

Contents for Annexure

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Annexure 1

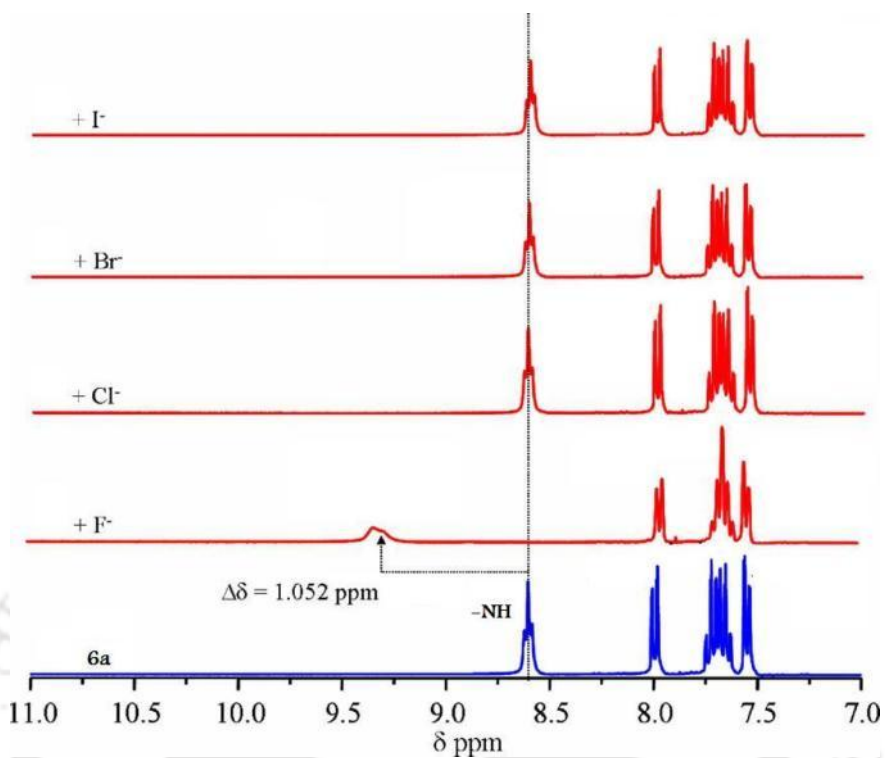


Figure A1.1 Partial ^1H NMR (300 MHz) spectra of receptor **6a** in $\text{DMSO-}d_6$ with $(\text{TBA})\text{X}^-$ (where $\text{X}^- = \text{F}^-$, Cl^- , Br^- and I^-) at 298 K.

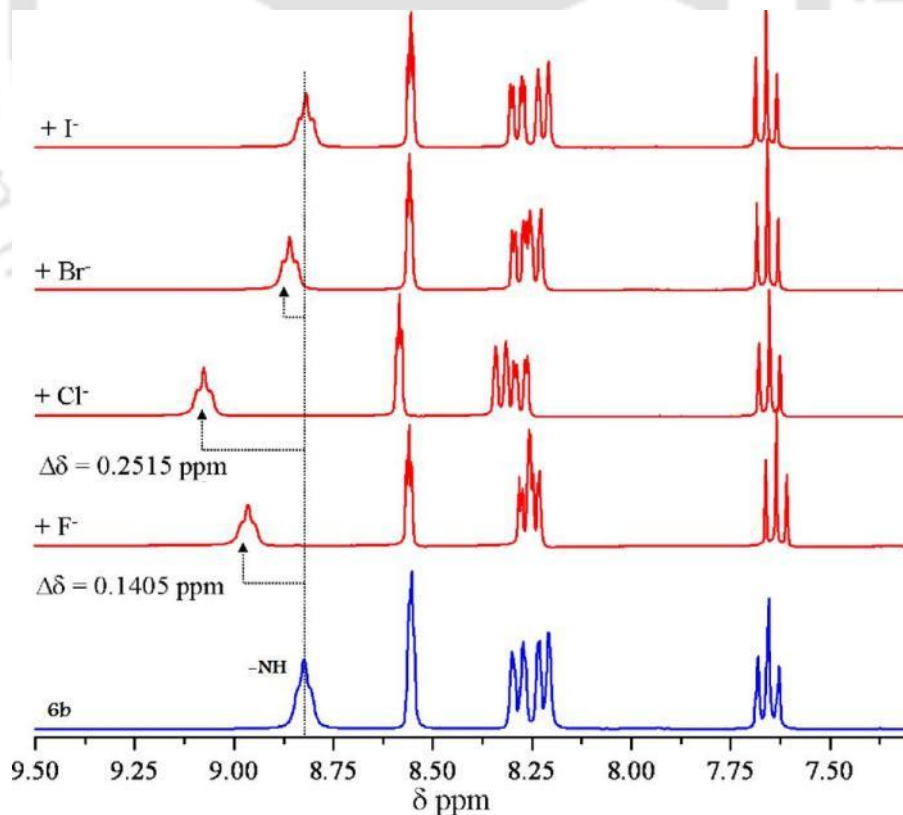


Figure A1.2 Partial ^1H NMR (300 MHz) spectra of receptor **6b** in $\text{DMSO-}d_6$ with $(\text{TBA})\text{X}^-$ (where $\text{X}^- = \text{F}^-$, Cl^- , Br^- and I^-) at 298 K.

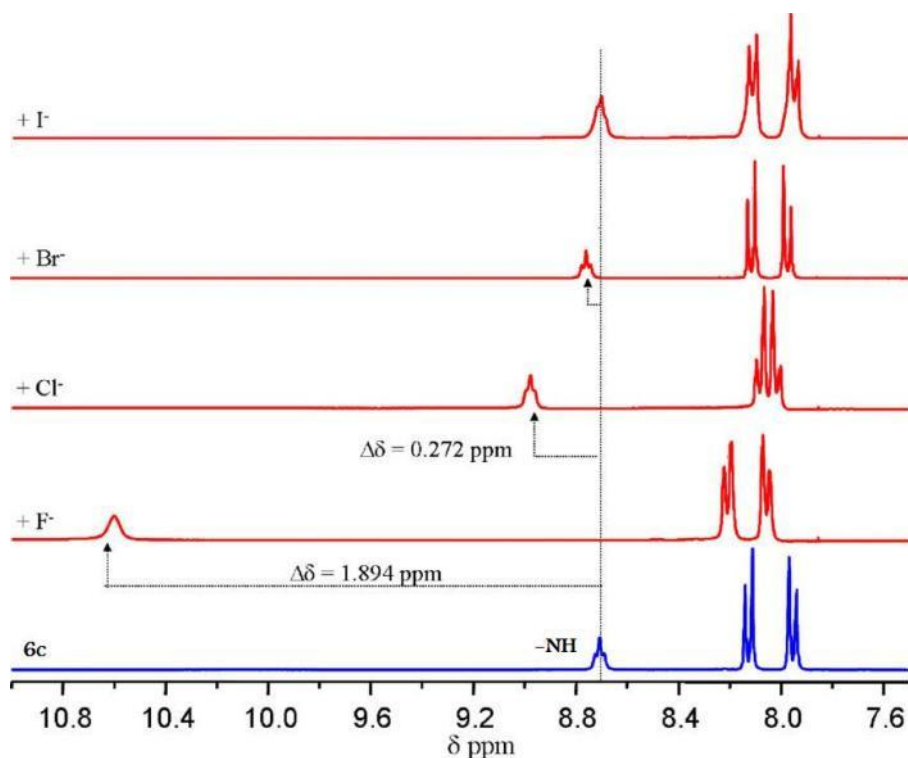


Figure A1.3 Partial ^1H NMR (300 MHz) spectra of receptor **6c** in $\text{DMSO-}d_6$ with $(\text{TBA})\text{X}^-$ (where $\text{X}^- = \text{F}^-, \text{Cl}^-, \text{Br}^-$ and I^-) at 298 K.

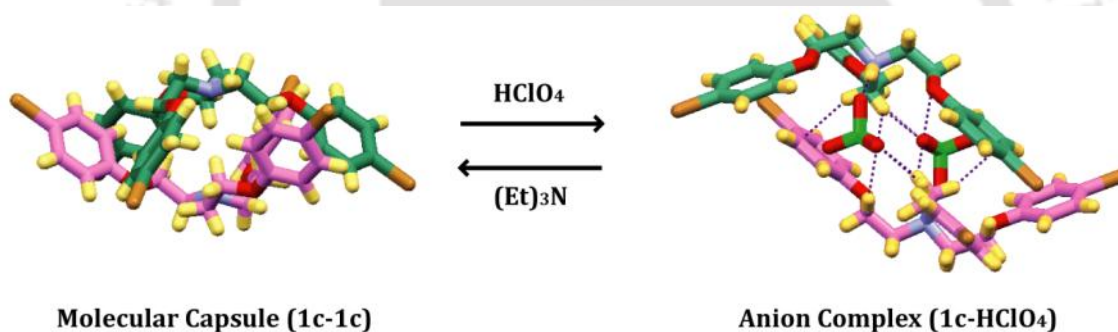


Figure A1.4 pH-modulated reversible exchange between molecular capsule (**1c•1c**) and its anion (ClO_4^-) complex.

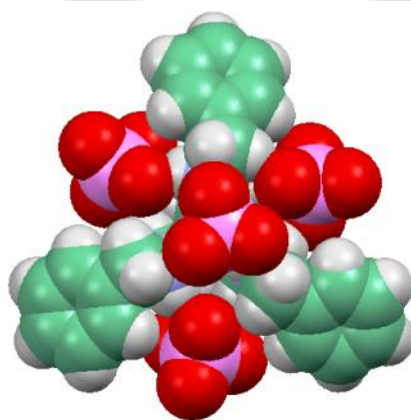


Figure A1.5 Spacefill model of the crystal structure of $[(\text{H}_3\mathbf{2a})(\text{H}_2\text{PO}_4^-)_3]\cdot\text{H}_3\text{PO}_4$ showing the binding of three H_2PO_4^- counter anions between each receptor side arms and an additional H_3PO_4 molecule above the quasi-planar tripod.

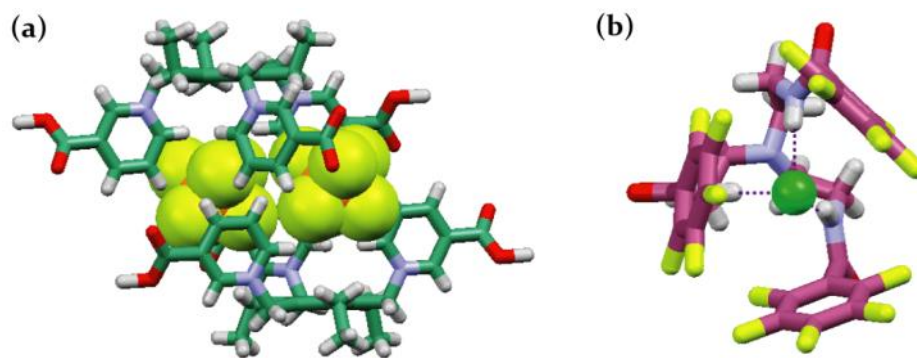


Figure A1.6 (a) Non-capsular aggregation of complex $[\text{H3a}\cdot\text{PF}_6\cdot 2\text{H}_2\text{O}]$, and (b) encapsulation of a naked Cl^- anion by receptor **6d**.

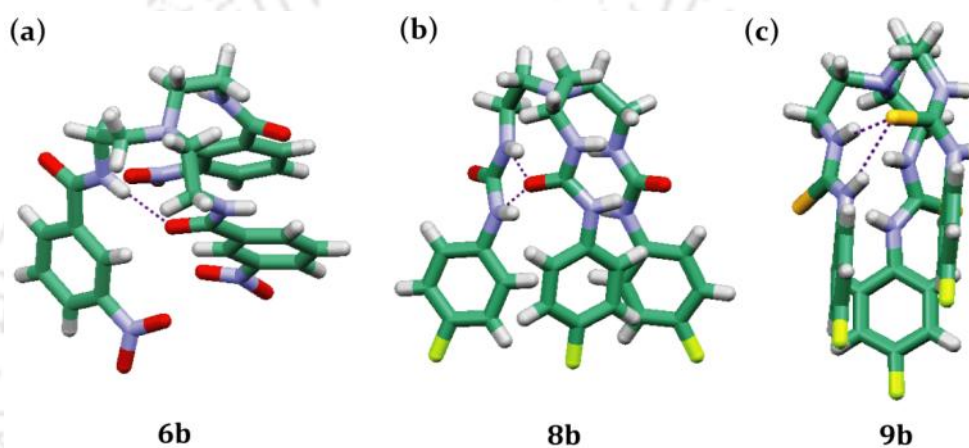


Figure A1.7 Intramolecular $\text{N-H}\cdots\text{X}=\text{C}$ ($\text{X} = \text{O/S}$) hydrogen bonding interactions between the receptor side arms of amide **6b**, urea **8b** and thiourea **9b**.

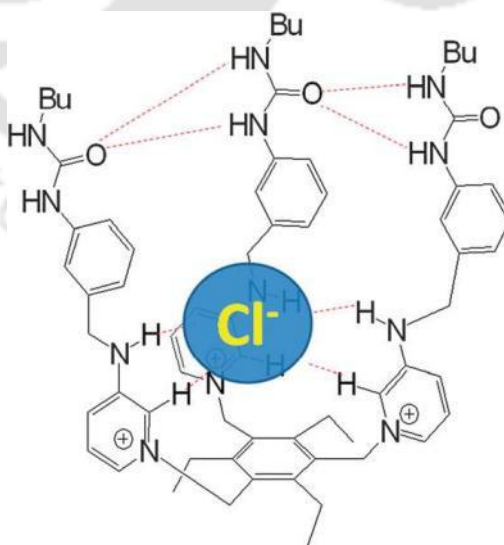
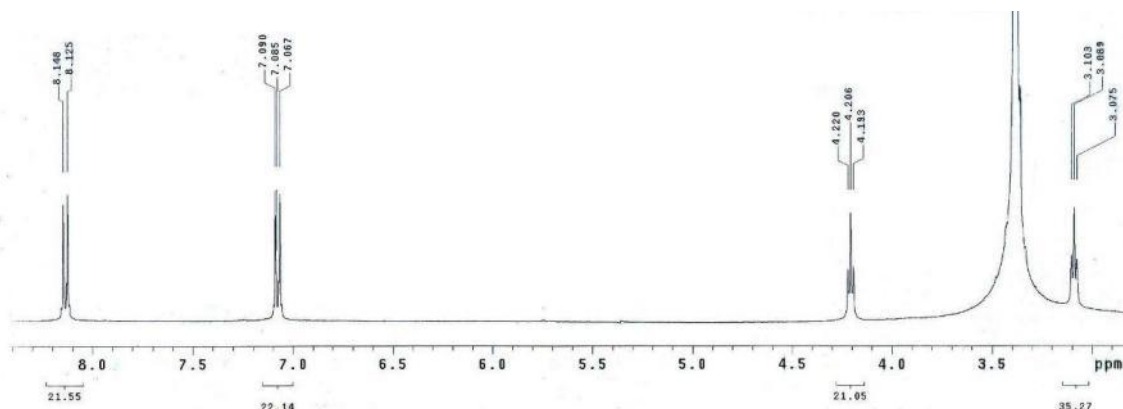
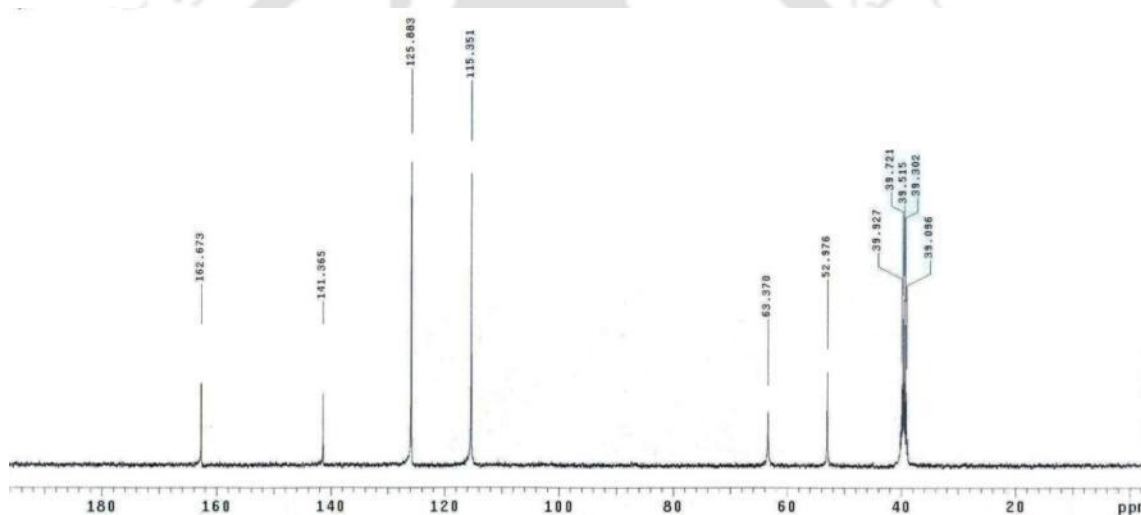
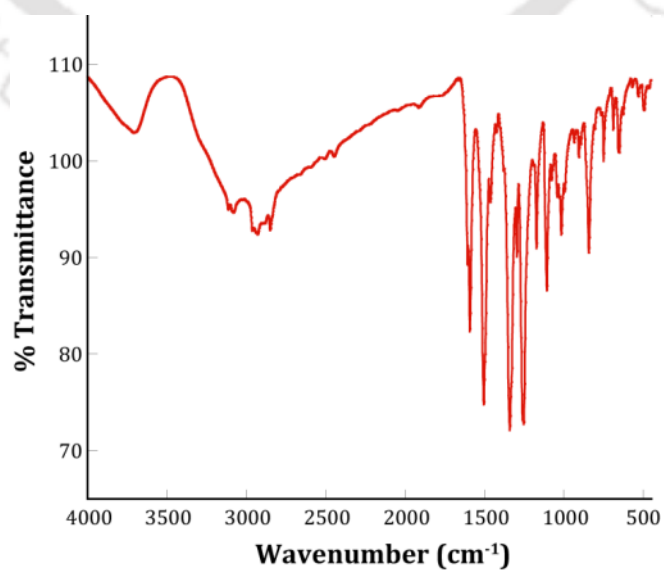


Figure A1.8 Chloride binding assisted formation of unimolecular capsule in receptor **12**.

Annexure 2

Characterization data of receptors, L₁-L₅:Figure A2.1 ¹H NMR spectrum of receptor L₁ (400 MHz, DMSO-d₆).Figure A2.2 ¹³C NMR spectrum of receptor L₁ (100 MHz, DMSO-d₆).Figure A2.3 FT-IR spectrum of L₁ recorded in KBr pellet.

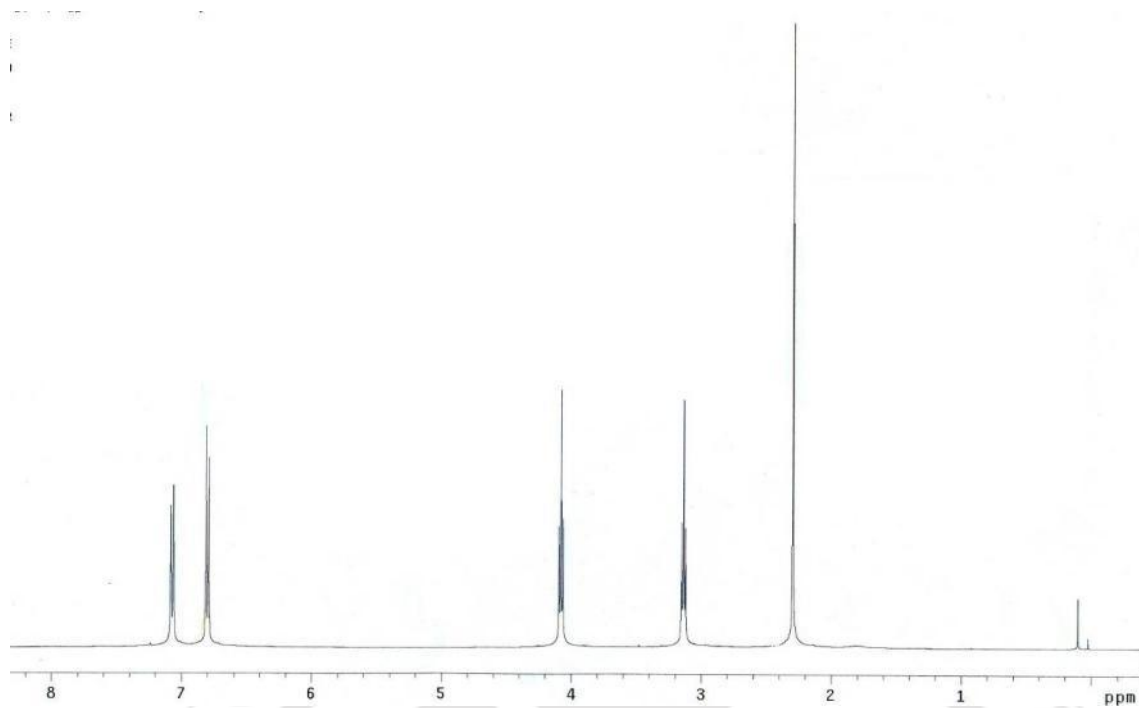


Figure A2.4 ^1H NMR spectrum of receptor **L**₂ (400 MHz, CDCl_3).

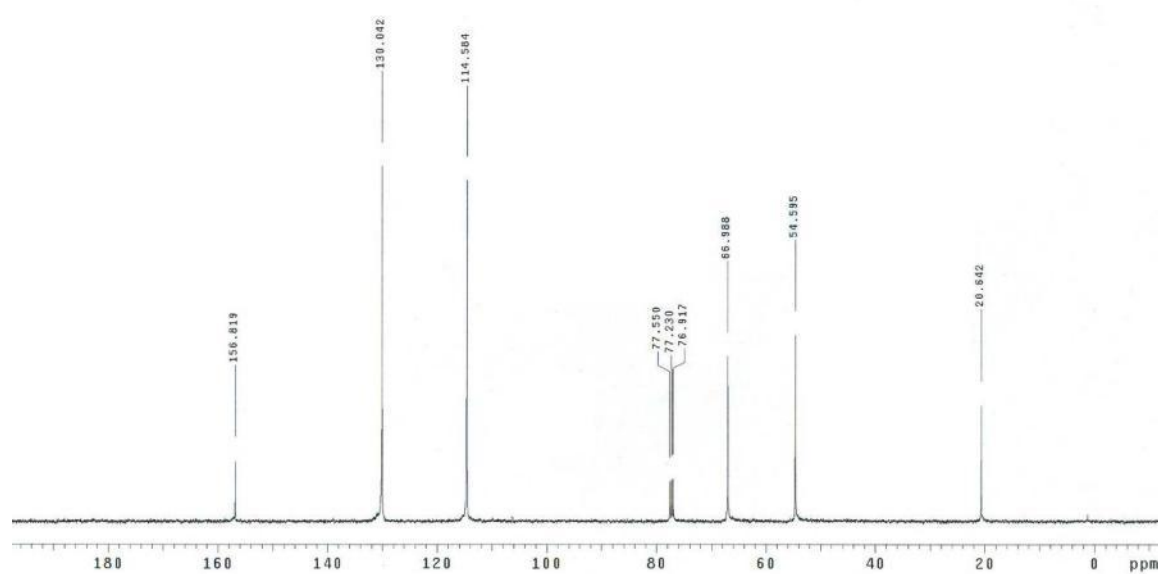


Figure A2.5 ^{13}C NMR spectrum of receptor **L**₂ (100 MHz, CDCl_3).

