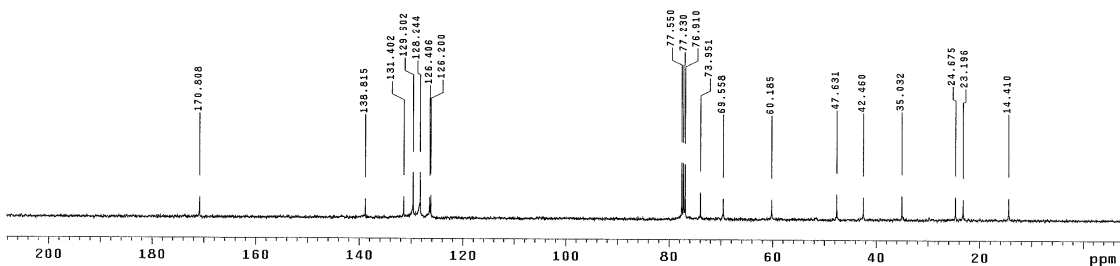
```

PS_595_M
expl s2pul

SAMPLE
date Feb 25 2012 temp not used
solvent CDCl3 gain not used
file /export/home/~ spin not used
ciftemp/AKS_PS_595- hst 0.008
M.C13.Fid pw90 18.600
ACQUISITION
25125.6 a1fa 20.000
sw
at 1.199 i1
np 68270 in
fb 13800 dp
bs 16 hs
d1 1.000 PROCESSING
nt 5000 lb 2.00
ct 2240 fn 65536

TRANSMITTER C13 sp DISPLAY
tn C13 sp -756.5
sfrq 180.554 wp 21691.1
tof 1536.3 rF1 9280.5
tpwr 61 rfp 7764.9
pw 9.300 rp -75.1
DECOUPLER H1 lp -301.3
dn
dof 0 wc PLOT 250
dm yvy sc 0
dmb w vs 13
dpwr 42 th 2
dmt 8900 nm no ph

```



List of Publications

1. A Novel Synthesis of Oxabicyclo[3.3.1]nonanone via (3,5)-Oxonium-Ene Reaction.
Saha, P.; Reddy, U. C.; Bondalapati, S.; Saikia, A. K. *Org. Lett.* **2010**, *12*, 1824-1826.
2. Synthesis of Oxabicyclo[3.3.1]nonenes and Substituted Tetrahydropyrans via (3,5)-Oxonium-Ene Reaction.
Saha, P.; Gogoi, P.; Saikia, A. K. *Org. Biomol. Chem.* **2011**, *9*, 4626-4634.
3. Stereoselective Synthesis of 6-Oxabicyclo[3.2.1]octene via (3,5)-Oxonium-Ene Reaction.
Saha, P.; Saikia, A. K. *Tetrahedron.* **2012**, *68*, 2261-2266.
4. Synthesis of 2,3,5,6-Tetrasubstituted Tetrahydropyrans via (3,5)-Oxonium-Ene Reaction.
Saha, P.; Bhunia, A.; Saikia, A. K. *Org. Biomol. Chem.* **2012**, *10*, 2470-2481.
5. Diastereoselective Synthesis of Substituted Dihydropyrans via Oxonium-Ene Cyclization Reaction.
Saha, P.; Ghosh, P.; Sultana, S.; Saikia, A. K. *Org. Biomol. Chem.* **2012**, *10*, 8730-8738.
6. An Efficient Synthesis of 10-Amino Artemisinin Derivative.
Saha, P.; Saikia, A. K. *J. Heterocycl. Chem.* **2012** (Article in press).
7. An Efficient Synthesis of Dihydro- and Tetrahydropyrans via Oxonium-Ene Cyclization Reaction.
Bondalapati, S.; Reddy, U. C.; **Saha, P.;** Saikia, A. K. *Org. Biomol. Chem.* **2011**, *9*, 3428-3438.
8. Application of a Novel 1,3-Diol with a Benzyl Backbone as Chiral Ligand for Asymmetric Oxidation of Sulfides to Sulfoxides.
Gogoi, P.; Kotipalli, T.; Indukuri, K.; Bondalapati, S.; **Saha, P.;** Saikia, A. K. *Tetrahedron Lett.* **2012**, *53*, 2726-2729.