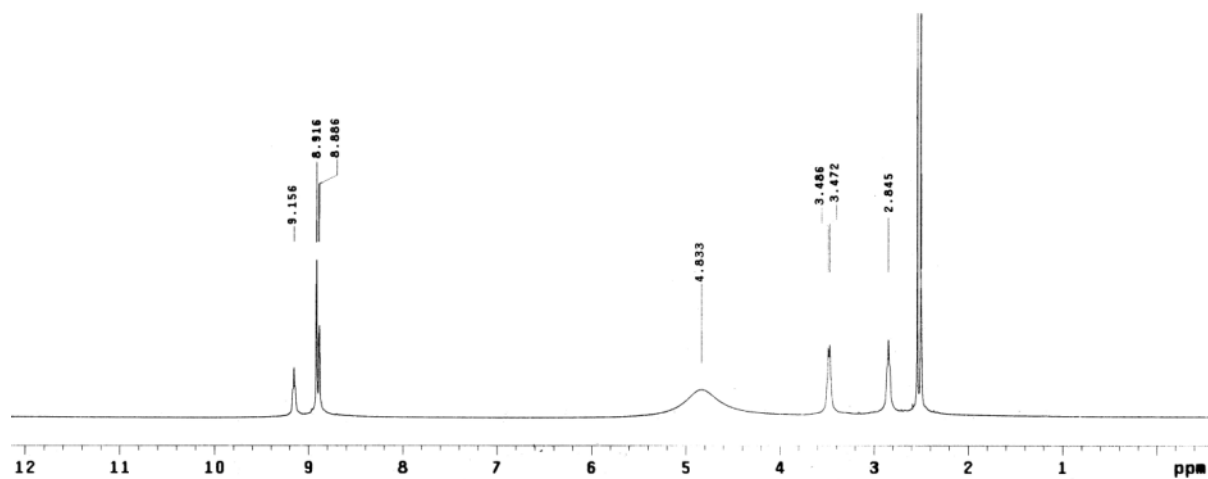


Figure A2.6 ¹H NMR spectrum of receptor L₃ (400 MHz, DMSO-*d*₆).

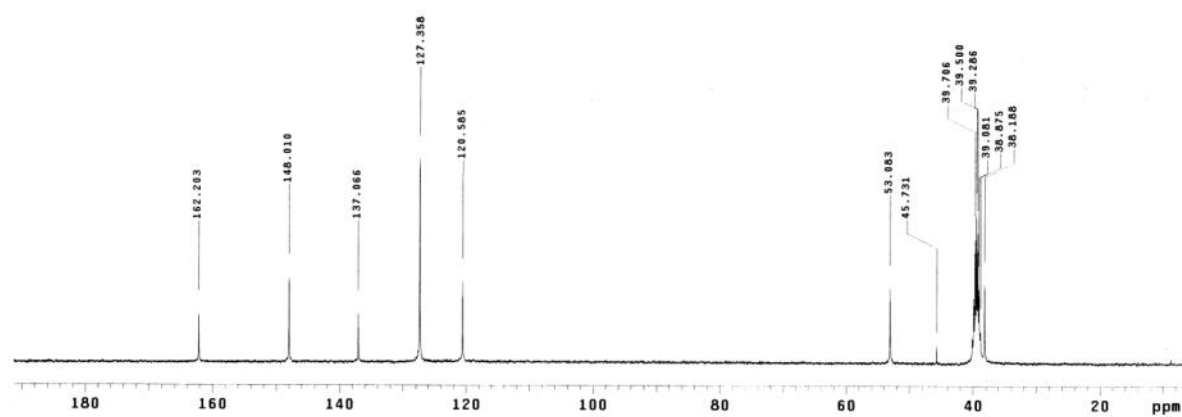


Figure A2.7 ¹³C NMR spectrum of receptor L₃ (100 MHz, DMSO-*d*₆).

TOF MS
SD_101 43 (0.730)

24-Jan-2011
TOF MS ES+
598

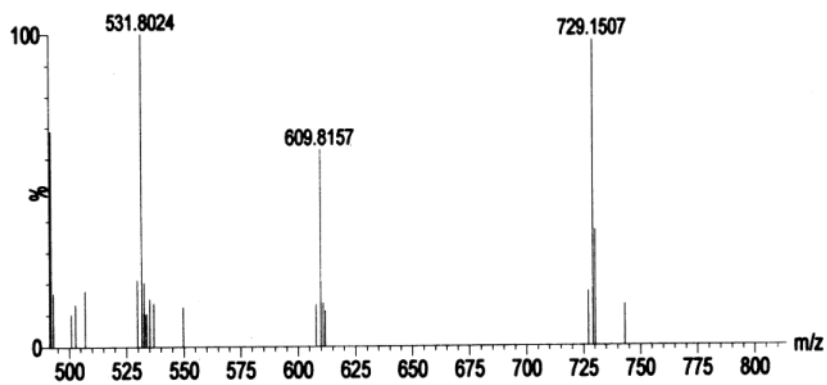


Figure A2.8 ESI-mass spectrum of L₃ showing m/z = 729.150 corresponds to [M+1]⁺.

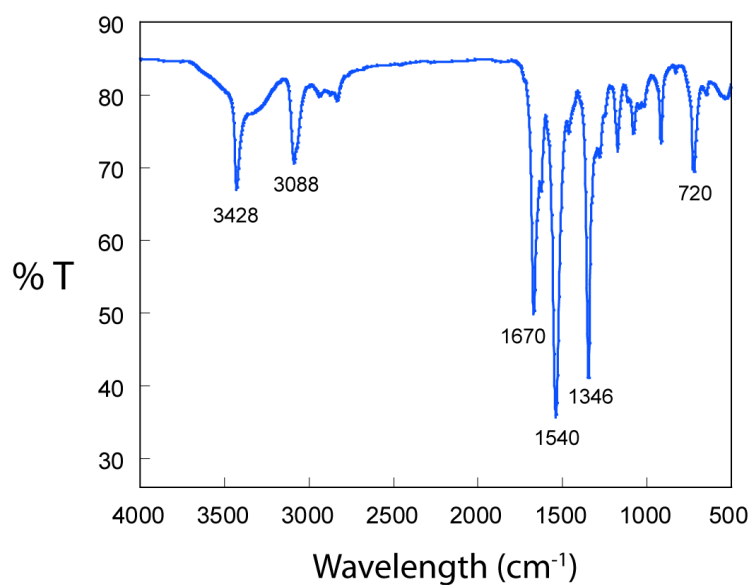


Figure A2.9 FT-IR spectrum of L_3 recorded in KBr pellet.

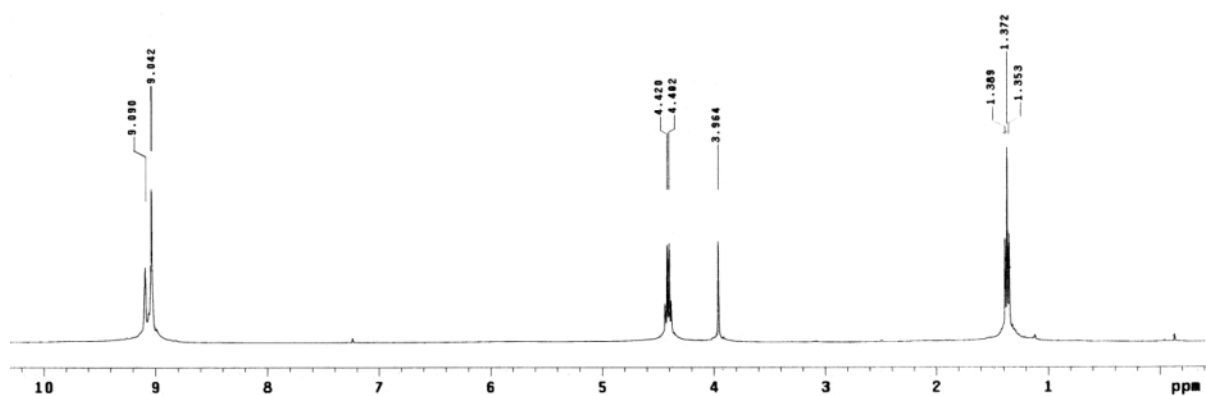


Figure A2.10 ^1H NMR spectrum of receptor CL_3 (400 MHz, CDCl_3).

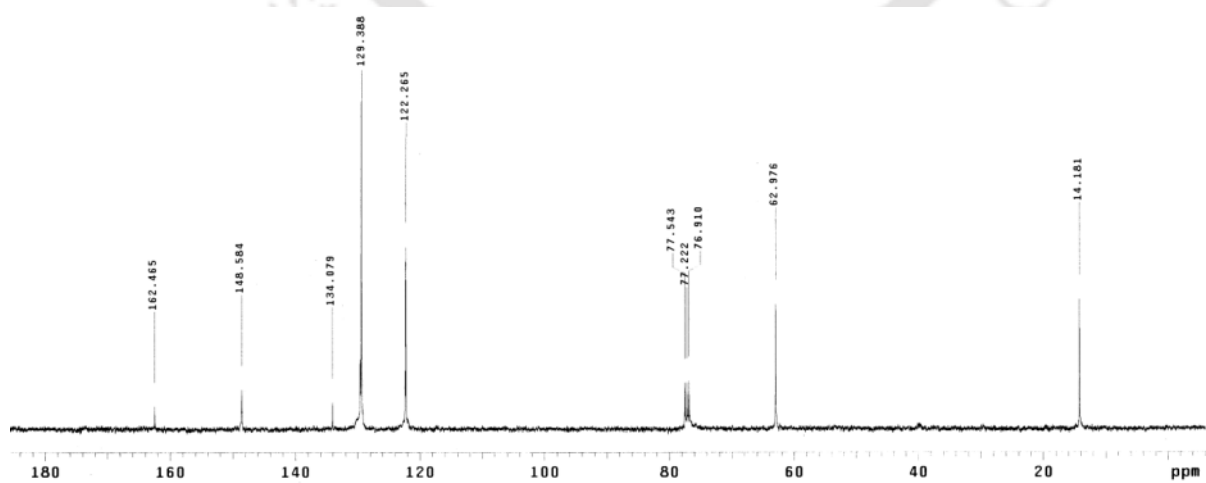


Figure A2.11 ^{13}C NMR spectrum of receptor CL_3 (100 MHz, CDCl_3).

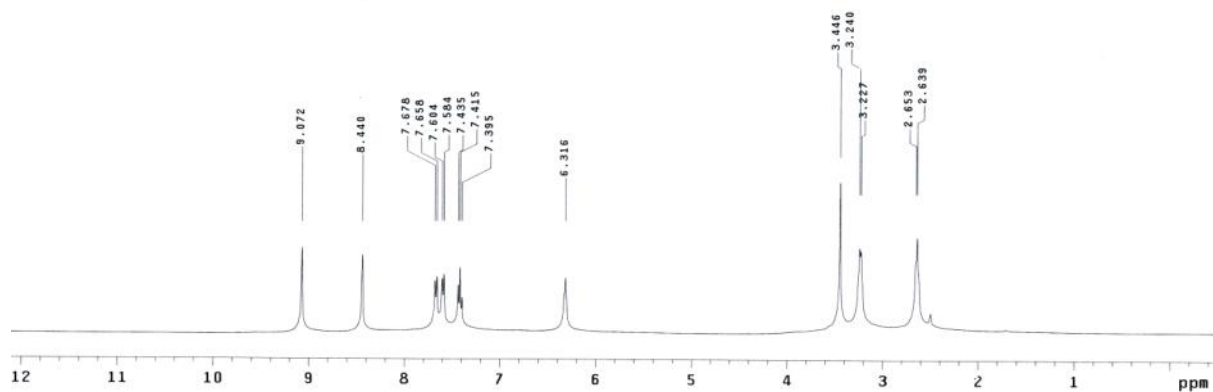


Figure A2.12 ^1H NMR spectrum of receptor L_4 (400 MHz, $\text{DMSO-}d_6$).

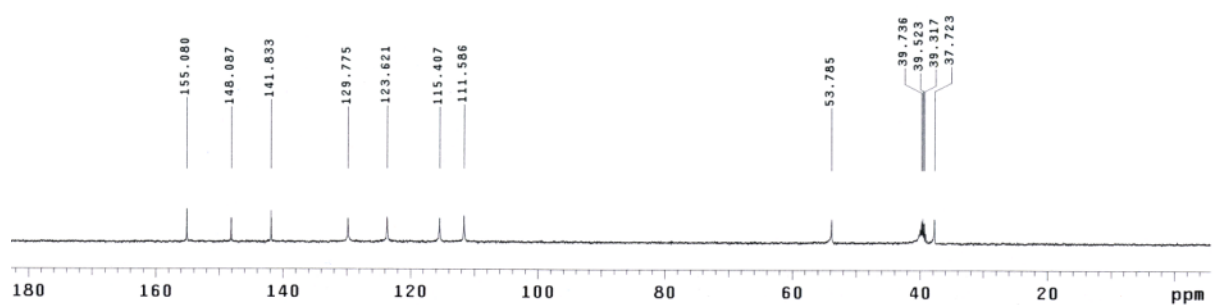


Figure A2.13 ^{13}C NMR spectrum of receptor L_4 (100 MHz, $\text{DMSO-}d_6$).

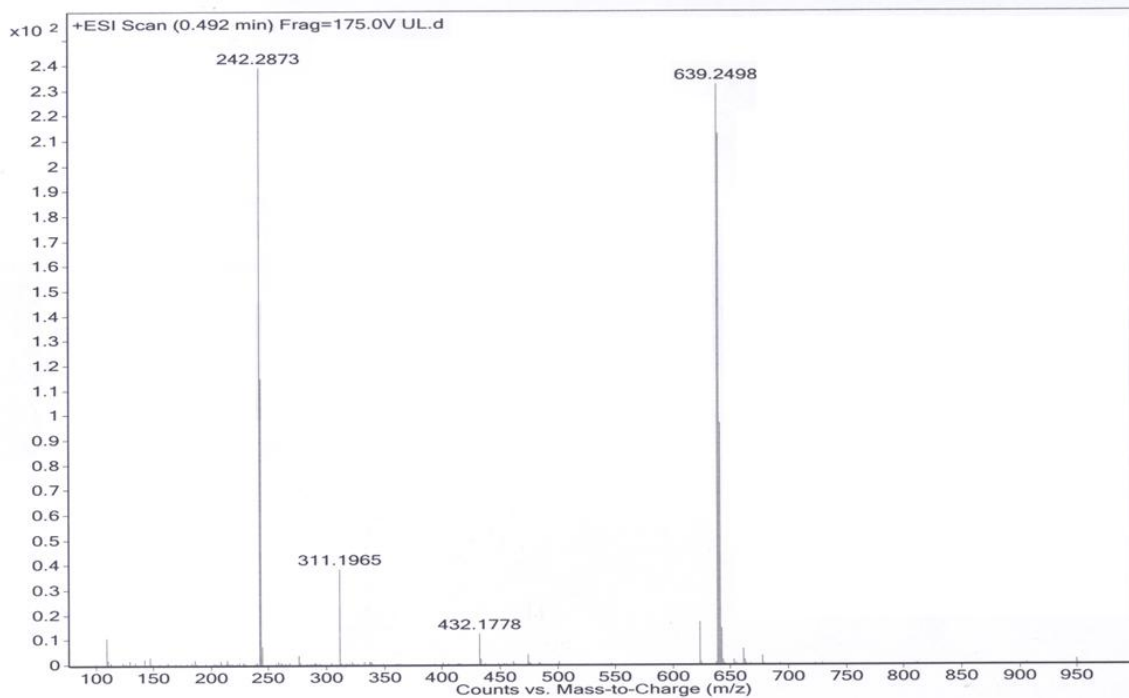


Figure A2.14 ESI-mass spectrum of L_4 showing $m/z = 639.249$ corresponds to $[\text{M}+1]^+$.

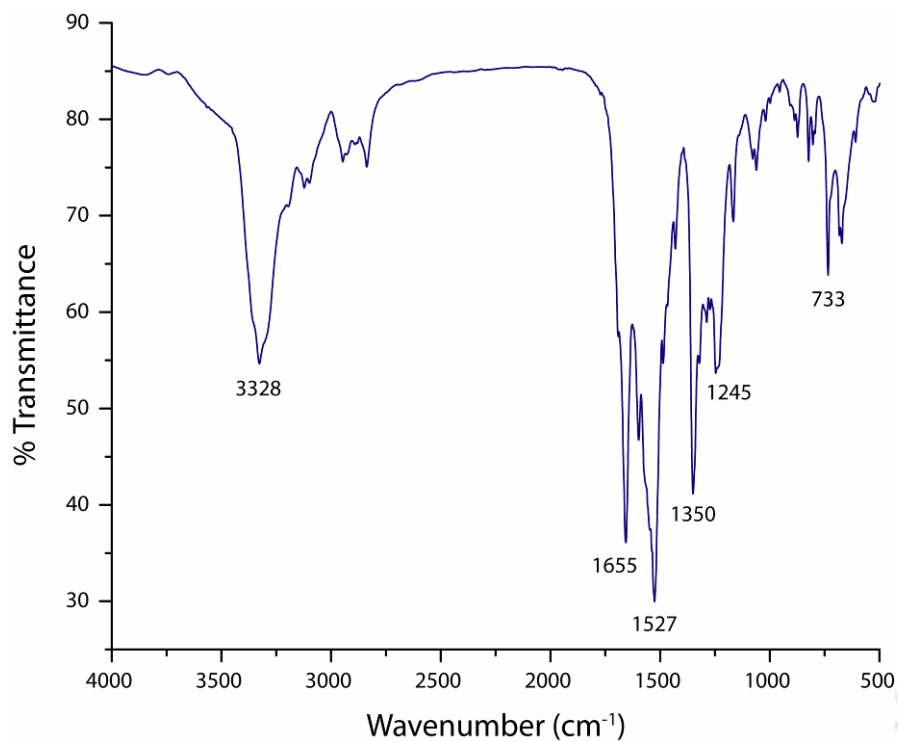


Figure A2.15 FT-IR spectrum of L_4 recorded in KBr pellet.

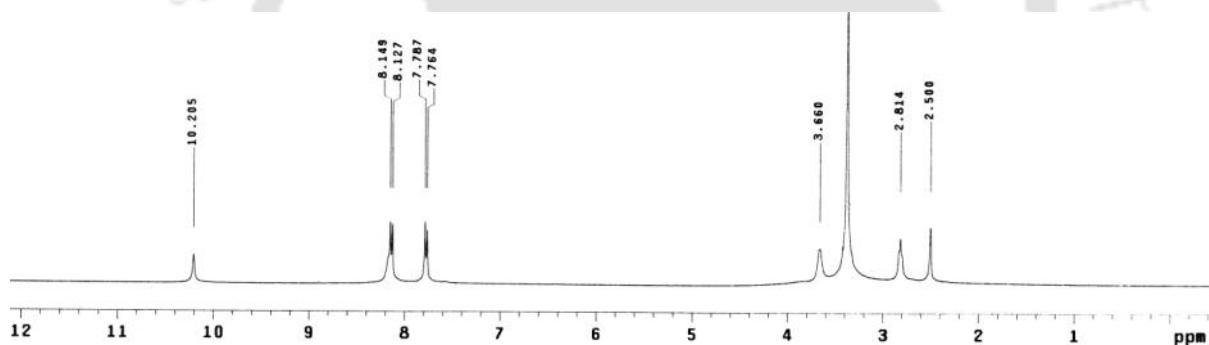


Figure A2.16 ^1H NMR spectrum of receptor L_5 (400 MHz, $\text{DMSO}-d_6$).

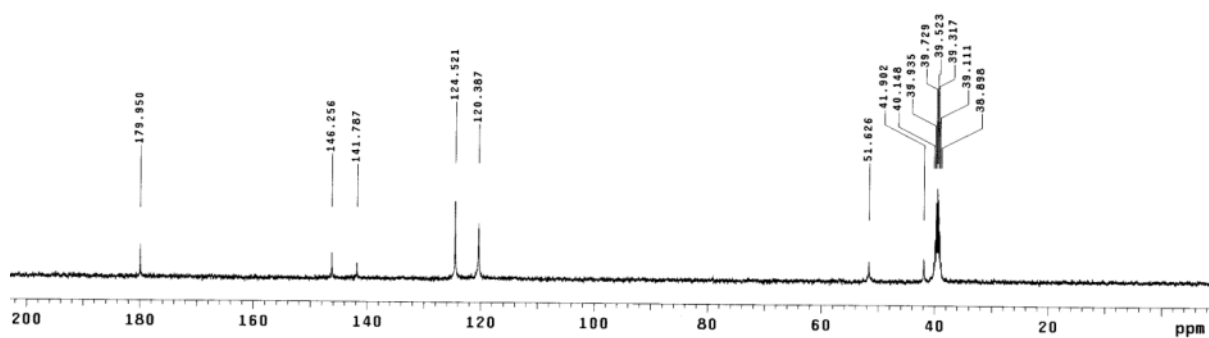


Figure A2.17 ^{13}C NMR spectrum of receptor L_5 (100 MHz, $\text{DMSO}-d_6$).

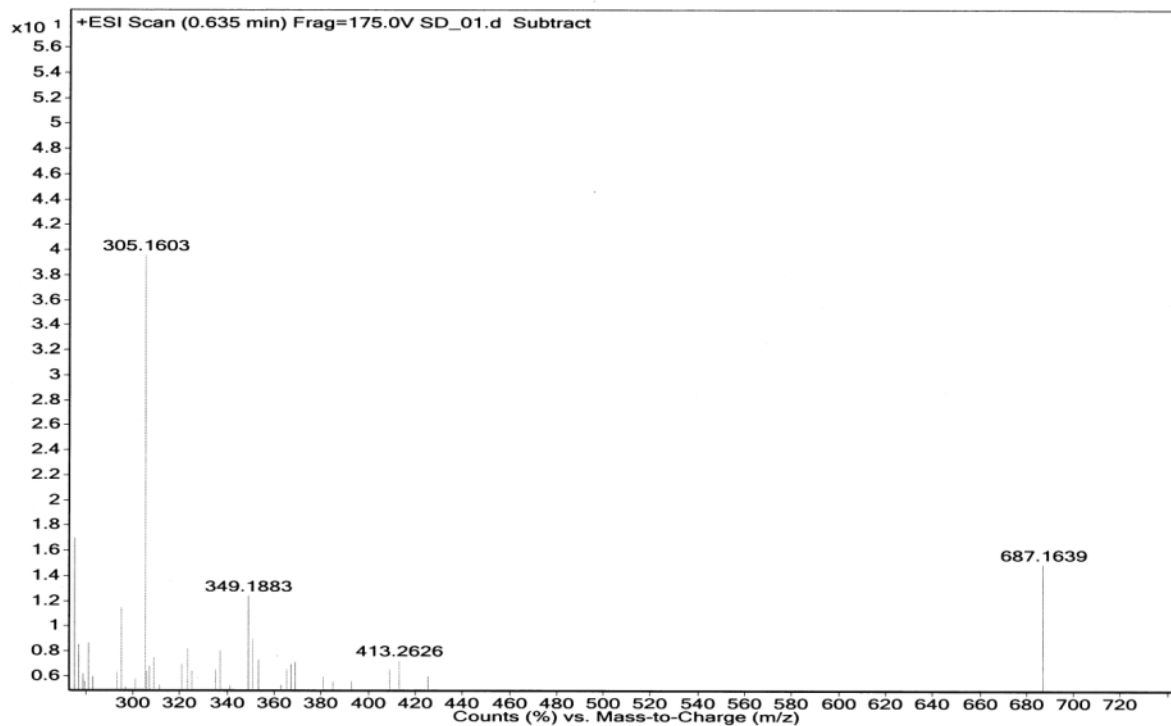


Figure A2.18 ESI-mass spectrum of L_5 showing $m/z = 687.164$ corresponds to $[M+1]^+$.

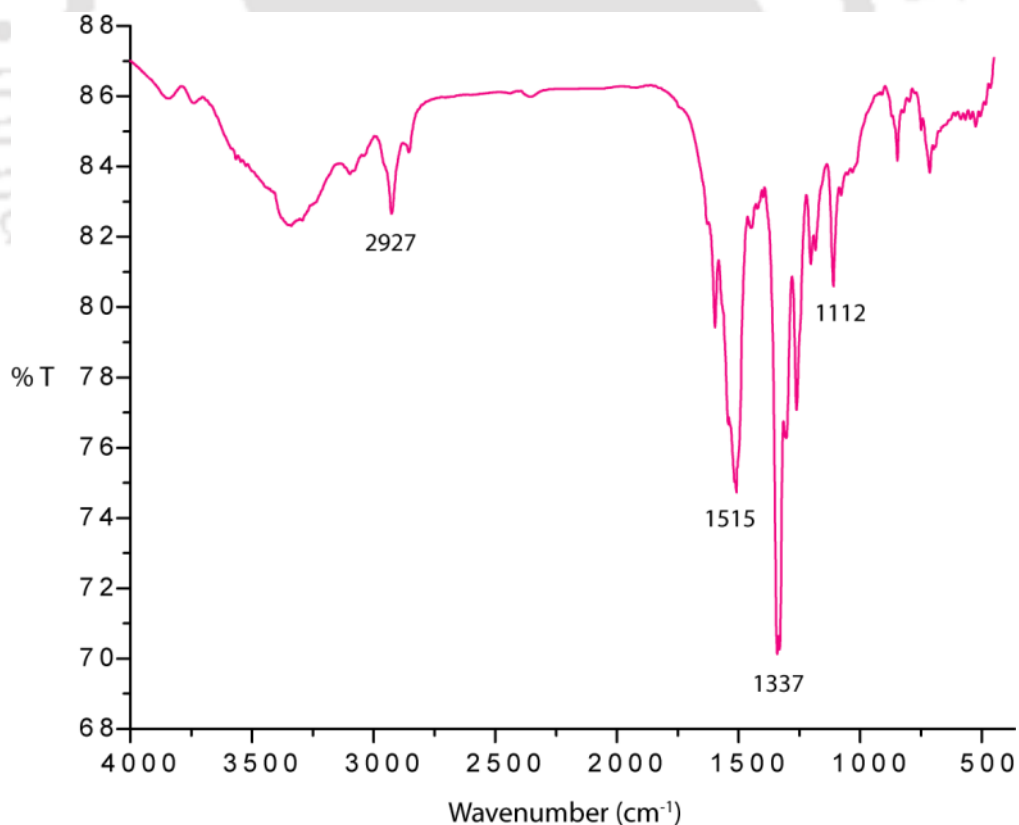
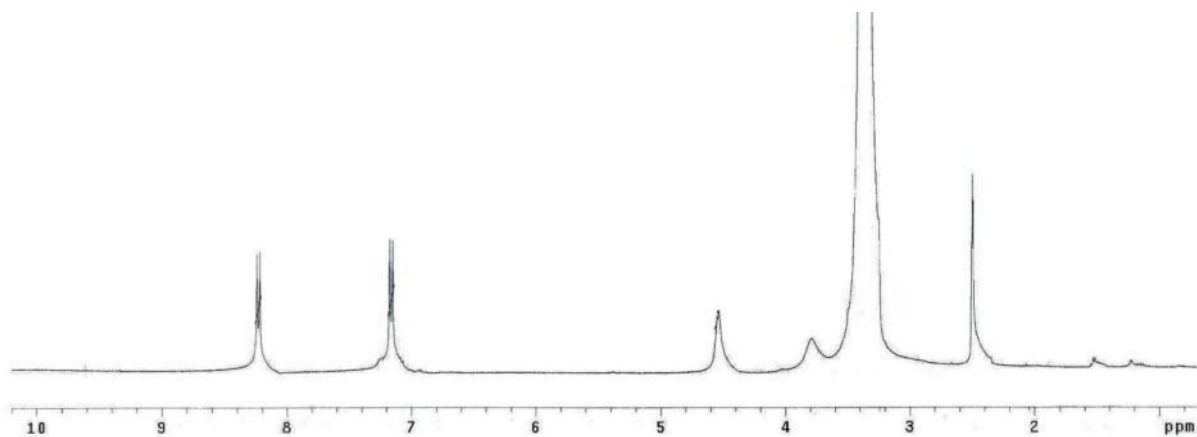
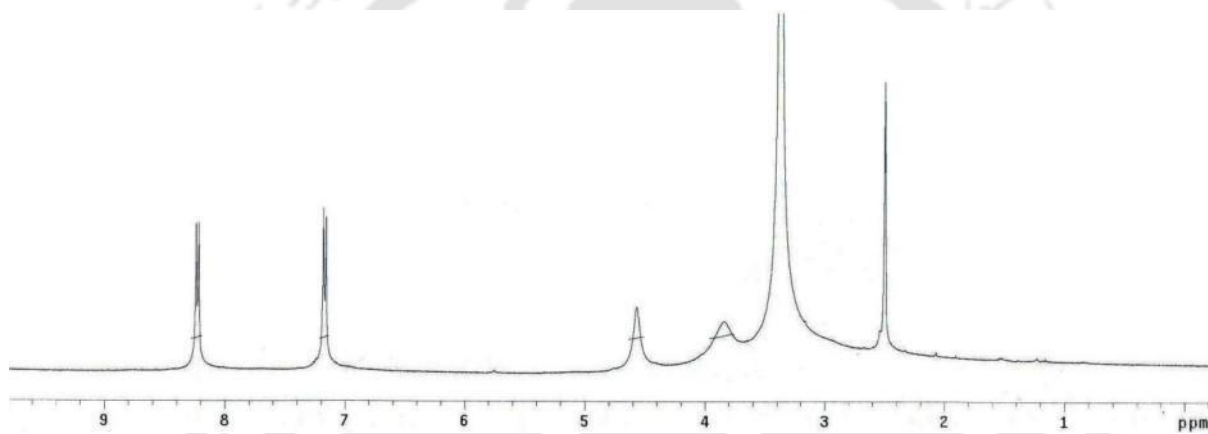
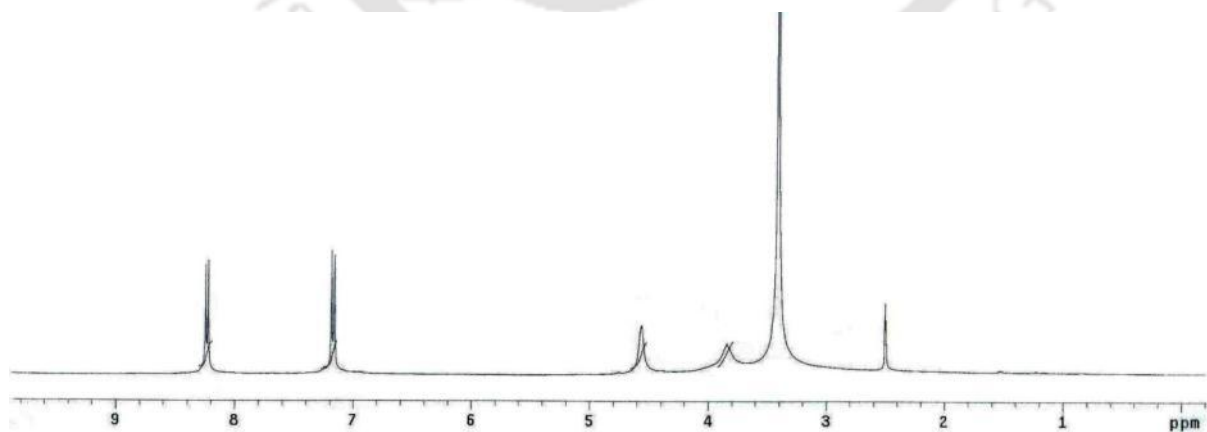


Figure A2.19 FT-IR spectrum of L_5 recorded in KBr pellet.

Characterization data of anion complexes of L₁ (1a-1e):**Figure A2.20** ¹H NMR spectrum of complex [(HL₁)⁺Cl⁻], **1a** (400 MHz, DMSO-*d*₆).**Figure A2.21** ¹H NMR spectrum of complex [(HL₁)⁺Br⁻], **1b** (400 MHz, DMSO-*d*₆).**Figure A2.22** ¹H NMR spectrum of complex [(HL₁)⁺NO₃⁻], **1c** (400 MHz, DMSO-*d*₆).

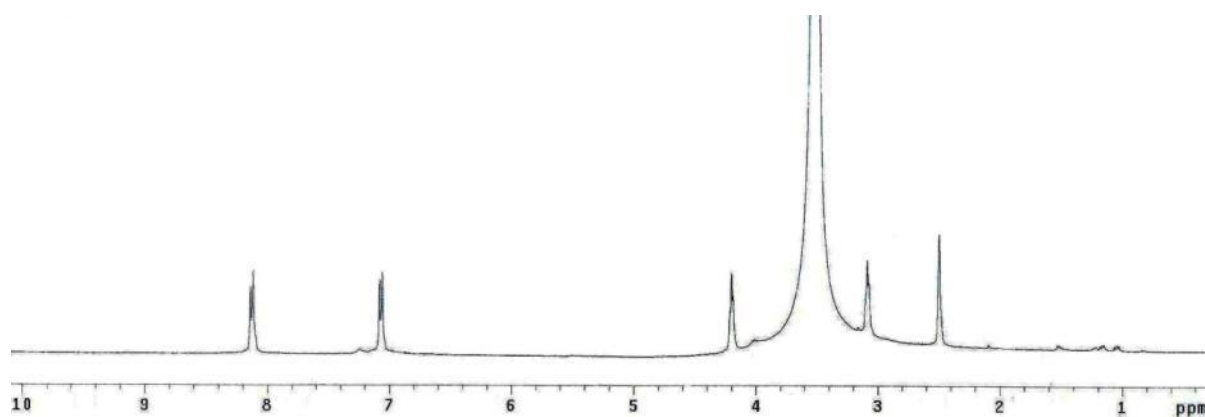


Figure A2.23 ¹H NMR spectrum of complex [(HL₁)⁺CF₃COO⁻], **1d** (400 MHz, DMSO-*d*₆).

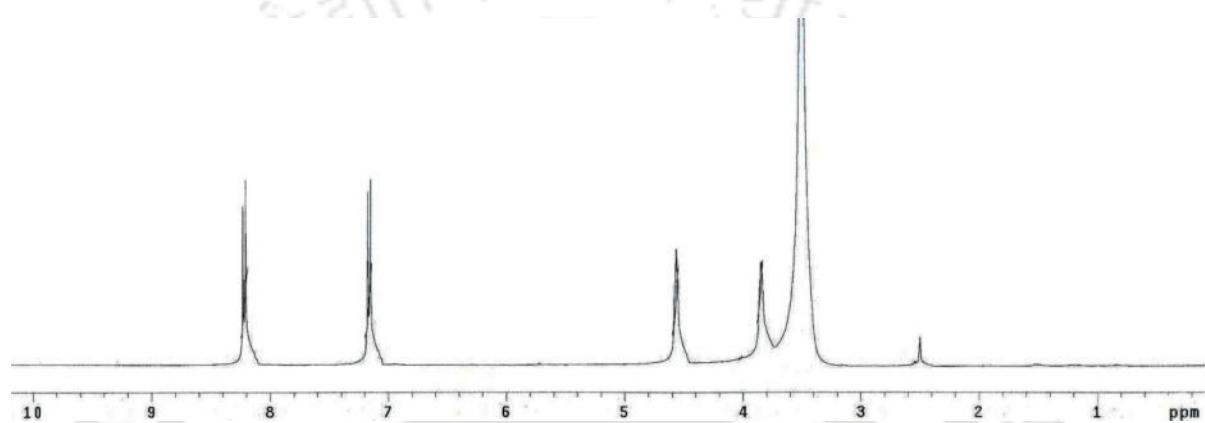


Figure A2.24 ¹H NMR spectrum of complex [(HL₁)⁺ClO₄⁻], **1e** (400 MHz, DMSO-*d*₆).

Characterization data of anion complexes of L₂ (2a-2d):

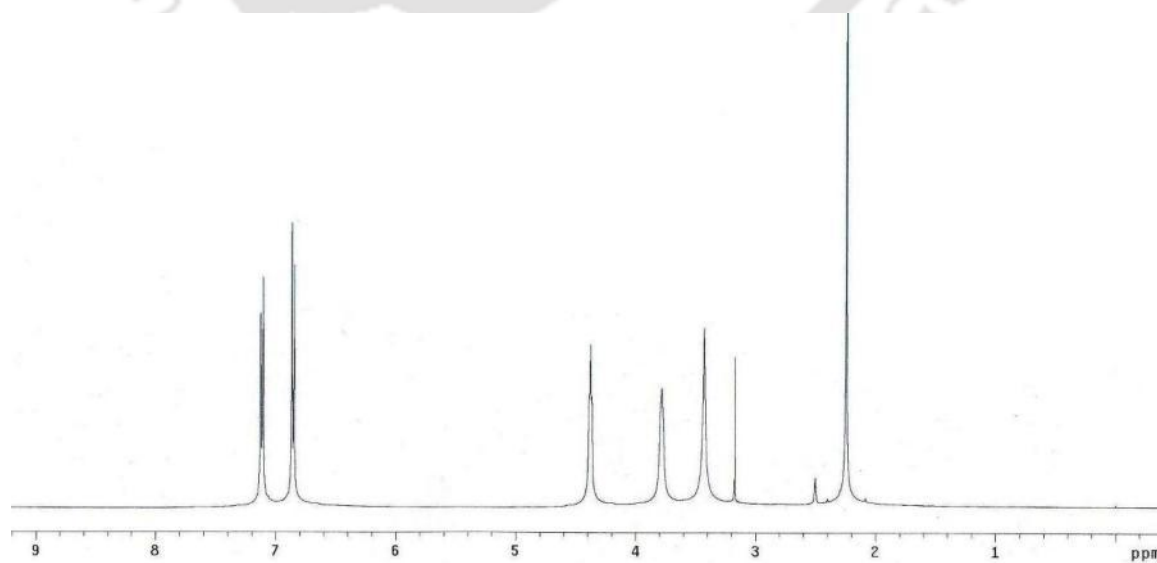


Figure A2.25 ¹H NMR spectrum of complex [(HL₂)⁺ClO₄⁻], **2a** (400 MHz, DMSO-*d*₆).

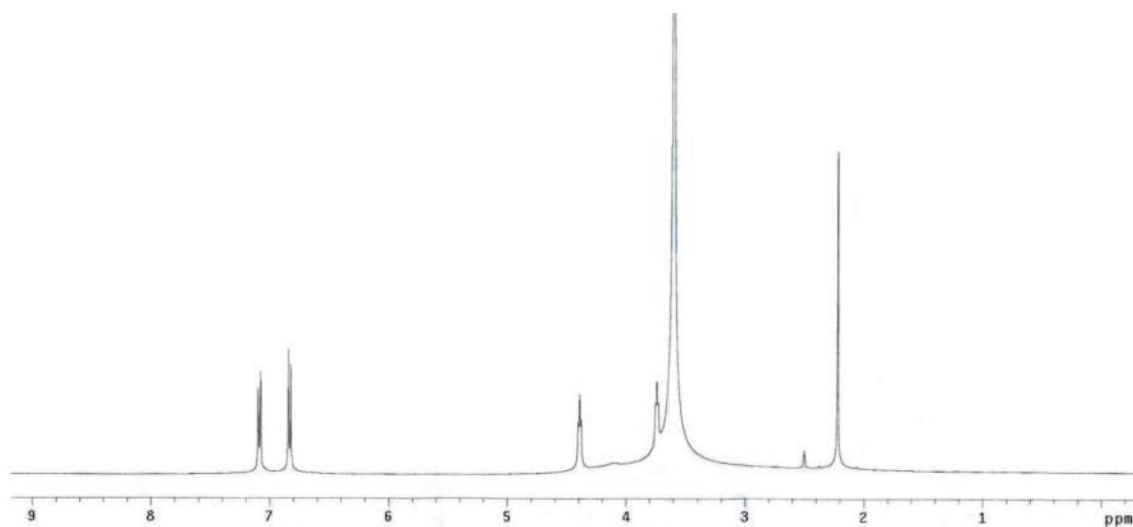


Figure A2.26 ¹H NMR spectrum of complex $[(HL_2)^+Br^-]2H_2O$, **2b** (400 MHz, $DMSO-d_6$).

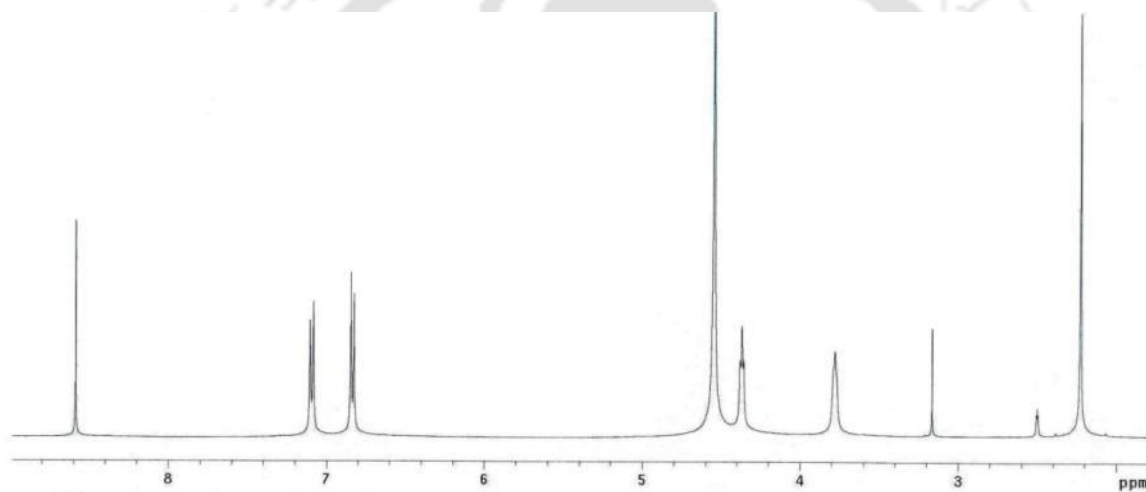


Figure A2.27 ¹H NMR spectrum of complex $[(HL_2)^+PA^-]$, **2c** (400 MHz, $DMSO-d_6$).

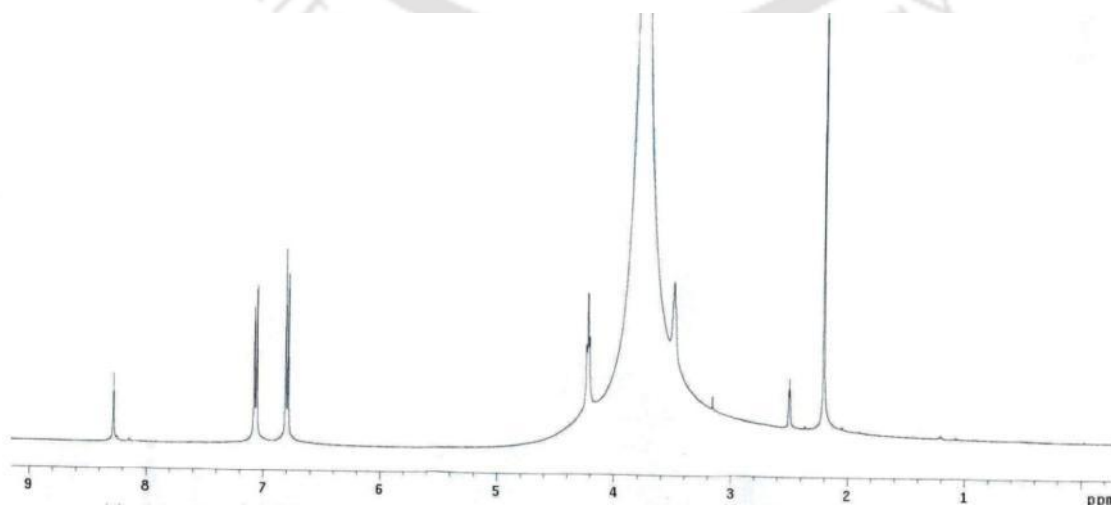


Figure A2.28 ¹H NMR spectrum of complex $[(HL_2)^+PMA^-]$, **2d** (400 MHz, $DMSO-d_6$).

Characterization data of anion complexes of L₃ (3a-3f):

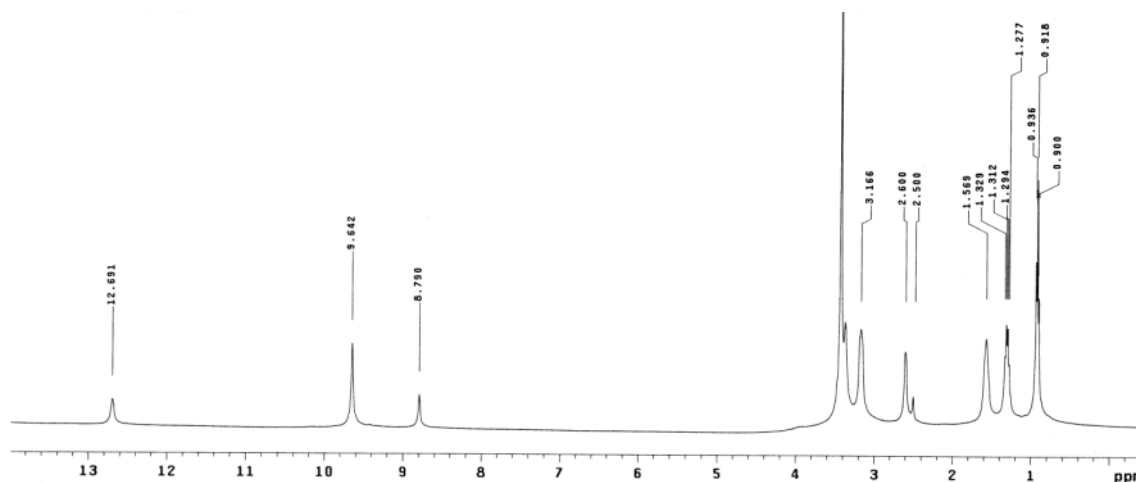


Figure A2.29 ¹H NMR spectrum of complex TBA[L₃(F⁻)], **3a-IV** (400 MHz, DMSO-*d*₆).

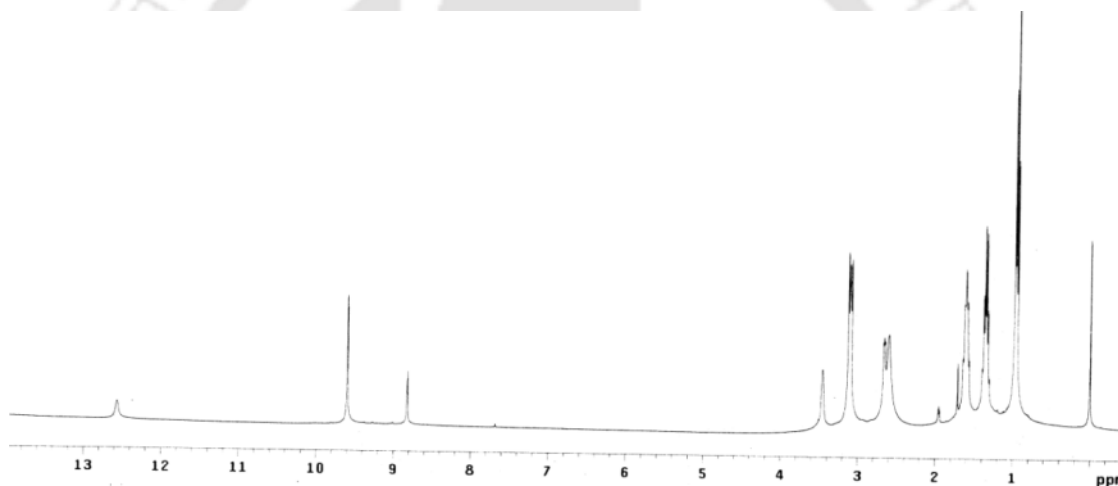


Figure A2.30 ¹H NMR spectrum of complex TBA[L₃(F⁻)], **3a-IV** (400 MHz, CD₃CN).

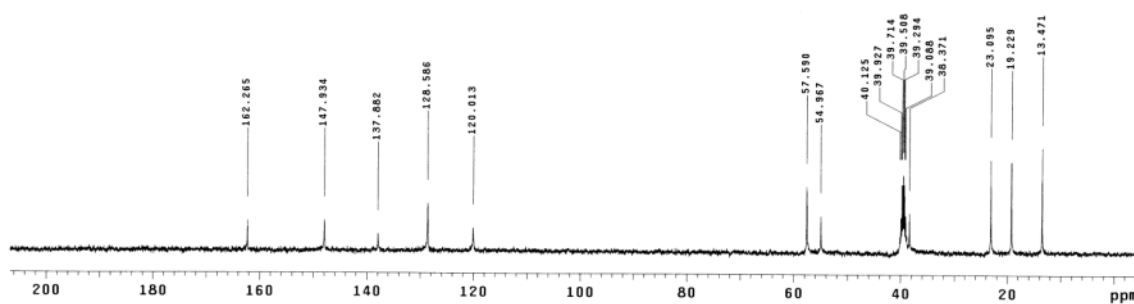


Figure A2.31 ¹³C NMR spectrum of complex TBA[L₃(F⁻)], **3a-IV** (100 MHz, DMSO-*d*₆).

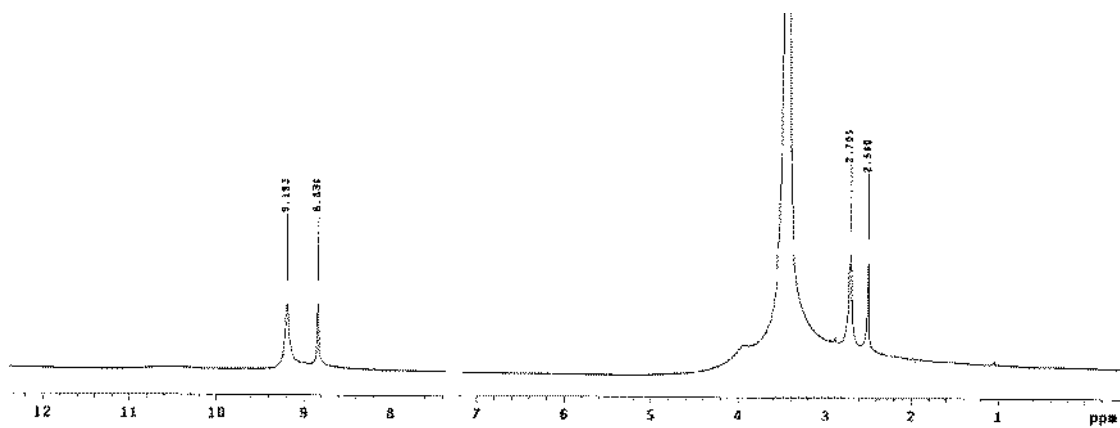


Figure A2.32 ^1H NMR spectrum of complex $[\text{L}_3\text{-}2\text{H}_2\text{OKF}]$, **3b** (400 MHz, $\text{DMSO-}d_6$).

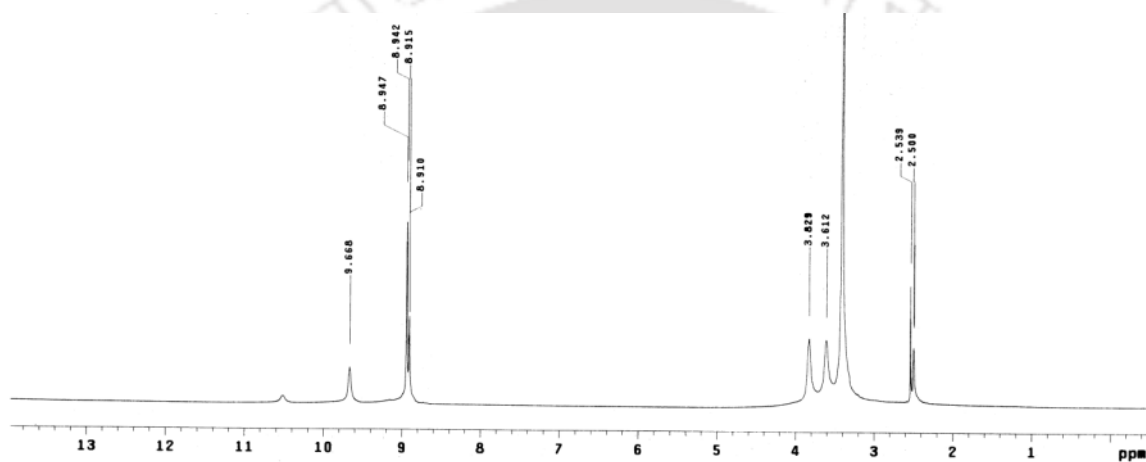


Figure A2.33 ^1H NMR spectrum of complex $[(\text{HL}_3)^+\text{Cl}^-]$, **3c** (400 MHz, $\text{DMSO-}d_6$).

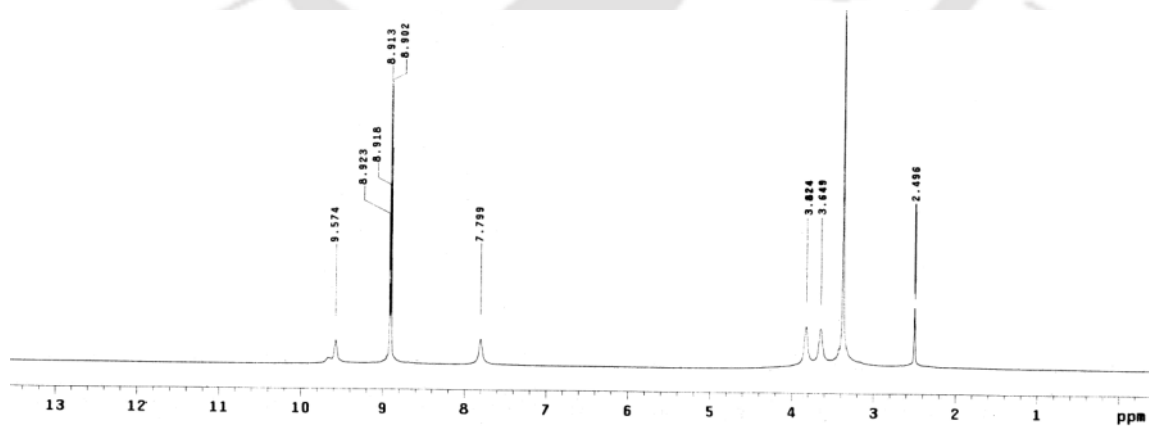


Figure A2.34 ^1H NMR spectrum of complex $[(\text{HL}_3)^+\text{Br}^-]2\text{H}_2\text{O}$, **3d** (400 MHz, $\text{DMSO-}d_6$).

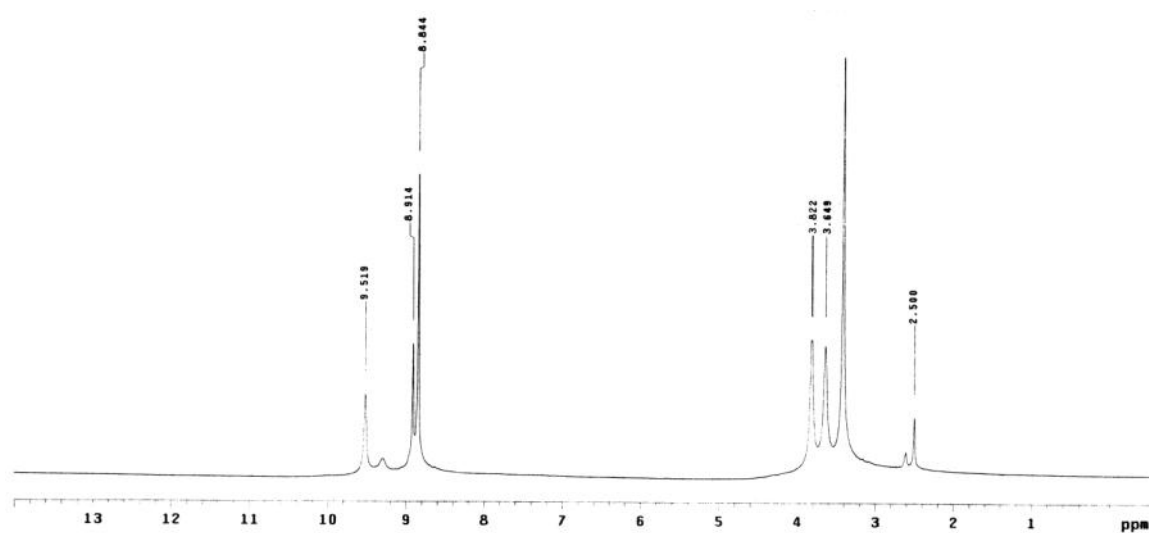


Figure A2.35 ¹H NMR spectrum of complex $[(HL_3)^+ClO_4^-]DMSO \cdot H_2O$, **3e** (400 MHz, DMSO-*d*₆).

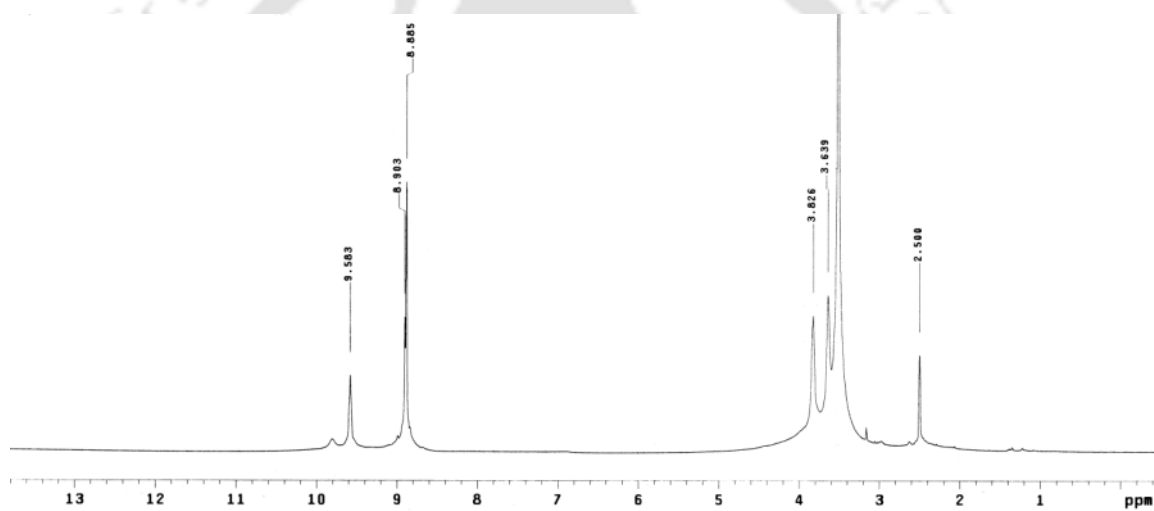
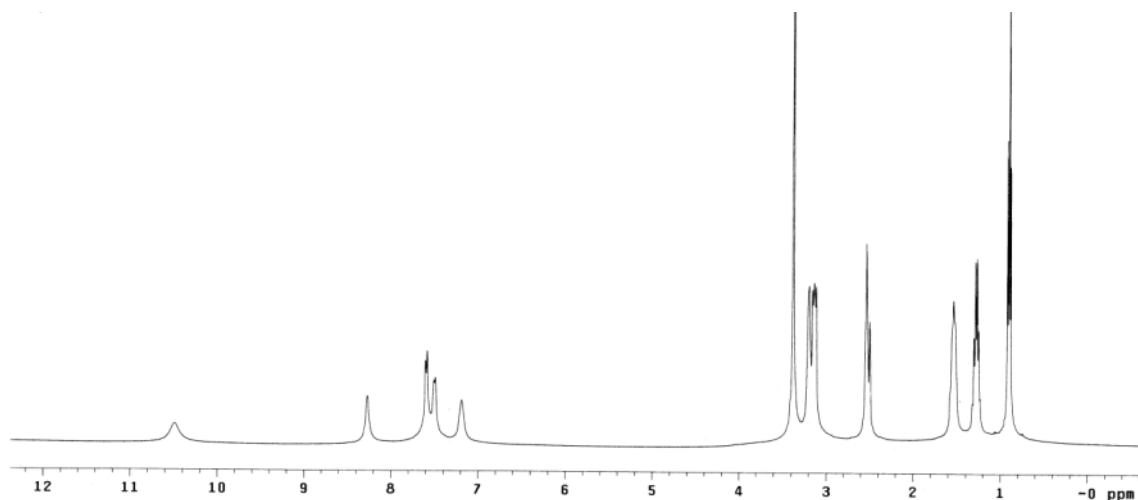
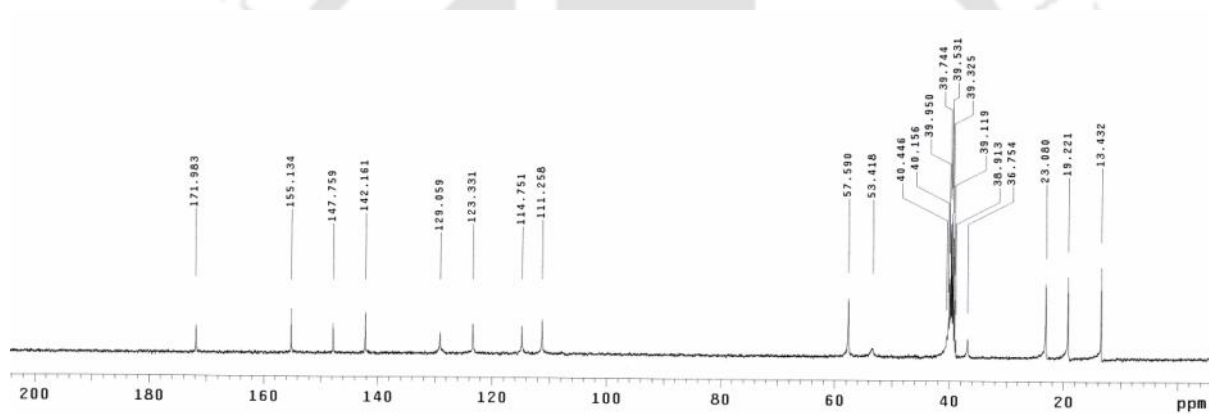
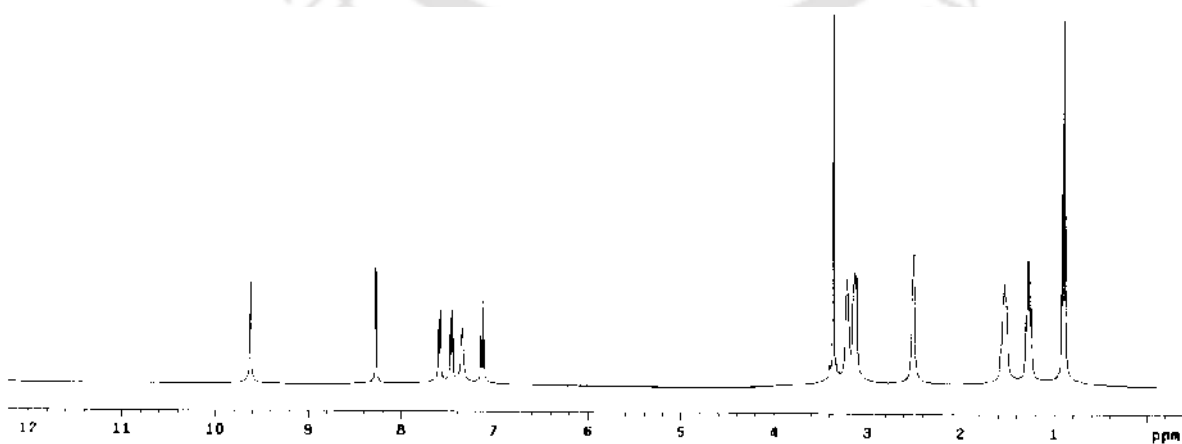


Figure A2.36 ¹H NMR spectrum of complex $[(HL_3)^+HSO_4^-]DMSO$, **3f** (400 MHz, DMSO-*d*₆).

Characterization data of anion complexes of L₄ (4a and 4b):**Figure A2.37** ¹H NMR spectrum of complex 2TBA[2L₄(CO₃)], **4a** (400 MHz, DMSO-*d*₆).**Figure A2.38** ¹³C NMR spectrum of complex 2TBA[2L₄(CO₃)], **4a** (100 MHz, DMSO-*d*₆).**Figure A2.39** ¹H NMR spectrum of complex 2TBA[2L₄(SO₄)], **4b** (400 MHz, DMSO-*d*₆).

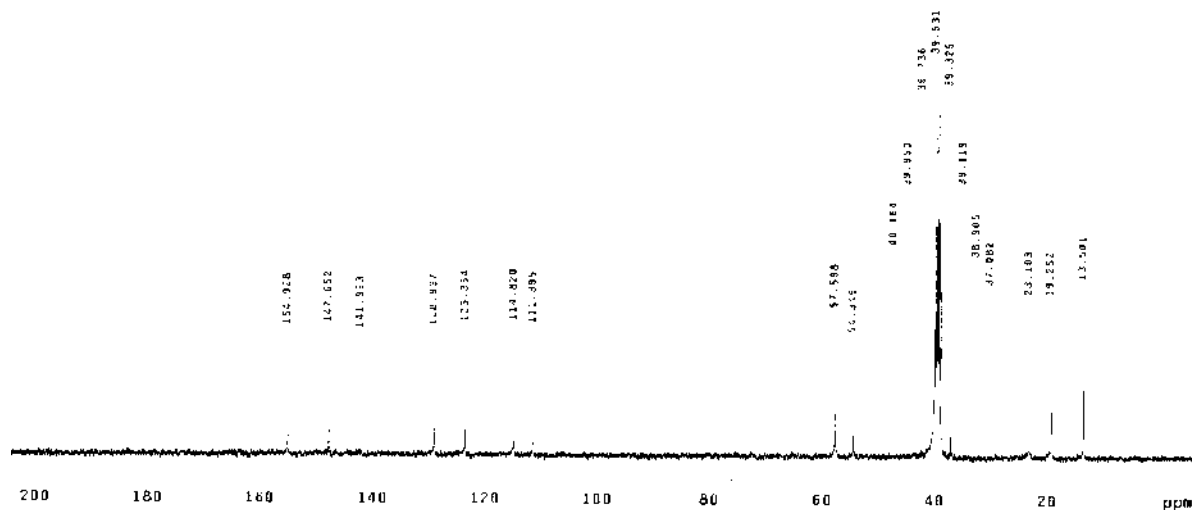


Figure A2.40 ^{13}C NMR spectrum of complex $2\text{TBA}[2\text{L}_4(\text{SO}_4)]$, **4b** (100 MHz, $\text{DMSO-}d_6$).

Characterization data of anion complexes of L_5 (5a-5f):

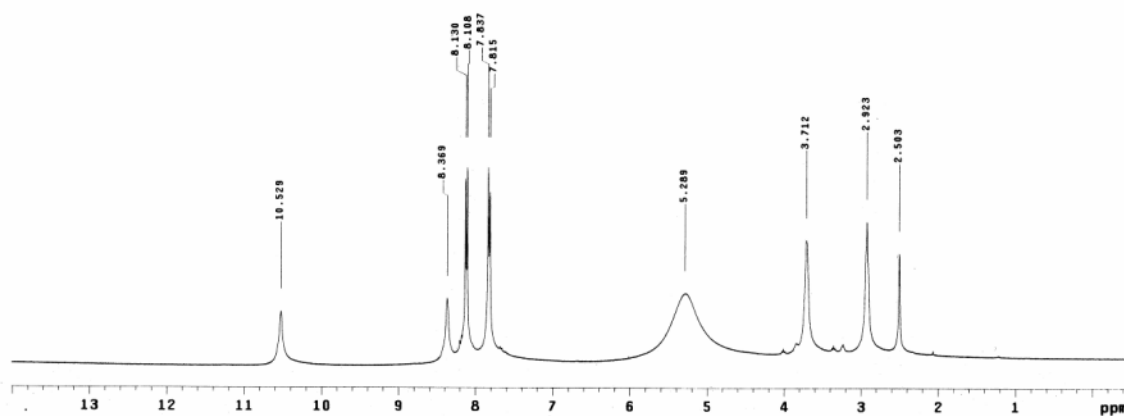


Figure A2.41 ^1H NMR spectrum of complex $[2(\text{HL}_5)^+-\text{HPO}_4^{2-}]\cdot 3\text{H}_2\text{O}$, **5a** (400 MHz, $\text{DMSO-}d_6$).

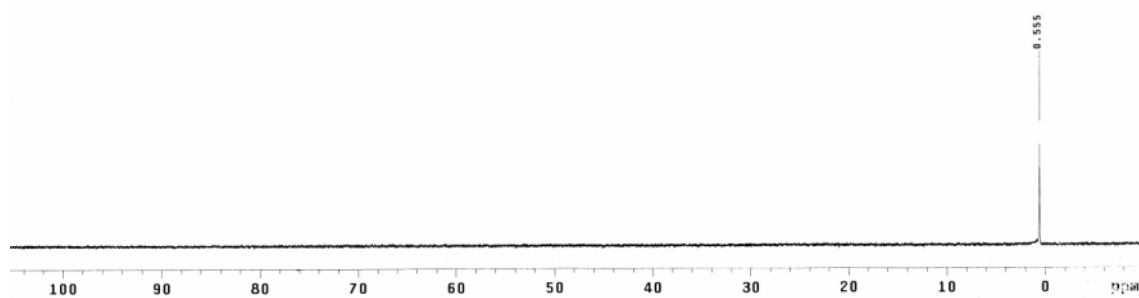


Figure A2.42 ^{31}P NMR spectrum of complex $[2(\text{HL}_5)^+-\text{HPO}_4^{2-}]\cdot 3\text{H}_2\text{O}$, **5a** (400 MHz, $\text{DMSO-}d_6$).

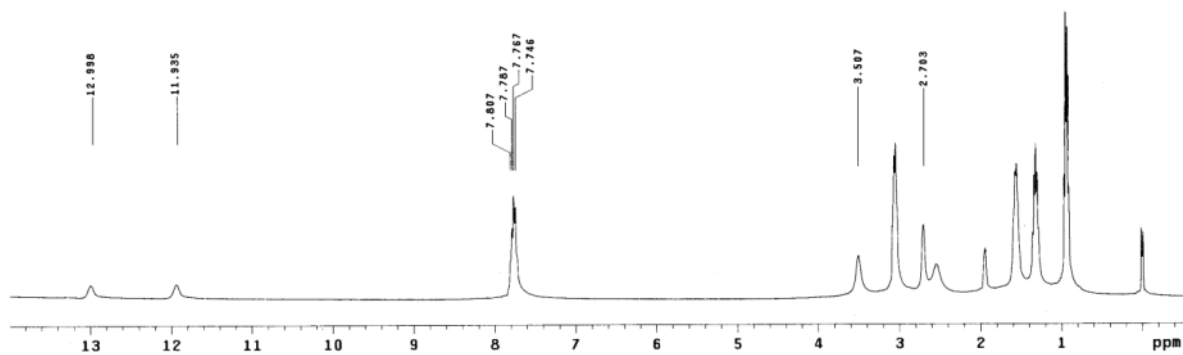


Figure A2.43 ^1H NMR spectrum of complex $3\text{TBA}[2\text{L}_5(\text{PO}_4)]2\text{MeCN}$, **5b-I** (400 MHz, $\text{DMSO-}d_6$).

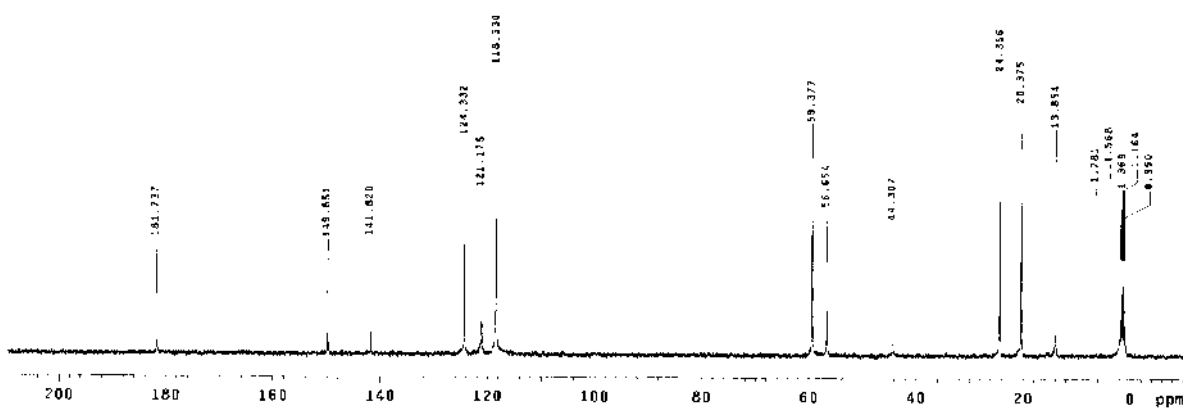


Figure A2.44 ^{13}C NMR spectrum of complex $3\text{TBA}[2\text{L}_5(\text{PO}_4)]2\text{MeCN}$, **5b-I** (100 MHz, $\text{DMSO-}d_6$).

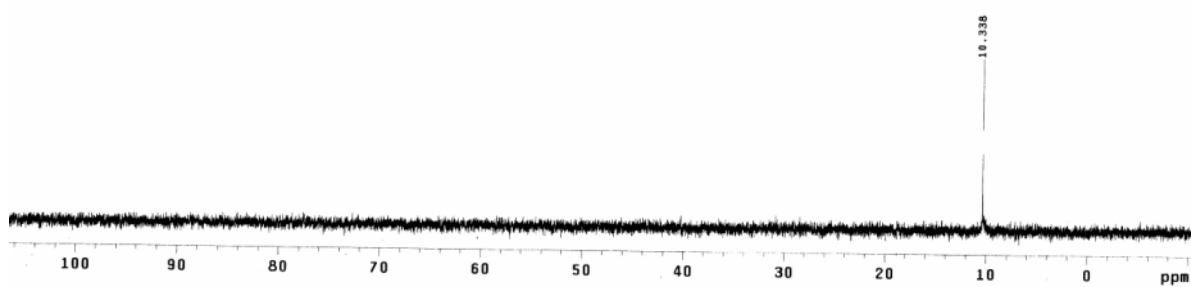


Figure A2.45 ^{31}P NMR spectrum of complex $3\text{TBA}[2\text{L}_5(\text{PO}_4)]2\text{MeCN}$, **5b-I** (400 MHz, $\text{DMSO-}d_6$).

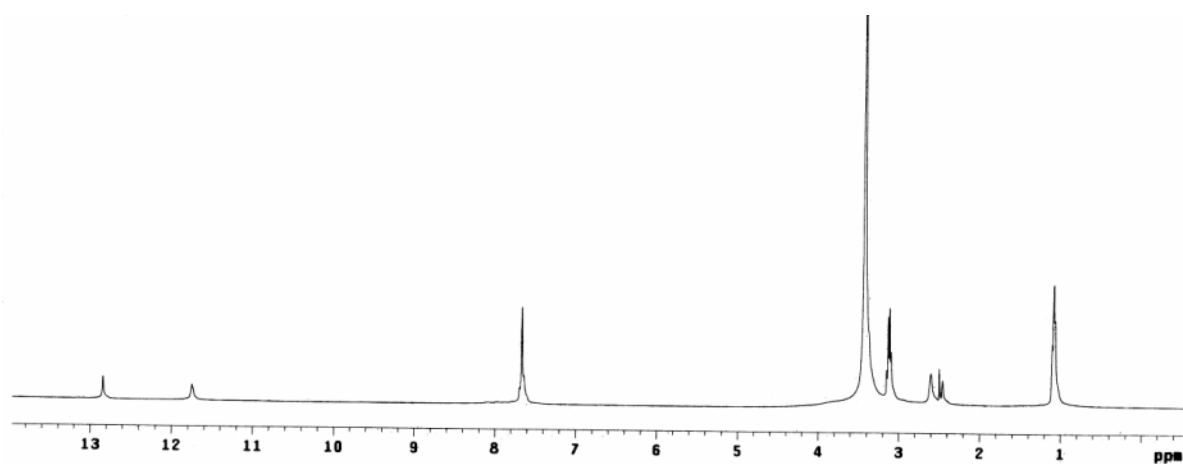


Figure A2.46 ^1H NMR spectrum of complex $3\text{TEA}[2\text{L}_5(\text{PO}_4)]$, **5b-II** (400 MHz, $\text{DMSO-}d_6$).

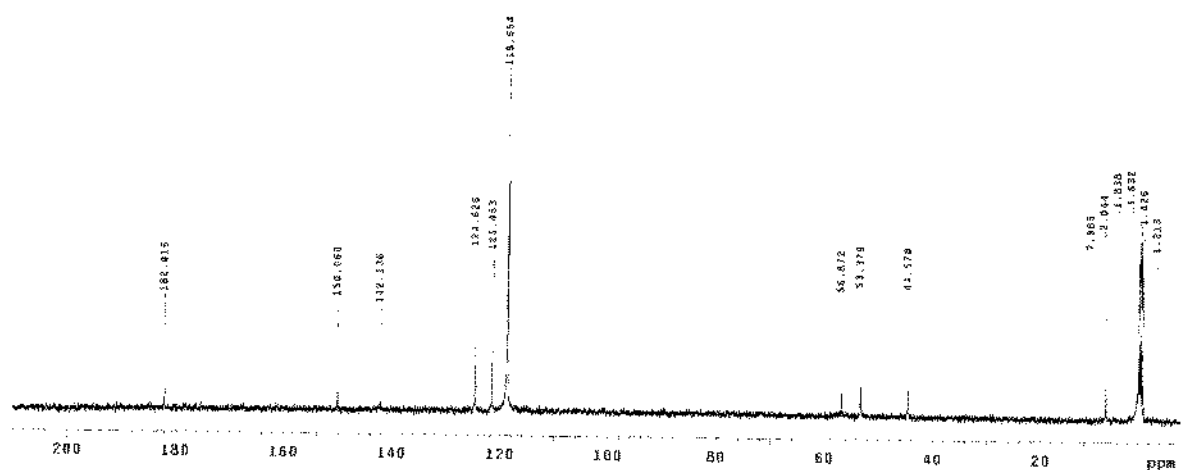


Figure A2.47 ^{13}C NMR spectrum of complex $3\text{TEA}[2\text{L}_5(\text{PO}_4)]$, **5b-II** (100 MHz, CD_3CN).

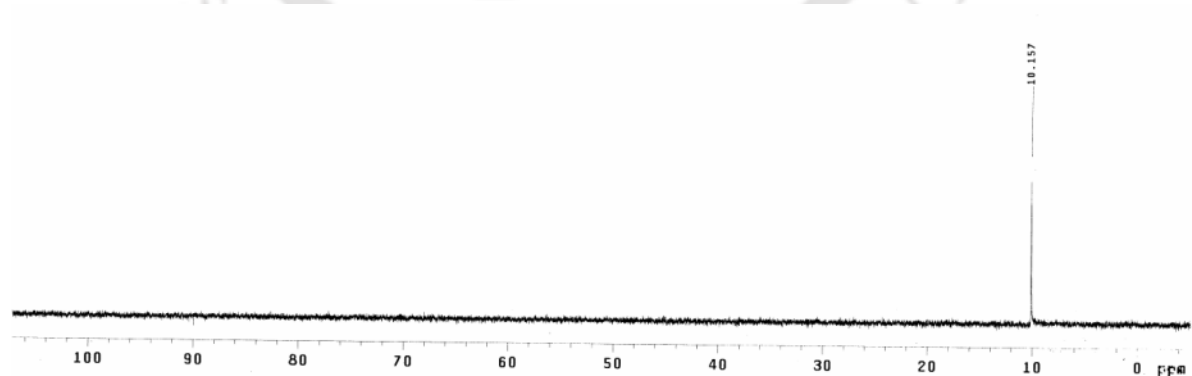


Figure A2.48 ^{31}P NMR spectrum of complex $3\text{TEA}[2\text{L}_5(\text{PO}_4)]$, **5b-II** (400 MHz, $\text{DMSO-}d_6$).

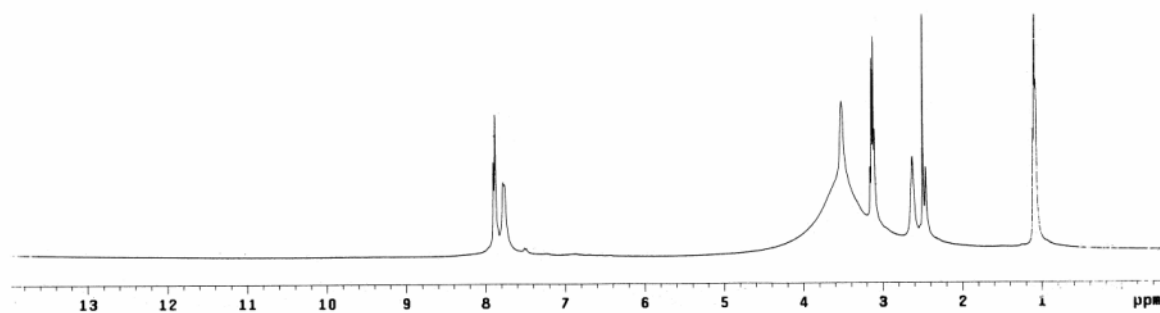


Figure A2.49 ^1H NMR spectrum of complex $2\text{TEA}[2\text{L}_5(\text{CO}_3)]$, **5c** (400 MHz, $\text{DMSO-}d_6$).

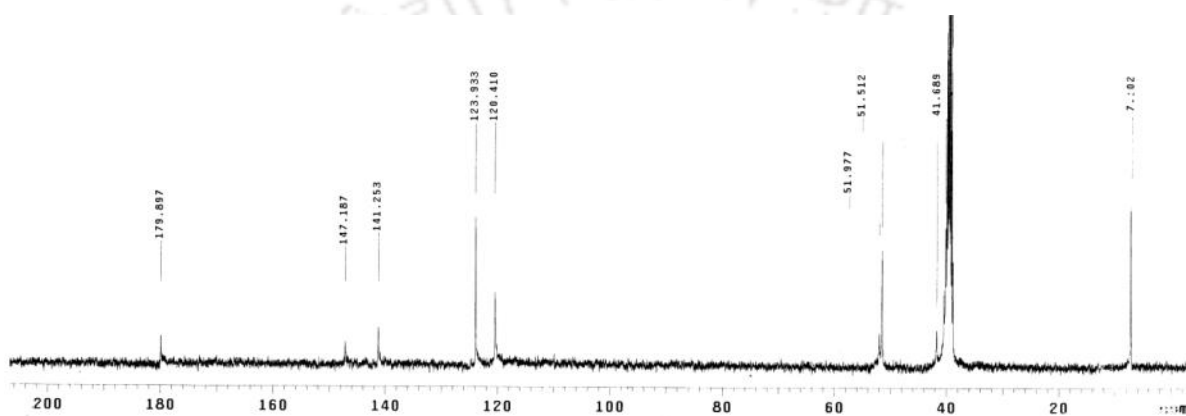


Figure A2.50 ^{13}C NMR spectrum of complex $2\text{TEA}[2\text{L}_5(\text{CO}_3)]$, **5c** (100 MHz, $\text{DMSO-}d_6$).

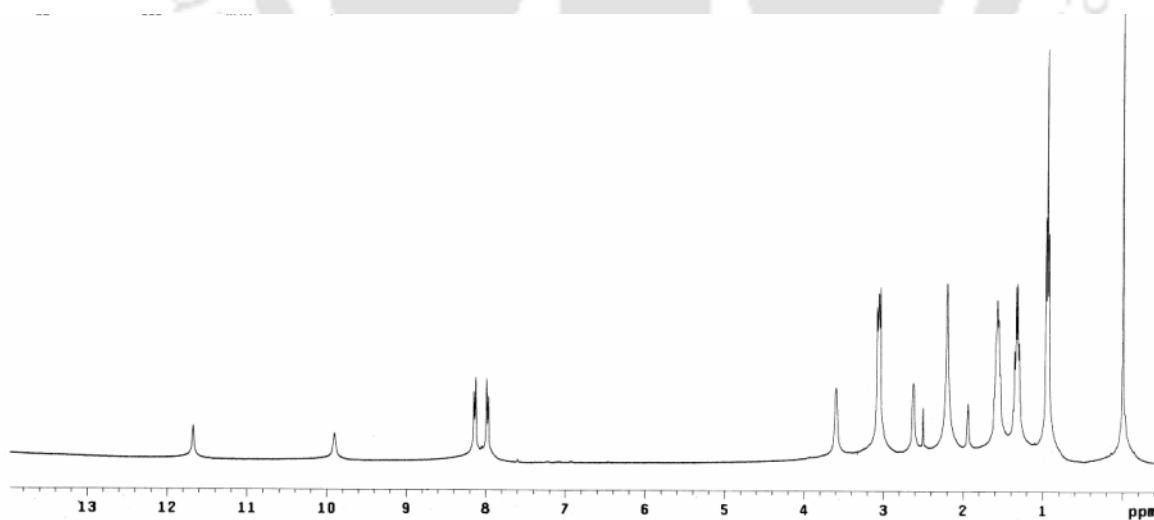


Figure A2.51 ^1H NMR spectrum of complex $2\text{TBA}[\text{L}_5(\text{SO}_4)]$, **5d** (400 MHz, $\text{DMSO-}d_6$).

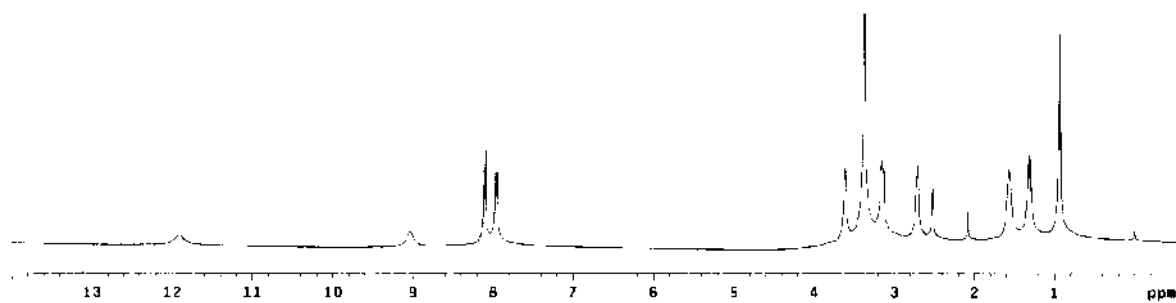


Figure A2.52 ^1H NMR spectrum of complex TBA[L₅(F)]₂DMSO, 5e-I (400 MHz, CD₃CN).

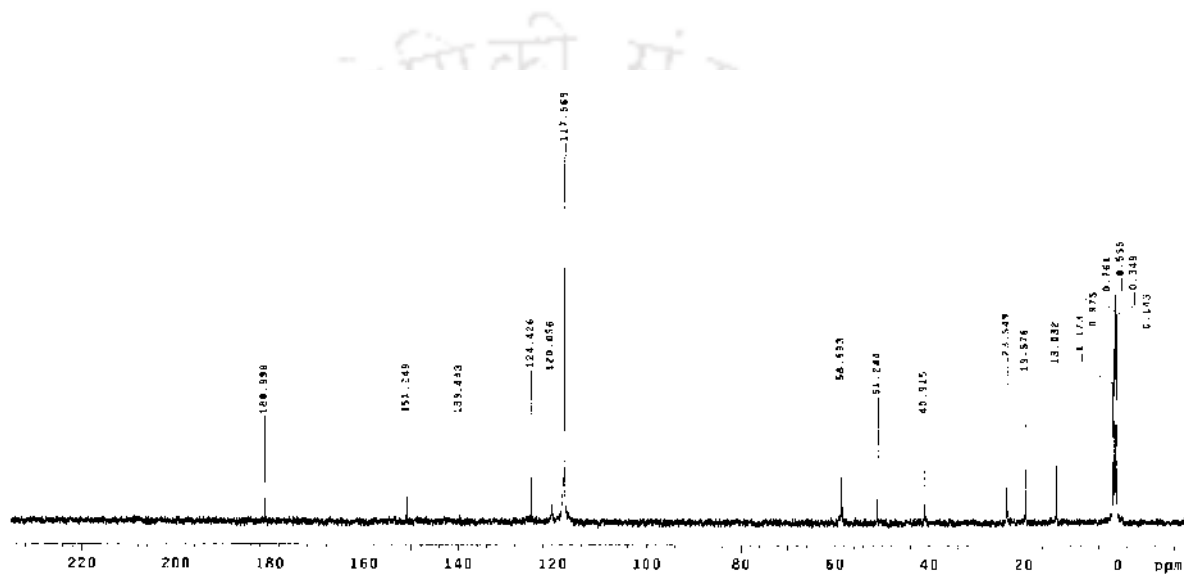


Figure A2.53 ^{13}C NMR spectrum of complex TBA[L₅(F)]₂DMSO, 5e-I (100 MHz, CD₃CN).

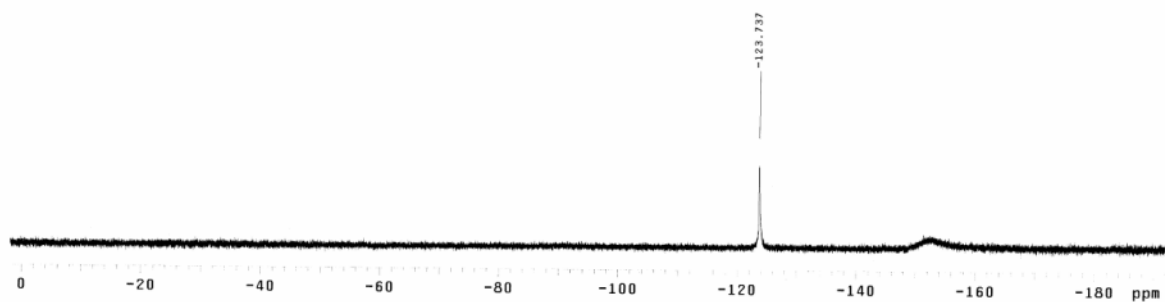


Figure A2.54 ^{19}F NMR spectrum of complex TBA[L₅(F)]₂DMSO, 5e-I (400 MHz, CD₃CN).

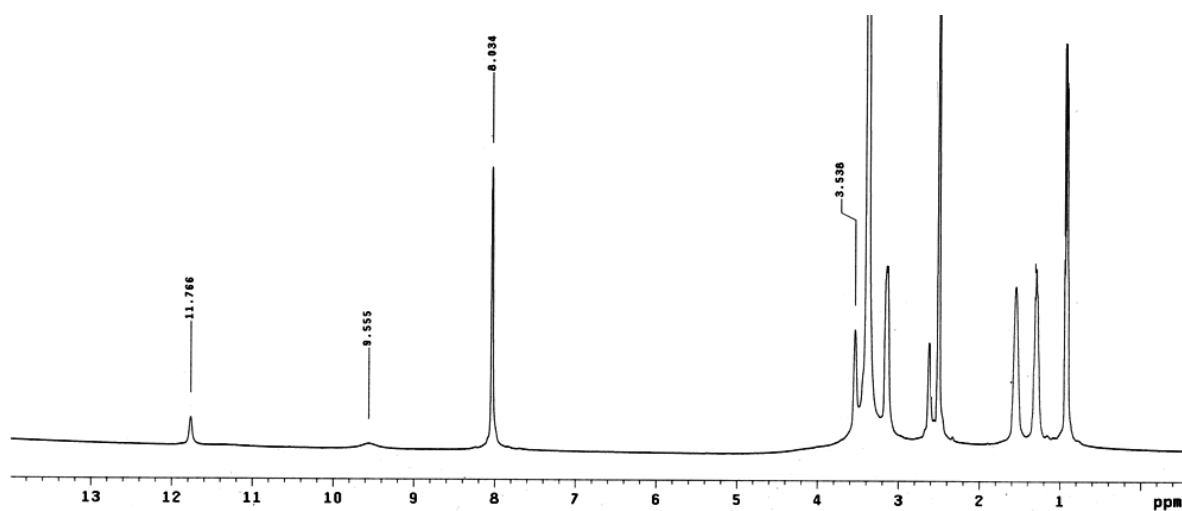


Figure A2.55 ^1H NMR spectrum of complex $\text{TBA}[\text{L}_5(\text{F})]\text{MeCN}$, **5e-II** (400 MHz, $\text{DMSO-}d_6$).

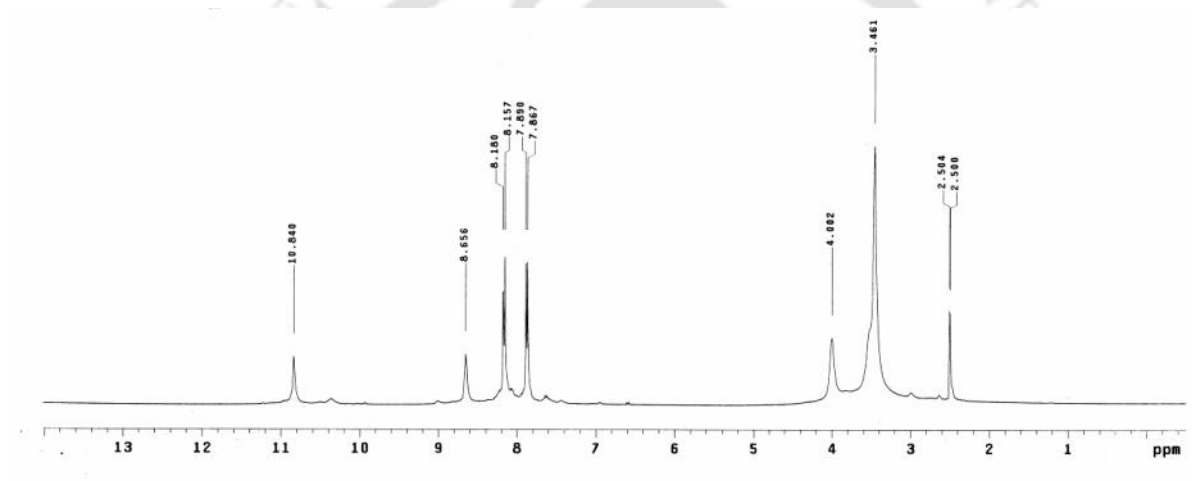


Figure A2.56 ^1H NMR spectrum of complex $[(\text{HL}_5)^+\text{Cl}^-]\text{DMF}$, **5f** (400 MHz, $\text{DMSO-}d_6$).

Annexure 3

Table A3.1 Intermolecular noncovalent interactions in polymorphs **I**, **II** and **III**.

| D-H...A | $d(\text{H}\cdots\text{A})$ (Å) | $d(\text{D}\cdots\text{A})$ (Å) | $\angle(\text{DHA})$ (deg) |
|----------------------|---------------------------------|---------------------------------|----------------------------|
| Polymorph-I | | | |
| C24-H...O3 | 2.674 | 3.557 | 158.63 |
| C17-H(A)...O5 | 2.566 | 3.520 | 167.72 |
| C4-H...O6 | 2.596 | 3.283 | 131.03 |
| C8-H...O8 | 2.431 | 3.156 | 134.86 |
| C15-H...O9 | 2.629 | 3.424 | 143.80 |
| C23-H...O9 | 2.570 | 3.135 | 119.51 |
| Polymorph-II | | | |
| C8-H...O6 | 2.443 | 3.349 | 164.28 |
| C9-H(A)...O8 | 2.587 | 3.506 | 158.35 |
| C16-H...O9 | 2.684 | 3.549 | 155.24 |
| Polymorph-III | | | |
| C5-H...O8 | 2.515 | 3.285 | 140.27 |
| C2-H(A)...O5 | 2.515 | 3.373 | 147.29 |
| C2-H(B)...O5 | 2.672 | 3.275 | 120.66 |
| C10-H(A)...O3 | 2.629 | 3.399 | 136.50 |
| C16-H...O3 | 2.631 | 3.221 | 121.87 |

Table A3.2 Polymorph Screening of podand **L₁**.

| Solvent(s) | Polymorph |
|-------------------------------|-------------|
| DMSO | I |
| Acetonitrile | I |
| Ethanol | I/II |
| Methanol | I/II |
| Ethyl acetate | I/II |
| DCM | I/II |
| Chloroform | I/II |
| THF | I/II |
| Ethanol/Acetic acid (3:1) | II |
| DMSO/Acetic acid (3:1) | III |
| Ethanol (refluxed and cooled) | III |

**Figure A3.1** Crystal habit of polymorphs **I**, **II** and **III**.

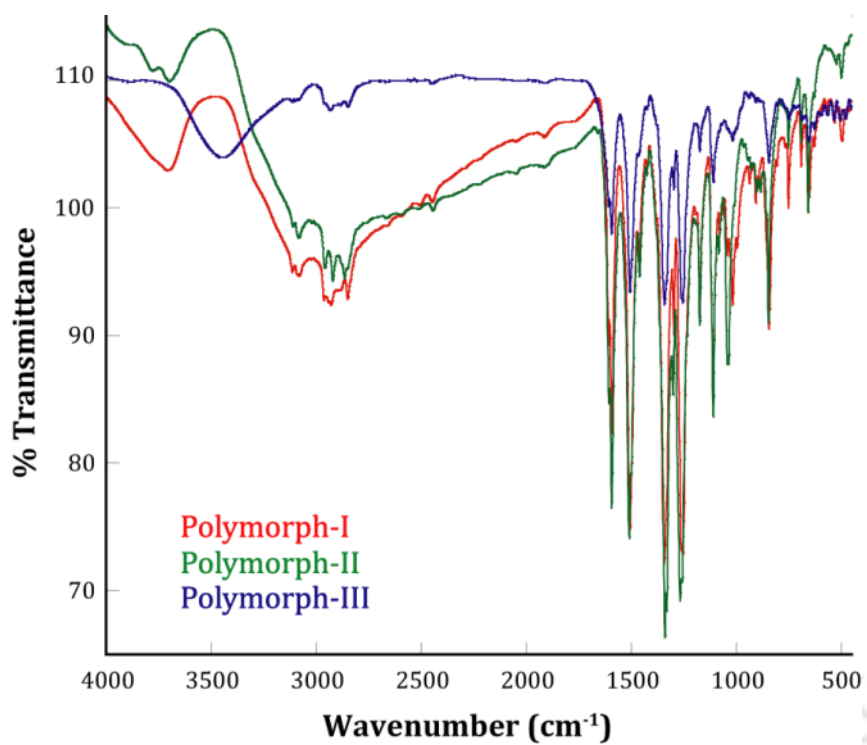


Figure A3.2 FT-IR spectra of polymorphs I, II and III.

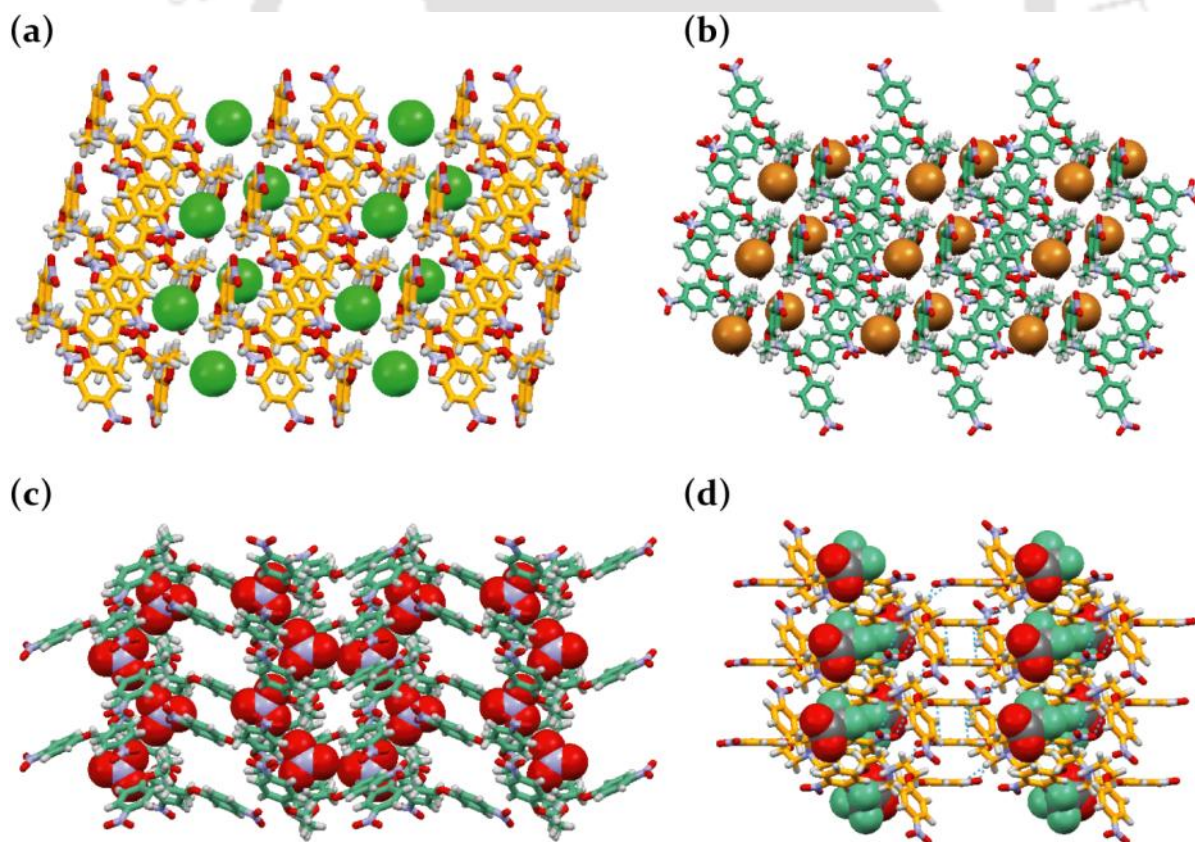


Figure A3.3 Crystal packing diagrams of anion complexes of protonated L₁, 1a-1d.

Table A3.3 Intermolecular C–H···O-(nitro), C–H··· π and π -stacking interactions in anion complexes of protonated **L**₁ (**1a-1e**).

| Complexes | D-H···A | <i>d</i> (H···A) (Å) | <i>d</i> (D···A) (Å) | \angle (DHA) (°) |
|---------------|---------------|----------------------|----------------------|--------------------|
| 1a | C16-H···O2 | 2.51 | 3.390 | 156 |
| | C2-H(B)···O3 | 2.60 | 3.229 | 121 |
| | C10-H(A)···O3 | 2.59 | 3.560 | 178 |
| | C8-H···O3 | 2.33 | 3.222 | 160 |
| | C17-H(B)···O5 | 2.70 | 3.349 | 124 |
| | C20-H···O6 | 2.52 | 3.428 | 164 |
| | C7-H···C2g | 3.03 | 3.970 | 157 |
| | C1g···C1g | | 4.183 | |
| | C3g···C3g | | 3.610 | |
| | C1g···C3g | | 3.858 | |
| 1b | C24-H···O3 | 2.63 | 3.499 | 162 |
| | C2-H(A)···O5 | 2.55 | 3.559 | 172 |
| | C10-H(A)···O5 | 2.59 | 3.263 | 125 |
| | C16-H···O5 | 2.42 | 3.328 | 163 |
| | C8-H···O6 | 2.52 | 3.415 | 160 |
| | C23-H···O9 | 2.70 | 3.504 | 150 |
| | C15-H···C1g | 3.05 | 3.924 | 156 |
| | C3g···C3g | | 3.601 | |
| | C3g···C2g | | 3.906 | |
| 1c | C1-H(B)···O3 | 2.70 | 3.594 | 152 |
| | C24-H···O3 | 2.60 | 3.323 | 134 |
| | C2-H(A)···O5 | 2.60 | 3.472 | 148 |
| | C17-H(B)···O5 | 2.67 | 3.644 | 175 |
| | C4-H···O6 | 2.48 | 3.399 | 168 |
| | C10-H(A)···O6 | 2.58 | 3.256 | 126 |
| | C8-H···O9 | 2.40 | 3.089 | 130 |
| | 1d | C9-H(A)···O2 | 2.56 | 3.324 |
| C10-H(A)···O3 | | 2.67 | 3.620 | 164 |
| C17-H(B)···O5 | | 2.68 | 3.615 | 160 |
| C17-H(B)···O6 | | 2.66 | 3.497 | 143 |
| C20-H···O6 | | 2.56 | 3.155 | 121 |
| C1g···C1g | | | 3.726 | |
| C2g···C2g | | | 3.518 | |
| 1e | C17-H(A)···O2 | 2.54 | 3.367 | 142 |
| | C10-H(B)···O3 | 2.71 | 3.594 | 151 |
| | C17-H(B)···O3 | 2.33 | 3.235 | 154 |
| | C9-H(B)···O5 | 2.51 | 3.373 | 147 |
| | C24-H···O6 | 2.54 | 3.431 | 144 |
| | C8-H···O8 | 2.68 | 3.563 | 157 |

Annexure 4

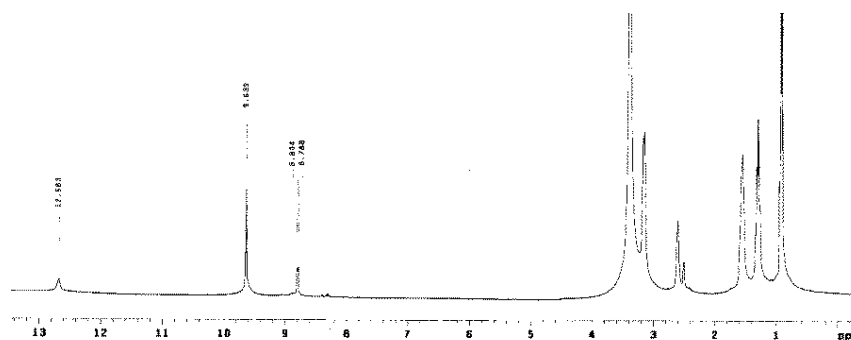


Figure A4.1 ¹H NMR spectrum of L₃ (DMSO-*d*₆) in presence of equivalent amount of TBAF.

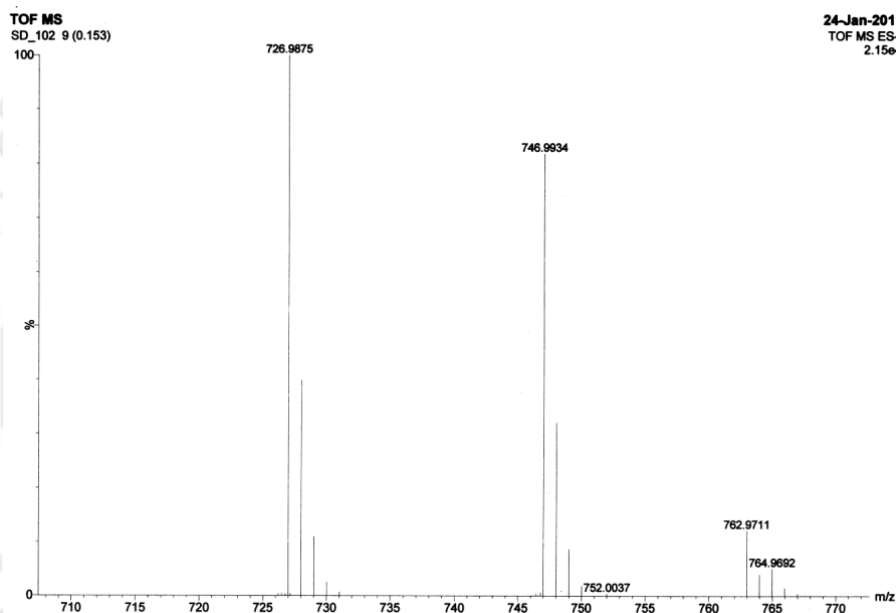


Figure A4.2 ESI-mass spectrum of L₃ in presence of excess TBAF in acetonitrile (MeCN) showing $m/z = 746.99$ for the fluoride-encapsulated complex.

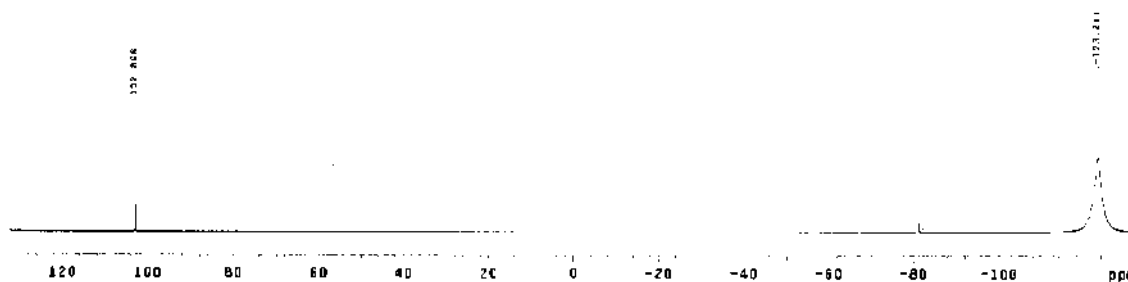


Figure A4.3 ¹⁹F NMR spectrum of TBAF in DMSO-*d*₆.

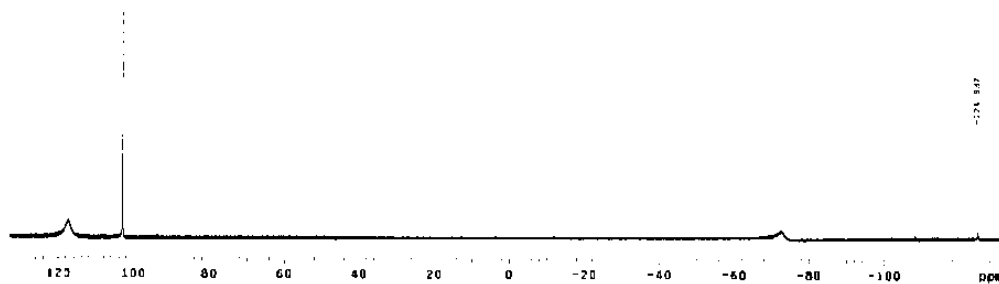


Figure A4.4 ^{19}F NMR spectrum of TBAF ($\text{DMSO-}d_6$) in presence of equivalent amount of L_3 .

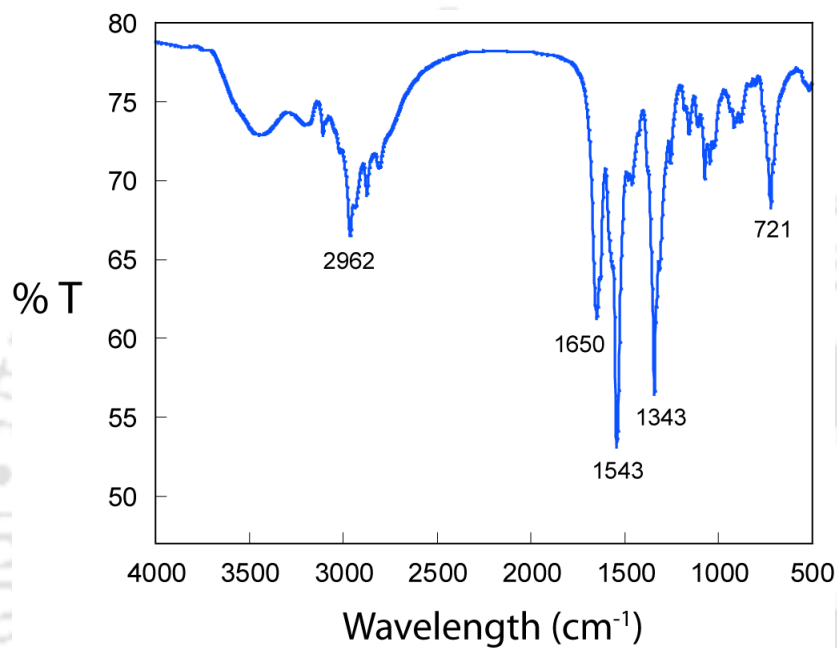


Figure A4.5 FT-IR spectrum of solvate $3a\text{-I}$ recorded in KBr pellet.

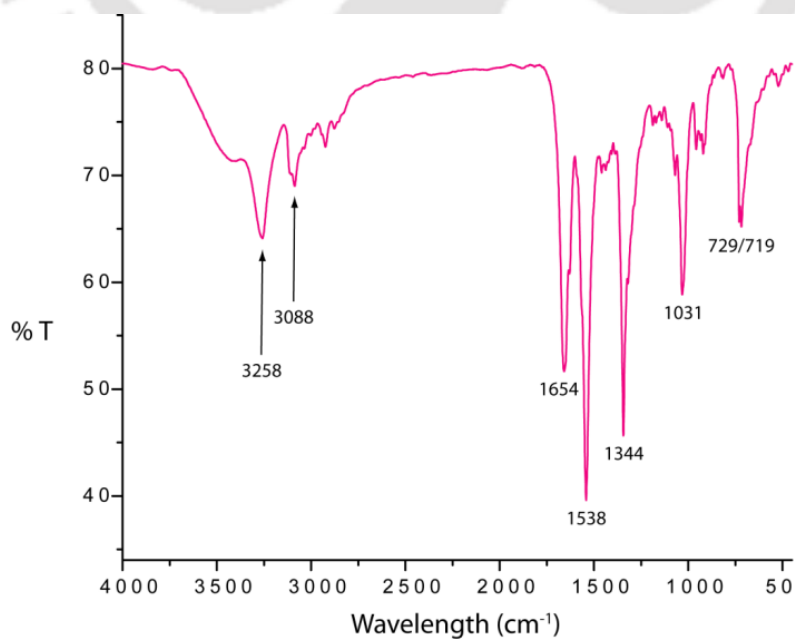


Figure A4.6 FT-IR spectrum of solvate $3a\text{-IV}$ recorded in KBr pellet.

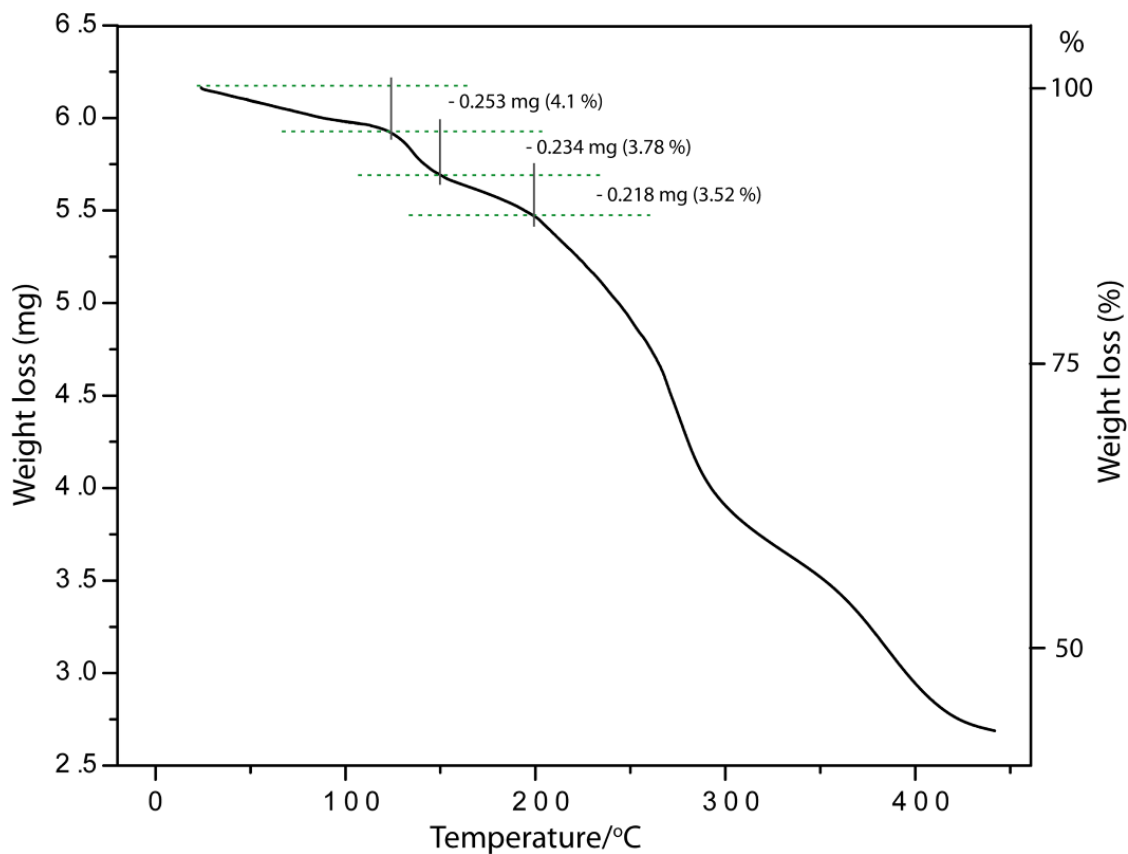


Figure A4.7 Thermo gravimetric curve of complex TBA[L(F)]•3THF (**3a-III**) at a heating rate of 10 °C/min in a N₂ atmosphere.

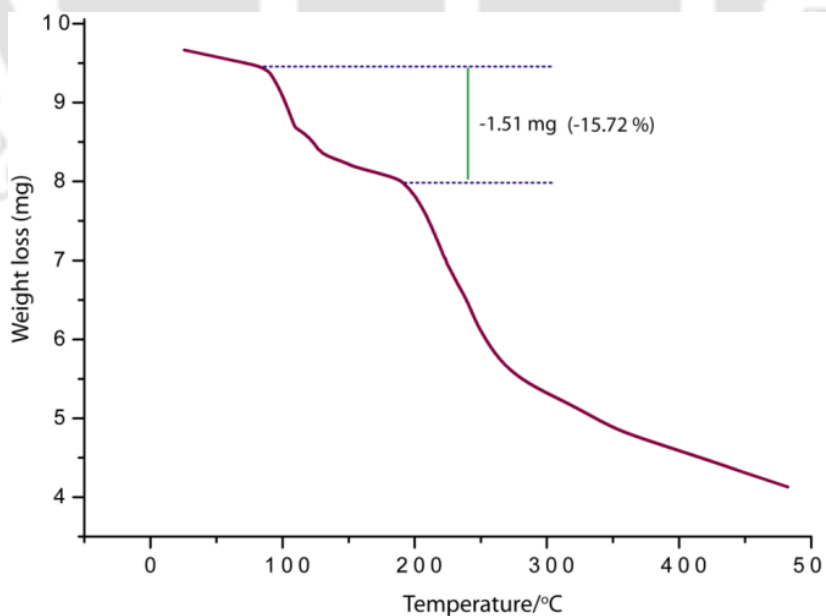


Figure A4.8 Thermo gravimetric curve of complex TBA[L(F)]•2DMSO•4H₂O (**3a-IV**) at a heating rate of 5°C/min in a N₂ atmosphere.

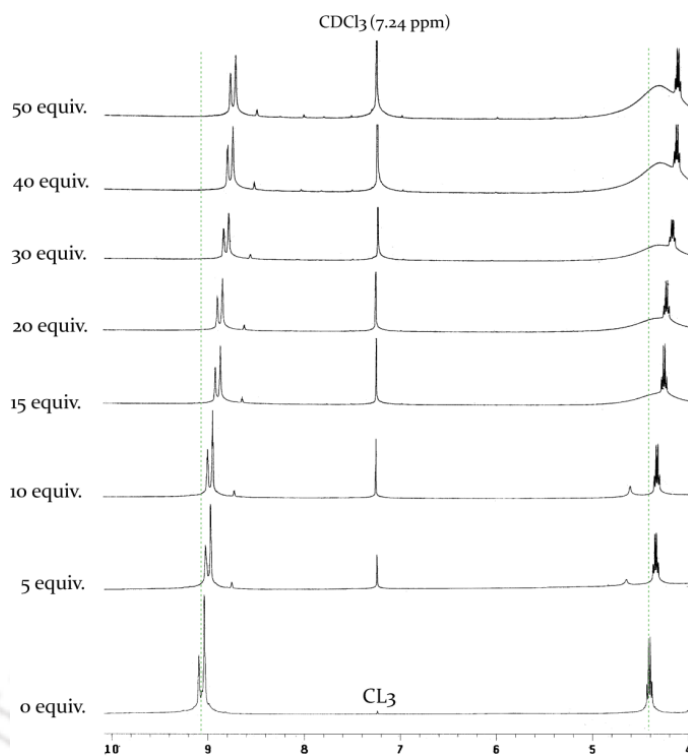


Figure A4.9 Partial ^1H NMR spectra showing the titration of CL_3 (CDCl_3) upon increasing equivalents of TBAF.

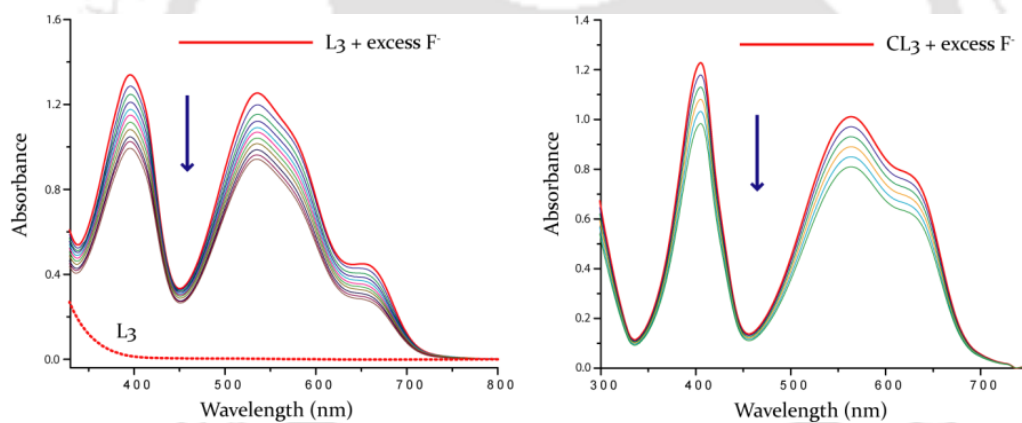


Figure A4.10 UV-Vis titration curves of L_3 (left) and CL_3 (right) charged with excess F^- in dry DMSO upon gradual addition of water (up to 1 mL).

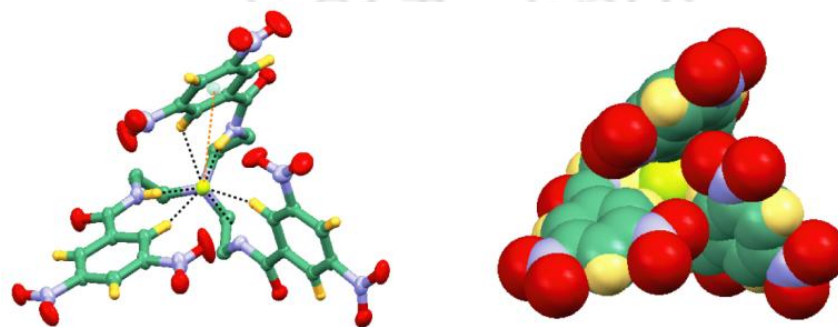


Figure A4.11 X-ray structure (thermal ellipsoid plot with 30% probability and space fill model) of F^- -encapsulated complex $\text{TBA}[\text{L}_3(\text{F})]2\text{DMSO}\cdot 4\text{H}_2\text{O}$ (**3a-IV**) showing binding and encapsulation of F^- within the receptor cavity. TBA cation and solvent molecule(s) are omitted for clarity in each case.

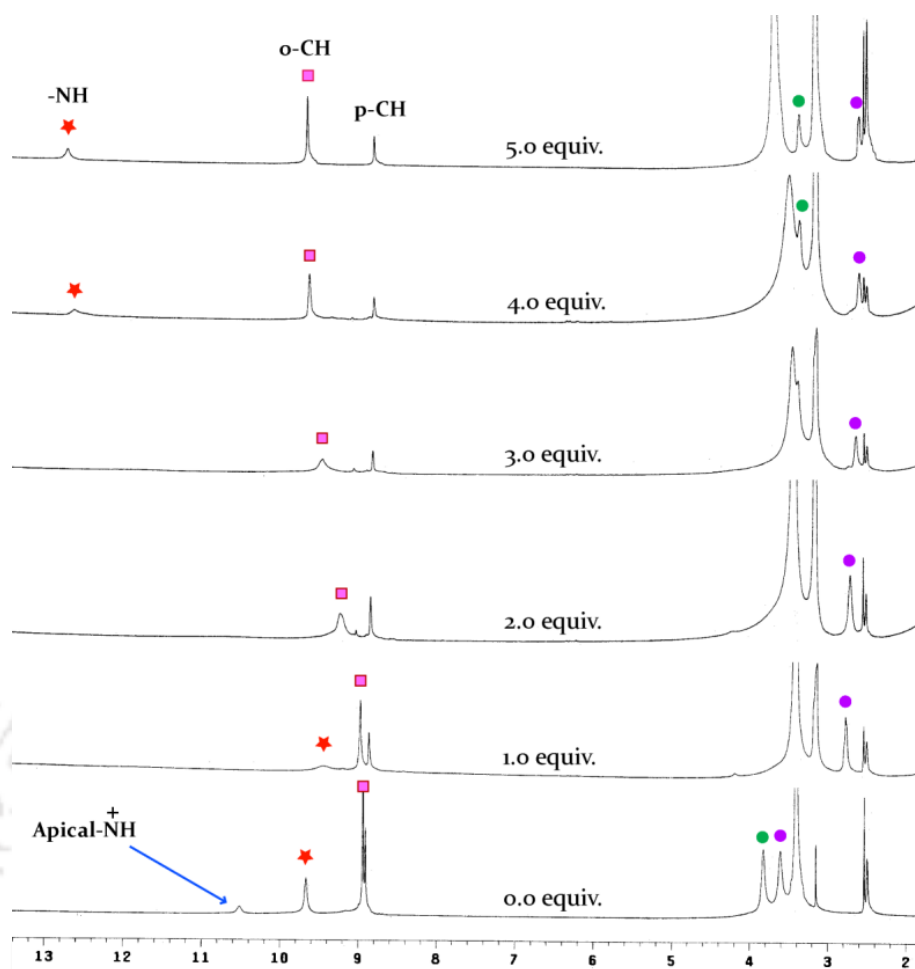


Figure A4.12 Partial ^1H NMR spectra showing the titration of complex **3c** (CDCl_3) with increasing equivalents of TBAF.

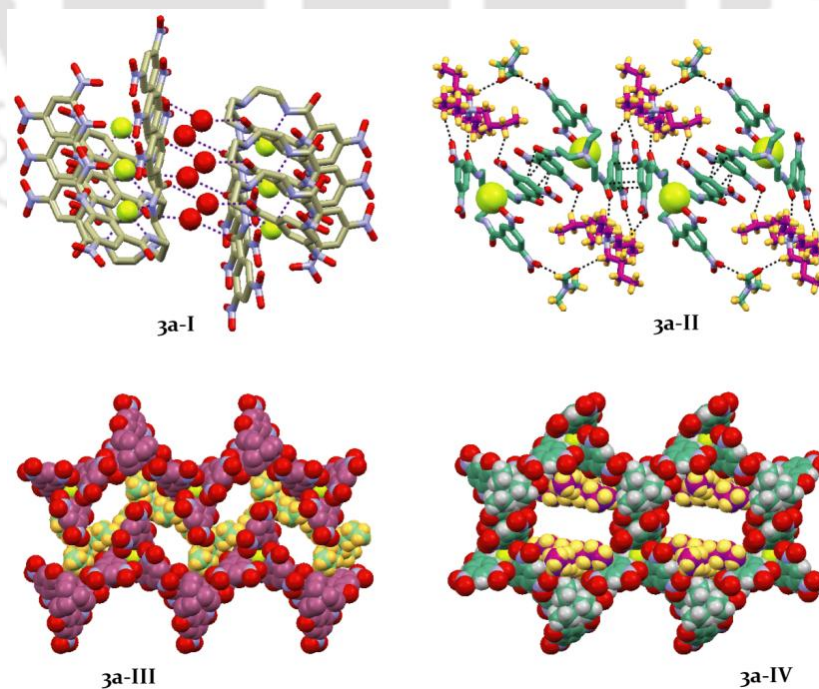
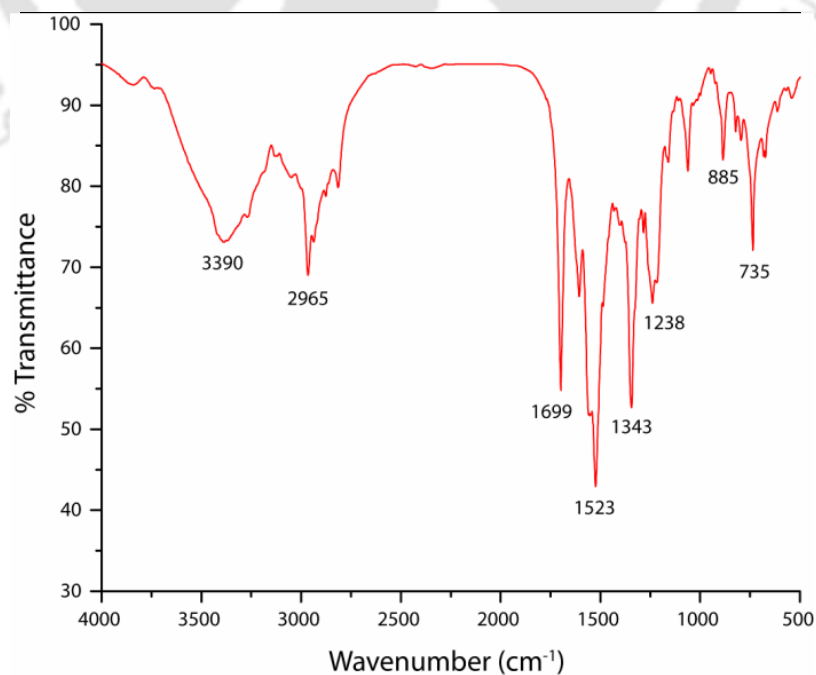


Figure A4.13 Crystal packing motifs of solvatomorphs of F^- -encapsulated complex, **3a I-IV**.

Annexure 5

Table A5.1 Hydrogen bonding interactions involved in the crystal structure of **L₄**.

| D-H···O | $d(\text{H}\cdots\text{O})/\text{\AA}$ | $d(\text{D}\cdots\text{O})/\text{\AA}$ | $\angle\text{D-H}\cdots\text{O}/^\circ$ |
|----------------|--|--|---|
| N2-H···O7 | 2.15(2) | 2.948(3) | 154(1) |
| N3-H···O7 | 2.04(1) | 2.850(2) | 156(1) |
| N12-H···O16 | 2.18(2) | 2.973(3) | 151(1) |
| N13-H···O16 | 2.00(1) | 2.822(2) | 159(1) |
| N5-H···O1 | 2.35(2) | 3.018(3) | 138(1) |
| N6-H···O1 | 2.10(2) | 2.910(3) | 156(1) |
| N8-H···O4 | 2.15(1) | 2.942(2) | 152(1) |
| N9-H···O4 | 2.14(1) | 2.943(2) | 154(1) |
| N15-H···O10 | 2.28(2) | 2.974(2) | 137(1) |
| N16-H···O10 | 2.04(1) | 2.862(2) | 159(1) |
| N18-H···O13 | 2.16(1) | 2.961(2) | 154(1) |
| N19-H···O13 | 2.23(1) | 3.030(2) | 153(1) |
| C1-H(A)···O12 | 2.71(2) | 3.631(3) | 157(1) |
| C20-H(A)···O12 | 2.66(2) | 3.491(3) | 143(1) |
| C35-H···O3 | 2.57(3) | 3.338(4) | 139(2) |
| C53-H···O9 | 2.66(2) | 3.388(3) | 135(2) |
| C34-H···C1g | 3.21 | 4.027 | 147 |

**Figure A.5.1** FT-IR spectrum of complex **4a** recorded in KBr pellet.

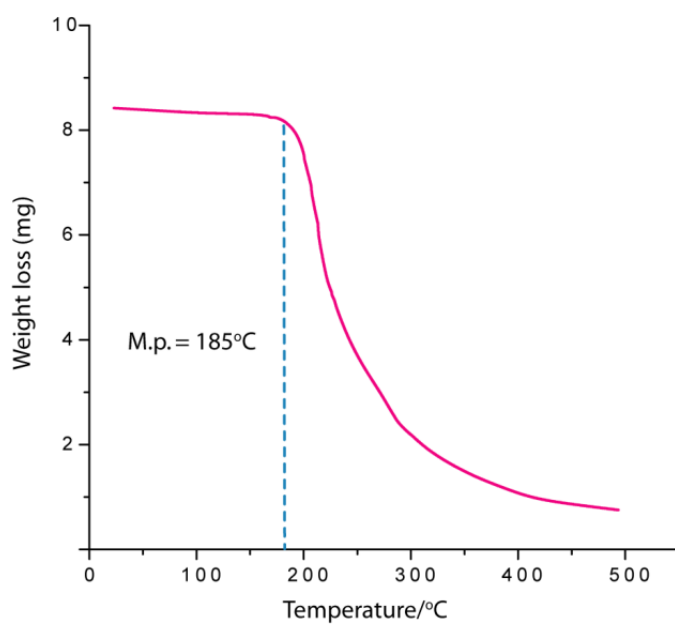


Figure A5.2 Thermo gravimetric (TGA) curve of complex **4a** obtained at a heating rate of 5°C/min in N₂ atmosphere.

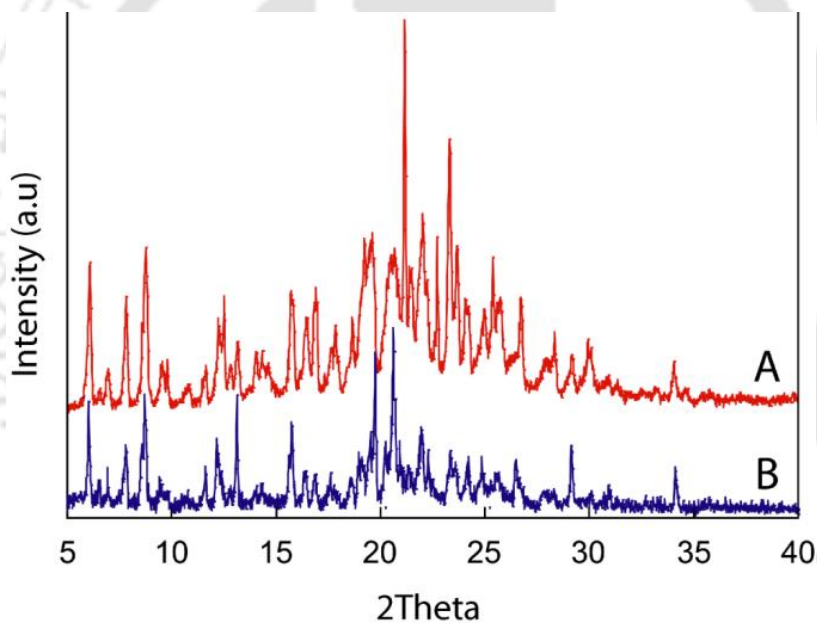


Figure A5.3 Powder X-ray diffraction patterns of isolated crystals of **4a** from (A) the receptor-OH solution and (B) receptor-F⁻ solution.

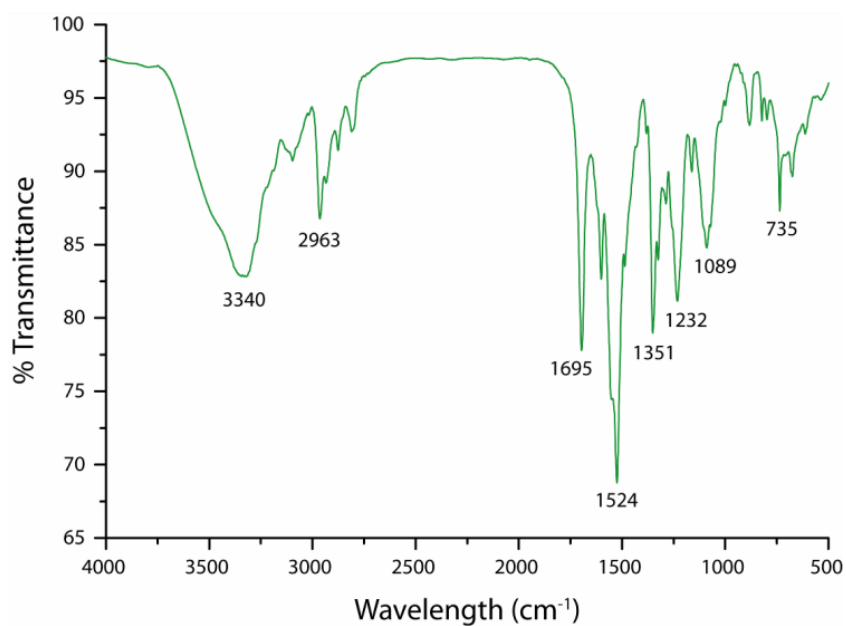


Figure A.5.4 FT-IR spectrum of complex **4b** recorded in KBr pellet.

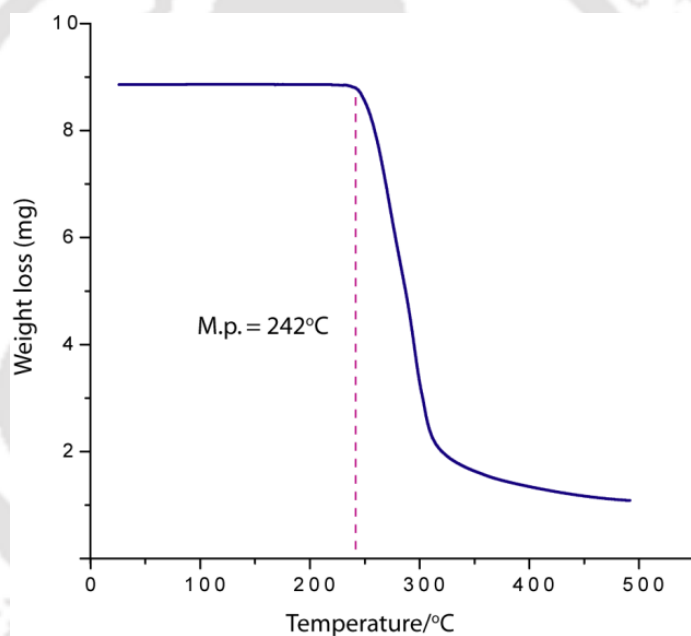


Figure A5.5 Thermo gravimetric (TGA) curve of complex **4b** obtained at a heating rate of 5°C/min in N₂ atmosphere.

Annexure 6

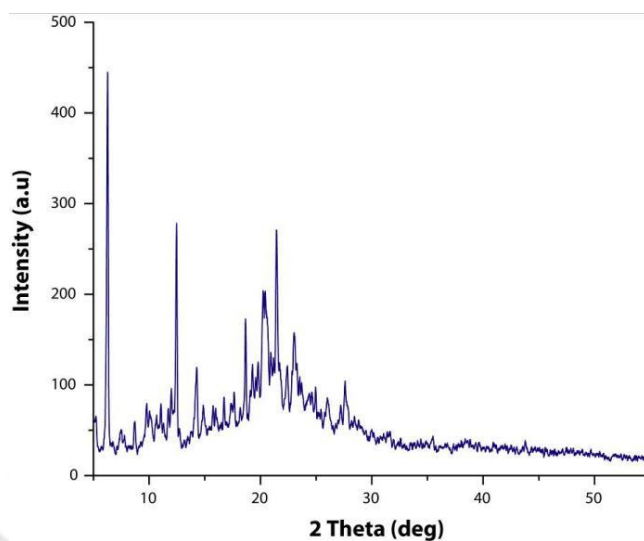


Figure A6.1 Powder XRD patterns of complex **5b-I** recorded with dried crystalline powders.

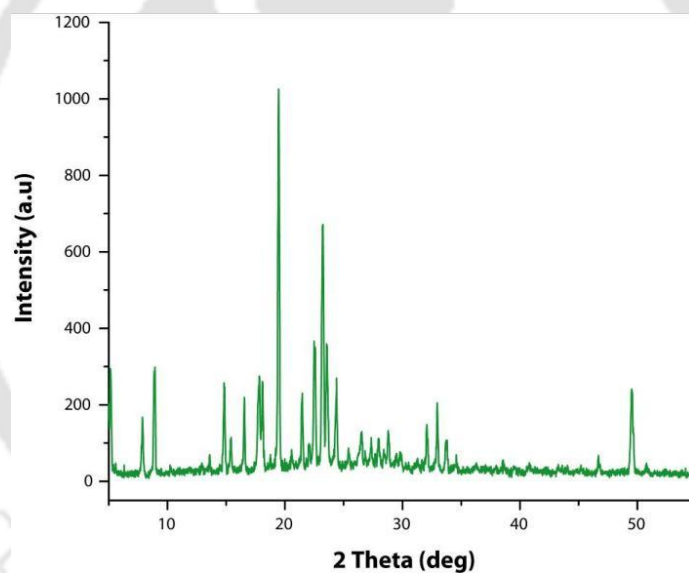


Figure A6.2 Powder XRD patterns of complex **5b-II** recorded with dried crystalline powders.

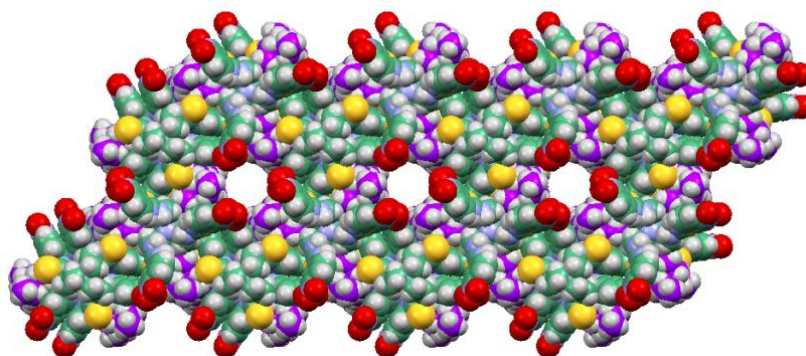


Figure A6.3 Crystal packing diagram of complex **5b-II**, showing the cylindrical voids of 568 \AA^3 (view down the crystallographic c-axis).

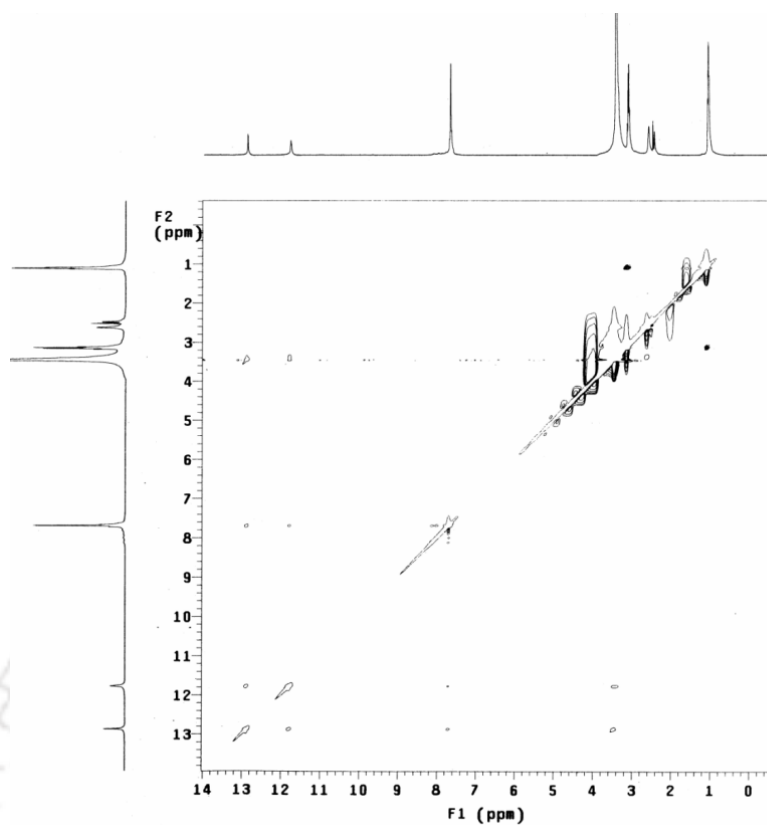


Figure A6.4 2D-NOESY NMR spectrum of complex **5b-II** in DMSO- d_6 .

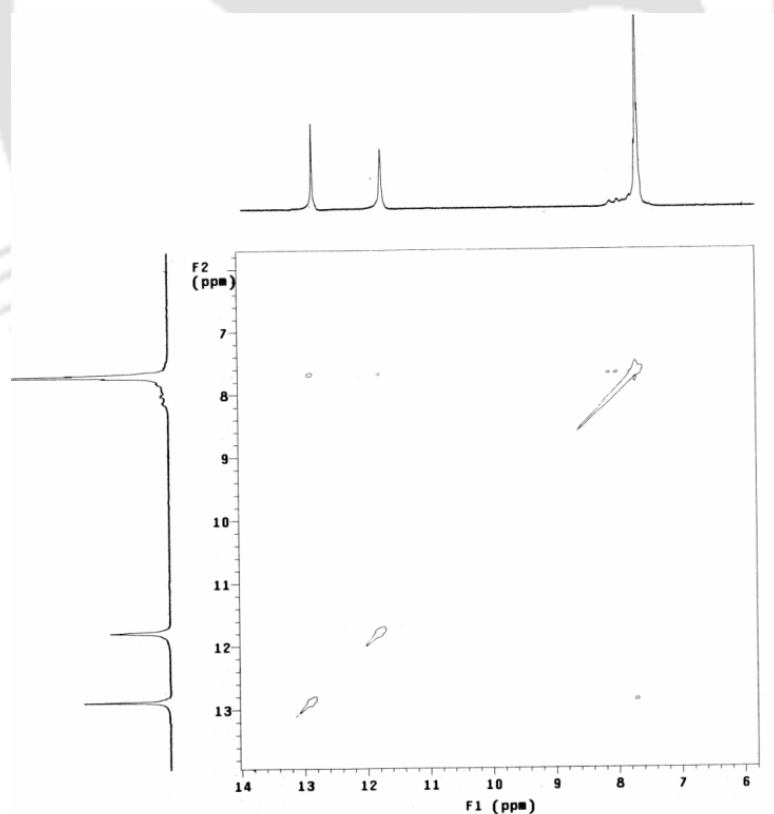


Figure A6.5 Partial (aromatic region) 2D-NOESY NMR spectrum of complex **5b-II** in DMSO- d_6 , in presence of 0.5 equivalent 2:1 mixture of TBA(OH) and TBA(H_2PO_4).

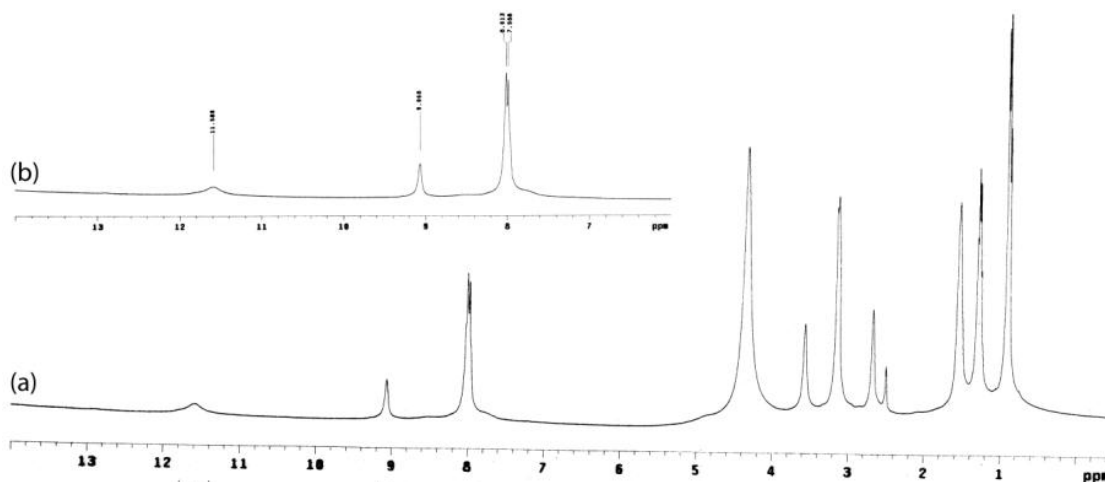


Figure A6.6 (a) ^1H NMR spectrum of L_5 in presence of 1 equivalent of $\text{TBAH}_2\text{PO}_4^-$ recorded after overnight equilibration in $\text{DMSO}-d_6$; (b) Partial ^1H NMR spectrum (aromatic region), suggesting that PO_4^{3-} -encapsulated complex does not form at equivalent stoichiometry of H_2PO_4^- .

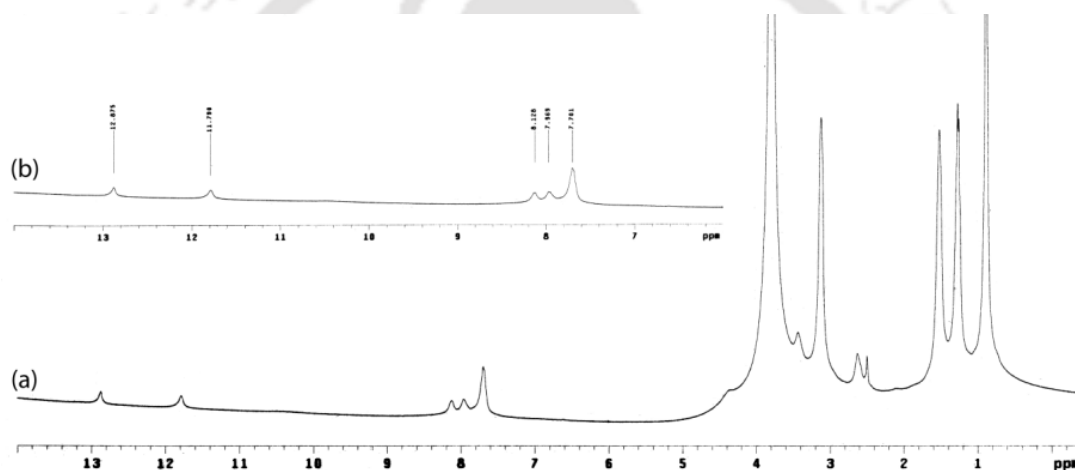


Figure A6.7 (a) ^1H NMR spectrum of L_5 in presence of 3 equivalent of $\text{TBAH}_2\text{PO}_4^-$ recorded after overnight equilibration in $\text{DMSO}-d_6$; (b) Partial ^1H NMR spectrum (aromatic region), suggesting the in situ formation of PO_4^{3-} -encapsulated complex in a comparatively greater percentage than the complex formed between added H_2PO_4^- and L_5 .

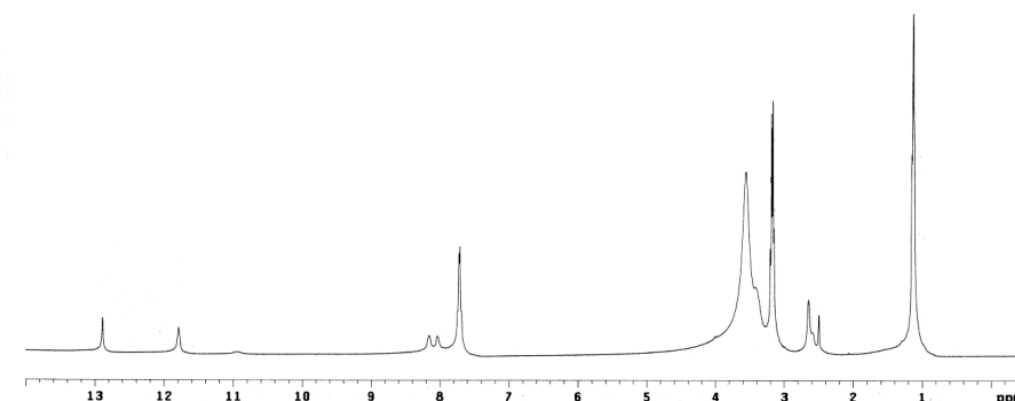


Figure A6.8 ^1H NMR spectrum of complex $\mathbf{5a}$ in presence of excess $\text{TEA}(\text{AcO})$ in $\text{DMSO}-d_6$ showing the selective formation of phosphate-encapsulated complex in solution.

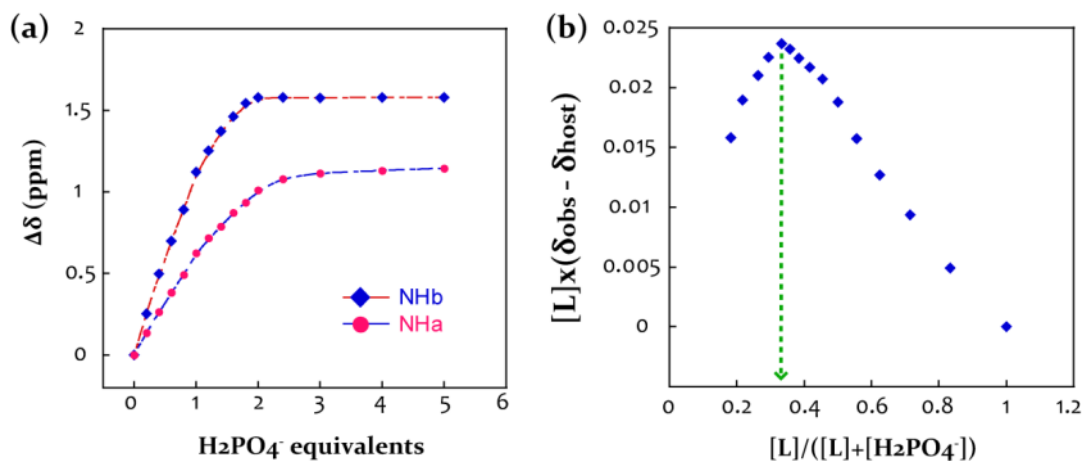


Figure A6.9 (a) Change in chemical shift of $-\text{NH}$ resonances of L_5 (10 mM) with increasing conc. of standard H_2PO_4^- solution (50 mM) in $\text{DMSO}-d_6$; (b) The Job's plot.

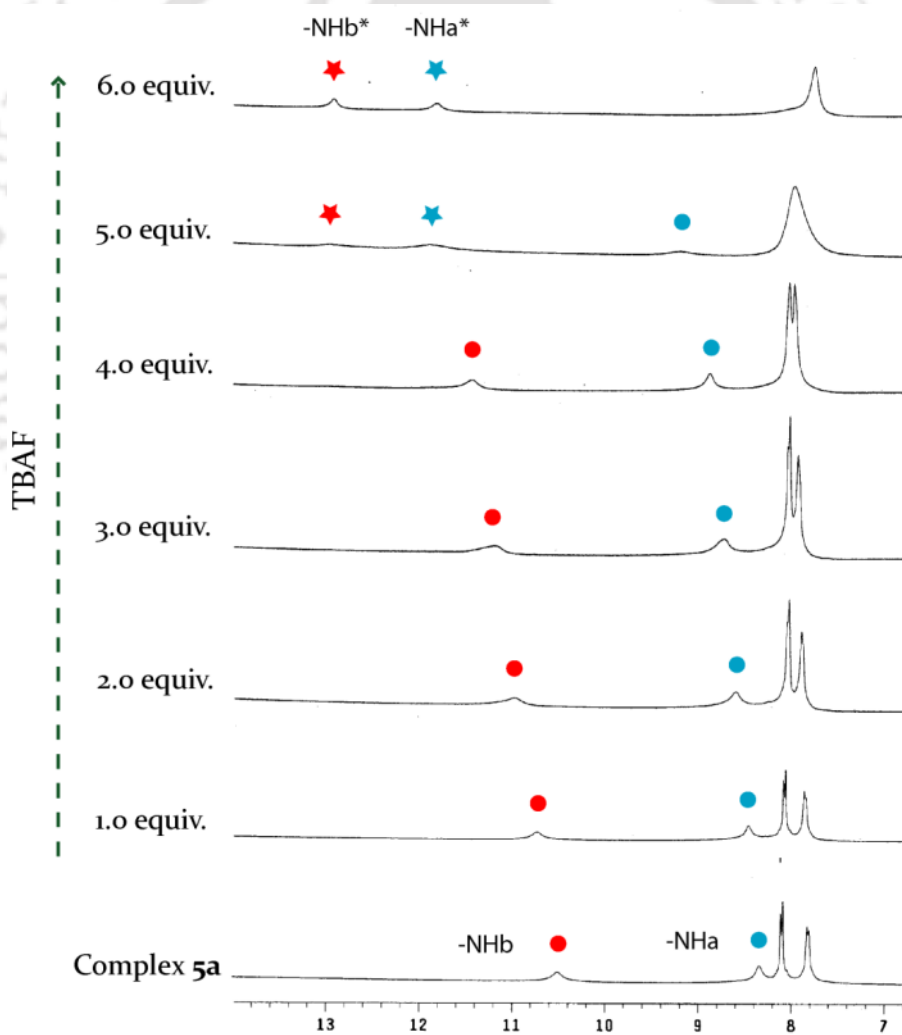


Figure A6.10 Partial ^1H NMR titration spectra of complex **5a** (10 mM) recorded with increasing equivalents of TBAF salt.